Editors

Prof. Dr. Gaston Berthier
Université de Paris
Institut de Biologie
Physico-Chimique
Fondation Edmond de Rothschild
13, rue Pierre et Marie Curie
F–75005 Paris

Prof. Dr. Michael J. S. Dewar
Department of Chemistry
The University of Texas
Austin, Texas 78712/USA

Prof. Dr. Hanns Fischer
Physikalisch-Chemisches Institut
der Universität Zürich
Rämistr. 76
CH–8001 Zürich

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Kyoto University
Dept. of Hydrocarbon Chemistry
Kyoto/Japan

Prof. Dr. George G. Hall
Department of Mathematics
The University of Nottingham
University Park
Nottingham NG7 2RD/Great Britain

Prof. Dr. Jürgen Hinze
Fakultät für Chemie
Universität Bielefeld
Postfach 8640
D-4800 Bielefeld

Prof. Dr. Hans H. Jaffé
Department of Chemistry
University of Cincinnati
Cincinnati, Ohio 45221/USA

Prof. Joshua Jortner
Institute of Chemistry
Tel-Aviv University
61390 Ramat-Aviv
Tel-Aviv/Israel

Prof. Dr. Werner Kutzelnigg
Lehrstuhl für Theoretische Chemie
der Universität Bochum
Postfach 102148
D-4630 Bochum 1

Prof. Dr. Klaus Ruedenberg
Department of Chemistry
Iowa State University
Ames, Iowa 50010/USA

Prof. Jacopo Tomasi
Dipartimento di Chimica e
Chimica Industriale
Universitá di Pisa
Via Risorgimento, 35
I-Pisa

Lecture Notes in Chemistry

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P. Vanýsek

Electrochemistry on Liquid/Liquid Interfaces

Springer-Verlag
Berlin Heidelberg New York Tokyo

Author

P. Vanýsek
Northern Illinois University, Department of Chemistry
DeKalb, IL 60115, USA

ISBN-13: 978-3-540-15677-2 e-ISBN-13: 978-3-642-48910-5
DOI: 10.1007/978-3-642-48910-5

2152/3140-543210

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P. Vanýsek

Northern Illinois Univeristy, The Michael Faraday Laboratories
Department of Chemistry, De Kalb, Illinois 60115, U.S.A.*

CONTENTS

* The manuscript was prepared at the Chemistry Department of the University of New Hampshire, Durham, NH 03824. Author gratefully acknowledges support received from the department and extends his thanks for the use of a word processor and other facilities during the past year.

List of symbols

a	activity
A	area
AC	alternating current
b	backward, bulk (in subscript)
Bu	butyl
c	concentration
C	capacitance
D	diffusion coefficient
DC	direct current
E	potential, energy
Et	ethyl
f	forward, frequency
F	Faraday constant
G	Gibbs energy of transfer
i	ion, current
I	current, imaginary (in subscript)
IR	voltage loss during current flow
ITIES	interface between two immiscible electrolyte solutions
j	current density, flux
k	rate constant
K	equilibrium constant
L	inductance
L/L	liquid/liquid
m	convolution integral
Me	methyl
n	nonaqueous, number of exchanged electrons
nb	nitrobenzene
o	oil (nonaqueous)
P	preexponential factor, phenyl
Pe	pentyl
Pr	propyl
PTC	phase transfer catalysis
PZC	potential of zero charge
r	radius
R	resistance, real (in subscript), gas constant
t	time
tr	transfer
T	absolute temperature
TBA	tetrabutylammonium
TEA	tetraethylammonium
TMA	(also TMeA) tetramethylammonium
TPAs	tetraphenylarsonium
TPB	tetraphenylborate
U	potential
V	potential
w	water, aqueous phase
z	charge on transporting ion
Z	impedance
α	charge transfer coefficient
β	$1 - \alpha$
Δ	difference
ε	permitivity
μ	chemical potential
$\tilde{\mu}$	electrochemical potential
τ	transition time, time constant
ϕ	Galvani (inner) potential, electrostatic potential, phase angle
ω	circular frequency