

REVIEW

Iron deficiency in critically ill patients: highlighting the role of hepcidin

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Introduction

Iron is a paradoxical element, essential for living organisms but also potentially toxic. Indeed, iron has the ability to readily accept and donate electrons, interconverting from soluble ferrous form (Fe^{2+}) to the insoluble ferric form (Fe^{3+}). This capacity allows iron to play a major role in oxygen transport (as the central part of hemoglobin) but also in electron transfer, nitrogen fixation or DNA synthesis, all essential reactions for living organisms. Indeed, iron deficiency is the main cause of anemia [1] as well as a cause of fatigue [2,3] and decreased effort capacity [4,5]. However, despite a high frequency of anemia among critically ill patients, with 60 to 66% being anemic at intensive care unit (ICU) admission [6,7], to date little is known about iron deficiency and iron metabolism in critically ill patients [8]. The interaction between inflammation and iron metabolism interferes with the usual iron metabolism variables and renders this metabolism difficult to investigate [9,10].

The recent discovery of hepcidin (the master regulator of iron metabolism) has shed new light on the regulation of iron homeostasis and has helped our understanding of complex clinical situations, such as those observed in critically ill patients, where several regulatory circuits interfere with iron metabolism [11]. The purpose of this article is to review iron metabolism and anemia in critically ill patients as well as the role of hepcidin, and to discuss the indications for iron supplementation in these patients.

Iron metabolism overview and the role of hepcidin

Although iron is essential for life, it may also be toxic because of its capacity to react with oxygen and to

promote the production of free radicals. This duality is found in human pathology: Iron deficiency (because of poor iron intake, abnormal blood losses etc...) presents with anemia and fatigue; whereas iron overload (mainly in hereditary hemochromatosis and following repeated blood transfusions) induces multiple organ dysfunctions (including liver fibrosis, cirrhosis, cardiomyopathy, diabetes...). This explains why iron homeostasis must be finely tuned to avoid both deficiency and excess.

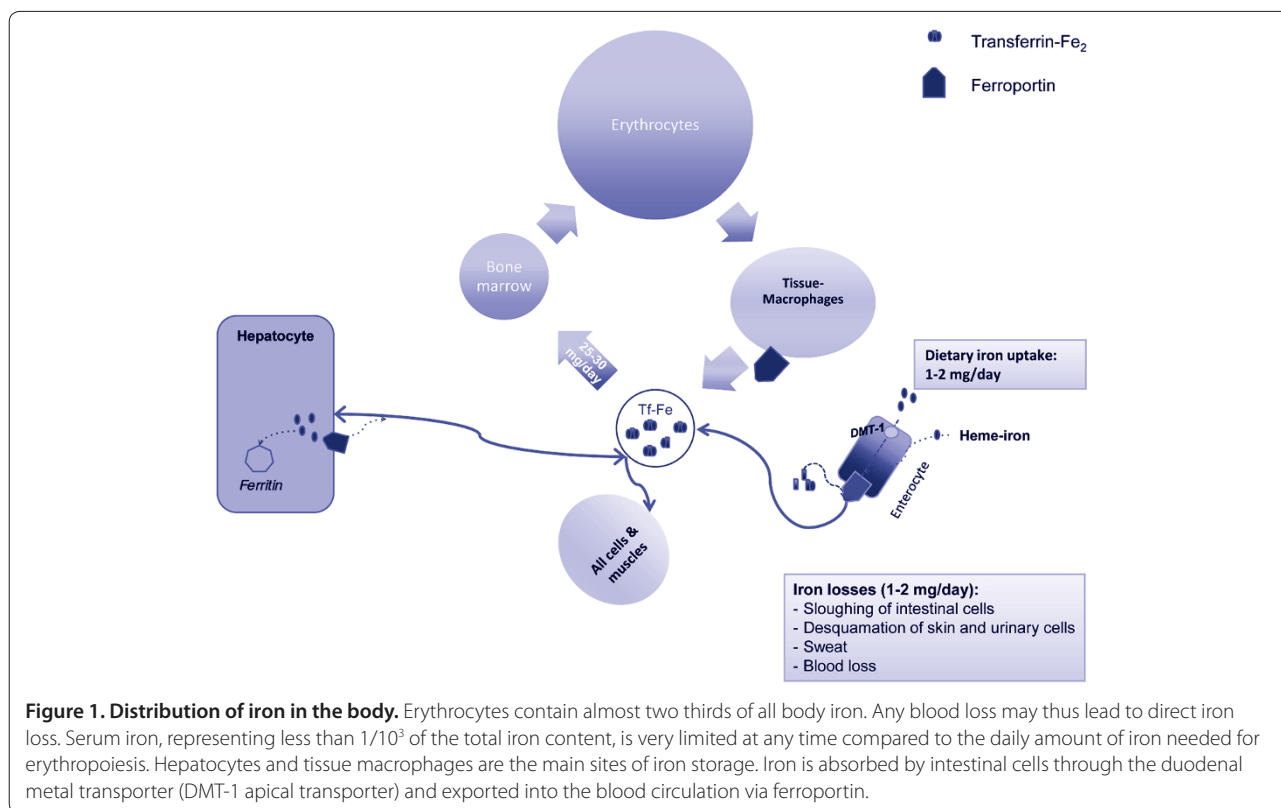
Iron turnover in the organism occurs almost in a closed circuit (Fig. 1). Indeed, global iron turnover through losses (because of bleeding or cell desquamation) and dietary uptake (by duodenal cells) is only 1 to 2 mg per day, compared to approximately 3 to 4 g of iron contained in the organism. In fact, most of the iron available for erythropoiesis comes from the catabolism of senescent red blood cells (RBCs) by macrophages in the reticulo-endothelial system (called erythrophagocytosis). As shown in Figure 1, more than two-thirds of the body's iron content is incorporated into hemoglobin, either in bone marrow erythroid progenitors or in circulating RBCs. Once aged, these RBCs are internalized and hemoglobin is degraded in tissue macrophages. Iron is then transferred to the macrophage cytosol and either released into the blood flow or stored in ferritin molecules. In the plasma, transferrin binds newly released iron to allow its mobilization from storage sites (mainly the spleen and to a lesser extent the liver) to utilization sites (mainly the bone marrow). Erythropoiesis requires about 25 to 30 mg of iron daily. It has to be stressed that the amount of iron present in the plasma at any time is small (about 3 mg) compared to the daily amount of iron needed for erythropoiesis. Iron metabolism is therefore finely tuned, with hepcidin being central to its regulation [12].

Hepcidin is a small 25 amino acid peptide mainly produced by the liver. It is produced as an 84 amino acid pre-pro-peptide. Pro-hepcidin has been shown to be

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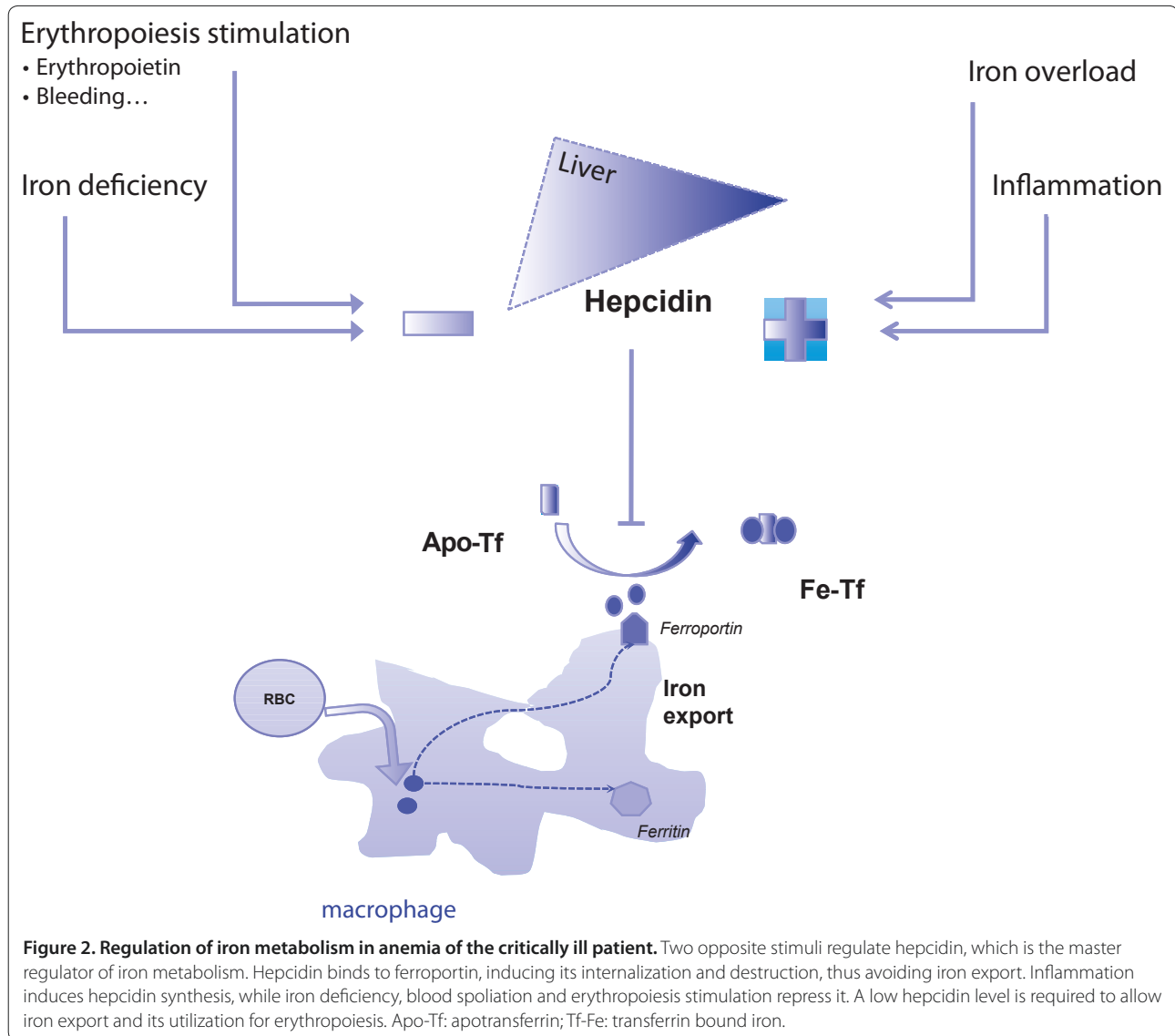
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biologically inactive. Hepcidin acts by binding to ferroportin, which is the sole known iron exporter [13]. The binding of hepcidin to ferroportin induces its internalization and degradation in the cytosol, which prevents the release of intracellular iron [13]. Ferroportin is mainly expressed in macrophages and duodenal cells, allowing, respectively, iron recycling (after erythrophagocytosis) and iron absorption from the digestive lumen (after internalization of iron through natural resistance-associated macrophage protein [nRAMP]/duodenal metal transporter [DMT1]). Induction of hepcidin synthesis may thus lead to iron-restricted erythropoiesis (by inhibiting the release of iron from macrophages to the bone marrow) and to dietary iron deficiency (by inhibiting the uptake of iron from the digestive duodenal cells). Hepcidin acts as a 'hyposideremic' hormone, aimed at inhibiting iron absorption and reducing the level of iron in the blood.

Because hepcidin plays this central role in iron metabolism regulation, its synthesis is finely regulated (Fig. 2) [11,12]. Hepcidin synthesis is induced by iron overload and inflammation, whereas iron deficiency, hypoxia and erythroid expansion repress its synthesis. The molecular mechanisms implicated in these complex regulations are not fully understood (see [12] for review), but the induction of hepcidin synthesis by inflammation

has been shown to be interleukin (IL)-6 dependent [14]. This interaction between hepcidin and inflammation makes hepcidin the principal agent responsible for the iron-restricted erythropoiesis observed during chronic diseases, ultimately leading to the 'anemia of chronic disease' (or anemia of inflammation) [15,16]. On the other hand, hepcidin synthesis is repressed by both iron deficiency and stimulation of erythropoiesis [11,12]. Although the precise mechanisms involved in the repression of hepcidin are not fully understood, it appears that matriptase 2, a membrane-bound serine protease expressed in hepatocytes, seems to play a key role in repressing hepcidin synthesis in iron deficiency conditions [17]. Repression of hepcidin by erythropoiesis stimulation is even less well understood, but seems to involve bone marrow erythropoietic activity rather than erythropoietin itself [18,19]. Hypoxia-inducible factor (HIF) or CCAAT enhancer binding protein- α pathways have also been proposed [12]. In human pathology, little is known. Growth differentiation factor 15, a member of the transforming growth factor (TGF)- β family produced by late erythroblasts, has been found in high levels in patients with beta-thalassemia syndromes and has been shown to repress hepcidin synthesis [20]. These two opposite stimuli are found in the anemia of critically ill patients, as discussed below.



Implication of iron metabolism in the anemia of the critically ill: hepcidin as a diagnostic tool?

Anemia is not only very frequent among critically ill patients, it is also associated with increased transfusion rates and worse outcomes (increased length of stay, increased mortality) [6,7]. However, recent recommendations have led to a decrease in transfusion triggers [21]. Nowadays, anemia is present at ICU discharge in at least 75% of all patients when considering their last measured hemoglobin levels [22]. Furthermore, anemia may also be prolonged after discharge, with a median time to recovery of 11 weeks and more than half of the patients still anemic 6 months after ICU discharge [23]. There is, therefore, need for a better understanding of the mechanisms of anemia in the critically ill and an evaluation of therapeutic options.

The two main contributing factors for anemia in the critically ill are inflammation and iron deficiency, which have opposite effects on iron metabolism (see above). Until recently, inflammation, rather than iron deficiency, was considered to play the major role. Indeed, the iron profile of critically ill patients constantly shows hallmarks of anemia of inflammation. However, this topic has not been considered a matter of great interest in the past, with few studies undertaken [9]. Inflammation is frequent in critical illness, whatever the underlying pathology. The anemia of critically ill patients is indeed similar to the anemia of inflammation, with blunted erythropoietic response and activation of RBC destruction by macrophages [15,24]. Low serum iron and high ferritin levels constitute the typical iron profile of critically ill patients and are indicative of an inflammatory iron profile [25,26].

Because ferritin synthesis is induced by inflammation (through IL-1) independently of the level of iron stores, elevated ferritin levels are no longer indicative of iron stores in the context of inflammation [10]. Thus, despite an iron profile that mimics iron overload (with high ferritin levels), iron deficiency may exist in these critically ill patients.

Indeed, daily blood losses are far from negligible, either through repeated blood sampling [6,27], surgical site bleeding, other invasive procedures (drainage, catheter placement, renal replacement therapy...) or occult bleeding [26]. The median blood loss for anemic critically ill patients has been estimated to be as high as 128 ml per day [26]. This may represent a median iron loss as high as 64 mg per day. As daily iron intake is less than 20 fold iron losses, iron deficiency could easily appear in critically ill patients.

Iron deficiency may thus coexist with inflammation. In addition, iron deficiency is not infrequent in the general population [28], and also in the elderly [3,29] or patients suffering from heart failure [30]. The frequency of iron deficiency on ICU admission may thus be around 35% [31,32]. However, the diagnosis of iron deficiency is difficult in the context of inflammation because the usual indicators of iron deficiency are no longer valid [9,10]. Because inflammation induces ferritin synthesis, serum ferritin levels are no longer indicative of iron stores. New biological markers are thus required for the diagnosis of iron deficiency in the context of inflammation (Figure 3) [10]. Below are the main biological markers that can be used:

- Percentage of hypochromic RBCs. These hypochromic RBCs result from iron-restricted erythropoiesis. Schematically, a value of > 10% hypochromic erythrocytes (normal < 2.5%) is indicative of iron-restricted erythropoiesis over the past 3 months (this being the RBC lifespan).
- Reticulocyte hemoglobin content. Reticulocyte hemoglobin content below 28 pg is also indicative of iron-restricted erythropoiesis over the past 2 to 3 days (this being the lifespan of reticulocytes). Recently, a low reticulocyte hemoglobin content on admission was shown to be associated with higher transfusion rates in critically ill patients [32].
- Erythrocyte zinc protoporphyrin (ZPP). During erythropoiesis, Fe is normally incorporated into protoporphyrin IX to form heme. In iron deficiency, zinc is substituted for iron, leading to the formation of ZPP. Increased erythrocyte ZPP is thus indicative of iron deficiency.
- Soluble transferrin receptor (sTfR). Transferrin receptors allow the internalization of iron into erythroid progenitor. Their synthesis is increased as bone marrow erythropoietic activity increases. When iron

supply is insufficient, a truncated form of transferrin receptor appears in the serum. sTfR is thus indicative of iron-deficiency anemia. This marker is widely proposed, however there is no gold standard for its measurement.

- sTfR/log ferritin ratio (called the ferritin index). This is proposed as a marker to differentiate between anemia of inflammation and the combined situation of iron deficiency and anemia of inflammation, taking into account the “uncovered need for iron” on the one hand and the “iron stores” on the other [15].

Complex algorithms combining all these variables have been proposed for the diagnosis of iron deficiency in the presence of inflammation [10,15]; however, none are clinically validated and the cut-off values for each variable are unknown. Moreover, all but sTfR cannot be used after recent blood transfusion.

Being central to iron metabolism, hepcidin may be a marker of iron deficiency, even in the presence of inflammation. Indeed, using animal models, we and others have demonstrated that hepcidin can be repressed despite inflammation [33–35] and that this repression is associated with spleen iron mobilization [34]. These observations reinforce the concept that iron deficiency may coexist with anemia of inflammation [15]. Measurement of hepcidin concentrations may thus be helpful for the diagnosis of iron deficiency in the context of inflammation. Additionally, many hepcidin assays have been recently developed [36]. Most studies evaluating the use of hepcidin concentrations to diagnose iron deficiency during inflammation have used ELISA-based values showing virtually undetectable levels [35] or normal values [37,38] of hepcidin despite inflammation (supposed to increase hepcidin synthesis). Measurement of hepcidin concentrations could be accurate in the diagnosis of iron deficiency in critically ill anemic patients using a cut-off value of less than 130 ng/l [38].

Is there a place for iron supplementation or treatment in critically ill patients?

Because iron deficiency may coexist with inflammation in critically ill patients [9,10,32,38] and because iron may be mobilized from spleen stores in the presence of inflammation [34,35], one could propose that iron be given to critically ill patients.

Because blood transfusion is not an option to fully correct the anemia in critically ill patients [6,21], the use of alternatives such as erythropoiesis-stimulating agents or iron has been suggested. Erythropoiesis-stimulating agents have already been studied in the critically ill. They have not been shown to be useful [39] and are beyond the scope of this review. In addition, iron deficiency may concern up to 40% of critically ill patients [10,31,32,38]. Iron may thus be needed not only for erythropoiesis but

	Iron deficiency	Anemia of inflammation	Iron deficiency and inflammation
Bone marrow iron	↓↓↓	↓	↓↓
Iron	↓↓	↓↓	↓↓
Transferrin	↗↗	↓↓	N to ↓
Transferrin saturation	↓↓	↓↓	↓↓
Ferritin	↓↓	↗↗	N to ↗
Percentage of hypochromic red blood cells	↗↗	N to ↗	↗↗
Reticulocyte hemoglobin content	↓↓	N or ↓	↓
Erythrocyte zinc protoporphyrin	↗↗	N to ↗	↗↗
sTfR	↗↗	↓↓	↗
sTfR/log ferritin	↗↗	↓	↗
Hepcidin	↓↓	↗↗	N to ↓
C-reactive Protein	N	↗↗	↗

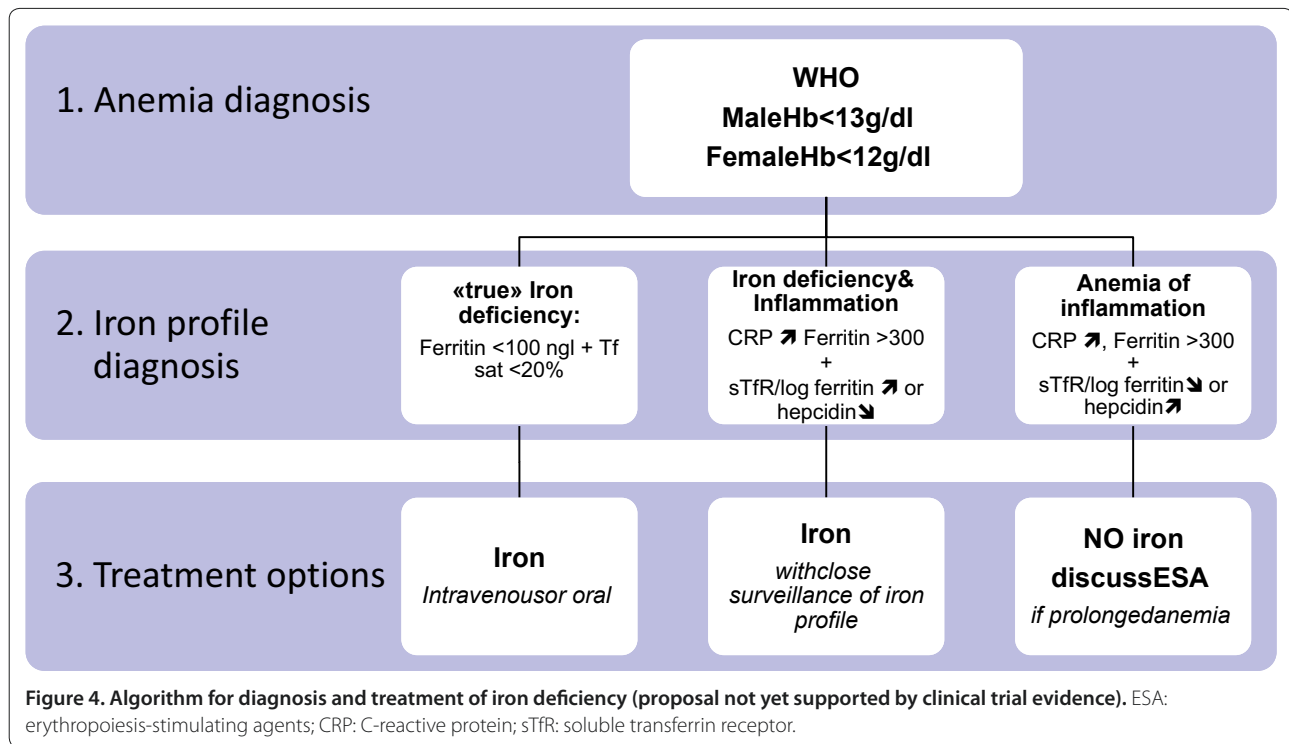
sTfR: soluble transferrin receptor; N: normal; ↓: decreased; ↗: increased

Figure 3 Biological variable of iron metabolism.

also to correct all the disorders associated with iron deficiency, having been shown to improve functional capacity in women [40] and in cardiac patients [41]. However, iron is also a toxic compound with the ability to induce oxidative stress or to promote bacterial growth and may thus not be suitable in the ICU context. Indeed, free iron may induce oxidative stress through the Fenton reaction. Large amounts of iron, exceeding the transferrin iron-binding capacity, may thus be toxic by inducing the release of free iron and causing oxidative stress. This probably explains the increased mortality associated with large amounts of iron administration (around the DL50) observed in an animal model of peritonitis [42]. However, no increase in oxidative stress has been demonstrated in human practice [43]. There is also a link between iron and infection, with iron being needed for bacterial growth. The decrease in serum iron concentration may be a defense mechanism against bacterial proliferation. However, bacteria have developed mechanisms for iron acquisition including the release of siderophores. The respective affinity for iron between transferrin and siderophores is probably what matters [44]. In clinical studies, this link between iron and infection has essentially been supported by experimental data on microorganisms and retrospective studies in hemodialysis patients showing an association between hyperferritinemia and the likelihood of infection. However, available observational studies in postoperative or critically ill patients show no association between intravenous iron administration and risk of infection [45]. Furthermore,

iron deficiency is associated with impaired immunity [46] and may, therefore, be responsible for increased susceptibility to infection [32] as well as being associated with increased length of stay in the ICU [31].

Iron may thus be suggested to correct iron deficiency, even in the presence of inflammation, similar to its proposed use in the treatment of patients with cancer-induced anemia [15,47]. Iron may be given using either intravenous or enteral routes. For the latter, ferrous iron is used. Iron absorption requires a mildly acidic medium (i.e., without concomitant use of proton pump inhibitors) and ascorbic acid. However, absorption may be reduced by inflammation because of the decrease in ferroportin levels induced by hepcidin, or because of frequent gastrointestinal adverse effects. The intravenous route allows administration of much higher doses with few adverse effects (with the notable exception of anaphylactic shock following iron dextran injections) and no difficulty of absorption. A recent meta-analysis showed that non-dextran iron was superior to enteral iron for the correction of anemia, with few adverse effects [48]. However, the only available study of intravenous iron showed no beneficial effect on erythropoiesis when used without erythropoiesis-stimulating agents [25]. The only study of iron deficiency treatment in critically ill patients is the study by Pieracci *et al.*, which showed a reduced transfusion rate in patients with baseline iron deficiency treated with enteral iron supplementation (ferrous sulfate 325 mg three times daily) [49]. In this study, oral iron supplementation was not associated with an increased risk of infection.



Iron may therefore be proposed either to correct iron deficiency and/or to enhance the response to erythropoiesis-stimulating agents in critically ill patients, but further studies are needed to rule out the potential risks of iron treatment (i.e., oxidative stress induction, increased risk of infection) and to define the best route of administration. In Figure 4, we propose an algorithm for iron deficiency diagnosis and treatment. We believe that iron should be given to critically ill patients only in cases of iron deficiency, at best defined according to a low hepcidin level. The dose of iron needed may be assessed using the following formula:

$$\text{iron deficit} = \text{body weight (kg)} \times (\text{target Hb} - \text{actual Hb}) \times 2.4.$$

Because elevated iron concentrations induce the synthesis of hepcidin, which in turn may reduce iron availability, the total dose of iron should be given using fractionated injections. Further clinical studies are needed to validate these propositions.

Conclusion

The discovery of hepcidin sheds new light on our knowledge of iron metabolism and may enable easier recognition of iron deficiency in the presence of inflammation in critically ill patients. This opens new areas of research exploring the role of iron treatment for these patients.

Competing interests

The authors declare that they have no competing interests.

List of abbreviations used

RBC: red blood cell; sTfR: soluble transferrin receptor; ZPP: zinc protoporphyrin.

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