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The chemical case for Mercury mantle stripping



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Abstract

Mercury, the Solar System's innermost planet, has an unusually massive core prompting speculation that the planet lost silicate after it formed. Using the unusually high sulfur and low iron composition of its surface and space geodetic constraints on its core composition, we show Mercury's chemistry to be compatible with formation in a larger planet at minimum 1.4–2.5 times Mercury's present mass and possibly 2–4 times its mass by similarity with other rocky Solar System bodies. To do this, we apply an experimentally determined metal-silicate partitioning model for sulfur to Mercury's silicate. The model is validated by applying it to Vesta, which, when evaluated at the conditions of Vestan self-differentiation, yields sulfur contents in its silicate in the range of HED meteorites. Mercury could have lost a substantial fraction of its rocky material through impacts or by being itself a remnant impactor. Independent of any stripping, because a significant amount of silicon resides in Mercury's core, silicate meteoritic debris from Mercury would likely be characterized by ³⁰Si isotopic enrichment > + 0.10% relative to parent sources that could aid identification of a new meteorite class.

Keywords: Mercury, Vesta, Sulfur, Silicon, N-body simulation, Meteorites, Stable isotopes

Introduction

Mercury is essentially a metal core draped with a thin silicate mantle. Detailed models of its interior structure (Rivoldini et al. 2009; Rivoldini and Van Hoolst 2013; Hauck II et al. 2013; Dumberry and Rivoldini 2015) indicate core radii of 2000-2060 km and a core proportionately 58–70% of the planetary mass (M_{igodot} , 3.302 $imes 10^{23}$ kg (Rivoldini et al. 2009)). By comparison, Earth's core is 32% of the planetary mass (Stacey 1992), Venus 24-32% (Fegley 2014), Mars' and Vesta's cores are 21 and 18%, respectively, of the planetary mass (Rivoldini et al. 2011; Russell et al. 2012), and the fractional mass of metallic elements in chondritic meteorites is $\sim 30\%$ (McDonough and Sun 1995). These facts suggest that Mercury might have lost part of its original mantle at some point in the Solar System's history through impacts with other bodies, stripping it of its silicate and leaving metal (Benz et al. 1988; Benz et al. 2007; Asphaug and Reufer 2014). Moreover, spectroscopic observations of Mercury's surface by

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the MESSENGER mission (Nittler et al. 2011; Zolotov et al. 2013; Weider et al. 2014) indicate high concentrations of sulfur (1-6 wt%) and low concentrations of FeO (< 1.9 wt%). From a planetological perspective, this rather weird chemical imprint compared to either the Earth or to CI carbonaceous chondrites (Earth: \sim 0.03 wt% S, \sim 8 wt% FeO; CI 5.4 wt% S, 15.1 wt% FeO (McDonough and Sun 1995)) suggests that it may have arisen under unusual conditions. Chabot et al. (2014) showed that it is difficult to reconcile the observed surface FeO content with equilibrium between silicate and metal at pressures up to that of Mercury's present core-mantle boundary (CMB). Other studies showed plausible ways to build Mercury from chondritic and non-chondritic precursors (Ebel et al. 2018; Cartier and Wood 2019). Here we approach the problem via accretion modeling. Accretion models for the Earth and other rocky planets use experimentally determined metal/silicate partition coefficients to constrain the physical conditions during formation (Li and Agee 1996; Wood et al. 2006; Rubie et al. 2015). These models assume that the composition of the metal and silicate is set by equilibrium partitioning of elements at the rheologically solid base of liquid silicate when a planet's surface is partly or completely molten due to impact

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heating (Kaula 1979; Rubie et al. 2015). Under these conditions, the liquid metal separates from the liquid silicate and accumulates in the core of the planet, ceasing to be in chemical equilibrium with silicate. Consequently, we use a model for sulfur, silicon, oxygen, carbon, and iron partitioning between metal and silicate to evaluate whether there are feasible conditions for Mercury that simultaneously satisfy the composition of the silicate mantle while yielding core compositions that satisfy space geodetic constraints. Our objective is to show that Mercury's surface chemistry is compatible with the formation in an originally larger body from which the mass was lost by impact processing.

Methods

Sulfur partitioning model

We use Boujibar et al.'s (2014) S partitioning model to determine the pressure- (P) and temperature- (T) dependent joint S and FeO content of the silicate. This reads,

$$\begin{split} \log_{10} D_{\rm S}^{\rm met/sil} &= \log_{10} \left[X_{\rm S}^{\rm met}/X_{\rm S}^{\rm sil} \right] = \\ \log_{10} W_{\rm FeO}^{\rm sil} - \log_{10} C_{\rm s} + b/T + cP/T \\ &+ d \log_{10} (1 - X_{\rm Si}^{\rm met}) + e \log_{10}{}^{2} (1 - X_{\rm Si}^{\rm met}) + f \log_{10}{}^{3} (1 - X_{\rm Si}^{\rm met}) \\ &+ g \log_{10} (1 - X_{\rm C}^{\rm met}) + h \log_{10} (1 - X_{\rm Fe}^{\rm met}) + i \log_{10} (1 - X_{\rm Ni}^{\rm met}) \\ &+ j \log_{10} (1 - X_{\rm O}^{\rm met}) + k \quad . \end{split}$$

$$(1)$$

b - k are experimentally determined model coefficients, listed in Table 1. The metal concentrations X_i^{met} and X_S^{sil} are weight fractions, whereas silicate FeO concentration $W_{\text{FeO}}^{\text{sil}}$ is weight percent. C_S is the sulfur carrying capacity, given by

$$\log_{10}C_{\rm S} = -5.704 + 3.15 \times X_{\rm FeO} + 2.65 \times X_{\rm CaO} + 0.12 \times X_{\rm MgO} + 0.77 \times X_{\rm TiO_2} + 0.75 \times (X_{\rm Na_2O} + X_{\rm K_2O}) \quad .$$
(2)

Here, by contrast with (1), the oxide concentrations X_i are mole fractions in the silicate.

In order to evaluate these expressions, we need the composition of core metal, in particular its concentrations of Fe, Ni, O, C, Si, and S. S is fixed by a family of solutions to the Mercury planetary geodesy data and by estimated mantle compositions which will be reviewed

Table 1 Sulfur solubility model parameters

Symbol	b	С	d	е	f	g	h	i	j	k
Value	405	136	32	181	305	30.2	1.13	10.7	31.4	- 3.72
±	150	25	5.5	45	380	6	0.14	2.5	18	0.07
Units	К	K GPa-1								

Used in Eq. (1). Source: Boujibar et al. (2014)

below. To obtain the rest, we use thermodynamic constraints on metal/silicate partitioning of Si, O, and C from various sources, assuming a single-stage core formation model. To obtain the oxygen content of the metal, we use Frost et al.'s (2010) model for Fe + O equilibrium between metal and silicate and the specified silicate mantle content. This gives the activity of an "FeO" component in the metal from which the O and Fe concentrations may be obtained by

$$X_{\rm Fe}^{\rm met} \times X_{\rm O}^{\rm met} = \frac{a_{\rm FeO}}{\gamma_{\rm Fe}^{\rm met} \times \gamma_{\rm O}^{\rm met}} \quad . \tag{3}$$

Ni is obtained by assuming a chondritic mass ratio $X_{\rm Ni}/X_{\rm Fe}$ = 0.06 (McDonough and Sun 1995). C is obtained by assuming saturation in the metal. We justify this choice based on the quite large C concentrations implied with Marcyr's high S abundances using the chondritic C/S ratio (0.648) (McDonough and Sun 1995), because models of Mercury formation invoke an early graphite crust (Kaaden and McCubbin 2016; Cartier and Wood 2019), and because graphite appears to be present on Mercury's surface (Peplowski et al. 2015). We use Wood's (1993) parameterization for C saturation in metal, and use the pressure of the depth of the base of the magma ocean to set the *P* and *T* conditions; see Additional file 1: Figure S1 for a plot of C saturation vs pressure. Si is obtained from Hirose et al.'s (2017) model for joint Si and O metal/silicate partitioning. The metal activity coefficients γ_i are evaluated using the ϵ formalism (Ma 2001) with interaction parameters given in Table 2. An iterative method is used to solve consistently for Si concentration and metal activity coefficients given the Si and O K_Ds and C concentration. Table 3 gives metal Si content at 2 wt% S.

Peridotite solidus and liquidus temperatures are from Fiquet et al. (2010), whose mean provides the temperature as a function of pressure, eliminating an independent variable. Pressures in the proto-Mercurys are calculated assuming an adiabatic temperature profile with the metal-silicate equilibration pressure fixed at $\frac{1}{2} P_{\text{CMB}}$ and no temperature discontinuity at the CMB. Internal

Table 2 Metal interaction parame	eters
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$\epsilon_j^i = \epsilon_i^j$	$i \rightarrow j\downarrow$	Ni	Si	С	0	S
S		- 0.05	9.16	6.12	0	- 5.66
0		1.4	7.73	- 20.08	- 9.16	
С		2.31	9.70	12.83		
Si		7.5	0			
Ni		0.12				

Sources: O-O, O-Si, Si-Si (Hirose et al. 2017); S-Ni, Si-Ni, S-S, Ni-Ni (Wade et al. 2012); Ni-O (Tuff et al. 2011). C-C, C-S, C-O, C-Si, C-Ni MetalAct Website, http://www.earth. ox.ac.uk/~expet/metalact/, accessed 2018/01/04 (Wade and Wood 2005)

Table 3 Mercury and Vesta mantle compositions

Oxide	Zolotov ^a	Chabot ^b	Nittler ^c	Vesta ^d
SiO ₂	56.90	49.02	52.83	44.47
TiO ₂	0.89	1.07	0.22	0.12
AI_2O_3	14.10	11.72	4.41	2.76
FeO	3.42	0	0.03	21.02
MnO	0.69	0	0	0
MgO	17.90	29.84	37.27	28.92
CaO	5.02	8.36	3.05	2.27
Na ₂ O	0.18	0	1.63	0.08
K ₂ O	0.15	0	0.04	0.01
S	1.90	-	-	-
$log_{10}C_5$	- 5.446	- 5.435	- 5.560	- 5.091
X ^{sil} wt% ^e	26.60	22.91	24.69	20.79
X ^{met} wt% ^f	28.5	28.5	27.9	-

All oxide values are wt%. ^{*a*}Mean of Zolotov et al. (2013) northern volcanic plains and intercrater plains compositions; ^{*b*}Chabot et al. (2014) experimental starting material; ^{*c*}Nittler et al. (2018) mean estimated mantle composition; ^{*d*}Average of H-chondrite and H+CM chondrite mix (Toplis et al. 2013) and OIM and BD compositions (*^d*Average) (see the start and start are start as the start and the start as the start as the start and the start as the start and the start as the start and th

(Lodders 2000). Average Vesta metal composition (wt%): Fe 76.32, Ni 9.63, Co 0.096 S 12.98; ^eMean value is 24.73 wt%; ^fAt 2 wt% S in metal

structure is calculated using the methodology and material properties of metal and silicate described elsewhere (Helffrich 2017). The choice of $\frac{1}{2} P_{\text{CMB}}$ is motivated by various studies of moderately siderophile element concentrations in the mantle (Tuff et al. 2011; Rubie et al. 2011) that find a best fit if magma ocean pressure P_{BMO} is $2/7 \leq P_{\text{BMO}}/P_{\text{CMB}} \leq 2/3$; $\frac{1}{2}$ is a useful approximation to the mean pressure.

Because they are cast as exchange reactions, our models are not explicitly dependent on oxygen fugacity (f_{O_2}), which, on Mercury, is quite low relative to the iron-wüstite buffer (Namur et al. 2016). The major element content, rather, implies an f_{O_2} . On the Earth, the variation in equilibrium partitioning conditions for various moderately siderophile elements leads one to invoke a change in the composition of accreting material (or, equivalently, f_{O_2}) (Wood et al. 2006; Rubie et al. 2011), but we lack the information required to evaluate this for Mercury. For the Earth, there was an evolution to + 1 or + 1.5 log units more oxidizing conditions through accretion, but at all times was more oxidizing than Mercury's present conditions (Wood et al. 2006; Namur et al. 2016).

In order to test the model and demonstrate its suitability for other silicate bodies, we recruit Vesta. Its total and core mass are known from the Dawn mission, and the silicate composition is represented by the HED meteorite suite, thought to represent Vesta fragments; Vesta is also believed to have self-differentiated and was not subject to significant mass loss by impact erosion (Russell et al. 2012) (but see Consolmagno et al. (2015) for an alternative view). Various efforts to model Vesta's differentiation using HED meteorites provide Vesta silicate and core compositions (Toplis et al. 2013; Mandler and Elkins-Tanton 2013) whose average is given in Table 3. We use the sulfur partitioning model to calculate the equilibrium sulfur content of Vestan silicate. We modify the solution method slightly for this case and fix the metal's carbon concentration to the low-pressure saturation value in metal, 3.5 wt% (Tafwidli and Kang 2017). The differentiation pressure is the CMB pressure (Fig. 1) and temperature is midway between the peridotite solidus and liquidus. Figure 1 shows the results. At Vestan differentiation pressures (< 0.1 GPa), $X_{\rm S}^{\rm sil}$ is limited to 0.47–0.55 wt% depending on the core's Si concentration, which is constrained by Si partitioning experiments between metal and silicate (Fischer et al. 2015). The sulfur contents of individual eucrite analyses (Kitts and Lodders 1998) is 0.21 ± 0.22 wt% (ranging from 0.03–1.02 wt%). Hence, metal-silicate equilibration in Vesta may be demonstrated at the 2σ level with the model. A further comparison, though requiring further inferences that weaken its verdict, may be made using the estimated S content of the whole HED meteorite suite (Steenstra et al. 2019). They estimate S in HED silicate to be 0.13 \pm 0.04 wt%. The depletion of volatile lithophile elements in HED meteorites appears to be 3-4 times relative to chondrites (Ruzicka et al. 2001). Hence, from these values, 3–4 times 0.13 wt% meteorite S yields 0.39-0.52 wt% S in bulk Vesta silicate, which is within our model estimate's range. This validates the model against the composition of a relatively well-known protoplanetary body.

Isotopic model

The ³⁰Si isotope balance for bulk Mercury (subscript *b*) may be written in terms of its partitioning between the core's metal (subscript *c*) and the mantle's silicate (subscript *m*)

$$\delta^{30} \mathrm{Si}_b = f_c \delta^{30} \mathrm{Si}_c + (1 - f_c) \delta^{30} \mathrm{Si}_m \quad , \tag{4}$$

where f_c is the Si mass fraction of the core relative to the planet (reckoned as 0.2918 for proto-Mercury core mass fraction of 0.26). To account for the silicate-metal partitioning during core formation, we use temperaturedependent fractionation factors $\Delta^{30}\text{Si}(T)$. Hence,

$$\delta^{30}\mathrm{Si}_c = \Delta^{30}\mathrm{Si}(T) + \delta^{30}\mathrm{Si}_m \quad . \tag{5}$$

The fractionation factor expressions $\Delta^{30}\text{Si}(T)$ are either $\Delta^{30}\text{Si}(T) = -7.45(\pm 0.41) \times 10^6/T^2$ (Shahar et al. 2011) or a composite value derived from various fractionation factor studies: Shahar et al. (2011), Ziegler et al. (2010), Georg et al. (2007), and Hin et al. (2014). To form the composite fractionation factor, we average using



(CMB) pressure is evident in knickpoint in P(R) curve. **b** Sulfur model applied to Vesta, using average estimates of bulk Vesta composition. Contours show wt% S in Vesta silicate assuming metal-silicate equilibrium at indicated pressure (*P*) and sulfur content of silicate X_{S}^{sil} . Dashed vertical line is Vestan CMB pressure from **a**. Sloped horizontal line is X_{S}^{sinet} calculated from metal-silicate equilibrium at the given pressure. For equilibration pressures at or below the Vestan CMB (lower left), silicate sulfur content is $0.47 \le X_{S}^{sil} \le 0.55$ wt%. For measured eucrites, $0.03 \le X_{S}^{sil} \le 1.02$ wt% (Kitts and Lodders 1998)

 $1/\sigma^2$ weights with reported uncertainties, where available. Georg et al.'s (2007) calculations yielded two separate fractionation factor values that we average and adopt half of their difference as the uncertainty. The uncertainty estimation procedure for the value derived from Hin et al.'s (2014) experiments is not clearly described, so we took a weighted average of their values at the two experimental temperatures and fit those two values to a $10^6/T^2$ model to obtain an estimated value and uncertainty. The individual values (and uncertainties) divided by 10^6 are, respectively, -7.64 ± 0.47 , -7.45 ± 0.41 , -5.65 ± 0.15 , and -5.05 ± 1.58 , which yield a composite fractionation factor of $\Delta^{30} \text{Si}(T) = -6.00(\pm 1.21) \times 10^6/T^2$.

The mean values for various chondrite classes provide δ^{30} Si_b. Combining (4) and (5), δ^{30} Si for the silicate of the proto-Mercury is

$$\delta^{30} \mathrm{Si}_m = \delta^{30} \mathrm{Si}_b - f_c \times \Delta^{30} \mathrm{Si}(T) \quad . \tag{6}$$

Results

The experimentally determined metal-silicate partition coefficient for sulfur (1) may be written as

$$D_S = f(P, T, C_S, W_{\text{FeO}}^{\text{sil}}, X^{\text{met}}) \quad , \tag{7}$$

where f is a function of pressure (P), temperature (T), the sulfur carrying capacity of the silicate C_S (2), the FeO concentration in the silicate, and the metal composition X^{met} . Mercury's silicate composition (Table 3) may then be used to find the P-T conditions where metal-silicate equilibrium is achieved if the metal composition is known.

The three silicate compositional estimates are derived in various ways from surface spectroscopic observations (Table 3). The most straightforward is to assume that the surface represents bulk Mercury. Zolotov et al.'s (2013) estimates of the composition of crustal lavas on Mercury provides this compositional choice. It neglects any differentiation that arises from partial melting of the mantlemelt fractions are thought to be 25-40% (Nittler et al. 2018)-but has the virtue of being purely observationally based. Chabot et al. (2014) attempted to reproduce Mercury's surface composition experimentally by melting mixtures of silicate and metal. We use the experimental starting silicate as a second estimate of the mantle composition. A virtue of this choice is that there is no model underlying it except for being a plausible parental source composition. However, the composition is simplified, lacking alkalis (Table 3). The final estimate is by a formal back-calculation from the surface composition of the mantle's composition assuming that it was formed by batch decompression melting as solid convectively rose along an adiabat (Nittler et al. 2018). For fidelity with the modeling assumptions, we use the model's reported output composition (0.02–0.04 wt% FeO, Table 3) rather than the one augmented by the addition of 1 wt% FeO reported by those authors. There is no significant difference in the results no matter which compositional model is used, so the results shown in Fig. 2 are using Nittler et al.'s (2018) estimate; see Additional file 1: Figure S2 for the results using the other starting compositions. Based on an assumption of an enstatite chondrite origin for Mercury and the content of its surface lavas, Namur et al. (2016) estimated the f_{O_2} of formation conditions



metal/silicate equilibrium. Colored box depicts limits on S and FeO in silicate discussed in text (see Table 3 for compositions). Lines of constant FeO content across box indicate estimates of Mercury compositions from two sources: Nittler et al. (2018) mantle composition (0.04 wt%) and Namur et al. (2016) ($\Delta/W = -5.4$). Equivalent oxygen fugacity in log units relative to the iron-wüstite buffer on right assuming 70 wt% Fe in metal. Hatched regions flag pressures outside Mercury's $2/7 \le P/P_{CMB} \le 2/3$ limits. Liquid immiscibility in Fe-Si-S (Morard and Katsura 2010) potentially affects only the lowest pressures at $X_5^{met} = 2$, but does not substantially affect the results

as $-5.4 \log_{10}$ units relative to iron-wüstite, which constrains FeO to ~ 0.3 wt%. We use this as an upper bound on the range of possible silicate FeO content. The same authors also analyzed Mercury surface spectroscopic data to estimate the surface S content, finding terrane mean ranges from 1.6 to 2.5 wt%. From melt modeling, they reckon the source of the surface lavas contains 2–12 wt% S, but favor 7–11 wt% on the basis of their f_{O_2} estimate and sulfur saturation model. The FeO estimates provided by the two studies are not mutually compatible; we adopt an inclusive range of 0.01–0.3 wt% for FeO (but show values up to 1 wt%) and, for S, adopt a narrower 1–7 wt% range because we do not assume S saturation in the silicate as Namur et al. (2016) does.

Studies of Mercury's gravity field and spin state provide limits on the sulfur content of its core through their compatibility with the planet's polar moment of inertia and its mantle's moment of inertia. Dumberry and Rivoldini (2015), extending Rivoldini and Van Hoolst's (2013) work, explored a range of ~ 1–15 wt% for the sulfur content of the core, but commonly found contents ~ 5 wt% compatible with the geodetic data. Hauck II et al. (2013) also explored compositional constraints on Mercury's core but used only a moment of inertia rotational data which do not provide as tight constraints. We therefore examine compositions with $X_{\rm S}^{\rm met}$ of 1, 2, 5, and 7 wt%. For each of these sulfur compositional ranges that simultaneously satisfy the S, Fe, Si, and O partitioning between core metal and mantle silicate at C saturation.

The pressure for metal-silicate equilibrium in Mercury depends on the core sulfur content (Fig. 2). At low values (1 wt%), the pressure is \sim 7–75 GPa and increases to \sim 11–103 GPa at 7 wt%. For most of the core's sulfur composition range, the equilibrium pressure exceeds the present CMB pressure, \sim 6 GPa (Rivoldini et al. 2009; Rivoldini and Van Hoolst 2013), suggesting that the metal-silicate equilibrium took place in a larger planet.

Some insights into how large the earlier planet may have been may be derived from planetary metal proportions f_{metal} (= M_{core} / M_{body}) and the planetary Fe/Mg mass ratio. Figure 3 shows the relations for the rocky planets, Vesta, and for enstatite and carbonaceous chondrites. Mercury is a clear outlier, even compared to metal-rich CB chondrites that are not thought to be viable Mercury building blocks (Cartier and Wood 2019). However, by adding back silicate with the composition of Mercury's bulk silicate, we see a trend leading back to the cluster that the Solar System's other rocky bodies define. Between 1 and 3 additional $M_{\mbox{Q}}$ of silicate would place it within the solar system's terrestrial grouping.



Fig. 3 Metal weight fraction (f_{metal}) versus planetary Fe/Mg mass ratio for various rocky bodies and chondritic meteorite classes (EC, enstatite; CC, carbonaceous). Bulk Mercury estimates from Nittler et al. (2018), Chabot et al. (2014), and Zolotov et al. (2013). Mercury is a singular outlier from the cluster defined by the other objects. Dashed line is trend from silicate added to Mercury with the composition of its bulk silicate (Table 3); points on line show total body mass in units of $M_{\mbox{Q}}$. An original mass 2–4 × $M_{\mbox{Q}}$ would put it within the rocky body cluster. Data are from Fegley (2014), Dumoulin et al. (2017), McDonough and Sun (1995), Rivoldini et al. (2011), Javoy et al. (2010), and Table 3

To explore the additional silicate added to Mercury needed to permit metal-silicate equilibrium, we adopt an approach used in many simulations of major- and traceelement partitioning between the metal and silicate during planetary accretion (Wade and Wood 2005; Rubie et al. 2011). We assume a planet-wide magma ocean whose base is at $\frac{1}{2} P_{CMB}$ and calculate the mass of the planet required to achieve the equilibration pressure, which may be viewed as equivalent to a single-stage core formation model or a continuous core formation model for elements whose metal/silicate concentration ratios do not change significantly during accretion (principally the lithophile elements, but also some moderately siderophile elements, e.g., Tuff et al. (2011)).

Figure 4 shows the results. The pressure at the magma ocean's base, $P_{BMO} = \frac{1}{2} P_{CMB}$, is the anticipated equilibration pressure whose uncertainty is taken from various accretion scenarios, showing that $2/7 \le P_{\text{BMO}}/P_{\text{CMB}} \le 2/3$ (Tuff et al. 2011; Rubie et al. 2011). Thus the intersection of the $P_{\rm BMO}$ curve with the minimum equilibration pressure for each core sulfur content yields the range of the minimum added silicate mass required to allow metal-silicate equilibrium. All $X_{
m S}$ ranges explored entail added silicate: 0.4–1.5 $M_{
m igodot}$ (1 wt%), 0.8–3.2 M_{\bigotimes} (2 wt%), and > 1.8 M_{\bigotimes} (5, 7 wt%). Thus, from a bulk chemistry standpoint alone, at least 0.4 M_{X} was lost from Mercury. An additional constraint is the planetary Fe/Mg ratio and its comparison to the ratio in terrestrial solar system bodies and chondritic precursors (Fig. 4). At low core S content (1 wt%), the Fe/Mg range lies outside the range, rendering it infeasible on compositional grounds. Only at a core sulfur content of ~ 2 wt% and higher does metal/silicate sulfur partitioning also satsify Fe/Mg constraints. A core sulfur content of $\sim~2$ wt% yields planetary Fe/Mg and metal ratios in the known range of the other rocky bodies. A minimum of 0.8–3.3 \times M_{\odot} added silicate is required to yield a Mercury bulk silicate composition controlled by metalsilicate equilibrium for this core composition. More S-rich cores for Mercury require higher equilibration pressures (Fig. 2) and suggest larger sizes for the proto-Mercury. Added silicate amounts of $\sim 1{-}3\,M_{\bigotimes}$ yield both Fe/Mg ratios in the chondritic range and metal mass fractions compatible with the other terrestrial planets.

It is worth highlighting some limitations of our approach. Firstly, we have no direct samples or observations of Mercury's mantle composition, so it is inferred. The surface compositions, from which all Mercury silicate estimates are derived, are obtained from re-emission of energy radiated during solar flares (Nittler et al. 2011), and strictly sample only the top tens of micrometers of Mercury's surface. If the surface were completely mantled by a late veneer, its bulk composition would

be masked. Mercury imaging shows extensive volcanic formations, however, whose compositions are individually distinct (Zolotov et al. 2013; Namur et al. 2016), making any overall mantling unlikely, but possible still.

Our modeling is essentially driven by the surface silicate composition, from which the core composition is derived consistent with thermodynamic and experimental data and models. If Mercury is chondritic, some key major element ratios obtained by proportionally combining the silicate and core compositions should also be chondritic. In particular, Fe/Si and Mg/Si should be. With Additional file 1: Figure S3, one can see that the comparison is imperfect but encouraging. Either the silicate estimates are insufficiently rich in Mg relative to Si or the modeled core metal Si content is too high. It is unclear where the problem lies, but, again, the surface composition estimate relies on more model-dependent assumptions than does the sulfur partitioning model.

The sulfur partitioning model we use is experimentally calibrated to only 23 GPa, yet we employ it to high pressures; the extrapolation is questionable. However, the constraints it provides are essentially low-pressure ones: for Vesta (< 1 GPa) and for the lowest pressures of metal-silicate equilibration (5–25 GPa). Hence, the main results derive from pressures within the model's applicability range.

We rely on geodetic studies that assume that S is the main constituent of Mercury's core. This is certainly wrong from a chemical perspective, because our models demand knowledge of the core's Ni, Si, O, and C contents too. Whether density profiles for chemically realistic Mercury core compositions would change the compositional constraints the geodetic studies provide is an open question.

Discussion

From N-body and SPH impact simulations, Benz et al. (1988, 2007) reckoned that Mercury's original mass must have been $\sim 2.25 \times M_{\odot}$, whereas Asphaug and Reufer (2014) found that impactors between $2.7-5.4 \times M_{\odot}$, striking a proto-Venus- or proto-Earth-sized target at glancing incidence (> 30°) shed sufficient mass to yield Mercury-like residues. These limits overlap with the chemical equilibration for $X_{\rm S}^{\rm met}$ of 2–5 wt% (Fig. 4). Though chemical equilibrium imposes no useful upper limit to Mercury's original size, similarity to the Solar System's other rocky bodies does. A Mercury constructed from added silicate could have had no more than about three masses added to it before it would exceed the Solar System's rocky body Fe/Mg ratios (Fig. 3).

The estimated compositions also place limits on the Si content of Mercury's core, X_{Si}^{met} , which is shown in Fig. 5a. The mean X_{Si}^{met} depends on the pressure, though Si solubility in metal is not strictly pressure-dependent (Fischer et al. 2015; Hirose et al. 2017) but other metal components are (e.g., O) that affect γ_{Si} (Table 2). The variation shown in Fig. 5a shows a further dependence on core sulfur content, justifying, post hoc, the geodetic modeling approach (Rivoldini et al. 2009; Rivoldini and Van Hoolst 2013; Hauck II et al. 2013; Dumberry and Rivoldini 2015). If so much silicate was lost from Mercury, there might be some as yet unidentified meteorite material derived from the debris of its stripped mantle (Gladman and Coffey 2009); Cartier and Wood (2019) note that aubrites would be good candidates for Mercury surface fragments. As an identification aid for Mercuryderived meteorites, stable silicon isotope analyses might be a useful discriminant due to silicon isotope fractionation during core formation (Georg et al. 2007; Fitoussi and Bourdon 2012; Hin et al. 2014). To draw up a silicon isotope budget, we need the core and silicate masses for the planet, and the concentration of silicon in the metal and silicate. We use the silicate composition from Table 3 to obtain the elemental Si content (24.73 wt%) and use



2.39

ECH

1.74

ទក្ខ

OCHCI

1.40

ocL

1.18 Fe/Mg

0.6

0.2

0.1

Рсм

PBMC

5 wť

2 w

15.79

60

50

4

30

20

9

C

P (GPa)

4.01



Fig. 5 a Silicon content of Mercury's core assuming metal-silicate equilibration at pressures prior to mantle stripping. Points show X_{Si}^{inc} for core formation at specified pressure for different metal S content (1, 2, 5, and 7 wt%). Horizontal dashed lines show the mean for each assumed core sulfur content. **b** Calculated δ^{30} Si_m for fractionation at various temperatures corresponding to the depth of a magma ocean and growing core mass on proto-Mercury. Colored bands show mean and $\pm 2\sigma$ uncertainties for δ^{30} Si for major chondritic meteorite classes. Points show predicted fractionation in silicate for ~ 2 wt% S content if δ^{30} Si_b corresponds to enstatite or ordinary and carbonaceous classes, and would be the predicted δ^{30} Si for meteorites derived from fragments of Mercury's stripped mantle. All estimates are $\sim 0.1\%$ higher than their chondritic parents' means. Circles and squares are from experimentally determined fractionation factors of Shahar et al. (2011) and the composite estimate (see the "Methods" section), respectively

the mean core Si content to calculate the silicate-metal partitioning of ³⁰Si during core formation. Lacking knowledge of the bulk Mercury δ^{30} Si, we use the mean values for various chondrite classes reported by Armytage et al. (2011): enstatite chondrites -0.63% ordinary and carbonaceous chondrites -0.45% The results are shown in Fig. 5b. We find that offsets of + 0.10% or more in δ^{30} Si relative to any chondritic parent would be characteristic of meteorites derived from the proto-Mercury body if either Shahar et al.'s (2011) fractionation factor is used or a composite factor derived from all sources. Due to Mercury's similarity to enstatite chondrites (Cartier and Wood 2019), these would be the likely parent material, and at low (15-20 GPa) magma ocean pressures corresponding to 2.5–3× M_{\odot} proto-Mercurys, δ^{30} Si offsets would be \sim + 0.20% (Fig. 5b). The trend with lower pressures also suggests that even if no stripping occurred, Mercury-derived meteorites would have significant δ^{30} Si offsets > + 0.3%. Hence, the S and FeO content of the silicate, along with the δ^{30} Si offset from enstatite chondrites, might prove to be a useful signature of Mercury provenance.

If Mercury was originally a few times its current mass and its mantle was lost through a giant impact (Benz et al. 1988; Benz et al. 2007; Asphaug and Reufer 2014), where did this material go? Benz et al. (1988) concluded that the large impact scenario could only work if the ejected material was not subsequently re-accreted by Mercury. From *N*-body simulations of the ejecta, Benz et al. (2007) concluded that only a small fraction of ejected material would collide with Venus or be removed from Mercury-crossing orbits through secular perturbations. Even if launch speeds of ejecta from Mercury are high enough to place them on Venus- and Earth-crossing orbits, only a few percent of this material collides with the other two planets (Gladman and Coffey 2009). Only if the debris was centimeter-sized and thus affected by Poynting-Robertson drag could material be efficiently removed from this region in a few million years (Benz et al. 2007; Gladman and Coffey 2009). Benz et al. (2007) compute that the size of ejecta material has a peak below 1 cm, making Poynting-Robertson an attractive way to remove material before most of it reaccretes onto Mercury.

Asphaug and Reufer (2014) proposed solving the problem of ejecta re-accretion onto Mercury by making Mercury the impactor rather than the target. If proto-Mercury formed in the outer, volatile-rich regions of the protoplanetary disc and, additionally, suffered multiple, lower speed collisions with other targets, it could also retain volatile sulfur (Hauck II et al. 2013; Asphaug and Reufer 2014; Cartier and Wood 2019), solving yet another Hermean chemical conundrum.

Conclusions

In this study, we used the concept of metal-silicate partitioning in magma ocean conditions to constrain the pressure at which Mercury's silicate and metal could have been in equilibrium. An experimentally derived model of partitioning for sulfur provided the required pressurecomposition relationship. The structure of Vesta and candidate Vestan silicate compositions provided a demonstration of the suitability of the model by showing that it yields, at Vestan equilibration pressures, silicate S content equivalent to Vesta's.

Applying the same model to Mercury, using its observed surface silicate composition and estimates of its core composition from space geodetic studies, we find that equilibration may be achieved if Mercury was originally a larger planet 1.4–2.5 times more massive, and possibly 2–4 times so. Hence Mercury's surface chemistry and geodetic parameters are compatible with silicate mass loss through collision with other objects, or by it having been an impactor itself.

We used the estimated compositions of Mercury's core to calculate the stable silicon isotope fractionation between core metal and mantle silicate. δ^{30} Si offsets of + 0.10% relative to ordinary, enstatite or carbonaceous chondrite precursors would arise, and, if Mercury's original mass was $2.5-3 \times M_{relevent}$, the shifts would be ~ + 0.20%. Even if no mantle stripping occurred, the fractionation trend implies offsets in excess of + 0.30%. These offsets could fingerprint meteorite fragments derived from Mercury.

The future mission to the asteroid Psyche (Lord et al. 2017) may yield data to decide whether the body represents the end-point of a silicate stripping process. If there is sufficient mantling silicate for estimating its mantle's composition and if the silicate retains sufficient sulfur, a chemical approach similar to that used here may reveal Psyche's original mass as well.

Supplementary information

Supplementary information accompanies this paper at https://doi.org/10.1186/s40645-019-0312-z.

Additional file 1: Discussion of f_{metal} for chondritic meteorites. Figure S1 Carbon saturation as a function of pressure. Figure S2a Compatibility of S and FeO content of Mercury's mantle with pressure of metal/silicate equilibration for various core S contents (1, 2, 5 and 7 wt%), using Chabot bulk silicate composition. Figure S2b Compatibility of S and FeO content of Mercury's mantle with pressure of metal/silicate equilibration for various core S contents (1, 2, 5 and 7 wt%), using Zolotov bulk silicate composition. Figure S3 Comparison of modeled bulk Mercury Fe, Mg and Si against various chondritic meteorite classes.

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Authors' contributions

GH designed the study, performed all calculations, and drafted the paper. RB and AS discussed the methods used and contributed to the submitted version's text. All authors read and approved the final manuscript.

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Availability of data and materials

All data used are in this paper; there are no additional experimental materials associated with the work.

Competing interests

The authors have no competing interests regarding the Hermean composition or resource exploitation nor in narratives of Mercury's history.

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