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Phase-change properties of GeSbTe thin films deposited by plasma-enchanced atomic layer depositon

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Abstract

Phase-change access memory (PCM) appears to be the strongest candidate for next-generation high-density nonvolatile memory. The fabrication of ultrahigh-density PCM depends heavily on the thin-film growth technique for the phase-changing chalcogenide material. In this study, $Ge_2Sb_2Te_5$ (GST) and $GeSb_8Te$ thin films were deposited by plasma-enhanced atomic layer deposition (ALD) method using $Ge [(CH_3)_2 N]_4$, $Sb [(CH_3)_2 N]_3$, Te $(C_4H_9)_2$ as precursors and plasma-activated H₂ gas as reducing agent of the metallorganic precursors. Compared with GST-based device, $GeSb_8Te$ -based device exhibits a faster switching speed and reduced reset voltage, which is attributed to the growth-dominated crystallization mechanism of the Sb-rich $GeSb_8Te$ films. These results show that ALD is an attractive method for preparation of phase-change materials.

Keywords: Phase-change memory; Atomic layer deposition; Microstructure; Electric properties

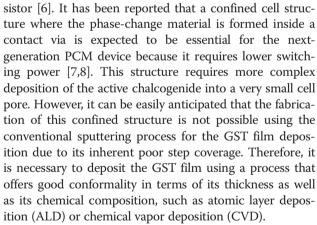
Background

Phase-change memory (PCM) has been regarded as one of the most promising nonvolatile memories for the next generation because of the advantages of high speed, low power, good endurance, high scalability, and fabrication compatibility with complementary metal-oxide semiconductor (CMOS) process [1-4]. PCM uses the reversible phase change between the crystalline and amorphous states of chalcogenide materials brought about by Joule heating. Ternary Ge₂Sb₂Te₅ (GST) compounds are widely regarded as the most commercially viable and practical phase-change family of materials for this application. These materials are currently being used commercially, and processes which deposit GST films by RF sputtering are being implemented into production lines [5].

In PCM cells, the high level of reset current, which is required for switching the GST material from the crystalline to the amorphous state in a planar cell structure, has been the major obstacle to the further scaling of PCM because

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of the limited on-current drive capability of the cell tran-

Recently, a number of studies have been carried out to investigate the deposition of phase-change materials by ALD and CVD techniques [9-15]. It has been reported that the precursor and substrates have an important influence on the microstructure and composition of Ge-Sb-Te films. Mikko Ritala *et al.* reported Sb₂Te₃, GeTe, and GST films were deposited by ALD at remarkably low temperature of 90°C using (Et₃Si)₂Te, SbCl₃, and GeCl₂ · C₄H₈O₂ as precursors [10]. Byung Joon Choi *et al.* reported the different



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nucleation and growth behaviors of the GST films deposited by the combined plasma-enhanced CVD/ALD on various types of substrates. The nucleation of the GST films on the SiO₂, Si₃N₄, and ZrO₂ substrates was seriously retarded compared to those on the TiN and TiO₂ substrates [11]. Adulfas Abrutis et al. reported the deposition of smooth GST films by using a hot-wire CVD technique [12]. Most of these works focus on the microstructure properties and the deposition process. However, the electric properties and switching properties of ALD/CVD-deposited phasechange materials (especially Sb-rich GeSbTe films) have rarely been reported in literature. In this study, ALD of Sb₂Te₃, GST, and GeSb₈Te thin films were attempted with Ge [(CH₃)₂ N]₄, Sb [(CH₃)₂ N]₃, Te(C₄H₉)₂ as Ge, Sb, and Te precursors, respectively, and plasma-activated H₂ gas as reducing agent of the metallorganic precursors. The microstructural and electric properties of these materials have been studied.

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Methods

The Sb₂Te₃ and Ge-Sb-Te films were deposited on Si₃N₄/Si substrates using a plasma-enhanced ALD reactor (Beneq TFS 500 ALD system) at wafer temperatures ranging from 190 to 250°C. The Si₃N₄ films were prepared by inductively coupled plasma CVD on Si substrate at temperature of 130°C. The SiH₄ and NH₃ were used as reactive gases. Ge $[(CH_3)_2 N]_4$, Sb $[(CH_3)_2 N]_3$, and $Te(C_4H_9)_2$ were used as the Ge, Sb, and Te precursors, respectively. Each precursor deposition cycle for the Ge, Sb, and Te process was composed of four consecutive pulses: (i) a pulse of precursor vapor, (ii) a purge pulse, (iii) a pulse for exposure to H_2 plasma, and (iv) another purge pulse. The sequence of precursor pulses was Sb-Te-Ge. To control the cation composition ratio in Ge-Sb-Te films, the pulse ratio of precursor cycles was changed. During the H₂ plasma pulse period, a radio frequency (rf) plasma was applied (rf power of 150 W, rf frequency of 13.56 MHz) to the reaction chamber. The

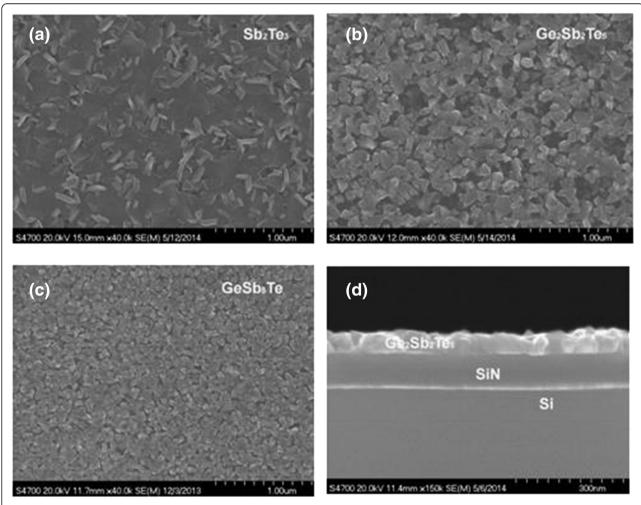
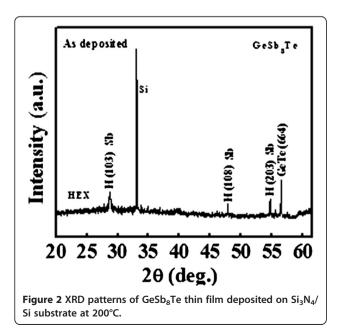


Figure 1 SEM surface micrographs. SEM surface micrographs of (a) a 70-nm Sb2Te3 film grown on Si3N4, (b) a 60-nm Ge2Sb2Te5 film grown on Si3N4, (c) a 70-nm GeSb8Te film grown on Si3N4, (d) cross-section structure of the 60-nm Ge2Sb2Te5 film grown on Si3N4.

growth rates of GST and Sb₂Te₃ films are about 0.2 and 0.1 nm/cycle, respectively. The uniformity of Sb₂Te₃ films was very good, but the uniformity of GST was very sensitive to the deposition process. Similar results have been found by other researchers [13]. The stoichiometry of the deposited films was confirmed by electron dispersive spectroscopy (EDS). The thickness and microstructure of the films were determined by field-emission scanning electron microscopy (FESEM). T-shaped PCM cells are fabricated using 0.18 µm CMOS technology to verify the electrical switching behaviors of the alloys. The bottom W electrode with diameter of 260 nm is covered by phase-change films of approximately 100 nm thickness, while TiN (20 nm) and Al (300 nm) are deposited sequentially as top electrodes. For comparison, physical vapor deposition (PVD) GST film was also fabricated with the same structure. The current-voltage (I-V) and resistance-voltage (R-V) characteristics of the PCM cells are monitored employing an arbitrary waveform generator (Tektronix AWG5002B) and a Keithley-2400 meter.

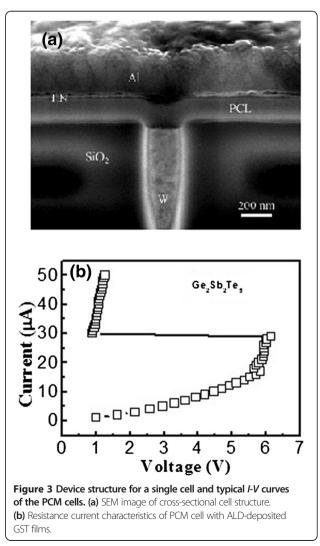
Results and discussion

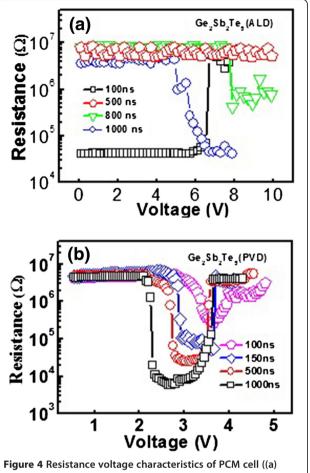
Figure 1a to d shows the SEM surface micrographs of a 70-nm Sb_2Te_3 film, a 60-nm GST film, a 70-nm $GeSb_8Te$ film, and the cross-section structure of the 60-nm GST film grown on Si_3N_4 surfaces, respectively. It can be seen in the figure that pure Sb_2Te_3 film shows much larger crystal grain size than those of the GST and $GeSb_8Te$ films. The GST film grown on a Si_3N_4 surface appeared to have many surface voids. The grain size of GST film is about 100 nm. The $GeSb_8Te$ film grown on a Si_3N_4 surface showed a similar surface morphology with a slightly smaller average grain size (approximately 70 nm) and without surface voids. The root-mean-square (rms)



surface roughness values of Sb₂Te₃, GST, and GeSb₈Te are 43.5, 30.4, and 21.8 nm, respectively, indicating rough surface morphology of these films. The cross-sectional image in Figure 1d confirms the rough surface morphology of the GST film. XRD patterns of GeSb₈Te thin film deposited on Si₃N₄/Si substrate at substrate temperature of 200°C are shown in Figure 2. Hexagonal Sb and additional GeTe phases were seen in GeSb₈Te film, which indicates that it crystallized during deposition. This is due to the low crystallization temperature (<200°C) of the GeSb₈Te film.

For further investigation of device performances, GST and $GeSb_8Te$ films were selected and inserted in PCM cells. The device structure for a single cell is shown in Figure 3a. Figure 3b shows the typical *I-V* curves of the PCM cells based on ALD-deposited GST films. Before each *I-V* test, the cell has been re-amorphized using electrical pulse. As shown in Figure 3b, large snapback of voltage and negative-resistance behavior were observed in the *I-V* curve, which



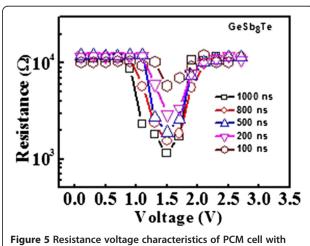


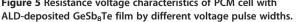
ALD-deposited GST, (b) PVD-deposited GST films by different voltage pulse widths).

indicates that the phase transition has occurred from amorphous state (high resistance) to crystalline state (low resistance). The threshold voltage for the cell based on ALDdeposited GST is about 6.1 V, which is much higher than that (3.5 V) of the device based on PVD-deposited GST with the identical cell architecture [16].

The phase transition of PCM cell can be characterized from the relation between the cell resistance and the corresponding amplitude of voltage pulse or current pulse (so-called R-V or R-I curve). In order to test the electrical phase-change ability of the ALD-deposited GST films, the SET and RESET operations of the PCM cells based on the ALD-deposited GST materials are realized by a stimulated voltage pulse with different pulse widths, as presented in Figure 4a. As indicated in Figure 4a, the resistance dramatically increases by two orders of magnitude at the reset voltage of around 6.7 V. The R-V curves of the PCM cells using PVD-deposited GST materials are also shown in Figure 4b. For the device based on PVD-deposited GST films with the identical cell architecture, the reset voltage is around 3.6 V, which is lower than that of the device based on ALD-deposited GST. It also can be seen from Figure 4b that a 200-ns-wide voltage pulse is able to set the cells based on PVD-deposited GST materials with the sensing margin ($R_{\text{RESET}}/R_{\text{SET}}$) of more than two orders of magnitude. For the device based on ALD-deposited GST, it is noted that a 500-ns-wide pulse fails to set the cell and a pulse width of 800 ns is insufficient for a complete set programming, suggesting that ALD process indeed leads to a slower crystallization process thus longer write time for the set operation. The measured set speed of the ALD-deposited GST films, which is supposedly due to the impurities. Especially N, C impurities in ALD-deposited GST films are known to increase the crystallization temperature and reduce the crystallization speed of the GST films [17,18].

The R-V curves of the PCM cells based on ALDdeposited GeSb₈Te material with various pulse widths are plotted in Figure 5. Reversible phase-change process has been observed. As revealed, once the programing voltage increases beyond the threshold voltage, the cell resistance starts to drop due to the crystallization of GST alloy and then reaches a minimum, which is corresponding to the set resistance. When the voltage is further increased, the resistance again rises and then returns to the reset state. The minimum set and reset voltages decrease with the increasing pulse width due to the equivalent energy required to crystallize and melt the programming region, respectively. The set resistance of the cells decreases with the pulse width, obtaining a lower resistance of approximately 1.1 k Ω for the 1,000ns-wide pulse, which is attributed to the larger grains and/or a more complete crystallization state. As indicated in the figure, the reset voltage of approximately 2.1 V of the cell is clearly lower than that of the ALD GST-based cells (approximately 6.7 V) and PVD GSTbased cells (approximately 3.7 V) with the identical cell





architecture. The resistance window could be obtained by an electric pulse as short as 100 ns. Compared with ALD GST requiring 1,000 ns in this study, $GeSb_8Te$ exhibits a faster switching speed of more than 10 times. The growth-dominated crystallization mechanism of the Sb-rich GeSb₈Te film could account for its ultra-fast speed, while GST requires much more time for nucleation due to its nucleation-dominated crystallization mechanism.

Conclusions

In this study, GST and GeSb₈Te thin films were deposited by plasma-enhanced ALD method using Ge $[(CH_3)_2 N]_4$, Sb $[(CH_3)_2 N]_3$, Te(C₄H₉)₂ as precursors and plasma-activated H₂ gas as reducing agent of the metallorganic precursors. The measured set speed of the ALD-deposited GST films is slower than that of PVD-deposited GST films, which is supposedly due to the N, C impurities. Compared with GST-based device, GeSb₈Te-based device exhibits a faster switching speed and reduced reset voltage, which is attributed to the growth-dominated crystallization mechanism of the Sb-rich GeSb₈Te films. These results show that ALD is an attractive method for preparation of phase-change materials.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

SS and ZS conceived the study. DY and ZZ participated in the sample preparation. LG and LL carried out the XRD and SEM characterizations. LS and LW participated in the fabrication of the device. YC carried out the electric properties measurement. BL and SF read the manuscript and contributed to its improvement. All the authors discussed the results and approved the final version of the manuscript. All the authors read and approved the final manuscript.

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