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Investigation of Energy Band at Atomic-Layer-Deposited ZnO/ β -Ga₂O₃ ($\bar{2}01$) Heterojunctions

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Abstract

The energy band alignment of ZnO/ β -Ga₂O₃ ($\bar{2}01$) heterojunction was characterized by X-ray photoelectron spectroscopy (XPS). The ZnO films were grown by using atomic layer deposition at various temperatures. A type-I band alignment was identified for all the ZnO/ β -Ga₂O₃ heterojunctions. The conduction (valence) band offset varied from 1.26 (0.20) eV to 1.47 (0.01) eV with the growth temperature increasing from 150 to 250 °C. The increased conduction band offset with temperature is mainly contributed by Zn interstitials in ZnO film. In the meanwhile, the acceptor-type complex defect $V_{Zn} + OH$ could account for the reduced valence band offset. These findings will facilitate the design and physical analysis of ZnO/ β -Ga₂O₃ relevant electronic devices.

Keywords: β -Ga₂O₃, Contacts, ZnO, ALD

Introduction

Gallium oxide (Ga₂O₃) has been widely investigated as a promising ultrawide bandgap semiconductor material for next generation power electronic devices due to its unique properties [1]. Among various polymorphs (α , β , γ , δ , and ϵ), monoclinic β -Ga₂O₃ has the most thermal stability [2]. In addition, β -Ga₂O₃ has a room temperature bandgap of 4.5~4.9 eV, and excellent chemical stability [3]. Especially, β -Ga₂O₃ has a high bulk electron mobility of ~100 cm²/V·s, much higher breakdown field of 8 MV/cm than that of SiC (3.18 MV/cm) or GaN (3 MV/cm) [4], and the carrier concentration can be easily modulated by doping Sn and Si [5, 6]. Therefore, β -Ga₂O₃-based devices including solar-blind photodetectors [7] and metal-oxide-semiconductor field-effect transistors (MOSFETs) [8] have been reported. However, limitations still exist in β -Ga₂O₃-based devices, such as the poor ohmic contact between the metal and β -Ga₂O₃ [9]. In recent year, inserting a high electron concentration metal-oxide-semiconductor interlayer, i.e., intermediate semiconductor layer (ISL) between the metal and Ga₂O₃, has been shown to be an

effective resolution because the modulation of energy barrier at the interface [10–12].

Zinc oxide (ZnO) has attracted much attention because it has a large exciton binding energy of 60 meV, a high electron concentration of $>10^{19}$ cm⁻³, and a strong cohesive energy of 1.89 eV. [13, 14] Additionally, the lattice mismatch between ZnO and Ga₂O₃ is within 5% [15]. Various deposition techniques have been developed to prepare ZnO film, including hydrothermal method [16, 17] and chemical vapor deposition (CVD). [18] However, hydrothermal method need a complicated process and the grow rate is quiet slow, and CVD generally requires quiet high growth temperature above 900 °C. These drawbacks make it challenging to be applied in devices. Recently, atomic layer deposition (ALD) has emerged as a promising technique, which exhibits excellent step coverage, atomic scale thickness controllability, good uniformity, and a relatively low deposition temperature. Consequently, atomic-layer-deposited ZnO on wide-bandgap semiconductors can reduce interface disorder and yield more controllable sample to examine the energy band alignment, which plays an important role in the carrier transport process [19]. Up to now, band alignment between Ga₂O₃ and atomic-layer-deposited ZnO has not been studied by experiment, although there are some reports about the

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theoretical band alignment of ZnO and Ga₂O₃. [20] Therefore, understanding the energy band alignment of atomic-layer-deposited ZnO/ β -Ga₂O₃ heterojunction is highly desirable for the design and physical analysis of relevant devices in the future. In this work, the energy band alignment of atomic-layer-deposited ZnO on β -Ga₂O₃ was characterized by X-ray photoelectron spectroscopy (XPS). Moreover, the influence of growth temperature of ZnO on the band alignment was also addressed.

Methods

β -Ga₂O₃ ($\bar{2}01$) substrates with a Sn doping concentration of $\sim 3 \times 10^{18}/\text{cm}^3$ were diced into small pieces with the size of $6 \times 6 \text{ mm}^2$. The diced samples were alternately cleaned in acetone, isopropanol by ultrasonic cleaning for each 10 min, subsequently rinsed with de-ionized water to remove residual organic solvents. After that, Ga₂O₃ substrates were transferred into an ALD reactor (Wuxi MNT Micro Nanotech co., LTD, China). The growth rate of ZnO films was $\sim 1.6 \text{ \AA}/\text{cycle}$. Both 40 and 5 nm ZnO films were grown on cleaned β -Ga₂O₃ using Zn (C₂H₅)₂ (DEZ) and H₂O at each temperature of 150, 200, and 250 °C, respectively. The thickness of prepared ZnO films was measured by Ellipsometer (Sopra GES-5E). The ZnO(40 nm)/ β -Ga₂O₃ was used as bulk standard, and the ZnO(5 nm)/ β -Ga₂O₃ was used to determine the band alignment, in the meanwhile the bare bulk β -Ga₂O₃ was used as the control sample. XPS (AXIS Ultra DLD, Shimadzu) measurements with a step of 0.05 eV were performed to measure the valence band maximum (VBM), Ga 2p and Zn 2p spectra. To avoid interference of surface oxidation and contamination, all samples were etched by Ar ion for 3 min with a voltage of 2 kV before XPS measurement. Note that all the XPS spectra were calibrated by C 1s peak at 284.8 eV for compensating the charging effect. To identify the band-gap, the optical transmittance spectra of Ga₂O₃ and ZnO were measured by ultraviolet-visible (UV-VIS) spectroscopy (Lambda 750, PerkinElmer, USA).

Results and Discussion

Figure 1 shows the variation of $(\alpha hv)^{1/n}$ as a function of photon energy for bulk β -Ga₂O₃ and the as-grown ZnO film deposited at 200 °C. The optical band gap (E_g) of the ZnO film and β -Ga₂O₃ can be determined by the Tauc's relation [21]: $(\alpha hv)^{1/n} = A(hv - E_g)$, where α is the absorption coefficient, A is a constant, $h\nu$ is the incident photon energy, E_g is the optical energy bandgap, n is 1/2 for the direct bandgap, and 2 for the indirect bandgap. Here, both ZnO and β -Ga₂O₃ have typical direct band gap that make the value of n is 1/2. Subsequently, E_g can be extracted by extrapolating the straight line portion to the energy bias at $\alpha = 0$. Therefore, the extracted E_g of

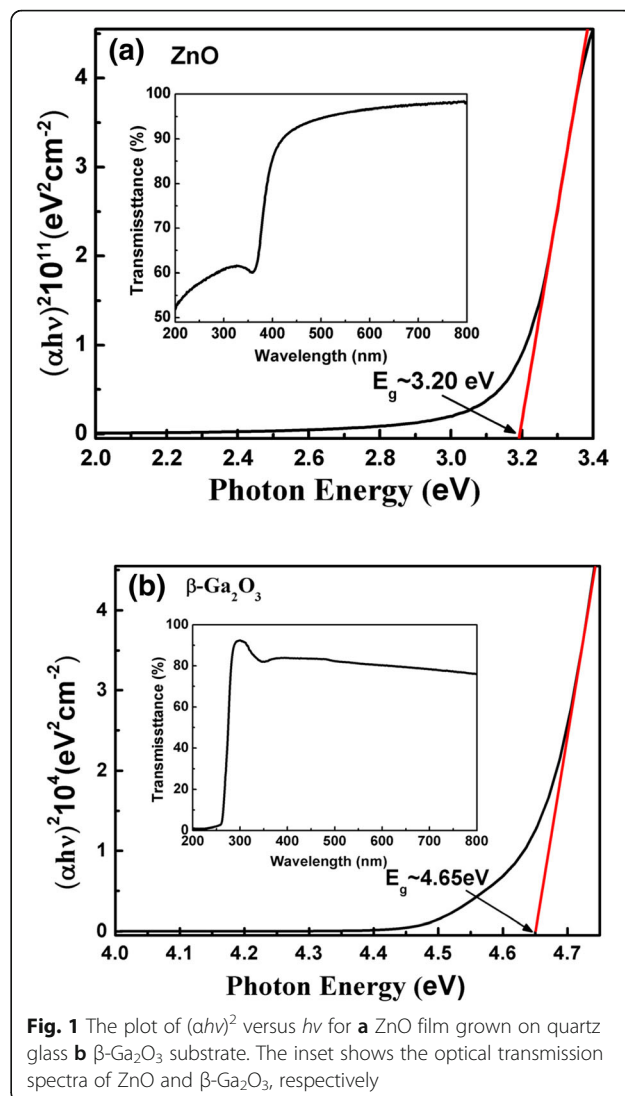


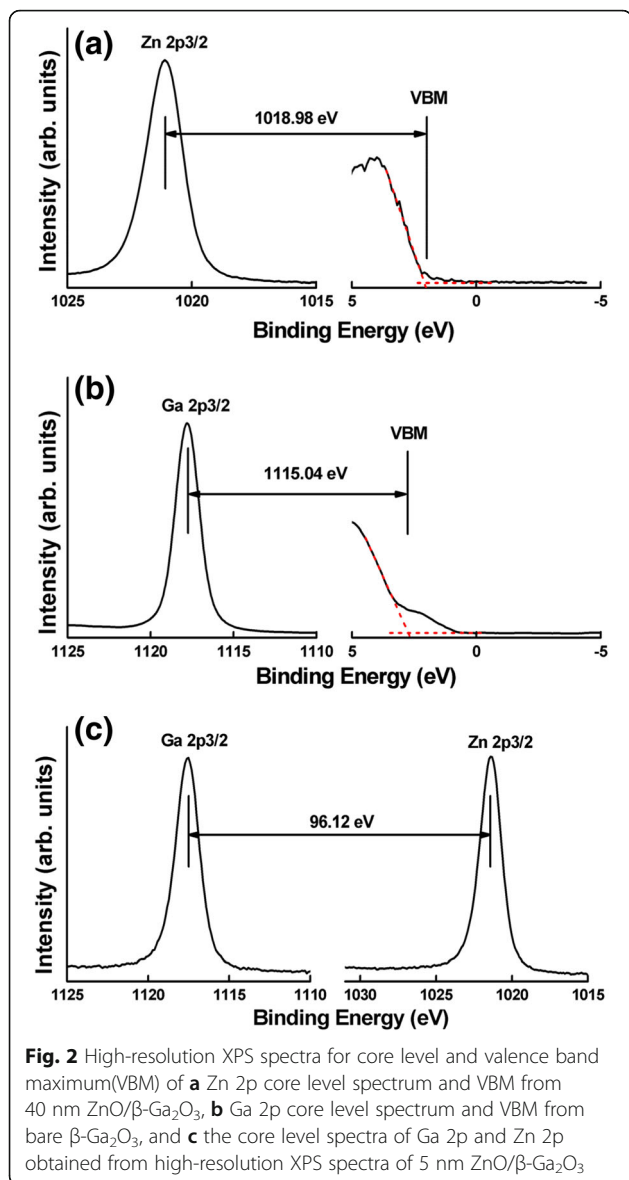
Fig. 1 The plot of $(\alpha hv)^2$ versus $h\nu$ for **a** ZnO film grown on quartz glass **b** β -Ga₂O₃ substrate. The inset shows the optical transmittance spectra of ZnO and β -Ga₂O₃, respectively

ZnO and β -Ga₂O₃ are 3.20 eV and 4.65 eV, respectively, in good agreement with the reported. [22, 23]

The valence band offset (VBO) can be determined by Kraut's method using the following formula [24]

$$\Delta E_V = \left(E_{Ga\ 2p}^{Ga_2O_3} - E_{VBM}^{Ga_2O_3} \right) - \left(E_{Zn\ 2p}^{ZnO} - E_{VBM}^{ZnO} \right) - \left(E_{Ga\ 2p}^{Ga_2O_3} - E_{Zn\ 2p}^{ZnO} \right), \quad (1)$$

where $E_{Ga\ 2p}^{Ga_2O_3} - E_{VBM}^{Ga_2O_3}$ ($E_{Zn\ 2p}^{ZnO} - E_{VBM}^{ZnO}$) represents to the energy difference between Ga 2p (Zn 2p) core level (CL) and VBM of bulk β -Ga₂O₃ (ZnO), and $E_{Ga\ 2p}^{Ga_2O_3} - E_{Zn\ 2p}^{ZnO}$ denotes as the energy difference between Ga 2p and Zn 2p core levels. Figure 2 shows all CL spectra including Zn 2p of ZnO (40 nm)/ β -Ga₂O₃ and ZnO (5 nm)/ β -Ga₂O₃, Ga 2p of bulk Ga₂O₃ and ZnO (5 nm)/ β -Ga₂O₃, as well as valence band spectra from bulk Ga₂O₃ and ZnO (40 nm)/ β -Ga₂O₃. Figure 2a presents the CL spectra of



Zn 2p on the ZnO (40 nm)/ β -Ga₂O₃, which is quite symmetrical indicating the uniform bonding state, and the peak locates at 1021.09 eV corresponds the Zn-O bond [25]. The VBM can be determined using a linear extrapolation method [26]. The VBM of ZnO is located at 2.11 eV. In Fig. 2b, the peak located at 1117.78 eV corresponds to the Ga-O bond [27] and the VBM of Ga₂O₃ is deduced to be 2.74 eV according to the method mentioned above. The CLs of Zn 2p and Ga 2p in the ZnO (5 nm)/ β -Ga₂O₃ are shown in Fig. 2c. According to Eq. (1), the VBO at the interface of ZnO/Ga₂O₃ is determined to be 0.06 eV.

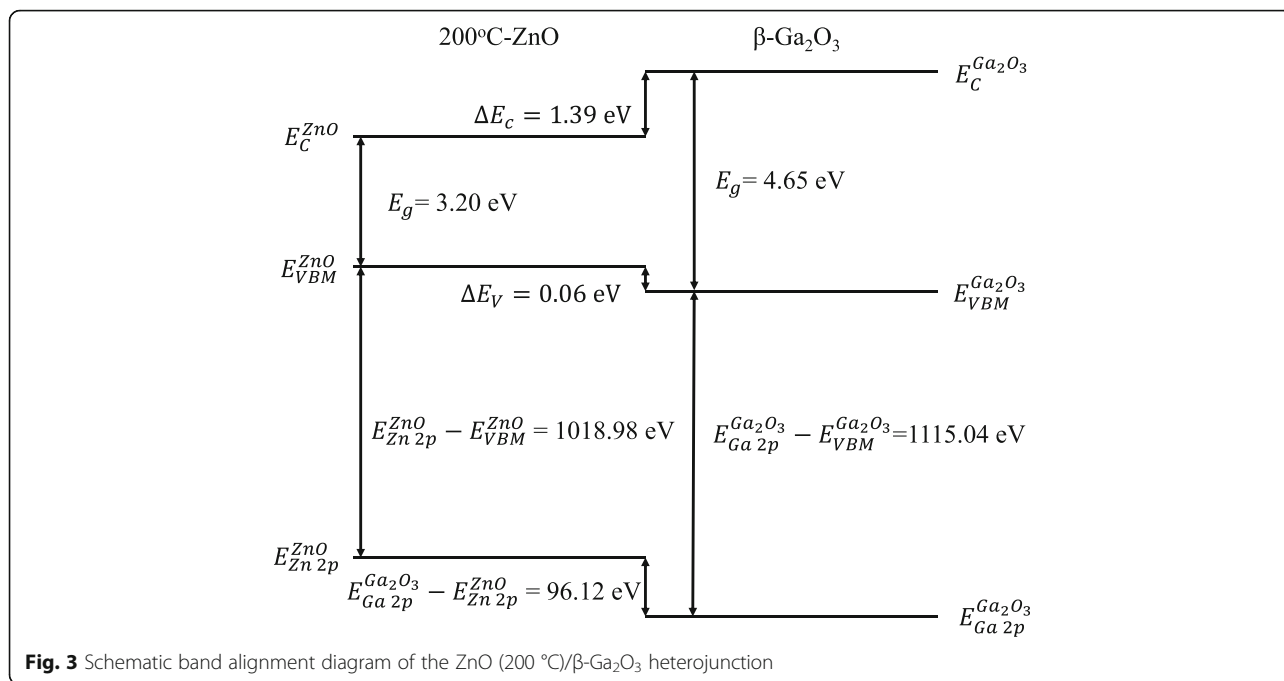
Based on the calculated E_g and ΔE_V , the conduction band offset (CBO) at the ZnO/Ga₂O₃ interface can be easily deduced from the following equation:

$$\Delta E_C = E_g^{Ga_2O_3} - E_g^{ZnO} - \Delta E_V, \quad (2)$$

where $E_g^{Ga_2O_3}$ and E_g^{ZnO} are the energy bandgap for β -Ga₂O₃ and ZnO, respectively. The detailed energy band diagram for ZnO/ β -Ga₂O₃ is depicted in Fig. 3. The interface has a type-I band alignment, where both conduction and valence band edges of ZnO are located within the bandgap of β -Ga₂O₃.

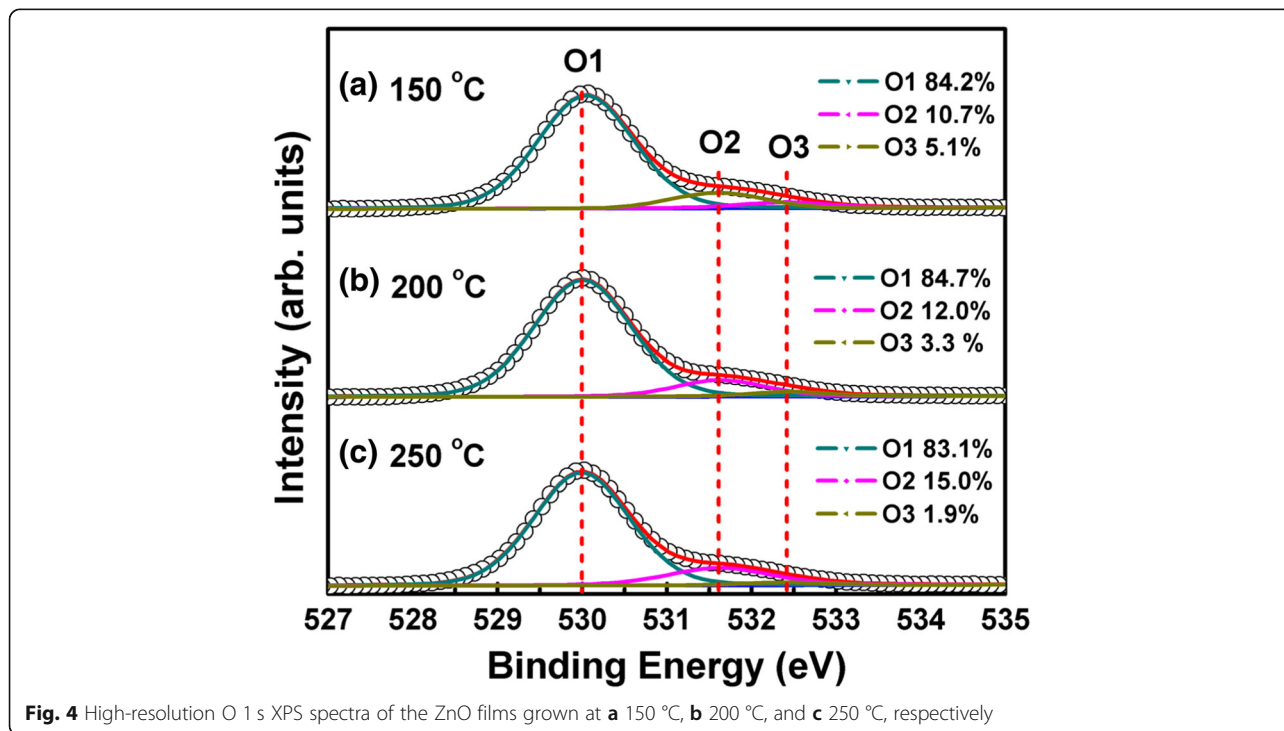
To further examine the effect of the growth temperature on the band alignment between ZnO and β -Ga₂O₃, the ZnO films are also grown at 150 and 250 °C. Note that ZnO films prepared by ALD at the temperatures of 150–250 °C have poly-crystalline nature. Figure 4 shows the high-resolution O 1s XPS spectra of the ZnO films grown at different temperatures. Each O 1s spectrum can be well separated into three components using Gaussian-Lorentzian function. The peaks centered at 530.0 (O1), 531.6 (O2), and 532.4 (O3) eV correspond to the Zn-O bands, oxygen vacancies, and -OH group [28, 29], respectively. The relative percentage of different components is also calculated according to the peak area, digested in Fig. 4. It shows that the relative content of oxygen vacancies increases from 10.7 to 15.0% due to the decomposition of precursors and the increase of Zn interstitials. However, the -OH counterpart reduces from 5.1 to 1.9% because of more complete reactions between DEZ precursors and surface -OH groups in this temperature range [30].

Figure 5 shows the band offsets of ZnO/ β -Ga₂O₃ heterojunctions as a function of growth temperature. The CBO increases from 1.26 to 1.47 eV with the growth temperature varying from 150 to 250 °C. The native donor defects include the Zn anti-position, oxygen vacancies, and Zn interstitials. However, the formation energy of anti-position atoms is so high that its concentration is extremely low. The Zn interstitials have more influence on the conduction band minimum (CBM) than that of the oxygen vacancy because the CBM is mainly dominated by the 4s orbit of Zn atom. [31] As a result, the increased CBO of 0.21 eV could be mainly contributed by Zn interstitials. On the other hand, the VBO reduces from 0.20 to 0.01 eV with the growth temperature increasing from 150 to 250 °C. The native acceptor defects include the O anti-position, Zn vacancies, and oxygen interstitials [32], whose formation energies are high and their number can be even negligible. Furthermore, the most native acceptor levels are deep within the ZnO bandgap, thus they have little effect on the VBM [33]. However, $V_{zn} + OH$ is favorable to be presented due to the low formation



energy, [34] $V_{Zn} + OH$ may occur with an electron belonging to OH bonds. The lattice hydrogen H^+ ion acts as a compensating center, and it can bind with the V_{Zn} around the dislocation and stacking faults core, ensuring the acceptor-type complex defect for p-type conductivity [35]. More residual $-OH$ groups in the ZnO film are obtained at a lower

growth temperature, i.e., 150 °C [36]. The acceptor level near the VBM reduces with the temperature, leading to an effectively downward shift in E_V of ZnO, thus the ΔE_V becomes lower. Therefore, the ZnO deposited at lower temperature could be more efficiently to reduce the barrier height at the interface between the metal and Ga₂O₃.



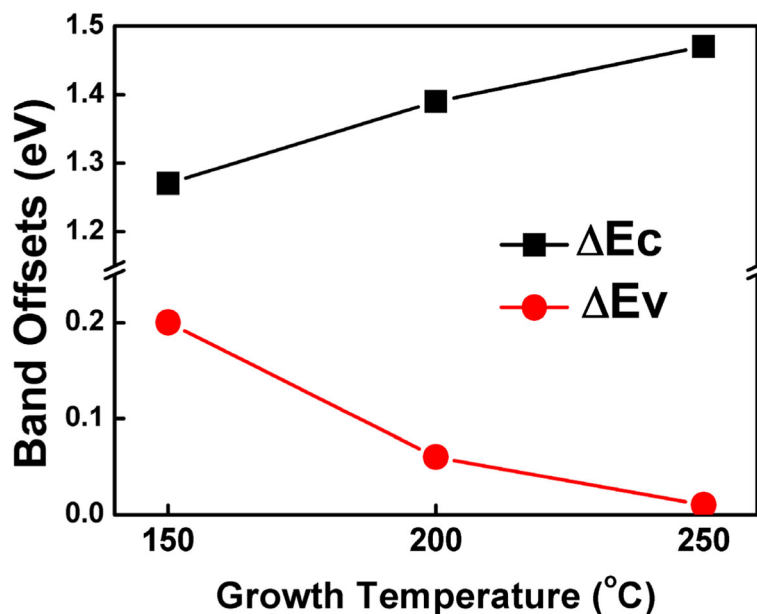


Fig. 5 The conduction and valence band offsets of atomic-layer-deposited ZnO/ β -Ga₂O₃ heterojunctions fabricated at different temperatures

Conclusions

In summary, the energy band alignment at atomic-layer-deposited ZnO/ β -Ga₂O₃ ($\bar{2}01$) was characterized by XPS. A type-I band alignment formed at the ZnO/ β -Ga₂O₃ interface. The conduction band offset increased from 1.26 to 1.47 eV while the valence band offset decreased from 0.20 to 0.01 eV with the temperature increasing from 150 to 250 °C. These observations suggest that the ZnO deposited at lower temperature is favorable to be a promising ISL to reduce the electron barrier height at the ZnO/ β -Ga₂O₃ interface.

Abbreviations

ALD: Atomic layer deposition; CBM: Conduction band minimum; CBO: Conduction band offset; CVD: Chemical vapor deposition; DEZ: Zn (C₂H₅)₂; Ga₂O₃: Gallium oxide; GaN: Gallium nitride; ISL: Intermediate semiconductor layer; MOSFETs: Metal-oxide-semiconductor field-effect transistors; OH: Hydroxyl; SiC: Silicon carbide; UV-VIS: Ultraviolet-visible spectroscopy; VBM: Valence band maximum; VBO: Valence band offset; XPS: X-ray spectroscopy; ZnO: Zinc oxide

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Availability of Data and Materials

The datasets supporting the conclusions of this manuscript are included within the manuscript.

Authors' Contributions

SMS, YFX, YWH, and HL conducted the extensive experiments and analyzed the data. WJL and SJD supervised the project and wrote the manuscript. DWZ helped to review and discuss the manuscript. All authors read and approved the final manuscript.

Competing Interests

The authors declare that they have no competing interests.

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