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Direct Growth of Al₂O₃ on Black Phosphorus by Plasma-Enhanced Atomic Layer Deposition

B. B. Wu^{1,2}, H. M. Zheng², Y. Q. Ding^{1*}, W. J. Liu^{2*}, H. L. Lu², P. Zhou², L. Chen², Q. Q. Sun², S. J. Ding² and David W. Zhang²

Abstract

Growing high-quality and uniform dielectric on black phosphorus is challenging since it is easy to react with O₂ or H₂O in ambient. In this work, we have directly grown Al₂O₃ on BP using plasma-enhanced atomic layer deposition (PEALD). The surface roughness of BP with covered Al₂O₃ film can reduce significantly, which is due to the removal of oxidized bubble in BP surface by oxygen plasma. It was also found there is an interfacial layer of PO_x in between amorphous Al₂O₃ film and crystallized BP, which is verified by both X-ray photoelectron spectroscopy (XPS) and transmission electron microscopy (TEM) measurements. By increasing temperature, the PO_x can be converted into fully oxidized P₂O₅.

Keywords: Black phosphorus, Plasma-enhanced atomic layer deposition, Oxygen plasma, Al₂O₃

Background

Two-dimensional (2D) semiconductor materials, such as graphene [1, 2], MoS₂ [3, 4], WSe₂ [5], WS₂ [6], MoTe₂ [7, 8], SnSe₂ [9], and black phosphorous (BP) [10–18], have been widely studied for the potential applications in the next generation devices including field emitters [10–12], gas sensors [13, 14], solar cells [17], field-effect transistors [18], optoelectronics [19], and light-emitting diodes [20]. Among these 2D materials, BP is found to be more thermodynamically stable under the ambient conditions [14]. BP is an anisotropic lamellar semiconductor and has a direct band gap of ~0.3–1.5 eV from the bulk to monolayer structure [21–25]. BP transistors also have high carrier mobility up to 1000 cm²/(V·s) at room temperature, on/off current ratio over 10⁴ [18, 26]. Thus, BP may be a promising candidate for electronic and optoelectronic device applications [27]. However, exfoliated BP films will degrade in the air ambient owing to the possible reactions between BP and the adsorbed

water and oxygen, thus leading to a significant reduction of BP carrier mobility and on/off current ratio [27, 28]. Therefore, efficient and reliable isolation/passivation layers are necessary for BP to preserve its inherent structure and property.

So far, many efforts on passivation for BP have been made, such as an encapsulation layer for BP with boron nitride (*h*-BN) [29–33], creating saturated P₂O₅ on BP surface [34–36], atomic layer deposited dielectric capping [27, 37–40]. Nevertheless, *h*-BN passivation requires complicated environmental conditions and has extremely low yield [29–33]. The P₂O₅ which was created on BP surface provides only the short-time protection since the oxygen and moisture in the air can erode it slowly [34–36]. It is also quite tough for atomic layer deposition (ALD) to form a high-quality and uniform top dielectric film on BP because of no dangling bonds. Therefore, it is important to prevent BP-based devices from degradation in the air ambient by covering a protective oxide dielectric. Moreover, uniform and reliable dielectrics are also essentially needed for the top-gate BP devices.

In this work, uniform Al₂O₃ was directly grown on BP flakes using plasma-enhanced atomic layer deposition (PEALD) with the help of O₂ plasma as an oxygen

* Correspondence: yding@jiangnan.edu.cn; wjliu@fudan.edu.cn

¹The Key Laboratory of Food Colloids and Biotechnology, Ministry of Education, School of Chemical and Material Engineering, Jiangnan University, Wuxi 214122, Jiangsu, China

²State Key Lab ASIC & Syst., Department Microelectronics, Fudan University, Shanghai 201203, China

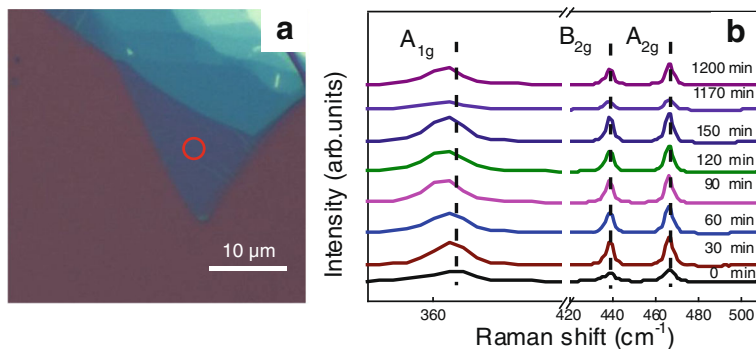


Fig. 1 **a** Optical microscope image of the BP sample acquired by Raman measurements. **b** Raman spectra of pristine BP for different exposure time. All Raman measurements were done in the air ambient with the same laser excitation

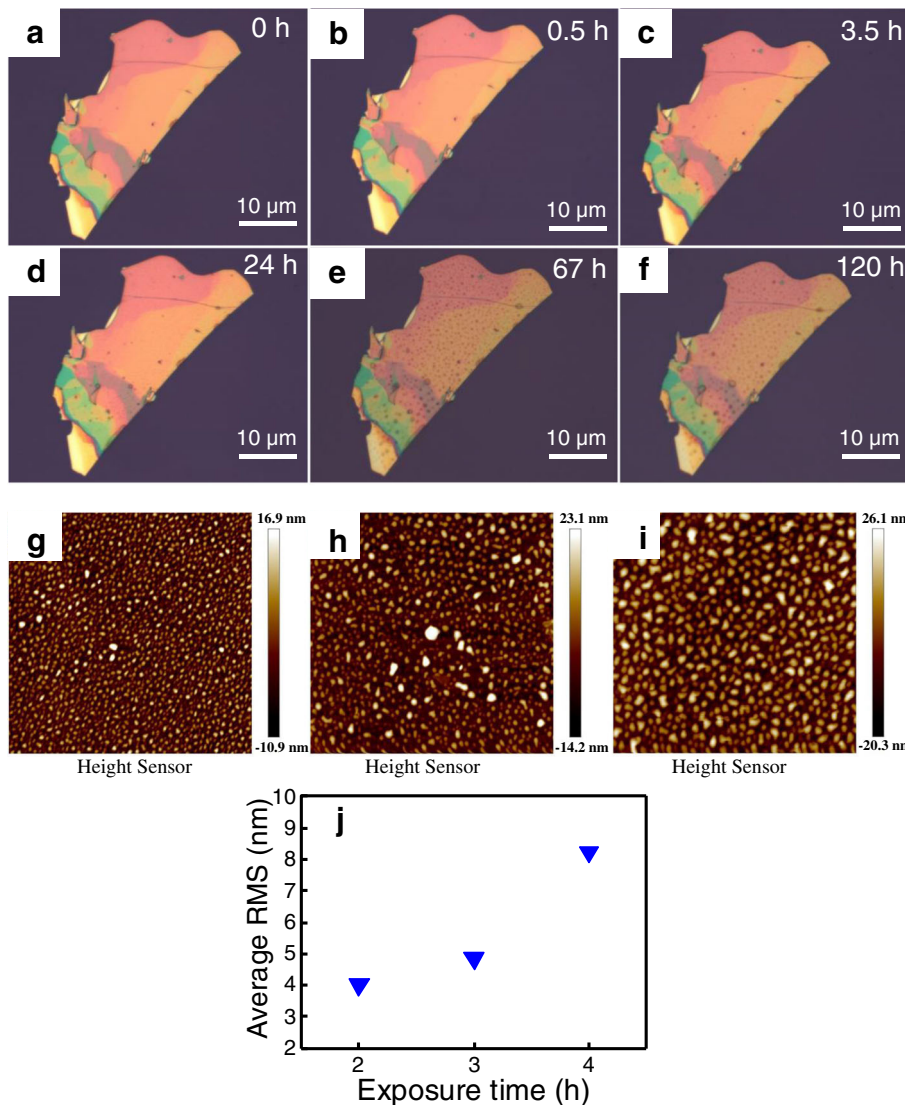


Fig. 2 **a-f** Optical images of accelerated BP degradation on SiO₂/Si exposed to air for different time. **g-i** AFM images of BP flake exposed to the air ambient for 2, 3, and 4 h. All three BP samples for AFM measurements were taken from the same batch. The average thickness of BP in **g-i** was 130 nm. **j** The average RMS roughness of BP samples versus the exposure time, the same samples as shown in **g-i**

precursor, instead of H_2O , to react with trimethylaluminum (TMA) [41]. The composition and properties of the interfacial layer between Al_2O_3 and BP have been examined by physical characterizations, and the mechanisms behind are analyzed.

Methods

Few-layer BP (purity: 99.998%, Smart Elements) was transferred onto a Si substrate with thermally grown 285 nm SiO_2 using a micromechanical method with polydimethylsiloxane (PDMS) [1, 28, 42]. Prior to transfer, the SiO_2 surface was ultrasonically cleaned in turn by acetone and isopropyl alcohol (IPA) and piranha solutions for 10 min each, followed by 100% O_2 annealing at 500 °C for 3 min using rapid thermal annealing (Annealsys As-One). The optical images of BP were acquired by an optical microscope (BA310Met, Motic) equipped with a camera. Raman spectroscopy measurements were performed using LabRam-1B (the Raman spectral resolution was 1.1 cm^{-1}) with an excitation wavelength of 532 nm at room temperature in the air ambient. The laser power was maintained at around 0.5 mW to prevent any heating-induced damage during the measurement. Al_2O_3 films on BP were deposited using PEALD with O_2 plasma and TMA precursors at different temperatures. The freshly exfoliated BP samples were transfer to Picosun 200R ALD chamber (the vacuum pressure was 12 hPa). PEALD of Al_2O_3 was carried out with successive cycles of O_2 plasma and TMA precursors, with an Ar carrier gas (99.9997%, Airgas) at a flow rate of 300 sccm, 15 s pulse + 10 s Ar purge time for O_2 plasma (The O_2 plasma RF Power was 2000 W), 0.1 s pulse + 5 s Ar purge time for TMA (the precursor temperature was 18 °C) at a substrate temperature of 200 °C. The surface and interfacial properties of Al_2O_3 on BP were physically characterized using atomic force microscopy (AFM, Dimension Edge, Bruker), XPS (AXIS ULDDLTRA, Shimadzu), and TEM (Tecnai $\text{G}^2\text{ F20 S-TWIN}$, FEI) measurements at room temperature.

Results and Discussion

Figure 1a shows the optical image of transferred BP sample prepared by mechanical exfoliation from its bulky crystalline. The Raman spectra of thin-layer BP, as denoted by a red circle in Fig. 1a, were examined as a function of exposure time in the air ambient at room temperature, as shown in Fig. 1b. It is noted that all Raman spectra measured in Fig. 1b are calibrated using a Si peak of 520 cm^{-1} . It can be clearly seen one out-of-plane modes (A_{1g}) and two in-plane modes (A_{2g} and B_{2g}) in thin-layer BP [43]. Both A_{2g} and B_{2g} peak positions keep almost unchanged. While for A_{1g} mode, it has redshifted as the exposure time goes up to 30 min and then seems to be stable up to 20 h. This may be

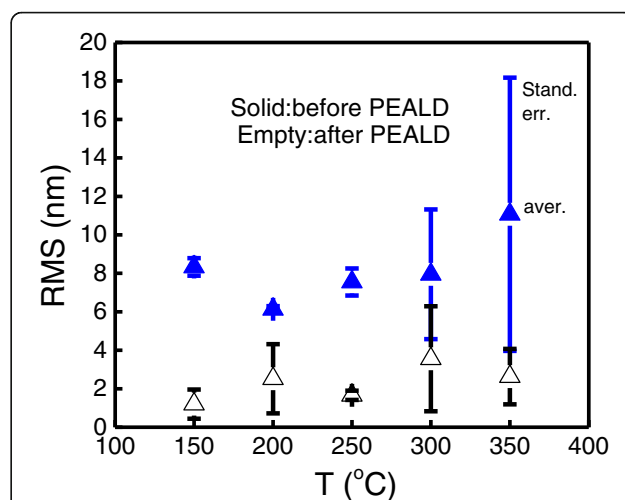


Fig. 3 RMS roughness of BP samples before and after PEALD deposition at various temperatures. For each deposition temperature, more than five samples were measured for an accurate assessment

attributed to the oxidation of surficial BP in the initial stage and a relatively saturation of oxidation up to 20 h. This is evidenced by the time evolution of BP surface morphology examined by optical microscopy, as shown in Fig. 2a–f. It was markedly observed that BP flake exposed to the air ambient degrades as the exposure time extended and then exhibited a fare rough BP surface with bubbles, as presented in Fig. 2d–f. Figure 2g–i shows AFM images of exfoliated BP flake exposed to the air ambient for 2, 3, and 4 h, respectively. All three BP samples for AFM measurements were taken from the same batch and their RMS roughness is summarized in Fig. 2j. The RMS roughness of BP surface increases as the exposure time increases, indicating the formation of oxidative phosphorus species.

To evaluate the surface quality of Al_2O_3 film on thin-layer BP, its roughness of RMS was compared quantitatively at different deposition temperatures, as shown in Fig. 3. The corresponding data were summarized in Table 1. Note that more than five samples were measured for each temperature. It was observed that prior to PEALD, the average RMS of thin BP surface is larger than 6 nm, large scattering is due to sample-to-sample

Table 1 The RMS roughness of BP samples before and after PEALD at different temperatures

Temperature (°C)	The average roughness (before/after) (nm)	Standard deviation (before/after) (nm)
150	8.33/1.20	0.46/0.76
200	6.13/2.52	0.16/1.80
250	7.55/1.66	0.71/0.24
300	7.95/3.56	3.34/2.73
350	11.05/2.63	7.10/1.44

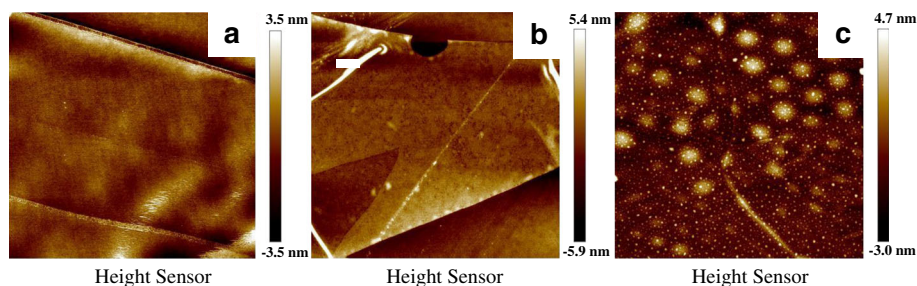


Fig. 4 AFM images of $\text{Al}_2\text{O}_3/\text{BP}$ samples with different deposition recipes at 200°C . **a** 100 cycles Al_2O_3 on BP grown by PEALD with the pretreatment of 20 cycles O_2 plasma. **b** 100 cycles Al_2O_3 on BP grown by PEALD without any pretreatment. **c** 100 cycles Al_2O_3 on BP grown by ALD with H_2O as an oxygen precursor

variations; however, the RMS reduces to ~ 3 nm after 100 cycles Al_2O_3 deposition. This infers that O_2 plasma as an oxygen precursor can effectively etch the oxidized bubbles in the top-layer BP thin film, thus leading to a significant reduction of RMS, while H_2O as an O source may not have this benefit (discuss later).

To understand the impact of the pretreatment of O_2 plasma and different oxygen precursors on Al_2O_3 growth in freshly exfoliated BP samples, Al_2O_3 deposition on BP was realized by three approaches: (1) 20 cycles O_2 plasma pretreatment + 100 cycles TMA/ O_2 plasma, (2) 100 cycles TMA/ O_2 plasma, and (3) 100 cycles TMA/ H_2O , as shown in Fig. 4a, b, and c, respectively. Figure 4a, b depicts AFM images of the 100 cycles Al_2O_3 grown on BP samples by PEALD with and without an oxygen plasma pretreatment, respectively. Using PEALD for Al_2O_3 growth in BP flakes, it has demonstrated a highly uniform surface morphology of $\text{Al}_2\text{O}_3/\text{BP}$. The average RMS roughness of $\text{Al}_2\text{O}_3/\text{BP}$ samples prepared by PEALD is only 0.4 nm regardless of an oxygen plasma pretreatment, as shown in Fig. 4a, b. For freshly exfoliated BP samples, PEALD (with and w/o pretreatment) can achieve a good uniformity and coverage of Al_2O_3 films. While for BP samples exposed to the air ambient for certain time, 4 h for example, PEALD with O_2 plasma pretreatment is much preferred. O_2 plasma pretreatment can create enough nucleation sites for ALD growth. On the other hand, it also has an “etching” effect for thinning BP samples. O_2 plasma may penetrate the PO_x layer and oxidize the underlying BP, then increase the thickness of PO_x layer [35]. On the contrary, Al_2O_3 films on freshly exfoliated BP grown by ALD with H_2O as an oxygen precursor nucleate to an isolated “island” and exhibit a remarkable nonuniform surface profile, resulting in a large RMS roughness of 0.8 nm, as shown in Fig. 4c. It is attributed to the insufficient dangling bonds or nucleation sites in BP surface for ALD growth with H_2O as an oxygen precursor [37]. It is worthwhile to mention that BP flake was covered uniformly by Al_2O_3 film and can prevent O_2 or H_2O in

air ambient further reacting with BP, thus protected BP from degradation. Otherwise, the uncovered portions of BP surface may react with H_2O and O_2 to produce many bubbles, as shown in Fig. 4c.

Next, chemical analysis of the interfacial characteristics near BP film was examined by XPS characterizations. Figure 5 shows photoelectron spectroscopy measurements of the P 2p core level at different deposition temperatures. Middle and top panels present

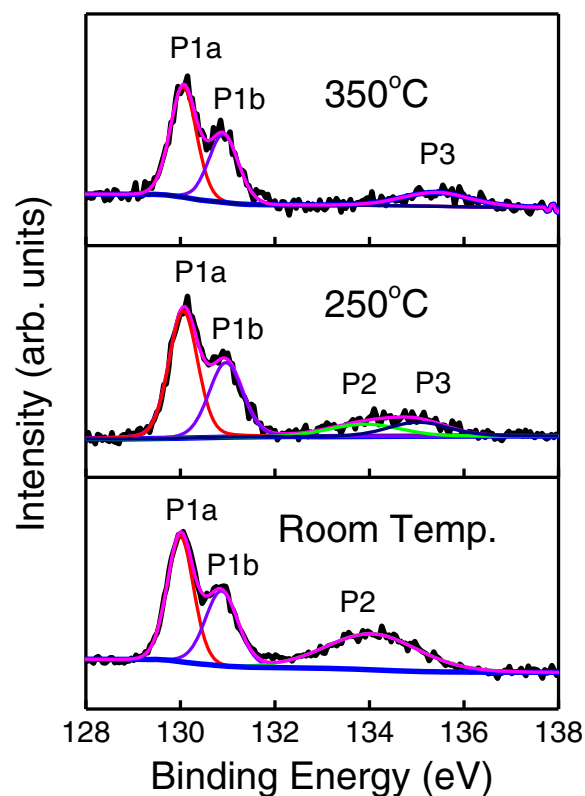


Fig. 5 P 2p XPS spectra of the interface between Al_2O_3 and BP of samples at different deposition temperatures. P1a($\text{P}2p_{3/2}$) and P1b($\text{P}2p_{1/2}$) represent phosphorus bonded to phosphorus; P2 and P3 represent the different oxidative species of PO_x and P_2O_5 , respectively

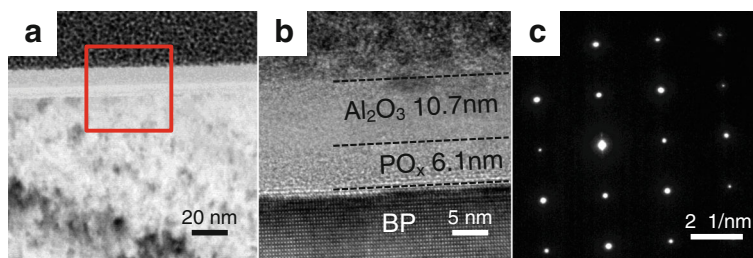


Fig. 6 **a** Low-magnification TEM image of $\text{Al}_2\text{O}_3/\text{BP}$ sample fabricated at 200°C . **b** High-resolution TEM image of $\text{Al}_2\text{O}_3/\text{BP}$ sample, same scanned region marked by a red square in **a**. The thickness of Al_2O_3 and PO_x layer is 10.7 and 6.1 nm, respectively. **c** Selected area electron diffraction (SAED) pattern for the BP crystalline in **b**

the P 2p core level after growth of 100 cycles Al_2O_3 by PEALD at 250 and 350°C , respectively. The peaks which labeled as P1a and P1b correspond to P-P bonds. The phosphorus core level contains only the characteristic doublet representing the P $2p_{3/2}$ (P1a) and P $2p_{1/2}$ (P1b) orbitals with peak positions of 130.06 eV and 130.92 eV, respectively, consistent with the report [34]. Two peaks of P2 and P3 were observed at 134.07 and 135.03 eV, corresponding to the PO_x and P_2O_5 , respectively, as shown in the middle panel. It can be seen that the PO_x peak locates at 134.07 eV, smaller than that of reported P_2O_3 (134.2 eV) [44], which may be caused by less O concentration in the interfacial PO_x layer. The P3 represents the most dominant oxide component and appears at 135.03 eV, in good agreement with the reported P_2O_5 binding energies which are between 135.0 eV [45] and 135.15 eV [37]. When the temperature goes up to 350°C , interestingly, P2 peak disappears. This is due to the conversion from PO_x to P_2O_5 , with the help of reactivity of O_2 plasma at high temperatures. However, there is no P3 peak for natively oxidized BP at room temperature and its peaks of P1a; P1b and P2 locate at 130.06 eV (P $2p_{3/2}$), 130.87 eV (P $2p_{1/2}$), and 134.05 eV, respectively. The absence of P3 peak is due to low temperature or insufficient exposure time for the formation of fully oxidative top layer, which may prevent PO_x from converting into P_2O_5 film.

Finally, the interface properties of $\text{Al}_2\text{O}_3/\text{BP}$ samples were also characterized by TEM measurements. It can be clearly seen for Fig. 6a that the interfacial PO_x layer between Al_2O_3 and BP was formed during PEALD process with the 20 cycles O_2 plasma pretreatment. Figure 6b shows high-resolution TEM (HRTEM) image of the $\text{Al}_2\text{O}_3/\text{BP}$ sample after the deposition of 100 cycles Al_2O_3 , same scanned region marked by a red square in Fig. 6a. The thickness of PO_x and Al_2O_3 is 6.1 and 10.7 nm, respectively. It is worth noting that Al_2O_3 and PO_x film is amorphous, while our BP sample is single crystalline which is verified by results of selected area electron diffraction (SAED) pattern, as seen from Fig. 6c. This interfacial layer PO_x was evidenced by TEM results,

indicative of O_2 plasma penetrating into PO_x layer and reacting with underlying BP.

Conclusions

In summary, we have demonstrated the direct growth of Al_2O_3 film on BP by using PEALD. The A_{1g} peak of freshly exfoliated BP sample shifts downwards owing to the formation of PO_x in the BP surface. The uniform Al_2O_3 film on BP can be achieved by PEALD with O_2 plasma and TMA precursors, which may be attributed to the etching and reactivity of O_2 plasma with BP at high temperatures. The interfacial layer of PO_x between Al_2O_3 and BP was converted into P_2O_5 as the deposition temperature increases to 350°C , revealed by XPS characterizations. These findings provide insightful information on passivation and top-gate dielectric integration for future applications in BP devices.

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Authors' Contributions

BBW and HMZ carried out the BP fabrication and Al_2O_3 growth and measurements. YQD and WJL supervised the work and drafted the manuscript. HLL, PZ, LC, QQS, SJD, and DWZ helped to analyze the experimental results. All authors read and approved the final manuscript.

Competing Interests

The authors declare that they have no competing interests.

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