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Interface properties of SiO_xN_y layer on Si prepared by atmospheric-pressure plasma oxidation-nitridation

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Abstract

 SiO_xN_y films with a low nitrogen concentration (< 4%) have been prepared on Si substrates at 400°C by atmospheric-pressure plasma oxidation-nitridation process using O_2 and N_2 as gaseous precursors diluted in He. Interface properties of SiO_xN_y films have been investigated by analyzing high-frequency and quasistatic capacitance-voltage characteristics of metal-oxide-semiconductor capacitors. It is found that addition of N into the oxide increases both interface state density (D_{it}) and positive fixed charge density (Q_f). After forming gas anneal, D_{it} decreases largely with decreasing N_2/O_2 flow ratio from 1 to 0.01 while the change of Q_f is insignificant. These results suggest that low N_2/O_2 flow ratio is a key parameter to achieve a low D_{it} and relatively high Q_f , which is effective for field effect passivation of n-type Si surfaces.

Keywords: SiO_xN_y film, Interface properties, Interface state density, Atmospheric-pressure plasma, Plasma oxidationnitridation

Background

Silicon oxynitride (SiO_xN_y) is a very useful material for applications in microelectronic and optoelectronic devices due to the possibility of tailoring the film composition and property according to the O/N ratio. Recently, considerable attention has been focused on SiO_xN_y for anti-reflection coatings and surface passivation films for thin crystalline Si solar cells [1-3]. It has been reported that SiO_xN_y films with high positive fixed charge density (Q_f) in the range of 10¹² cm⁻² is effective for field-effect passivation of n-type Si surfaces [2].

So far, several methods have been applied to grow SiO_xN_y films. For example, high-temperature (>900°C) processes such as the direct thermal oxynitridation of Si in NO or N₂O ambient [4,5] and the annealing of SiO₂ in nitrogencontaining ambient [6,7] have been widely used. However, the high-temperature processes suffer a large thermal budget and a redistribution problem of dopant atoms. Plasma-enhanced chemical vapor deposition (PECVD)

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process is a low-temperature alternative below 400°C [8-10]. However, the PECVD method needs toxic precursor gases, and it is also noted that the interfacial properties prepared by this method are usually inferior to those of thermal oxides [11], because the deposition method does not consume the substrate Si unlike thermal oxidation. Moreover, in the films prepared by low-temperature PECVD, the concentration of hydrogen atoms in the form of Si-OH and Si-H bonds is high, which are responsible for poor dielectric properties [12]. Nitridation of silicon oxide in lowpressure nitrogen plasma has also been investigated to fabricate SiO_xN_y at low temperatures [13,14]. In the case of low-pressure nitrogen plasma, the ion bombardment of the film surface is a serious problem to develop highly reliable ultra-large-scale integrated circuits [15]. Recently, we have studied the plasma oxidation of Si wafers to grow SiO₂ films using atmospheric-pressure (AP) plasma generated by a 150-MHz very-high-frequency (VHF) electric field and demonstrated that high-quality SiO_2 films can be obtained using He/O₂ or Ar/O₂ plasma at 400°C [16,17]. We have also reported that the AP VHF plasma oxidation process at 400°C is capable of producing material quality of SiO₂ films comparable to those of

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high-temperature (>1,000°C) thermal oxides. The SiO₂/Si structure with low interface state density (D_{it}) around the midgap of 1.4×10^{10} cm⁻² eV⁻¹ and moderately high Q_f of 5.3×10^{11} cm⁻² has been demonstrated [18]. Therefore, addition of N into the SiO₂ film by AP plasma oxidation-nitridation using O₂ and N₂ precursor gas mixture is an alternative approach for obtaining SiO_xN_y films at a low temperature of 400°C.

The purpose of this work is to present a method for preparing SiO_xN_y films by AP VHF plasma oxidationnitridation with a detailed analysis of interface properties of SiO_xN_y layer by capacitance-voltage (*C*-*V*) measurements on metal-SiO_xN_y-Si capacitors.

Methods

The details of the AP VHF plasma apparatus have been reported previously [18]. A schematic illustration of an electrode for AP VHF plasma oxidation-nitridation is shown in Figure 1. In the gap between the substrate and parallel-plate electrode, stable plasma is generated at atmospheric pressure with 150-MHz VHF power using a gas mixture of 1% O_2 /He. N_2 gas was simultaneously introduced into the AP VHF plasma with gas flow rates of 1, 10, and 100 sccm. The N_2/O_2 gas flow ratios were 0.01, 0.1, and 1. The temperature of the Si wafer was fixed at 400°C by monitoring by a thermocouple embedded in the substrate heating stage. The detailed experimental conditions are shown in Table 1.

The substrates used in the present experiments were ntype (001) CZ-Si wafers (4-in. diameter) with a resistivity of 1 to 10 Ω cm. They were cleaned by a roomtemperature chemical cleaning method [19] and were finished by a diluted HF treatment. After AP plasma oxidation-nitridation, some of the samples were subjected to a forming gas anneal (FGA) in 10% H₂/He for 30 min at 400°C. In order to investigate Q_f and D_{it} of the SiO_xN_y film, Al/SiO_xN_y/Si metal-oxide-semiconductor (MOS) capacitors were fabricated with 0.5-mm-diameter Al pads by vacuum deposition. A back contacting electrode at the rear Si surface was also made by Al deposition.



 Table 1 Oxidation-nitridation conditions for Si wafer

Condition	Value
Pressure (Torr)	760
O ₂ concentration (%)	1
He flow rate (slm)	10
O ₂ flow rate (sccm)	100
N ₂ flow rate (sccm)	1,10, and 100
VHF (MHz)	150
VHF power (W)	1,000 to 1,500
Plasma gap (mm)	0.8 to 1
Substrate temperature (°C)	400
Oxidation-nitridation time (min)	9 to 25

The thickness of the SiO_xN_y layer was determined by ellipsometry (Rudolph Auto EL III) with a wavelength of 632.8 nm. The chemical bonding in the material was investigated by Fourier transform infrared absorption (FTIR) spectrometry (Shimadzu FTIR–8600PC) in the wave number range of 400 to 4,000 cm⁻¹. X-ray photoelectron spectroscopy (XPS; ULVAC-PHI Quantum 2000) was used to investigate the depth profile of atomic composition and bonding of atoms in SiO_xN_y films. High-frequency (HF) and quasistatic (QS) *C-V* measurements were performed using a 1-MHz C meter/CV plotter (HP 4280A) and quasistatic CV meter (Keithley 595), respectively.

Results and discussion

Thicknesses of films prepared at 400°C for 9 min under N_2/O_2 flow ratios of 0.01, 0.1, and 1 were 20.8, 19.5, and 18.9 nm, respectively. (The film thickness was a mean value for measurements of eight different sites on the sample.) Since the difference in the film thickness is





small (<±5%), its effect on the interface state properties may be negligible. Figure 2 shows FTIR spectra of the films prepared at 400°C for 9 min under different N₂/O₂ flow ratios. The dotted lines in Figure 2 indicate the stretching and bending vibration modes of Si-O-Si bonds at the wave numbers of 1,075 and 810 cm⁻¹, respectively. Almost no apparent peak for Si-N stretching mode at 835 cm⁻¹ is observed [1], which may be related with the larger dissociation energy of N₂ than that of O₂ molecules.

In Figure 2, the strongest peak in IR spectra corresponds to Si-O-Si stretching mode, indicating that the film consists predominantly of SiO₂. The dielectric constant of the film was calculated using the maximum accumulation capacitance obtained by *C-V* curves. The result showed that the dielectric constant was fairly uniform over the sample area with a variation of about 2% and that the average dielectric constants of the films were 4.26 and 4.01 for N₂/O₂ flow ratios of 0.01 and 1, respectively. Since the dielectric constants of SiO₂ and Si₃N₄ are 3.9 and 7.5, respectively, nitrogen atoms are considered to be incorporated in the SiO₂ structure.

XPS spectra in the Si 2p region for the SiO_xN_y layer formed at 400°C for 9 min with a N₂/O₂ gas flow ratio of 0.1 are shown in Figure 3. The Si 2p peak observed at 99.7 eV is from the Si substrate and the one at 103.5 eV



from Si-O-Si bonding. On the as-grown sample, as shown in Figure 3a, after five times of surface layer sputtering by 10-keV Ar ions (duration of one sputtering is 10 s), Si-O-Si bonding peak is strong, but a small peak from the Si substrate is also seen. By the sixth and seventh sputtering, the Si-O-Si peak decreases and the bulk Si peak increases. It is noteworthy that Si-N bonding at 102.4 eV is also detected. Since the Si-N peak becomes clear before the Si-O-Si peak vanishes, Si-N bonding is supposed to be located at the SiO₂/Si interface region. In the annealed sample, as shown in Figure 3b, the decrease of the Si-O-Si peak after the sixth sputtering is not significant as compared to that in the as-grown sample and the Si-O-Si peak still remains after the seventh sputtering. The Si-N peak becomes well observable after the seventh sputtering in the annealed sample instead of the sixth sputtering for the as-grown case. However, the tendency of decreasing Si-O-Si peak and increasing



bulk Si peak with increasing sputtering time is the same for both as-grown and annealed samples. These results can be understood by considering the increase in SiO₂ thickness by the annealing and the presence of Si-N bonding at the SiO₂/Si interface region. The thickness increase in the annealed SiO₂ sample is considered to be due to the density relaxation of SiO₂ by the thermal annealing [20,21].

Figure 4 shows depth profiles of Si, O, and N atom concentrations in SiO_xN_y films measured by XPS as a function of sputtering time, which reveals that incorporated N atoms (approximately 4%) locate at the film/substrate interface for all the samples. These results are similar to those by the high-temperature process, such as the direct thermal oxynitridation of Si in N₂O ambient at 1,000°C [5]. According to the thermodynamics of Si-N-O system, nitrogen in bulk SiO₂ is not thermodynamically stable but may be stable at the interface [22]; therefore, it has been assumed that nitrogen incorporated into the film during oxynitridation (especially in high-temperature N_2O or NO process) reacts only with Si-Si bonds at or near the interface, not with Si-O bonds in the bulk of the SiO₂ overlayer. Similarly, we suppose that since the dissociation of nitrogen molecules is not significant in the present case, nitrogen migrates to the Si/SiO₂ interface during AP plasma oxidation-nitridation.

Finally, the interface electrical quality of SiO_xN_y layers prepared by AP VHF plasma oxidation-nitridation process has been investigated. Figure 5 shows typical HF *C-V* curves of the MOS capacitors utilizing SiO_xN_y layers formed by various N_2/O_2 flow ratios. The HF *C-V* curve shifts to a negative gate bias direction with increasing N_2/O_2 flow ratios, which shows an increase in





positive Q_f with incorporation of more N atoms into the SiO₂ film (Figure 4). The values of Q_f have been estimated by flat-band voltage shift to be 5.1×10^{11} , 8.1×10^{11} , and 8.4×10^{11} cm⁻² for N₂/O₂ flow ratios of 0.01, 0.1, and 1, respectively.

The HF (blue) and OS (cvan) C-V curves for Al/ SiO_xN_y/Si MOS capacitors before and after FGA are shown in Figures 6 and 7, respectively. The annealed Al/ SiO_xN_y/Si MOS capacitors show better interface properties compared with those without FGA. Dit after FGA were 6.1×10^{11} , 1.2×10^{12} , and 2.3×10^{12} cm⁻² eV⁻¹ for N_2/O_2 flow ratios of 0.01, 0.1, and 1, respectively. It is well known that an introduction of a small amount of nitrogen into the SiO₂ gate oxide leads to an enhanced defect density in the case of N pileup at the Si/SiO₂ interface [23]. From our XPS results, when the N_2/O_2 gas flow ratio increases, the more N atoms pileup at the Si/SiO₂ interface during AP plasma oxidation-nitridation; therefore, D_{it} increases largely with increasing N_2/O_2 flow ratio from 0.01 to 1. The corresponding values of $Q_{\rm f}$ were 1.2×10^{12} , 1.4×10^{12} , and 1.5×10^{12} cm⁻², respectively. It is noted that D_{it} decreases largely with decreasing N₂/O₂ flow ratio from 1 to 0.01, while the decrease of $Q_{\rm f}$ is insignificant. These results suggest that a significantly low N_2/O_2 flow ratio is a key parameter to achieve a small D_{it} and relatively large Q_{fr} which is effective for field-effect passivation of n-type Si surfaces.

Conclusions

SiO_xN_y films with a low nitrogen concentration (approximately 4%) have been prepared on n-type (001) Si wafers at 400°C for 9 min by oxidation-nitridation process in AP plasma using O₂ and N₂ diluted in He gas. Interface properties of SiO_xN_y films have been investigated by *C*-*V* measurements, and it is found that addition of N into the oxide increases both the values of D_{it} and Q_{f} . After FGA, D_{it} at midgap decreases from 2.3 × 10^{12} to 6.1×10^{11} cm⁻² eV⁻¹ with decreasing N₂/O₂ flow ratio from 1 to 0.01, while the decrease of Q_{f} is insignificant from 1.5×10^{12} to 1.2×10^{12} cm⁻². These results suggest that a low N₂/O₂ flow ratio is a key parameter to achieve a low D_{it} and relatively high Q_{f} which is useful to realize an effective field-effect passivation of n-type Si surfaces.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

ZZ helped in the oxidation-nitridation experiments and sample characterization, and wrote the manuscript. YS and YK performed the atmospheric-pressure plasma oxidation-nitridation of Si wafers and XPS, FTIR, and *C-V* measurements. TY, HO, and HK helped in designing the work. KY discussed the results and proofread the manuscript. All authors read and approved the final manuscript.

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