Photovoltaic efficiency enhancement of polycrystalline silicon solar cells by a highly stable luminescent film

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ABSTRACT Si-based solar cells have dominated the entire photovoltaic market, but remain suffering from low power conversion efficiency (PCE), partly because of the poor utilization of ultraviolet (UV) light. Europium(III) (Eu^{3+}) complexes with organic ligands are capable of converting UV light into strong visible light, which makes them ideal light converter to increase the efficiency of solar cells. However, the low stability of such complexes seriously hampers their practical applications. In this work, we report a highly stable and luminescent ethylene-vinyl acetate (EVA) copolymer film consisting of a Eu^{3+} complex as a down-shift material, $\text{Eu}(\text{ND})_4\text{CTAC}$ (ND = 4-hydroxy-2-methyl-1,5-naphthyridine-3-carbonitrile, CTAC = hexadecyl trimethyl ammonium chloride), coating of which onto the surface of large area polycrystalline silicon solar cells (active area: 110 cm^2) results in an increase of PCE from 15.06% to 15.57%. Remarkable stability of the luminescent film was also demonstrated under light-soaking test for 500 h, and no obvious luminescence degradation can be observed. The remarkable enhancement of the conversion efficiency by 0.51% absolute on such a large active area, together with the high stability of the luminescent film, demonstrates a prospect for the implementation of the films in photovoltaic industry.

Keywords: ethylene-vinyl acetate, lanthanide complexes, solar cell, stability

INTRODUCTION

Nowadays, fossil fuels are still the main source for covering the energy demands for humanity's economic development. As the increasing energy requirement will

lead to the ultimate depletion of fossil fuels, there is an urgent need for exploring renewable energy sources. Therefore, scientists are turning their attention to solar energy for its inexhaustibility and environmental friendliness [1,2]. The utilization of solar energy includes both photo-thermal conversion and photoelectric conversion, in which photovoltaic cells that employ semiconductor materials to convert light into electricity are a promising candidate for the new dominant energy source in the future [3,4]. Presently, silicon-based solar cells account for 95% of the photovoltaic market. According to the Shockley-Queisser efficiency limitation, the maximum efficiency value of silicon-based solar cells is about 31%. However, the actual power conversion efficiency (PCE) is relatively lower than the theoretical value, limiting their long-term development [5–7]. In general, there are three main factors that limit the energy conversion efficiency: a) Lattice thermalization loss, where high-energy photons are absorbed by solar cells and then produce one electron-hole pair, while the excess energy is dissipated through non-radiative relaxation as heat. b) Sub-bandgap photon loss, where photon energy is too small to cross the energy bandgap threshold energy, so that no electron-hole pairs are generated. c) The loss from electronic properties of the device, such as the loss of recombination of electron-hole pairs from surface, contact voltage loss, and junction loss [5,8]. The first two energy-conversion loss mechanisms are directly related to the spectral mismatch between solar spectrum and bandgap of the silicon-based material. The last loss process depends on the fabrication of solar cell devices. The strategy of spectrum

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modification associated with spectral mismatch like down-conversion, down-shift or up-conversion can optimize the conversion efficiency of solar cells, which also reduces the long-term cost of solar cell construction [9–14].

The technology of down-shift that involves absorbing one higher energy photon and releasing one photon of lower energy is considered as an effective approach to reduce the thermalization loss [6,15]. Usually, a practical and intriguing method to utilize this technology is to embed down-shift materials into polymer matrices such as polyvinyl alcohol, ethylene-vinyl acetate (EVA) and polymethyl methacrylate, after which the polymer matrices are attached on the surface or encapsulated inside of photovoltaic modules for enhancing the conversion efficiency [16–18]. In this context, lanthanide complexes constituted of lanthanide ions and organic ligands, exhibiting favorable luminescent features such as a large Stokes shift, narrow emission bands and longer decay times, are promising down-shift candidates [19,20]. A wide range absorption of organic ligands in the UV region can lead to efficient energy transfer to the lanthanide ions and trigger strong luminescence, simultaneously also attenuating polymer matrix long-term degradation caused by UV irradiation [21–24]. In particular, the dominant emission of Eu^{3+} at 612 nm is very close to the maximum spectral sensitivity of crystalline Si-based solar cells [25–32]. For instance, Fix and co-workers [33] have summarized a series of promising Eu^{3+} complexes as down-shift species that exhibit an improvement of PCE, but the stability issue remains a challenge. In terms of the $[\text{Eu}(\text{tta})_3(\text{tppo})_2]$ complex, it was found to be only stable for several hours upon illumination with 350 nm UV light [34]. The intrinsic disadvantages of lanthanide complexes such as low thermal and photochemical stability, especially photodegradation of the sensitizer upon UV exposure, limit their photovoltaic application. Hence, engineering of lanthanide complexes with impressive stability and high quantum efficiency is paramount for further development of this technology.

Herein, we demonstrate the increase in PCE of polycrystalline silicon solar cells by coating EVA film containing europium(III) complex ($\text{Eu}(\text{ND})_4\text{CTAC}$) [35] on the surface of the devices (active area: 110 cm^2). By down-shifting higher energy photons, the obtained luminescent film (denoted as $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film) can be better to utilize the AM 1.5G spectrum. A series of luminescent films with varying complex concentrations have been compared and studied in photovoltaic characterization. The results show that in the best case, the PCE of poly-

crystalline silicon solar cells increases from 15.06% to 15.57% by using the luminescent film. The obtained luminescent films display excellent photostability, almost keeping the luminescence intensity unchanged even under continuous UV radiation at 30°C for 500 h. The realization of employing stable $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film as a medium to enhance the PCE of Si-based solar cells demonstrated here makes it attractive for photovoltaic industry.

EXPERIMENTAL SECTION

Materials

EVA copolymer (LG ES28005), toluene and dichloromethane were used as received. $\text{Eu}(\text{ND})_4\text{CTAC}$ (ND = 4-hydroxy-2-methyl-1,5-naphthyridine-3-carbonitrile, CTAC = hexadecyl trimethyl ammonium chloride) was obtained from Beijing Boda Green Technology Co., Ltd. and the structure was confirmed by elemental analysis and ^1H NMR spectrum as shown in Table S1 and Fig. S1. Si-based solar cells were provided by Xinao Photovoltaic Energy Co., Ltd.

Preparation of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films

Dichloromethane solutions of $\text{Eu}(\text{ND})_4\text{CTAC}$ complexes (7.2–54 mg) at different concentrations were prepared. 3.6 g of EVA pellets was dissolved in toluene (24 g) at 60°C for 2 h with stirring. The aforementioned solutions were separately mixed in an ultrasonic bath, after which the solution was poured into a Petri dish (170 mm) and dried at 80°C for several hours, resulting in luminescent polymer composite films. Pure EVA film was obtained using the same method as described above, without the addition of $\text{Eu}(\text{ND})_4\text{CTAC}$ complex. Subsequently, the weight% relative value was used to indicate the concentration of $\text{Eu}(\text{ND})_4\text{CTAC}$ complex relative to EVA (0.2%, 0.5%, 0.7%, 1% and 1.5%).

Characterizations

^1H NMR spectrum was recorded in CDCl_3 using a Bruker ARX 400 spectrometer. Elemental analyses (C, H and N) were measured with an Elementary Vario EI. The fluorescence data were obtained by using an Edinburgh Instruments FS920P spectrophotometer. Scanning electron microscopy (SEM) measurement was carried out with an S-4800 at a voltage of 3 kV. Light transmittance of the films and UV absorbance spectra of the samples were measured with an Agilent Cary 100 UV-Vis spectrometer. The thermal stability was studied by thermogravimetric analysis by using a SDT Q600 TGA in ambient air at-

mosphere at a heating rate of $10^{\circ}\text{C min}^{-1}$. Photo stability was studied by a UV aging test under $160 \mu\text{W cm}^{-2}$ irradiance at 30°C for 500 h. For the characterization of the photovoltaic solar cell performance, different EVA/complex films were directly coated on top of a Si-based solar cell (active area: 110 cm^2). External quantum efficiency (EQE) measurement and photocurrent-voltage characteristics were recorded by using a QEX10 PV measurement system and a Newport IV30 solar simulator at AM 1.5G conditions, respectively.

RESULTS AND DISCUSSION

The structure of the complex $\text{Eu}(\text{ND})_4\text{CTAC}$ is shown in Scheme 1, where each Eu atom is linked to four ND ligands and one hexadecyltrimethylammonium ion. Transparent $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films containing different concentrations of the $\text{Eu}(\text{ND})_4\text{CTAC}$ complex (0.2%, 0.5%, 0.7%, 1% and 1.5%) were prepared by the solution-casting approach [25]. Selecting EVA as the matrix for down-shift materials is motivated by its strong adhesion and excellent light transparency in relation to glass. In fact, EVA usually plays the role of sealer material in the whole process of encapsulating Si-based solar

Scheme 1 Structural formula of the Eu^{3+} -compound used in this study.

modules in photovoltaic industry, which provides an opportunity to facilitate the application at production scale [36].

As the first step, the photoluminescent properties of the resulting films were studied. As shown in Fig. 1a, the emission spectrum of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film, excited at 356 nm, is composed of five sharp bands at 579, 590, 613, 652 and 702 nm, ascribed to the $^5\text{D}_0 \rightarrow ^7\text{F}_J$ ($J = 0-4$) transitions, respectively. The presence of the $^5\text{D}_0 \rightarrow ^7\text{F}_0$ transition indicates a low point symmetry of the surroundings of the Eu^{3+} ion (C_{nv} , C_n or C_s symmetry). On the other hand, the low intensity of the $^5\text{D}_0 \rightarrow ^7\text{F}_3$ transition, which is forbidden as an electric dipole and

Figure 1 (a) Excitation and emission spectra of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film (0.7%). (b) Emission spectra of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films with different concentrations of $\text{Eu}(\text{ND})_4\text{CTAC}$ complex. Digital photographs of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film (0.7%) under daylight (c) and UV irradiation at 365 nm (d).

magnetic dipole transition, indicates a small participation of J-mixing in the intensity of f-f transitions in this system. The main band dominated by the $^5D_0 \rightarrow ^7F_2$ transition at 613 nm is responsible for the strong red luminescence [37]. The excitation spectrum obtained by monitoring the absorbance under the 613 nm wavelength exhibits a broad band in the region from 265 to 388 nm attributed to the absorption of the ND ligand, which is almost overlapped with the $n-\pi^*$ transition present in the absorption spectrum (Fig. 2 and Fig. S2). This observation confirms the occurrence of energy transfer from the ND ligand to Eu^{3+} . In addition, the emission spectra of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films with different $\text{Eu}(\text{ND})_4\text{CTAC}$ concentrations from 0.2% to 1.5% were measured as shown in Fig. 1b, revealing that the luminescence intensity increases along with the concentration of the $\text{Eu}(\text{ND})_4\text{CTAC}$ complex. As shown in Fig. S3, the lifetime of the Eu^{3+} excited state for $\text{Eu}(\text{ND})_4\text{CTAC}$ complex is determined to be 0.98 ms, and that for $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film (0.7%) is 0.96 ms. Interestingly, the resultant films exhibit obviously red light even under daylight (Fig. 1c), while very strong luminescence is also observed upon illumination with 365 nm light source (Fig. 1d), which implies that this luminescent film shows promising prospect as a light-converting material in enhancing the PCE of Si-based solar cells.

The absorption spectra shown in Fig. 2 reveal that light transmittance of the film with $\text{Eu}(\text{ND})_4\text{CTAC}$ complex (0.2%) is higher than 82% in the visible light region. In addition, the transparency of the films slightly decreases in the visible region with the increase of complex amount. However, the transparency of luminescence films drastically declines in both the UV and visible light regions after the doping concentration exceeds 0.7%, which is attributed to excessive aggregation of $\text{Eu}(\text{ND})_4\text{CTAC}$ in EVA matrix. The morphologies of a series of films were studied by SEM. As shown in Fig. S4a–f, the surface morphology of the EVA film is almost uniform. When the EVA film is doped with $\text{Eu}(\text{ND})_4\text{CTAC}$ complexes (0.2%, 0.5% and 0.7%), there are many tiny and uneven protrusions on the surface. However, these protrusions' distribution becomes tighter to form agglomeration zones after the doping concentration of $\text{Eu}(\text{ND})_4\text{CTAC}$ complex exceeds 0.7%, which suggests that increasing the doping concentration of $\text{Eu}(\text{ND})_4\text{CTAC}$ complex easily promotes the $\text{Eu}(\text{ND})_4\text{CTAC}$ complex to gather together and results in poor compatibility between the $\text{Eu}(\text{ND})_4\text{CTAC}$ complex and EVA matrix. This result is in consistent with the transmittance spectrum. The thickness of the film was estimated at about 141.6 μm (Fig. S4g).

Figure 2 UV-vis transmittance spectra of pure EVA and $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films with different concentrations of $\text{Eu}(\text{ND})_4\text{CTAC}$.

Moreover, we also characterized the EDS mapping in order to further demonstrate the uniformity of $\text{Eu}(\text{ND})_4\text{CTAC}$ complex in EVA matrix (Fig. S4h). Considering the luminescence intensity and transparency of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films, the 0.7% $\text{Eu}(\text{ND})_4\text{CTAC}$ complex in EVA matrix can be considered as the optimal choice for application in Si-based solar cells.

In order to confirm the beneficial effect of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films on the photovoltaic response characteristic of polycrystalline silicon solar cells, the EQE measurement was carried out in the wavelength range from 300 to 1100 nm. EQE is defined as the ratio between the number of charge carriers of the solar cell and the number of photons on its surface at each wavelength [16]. Measurements were taken after covering the photovoltaic module with a $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film, as demonstrated in Fig. 3. In Fig. 4a, the EQE curves of a bare cell, and the cell covered with pure EVA film and $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film are shown.

Figure 3 Si-based solar cell module coated with the luminescent film.

Figure 4 (a) The EQE curves of the bare cell, cell covered with pure EVA film and Eu(ND)₄CTAC/EVA films with different complex concentrations. (b) Current-voltage curves of the bare cell, cells covered with pure EVA film and Eu(ND)₄CTAC/EVA films with increasing complex concentration.

CTAC/EVA films with varying complex concentrations (from 0.2% to 1.5%) are studied. For the setups with the cell uncoated and coated with pure EVA film, the EQE profiles are very similar, illustrating that the EQE effect of pure EVA film on the cell is negligible. However, the presence of the Eu(ND)₄CTAC complex leads to an obvious EQE enhancement in the UV spectrum range from 300 to 370 nm, which is ascribed to the photons undergoing down-shift *via* the lanthanide complex incorporated within the EVA films. On the other hand, the EQE of the cells coated with Eu(ND)₄CTAC/EVA films in the visible region decreases gradually with the increase of the doping concentration from 0.2% to 1.5%, with the effect possibly attributed to the photon absorption loss in visible wavelength range caused by the overall transparency decrease of the luminescent films [33,38].

The photocurrent-voltage characteristic test was carried out under an AM 1.5G solar illumination at 100 mW cm⁻² (1 sun) as shown in Fig. 4b. The detailed photovoltaic parameters such as short-circuit current density (J_{SC}), open-circuit photovoltage (V_{OC}), fill factor (FF) and PCE derived from these curves are summarized

in Table 1. All acquired data result from averaging the outcomes of 5 measurements each, to ensure the reliability of the obtained values. It is observed that the PCE displays an improvement (from 15.06% to 15.14%) after coating with a pure EVA film in comparison with the bare cell, which mainly arises from the decrease of light reflection and some small error [33,39]. For the Eu(ND)₄CTAC/EVA films, the PCE prominently increases with the increase of doping concentration of the complex reaching a maximum at 0.7%, which is then followed by a decrease (Fig. S5b). Furthermore, the J_{SC} values show the similar variation trend (Fig. S5a). The decrease of performance might be caused by the fact that the severe diminishing transparency of the films (1.0% and 1.5%) could obstruct the absorption of solar cells in the visible region, leading to the decrease of PCE, despite their increased EQE response in the ultraviolet region. As such, the optimal performance of a solar cell coated with the luminescent film is achieved when the concentration of the complex in the EVA film equals 0.7%, in which case the value of PCE is increased from 15.06% to 15.57%, with a relative increase by 3.42%.

Table 1 Photovoltaic parameters of a bare cell, cells covered with pure EVA film and Eu(ND)₄CTAC/EVA films with different complex concentrations

Silicon solar cell with and without Eu(ND) ₄ CTAC coating	V_{OC} (V)	J_{SC} (mA cm ⁻²)	FF (%)	PCE (%)
Bare solar cell	0.599±0.001	34.14±0.02	73.77±0.05	15.06±0.01
Pure EVA film coating	0.601±0.001	34.16±0.09	73.93±0.11	15.14±0.02
0.2% Eu(ND) ₄ CTAC in EVA	0.606±0.001	34.18±0.05	74.16±0.04	15.33±0.01
0.5% Eu(ND) ₄ CTAC in EVA	0.608±0.001	34.36±0.08	74.23±0.08	15.47±0.02
0.7% Eu(ND) ₄ CTAC in EVA	0.610±0.001	34.47±0.05	74.22±0.08	15.57±0.02
1.0% Eu(ND) ₄ CTAC in EVA	0.608±0.001	34.43±0.09	74.21±0.05	15.51±0.01
1.5% Eu(ND) ₄ CTAC in EVA	0.605±0.002	34.26±0.14	74.09±0.10	15.32±0.01

In addition to the excellent performance in terms of luminescent properties, high stability is highly desired for luminescent films to be successfully used in enhancing the efficiency of solar cells. Interestingly, the luminescent films investigated in this work show high thermal stability and remarkable photostability as shown in Fig. 5 and Fig. S6, respectively. From Fig. S6, it can be found that the thermal stability of the $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film is higher than that of the pure EVA film and the main weight loss starts to occur at around 350°C . The presence of the $\text{Eu}(\text{ND})_4\text{CTAC}$ complex can apparently improve the thermostability of the EVA film. Fig. 5 reveals that the emission intensity of 0.7% $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film remains unchanged even after the film is subjected to a long-term accelerated aging at 30°C for 500 h. No obvious changes in the quantum yield can be observed, with its value remaining constantly at about 52%, unaffected by the long-term accelerated aging treatment. Moreover, the transparency of $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film is also well maintained after the long-term UV light soaking test, as revealed in Fig. S7. These results demonstrate that $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ films possess the remarkable stability, namely strong resistance to photodegradation and high thermal stability.

CONCLUSION

In conclusion, a feasible method of increasing the PCE of polycrystalline silicon solar cells through spectra modification technology, achieved by employing a down-shift material ($\text{Eu}(\text{ND})_4\text{CTAC}$ complex) embedded in EVA matrix, was demonstrated. In the optimal case (0.7% concentration of the complex in EVA matrix), the PCE of

the polycrystalline silicon solar cells coated with the luminescent film was improved from 15.06% to 15.57%. Owing to the remarkable anti-photobleaching properties of the $\text{Eu}(\text{ND})_4\text{CTAC}$ complex, the $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film displayed a highly satisfactory stability under the 500 h light soaking measurement. Therefore, the combination of favorable factors in the form of PCE enhancement and anti-photodegradation properties makes the $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film an attractive candidate for the development of low-cost and highly efficient photovoltaic devices.

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Figure 5 Emission spectra of the $\text{Eu}(\text{ND})_4\text{CTAC}/\text{EVA}$ film (0.7% concentration) before and after accelerated UV aging.

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Conflict of interest The authors declare no conflict of interest.

Supplementary information Supporting data are available in the online version of the paper.

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利用高度稳定的发光薄膜提高多晶硅太阳能电池的光伏效率

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摘要 硅基太阳能电池已经主导了整个光伏市场,但是仍然面临着光电转化效率低的问题,其中部分原因是其对紫外线的利用率较低。稀土铕配合物能够将紫外光转化为可见光,有望提高硅基太阳能电池的光电转化效率。然而,这类配合物较低稳定性限制了它们的实际应用。本文中,我们制备了一种高度稳定的EVA/Eu(ND)₄-CTAC发光薄膜,将其覆盖在大尺寸的多晶硅太阳能电池表面(110 cm²)可以使得光电转化效率从15.06%提高到15.57%。在500 h的加速老化实验中荧光性能几乎没有下降,证明了发光薄膜的超强稳定性。在如此大的有效面积上,发光薄膜使硅基太阳能电池的转换效率提高0.51%的绝对值,同时实现超高的稳定性,说明该发光膜在光伏工业上具有广阔的应用前景。