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Electronic Structure and Charge-Trapping Characteristics of the Al₂O₃-TiAlO-SiO₂ Gate Stack for Nonvolatile Memory Applications

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Abstract

In this work, high-*k* composite TiAlO film has been investigated as charge-trapping material for nonvolatile memory applications. The annealing formed Al₂O₃-TiAlO-SiO₂ dielectric stack demonstrates significant memory effects and excellent reliability properties. The memory device exhibits a large memory window of ~2.6 V under ±8 V sweeping voltage, and it shows only ~14% charge loss after more than 10 years' retention, indicating excellent charge retention properties. The electronic structures of the Al₂O₃-TiAlO-SiO₂ have been studied by X-ray photoelectron spectroscopy measurements, and it reveals that the quantum well and the defect traps in TiAlO film can provide a >1.8 eV deep barrier for charge confinement in the TiAlO layer. The mixing between Al₂O₃ and TiO₂ can increase the defects related to the under-coordinated Ti³⁺ atoms, thereby enhancing the charge-trapping efficiency of the device. Our work implies that high-*k* TiAlO composite film is promising for applications in future nonvolatile charge-trapping memories.

Keywords: Charge trapping, Nonvolatile memory, High-*k* dielectrics, TiAlO

Background

Among the family of nonvolatile flash memories, charge-trapping memory (CTM) devices such as silicon-oxide-nitride-oxide-silicon (SONOS)-type memory device receive a lot of attention due to its low-operating voltage, fast program/erase (P/E) speed, good endurance, and retention characteristics over the floating-gate devices [1–3]. For these advantages, SONOS-type CTM device has been considered as a promising candidate for the next-generation nonvolatile flash memory. However, some performance and reliability issues such as low charge-trapping efficiency and poor retention characteristics still exist in SONOS-type CTM devices with the continual scaling. To overcome these disadvantages, various charge-trapping materials as well as blocking and tunneling materials were extensively investigated. Among which, high-*k* materials including thin films and

their nanocrystals, for example HfO₂, TiO₂, and ZrO₂, have been proposed as the charge-trapping layer in the CTM devices to achieve better storage performance and retention characteristics [4–7]. In these high-*k* dielectrics, oxygen vacancy is verified as the main origin of the defects in the film [8]. Depending on the crystal structure and method of deposition [9], TiO₂ has a high permittivity of 80–110, which is favorable for low-voltage operation of the memory device. Another feature of TiO₂ is that Ti has several stable oxidation states of Ti³⁺ and Ti⁴⁺, which leads to a well-known phenomenon with materials containing Ti-O bonds: a reduced oxide. Such a reduced oxide has many oxygen vacancies, which can act as charge-trapping centers [10]. Therefore, it is expected that TiO₂ will be a good candidate used as the charge-trapping material, which can provide high charge-trapping ability as well as low device operation voltage. However, TiO₂ does not have good insulating quality due to a small bandgap and low crystallization temperature [9], which is not favorable for long-term stability of the memory device. Alumina (Al₂O₃) has a large bandgap (~8.7 eV) and large band offsets with Si substrate [9, 11–14] and is amorphous up to high

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temperatures. The drawback of Al_2O_3 is that it only has a $k \sim 8\text{--}10$ [9], which is not favorable for the huge reduction of the operation voltage. Therefore, it is expected that the composite film of TiAlO may combine the advantages of TiO_2 and Al_2O_3 , which can have high charge-trapping ability, high permittivity, high thermal stability, and low leakage current at the same. Compared with the commonly studied charge-trapping materials like metal nanocrystals or monophasic high- k layer, the above mentioned advantages of TiAlO composite film show potentially more favorable for the high-performance operation in the future charge-trapping flash memory devices.

In this work, we first fabricated the high- k $\text{Al}_2\text{O}_3/\text{TiO}_2/\text{Al}_2\text{O}_3$ stacking structure by electron beam evaporation. The TiAlO composite film will form by high-temperature annealing of the nominated $\text{Al}_2\text{O}_3/\text{TiO}_2/\text{Al}_2\text{O}_3$ dielectric stack, and finally, we got the $\text{Al}_2\text{O}_3/\text{TiAlO}/\text{SiO}_2$ memory device structure. The nonvolatile memory device using TiAlO composite film as charge-trapping material shows significant memory effect and excellent long-term charge stability. Although further work is still necessary to improve the overall device properties like increasing the programming speed, the TiAlO composite film is very promising for its applications in future high-performance nonvolatile memory devices.

Methods

P-type Si (100) substrates with $\rho = 1\text{--}10 \Omega \text{ cm}$ were first cleaned by wet-chemical solution and then dipped in a diluted HF solution (1%) to remove the surface oxide. The wafers were then immediately loaded into a vacuum chamber for deposition. The nominated Al_2O_3 (5 nm)- TiO_2 (10 nm)- Al_2O_3 (15 nm) structure was deposited by electron beam evaporation at a substrate temperature of 300 °C. After deposition, the films were annealed at a high temperature of 900 °C in O_2 atmosphere for 5 min by rapid thermal annealing. Dot-shaped Au top electrodes with an area of $\sim 3.14 \times 10^{-4} \text{ cm}^2$ were deposited on the surface of the samples using a shadow mask by vacuum evaporation. The electrical properties of the CTM devices were characterized by an Agilent E4980A impedance analyzer and an Agilent B1500A high-precision semiconductor analyzer at room temperature. High-resolution transmission electron microscopy (HRTEM) was used to study the cross-sectional microstructures of the tri-layer dielectric stack (Tecnai G2 F20 S-Twin). The electronic structure of the memory dielectric stack was investigated by using X-ray photoelectron spectroscopy (ULVAC-PHI, PHI 5000 Versa Probe) with Al $K\alpha$ X-ray source (1486.6 eV). The defect states and defect levels in our electron beam evaporation deposited TiAlO films were studied by photoluminescence (PL) (Horiba HR Revolution) measurements under 325-nm excitation wavelength.

Results and Discussion

Figure 1a, b shows the schematic diagrams of the tri-layer dielectric stack before and after high-temperature annealing. Before annealing, a nominated Al_2O_3 (5 nm)- TiO_2 (10 nm)- Al_2O_3 (15 nm) charge-trapping memory structure was deposited by electron beam evaporation. After 900 °C annealing in oxygen for 5 min, the memory structure will change to a SiO_2 (~ 4.2 nm)-TiAlO (~ 14 nm)- Al_2O_3 (~ 16.5 nm) structure, as shown in Fig. 1b. Figure 1c shows the cross-sectional HRTEM image of the actual memory structure after annealing. One amorphous SiO_2 layer of around 4.2 nm can be clearly observed. Under high-temperature annealing, the formation of interfacial SiO_2 layer has been often observed [7]. Noticeably, the 5-nm-thick tunneling layer of Al_2O_3 disappeared, which was ascribed to the mixture of the tunneling Al_2O_3 layer and TiO_2 . It is most probably that the mixture in the interface between the blocking Al_2O_3 layer and TiO_2 also occurred. The thickness of the composite TiAlO layer and Al_2O_3 blocking layer are ~ 14 and 16.5 nm, respectively. The $\text{TiO}_2/\text{Al}_2\text{O}_3$ mixture under high temperature has been reported by other researchers [15–17]. For example, in work by Mikhelashvili et al. [16] after annealing at 950 °C, the Al_2O_3 - TiO_2 nanolaminates transformed into a TiAlO layer with a nearly uniform distribution of Al and Ti oxides across the structure. Actually, this behavior is initiated already at 550 and at 750 °C in their work. Further, XPS study in our work also confirmed this mixture between TiO_2 and Al_2O_3 . Therefore, the actual high-temperature annealed memory dielectric stack in this work is Al_2O_3 -TiAlO- SiO_2 .

Figure 2a shows the typical high-frequency (1 MHz) capacitance-voltage (CV) characteristics of the Au- Al_2O_3 -TiAlO- SiO_2 -Si devices under ± 4 and ± 16 V. The anticlockwise direction of the CV hysteresis demonstrates the typical hysteresis loop directions of charge trapping. To confirm that memory effect is really from TiAlO composite film, we also fabricated a single layer of 20-nm-thick Al_2O_3 using electron beam evaporation, and it also receives an annealing at 900 °C for 5 min in O_2 . Figure 2b shows the typical high-frequency (1 MHz) CV characteristics of the Au- Al_2O_3 - SiO_2 -Si devices under ± 4 and ± 10 V. It was clear that no obvious memory effect can be observed for the diode device based on single-layer Al_2O_3 film. Only a hysteresis loop around 0.21 V was observed for both scanning voltages of ± 4 and ± 10 V. The results shown in Fig. 2b proved that the significant memory effect of Al_2O_3 -TiAlO- SiO_2 really comes from the charge trapping of TiAlO layer. For both of the Al_2O_3 -TiAlO- SiO_2 and Al_2O_3 - SiO_2 gate stack structure, a small hysteresis loop can be observed under ± 4 V scanning voltages, which is most probably due to charge trapping/de-trapping in the shallow traps in the Si/ SiO_2 interface.

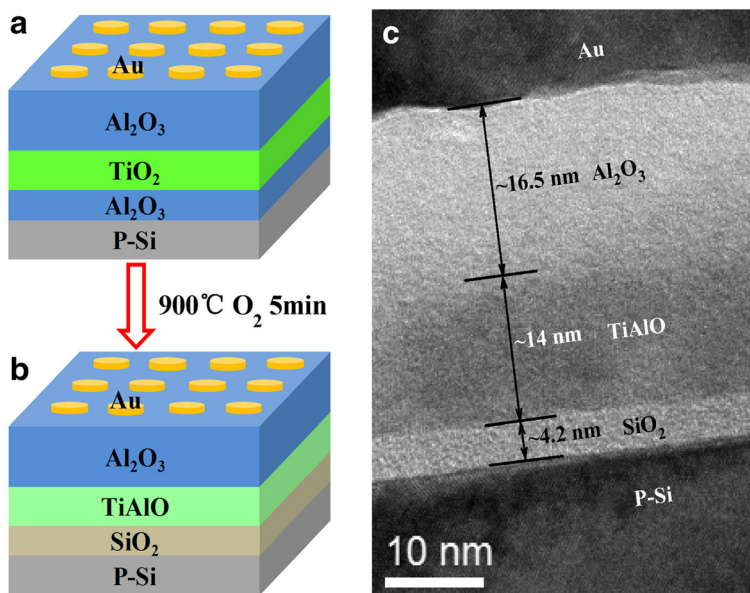


Fig. 1 (Color online) **a** The schematic diagram of the nominated Al₂O₃/TiO₂/Al₂O₃ charge-trapping memory device structure. **b** The schematic diagram of the charge-trapping memory device structure after 900 °C annealing. **c** The HRTEM cross-sectional image of the actual charge-trapping memory device used in our work

Figure 2c shows the dependence of the memory window on the different sweeping voltages. The memory window increases with the increase of the sweeping voltages, and it tends to be saturated around 16 V sweeping voltages. A large memory window of 5.5 V under ±16 V sweeping voltage indicates significant charge-trapping effect in the TiAlO

charge-trapping layer. Even for small scanning gate voltages of ±6 and ±8 V, comparatively large memory windows of 1.8 and 2.6 V can be observed. The P/E characteristics of the memory device were also investigated by measuring the flat voltage shift (ΔV_{fb}) induced by a pulsed P/E voltage. Pulses with different pulse height (ranging from 3 to 12 V)

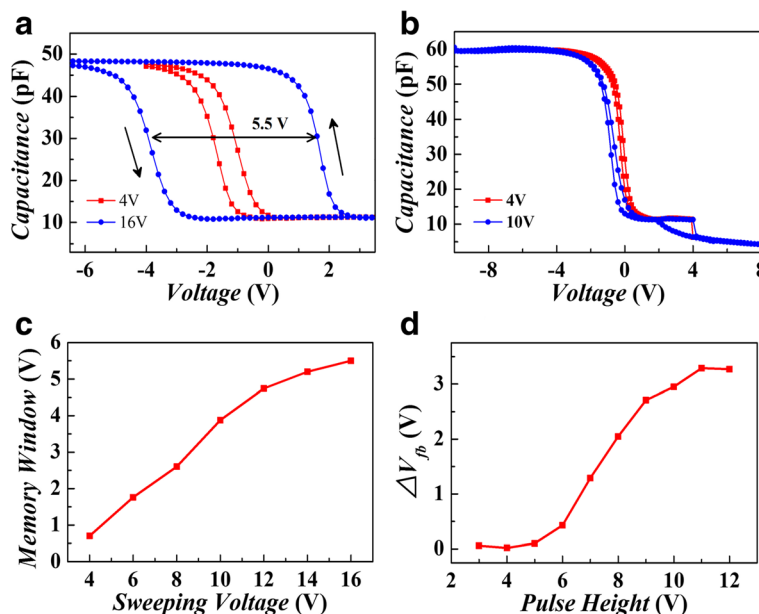


Fig. 2 (Color online) Typical high-frequency (1 MHz) CV characteristics of the Au-Al₂O₃-TiAlO-SiO₂-Si device (**a**) and Au-Al₂O₃-SiO₂-Si device (**b**). **c** Memory window width dependence of the Au-Al₂O₃-TiAlO-SiO₂-Si device under different sweeping voltages. **d** Dependence of the memory window width on the pulsed writing voltage height, the program/erase pulse width is fixed to be 1 s

and same pulse width (1 s) were applied onto the diode. The change of the flat-band voltage (V_{fb}) shift with the pulse height was shown in Fig. 2d. The V_{fb} calculation method is the same to that we used in previous work [18]. Memory effect can be clearly observed even under a small P/E voltage of 6 V. Large ΔV_{fb} value of ~ 3.0 V can be observed under 10 V P/E voltages, demonstrating its low-voltage operations compared with the conventional flash memories. It should be mentioned that the operation speed of the present device is still low. One of the reasons may be that the films were deposited by electron beam evaporation, which is likely to have lots of defects that could limit the P/E performance of the memory device. Charge-trapping memory devices fabricated by atomic layer deposition (ALD) exhibits excellent electrical performances [12, 19]. In the future, if we can use high-quality film growth method such as ALD to fabricate the dielectric stack, the electrical performance of the present TiAlO charge-trapping memory device may have a lot of space to be further improved.

Figure 3a illustrates the endurance property of the memory device at room temperature. After 10^3 P/E processes, the memory window hardly changes, indicating good endurance characteristics. As shown in Fig. 3b, the memory device also exhibits excellent retention characteristics. The high and low capacitance demonstrates to be very stable under -1 V read voltage. After 10^4 s' retention, only 2% of the high capacitance was lost. Even extrapolated for 10 years, only 14% loss of the high capacitance was observed. In the low capacitance remains nearly unchanged during all the measurement time. Figure 3c shows the leakage current characteristics of the memory device. The leakage current under -10 V bias voltage is determined to be 8.0×10^{-5} A/cm². For the P/E process by using pulse with 10 V in height and 1 s in width, the tunneling current responsible for charge-trapping memory effect can be estimated to be 4.78×10^{-7} A/cm², which is much smaller than that of the leakage current. This might be the main reason of the long program and erase time. The reason for this big difference is not very clear now, which is assumed to be closely related to the film quality deposited by electron beam evaporation. If high-quality film growth method such

as atomic layer deposition is used to deposit TiAlO charge-trapping material and Al₂O₃ blocking oxide, the P/E speed of the memory device can be expected to have a big improvement.

Considering the charge-trapping memory devices, electronic structure of the tri-layer dielectric stack (Al₂O₃-TiAlO-SiO₂) should be an important mechanism to affect the memory characteristics. The band alignment of the Al₂O₃-TiAlO-SiO₂ dielectric stack was investigated by valence band and energy loss spectra by using XPS, as shown in Fig. 4. The XPS measurement results of valence band spectra are shown in Fig. 4a. By using a linear extrapolation method [20], the valence band maximums (VBM) of P-Si substrate, SiO₂/Si, TiAlO/SiO₂/Si, and Al₂O₃/SiO₂/Si structure were determined to be 0.22 eV (E_{VBM}^{Si}), 3 eV ($E_{VBM}^{SiO_2}$), 0.8 eV (E_{VBM}^{TiAlO}), and 1.98 eV ($E_{VBM}^{Al_2O_3}$), respectively. Therefore, the valence band offsets of SiO₂/Si ($\Delta E_V^{SiO_2/Si}$), TiAlO/SiO₂ ($\Delta E_V^{TiAlO/SiO_2}$), and Al₂O₃/TiAlO ($\Delta E_V^{Al_2O_3/TiAlO}$) were calculated as 2.78, -2.2 , 1.18 eV, respectively, by using the following equations:

$$\Delta E_V^{SiO_2/Si} = E_{VBM}^{SiO_2} - E_{VBM}^{Si}, \quad (1)$$

$$\Delta E_V^{TiAlO/SiO_2} = E_{VBM}^{TiAlO} - E_{VBM}^{SiO_2}, \quad (2)$$

$$\Delta E_V^{Al_2O_3/TiAlO} = E_{VBM}^{Al_2O_3} - E_{VBM}^{TiAlO}, \quad (3)$$

The bandgaps of SiO₂, Al₂O₃, and TiAlO were determined by the onset of O 1 s electron energy loss spectra [20], as shown in Fig. 4b. For SiO₂, Al₂O₃, and TiAlO, the bandgaps were determined as 9 eV ($E_g^{SiO_2}$), 7 eV ($E_g^{Al_2O_3}$), and 5 eV (E_g^{TiAlO}), respectively. The conduction band offset (CBO) of SiO₂/Si ($\Delta E_C^{SiO_2/Si}$), TiAlO/SiO₂ ($\Delta E_C^{TiAlO/SiO_2}$), and Al₂O₃/TiAlO ($\Delta E_C^{Al_2O_3/TiAlO}$) could be deduced by using the following formulas:

$$\Delta E_C^{SiO_2/Si} = E_g^{SiO_2} - E_g^{Si} - \Delta E_V^{SiO_2/Si}, \quad (4)$$

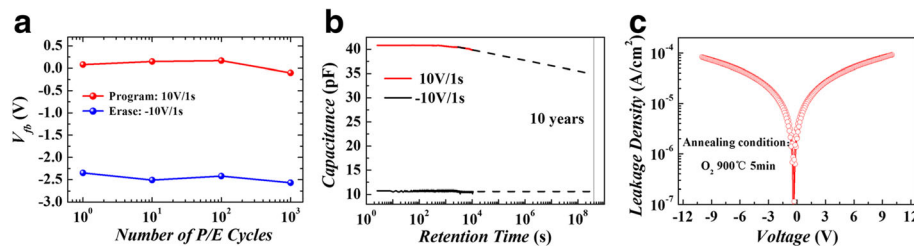
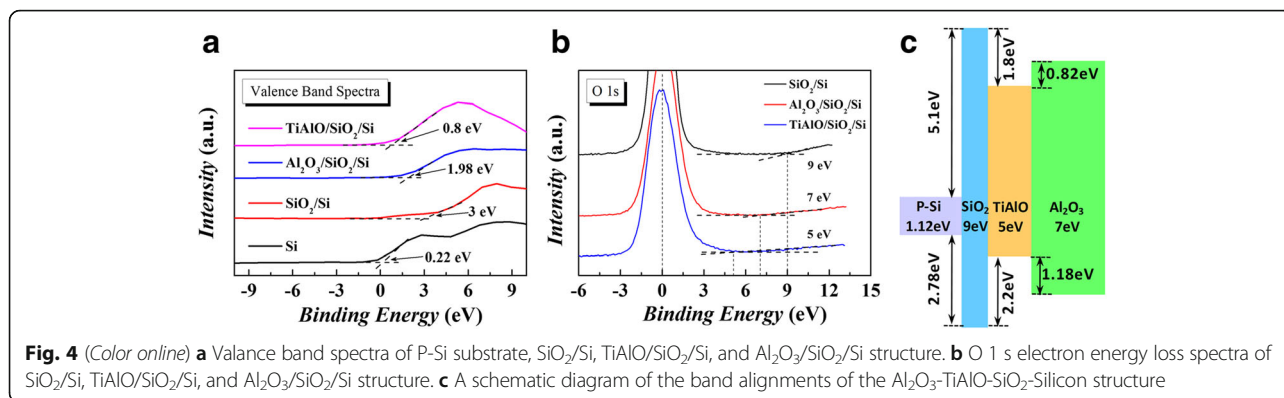


Fig. 3 (Color online) The endurance (a), retention (b), and leakage current-voltage (c) characteristics for the Au-Al₂O₃-TiAlO-SiO₂-Si device



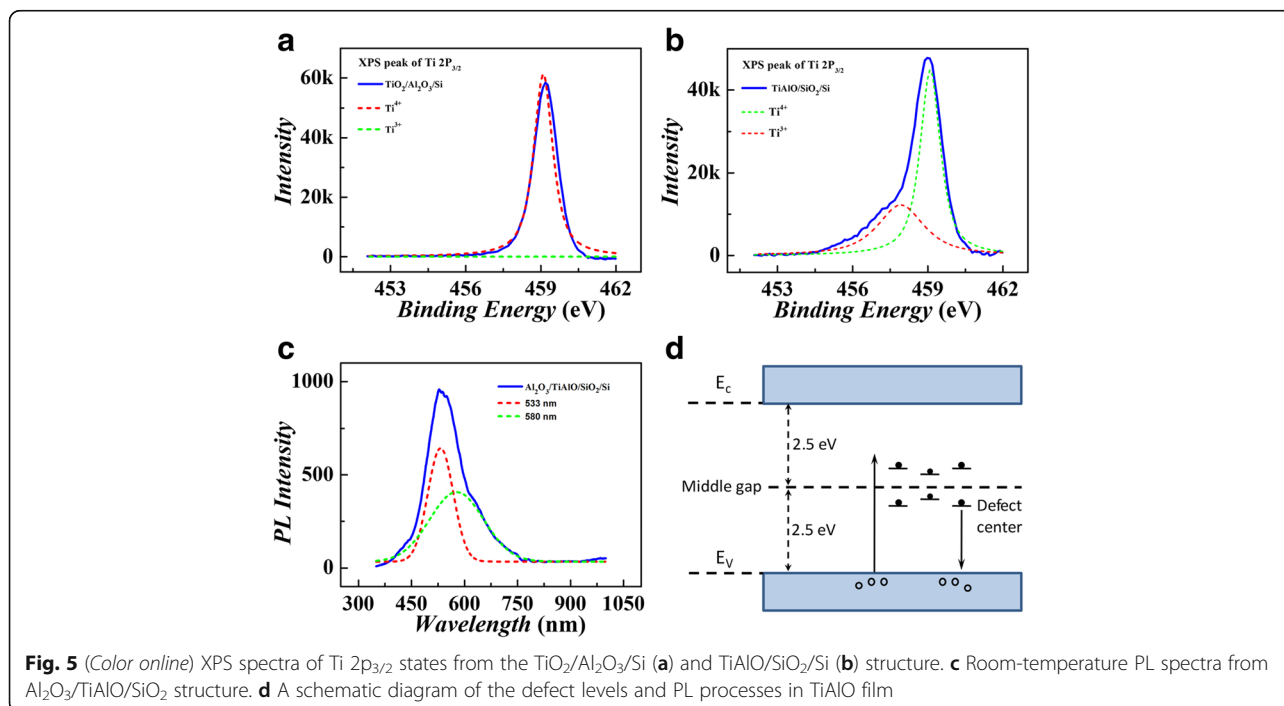
$$\Delta E_C^{\text{TiAlO/SiO}_2} = E_g^{\text{TiAlO}} - E_g^{\text{SiO}_2} - \Delta E_V^{\text{TiAlO/SiO}_2}, \quad (5)$$

$$\Delta E_C^{\text{Al}_2\text{O}_3/\text{TiAlO}} = E_g^{\text{Al}_2\text{O}_3} - E_g^{\text{TiAlO}} - \Delta E_V^{\text{Al}_2\text{O}_3/\text{TiAlO}}, \quad (6)$$

where $E_g^{\text{SiO}_2}$ is 1.12 eV. The deduced values of CBO were 5.1, -1.8, and 0.82 eV for SiO₂/Si, TiAlO/SiO₂, and Al₂O₃/TiAlO, respectively. Based on the above data, the complete band alignments of our memory devices can be established, as schematically shown in Fig. 4c. After programming, the charges will be trapped in the SiO₂-TiAlO-Al₂O₃ quantum well and the defect states of TiAlO film. Since $\Delta E_C^{\text{TiAlO/SiO}_2}$ is 1.8 eV, the electrons need to overcome at least a barrier of 1.8 eV to tunnel from the TiAlO charge-trapping layer back to the Si substrate. For the charges trapped in the defect states, for example oxygen vacancy, in the forbidden band of

TiAlO, the barrier should be higher than 1.8 eV. This deep barrier for charge trapping is assumed to be the most important reason for the excellent retention properties of the present memory device.

To investigate the charge-trapping mechanisms in our device, electron beam evaporation deposited TiAlO films were further studied by XPS measurement. Figure 5a, b shows the Ti 2p_{3/2} XPS spectra of the TiO₂/Al₂O₃/Si structures with and without 900 °C annealing, respectively. By fitting the spectrum with a Gaussian function [10, 21], the Ti 2p_{3/2} XPS spectrum of annealed structure (TiAlO/SiO₂/Si) can be resolved into two individual peaks, as shown by dotted lines in Fig. 5b. These two peaks could be ascribed to the contributions of Ti³⁺ 2p_{3/2} and Ti⁴⁺ 2p_{3/2}. Jiang et al. [10] had reported that in (TiO₂)_{0.8}(Al₂O₃)_{0.1} film part of Ti⁴⁺ ions were transformed into Ti³⁺ ions due to the diffusion between TiO₂ and



Al₂O₃. Therefore, in the TiAlO/SiO₂/Si structure, a small amount of trivalent Ti³⁺ ions formed because of the mixing of Al₂O₃ and TiO₂. Jin et al. [21] had found that the electron traps associated with the under-coordinated Ti³⁺ atoms can capture electrons. Thus, the TiAlO mixture formed by annealing can increase the defect density and may enhance the efficiency of the electron trapping. Different from Fig. 5b, the XPS spectra of the un-annealed TiO₂/Al₂O₃/Si structure shown in Fig. 5a can be well fitted by Ti⁴⁺ 2p_{3/2} peak, and Ti³⁺ 2p_{3/2} signal cannot be observed. The present Ti 2p_{3/2} XPS results proved again the mixture of alumina and titanium oxide after high-temperature annealing, and the mixed TiAlO composite film is expected to improve the efficiency of charge trapping.

The defect level in TiAlO composite film was characterized by the PL spectra, as shown in Fig. 5c. Through fitting the spectrum with a Gaussian function, the PL spectra can be divided into two separate peaks. The red dotted fitting line shows a PL peak at 533 nm, and the green dotted fitting line shows a PL peak at 580 nm. The PL peak at ~580 nm is believed to have a strong correlation with the defects associated with under-coordinated Ti³⁺ ions, and the PL peak at ~533 nm is related to the oxygen vacancies [21–23]. In addition to the defects caused by the under-coordinated Ti³⁺ ions, the charge-trapping effects in TiAlO composite film may be also partly from oxygen vacancy-related defects. Figure 5d shows a schematic diagram of the PL process in our devices. Since the bandgap of TiAlO was determined to be 5 eV by XPS, the PL emission should not come from the band to band emission. The electrons may be excited to the defect levels in the bandgap of TiAlO, whose center position is around 2.3 eV (533 nm) above the top of valence band. The excited electrons in defect levels then recombined with the holes in the valence band. Based on the PL results, we believe that the center of the defect levels in TiAlO film is located close to the middle gap of TiAlO, which are deep traps

for charge trapping in our memory devices. The deep trap levels in TiAlO composite film are expected to be one of the critical reasons for the excellent charge retention in the present memory devices.

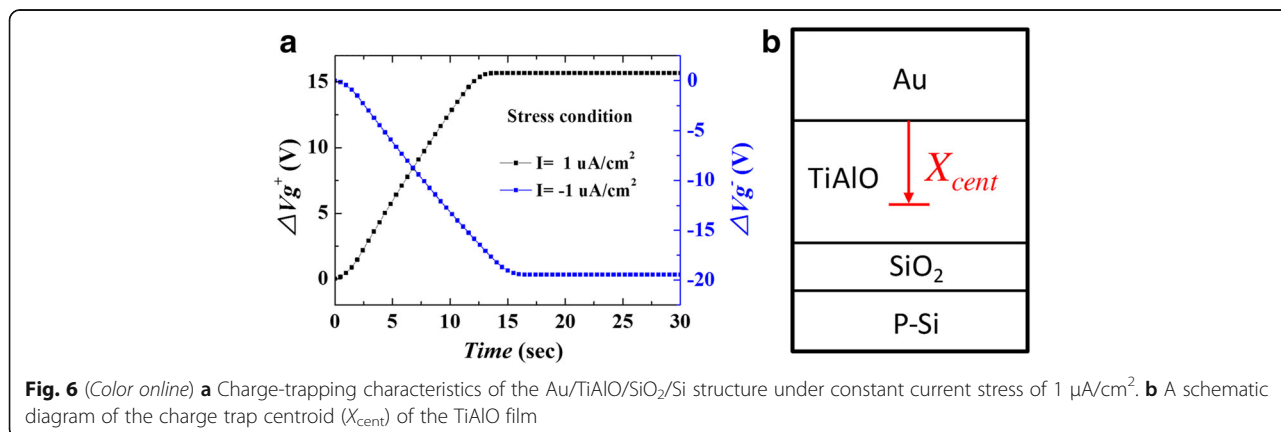
The charge trap centroid of TiAlO was evaluated using the constant current stress (CCS) method [24]. Figure 6a shows the charge-trapping characteristics under a constant current of 1 μA/cm². The voltage drop at TiAlO/SiO₂-stacked layer was measured as the gate voltage shift varied with increasing stress time. The shift in gate voltage is attributed to the charge trapping in TiAlO layer. The voltage shift was observed from the TiAlO/SiO₂-stacked layer, and the amount of voltage shift increased with stress time. Accordingly, we can conclude that the TiAlO layer has good charge-trapping characteristics. The charge trap centroid (X_{cent}) was extracted by using the CCS measurement method [24],

$$X_{cent} = \frac{t_{stack}}{\left[1 - \left(\frac{\Delta V_g^-}{\Delta V_g^+}\right)\right]}, \tag{7}$$

where X_{cent} is measured from the metal gate/oxide interface and ΔV_g^- and ΔV_g^+ are the negative and positive gate voltage shifts, respectively. The calculated X_{cent} is 6.7 nm, which is nearly close to the middle of TiAlO charge-trapping layer. Figure 6b shows the schematic diagram of X_{cent} position of the Au/TiAlO/SiO₂/Si structure.

Conclusions

In this paper, the nominated Al₂O₃-TiO₂-Al₂O₃ tri-layer charge-trapping memory structure was fabricated by electron beam evaporation, and the tri-layer dielectric stack changed to Al₂O₃-TiAlO-SiO₂ structure after annealing at 900 °C. The annealing formed memory devices with high-*k* TiAlO charge-trapping layer exhibit significant memory effects and excellent reliability properties. The electronic structures of the tri-layer dielectric



stack ($\text{Al}_2\text{O}_3\text{-TiAlO-SiO}_2$) were investigated by valence band and energy loss spectra measurements of XPS. The deep barrier height for charge confinement in TiAlO layer and good insulating properties of the gate dielectric were believed to be the reasons for the excellent retention and endurance properties of the memory device. The mixing between Al_2O_3 and TiO_2 can increase the defects related to the under-coordinated Ti^{3+} atoms, thereby enhancing the charge-trapping efficiency of the device. The defect level center of the high- k TiAlO is determined to be located at the middle gap of TiAlO film by PL measurement. Our results imply that the high-temperature annealing formed high- k TiAlO composite film is promising for applications in the future nonvolatile memories.

Abbreviations

ALD: Atomic layer deposition; CBO: Conduction band offset; CCS: Constant current stress; CTM: Charge-trapping memory; CV: Capacitance-voltage; P/E: Program/erase; PL: Photoluminescence; VBM: Valence band maximums; X_{cent} : Charge trap centroid; XPS: X-ray photoelectron spectroscopy

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Authors' Contributions

WCX and YZ performed the experimental works and drafted the manuscript. XBL and JML contributed to the design of the experiment, data analysis, and manuscript revision. ZJT, ZJS, MHQ, MZ, SJW, ZZ, JWG, and GFZ helped to analyze the data. All authors read and approved the final manuscript.

Competing Interests

The authors declare that they have no competing interests.

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