NANO EXPRESS

Open Access

Tuning the electronic transport properties of graphene through functionalisation with fluorine

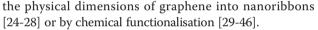
Freddie Withers¹, Saverio Russo¹, Marc Dubois² and Monica F Craciun^{3*}

Abstract

We demonstrate the possibility to tune the electronic transport properties of graphene mono-layers and multilayers by functionalisation with fluorine. For mono-layer samples, with increasing the fluorine content, we observe a transition from electronic transport through Mott variable range hopping (VRH) in two dimensions to Efros-Shklovskii VRH. Multi-layer fluorinated graphene with high concentration of fluorine show two-dimensional Mott VRH transport, whereas CF_{0.28} multi-layer flakes exhibit thermally activated transport through near neighbour hopping. Our experimental findings demonstrate that the ability to control the degree of functionalisation of graphene is instrumental to engineer different electronic properties in graphene materials.

1 Introduction

Graphene, a mono-layer of sp2 bonded carbon atoms arranged in a honeycomb pattern (Figure 1a), is a twodimensional semi-metal where the valence and conduction bands touch in two independent points at the border of the Brillouin zone, named K and K' valleys [1-5]. This material has remarkable electronic, optical and mechanical properties which can be used in a new generation of devices [6,7]. For instance, the high mobility of charge carriers is attracting considerable interest in the realm of high-speed electronics [8]. Furthermore, thanks to the unique combination of high electrical conductivity [4,5] and optical transparency [9], graphene is a promising material for optoelectronic applications such as displays, photovoltaic cells and light-emitting diodes. Few-layer graphene are yet unique materials [10] with unprecedented physical properties: bilayers are semiconductors with a gate-tuneable band gap [11-21], whereas trilayers are semi-metals with a gate-tuneable overlap between the conduction and valence bands [22,23]. However, the use of graphene for applications in daily-life electronics suffers from a major drawback, i.e. the current in graphene cannot be simply pinched off by means of a gate voltage. A valuable solution to this problem is to engineer a band gap in the energy spectrum of graphene for example confining



When chemical elements, e.g. oxygen, hydrogen or fluorine, are adsorbed on the surface of graphene, they form covalent bonds with the carbon atoms. As a result, the planar crystal structure of graphene characterised by sp2 bonds between the carbon atoms is transformed into a three-dimensional structure with sp3 bonds (see Figure 1b). The adsorbed elements can attach to graphene in a random way, as it is the case in graphene oxide [44-46], or they can form ordered patterns as it has been found for hydrogen [33-35] and fluorine [36-43] adsorbates. *Ab initio* calculations performed within the density functional theory formalism predict that functionalisation with hydrogen and fluorine should lead, respectively, to a band gap of 3.8 and 4.2 eV for full functionalisation [29-32].

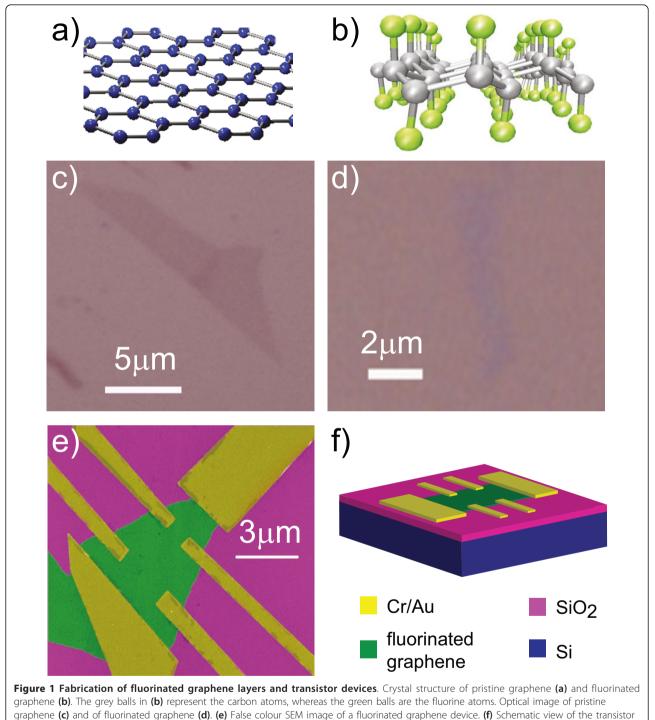
Successful hydrogenation and fluorination of graphene have been recently achieved by several groups [33-43]. Hydrogenation is usually carried out in a remote plasma of H_2 [33-35] which makes it difficult to control the degree of induced atomic defects as well as the stoichiometry of the functionalisation. Furthermore, hydrogenated graphene can loose H at moderate temperatures [33], which limits the use of this material in applications where high-temperature stability is required. On the other hand, fluorine has higher binding energy to carbon and higher desorption energy than hydrogen [29-32]. Opposed to hydrogenation, the process of fluorination is



© 2011 Withers et al; licensee Springer. This is an Open Access article distributed under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/2.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

^{*} Correspondence: m.f.craciun@exeter.ac.uk

³Centre for Graphene Science, College of Engineering, Mathematics and Physical Sciences, University of Exeter, Harrison building, Exeter EX4 4QL, UK Full list of author information is available at the end of the article



structure fabricated on fluorinated graphene.

easy to control, e.g. *via* temperature and reactant gases, leading to reproducibly precise C/F stoichiometries.

Here, we explore the electronic transport properties of functionalised graphene with a fluorine content ranging from 7% (i.e. $CF_{0.07}$ or F/C atomic ratio of 0.07) to 100% (CF₁). We have fabricated transistor structures

with fluorinated graphene mono-layers and multi-layers and studied their electrical transport properties in the temperature range from 4.2 to 300 K. We show that the electronic transport properties of fluorinated graphene can be tuned by adjusting the fluorine content, so that different transport regimes can be accessed, like Mott variable range hopping (VRH) in two dimensions [47,48], Efros-Shklovskii VRH [49] and nearest neighbour hopping (NNH) transport.

2 Experimental details

Fluorinated graphene mono-layers and multi-layers were mechanically exfoliated from fluorinated graphite and deposited onto conventional Si/SiO₂ (275 nm) substrates. The fluorinated graphite was synthesised via two routes: graphite fluorides (using F_2 gas) and fluorine graphite intercalation compounds (FGIC) (using XeF₂ as fluorinating agent), see Section 6. The samples produced using F_2 gas that we investigate here are multi-layers and have the concentration of fluorine of 28 and 100%, whereas the samples synthesised using XeF₂ gas are all mono-layers and have the fluorine content of 7, 24 and 28%.

Flakes of fluorinated graphene are located using an optical microscope (see Figure 1d) and subsequently characterised by Raman spectroscopy. Mono-layer graphene flakes were identified by fitting the 2D peak of the Raman spectra by a single Lorentzian function (see Figure 2b), with a full width at half maximum (FWHM) of 30-45 cm⁻¹ which is typical for pristine mono-layer graphene [50]. The height of the studied multi-layer flakes is determined by Atomic Force Microscopy: 10nm height for flakes exfoliated from the CF and 0.86-6.1 nm for flakes obtained from CF_{0.28}. In total, four monolayer, five CF and five CF_{0:28} multi-layer flakes were processed into four-terminal transistor devices, where the electrical contacts were defined by e-beam lithography, deposition of Cr/Au (5/50 nm) and lift-off procedure, see Figure 1e,f.

The typical optical contrast of fluorinated graphene is \sim 2-6%, which is systematically lower than what we observe on pristine graphene (\sim 9%), see Figure 1c,d. The reduced contrast in fluorinated graphene has to be expected, since the opening of a large energygap in the energy dispersion of fluorinated graphene lowers the optical absorption transitions between conduction and valence bands.

3 Raman spectroscopy

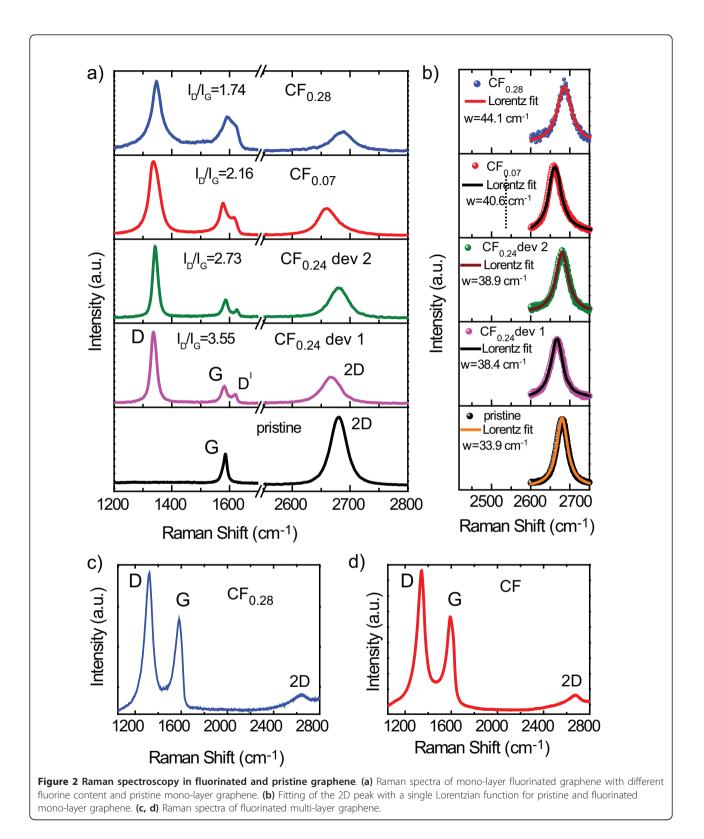
Figure 2 shows the Raman spectra of a mechanically exfoliated pristine graphene flake, with the G and 2D (also known as G') bands at 1580 and 2700 cm⁻¹. The G band is associated with the double degenerate E2g phonon mode at the Brillouin zone center, while the 2D mode originates from a second-order process, involving two intervalley phonons near the K point, without the presence of any kind of disorder or defect [50]. In the fluorinated graphene, additional peaks are activated in the Raman spectra (see Figure 2), the D and D' peaks that appear at 1350 and 1620 cm⁻¹. These Raman peaks

originate from double-resonance processes at the K point in the presence of defects, involving, respectively, intervalley (D) and intravalley (D') phonons [51-54].

In exfoliated pristine graphene, the D peak can only be observed at the edges of the flakes where there is a large concentration of structural defects and its intensity is typically much lower than the intensity of the G peak [55,56]. In our studies performed on pristine graphene flakes with similar size as the fluorinated graphene flakes, the intensity of the D peak is typically well below the sensitivity of our Raman setup, i.e. we are usually not able to detect any D peak because of the edges of the flakes. Therefore, the observed D peak in our fluorinated graphene samples must originate from other defects than simply the edges of the samples. As all our samples contain networks of sp2 bonded carbon atom rings, we believe that the D peak is mainly activated by the F atoms which act as vacancies in these sp2 rings.

A better understanding of the level of disorder in our samples is reached when analysing the intensity ratio I_D/I_G for the D and G bands. It has recently been shown that in graphene I_D/I_G has a non-monotonic dependence on the average distance between defects L_D, increasing with increasing L_D up to $L_D \sim 4$ nm and decreasing for $L_D >$ 4 nm [53,54,57]. Such behavior has been explained by the existence of two disorder-induced regions contributing to the D peak: a structurally disordered region of a radius ~1 nm around the defect and a larger defect-activated region which extends to ~3 nm around the defect. In the defect-activated region, the lattice structure is preserved, but the proximity to a defect causes a mixing of Bloch states near the K and K' valleys. Consequently, the breaking of the selection rules leads to an enhancement of the D peak. Furthermore, it was shown that in the structurally disordered region, the G and D' peaks overlap.

The Raman spectra of fluorinated mono-layer samples produced from graphite with fluorine content of 7 and 28% (see Figure 2a) systematically show that the G and D' peaks have a significant overlap. On the other hand, the samples exfoliated from CF_{0.24} exhibit very distinct G and D' peaks. Based on the aforementioned phenomenological model [53,54,57], we can state that the $CF_{0.07}$ and CF_{0.28} samples are in the regime where the intensity ratio I_D/I_G increases with increasing L_D (i.e. decreasing the concentration of F) whereas the $CF_{0.24}$ samples are in the opposite regime. This scenario is confirmed when comparing the intensity ratio I_D/I_G for fluorinated mono-layers extracted from graphite with different fluorine content: $I_D/I_G = 1.74$ for $CF_{0:28}$ and $I_D/I_G =$ 2.16 for $CF_{0.07}$. From the L_D dependence on I_D/I_G we estimate $L_{\rm D}$ ~ 1.5 nm for $CF_{0.28}$ and $L_{\rm D}$ ~2 nm for $CF_{0.07}$ [53,54,57]. For the $CF_{0.24}$ samples, we estimate L_D \sim 5.3 nm for device 1 and $L_{\rm D}$ \sim 6.1 nm for device 2. These values of L_D are in agreement with the observed



frequencies of the D, G, D' and 2D peaks as well as with the FWHM values of the 2D peaks [53].

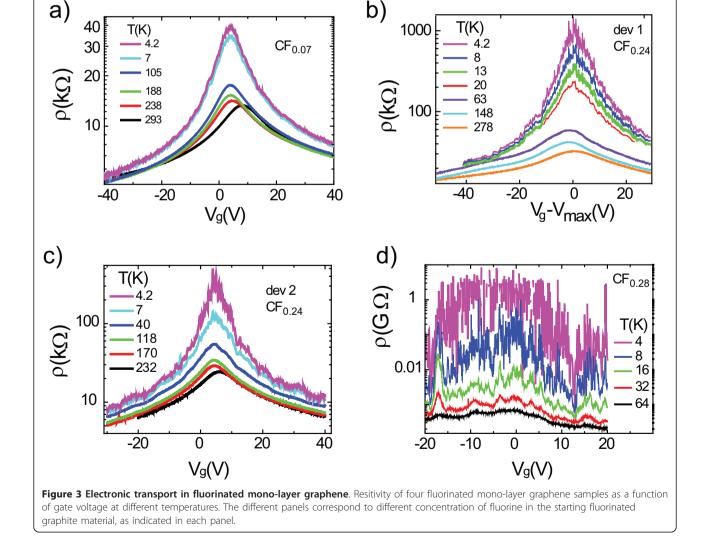
In the case of the fluorinated multi-layer flakes, see Figure 2c,d, it is difficult to perform a similar analysis, as the intensity of the G band depends on the number of graphene layers present in the sample [50]. For samples thicker than three to four layers, the structure of the 2D peak does not provide an accurate estimation for the number of layers because of the large number of fitting parameters.

4 Electrical transport measurements

Having characterised the level of disorder from Raman spectroscopy, we now proceed to address the role of disorder on the electrical transport properties of fluorinated graphene materials. Figure 3 shows the resistivity (ρ) as a function of gate voltage (V_g) for the fluorinated mono-layer samples for different temperatures. The

resistivity exhibits a non-monotonous dependence on $V_{\rm g}$ with a maximum at $V_{\rm g}$ = +10 V, stemming for a doping level of $n = 0.74 \ 10^{12} \ {\rm cm}^{-2}$ commonly seen also in pristine graphene devices and attributed to doping by atmospheric water. In all cases, the resistivity of fluorinated graphene shows a pronounced temperature dependence. Indeed, the maximum of resistivity changes over two orders of magnitude as *T* decreases from 300 to 4.2 K. Away from the maximum of resistivity region, the temperature dependence remains weak, with the mobility of carriers of 150 cm²/Vs. Furthermore, at low temperature the resistance shows strong mesoscopic fluctuations, as expected for samples of small size [58]. In the analysis of the maximum of resistivity, we smooth the $\rho(V_{\rm g})$ curves using a moving average filter.

To examine the presence of the energy gap, we analyse the temperature dependence of the maximum of resistivity by an exponential law describing thermal



activation of carriers across an energy gap $\Delta \varepsilon$: $\rho(T) =$ $\rho_0 exp(\Delta \varepsilon/2k_{\rm B}T)$, see Figure 4a. This analysis clearly shows that our data are not described by the thermal activation law over the whole temperature range. We note that the slope of $ln\rho$ (1/T) versus 1/T decreases with decreasing T, which is a signature of hopping conduction via localised states [48]. The fact that in the whole range of studied temperatures electron transport is not due to thermal activation across the gap but due to hopping becomes clear when re-analysing the temperature dependence in terms of the 2D Mott VRH (2D-VRH) [47,48]. In this model, the functional dependence of ρ on temperature is $\rho(T) = \rho_0 exp(T_0/T)^{1/3}$, where $k_{\rm B}T_0 = 13.6/a^2 g(\varepsilon_F)$, g is the density of localised states at the Fermi level ε_F and a is the localisation length [47,48]. Experimentally we find that the measured $\rho(T)$ for the samples produced from CF_{0.07} and CF_{0.24} graphite (see Figure 3b for $CF_{0.24}$) is described well by the 2D-VRH model.

Figure 4c shows the hopping parameter T_0 as a function of carrier concentration for these three samples. The value of T_0 approaches zero at a carrier concentration of $\pm 1.2 \ 10^{12} \ \text{cm}^{-2}$. This value gives the concentration of the localised electron states in the energy range from $\varepsilon = 0$ to the mobility edge, see Figure 4d. The mobility edge occurs at $V_g \pm 20V$ and indicates the transition from hopping to metallic conduction.

In order to relate the obtained concentration of the localised states to the energy gap $\Delta\varepsilon$, one needs to know the exact energy dependence of the density of states in the gap. For estimations, we will use the linear relation for the density of extended states above the mobility edge $g(\varepsilon) = 2\varepsilon/\pi\hbar^2v^2$ ($v = 10^6$ m/s is the Fermi velocity), and a constant value for the density of localised states below the mobility edge, Figure 4d. This gives $\Delta\varepsilon = 60$ meV and twice this value for the full mobility gap. In this approximation, the density of the localised states in the gap is $\sim 10^{36}$ J/m². Using the obtained value of the hopping parameter at the maximum of resistivity, we can then estimate the localisation length at $\varepsilon = 0$ as a = 40 nm for CF_{0.24} (device 1), a = 81 nm for CF_{0.24} (device 2) and a = 265 nm for CF_{0.07}.

Figure 4e shows the analysis of the temperature dependence of the resistivity for fluorinated mono-layer graphene exfoliated from CF_{0.28} graphite. For this sample, characterised by the largest disorder L_D ~1.5 nm, the experimental data cannot be described by thermally activated law nor Mott VRH. In this case, the *ln*(ρ) follows a $T^{-1/2}$ dependence characteristic of the Efros-Shklovskii VRH in the presence of Coulomb interaction between the localised states ($\rho(T) = \rho_0 exp(T_0/T)^{1/2}$) [49]. T_0 is related to the localisation lengths through $T_0 = 2.8e^2 / 4 \pi \varepsilon_r \varepsilon_0 k_B a$ and for our sample we estimate $T_0 = 52$ K. Assuming that

 ε_r is the dielectric constant of SiO₂ we obtain the localisation length *a* = 282 nm.

We turn now our discussion to multi-layer fluorinated graphene exfoliated from fully fluorinated graphite and from CF_{0.28} prepared by exposure to fluorine gas. The fully fluorinated multi-layer show systematically a very large resistance (more than 100 G Ω) independently of the specific thickness and no gate-voltage control of the resistivity. To achieve gate modulation in these samples, we reduced the fluorine content by annealing the samples at 300°C, in a 10 % atmosphere of H₂/Ar for 2 h. This annealing process restores a partial gate-voltage control of the resistance (Figure 5a) while leaving unchanged the Raman spectrum in Figure 2d.

Resistance measurements of CF_n flakes after annealing show a strong temperature dependence, Figure 5a. Analysis of the temperature dependence of the resistance in terms of the activation law at the highest gate voltage $V_g =$ 50 V (which is still far from the Dirac point) gives a gap of only 25 meV, which is significantly smaller than the expected energy gap for fully fluorinated graphene. Similarly to the fluorinated mono-layer graphene, the resistivity dependence on temperature is fitted well by VRH with the value of $T_0 = 20000$ K. This confirms that the previously found activation energy of 25 meV is not the activation energy $\Delta\varepsilon$ that separates the localised states from extended states at the mobility edge, but is an activation energy $\delta\varepsilon$ of hopping between localised states within the mobility gap, see Figure 4d.

Figure 5c,d shows the transport data for the multi-layer fluorinated graphene $CF_{0.28}$ prepared by exposure to fluorine gas. The I-V characteristics of these samples are strongly non-linear (see Figure 5c) with resistances of more than 1 G Ω . In this case, the dependence of the resistivity on temperature cannot be described neither by 2D-VRH nor by Efros-Shklovskii VRH (see Figure 5d). The $\ln(\rho)(T)$ dependence is rather well described by a thermally activated law at elevated temperatures (regime A), with a $\Delta \varepsilon = 0.25$ eV, followed by a temperature regime where the resistivity decreases with lowering the temperature (regime B). A 1/T dependence of $\ln(\rho)$ could be the consequence of either intrinsic transport through thermally excited carriers above the bandgap or conduction through NNH via localised states within the gap. Since the I-V characteristics of our devices show a non-linearity on a $V_{\rm sd}$ range much larger than the estimated $\Delta \varepsilon = 0.25$ eV (see Figure 5c), we conclude that transport occurs via NNH.

Finally, for the same degree of fluorination (i.e. $CF_{0.28}$) the graphene multi-layers exhibit a stronger temperature dependence and a larger transport gap than what is observed in mono-layers. This difference could originate from the different fluorination processes used for the

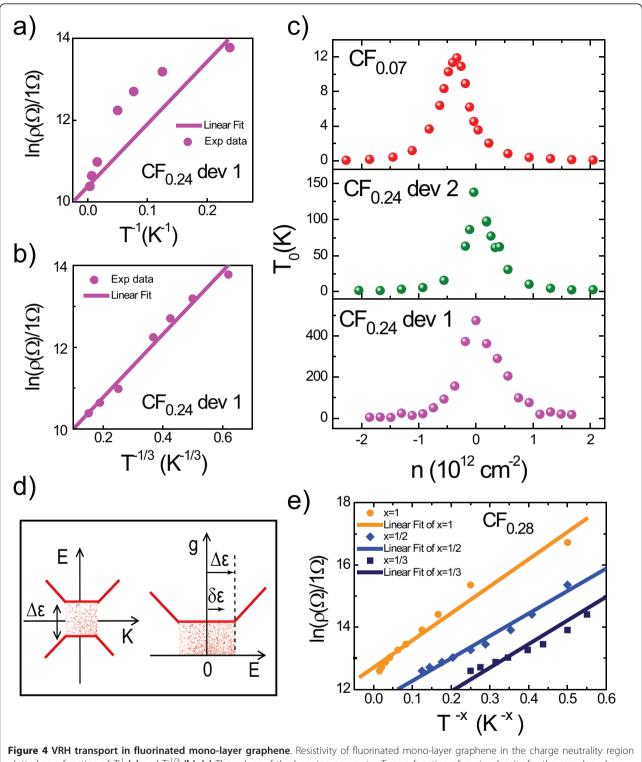
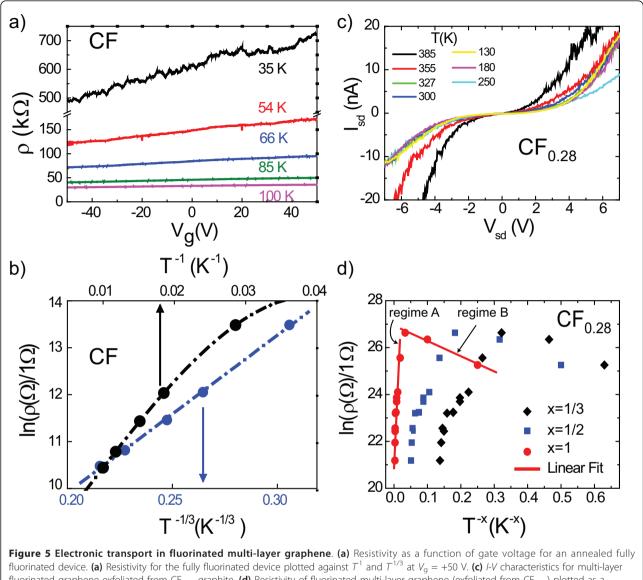


Figure 4 VRH transport in fluorinated mono-layer graphene. Resistivity of fluorinated mono-layer graphene in the charge neutrality region plotted as a function of T^1 (a) and $T^{1/3}$ (b). (c) The values of the hopping parameter T_0 as a function of carrier density for the samples where transport occurs by two-dimensional Mott VRH. (d) Schematic diagrams of the energy dispersion of fluorinated mono-layer graphene (left panel) and of the energy dependence of the density of electron states, with the Fermi level at zero energy (right panel). The localised states are shown by the shaded area. (e) Resistivity of fluorinated mono-layer graphene (exfoliated from CF_{0.28} graphite) in the charge neutrality region plotted as a function of T^x . The solid lines represent fits to the experimental data where x = 1 for thermally activated transport, x = 1/2 for Efros-Shklovskii VRH and x = 1/3 for 2D Mott VRH. The best t is obtained for x = 1/2.



fluorinated device. (a) Resistivity for the fully fluorinated device plotted against T^{-1} and $T^{-1/3}$ at $V_g = +50$ V. (c) *I-V* characteristics for multi-layer fluorinated graphene exfoliated from CF_{0.28} graphite. (d) Resistivity of fluorinated multi-layer graphene (exfoliated from CF_{0.28}) plotted as a function of T^{x} , where x = 1 for thermally activated transport, x = 1/2 for Efros-Shklovskii VRH and x = 1/3 for 2D Mott VRH.

multi-layers (direct fluorination with F_2 gas at high temperature to produce graphite fluorides) and for the monolayers (FGIC synthesised at lower temperatures), see Section 6. Indeed, different fluorination processes may lead to different concentrations of localised states. Even though in both materials, F-GIC and graphite fluorides, the nature of the bonding between fluorine and carbon atoms is covalent, the C-F bond order is slightly lower in F-GIC [59]. The lower C-F bond order in F-GIC is due to the hyper-conjugation which occurs between the C-F and the C-C single bonds around the C-F bonds. In particular, the C-F and Shorter in F-GIC than those in graphite fluorides. As a result, the

electrons involved in the covalent C-F bonds in F-GIC are slightly delocalised by this hyperconjugation, which may result in a smaller transport gap.

5 Conclusions

In conclusion, we have demonstrated the possibility to tune the band structure and therefore the electronic transport properties of graphene through functionalisation with fluorine. In particular, depending on the fluorine concentration different transport regimes can be accessed. For mono-layer samples, we observe a transition from 2D Mott VRH to Efros-Shklovskii VRH with increasing the fluorine content. Multi-layer fluorinated graphene with high concentration of fluorine shows 2D Mott VRH, whereas $CF_{0:28}$ multi-layer flakes exhibit NNH transport. Our experimental findings demonstrate that the ability to control the degree of functionalisation of graphene is instrumental to engineer different electronic properties in graphene materials. In all cases, fluorinated graphene transistors exhibit a large on/off ratio of the current, making this material of interest for future applications in transparent and bendable electronics.

6 Methods

6.1 Fluorination of graphite

To produce fluorinated graphite, we have used two distinct methods [59-62]. In the first method, graphite is heated in the presence of F_2 to temperatures in excess of 300°C, so that covalent C-F bonds are formed and modify the carbon hybridisation [60]. The layered structure of graphite is then transformed into a 3D arrangement of carbon atoms (Figure 1a,b). In this article, we present the studies on graphene exfoliated from fully fluorinated HOPG graphite CF_n (obtained at 600°C) and $CF_{0.28}$ synthesised with this method at 530°C. However, due to the harsh fluorination conditions, many structural defects are formed, which makes it very difficult to exfoliate large enough mono-layer flakes that can be identified by optical microscopy and easily processed into devices. To prepare larger fluorinated graphene samples, we have used a second fluorination method where graphite is exposed to a fluorinating agent, i.e. XeF₂. In this case, the functionalisation process is carried out at $T \leq 120^{\circ}$ C, as XeF₂ easily decomposes on the graphite surface into atomic fluorine [59]. The mixture of natural graphite and XeF_2 was prepared in a glove box in an Ar atmosphere. Owing to its reactivity and diffusion, the fluorination results in a homogenous dispersion of fluorine atoms that become covalently bonded to carbon atoms [59,62,63]. At low fluorine content, the F/C atomic ratio is \leq 0.4. In this case, the conjugated C-C double bonds in the nonfluorinated parts and covalent C-F bonds in corrugated fluorocarbon regions coexist [62,64]. The concentration of the covalent bonds increases with increasing the concentration of fluorine. The samples produced using the XeF₂ gas that we investigate here have the concentration of fluorine of 7, 24 and 28%.

6.2 Determination of fluorine concentration

The fluorine concentration (i.e. F:C molar ratio) of fluorinated graphite was determined by gravimetry (weight uptake). The concentration obtained by weight uptake was confirmed by solid-state NMR measurements on samples fabricated under identical conditions and the accuracy of gravimetry was estimated to be 0.02 [65-67]. The fluorine concentration measured by gravimetry can be under-estimated due to the decomposition of graphite under fluorine gas at high temperature, which results in the formation of carbene (CF₂) and C_2F_4 . However, the decomposition of graphite was found to start as a secondary reaction close to 600°C, with fluorination being the main reaction. Since the reactions with F₂ and XeF₂ have been carried at lower temperatures than the graphite decomposition temperature, the underestimation of F:C ratio in our fluorinated graphite samples is likely to be less than 0.02.

6.3 Raman spectroscopy characterisation

We have characterised all the exfoliated flakes by Raman spectroscopy using an excitation light with a wavelength of 532 nm and a spot size of 1.5 μ m in diameter. An incident power of 5 mW was used. We ensured that this power does not damage the graphene by performing Raman measurements on a similarly sized pristine graphene flake which shows the common spectra of mechanically exfoliated graphene: the G band and 2D band (also known as G') at 1580 and 2700 cm⁻¹, see Figure 2.

6.4 Electrical characterisation

The resistance of the transistor devices was measured both in dc, by means of Keithley 2400 Source-meter, and in ac at low frequency (34 Hz) with a lock-in amplifier in a voltage-biased configuration. For the ac-measurements, the excitation current was varied to ensure that the resulting voltage was smaller than the temperature to prevent heating of the electrons and the occurrence of nonequilibrium effects. The comparison of 2and 4-probe measurements shows that the contact resistance in our devices is negligible as compared to the sample resistance. This experimental finding insures that even 2-probe transport measurements are probing the electrical properties of the bulk fluorinated graphene rather then simply the Cr/graphene interface.

Abbreviations

2D-VRH: two-dimensional Mott variable range hopping; AFM: atomic force microscopy; F-GIC: fluorine graphite intercalation compounds; FWHM: full width at half maximum; NMR: nuclear magnetic resonance; NNH: nearest neighbour hopping; VRH: variable range hopping.

Acknowledgements

We acknowledge T.H. Bointon for performing the AFM measurements on the multi-layer graphene. SR and MFC acknowledge the financial support from the EPSRC (Grant nos. EP/G036101/1 and EP/J000396/1). SR acknowledges the financial support from the Royal Society Research Grant 2010/R2 (Grant no. SH-05052).

Author details

¹Centre for Graphene Science, College of Engineering, Mathematics and Physical Sciences, University of Exeter, Physics building, Exeter EX4 4QF, UK ²Clermont Université, UBP, Laboratoire des Matériaux Inorganiques, CNRS-UMR 6002, 63177 Aubière, France ³Centre for Graphene Science, College of Engineering, Mathematics and Physical Sciences, University of Exeter, Harrison building, Exeter EX4 4QL, UK

Authors' contributions

FW fabricated the fluorinated graphene devices and carried out the measurements. MD synthesised the fluorinated graphite. FW and MFC performed the data analysis. FW, SR and MFC wrote the manuscript. All authors discussed the results, read and approved the final manuscript.

Competing interests

The authors declare that they have no competing interests.

Received: 18 May 2011 Accepted: 12 September 2011 Published: 12 September 2011

References

- 1. Wallace PR: The band theory of graphite. Phys Rev 1947, 71:622-634.
- Castro Neto AH, Guinea F, Peres NMR, Novoselov KS, Geim AK: The electronic properties of graphene. *Rev Mod Phys* 2009, 81:109-162.
- Novoselov KS, Geim AK, Morozov SV, Jiang D, Zhang Y, Dubonos SV, Grigorieva IV, Firsov AA: Electric field effect in atomically thin carbon films. Science 2004, 306:666-669.
- Novoselov KS, Geim AK, Morozov SV, Jiang D, Katsnelson MI, Grigorieva IV, Dubonos SV, Firsov AA: Two-dimensional gas of massless Dirac fermions in graphene. Nature 2005, 438:197-200.
- Zhang YB, Tan YW, Stormer HL, Kim P: Experimental observation of the quantum Hall effect and Berry's phase in graphene. *Nature* 2005, 438:201-204.
- 6. Geim AK, Novoselov KS: The rise of graphene. Nat Mater 2007, 6:183-191.
- 7. Geim AK: Graphene: status and prospects. Science 2009, 324:1530-1534.
- Morozov SV, Novoselov KS, Katsnelson MI, Schedin F, Elias DC, Jaszczak JA, Geim AK: Giant intrinsic carrier mobilities in graphene and its bilayer. *Phys Rev Lett* 2008, 100:016602-016606.
- Nair RR, Blake P, Grigorenko AN, Novoselov KS, Booth TJ, Stauber T, Peres NMR, Geim AK: Fine structure constant defines visual transparency of graphene. *Science* 2008, 320:1308.
- 10. Craciun MF, Russo S, Yamamoto M, Tarucha S: **Tuneable electronic** properties in graphene. *Nano Today* 2011, **6**:42-60.
- 11. Ohta T, Bostowick A, Seyller T, Horn K, Rotenberg E: Controlling the electronic structure of bilayer graphene. *Science* 2006, **313**:951-954.
- Castro EV, Novoselov KS, Morozov SV, Peres NMR, Lopes dos Santos JMB, Nilsson J, Guinea F, Geim AK, Castro Neto AH: Biased bilayer graphene: semiconductor with a gap tunable by the electric field effect. *Phys Rev Lett* 2007, 99:216802-216806.
- Oostinga JB, Heersche HB, Liu X, Morpurgo AF, Vandersypen LMK: Gateinduced insulating state in bilayer graphene devices. *Nat Mater* 2008, 7:151-157.
- Zhang LM, Li ZQ, Basov DN, Fogler MM, Hao Z, Martin MC: Determination of the electronic structure of bilayer graphene from infrared spectroscopy. *Phys Rev B* 2008, 78:235408-235419.
- Zhou SY, Siegel DA, Fedorov AV, Lanzara A: Metal to insulator transition in epitaxial graphene induced by molecular doping. *Phys Rev Lett* 2008, 101:086402-086406.
- Zhang YB, Tang TT, Girit C, Hao Z, Martin MC, Zettl A, Crommie MF, Shen YR, Wang F: Direct observation of a widely tunable bandgap in bilayer graphene. *Nature* 2009, 459:820-823.
- Mak KF, Lui CH, Shan J, Heinz TF: Observation of an electric-field-induced band gap in bilayer graphene by infrared spectroscopy. *Phys Rev Lett* 2009, 102:256405-256409.
- Kuzmenko AB, Crassee I, van der Marel D, Blake P, Novoselov KS: Determination of the gate-tunable band gap and tight-binding parameters in bilayer graphene using infrared spectroscopy. *Phys Rev B* 2009, 80:165406-165418.
- Russo S, Craciun MF, Yamamoto M, Tarucha S, Morpurgo AF: Double-gated graphene-based devices. New J Phys 2009, 11:095018-095029.
- Xia F, Farmer DB, Lin Y, Avouris P: Graphene field-effect transistors with high on/off current ratio and large transport band gap at room temperature. Nano Lett 2010, 10:715-718.
- Zou K, Zhu J: Transport in gapped bilayer graphene: the role of potential fluctuations. *Phys Rev B* 2010, 82:081407-081411.
- Craciun MF, Russo S, Yamamoto M, Oostinga JB, Morpurgo AF, Tarucha S: Trilayer graphene is a semimetal with a gate-tunable band overlap. Nat Nanotechnol 2009, 4:383-388.

- 23. Koshino M, McCann E: Gate-induced interlayer asymmetry in ABA-stacked trilayer graphene. *Phys Rev B* 2009, **79**:125443-125448.
- 24. Son YW, Cohen ML, Louie SG: Energy gaps in graphene nanoribbons. *Phys Rev Lett* 2006, **97**:216803-216807.
- Han MY, Ozyilmaz B, Zhang YB, Kim P: Energy band-gap engineering of graphene nanoribbons. *Phys Rev Lett* 2007, 98:206805-206809.
- Li XL, Wang XR, Zhang L, Lee SW, Dai HJ: Chemically derived, ultrasmooth graphene nanoribbon semiconductors. *Science* 2008, 319:1229-1232.
- 27. Jiao LY, Zhang L, Wang XR, Diankov G, Dai H: Narrow graphene nanoribbons from carbon nanotubes. *Nature* 2009, **458**:877-880.
- Oostinga JB, Sacepe B, Craciun MF, Morpurgo AF: Magnetotransport through graphene nanoribbons. *Phys Rev B* 2010, 81:193408-193412.
- 29. Sofo OJ, Chaudhari AS, Barber GD: Graphane: A two-dimensional hydrocarbon. *Phys Rev B* 2007, **75**:153401-153405.
- Boukhvalov DW, Katsnelson MI: Chemical functionalization of graphene. J Phys Condens Matter 2009, 21:344205-344217.
- Leenaerts O, Peelaers H, Hernandez-Nieves AD, Partoens B, Peeters FM: First-principles investigation of graphene fluoride and graphane. *Phys Rev B* 2010, 82:195436-195442.
- 32. Sahin H, Topsakal M, Ciraci S: Structures of fluorinated graphene and their signatures. *Phys Rev B* 2011, 83:115432-115438.
- Elias DC, Nair RR, Mohiuddin TMG, Morozov SV, Blake P, Halsall MP, Ferrari AC, Boukhvalov DW, Katsnelson MI, Geim AK, Novoselov KS: Control of graphene's properties by reversible hydrogenation: evidence for graphane. *Science* 2009, 323:610-613.
- Han Y, Maultzsch J, Heinz TF, Kim P, Steigerwald ML, Brus LE: Reversible basal plane hydrogenation of graphene. *Nano Lett* 2008, 8:4597-4602.
- Balog R, Jorgensen B, Nilsson L, Andersen M, Rienks E, Bianchi M, Fanetti M, Lagsgaard E, Baraldi A, Lizzit S, Sljivancanin Z, Besenbacher F, Hammer B, Pedersen TG, Hofmann P, Hornekar L: Bandgap opening in graphene induced by patterned hydrogen adsorption. *Nat Mater* 2010, 9:315-319.
- Worsley KA, Ramesh P, Mandal SK, Niyogi S, Itkis ME, Haddon RC: Soluble graphene derived from graphite fluoride. *Chem Phys Lett* 2007, 445:51-56.
- Bon SB, Valentini L, Verdejo R, Garcia Fierro JL, Peponi L, Lopez-Manchado MA, Kenny JM: Plasma fluorination of chemically derived graphene sheets and subsequent modification with butylamine. *Chem Mater* 2009, 21:3433-3438.
- Withers F, Dubois M, Savchenko AK: Electron properties of fluorinated single-layer graphene transistors. *Phys Rev B* 2010, 82:073403-073407.
- Nair RR, Ren W, Jalil R, Riaz I, Kravets VG, Britnell L, Blake P, Schedin F, Mayorov AS, Yuan S, Katsnelson MI, Cheng HM, Strupinski W, Bulusheva LG, Okotrub AV, Grigorieva IV, Grigorenko AN, Novoselov KS, Geim AK: Fluorographene: a two-dimensional counterpart of teflon. Small 2010, 6:2877-2884.
- Robinson JT, Burgess JS, Junkermeier CE, Badescu SC, Reinecke TL, Perkins FK, Zalalutdniov MK, Baldwin JW, Culbertson JC, Sheehan PE, Snow ES: Properties of fluorinated graphene films. *Nano Lett* 2010, 10:3001-3005.
- Cheng SH, Zou K, Okino F, Gutierrez HR, Gupta A, Shen N, Eklund PC, Sofo JO, Zhu J: Reversible fluorination of graphene: evidence of a twodimensional wide bandgap semiconductor. *Phys Rev B* 2010, 81:205435-205440.
- Jeon KJ, Lee Z, Pollak E, Moreschini L, Bostwick A, Park CM, Mendelsberg R, Radmilovic V, Kostecki R, Richardson TJ, Rotenberg E: Fluorographene: a wide bandgap semiconductor with ultraviolet luminescence. ACS Nano 2011, 5:1042-1046.
- Hong X, Cheng SH, Herding C, Zhu J: Colossal negative magnetoresistance in dilute fluorinated graphene. *Phys Rev B* 2011, 83:085410-085415.
- Dikin DA, Stankovich S, Zimney EJ, Piner RD, Dommett GHB, Evmenenko G, Nguyen ST, Ruoff RS: Preparation and characterization of graphene oxide paper. *Nature* 2007, 448:457-460.
- 45. Park S, Ruoff RS: Chemical methods for the production of graphenes. Nat Nanotechnol 2009, 4:217-224.
- Eda G, Chhowalla M: Chemically derived graphene oxide: towards largearea thin-film electronics and optoelectronics. *Adv Mater* 2010, 22:2392-2415.
- 47. Mott NF: Conduction in non-crystalline materials III. Localized states in a pseudogap and near extremities of conduction and valence bands. *Philos Mag* 1969, **19**:835-852.

- Shklovskii Bl, Efros AL: Electronic Properties of Doped Semiconductors Springer Series in Solid State Sciences, vol 45. Berlin: Springer; 1984.
- Efros AL, Shklovskii BI: In *Electron-Electron Interactions in Disordered Systems*. Volume 409. Amsterdam: North-Holland; 1985.
- Ferrari AC, Meyer JC, Scardaci V, Casiraghi C, Lazzeri M, Mauri F, Piscanec S, Jiang D, Novoselov KS, Roth S, Geim AK: Raman spectrum of graphene and graphene layers. *Phys Rev Lett* 2006, 97:187401-187405.
- Dresselhaus MS, Jorio A, Hofmann M, Dresselhaus G, Saito R: Perspectives on carbon nanotubes and graphene raman spectroscopy. *Nano Lett* 2010, 10:751-758.
- Pimenta MA, Dresselhaus G, Dresselhaus MS, Cancado LG, Jorio A, Saito R: Studying disorder in graphite-based systems by Raman spectroscopy. *Phys Chem Chem Phys* 2007, 9:1276-1291.
- Martins Ferreira EH, Moutinho MVO, Stavale F, Lucchese MM, Capaz RB, Achete CA, Jorio A: Evolution of the Raman spectra from single-, few-, and many-layer graphene with increasing disorder. *Phys Rev B* 2010, 82:125429-125438.
- Lucchese MM, Stavale F, Martins Ferreira EH, Vilania C, Moutinho MVO, Capaz RB, Achete CA, Jorio A: Quantifying ion-induced defects and Raman relaxation length in graphene. *Carbon* 2010, 48:1592-1597.
- Casiraghi C, Hartschuh A, Qian H, Piscane S, Georgi C, Fasoli A, Novoselov KS, Basko DM, Ferrari AC: Raman spectroscopy of graphene edges. Nano Lett 2009, 9:1433-1441.
- Malard LM, Pimenta MA, Dresselhaus G, Dresselhaus MS: Raman spectroscopy in graphene. *Phys Rep* 2009, 473:51-87.
- Cancado LG, Jorio A, Martins Ferreira EH, Stavale F, Achete CA, Capaz RB, Moutinho MVO, Lombardo A, Kulmala T, Ferrari AC: Quantifying defects in graphene via Raman spectroscopy at different excitation energies., arXiv:1105.0175.
- Kechedzhi K, Horsell DW, Tikhonenko FV, Savchenko AK, Gorbachev RV, Lerner IV, Falko VI: Quantum transport thermometry for electrons in graphene. *Phys Rev Lett* 2009, 102:066801-066805.
- Sato Y, Itoh K, Hagiwara R, Fukunaga T, Ito Y: On the so-called "semi-ionic" C-F bond character in fluorine-GIC. *Carbon* 2004, 42:3243-3249.
- 60. Nakajima T: Synthesis, structure and physicochemical properties of fluorine-graphite intercalation compounds. *Fluorine-Carbon and Fluoride-Carbon Materials* New York: Marcel Dekker Inc; 1995, 1-33.
- 61. Touhara K, Okino K: Property control of carbon materials by fluorination. *Carbon* 2000, **38**:241-267.
- Zhang W, Spinelle L, Dubois M, Guerin K, Kharbache H, Masin F, Kharitonov AP, Hamwi A, Brunet J, Varenne C, Pauly A, Thomas P, Himmel D, Mansot JL: New synthesis methods for fluorinated carbon nanofibres and applications. J Fluorine Chem 2010, 131:676-683.
- Zhanga W, Guerina K, Dubois M, Fawalb ZE, Ivanovc DA, Vidalc L, Hamwia A: Carbon nanofibres fluorinated using TbF4 as fluorinating agent. Part I: structural properties. *Carbon* 2008, 46:1010-1016.
- Giraudet J, Dubois M, Guerin K, Delabarre C, Hamwi A, Masin F: Solid-state NMR study of the post-fluorination of (C2.5F)n fluorine-GIC. J Phys Chem B 2007, 111:14143-14151.
- Chamssedine F, Dubois M, Guerin K, Giraudet J, Masin F, Ivanov DA, Vidal L, Yazami R, Hamwi A: Reactivity of carbon nanofibers with fluorine gas. *Chem Mater* 2007, 19:161-172.
- Dubois M, Giraudet J, Guerin K, Hamwi A, Fawal Z, Pirotte P, Masin F: EPR and solid-state NMR studies of poly(dicarbon monofluoride) (C2F)n. J Phys Chem B 2006, 110:11800-11808.
- Zhang W, Dubois M, Guerin K, Bonnet P, Kharbache H, Masin F, Kharitonovd AP, Hamwi A: Effect of curvature on C-F bonding in fluorinated carbons: from fullerene and derivatives to graphite. *Phys Chem Chem Phys* 2010, 12:1388-1398.

doi:10.1186/1556-276X-6-526

Cite this article as: Withers *et al.*: **Tuning the electronic transport properties of graphene through functionalisation with fluorine.** *Nanoscale Research Letters* 2011 **6**:526.

Submit your manuscript to a SpringerOpen[™] journal and benefit from:

- Convenient online submission
- ► Rigorous peer review
- Immediate publication on acceptance
- Open access: articles freely available online
- ► High visibility within the field
- Retaining the copyright to your article

Submit your next manuscript at > springeropen.com