RESEARCH ARTICLE







Synthesis and antibacterial activity of novel 5,6,7,8-tetrahydroimidazo[1,2-a] pyrimidine-2-carbohydrazide derivatives

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Abstract

Background: The intensely increasing multi-drug resistant microbial infections have encouraged the search for new antimicrobial agents. Hydrazone derivatives are known to exhibit a wide variety of biological activities including antimicrobial. In heterocyclic moiety, imidazo[1,2-a]pyrimidines are the subject of immense interest for their antimicrobial activity and also for their analgesic, antipyretic and anti-inflammatory properties.

Results: Condensation of 5,6,7,8-tetrahydroimidazo[1,2-a]pyrimidine-2-carbohydrazide **7** with aromatic aldehydes **a–k** in ethanol at reflux led to the generation of hydrazone derivatives **8a–k** in 80–92% yield. The synthesis of carbohydrazide **7** was accomplished in six steps from commercially available 2-amino pyrimidine. The structures of the synthesized compounds were confirmed by ¹H, ¹³C NMR, Mass and IR spectral data. All the synthesized hydrazone derivatives **8a–k** were tested in vitro for their antibacterial activity. Compounds **8d**, **8e** and **8f** exhibited excellent antibacterial activity with zone of inhibition 30–33 mm against *E. coli* (Gram negative bacteria) and *S. aureus* (Gram positive bacteria). These compounds also exhibited excellent antibacterial activity with zone of inhibition 22–25 mm against *P. aeruginosa* (Gram negative bacteria) and *S. pyogenes* (Gram positive bacteria).

Conclusion: Synthesized and recorded antibacterial activity of some new 5,6,7,8-tetrahydro-imidazo[1,2-a]pyrimi-dine-hydrazone derivatives.

Keywords: Antibacterial activity, 2-Amino pyrimidine, Imidazo[1,2-a]pyrimidine-hydrazones, Norfloxacin

Background

The fast resistance of bacteria against antibiotics has become a prevailing medical problem. Treatment options for these infections are often inadequate especially in immune compromised patients. The intensely increasing multi-drug resistant microbial infections in the past few decades have become a serious health issue. The exploration of new antimicrobial agents will always remain as a major challenging task.

Hydrazones constitute a versatile compound of organic class having the basic structure $R_1R_2C=NNH_2$. Hydrazones are formed by the replacement of the oxygen of carbonyl compounds with the $-NNH_2$ functional group.

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Imidazo[1,2-a]pyrimidines are evolving as potentially interesting drugs particularly with regard to their antimicrobial, analgesic, antipyretic, anti-inflammatory properties and also with regard to their activity against ulcers [13]. They are also important as benzodiazepine receptor



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agonists, antiviral agents and calcium channel blockers [14].

Keeping in view of the biological importance of hydrazones and Imidazo[1,2-a]pyrimidines, in continuation to our research program about a potent antimicrobial agent, we report herein the synthesis, characterization and antibacterial activity of some new 5,6,7,8-tetrahydro-imidazo[1,2-a]pyrimidine-hydrazone derivatives. It is interesting to note that, a class of 5,6,7,8-tetrahydroimidazo[1,2-a]pyrimidine compounds is prescribed for treatment to reduce neurotoxic injury associated with anoxia or ischemia which typically follows stroke, cardiac arrest, hypoglycemia or perinatal asphyxia [11].

Results and discussions

Chemistry

The synthesis of 5,6,7,8-tetrahydro-imidazo[1,2-a]pyrimidine-hydrazone derivatives **8a–k** described in this paper were prepared according to the synthetic Scheme 1. The cyclo condensation reactions of 1,1,3-trichloro acetone with 2-aminopyrimidine 1 to 2-(dichloromethyl)imidazo[1,2-a]pyrimidine 2 was carried out in ethanol at reflux temperature for 10 h. Dichloromethyl imidazo[1,2-a]pyrimidine 2 was treated with CaCO₃ in water at reflux for 1 h to produce imidazo[1,2-a] pyrimidine-2-carbaldehyde **3**. The oxidation of aldehyde **3** was carried out in presence of oxone to obtain the desired imidazo[1,2-a]pyrimidine-2-carboxylic acid **4**. Esterification of carboxylic acid **3** followed by the hydrogenation in presence of PtO₂ at 30 psi resulted in the formation of ethyl 5,6,7,8-tetrahydroimidazo[1,2a]pyrimidine-2-carboxylate 6. Treatment of compound 6 with hydrazine-hydrate in ethanol at reflux temperature enabled in forming the key intermediate 5,6,7,8-tetrahydroimidazo[1,2-a]pyrimidine-2-carbohydrazide 7. Condensation of carbohydrazide 7 with various aromatic aldehydes $\mathbf{a}-\mathbf{k}$ in ethanol at reflux led to the generation of hydrazone derivatives 8a-k in 80-92% yield (Scheme 1). The structures of the synthesized compounds were confirmed by ¹H NMR, Mass and IR spectral data. As a representative example, the ¹H NMR spec-(E-N'-(4-(trifluoromethyl)benzylidene)-5,6,7,8tra of tetrahydroimidazo[1,2-a]pyrimidine-2-carbohydrazide 8d is characterized as follows: the four singlets at 11.26, 8.52, 7.36 and 6.32 ppm with one proton integration corresponds to -CO-<u>NH</u>-N-, -N=<u>HC</u>-Ar, Imidazo-<u>HC</u>=Cand $-CH_2 - \underline{NH}$ - groups, respectively. The doublet signals at 7.85 and 7.78 ppm with two proton integration represents to 4-CF₃ aromatic ring. The broad singlets with two proton integrations at 3.94, 3.35 and 1.95 ppm corresponds to the cyclic 5,6,7,8-tetrahydroimidazo ring. The mass spectra of the synthesized compounds showed (M+H) peaks, in agreement with their molecular formula. The IR data provided functional group evidence for the formation of the expected structures.

Biological assay

The synthesized (E)-*N*′-(benzylidene)-5,6,7,8-tetrahydroimidazo[1,2-a]pyrimidine-2-carbohydrazide derivatives



8a–k were dissolved in dimethylsulphoxide at 250 μ g/ mL concentration (standard antibacterial drug, Nor-floxacin was used as the reference antibiotic) and tested against Gram negative strains of (1) *Escherichia coli* (MTCC 443), (2) *Pseudomonas aeruginosa* (MTCC 424) and Gram positive strains of (3) *Staphylococcus aureus* (MTCC 96) and (4) *Streptococcus pyogenes* (MTCC 442) using agar well diffusion method according to the literature protocol [12, 13]. Activity was determined by zones showing complete inhibition (mm). Growth inhibition was calculated with reference to positive control. All the samples were taken in triplicates.

Antibacterial activity

The screening results of antibacterial activity of hydrazone derivatives 8a-k are summarized in Table 1. It is observed that compounds 8d, 8e and 8f revealed excellent antibacterial activity with zone of inhibition 30-33 mm against E. coli (Gram negative bacteria) and S. aureus (Gram positive bacteria) even in the case of P. aeruginosa (Gram negative bacteria) and S. pyogenes (Gram positive bacteria), compounds 8d, 8e and 8f displayed excellent anti-bacterial activity with zone of inhibition 22-25 mm. Compounds 8a, 8b and 8c showed appreciable activity while the compounds 8g and 8h showed moderate activity against all the tested bacterial strains. From the structural point of view of the scaffold, it may be generalized that scaffold with $R = 4-CF_3$, $4-OCF_3$ and 4-OCHF₂ showed excellent antibacterial activity, while R = 4-F, 2-CF₃ and 3-CF₃ showed good antibacterial activity and R = 2,4-difluoro and 3,4-difluoro showed moderate antibacterial activity. Furthermore, it is observed that within the series of the hydrazone derivatives 8a-k, compounds 8i (R = 2-Me, 3-F), 8j (R = 2-Cl, 3-F) and 8k (R = benzo[b]furan) showed no activity.

Experimental section

All commercial chemicals were taken as they are. The solvents underwent purification process as per standard procedures. For thin-layer chromatography (TLC) analvsis, Merck pre-coated Plates (silica gel 60 F254) were used and spots were visualized with UV light. Merck silica gel 60 (230-400 mesh) was used for flash column chromatography and the eluting solvents were indicated in the procedures. Melting point (mp) determinations were performed by using Mel-temp apparatus and are uncorrected. ¹H NMR spectra were recorded in Varian MR-400 MHz instrument. Chemical shifts were reported in δ parts per million (ppm) downfield from tetramethylsilane (TMS) with reference to internal standard and the signals were reported as s (singlet), d (doublet), dd (doublet of doublet), t (triplet), g (guartet), m (multiplet) and coupling constants in Hz. The mass spectra were recorded on Agilent ion trap MS. Infrared (IR) spectra were recorded on a Perkin Elmer FT-IR spectrometer. The benzaldehydes **a**-**k** utilized for the synthesis of **8a**-**k** were purchased from commercial sources.

Synthesis of 2-(dichloromethyl)imidazo[1,2-a]pyrimidine 2

To a mixture of 2-aminopyrimidine (5 g, 52.57 mmol), sodium bicarbonate (11.34 g, 105.14 mmol) in ethylalcohol (20 mL) was added 1,1,3-trichloro acetone (12.70 g, 78.85 mmol) and stirred at reflux for 10 h. The completion of the reaction was monitored by TLC, the reaction mixture was poured into ice-water (30 mL) with vigorous

Table 1 Antibacterial activity of compounds 8a-k (concentration used 250 μ g/ml	. of DMSO)
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Compound no.	Zones of inhibition of compounds 8a-k in mm				
	Gram negative		Gram positive		
	E. coli MTCC 443	P. aeruginosa MTCC 424	S. aureus MTCC 96	S. pyogenes MTCC 442	
8a	25	19	24	17	
8b	27	20	23	18	
8c	26	19	23	17	
8d	33	24	29	24	
8e	30	25	30	22	
8f	30	23	31	22	
8g	20	16	19	13	
8h	21	15	19	14	
8i	-	_	-	-	
8j	-	_	-	-	
8k	-	_	-	-	
Standard drug: Norfloxacin	29	23	28	21	

stirring to precipitate the product, 2-(dichloromethyl) imidazo[1,2-a]pyrimidine **2**. The crude compound was taken to the next step without further purification.

Synthesis of imidazo[1,2-a]pyrimidine-2-carbaldehyde 3

The compound **2** (1 g, 4.95 mmol) was dissolved in water (30 mL) and CaCO₃ was added (5 g, 24.75 mmol) and refluxed for 1 h. The reaction mixture was cooled to room temperature and extracted with ethylacetate and evaporated under reduced pressure to obtain the crude compound **3**, which was purified by column chromatography (silica gel: 60–120 mesh, eluent: 1% MeOH/CHCl₃) to afford compound **3**. Pale yellow solid, Yield: 60%; M.p: 209–210°C. ¹H NMR (400 MHz, CDCl₃): δ 9.96 (s, 1H), 9.66 (q, J = 6.8 Hz, 1H), 8.88 (q, J = 4.0 Hz, 1 H), 8.70 (s, 1H), 7.43 (q, J = 6.4 Hz, 1 H). ESI–MS: m/z, 148.2 (M+H)⁺.

Synthesis of imidazo[1,2-a]pyrimidine-2-carboxylic acid 4

To a solution of compound **3** (2 g, 13.60 mmol) in DMF (15 vol), Oxone was added (40.80 mmol) at 5°C and stirred for 2 h. The completion of the reaction was monitored by TLC. As soon as the reaction was completed, the reaction mixture was diluted with water (45 mL) followed by ethyl acetate. The organic layer was washed with water followed by brine solution, dried over sodium sulphate, filtered and evaporated to produce compound **4** as pale yellow syrupy liquid. The crude compound was used in the next step without further purification. Yield: 1.10 g, 50%.

Synthesis of ethyl imidazo[1,2-a] pyrimidine-2-carboxylate 5

To a solution of compound **4** (1.10 g, 6.75 mmol) in ethanol (12 mL), Conc. H_2SO_4 (catalyst) was added and refluxed for 16 h. The reaction mixture was evaporated and the residue was extracted with EtOAc (2 × 15 mL). The organic layer was washed with water followed by brine solution, dried over anhydrous sodium sulphate, filtered and evaporated to isolate compound **5** as a yellow viscous liquid. Yield: 1.06 g, 85%. M.p. 180–183°C; ¹H NMR (400 MHz, CDCl₃): δ 8.66 (dd, *J* = 2.1, 3.9 Hz, 1H), 9.50 (dd, *J* = 1.8, 6.9 Hz, 1H), 8.15 (s, 1H), 6.96 (dd, *J* = 4.2, 6.9 Hz, 1H), 4.45 (q, *J* = 7.2 Hz, 2H), 1.44 (t, *J* = 7.2 Hz, 3H). ESI–MS: *m/z*, 192.0 (M+H)⁺

Synthesis of ethyl 5,6,7,8-tetrahydroimidazo[1,2-a] pyrimidine-2-carboxylate 6

A solution of compound 5 (1.06 g, 5.55 mmol) in ethanol (20 mL): conc. HCl (0.5 mL) was hydrogenated at 30 psi in the presence of 20% PtO_2 (200 mg) for 3 h. The reaction mixture was filtered under nitrogen atmosphere and the filtrate was evaporated to obtain compound 6. The pH of the solution was adjusted to 9–10 with aqueous sodium carbonate solution and extracted with

chloroform (2 × 25 mL). The combined organic extracts were washed with water (50 mL) followed by brine solution. The organic layer was dried over anhydrous sodium sulphate and concentrated in vacuum to isolate compound **6** as yellow solid. Yield: 0.97 g, 90%; M.p: 170–171°C; ¹H NMR (400 MHz, CDCl₃): δ 7.18 (s, 1H), 6.25 (brs, 1H), 1.36 (t, *J* = 7.2 Hz, 3H), 2.07 (t, *J* = 5.7 Hz, 2H), 3.45 (t, *J* = 5.4 Hz, 2H), 3.90 (t, *J* = 6.0 Hz, 2H), 4.32 (q, *J* = 7.2 Hz, 2H), ESI–MS: *m/z*, 196.2 (M+H)⁺.

Synthesis of 5,6,7,8-tetrahydroimidazo[1,2-a] pyrimidine-2-carbohydrazide 7

A mixture of compound **6** (0.95 g, 4.87 mmol) and hydrazine hydrate (5.0 mmol) in ethanol (4 mL) was refluxed for 6 h. The reaction mixture was evaporated and the residue was diluted with EtOAc and water. The organic layer was washed with water followed by brine solution, separated, dried over anhydrous Na₂SO₄, filtered and evaporated to obtain compound 7 as a pale yellow solid. Yield: 80%; M.p: 116–117°C; ¹H NMR (400 MHz, CDCl₃): δ 8.50 (s, 1H), 7.09 (s, 1H), 6.34 (s, 1H), 4.20 (brs, 1H), 3.86 (t, J = 6.0 Hz, 2H), 3.21–3.17 (m, 2H), 1.93–1.87 (m, 2H); ESI–MS: m/z, 181.8 (M+H)⁺.

General experimental procedure for the synthesis of hydrazone derivatives (8a-k)

Respective benzaldehydes $\mathbf{a}-\mathbf{k}$ (1.0 mmol) were added to an ethanol solution containing 7(100 mg, 0.553 mmol) and the contents were stirred at reflux temperature for 6 h. Ethanol was evaporated from the reaction mixture and the residue was extracted with pet-ether, to obtain pure compounds. Yields of the products varied between 80 and 92%.

(E)-N'-(4-fluorobenzylidene)-5,6,7,8-tetrahydroimidazo [1,2-a]pyrimidine-2-carbohydrazide (8a)

M.p: 84–85°C; IR(KBr): υ_{max} 3,443, 3,303, 3,275, 3,243, 3,187, 2,981, 1,667, 1,612, 1,602, 1,565, 1,506, 1,495, 1,425, 1,415, 1,370, 1,350, 1,319, 1,290, 1,275, 1,236, 1,209, 1,192, 1,183, 1,156, 1,135, 1,112, 1,101, 1,064, 998, 959, 930, 901, 854, 837, 819, 788, 751, 629, 612, 566, 519, 505 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 11.01 (s, 1H), 8.43 (s, 1H), 7.71–7.68 (m, 2H), 7.32–7.25 (m, 3H), 6.30 (s, 1H), 3.39 (t, *J* = 5.6 Hz, 2H), 3.24 (brm, 2H), 1.94 (t, *J* = 5.6 Hz, 2H); ESI–MS: *m/z*, 287.9 (M+H)⁺.

(E)-N'-(2-(trifluoromethyl) benzylidene)-5,6,7,8-tetrahydroim idazo[1,2-a]pyrimidine-2-carbohydrazide (8b)

M.p: 104–105°C; ¹H NMR (400 MHz, DMSO-d₆): δ 11.22 (s, 1H), 8.86 (s, 1H), 8.16 (d, *J* = 8.4 Hz, 1H), 7.76 (d, *J* = 8.4 Hz, 1H), 7.74–7.59 (m, 2H), 7.35 (s, 1H), 6.28 (s, 1H), 3.93 (brs, 2H), 3.24 (brs, 2H), 1.94 (brs, 2H); ESI–MS: *m*/*z*, 337.9 (M+H)⁺.

(E)-N'-(3-(trifluoromethyl)benzylidene)-5,6,7,8-tetrahydroimi dazo[1,2-a]pyrimidine-2-carbohydrazide (8c)

M.p: 98–99°C; ¹H NMR (400 MHz, DMSO-d₆): δ 11.22 (s, 1H), 8.86 (s, 1H), 8.16 (d, J = 8.4 Hz, 1H), 7.76 (d, J = 8.4 Hz, 1H), 7.74–7.59 (m, 2H), 7.35 (s, 1H), 6.28 (s, 1H), 3.93 (brs, 2H), 3.24 (brs, 2H), 1.94 (brs, 2H); ESI–MS: m/z, 337.9 (M+H)⁺.

(E)-N'-(4-(trifluoromethyl) benzylidene)-5,6,7,8-tetrahydroimi dazo[1,2a]pyrimidine-2-carbohydrazide (8d)

M.p: $120-121^{\circ}$ C; IR(KBr): υ_{max} 3,443, 3,261, 3,242, 3,184, 3,147, 3,131, 3,085, 3,011, 2,982, 2,959, 2,936, 2,851, 2,206, 1,663, 1,625, 1,560, 1,522, 1,507, 1,494, 1,463, 1,445, 1,433, 1,413, 1,364, 1,331, 1,321, 1,300, 1,293, 1,271, 1,235, 1,210, 1,193, 1,182, 1,158, 1,134, 1,119, 1,102, 1,068, 1,017, 994, 967, 959, 939, 898, 895, 842, 832, 784, 762, 720, 709, 672, 644, 630, 615, 594, 508, 503 cm⁻¹ ¹H NMR (400 MHz, DMSO-d₆): δ 11.26 (s, 1H), 8.52 (s, 1H), 7.85 (d, J = 8.4 Hz, 2H), 7.78 (d, J = 8.0 Hz, 2H), 7.36 (s, 1H), 6.32 (s, 1H), 3.94 (brs, 2H), 3.35 (brs, 2H), 1.95 (brs, 2H); ESI-MS: m/z, 337.8 (M+H)⁺.

(E)-N/-(4-(trifluoromethoxy)benzylidene)-5,6,7,8-tetrahydro imidazo[1,2-a]pyrimidine-2-carbohydrazide (8e)

M.p: 126–127°C; IR(KBr): υ_{max} 3,243, 3,224, 3,182, 3,099, 3,054, 2,972, 2,883, 2,847, 1,653, 1,631, 1,605, 1,555, 1,535, 1,506, 1,492, 1,441, 1,414, 1,386, 1,367, 1,351, 1,325, 1,301, 1,270, 1,243, 1,217, 1,193, 1,166, 1,099, 1,090, 1,072, 1,017, 1,002, 964, 944, 918, 902, 878, 850, 793, 756, 715, 677, 650, 662, 600, 532 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 11.10 (s, 1H), 8.46 (s, 1H), 7.76 (d, *J* = 8.4 Hz, 2H), 7.42 (d, *J* = 8.4 Hz, 2H), 7.33 (s, 1H), 6.30 (s, 1H), 3.92 (t, *J* = 6.0 Hz, 2H), 3.24 (brs, 2H), 1.94 (brs, 2H); ESI–MS: *m/z*, 354.0 (M+H)⁺.

(E)-N/-(4-(difluoromethoxy)benzylidene)-5,6,7,8-tetrahydro imidazo[1,2-a]pyrimidine-2-carbohydrazide (8f)

M.p: 119–120°C; IR(KBr): υ_{max} 3,429, 3,272, 3,099, 3,056, 3,041, 2,972, 2,920, 2,886, 2,847, 1,644, 1,603, 1,575, 1,538, 1,508, 1,456, 1,440, 1,417, 1,372, 1,323, 1,295, 1,278, 1,231, 1,214, 1,203, 1,189, 1,174, 1,124, 1,115, 1,100, 1,023, 961, 936, 903, 864, 836, 789, 717, 702, 679, 647, 617, 582, 561, 529, 492, 479, 459 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 11.02 (s, 1H), 8.43 (s, 1H), 7.70 (d, *J* = 8.8 Hz, 2H), 7.31 (s, 1H), 7.22 (d, *J* = 8.4 Hz, 2H), 6.30 (s, 1H), 3.92 (t, *J* = 5.6 Hz, 2H), 3.24 (brs, 2H), 1.94 (t, J = 5.8 Hz, 2H); ESI–MS: *m/z*, 336.0 (M+H)⁺.

(E)-N'-(2,4-difluorobenzylidene)-5,6,7,8-tetrahydroimidazo [1,2-a]pyrimidine-2-carbohydrazide (8 g)

M.p: 138–139°C; IR(KBr): υ_{max} 3,429, 3,265, 3,193, 3,139, 3,120, 3,097, 3,008, 2,981, 2,962, 2,882, 2,562, 1,665,

1,617, 1,561, 1,517, 1,492, 1,459, 1,427, 1,372, 1,316, 1,282, 1,214, 1,190, 1,175, 1,140, 1,097, 1,066, 995, 945, 897, 854, 828, 820, 782, 744, 728, 718, 707, 638, 629, 610, 577, 542, 509; cm⁻¹; ¹HNMR (400 MHz, DMSO-d₆): δ 11.32 (s, 1H), 8.64 (s, 1H), 7.96–7.90 (m, 1H), 7.35 (d, J = 2.8 Hz, 1H), 7.18 (t, J = 8.8 Hz, 1H), 6.27 (s, 1H), 3.92 (t, J = 6.0 Hz, 2H), 3.24 (brs, 2H), 1.94 (t, J = 5.2 Hz, 2H); ESI–MS: m/z, 305.0 (M+H)⁺.

(E)-N'-(3,4-difluorobenzylidene)-5,6,7,8-tetrahydroimidazo [1,2-a]pyrimidine-2-carbohydrazide (8h)

M.p: 143–144°C; IR(KBr): υ_{max} 3,415, 3,307, 3,249, 3,114, 2,884, 2,559, 1,672, 1,617, 1,513, 1,437, 1,423, 1,370, 1,301, 1,283, 1,231, 1,205, 1,137, 1,110, 1,083, 988, 937, 918, 889, 876, 796, 767, 716, 636, 578, 537, 515 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 11.30 (s, 1H), 8.48 (s, 1H), 7.88 (s, 1H), 7.68-7.19 (m, 3H), 6.30 (s, 1H), 3.93 (brs, 2H), 3.40 (brs, 2H), 1.95 (brs, 2H); ESI–MS: *m/z*, 306.0 (M+H)⁺.

(E)-N'-(4-fluoro-2-methylbenzylidene)-5,6,7,8-tetrahydroimi dazo[1,2-a]pyrimidine-2-carbohydrazide (8i)

M.p: 81–82°C; IR(KBr): υ_{max} 3,419, 3,170, 3,097, 3,060, 2,967, 2,906, 2,835, 1,650, 1,633, 1,594, 1,556, 1,492, 1,456, 1,442, 1,417, 1,391, 1,379, 1,368, 1,352, 1,324, 1,300, 1,266, 1,238, 1,225, 1,194, 1,152, 1,099, 1,067, 1,004, 992, 962, 955, 896, 826, 811, 717, 698, 617, 577, 517 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 11.00 (s, 1H), 8.72 (s, 1H), 7.86 (d, *J* = 8.4 Hz, 1H), 7.30 (s, 1H), 7.10-7.05 (m, 2H), 6.28 (s, 1H), 3.92 (t, *J* = 5.6 Hz, 2H), 3.24 (brs, 2H), 1.94 (brs, 2H); ESI–MS: *m/z*, 302.10 (M+H)⁺.

(E)-N'-(2-chloro-3-(trifluoromethyl)benzylidene)-5,6,7,8-tetra hydroimidazo[1,2-a]pyrimidine-2-carbohydrazide (8j)

M.p: 117–119°C; ¹H NMR (400 MHz, DMSO-d₆): δ 11.58 (s, 1H), 8.88 (s, 1H), 8.20 (d, J = 8.4 Hz, 1H), 7.88 (d, J = 8.6 Hz, 1H), 7.58 (t, J = 7.8 Hz, 1H), 7.30 (s, 1H), 6.25 (s, 1H), 3.94 (t, J = 5.8 Hz, 2H), 3.22 (brs, 2H), 1.94 (t, J = 5.4 Hz, 2H); ESI–MS: m/z, 372.13 (M+H)⁺.

(E)-N'-((benzofuran-2-yl)methylene)-5,6,7,8-tetrahydroimida zo[1,2-a]pyrimidine-2-carbohydrazide (8k)

M.p: 117–119°C; IR(KBr): υ_{max} 3,366, 3,239, 3,135, 2,964, 2,930, 2,881, 1,659, 1,607, 1,574, 1,560, 1,506, 1,492, 1,474, 1,448, 1,406, 1,366, 1,331, 1,317, 1,290, 1,242, 1,219, 1,196, 1,158, 1,144, 1,131, 1,106, 1,095, 1,073, 1,004, 992, 952, 943, 891, 873, 839, 824, 764, 715, 659, 613, 580, 544, 468, 459, 440 cm⁻¹; ¹H NMR (400 MHz, DMSO-d₆): δ 11.30 (s, 1H), 8.50 (s, 1H), 7.60–7.48 (m, 2H), 7.32–7.28 (m, 1H), 7.18 (s, 1H), 6.30 (s, 1H), 3.94 (t, J = 5.8 Hz, 2H), 3.40 (brs, 2H), 1.80 (brs, 2H); ESI–MS: m/z, 310.14 (M+H)⁺.

Conclusion

The synthesis of 5,6,7,8-tetrahydro-imidazo[1,2-a] pyrimidine-hydrazone derivatives **8a–k** described in this paper was prepared in six elaborate steps from commercially available 2-aminopyrimidine as starting material. All the synthesised compounds 8a–k were investigated for their in vitro antimicrobial activity. Compounds **8d**, **8e** and **8f** demonstrated excellent antibacterial activity with zone of inhibition 30–33 mm against *E. coli* (Gram negative bacteria) and *S. aureus* (Gram positive bacteria) These compounds also displayed excellent antibacterial activity with zone of inhibition 22–25 mm even in the case of *P. aeruginosa*(Gram negative bacteria) and *S. pyogenes*(Gram positive bacteria).

Authors' contributions

KS carried out the total experimental work under the guidance of MT. ELS performed the statistical analysis and drafted the manuscript. All authors read and approved the final manuscript.

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Competing interests

The authors declare that they have no competing interests.

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