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Realization of ZnO/PVK transparent heterojunction by solution processing routes

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Abstract

In this work, transparent heterojunction between zinc oxide (ZnO) and poly(*N*-vinyl carbazole) (PVK) was fabricated by solution processing techniques such as spin-coating and dip-coating techniques; then, its performance was studied using current (*I*)-voltage (*V*) measurement at room temperature. Before fabricating the heterojunction, initially, the growth characteristics of both thin films were independently optimized on a well-cleaned glass substrate, then its structural properties, optical properties, and surface topography were characterized using an X-ray diffractometer, UV-VIS-NIR spectrophotometer, and atomic force microscope, respectively. The structural analysis confirms the existence of a PVK thin film in amorphous nature and ZnO thin film in hexagonal crystal structure. The transparent nature of the heterojunction was found to be more than 85% in the visible and NIR regions with the absorption onset in the ultraviolet region. The observed experimental results explored the possibilities of fabricating ZnO/PVK transparent heterojunction by solution-based routes on a transparent fluorine-doped tin oxide substrate for transparent electronics applications.

Keywords: Transparent electronics, Atomic force microscopy, Heterojunction

Background

Materials which are displaying the remarkable combination of high electrical conductivity and optical transparency already form the basis of many important technological applications including flat panel displays, solar energy capture, and other optoelectronic devices [1]. Transparent electronic devices such as bipolar transistors and diodes can be fabricated with the help of p-n transparent junction. After the discovery of the first transparent semiconductor delafossite CuAlO₂ in 1997 by Kawazoe et al. [2], the development of a series of ptype transparent materials and transparent p-n junction devices such as UV-light-emitting diodes was triggered. Very recently, Snure et al. [3] discovered another p-type transparent oxide (CuBO₂) that also belongs to the delafossite family but exhibits better room-temperature electrical conductivity (1.65 S/cm). Zinc oxide (ZnO) is a transparent n-type semiconductor material having a wide bandgap (e.g., approximately 3.37 eV at room temperature) with large exciton binding energy of approximately 60 meV [4]. Such properties make them well suited for the realization of many optoelectronic applications. Poly(N-vinyl carbazole) (PVK) is one of the famous holes conducting polymeric material with amorphous nature and widely utilized in the fabrication of blue light-emitting diodes [5]. It has, by linear chains of repeated molecular groups $(H_2C-HC)_n$ with pendant carbazole side groups [(C₆H₄)₂NH], arranged randomly around the same chain. Since the electrical and optical properties of films strongly depend on their microstructure, stoichiometric nature, and impurities present, each deposition technique with its associated parameters yields films of different properties. Depending on the deposition method, the substrate can have a significant influence on the properties of the films. Particularly, the thermal expansion coefficient of the substrate affects the orientation of crystallites in the films and its electrical properties. Solution processing is envisioned as a key technology in the low-cost production of electronic devices on both conducting and non-conducting substrates. In this report, the formation of transparent heterojunction between ZnO and PVK thin films by solution processing routes and their performance were discussed.

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Methods

All the precursor and solvent materials were used in asreceived condition without any further purification. Transparent PVK solution was prepared in dichloromethane (DCM) by dissolving 0.5 g of PVK in 20 ml of DCM at room temperature. The well-cleaned substrates (glass or fluorine-doped tin oxide (FTO)) were dip-coated using freshly prepared polymer sol and then baked at 80°C for 15 min to get the PVK films. In the case of ZnO thin films, the sol was prepared [6] using the following procedure: isopropyl alcohol (136.4 ml) and diethanolamine (5.2 ml) were stirred for 15 min at room temperature; then, zinc acetate dehydrate (13.38 g) was added and stirred once again for 60 min then refluxed at 70°C for 30 min, cooled, and aged for 6 h. The use of metal acetate instead of the commonly used alkoxides will reduce the problem of moisture sensitivity like significant factors in assessing the effectiveness of a process. A commercially available flat motor with a fan which is able to rotate at 2,000 rpm on 12 V DC supply was used to spin coat the ZnO sol on the substrates. After deposition, the films were baked in a box furnace at 450°C for 10 min and allowed to reach room temperature before introducing the next layer. The coating process was repeated thrice to improve the crystalline nature of the ZnO film. Structural analysis of both thin films was carried out using a Philips X-ray diffractometer (Philips, Amsterdam, Netherlands). The atomic force microscopy studies were carried out in contact mode using Veeco Dimension 3100 SPM unit (Veeco Instruments Inc., Plainview, NY, USA), and a UV-VIS-NIR spectrophotometer was used to study the optical behavior of thin films between the wavelength region 300 to 1,100 nm. The commercial scientific software Origin 6.1 was used to plot the spectra and, in some cases, even for the analysis of spectra.

Results and discussions

X-ray diffraction patterns of the dip-coated PVK, Joint Committee on Powder Diffraction Standards (JCPDS) pattern (89-1397) of the ZnO sample, and spin-coated ZnO thin films on glass substrates are shown in Figure 1. The amorphous nature of the PVK thin film was revealed from the analysis. All the peaks of the spincoated ZnO film XRD pattern are perfectly indexed to the hexagonal phase of ZnO and are in good agreement with the JCPDS file 89-1397. The peaks which are shown in Figure 1c with 2θ values of 31.73°, 34.37°, 36.21°, 47.48°, 56.53°, 62.77°, and 67.86° correspond to the crystal planes of (100), (002), (101), (102), (110), (103), and (112), respectively. Both of the samples were further examined for their microstructural and optical properties using atomic force microscopy (AFM) and UV-VIS-NIR spectrophotometer. The AFM images of both films are shown in Figure 2. Within the scanned region, both films have an appreciable network structure, without porosity, pin holes, and cracks. As compared with the ZnO thin film, PVK does not have any features in the AFM image which also confirm the amorphous nature of the film. In general, films obtained by sol—gel coating with different microstructural and physical properties are achievable by varying the solution preparation parameters such as aging the solution prior to the coating.

UV–VIS-NIR spectrophotometer transmittance spectra of both thin films are shown in Figure 3. The transparent nature was found to be more than 85% in the visible and near infrared region for both films with the absorption onset in the ultraviolet region. The shape of the absorption edge in both thin films are solely due to the electronic transition from the top of the valence band to the bottom of the conduction band. For a direct bandgap semiconductor, such as ZnO, the absorbance in the vicinity of the onset is given [7] by

$$\alpha = \frac{C\left(hv - E_g^{\text{bulk}}\right)^{1/2}}{hv}$$

where α is the absorption coefficient, C is a constant, $h\nu$ is the photon energy, and $E_{\rm g}^{\rm bulk}$ is the bandgap. Hence, the direct bandgap energy calculated using the above formula with its absorption onset wavelength of 390 nm is 3.2 eV, which agrees well with the value reported in the literature [8].

Realization of heterojunction formation between n-ZnO thin film with other materials such as p-SrCu₂O₂ [9], p-SiC [10], p-Si [11], and p-GaN [12] initiated us to explore

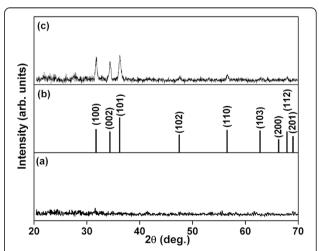
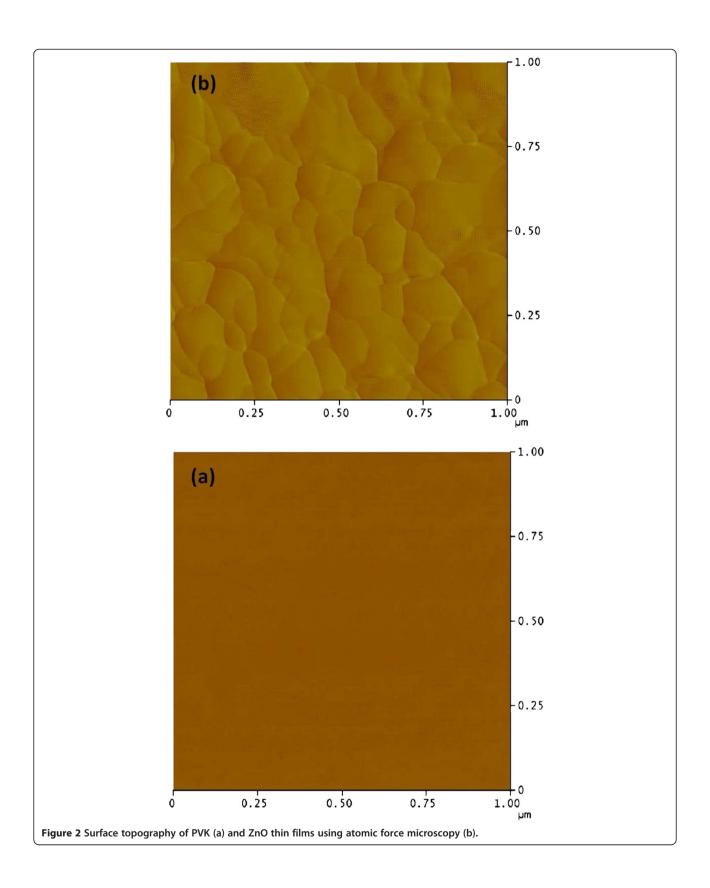
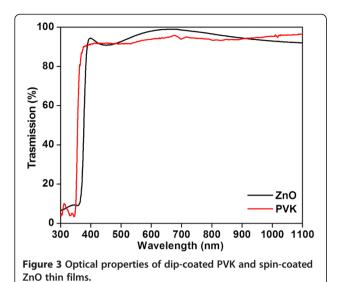


Figure 1 XRD patterns. (a) Dip-coated PVK thin film, (b) JCPDS standard pattern (89–1397) of ZnO sample, and (c) spin-coated ZnO thin film grown on glass substrate.





0.010
0.005
0.005
0.005
0.005
0.005
0.005
0.010
2 3
Voltage (V)
Figure 5 Current-voltage characteristics of the ZnO/PVK transparent heterojunction.

similar possibilities with polymer films. In order to fabricate a transparent heterojunction, ZnO thin film was deposited first on FTO substrate then characterized; the observed XRD pattern is shown as an inset in Figure 4. Substrate contribution due to the crystalline nature of the FTO substrate was evident from the XRD measurement. The optical quality was investigated using UV–VIS-NIR spectrophotometer, and it was shown in Figure 4. The oscillatory character of the films grown on FTO substrate may be due to interference effects between the sample and the FTO layer in the visible region. Then, on the top of ZnO film, a thin layer of PVK was deposited using the same protocol mentioned previously. *I-V* measurement was performed to study the junction formation; the observed

100

80

60

40

20

400

500

rasmission (%)

spectra are shown in Figure 5. The obtained forward and reverse bias breakdown characteristics confirm the junction formation between these layers. It helped us to realize the successful fabrication of a transparent diode by a simple solution-based film growth process.

Conclusions

Transparent ZnO/PVK heterojunction by solution processing techniques was successfully fabricated on solid substrates. Both of the films are uniform; more than 85% of optical transparency was achieved in the visible region, and atomic force microscopic investigation reveals our films are pin-hole and crack-free. Transparent heterojunction formation with low breakdown voltage (approximately 2.5 V) between ZnO and PVK films was confirmed by *I-V* measurement. All these experimental results confirm the possibilities of achieving heterojunction by solution-based routes.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

Both authors contributed equally to this manuscript and read and approved its final version.

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Wavelength (nm)

Figure 4 Optical properties of spin-coated ZnO thin film on FTO substrate. (Inset: XRD pattern of ZnO coated on the FTO substrate)

40

Intensity vs. 20 (deg.)

600 700 800

50

O - FTO

900 1000 1100

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