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# Construction of a high-performance magnetic enzyme nanosystem for rapid tryptic digestion

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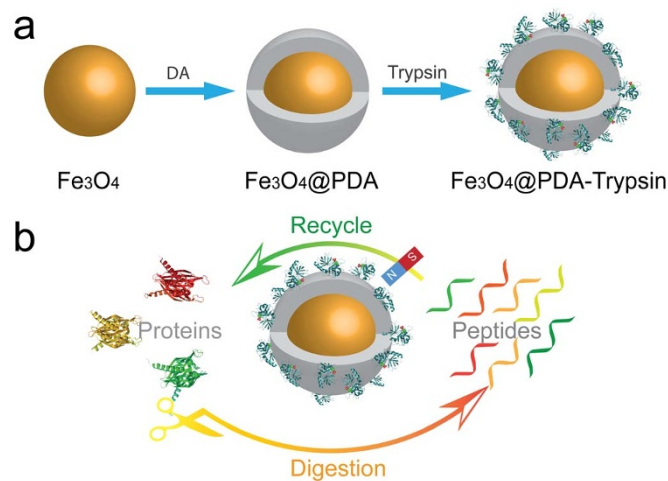
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A magnetic enzyme nanosystem have been designed and constructed by a polydopamine (PDA)-modification strategy. The magnetic enzyme nanosystem has well defined core-shell structure and a relatively high saturation magnetization ( $M_s$ ) value of  $48.3 \text{ emu g}^{-1}$ . The magnetic enzyme system can realize rapid, efficient and reusable tryptic digestion of proteins by taking advantage of its magnetic core and bifunctional shell. Various standard proteins (e.g. cytochrome C (Cyt-C), myoglobin (MYO) and bovine serum albumin (BSA)) have been used to evaluate the effectiveness of the magnetic enzyme nanosystem. The results show that the magnetic enzyme nanosystem can digest the proteins in 30 minutes, and the results are comparable to conventional 12 hours in-solution digestion. Furthermore, the magnetic enzyme nanosystem is also effective in the digestion of low-concentration proteins, even at as low as  $5 \text{ ng } \mu\text{L}^{-1}$  substrate concentration. Importantly, the system can be reused several times, and has excellent stability for storage. Therefore, this work will be highly beneficial for the rapid digestion and identification of proteins in future proteomics.

Along with the rapid development of nanotechnology, functional nanomaterials with desirable components, designed structures, and controlled morphologies have attracted much attention in material science<sup>1-3</sup>. As a type of important functional material, magnetic nanostructures have been widely applied in biomedicine, environmental remediation, energy generation, and storage because of their unique physical and chemical properties (e.g. high surface to volume ratios, adjustable size, size-dependent magnetic properties, chemical stability and variable surface modifications, etc.)<sup>4-6</sup>. Recently, magnetic nanostructures with core-shell architecture have aroused intense interests in various biomedical applications including cancer therapeutics, disease detection and biofunctional carrier, due to their enhanced properties and the synergistic effect between multiple discrete components<sup>7-10</sup>. Specifically, by taking advantage of their unique magnetic properties, the high water dispersion and chemical durability, magnetic core-shell nanoparticles can be rapidly collected and dispersed in aqueous solution, and are extremely favorable for collecting and carrying functional biomolecules (e.g. DNA, protein and peptide)<sup>11-13</sup>, which endow them with specific bioactivities and extend their biomedical applications.

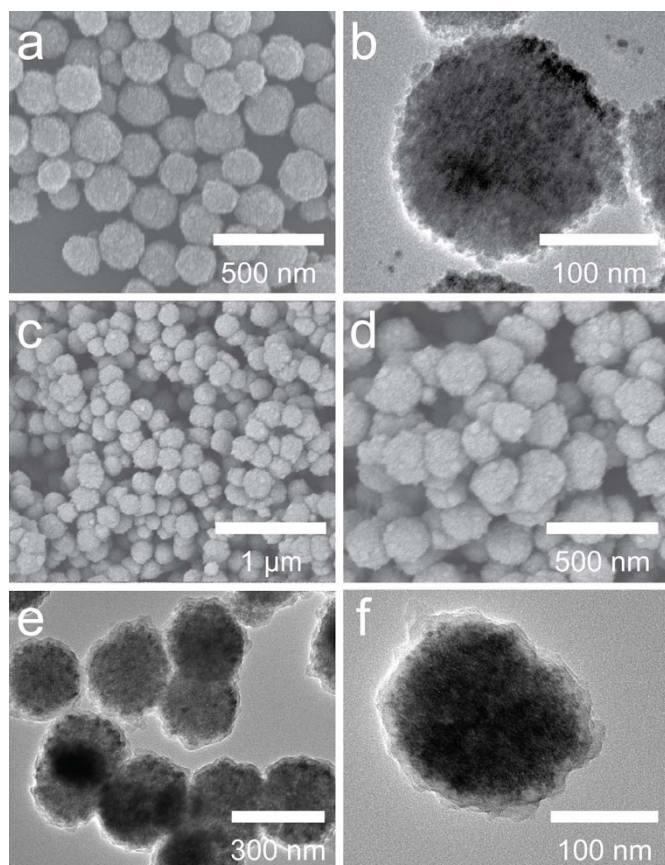
With the advent of the post-genomic era, proteomics has drawn increasing attention in protein interaction studies, biomarker discovery, and disease diagnosis and prevention<sup>14-16</sup>, because it focuses on the study of large scale proteins expression changes in structure and function via a global perspective. Tryptic proteolysis by sequence-specific proteases is a necessary and critical step in proteomics. Conventional in-solution or in-gel digestion by trypsin under certain restricted conditions is the commonly used proteolysis method<sup>17,18</sup>. However, these procedures often suffer from some disadvantages such as the low stability of trypsin due to the experimental condition changes, difficulty in the recovery of trypsin, prolonged digestion time, and limitations for the digestion of low concentration proteins<sup>19-21</sup>. These problems can be partly addressed by the immobilization of enzymes on solid supports, which can enhance enzyme stability, increase enzyme to substrate ratio, improve enzymatic efficiency, and facilitate separation and recovery for reuse<sup>22</sup>.

With the rapid development of nanotechnology, nanomaterials can provide an attractive solution due to their high surface area and designable functions<sup>23</sup>. Various mesoporous materials with nanostructures have been explored to carry enzymes by physical adsorption, and they showed better performance than those using free enzymes alone<sup>24,25</sup>. Although promising, these enzyme-immobilized nanostructures are compromised by enzyme leaching because of the lack of specific interaction between the substrates and the target enzyme molecules, and the centrifugal separation of the enzyme carriers after each step is inconvenient<sup>26,27</sup>. Therefore, the efficiency of the

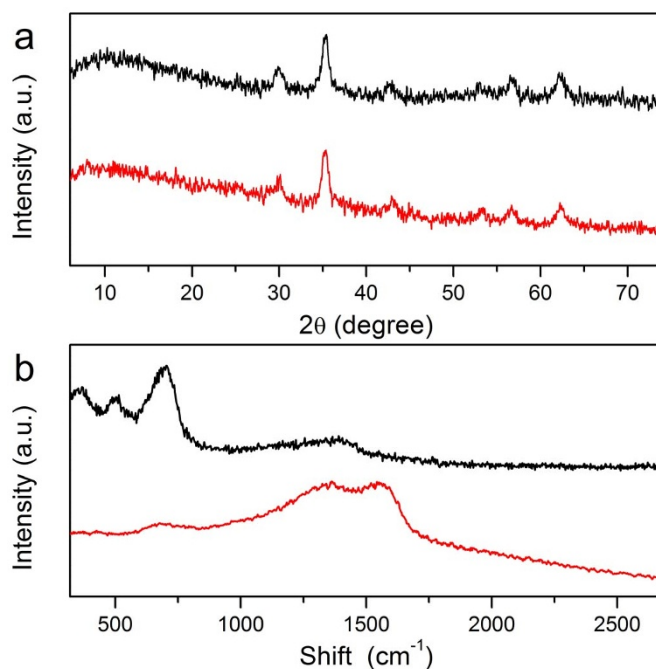


**Figure 1** | Schematic illustration of (a) the synthesis strategy for magnetic nanoreactor and (b) application as a multiplatform in rapid and recyclable digestion of proteins.

nanosystem would be negatively affected, while the recovery of these nanostructures with immobilized enzymes can hardly be achieved. To overcome these shortcomings, the proposed magnetic nanostructure is an ideal candidate by combing the merits of the magnetic property and the catalysis of trypsin. Normally, the chemical linkage based strategy was used to graft trypsin molecules, and the magnetic particles were first modified with silane reagents, followed by the grafting of enzyme molecules under strict conditions<sup>28</sup>. However, the whole preparation process focuses on monolayer modification



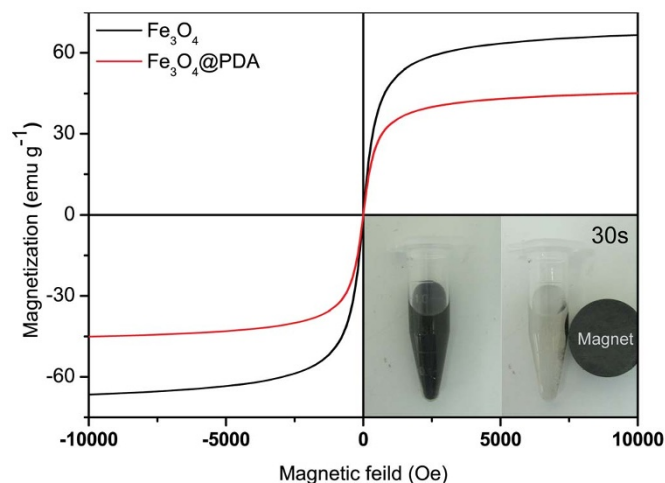
**Figure 2** | SEM (a, c and d) and TEM (b, e and f) images of prepared  $\text{Fe}_3\text{O}_4$  (a and b) and  $\text{Fe}_3\text{O}_4$ @PDA microspheres (c-f).



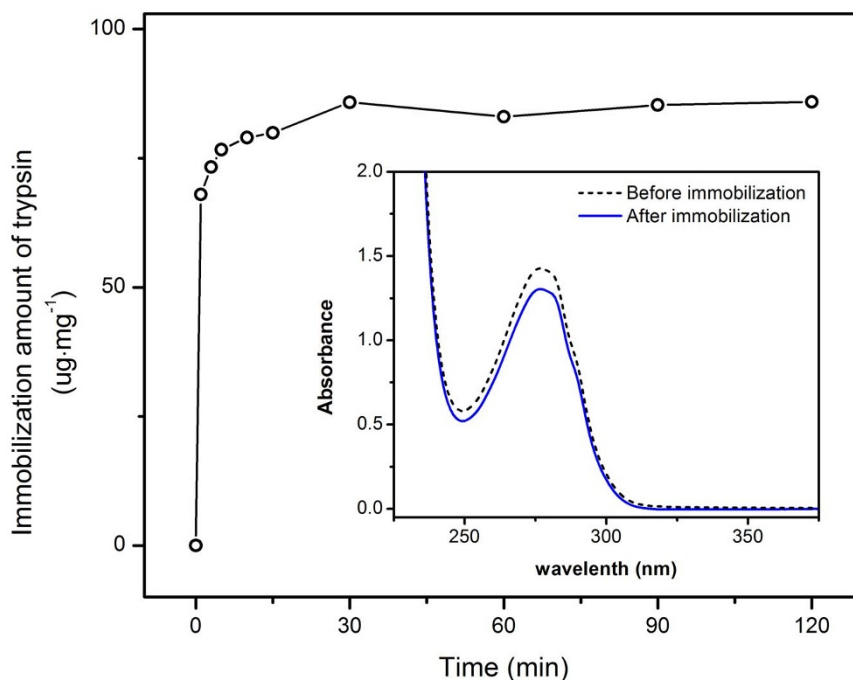
**Figure 3** | XRD patterns (a) and Raman spectra of  $\text{Fe}_3\text{O}_4$  (black) and  $\text{Fe}_3\text{O}_4$ @PDA microspheres (red).

of the particles with small molecules, which would limit the immobilization amount and reduce the accessibility of protein substrates to the immobilized enzyme<sup>24,29</sup>. Although recent works attempted to increase the amount of the binding enzyme by coating a layer of functional polymer, complex, rigorous and longtime modification procedures were required<sup>30</sup>, which might change the conformation of the enzymes and cause a reduction of their catalytic activity. Thus, it is highly desirable to construct an enzyme-immobilized magnetic nanosystem for rapid protein digestion under mild conditions via a simple method.

Inspired by the bio-adhesion properties of marine mussels, poly-dopamine, a novel coating material, has sparked research interest and been moved into the spotlight<sup>31,32</sup>. It can be easily coated on various inorganic and organic substrates with controllable film thickness and high stability<sup>33</sup>. More importantly, the chemical versatility, excellent dispersibility and extraordinary biocompatibility of poly-



**Figure 4** | Magnetic hysteresis loops of the  $\text{Fe}_3\text{O}_4$  and the  $\text{Fe}_3\text{O}_4$ @PDA microspheres at 300 K and rapid separation of the  $\text{Fe}_3\text{O}_4$ @PDA microspheres (inset).



**Figure 5** | The amount of enzymes immobilized onto  $\text{Fe}_3\text{O}_4$ @PDA-trypsin microspheres as a function of time. Inset shows the UV absorption spectra of the supernatant before and after immobilization.

dopamine enable it to be a platform for diverse secondary reactions in biological applications (e.g. biomineralization, cell encapsulation, biomolecule immobilization)<sup>34–36</sup>. Herein, we propose to construct a magnetic enzyme nanosystem (denoted as MEN) via a PDA-modification strategy for the rapid digestion of proteins. As shown in Figure 1a, the  $\text{Fe}_3\text{O}_4$  magnetic particles were coated with a layer of polydopamine after being dispersed in an alkaline solution of dopamine hydrochloride, after which the trypsin molecules can be easily immobilized on the surface of the magnetic particles by simply mixing under mild conditions. The composite nanosystem combines the merits of the easy separation of  $\text{Fe}_3\text{O}_4$  particles and the specific activity of trypsin functionalized polymer shells. They show the following advantages: (1) a polymer shell with a high density of functional groups would ensure the large binding capacity of trypsin molecules; (2) the large exposed surface area allows rapid mass transfer and diffusion of reactants and products; (3) the stability of the magnetic enzyme system under typical proteolytic digestion conditions and fast magnetic separation enables the regeneration and reuse of the nanosystem.

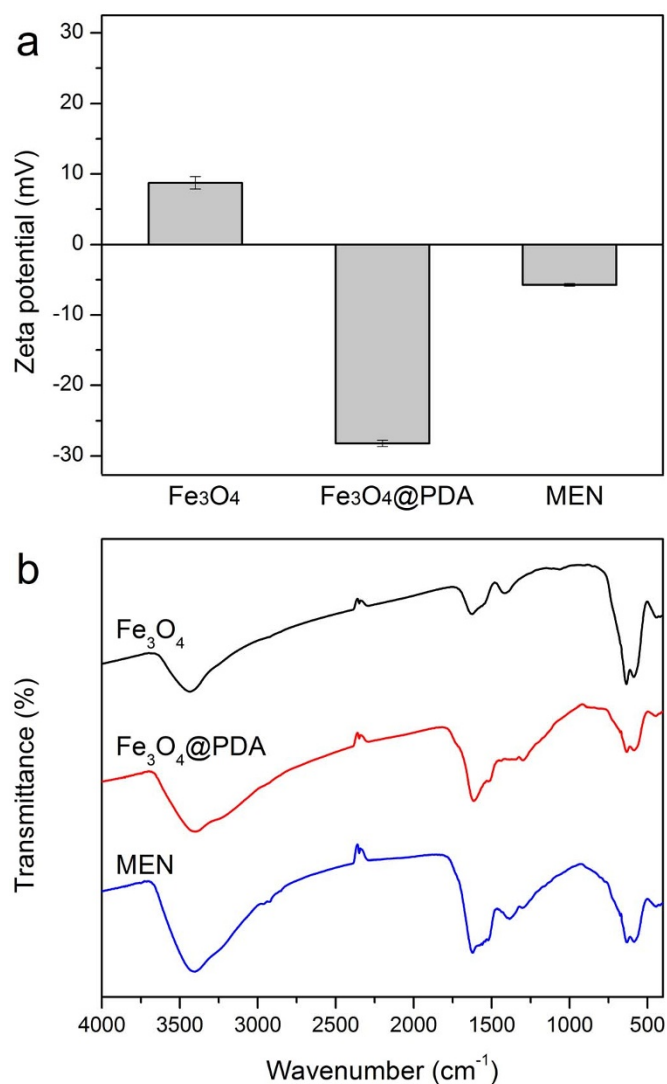
## Results

Figure 1a illustrates the construction strategy of the MEN. Well-dispersed  $\text{Fe}_3\text{O}_4$  microspheres were synthesized by a previously reported solvothermal method. Then, the prepared magnetic particles were coated with a highly uniform polymeric shell by self-assembly of dopamine (DA) molecules in alkaline aqueous solution using a biomimetic adhesive method. In virtue of the excellent water-dispersibility and the presence of many catechol hydroxyl-groups on the PDA layer, the trypsin molecules can be easily grafted onto the polydopamine coated magnetic microspheres (denoted as  $\text{Fe}_3\text{O}_4$ @PDA) by taking advantage of the unique adhesive properties of PDA, thereby forming the multifunctional MEN<sup>31</sup>.

Figure 2a shows the scanning electron microscopy (SEM) image of the prepared  $\text{Fe}_3\text{O}_4$  particles. Apparently, the well-dispersed  $\text{Fe}_3\text{O}_4$  particles have a relatively coarse surface and regular spherical shape with an average size of 180 nm. Transmission electron microscopy (TEM) images (Figure 2b) further reveal that the  $\text{Fe}_3\text{O}_4$  microspheres

consist of many reunited nanoparticles. As shown in Figure 2c and d, after further coating of PDA shells, the  $\text{Fe}_3\text{O}_4$ @PDA microspheres are relatively dispersed, but are slightly larger in diameter and present different surfaces in comparison to the precursor  $\text{Fe}_3\text{O}_4$  particles. The core-shell structure of the  $\text{Fe}_3\text{O}_4$ @PDA can be clearly observed in the TEM image (Figure 2e and f) and the thickness of the PDA shell is about 20 nm, confirming the formation of the uniform PDA shell. It should be noted that the coating time plays an important role in the thickness of PDA shell. As shown in Fig. S1, with the extension of the coating time, the thickness of polydopamine would increase; however, the microspheres also slightly aggregate. Therefore, magnetic microspheres with thin PDA layer are favorable due to well dispersibility and shorter reaction time without sacrificing the surface functionalization capability of the PDA layer. Thus thin PDA layer was used in the following application.

The crystal structure of the magnetic composite microspheres was investigated by powder X-ray diffraction (XRD), as shown in Figure 3a. The XRD pattern of the  $\text{Fe}_3\text{O}_4$  microspheres indicates that all diffraction peaks in the XRD pattern of  $\text{Fe}_3\text{O}_4$  microspheres are consistent with the magnetite (JCPDS card No. 19-0629). Notably, the XRD pattern of the  $\text{Fe}_3\text{O}_4$ @PDA microspheres has similar diffraction peaks as those of precursor  $\text{Fe}_3\text{O}_4$  microspheres, indicating that the polymer shell is amorphous in nature and the main magnetite phase of magnetic cores was not destroyed during the fabrication procedure. To evidently confirm the formation of PDA shell, Raman spectroscopy was used in investigating the changes of the interface of magnetic composite microspheres. As shown in Figure 3b, three strong and broad bands around 370, 489, and 681  $\text{cm}^{-1}$  can be observed in the Raman spectrum of bare magnetic cores, which correspond to the T<sub>2g</sub> and A<sub>1g</sub> modes of symmetry<sup>37,38</sup>. However, after coating with the PDA polymer shell, the intensity of these characteristic bands decreased, while two new broad bands around 1355  $\text{cm}^{-1}$  and 1577  $\text{cm}^{-1}$  are present in the Raman spectrum. They are characteristic bands of polydopamine, which can be attributed to the deformation of the catechol group<sup>39</sup>. It should be noted that catechol groups serve as active sites for the grafting of targets, and therefore the high abundance of residual catechol groups



**Figure 6** | Zeta potentials (a) and FTIR spectra (b) of Fe<sub>3</sub>O<sub>4</sub>, Fe<sub>3</sub>O<sub>4</sub>@PDA and Fe<sub>3</sub>O<sub>4</sub>@PDA-trypsin microspheres.

would contribute to the highly efficient immobilization of trypsin molecules.

The magnetic properties of MEN microspheres are beneficial for their convenient and fast separation, removing the need for repeated centrifugation in practical applications. As shown in Figure 4, the hysteresis loops of the MEN microspheres and the precursor Fe<sub>3</sub>O<sub>4</sub> microspheres were recorded by a superconducting quantum interface device (SQUID) magnetometer. Apparently, both the Fe<sub>3</sub>O<sub>4</sub> and Fe<sub>3</sub>O<sub>4</sub>@PDA microspheres have strong magnetism at room temperature. The saturation magnetization ( $M_s$ ) values of the Fe<sub>3</sub>O<sub>4</sub> and Fe<sub>3</sub>O<sub>4</sub>@PDA microspheres are 66.6 emu g<sup>-1</sup> and 45.0 emu g<sup>-1</sup>, respectively. Because of the introduction of nonmagnetic species, the saturation magnetization ( $M_s$ ) values of Fe<sub>3</sub>O<sub>4</sub>@PDA microspheres are smaller than that of Fe<sub>3</sub>O<sub>4</sub> microspheres; however, they can still be rapidly separated by a magnet. As shown in the Figure 5 inset, the MEN microspheres can be easily dispersed in aqueous solution without an external magnetic field, and they can then be rapidly collected from the mixture within 30 s by a magnet. These results demonstrate that the MEN microspheres simplify the separation process in practical applications, because of their excellent magnetic response.

The construction of a magnetic enzymatic nanosystem was easily conducted at room temperature, and the amount of immobilized

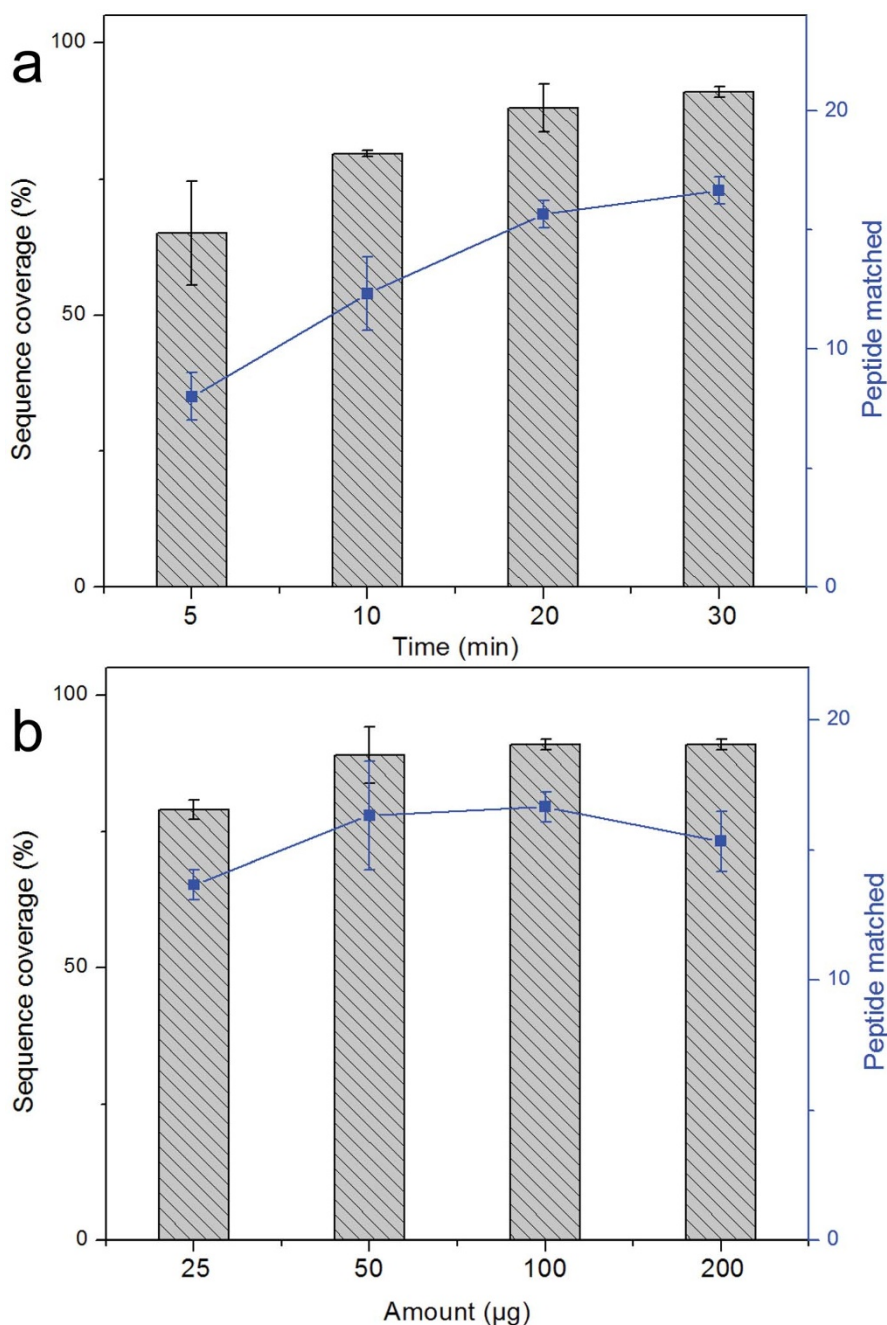
trypsin was monitored by an UV/Vis spectrophotometer. Figure 5 shows the time-dependent curve of trypsin immobilization. It is apparent that the maximum immobilization of trypsin can be realized in less than 30 minutes. This could be ascribed to the large exposed surface area of Fe<sub>3</sub>O<sub>4</sub>@PDA composite microspheres, which may provide more chances to interact with large guest molecules. The immobilization capacity of trypsin was calculated to be 85.9 μg mg<sup>-1</sup> (enzyme/carrier).

The successful grafting of trypsin molecules was confirmed by the zeta-potential analysis of the resulting composite microspheres. As shown in Figure 6a, the naked magnetic cores show positive values (8.74 ± 0.87 mV), while the Fe<sub>3</sub>O<sub>4</sub>@PDA composite microspheres present a negative zeta-potential of -28.23 ± 0.47 mV because of the deprotonation of the phenolic group on the polydopamine shells<sup>40</sup>. However, after immobilization of trypsin, the zeta potential values of the microspheres increased to -5.73 ± 0.20 mV because of the introduction of positive trypsin molecules<sup>41</sup>. In addition, FT-IR spectra were also recorded to study the transformation of surface properties and the evidence of the successful immobilization of trypsin. As shown in Figure 6b, the strong absorption peak at around 590 cm<sup>-1</sup> in FT-IR of Fe<sub>3</sub>O<sub>4</sub> microspheres can be assigned to the stretching vibration of Fe-O from the Fe<sub>3</sub>O<sub>4</sub>. For the Fe<sub>3</sub>O<sub>4</sub>@PDA composite microspheres, new strong and broad absorption bands can be observed in the range of 1800–1000 cm<sup>-1</sup>, which can be attributed to the stretching vibration of the aromatic rings and the C-O stretching of phenol compounds and further confirms the formation of the PDA shell. After the grafting of the trypsin, besides the characteristic adsorption peaks of the Fe<sub>3</sub>O<sub>4</sub> core and the PDA shell, several characteristic absorption peaks of trypsin with weak intensities were also observed in the range of 1000–1800 cm<sup>-1</sup> and 2750–3000 cm<sup>-1</sup> in the FTIR of MEN microspheres (Figure S2), compared to the Fe<sub>3</sub>O<sub>4</sub>@PDA microspheres. The above results demonstrate the successful construction of the MEN system.

As shown in Figure 1b, the MEN can be used as a digestion platform for rapid protein digestion as a result of its unique enzymatic activity; and the MEN microspheres can be easily separated and collected from the digestion solution for recycling by taking advantage of their magnetic properties. The MEN microspheres were mixed with the protein solution. Then, with the temperature at 37°C for 30 min, the proteins were digested. Notably, all the trypsin molecules were presented on the large exposed surface of the magnetic composite microspheres, and therefore the target proteins could easily interact with them. After digestion, the resulting peptides could diffuse rapidly out of the surface so as to expose the active sites for interaction with undigested proteins, which would promote catalytic efficiency. More importantly, the MEN microspheres can be conveniently isolated from the mixture by a magnet, while leaving the protein digest ready for detection. Being washed with buffer several times can regenerate the MEN, at which point it is ready for reuse.

The effectiveness of the constructed MEN system was evaluated by digesting the model protein cytochrome c under heating at 37°C. As the digestion time and the amount of trypsin are critical for protein digestion, herein, the digestion time and the amount number of MEN microspheres were optimized. Figure 7a presents the identification results of Cyt-c digested by the MEN system under different heating times. The proteins can be effectively digested and identified with relatively high sequence coverage (>70%) even though the proteins were only treated with the MEN system for 5 min. Further increasing the digestion time would lead to an increased number of matched peptide and sequence coverage, and excellent performance of protein digestion and identification can be realized in half an hour.

The amount of MEN microspheres can also affect the performance of protein digestion. As shown in Figure 7b, the MEN digestion system was highly effective, as the Cyt-c could be digested by 25 μg MEN microspheres, and the protein could be identified with sequence coverage of 79% and 14 matched peptides. Better perform-

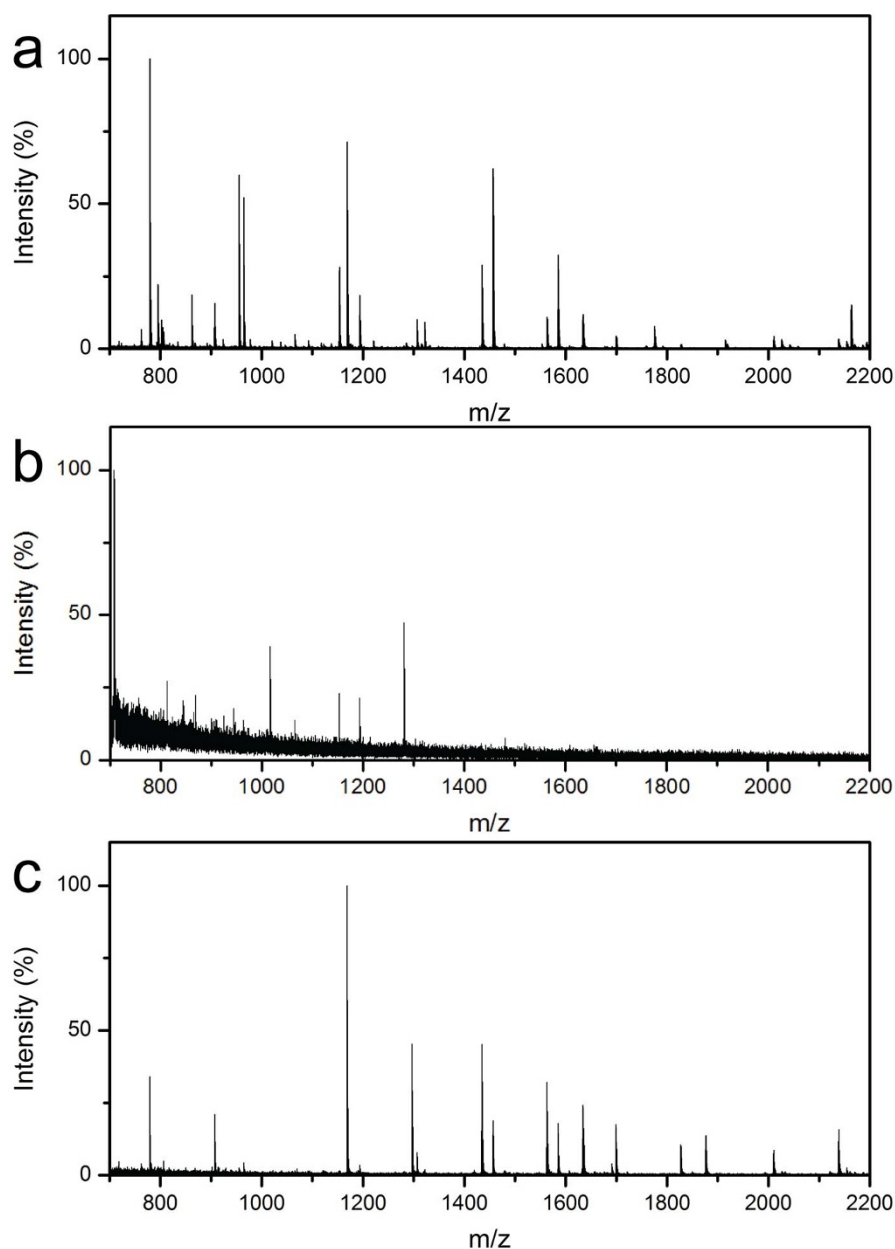


**Figure 7** | (a) Effect of digestion time on the sequence coverage (bar diagram) and the number of identified peptides (curve) of Cyt-c (50 ng  $\mu\text{L}^{-1}$ , 50  $\mu\text{L}$ ); (b) Effect of the amount of MEN microspheres for digestion.

ance could be further realized via increasing the amount of MEN microspheres, and the highest number of matched peptides and amount of sequence coverage was observed when 100  $\mu\text{g}$  MEN microspheres were used. However, the number of matched peptides decreased slightly while sequence coverage was still high when the amount of MEN microspheres was increased to 200  $\mu\text{g}$ . This could be ascribed to some peptide fragments being absorbed by the redundant microspheres. According to the above results, the optimal time and enzymatic microsphere quantity are 30 min and 100  $\mu\text{g}$  respectively and should be used for the MEN digestion system.

Figure 8a shows the MALDI-TOF mass spectrum of Cyt-c digested by the MEN digestion system. Apparently, the target proteins were well digested and detected with a high intensity and signal to noise (S/N) ratio, and 17 peptides can be matched and identified

with a sequence coverage of 92%. Table S1 lists the detailed information for the identified peptides. For comparison, conventional in-solution digestions of 30 minutes and 12 hours were also conducted. As shown in Figure 8b, for the short time (30 minutes) in-solution digestion, only a few peptides from Cyt-c could be detected and the S/N ratio and intensity were very low, leading to the poor quality of the MS. As expected, mass spectra with enhanced S/N ratios and intensity can be obtained (Figure 8c) when the proteins were digested for a long-time (12 hours) in-solution. 15 target peptides with sequence coverage of 83% can be identified, which were comparable to but slightly lower than those obtained using the MEN system. Reduction of protein digestion time from 12 hours to 30 minutes by the MEN system can significantly improve the throughput and turnaround time of proteomic analysis. However, the digestion time was



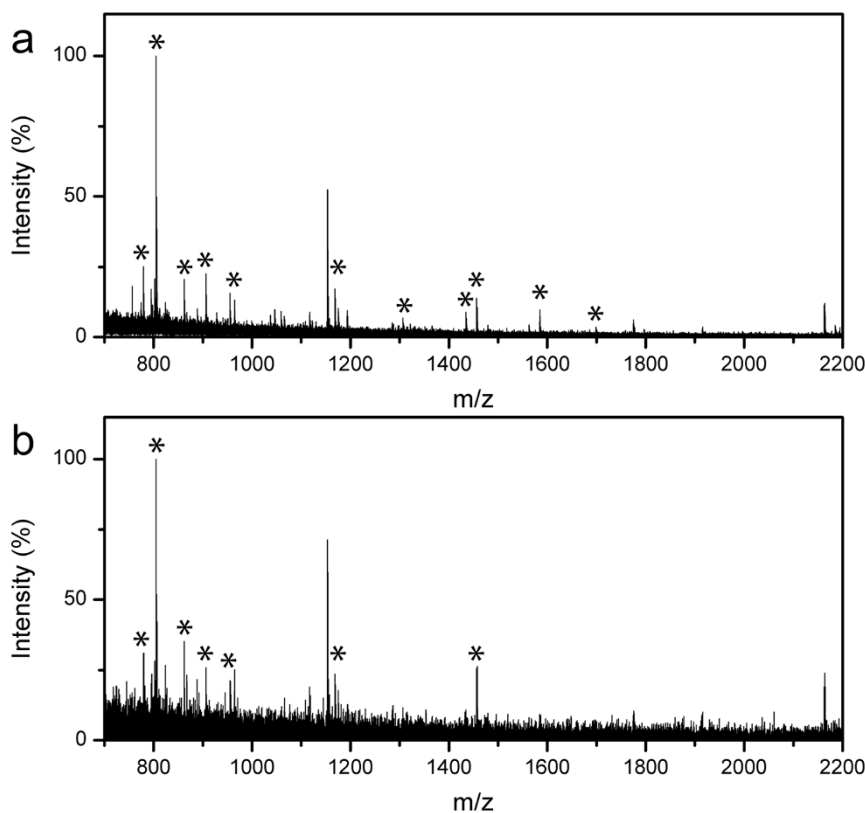
**Figure 8** | Mass spectra of Cyt-c ( $50 \text{ ng } \mu\text{L}^{-1}$ ,  $50 \mu\text{L}$ ) digested by the MEN system for 30 minutes (a), and the in-solution digestion for 30 minutes (b) and 12 hours (c).

extended to 12 hours when using the in-solution technique, and it was difficult to separate or recover the free trypsin molecules in the mixture solution afterwards.

Compared with the high-abundance proteins, the concentrations of most protein biomarkers are extremely low, and rapid digestion and detection of low concentration protein samples is critical and highly desirable in practical proteomics research and biomedical diagnosis. However, quite notoriously, in-solution digestion of small quantity proteins is often compromised by the low concentration, small sample volume and low enzyme/protein ratio. Figure S3 shows the MS of low concentration ( $10 \text{ ng } \mu\text{L}^{-1}$  and  $5 \text{ ng } \mu\text{L}^{-1}$ , respectively) Cyt-c digested using conventional in-solution digestion. It is apparent that only a few peaks with low S/N ratio and intensity were detected, while it is hard to identify the target protein. However, as shown in Figure 9a, when it was digested using the MEN system, peptides (marked with “\*”) with sequence coverage of 60% could be detected and identified for the target protein by the MS.

Furthermore, even when the protein concentration was lowered to  $5 \text{ ng } \mu\text{g}^{-1}$ , 6 peptides (marked with “\*”) from Cyt-c could still be detected after digestion using the MEN system (Figure 9b). This could be attributed to high local concentrations of substrate proteins near immobilized trypsin on the surface of MEN microspheres, which was absent from the in-solution digestion system.

Magnetic microspheres are characterized by easy separation and collection by an external magnetic field, and therefore the MEN microspheres can be conveniently collected and regenerated. Mass spectra and identification results of digested proteins were used to evaluate the reusability of the MEN. Figure 10a shows the results of protein digestion using the recycled MEN microspheres. The target proteins could be effectively identified even when the MEN microspheres were reused 5 times, and 15 peptides with a sequence coverage of 81% were matched on the fifth trial. Figure 10b and c represent the typical MS obtained from the third and fifth recycling digestion. Apparently, high quality mass spectra were observed, and



**Figure 9** | Mass spectra of low concentration Cyt-c digested MEN: (a)  $10 \text{ ng } \mu\text{g}^{-1}$ , (b)  $5 \text{ ng } \mu\text{g}^{-1}$ .

most of the target peptides were detected with strong intensities. These results demonstrate that the MEN microspheres have excellent recyclability.

Normally, the trypsin solution is very stable in the refrigerator. The digestion result using trypsin solution stored for 25 days is very similar to that obtained using the freshly prepared trypsin solution (Figure S4). In order to investigate their stability, MEN microspheres were stored in a refrigerator for different time intervals and then used for protein digestion. The digests were submitted for MS detection, and the number of matched peptides and sequence coverages for protein identification were summarized in Figure 11. Interestingly, even with the prolongation of storage time, the digestion and detection results were not greatly deteriorated. Even for the MEN microspheres stored for 50 days, 16 peptides with sequence coverage of 87% can still be observed, indicating the high proteolytic activity of the MEN microspheres after long storage time. Therefore, it can be safely concluded that the MEN microspheres have excellent stability.

To confirm the universality of the MEN system for digestion of various proteins, bovine album serum (BSA) and myoglobin (MYO) were also digested using the MEN microspheres. As shown in Figure 12a, many peptides with S/N ratios and high intensities can be observed in the MS of BSA. After data search by the mascot, 36 peptides with sequence coverage 55% could be matched for the identification of the BSA protein. High quality MS could also be observed for the MYO proteins after MEN digestion and detection (Figure 12b), and 14 peptides are matched while the sequence coverage is high at 83%. Table S2 lists the detailed information regarding the identified peptides. For comparison, in-solution digestion was also conducted (Figure S5). 35 matched peptides with sequence coverage of 42% can be identified for BSA, and 15 matched peptides with sequence coverage of 77% can be identified for MYO. Apparently, the results obtained using the 30 min MEN digestion approach is quite comparable to and in general better than the results obtained using the 12-hour in-solution method. The above results further

demonstrate that the MEN system can be used for rapid digestion of proteins.

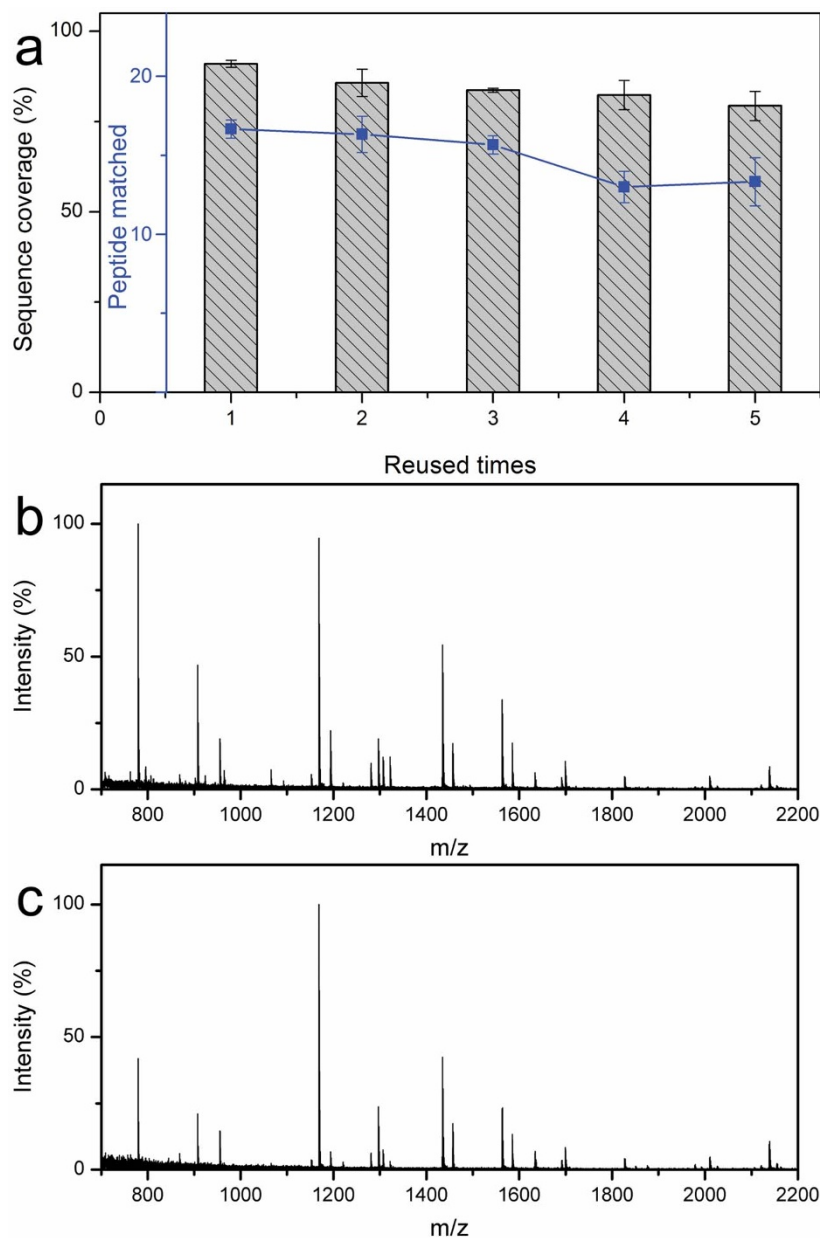
## Discussion

A novel magnetic enzymatic nanosystem was constructed via a simple mussel-inspired polydopamine linking method. The MEN system can realize the rapid digestion of proteins, convenient magnetic separation and MEN recycling, thanks to the exposed tryptic surface, the  $\text{Fe}_3\text{O}_4$  core, and the stable PDA linker. Various proteins were used to evaluate the performance of the MEN system. The proteins can be digested in 30 minutes, and the system is also effective for the digestion of extremely low-concentration proteins. Furthermore, the MEN system can be recycled several times with high digestion efficiency being maintained. In addition, the MEN microspheres are very stable in refrigerator storage, as the digestion efficiency is almost unchanged even after storage for 50 days. Therefore, the MEN system has high potential for rapid and high throughput protein digestion in proteomics.

## Methods

**Chemicals.** Ferric chloride ( $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ ), ethylene glycol (EG), trisodium citrate ( $\text{H}_3\text{Cit}$ ), tris (hydroxymethyl) aminomethane hydrochloride (Tris-HCl), ethanol (EtOH), sodium acetate (NaAc), ammonium bicarbonate ( $\text{NH}_4\text{HCO}_3$ ), acetonitrile (ACN), trifluoroacetic acid (TFA) and ammonium hydroxide ( $\text{NH}_3 \cdot \text{H}_2\text{O}$ ) were obtained from Alfa Aesar. Dopamine (DA), 2, 5-dihydroxybenzoic acid (2, 5-DHB), myoglobin from equine heart (MYO), cytochrome C (Cty-c), bovine serum albumin (BSA), and trypsin (from bovine pancreas, TPCK treated) were purchased from Sigma-Aldrich (St. Louis, MO, USA).

**Preparation of  $\text{Fe}_3\text{O}_4$  and  $\text{Fe}_3\text{O}_4$ @PDA microspheres.** The magnetic  $\text{Fe}_3\text{O}_4$  particles were prepared according to a reported solvothermal approach with minor modification<sup>42</sup>. Typically,  $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$  (0.81 g) and trisodium citrate (0.20 g) were first dissolved in ethylene glycol (20 mL). Then, NaAc (1.20 g) was added, and the mixture was vigorously stirred to form a transparent solution. Afterward, the solution was transferred to a 50 mL Teflon-lined stainless-steel autoclave. The autoclave was sealed and heated to  $200^\circ\text{C}$  and maintained for 8 hours, and then allowed to cool

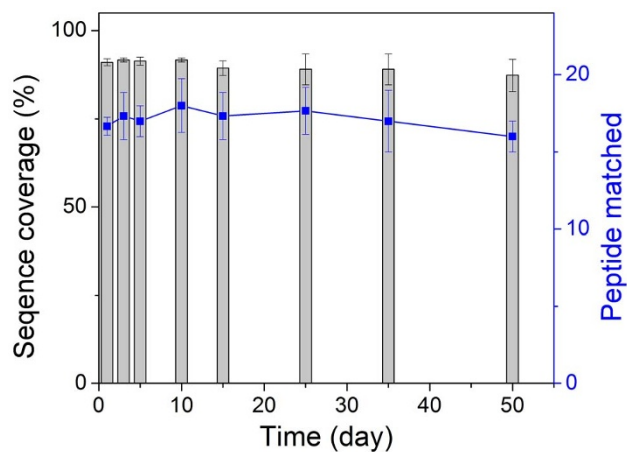


**Figure 10** | The identification results of the Cyt-c digested using the recycled MEN microspheres (a). Mass spectra of Cyt-c digested by the recycled MEN microspheres being used for the third time (b) and the fifth time (c).

down to room temperature. The products were washed with ethanol and deionized water several times and dried at 60 °C for 12 hours.

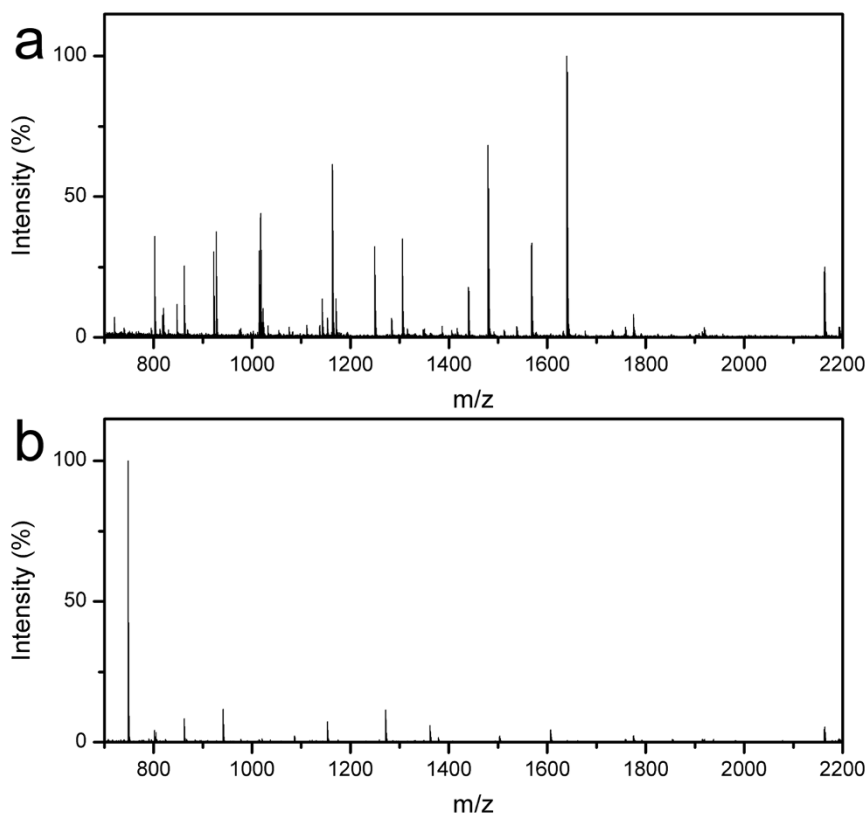
For the preparation of Fe<sub>3</sub>O<sub>4</sub>@PDA core-shell microspheres, the as-synthesized Fe<sub>3</sub>O<sub>4</sub> particles (25 mg) were fully dispersed in 25 mL of 20 mM tris-HCl (pH=8.0) by ultrasonication for 30 min. Dopamine (50 mg) was dissolved into 25 mL deionized water. The dopamine solution was quickly injected into the Fe<sub>3</sub>O<sub>4</sub> dispersion under continuous magnetic stirring at 30 °C for 2 hours. After that, the products were collected and separated with the help of a magnet, and then washed several times with deionized water.

**Construction of magnetic enzymatic nanosystem (MEN).** Fe<sub>3</sub>O<sub>4</sub>@PDA microspheres (10 mg) were added into 2.5 mL of 25 mM NH<sub>4</sub>HCO<sub>3</sub> buffer to form a homogeneous suspension by ultrasonication for 2 hours, while trypsin (5 mg) was dissolved in another 2.5 mL of 25 mM NH<sub>4</sub>HCO<sub>3</sub> buffer. Then, the particle suspension was added into the trypsin solution, and the mixture was shaken for varying times. Subsequently, the microspheres loaded with trypsin were collected and isolated from the mixture by magnetic sorting, and then the microspheres were washed three times with the NH<sub>4</sub>HCO<sub>3</sub> buffer to remove any weakly bound trypsin. The amount of trypsin immobilized was measured using an UV/visible absorption spectrophotometer at 280 nm for comparing the protein concentration before and after immobilization. The obtained microspheres were redispersed into NH<sub>4</sub>HCO<sub>3</sub> buffer (10 mg mL<sup>-1</sup>) via vibration to form a homogeneous suspension.



**Figure 11** | The identification results of the Cyt-c digested using the stored MEN microspheres with different time.





**Figure 12** | Mass spectra of BSA (a) and MYO (b) digested by the MEN system.

**Rapid digestion of proteins using MEN.** Cytochrome C (Cyt-c), myoglobin (MYO) and bovine serum albumin (BSA) were used to evaluate the efficiency of protein digestion by MEN. The proteins were first dissolved into  $\text{NH}_4\text{HCO}_3$  buffer (25 mM) to form protein solutions of different concentrations. 100  $\mu\text{g}$  of the MEN microspheres suspension was added into 50  $\mu\text{L}$  of the protein solution, and the mixture was then shaken at  $37^\circ\text{C}$  for 30 minutes. Subsequently, the supernatant was isolated and collected by magnetic separation for the following detection process. For regeneration, the collected MEN microspheres were washed with the  $\text{NH}_4\text{HCO}_3$  buffer three times. The in-solution digestion with free trypsin was also conducted for comparison. Trypsin was then added into the protein solution with an optimized enzyme/substrate ratio of 1:50 (w/w), and the solution was incubated at  $37^\circ\text{C}$  for 12 hours.

**MALDI-TOF MS analysis.** 1  $\mu\text{L}$  of proteolytic digest was mixed with 1  $\mu\text{L}$  of matrix solution containing 20  $\text{mg mL}^{-1}$  DHB (in 50% acetonitrile aqueous solution, v/v) and 1% (v/v)  $\text{H}_3\text{PO}_4$  aqueous solution by pipetting and 0.5  $\mu\text{L}$  of mixture was deposited onto the MALDI target. MALDI-TOF mass spectrometry analysis was performed on an AB SCIEX MALDI-TOF/TOF 5800 mass spectrometer (Foster City, CA, USA) in positive ion mode with a 355 nm Nd:YAG laser, 200 Hz repetition rate, and 20 kV acceleration voltage. Search parameters of fragment ion spectra were submitted to MASCOT (<http://www.matrixscience.com/>) for a database search and identification of corresponding peptides.

**Characterization.** Scanning electron microscopy (SEM) images were obtained on a field emission scanning electron microscope (FESEM; NanoSEM 630, NOVA). Transmission electron microscopy (TEM) images were taken with a JEOL-2010 microscope at the accelerating voltage of 200 kV. Powder X-ray diffraction (XRD) patterns were collected on a PANalytical Empyrean X-ray powder diffractometer (Cu  $K\alpha$  radiation, 45 kV, 40 mA) with the detective range from 5 to 80 degrees. Raman spectra were recorded on a WITec Confocal Raman instrument with a 514 nm laser wavelength. The zeta potential of the particles was recorded using a Malvern Zetasizer ZS. Fourier transform infrared spectra were determined on a Bruker Vertex V70 FTIR spectrometer over a potassium bromide pellet and then scanned from 400 to  $4000\text{ cm}^{-1}$  at a resolution of  $6\text{ cm}^{-1}$ . Magnetization measurement was carried out with a superconducting quantum interface device (SQUID) magnetometer at 300 K. The UV-Vis adsorption spectral values were performed with a Perkin-Elmer Lambda 950 Spectrophotometer.

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## Author contributions

G.C. designed and performed the experiments. G.C. and S.Y.Z. conceived the idea, discussed and wrote the paper.

## Additional information

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