

NANO EXPRESS

Open Access

Charge storage characteristics of Au nanocrystal memory improved by the oxygen vacancy-reduced HfO₂ blocking layer

Ruifan Tang, Kai Huang*, Hongkai Lai*, Cheng Li, Zhiming Wu and Junyong Kang

Abstract

This study characterizes the charge storage characteristics of metal/HfO₂/Au nanocrystals (NCs)/SiO₂/Si and significantly improves memory performance and retention time by annealing the HfO₂ blocking layer in O₂ ambient at 400°C. Experimental evidence shows that the underlying mechanism can be effectively applied to reduce oxygen vacancy and suppress unwanted electron trap-assisted tunneling. A memory window of 1 V at an applied sweeping voltage of ±2 V is also shown. The low program/erase voltage (±2 V) and the promising retention performances indicate the potential application of NCs in low-voltage, non-volatile memory devices.

Keywords: Memory performance; Oxygen deficiency; Annealing; Non-volatile memory

PACS: 85.35.-p; 61.72.Cc

Background

Nanocrystal (NC) floating gate memory devices have recently attracted much attention as a strong candidate for non-volatile memories given their scalability, fast write/erase speeds, low operating voltages, and long retention times [1-4]. Numerous attempts have been made to develop non-volatile memory devices using metal NCs, such as Ni [5], Au [6], Ir [7], and Pt [8], because metal NCs have a higher density of states around the Fermi level, a wider range of available work functions, and smaller energy perturbation compared with their semiconductor counterparts [9]. Further improvement in memory performance can be achieved through the integration of metal NCs with high- κ dielectric materials, such as HfO₂ [10] and Al₂O₃ [11]. The use of high- κ dielectric materials as blocking layers decreases the electric field at the top dielectric and program/erase (P/E) voltages, which also supports the demand for small effective oxide thickness [12]. Au NCs with high work functions (5.1 eV) enable the creation of a deep potential well to trap charge carriers, such as HfO₂, with high dielectric constants (20 to 25) and relatively high barrier heights (-5.7 eV). The structure

of metal/HfO₂/Au NCs/SiO₂/Si shows a strong potential for application in non-volatile memory devices [13,14].

Metal/HfO₂/Au NCs/SiO₂/Si is fabricated in this study. The capacitance-voltage (C-V) characteristics show that the main storage consists of holes. However, electron trapping is seldom achieved because of the HfO₂ blocking layer. X-ray photoelectron spectroscopy (XPS) confirms that the oxygen deficiency within the HfO₂ layer is caused by the presence of Hf-Hf bonding. The energy band diagram shows that electrons trapped in the NCs tend to leak into the gate electrode through trap-assisted tunneling, which is supported by the oxygen vacancy-related levels during programming. However, Hf-Hf bonding disappears after HfO₂ is annealed at 400°C for 10 min in O₂ ambient. The structure of metal/HfO₂ (as-annealed)/Au NCs/SiO₂/Si shows that both electrons and holes are stored. Given their memory window of 1 V at an applied sweeping voltage of ±2 V, low P/E voltage (±2 V), and promising retention performances, low-voltage NC memories have a strong potential for application in non-volatile memory devices.

Methods

A metal/HfO₂/Au NCs/SiO₂/Si (A₁) structure was fabricated. P-type Si with a doping level of $8.33 \times 10^{17} \text{ cm}^{-3}$ was used as a substrate. A 3-nm-thick thermal SiO₂ oxide was fabricated using a rapid thermal annealing (RTA) device

* Correspondence: k_huang@xmu.edu.cn; hkLai@xmu.edu.cn
Semiconductor Photonics Research Center, Department of Physics, Xiamen University, Xiamen 361005, China

after pre-gate cleaning. An Au film with a thickness of approximately 1 nm was sputtered using SCD005 (Balzers Union, Balzers, Liechtenstein) with a sputtering time of 2 s. The sample was then annealed in N₂ ambient using the RTA device. Annealing was performed at 600°C for 10 s to form Au NCs. A 30-nm HfO₂ film deposited by the electron beam (E-beam) evaporation system with a base pressure of 3.6×10^{-6} Torr served as the blocking layer. After depositing the TaN/Al metal gate electrode with thicknesses of 50/300 nm and the Cr/Au bottom electrode with thicknesses of 20/200 nm through magnetron sputtering, the capacitive structure of the NC memory device was finally completed. Metal/HfO₂/SiO₂/Si (A₂), metal/SiO₂/Au NCs/SiO₂/Si (A₃), and metal/HfO₂ (PDA)/Au NCs/SiO₂/Si (A₄) were fabricated using the same process, with the exception of a 20-nm SiO₂ film deposition using the E-beam for sample A₃ and the annealing of HfO₂ after deposition at 400°C for 10 min in the O₂ ambient for sample A₄. XPS with a 1,486.6-eV Al K α source was used to obtain composition information about the as-deposited and annealed HfO₂ film. The electrical characteristics of the NC memory devices were measured in the parallel mode using a Keithley 4200 semiconductor characterization system (Cleveland, OH, USA) and a Keithley 590 C-V analyzer at room temperature.

Results and discussion

Figure 1 shows the cross-sectional high-resolution transmission electron microscopy (HRTEM) micrograph of the A₁ device. The Au NCs formed on the 3-nm thermal SiO₂ are covered with a 30-nm HfO₂ layer. The NC density is approximately 8×10^{11} cm⁻², wherein the size is mainly distributed from 6 to 8 nm. The charging properties are described from the C-V measurements at 1 MHz with a step of 0.1 V/s for A₁ (Figure 2a). Double C-V sweeps are performed with voltage sweeps from inversion to accumulation, i.e., from positive to negative bias and back

to inversion to give prominence to the charge trapping in the Au NCs. Electron and hole trapping in the NCs are enabled by the positive and negative biases, respectively. The positive flat band voltage shifts (ΔV) correspond to an increase in electron trapping, whereas the negative ΔV corresponds to the increase in hole trapping given the increasing sweep voltage range. Figure 2a shows that the negative ΔV is about 1.05 V, whereas the positive ΔV is close to 0, which indicates that no additional electrons can be trapped with the increase in the sweep range. The inset plot in Figure 2a shows the C-V curves of sample A₂. Sample A₂ showed no apparent hysteresis loop both at the ± 2 and ± 4 V bias sweep, indicating that a charging effect only occurs with Au NCs. The electron's energy barrier of 3.2 eV between Si and SiO₂ is known to be much less than that of the hole (4.7 eV). Electron tunneling is expected to be easier than hole tunneling. However, the C-V characteristic shown here indicates that electron trapping is more difficult than hole trapping. One possible reason is because the electrons trapped in the Au NCs leak back to the substrate and result in lessened electron trapping, which is similar to previous reports [15]. In previous reports, a band offset exists at the valence band between Ge and Si. Holes can be trapped in Ge_{1-x}Si_x/Si heteronanocrystals, whereas electrons tunnel back to the substrate directly through the ultrathin tunnel oxide. However, these reports are inconsistent with our experiments because no additional barrier layer for holes exists in our experiments; thus, lessened electron trapping cannot be attributed to electron loss in thin tunnel oxide.

Another possible mechanism leading to electron injection from the inverted substrate into the Au NCs during programming is the positive gate bias. Electrons are emitted from the NCs, which cross the HfO₂ blocking layer to the gate electrode [16]. Sample A₃ is fabricated with SiO₂ as the blocking layer to investigate the effect of HfO₂ and the possible mechanism. The control oxide

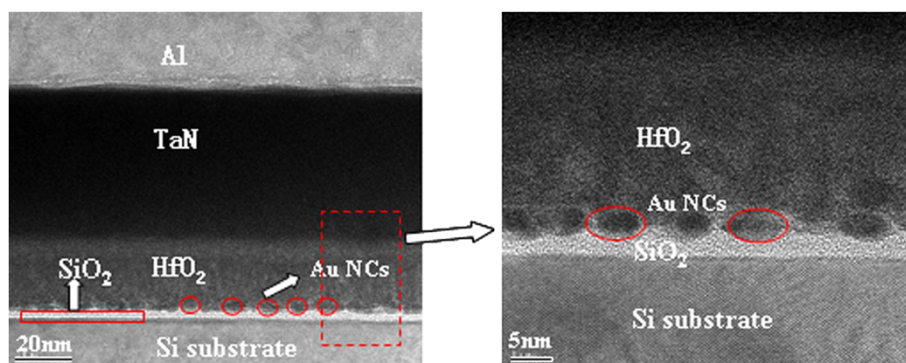
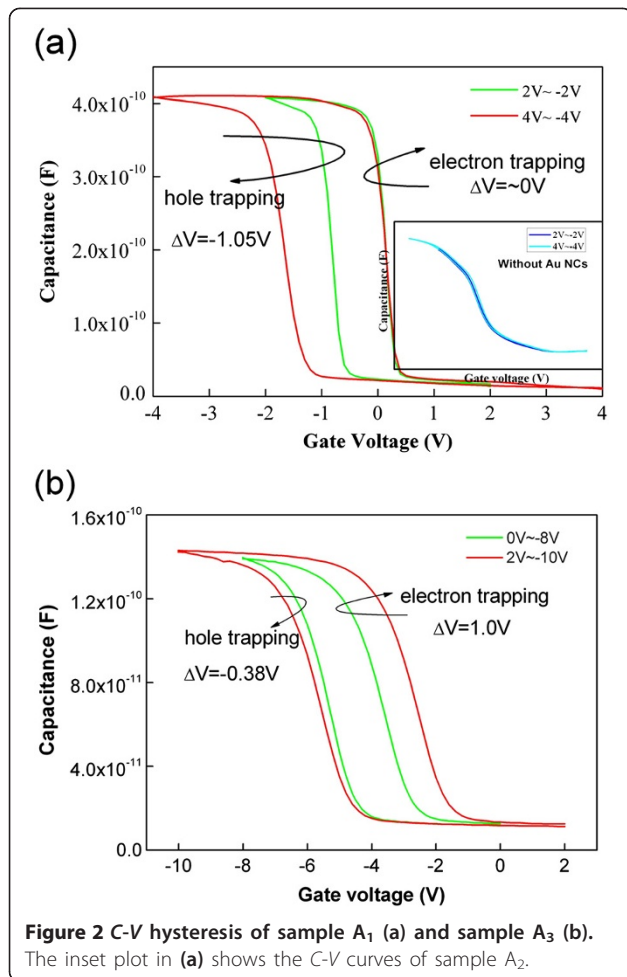
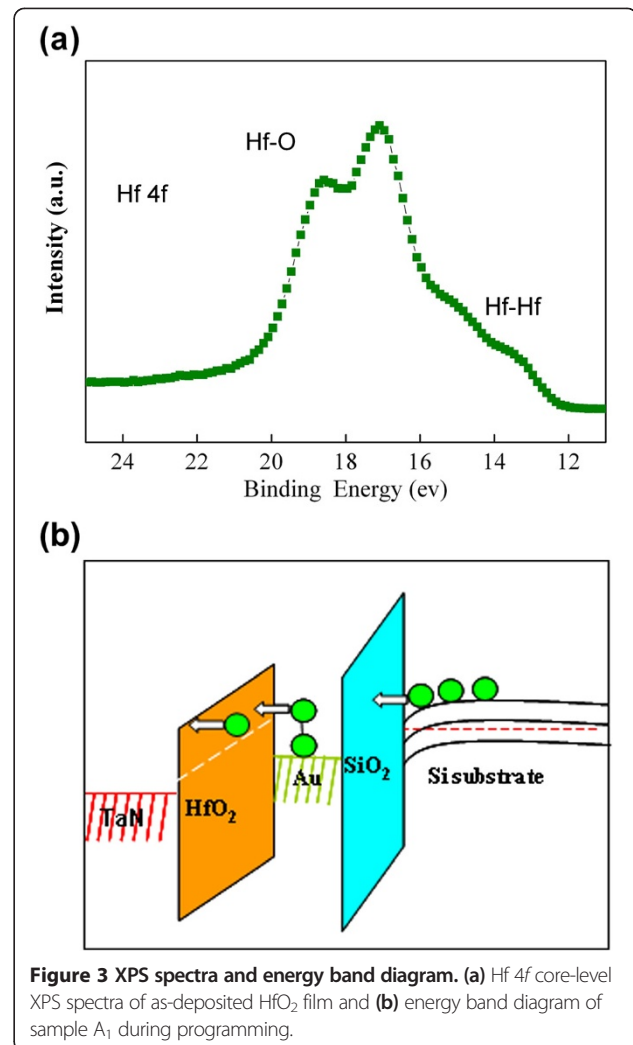


Figure 1 Cross-sectional HRTEM micrograph of sample A₁.

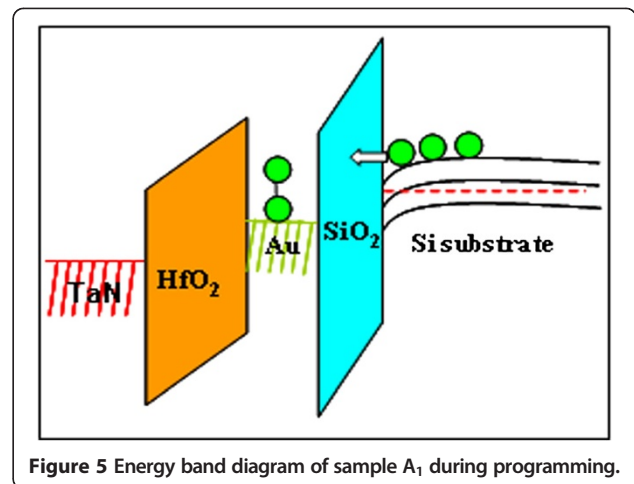
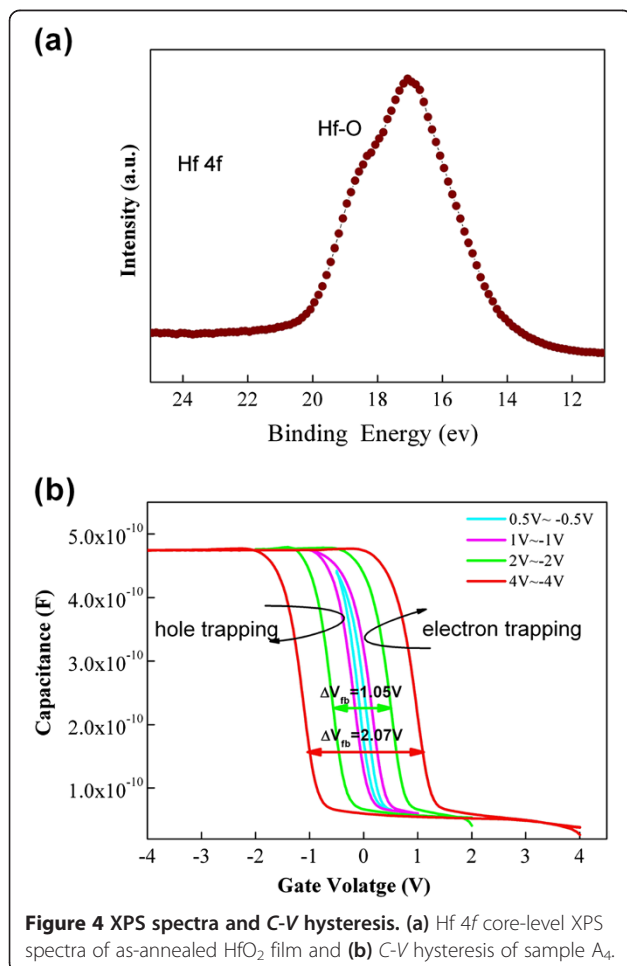


thickness of SiO₂ in sample A₃ is noted to be about 20 nm; to lessen the electric field differences between samples A₁ and A₃ during the sweep process, the sweeps are performed from -8 to 0 V and -10 to 2 V. Figure 2b shows the C-V hysteresis curves for A₃ with sweep ranges of -8 to 0 V and -10 to 2 V. The positive ΔV is approximately 1 V and is greater than the negative ΔV (0.38 V) with the increase in sweep range. A high positive ΔV value indicates that both electrons and holes can be stored in NCs. Electron trapping is also easier than hole trapping, which is consistent with previously reported theories and results [17,18]. Therefore, the asymmetric C-V hysteresis curve of A₁ is reasonably caused by the HfO₂ blocking layer. The HfO₂ films prepared using different growth methods have different microstructures and properties [19]. XPS measurements are performed using our E-beam device to investigate the composition information of the as-deposited HfO₂ film. About 2 nm of the sample top layer was removed using Ar ion bombardment to remove surface contaminants. Figure 3a shows the two peaks at 17.1 and 18.6 eV, which correspond to the Hf 4f



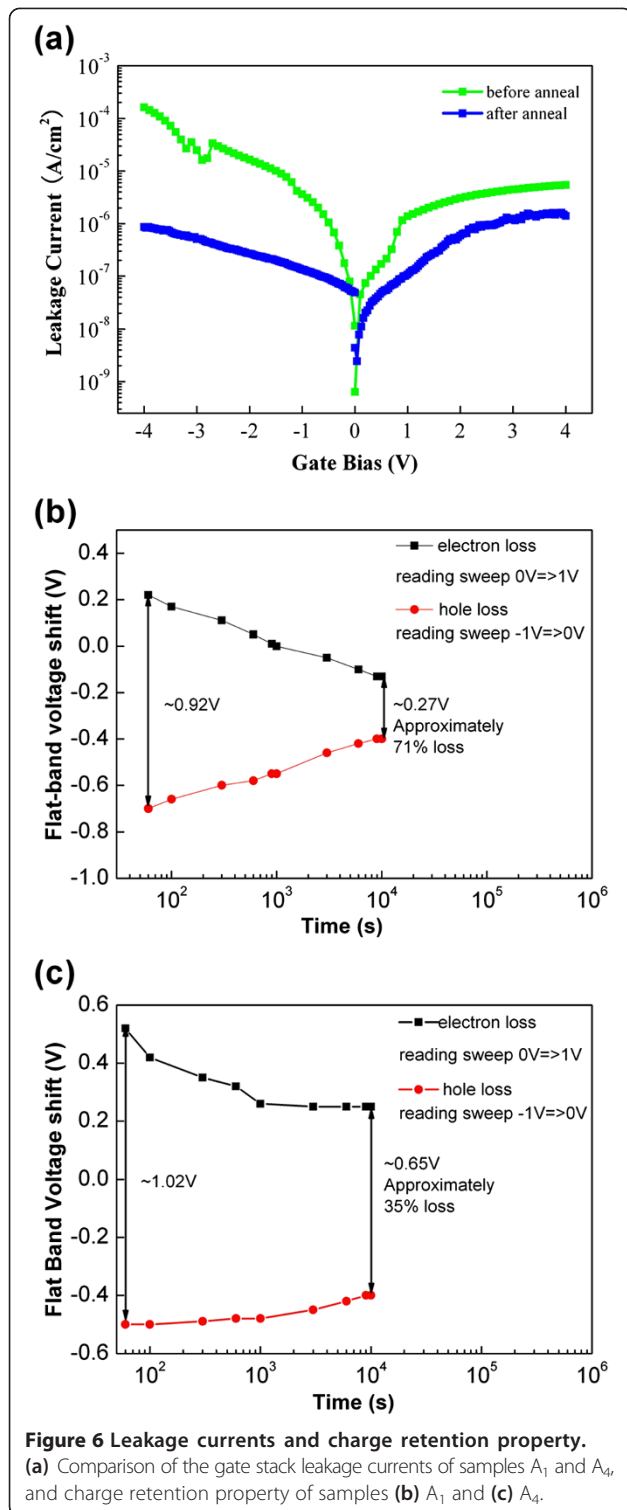
and Hf 4f peaks from HfO₂. Small but noticeable shoulders at the lower binding energy side of the main peak were also observed, which can be attributed to Hf-Hf bonding and indicate the existence of oxygen vacancy within the HfO₂ film [20]. Oxygen vacancy reportedly results in oxygen vacancy-related levels within the bandgap [21]. Takeuchi et al. used spectroscopic ellipsometry to demonstrate the existence of shallow oxygen vacancy-related defects 1.2 eV below the HfO₂ conduction band [22]. Given the existence of an oxygen vacancy-related level below the conduction band and the rise of electron potential because of electron trapping in the NCs [23], electrons trapped in Au NCs could possibly leak into the gate electrode through the trap-assisted tunneling method during the programming operation (Figure 3b). This method is similar to the multi-phonon-assisted tunneling model described in previous reports [24]. The trap-assisted tunneling effect may be responsible for the minimal electron storage.

HfO₂ was annealed after deposition at 400°C in O₂ ambient to verify this assumption. XPS analysis was performed on the O₂-annealed HfO₂ film after 2 nm of the HfO₂ top layer was removed by Ar ion bombardment to remove the surface contaminants. Figure 4a shows that no evidence of Hf-Hf bonding was observed, with the exception of the characteristic peak attributed to Hf-O bonds. This lack of evidence suggests that the annealing process can effectively reduce the oxygen vacancy of HfO₂ films. Sample A₄ was fabricated using the O₂-annealed HfO₂ as blocking layer. Figure 4b shows the C-V characteristics of A₄. The positive ΔV is almost similar to the negative ΔV with the increase in the sweep voltage range, thereby indicating that both electrons and holes can be easily stored in the Au NCs. The ease of electron and hole storage is caused by the reduced oxygen vacancy levels and the suppressed unwanted electron trap-assisted tunneling performed during programming, which leads to electron storage (Figure 5). Electron storage can be confirmed further through a comparison of



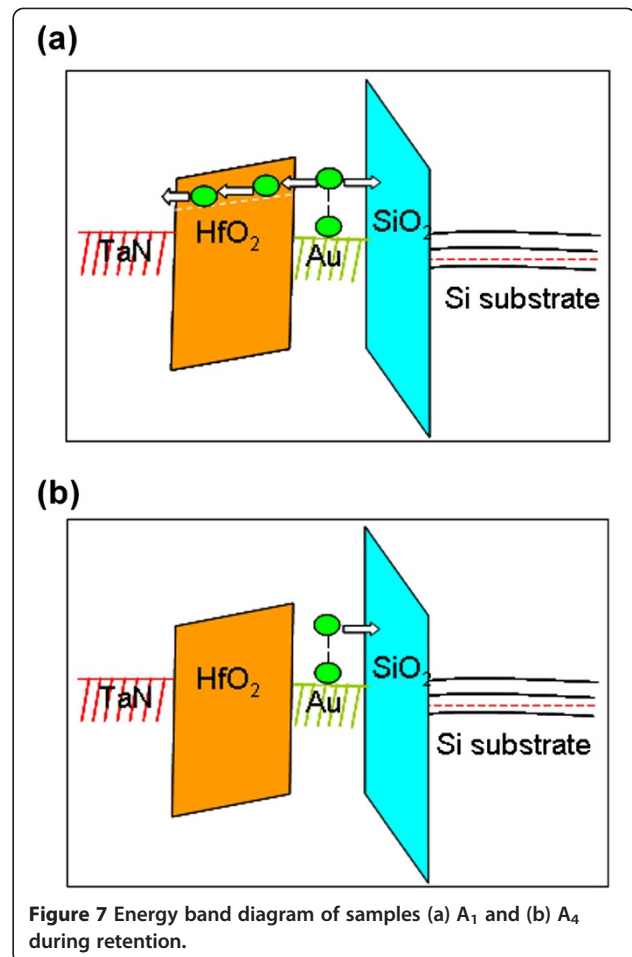
A₁ and A₄'s gate current characteristics. Figure 6a shows that sample A₄, with an O₂-annealed HfO₂, shows lower leakage current density at all regimes of the gate voltage compared with sample A₁, with an as-deposited HfO₂. The lower leakage current indicates that the reduced oxygen vacancy-related levels suppress electron injection from both the substrate and gate given that the positive gate voltage corresponds to substrate injection and the negative gate voltage corresponds to gate injection. Figure 6b,c shows the retention properties of A₁ and A₄. The initial memory windows are 0.92 and 1.02 V for A₁ and A₄, respectively. The windows are followed using a suitable reading condition. The decayed charges for sample A₄ with O₂-annealed HfO₂ were only 35% within a 10⁴-s span, which is much better than that of A₁ (approximately 71% loss). The difference between the observed retention behavior of A₁ and A₄ could be explained by the energy band diagram, which is based on the existence of oxygen vacancy-related levels. Figure 7a shows that the electrons trapped in the Au NCs leak into the gate electrode through the HfO₂ layer via electron tunneling to the oxygen vacancy-related level, as proposed in [24]; therefore, discharging easily occurs. However, the reduced oxygen-related levels in sample A₄ HfO₂ layer suppress the unwanted trap-assisted tunneling (Figure 7b); thus, electron loss rate is reduced.

A 1-V memory window was observed for A₄ at the ± 2 -V sweep (Figure 8), which shows the potential to prepare a low-voltage NC memory. The P/E operation was also performed by applying ± 2 -V pulses to the gate electrode. Figure 8 shows that a 1-V memory window can be obtained at P/E times of 10/10 ms, which shows a sufficient memory window even at a ± 2 -V applied pulse voltage. Given the improvements in the retention performances (Figure 6c), sample A₄ shows promise for application in low-voltage NC memory.

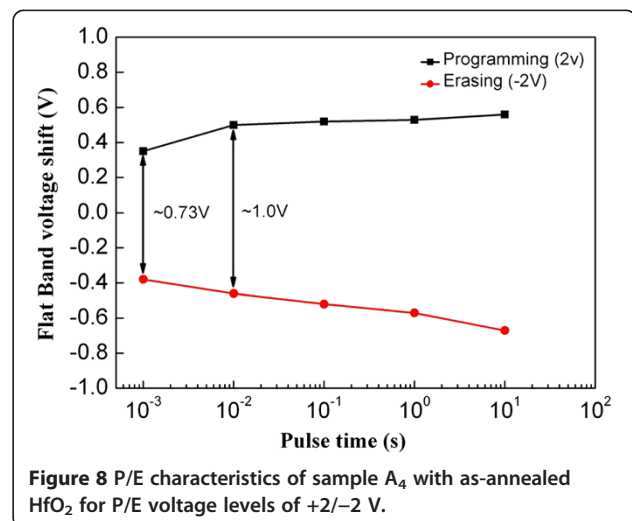


Conclusions

Electrons trapped in Au NCs tend to tunnel into the gate electrode through the oxygen vacancy-related levels of the HfO₂ blocking layer and tend to degrade memory performance because of the existence of oxygen vacancy.



Annealing the HfO₂ blocking layer at 400°C in O₂ ambient decreases oxygen vacancy and suppresses unwanted electron trap-assisted tunneling. Given their memory window of 1 V at an applied sweeping voltage of ± 2 V, low P/E voltage of ± 2 V, and improved retention performances,



low-voltage NC memories show promise for application in non-volatile memory devices.

Abbreviations

E-beam: Electron beam; HRTEM: High-resolution transmission electron microscopy; NCs: Nanocrystals; P/E: Programming/erasing; RTA: Rapid thermal annealing; XPS: X-ray photoelectron spectroscopy.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

RT carried out the experiments studied on the device fabrication and drafted the manuscript. KH designed the research programs and guided the experiment's progress. HL, CL, ZW, and JK participated in the mechanism development. All authors read and approved the final manuscript.

Acknowledgements

This work was supported by the National Basic Research Program of China under grant numbers 2011CB301905 and 2012CB933503; National Natural Science Foundation of China under grant numbers 61108064, 61036003, and 61176092; the Fundamental Research Funds for the Central Universities (2011120143); and Ph.D. Programs Foundation of Ministry of Education of China (20110121110025).

Received: 2 May 2013 Accepted: 21 August 2013

Published: 28 August 2013

References

1. Yang FM, Liu PT, Chang TC: Using double layer CoSi nanocrystals to improve the memory effects of nonvolatile memory devices. *Appl Phys Lett* 2007, **90**:212108.
2. Yang HG, Shi Y, Bu HM, Wu J, Zhao B, Yuan XL, Zheng YD: Simulation of electron storage in Ge/Si hetero-nanocrystal memory. *Solid-State Electron* 2001, **45**:767.
3. Lee C, Kwon JH, Sohn BH: Nonvolatile nanocrystal charge trap flash memory devices using a micellar route to ordered arrays of cobalt nanocrystals. *Appl Phys Lett* 2007, **91**:153506.
4. Lee JS, Yang JY, Hong JP: Charge trap memory characteristics of AlO_x shell-Al core nanoparticles embedded in HfO₂ gate oxide matrix. *Appl Phys Lett* 2009, **95**:052109.
5. Tan Z, Samanta SK, Yoo WJ, Lee S: Self-assembly of Ni nanocrystals on HfO₂ and N-assisted Ni confinement for nonvolatile memory application. *Appl Phys Lett* 2005, **86**:013107.
6. Mikhelashvili V, Meyler B, Yoffis S, Garbrecht M: A nonvolatile memory capacitor based on Au nanocrystals with HfO₂ tunneling and blocking layers. *Appl Phys Lett* 2011, **98**:212902.
7. Wang TT-J, Chu CL, Hsieh IJ, Tseng WS: Formation of iridium nanocrystals with highly thermal stability for the applications of nonvolatile memory device with excellent trapping ability. *Appl Phys Lett* 2010, **97**:143507.
8. Jeff RC Jr, Yun M, Ramalingam B, Triplett G, Gangopadhyay S: Charge storage characteristics of ultra-small Pt nanoparticle embedded GaAs based non-volatile memory. *Appl Phys Lett* 2011, **99**:212902.
9. Liu ZT, Lee C, Narayanan V, Pei G, Kan EC: Metal nanocrystal memories—part I: device design and fabrication. *IEEE Trans Electron Devices* 2002, **49**:9.
10. Kim JH, Yang JY, Lee JS, Hong JP: Memory characteristics of cobalt-silicide nanocrystals embedded in HfO₂ gate oxide for nonvolatile nanocrystal flash devices. *Appl Phys Lett* 2008, **92**:013512.
11. Wang C-C, Chiou Y-K, Chang C-H, Tseng J-Y, Wu L-J, Chen C-Y, Wu T-B: Memory characteristics of Au nanocrystals embedded in metal-oxide-semiconductor structure by using atomic-layer-deposited Al₂O₃ as control oxide. *J Phys D: Appl Phys* 2007, **40**:1673.
12. Lee C-H, Hur S-H, Shin Y-C, Choi J-H, Park D-G, Kim K: Charge-trapping device structure of SiO₂/SiN/high-*k* dielectric Al₂O₃ for high-density flash memory. *Appl Phys Lett* 2005, **86**:152908.
13. Mikhelashvili V, Meyler B, Yoffis S, Shneider Y, Salzman J, Eisenstein G: Nonvolatile low-voltage memory transistor based on SiO₂ tunneling and HfO₂ blocking layers with charge storage in Au nanocrystals. *Appl Phys Lett* 2011, **98**:212902.
14. Mikhelashvili V, Meyler B, Yoffis S, Shneider Y, Salzman J, Eisenstein G: Optical properties of nonvolatile memory capacitors based on gold nanoparticles and SiO₂-HfO₂ sublayers. *Appl Phys Lett* 2011, **98**:022905.
15. Lu J, Zuo Z, Chen YB, Shi Y: Charge storage characteristics in metal-oxide-semiconductor memory structure based on gradual Ge_{1-x}Si_x/Si heteronanocrystals. *Appl Phys Lett* 2008, **90**:013105.
16. Compagnoni C-M, Iemini D, Spinelli A, Lacaíta A-L: Modeling of tunneling P/E for nanocrystal memories. *IEEE Trans Electron Devices* 2005, **52**:569.
17. Dufourcq J, Bodnar S, Gay G, Lafond D, Vandroux L: High density platinum nanocrystals for non-volatile memory applications. *Appl Phys Lett* 2008, **92**:073102.
18. Lee JJ, Kwong DL: Metal nanocrystal memory with high-*k* tunneling barrier for improved data retention. *IEEE Trans Electron Devices* 2005, **52**:507.
19. Filatova EO, Sokolov AA, Kozhevnikov IV, Taracheva EY, Braun W: Investigation of the structure of thin HfO₂ films by soft x-ray reflectometry techniques. *J Phys Condens Matter* 2009, **21**:180512.
20. Chen B, Jha R, Misra V: Work function tuning via interface dipole by ultrathin reaction layers using AlTa and AlTaN alloys. *IEEE Trans Electron Devices* 2006, **27**:731.
21. Ramo D-M, Gavartin J-L, Shluger A-L: Spectroscopic properties of oxygen vacancies in monoclinic HfO₂ calculated with periodic and embedded cluster density functional theory. *Phys Rev B* 2007, **75**:205336.
22. Takeuchi H, Ha D, King T-J: Observation of bulk HfO₂ defects by spectroscopic ellipsometry. *J Vac Sci Technol A* 2004, **22**:1337.
23. She M, King T-J: Impact of crystal size and tunnel dielectric on semiconductor nanocrystal memory performance. *IEEE Trans Electron Devices* 1934, **2003**:50.
24. Lwin ZZ, Pey KL, Zhang Q, Bosman M, Liu Q, Gan CL, Singh PK, Mahapatra S: Study of charge distribution and charge loss in dual-layer metal-nanocrystal-embedded high-*k*/SiO₂ gate stack. *Appl Phys Lett* 2012, **100**:193109.

doi:10.1186/1556-276X-8-368

Cite this article as: Tang et al.: Charge storage characteristics of Au nanocrystal memory improved by the oxygen vacancy-reduced HfO₂ blocking layer. *Nanoscale Research Letters* 2013 **8**:368.

Submit your manuscript to a SpringerOpen® journal and benefit from:

- Convenient online submission
- Rigorous peer review
- Immediate publication on acceptance
- Open access: articles freely available online
- High visibility within the field
- Retaining the copyright to your article

Submit your next manuscript at ► springeropen.com