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Unified and practical access to α -alkynylated carbonyl derivatives via streamlined assembly at room temperature

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The development of a unified and straightforward method for the synthesis of α -alkynylated ketones, esters, and amides is an unmet challenge. Here we report a general and practical protocol to access α -alkynylated esters, ketones, and amides with diverse substitution patterns enabled by dual-catalyzed spontaneous formation of $C_{sp^3-sp^3}$ and C_{sp^3-sp} bond from alkenes at room temperature. This directing-group-free strategy is operationally simple, and allows for the straightforward introduction of an alkynyl group onto α -position of carbonyl group along with the streamlined skeleton assembly, providing a unified protocol to synthesize various α -alkynylated esters, acids, amides, ketones, and aldehydes, from readily available starting materials with excellent functional group compatibility.

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Alkynes and carbonyl derivatives are among the most important functional groups since they are ubiquitous in organic compounds as well as they serve as useful chemical handles transformed to other diverse functional groups^{1–4}. The introduction of alkyne onto a specific remote position to carbonyl groups is of interest to organic chemists^{5–14}. In particular, the alkylation at α -position of carbonyl functional group is challenging. Yu and Chatani reported a coordination-assisted strategy for alkylation of amides via Pd/Rh catalyzed C–H activation (Fig. 1a)^{15,16}. This strategy gave only two examples for α -alkynylation with poor results (<40% yield) and is only applicable to the terminal methyl group of amide with directing group. Recently, Zhu and Studer independently developed an elegant alkynyl migration of propargyl alcohol with a pendant olefin via a radical initiated chain reaction to give α -alkynylated ketones (Fig. 1b)^{17–19}. Despite significant progress in this field²⁰, the existing methods suffer from several major limitations and drawbacks: (1) Requiring additional steps to synthesize the backbone of sophisticated substrates. (2) Restricted to specific carbonyl functional groups (amide or ketone). (3) Limited substitution patterns at α -position (R=H, or C_nF_mCH₂). Thus, a general, practical, and straightforward method to introduce α -alkynylation for diverse carbonyl derivatives, including ketones,

esters, and amides, from easily available and cost-effective starting materials is highly desirable. On the other hand, intermolecular carbo-difunctionalization of alkenes is unarguably an attractive alternative to build molecular complexity via simultaneous formation of two C–C bonds by backbone assembly^{21–35}.

Herein, we establish a unified and general protocol for the direct synthesis of α -alkynylated aryl or alkyl ketones, aldehydes, esters, acids, secondary and tertiary amides using α -bromomethyl carbonyl precursors with alkynes in the presence of alkenes at room temperature (Fig. 1c). Over the past years, reports disclosed that α -halomethyl carbonyl compounds could be converted into alkyl radicals (A)^{36–46} to initiate coupling with unsaturated systems, such as alkenes and alkynes, to produce C-centered radical (B) (Fig. 1c). We hypothesize that the direct utilization of the radical intermediate (B) with copper catalysis^{47–51} by the merger of photocatalysis^{52–57} to give desired α -alkynylated carbonyl compounds. However, several highly competitive reactions have to be suppressed. First, radical B is facile to undergo atom transfer^{39–42} or single electron oxidation^{43–46} by transition metal or photocatalyst ($k = 10^5–10^9 \text{ M}^{-1} \text{ s}^{-1}$)^{58,59}. Moreover, A would directly undergo metal-catalyzed (Pd, Cu) cross-coupling to give α -alkynylated carbonyl derivatives C in the presence of alkyne^{60,61}.

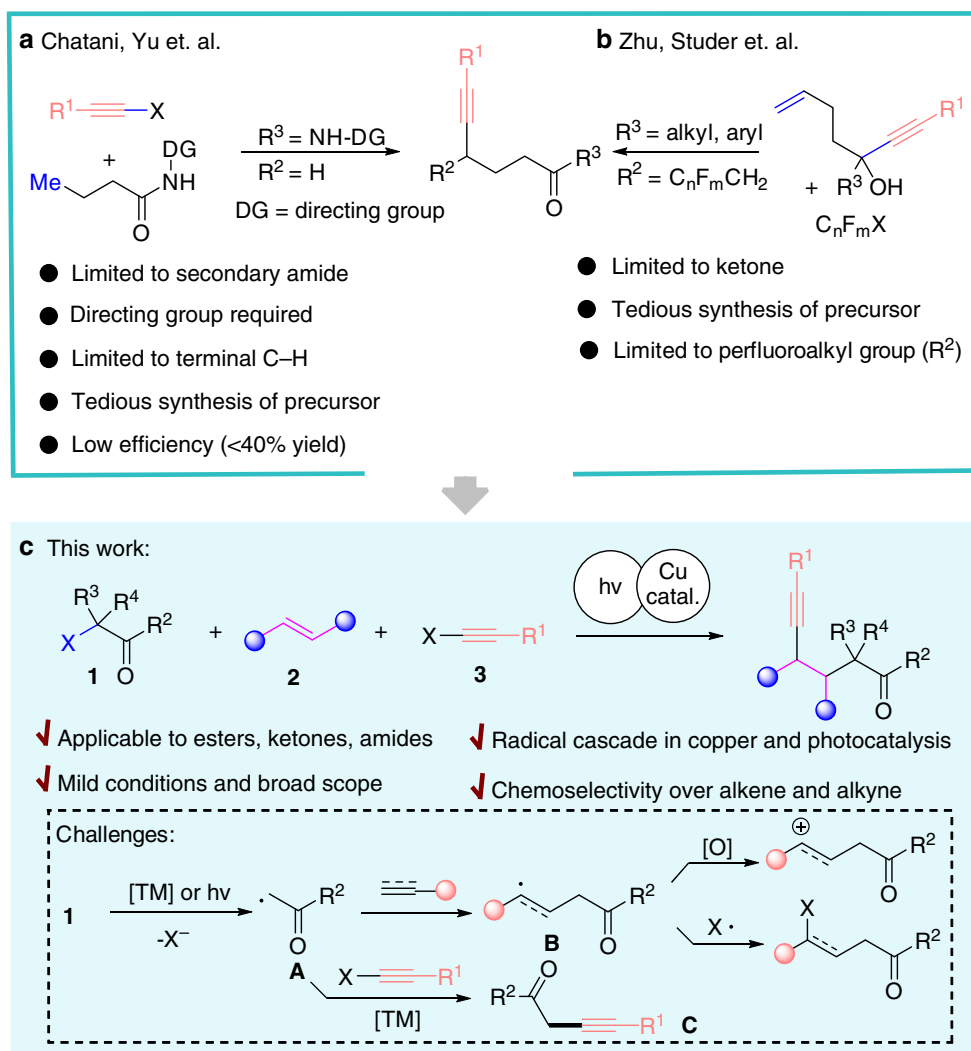
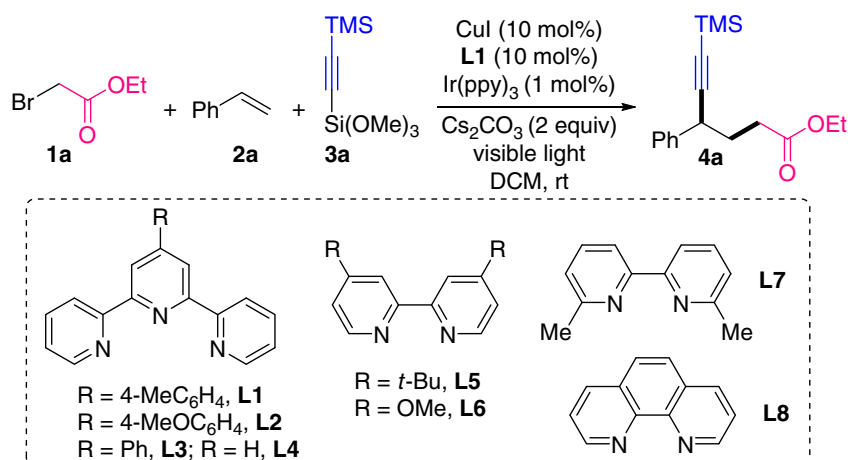


Fig. 1 Synthesis of α -alkynylated carbonyl derivatives. **a** Coordination-assisted strategy for alkylation of amides via Pd/Rh catalyzed C–H activation. **b** Alkynyl migration of propargyl alcohols with a pendant olefin by radical initiated chain reaction. **c** This work

Table 1 Effect of the reaction parameters^a

Entry	Variation from the "standard" conditions	Yield of 4a ^b
1	None	82% (78%) ^c
2	No CuI	0%
3	No light	Trace
4	No Ir(ppy) ₃	35%
5	L2 instead of L1	75%
6	L3 instead of L1	70%
7	L4 instead of L1	36%
8	L5 instead of L1	23%
9	L6 instead of L1	28%
10	L7 instead of L1	20%
11	L8 instead of L1	18%
12	CuTc instead of CuI	66%
13	K ₂ CO ₃ instead of Cs ₂ CO ₃	Trace
14	<i>t</i> -BuOLi instead of Cs ₂ CO ₃	5%
15	DCE instead of DCM	40%
16	Chloroform instead of DCM	Trace

^aThe reaction was carried out on 0.1 mmol scale of **2a**, using **1a** (2 equiv) and **3a** (2 equiv) in DCM (3 mL) under the irradiation of 30 W blue LED for 12 h

^bThe yield was determined by the ¹H NMR of crude mixture using mesitylene as internal standard

^cIsolated yield after flash chromatography

Results

Reaction conditions evaluation. With these concerns in mind, we set out to explore the possibility of direct incorporating radical intermediate **B** into copper-catalyzed C_{sp3}-sp bond-forming to furnish γ -alkynylated carbonyl derivatives using ethyl α -bromoacetate **1a**, styrene **2a**, and trimethylsilylacetylene. After some initial trials, we found the use of alkynyl silicate **3a** significantly improve the radical cascade coupling efficiency. Upon intensive examining a wide range of reaction parameters, we determined that CuI (10 mol%), a tridentate ligand **L1** (10 mol%) and a photocatalyst Ir(ppy)₃ (1 mol%) can accomplish the desired reaction in the presence of Cs₂CO₃ using DCM as solvent with blue LED irradiation, affording γ -ethynyl ester **4a** in 78% isolated yield (Table 1, entry 1). No desired product was obtained in the absence of copper or light (entries 2 and 3). The absence of photocatalyst dramatically decreased the reaction efficiency, albeit delivering **4a** in 35% yield (entry 4). The selection of ligand has significant impact on the radical cascade process. The employment of tridentate ligand is crucial for the success (entries 5–7). The use of bidentate ligand gave **4a** in low yields (entries 8–11). When CuTc was used, **4a** was obtained in 66% yield (entry 12). When the reaction was carried out with other bases, such as potassium carbonate or lithium *tert*-butoxide, no substantial

amount of **4a** was detected (entries 13 and 14). The use of other chlorine-containing solvents led to diminished yields (entries 15 and 16).

Reaction scope. With the optimal conditions in hand, we turned to test the generality of this reaction. We first evaluated different types of carbonyl derivatives (Fig. 2a). The reaction is applicable to a variety of α -bromomethyl carbonyl functional groups, affording γ -alkynylated esters, ketones, secondary and tertiary amides, which are inaccessible otherwise. γ -Alkynylated esters with two or one enolizable proton can be obtained in good yields (**4a–4e**). γ -Alkynylated lactone could be isolated in moderate yield (**4e**). Notably, this method is applicable to synthesize α,α -disubstituted γ -alkynylated esters in good yields (**4f** and **4g**). It is delighting that γ -alkynylated alkyl or aryl ketones could be furnished in good yields (**4h–4j**). This reaction also tolerates secondary and tertiary amides to afford corresponding γ -alkynylated amides in synthetically useful yields (**4k** and **4l**). This protocol tolerates a wide variety of carbonyl derivatives, providing a unified directing-group-free alternative to access γ -alkynylation of esters, ketones, amides with diverse substitution patterns under mild conditions.

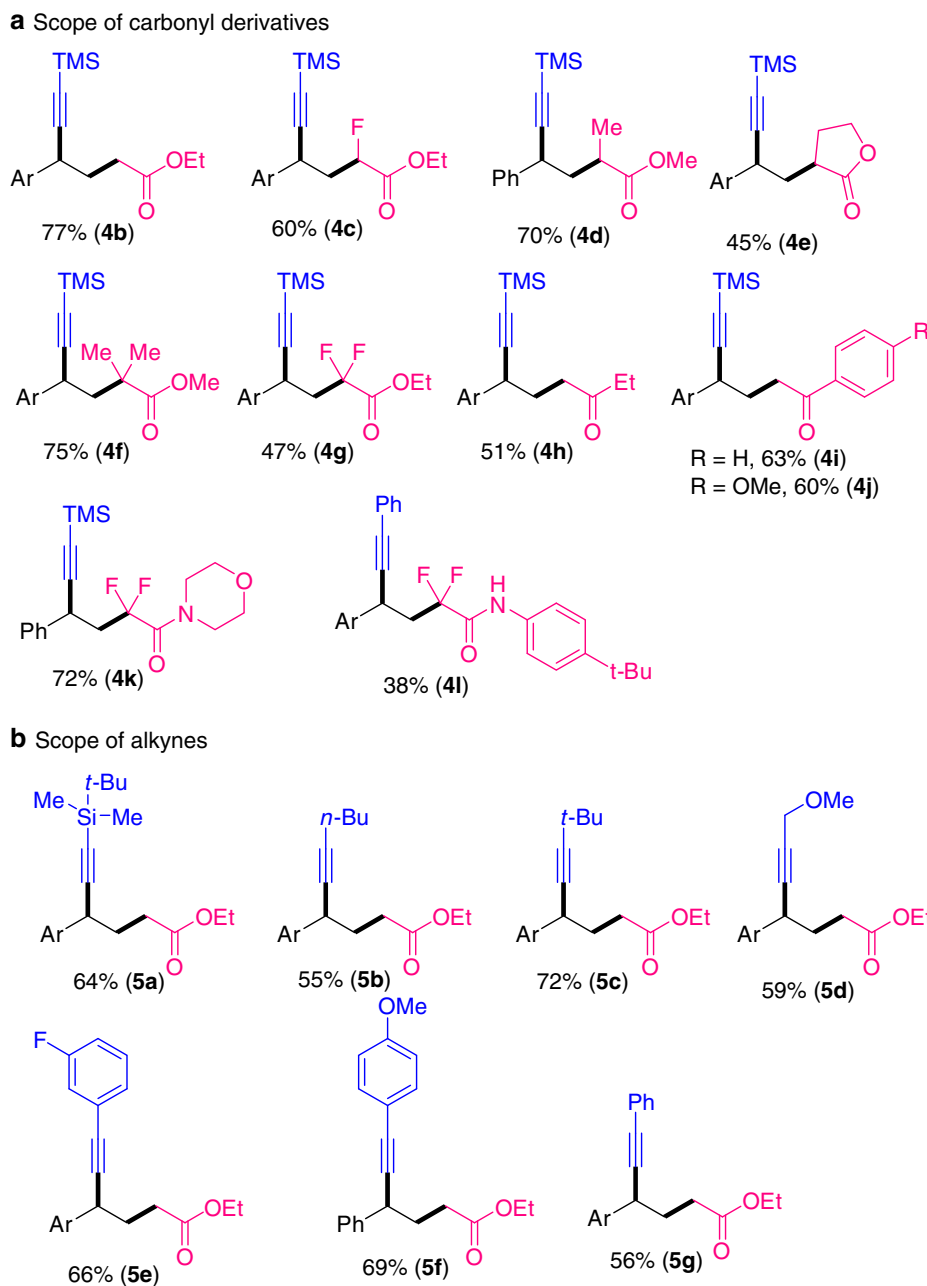


Fig. 2 Scope of different carbonyl derivatives and alkynes. **a** Scope of carbonyl derivatives. **b** Scope of alkynes. For reaction conditions, see Table 1, entry 1. Ar = 4-MeOC₆H₄

Next, we tested the scope of alkyne precursors using ethyl α -bromoacetate **1a** (Fig. 2b). The reaction tolerates silyl and alkyl alkynes, giving corresponding γ -alkynylated esters in good yields (**5a–5d**). Aryl alkynes with electron-donating or -withdrawing group proceeded smoothly under the reaction conditions, affording desired product in good yields (**5e–5g**).

We also examined the scope of alkenes (Fig. 3), which corresponds to the substitution patterns of γ - and β -position of the esters using ethyl α -bromoacetate (**1a**) and trimethylethynyl silicate (**3a**). Styrenes with electron-donating or -withdrawing substituents at *para*-, *meta*- or *ortho*-position could be all applied in the reaction conditions, affording corresponding γ -alkynylated esters in good yields with diverse aryl substitutions at γ -position (**4a**, **4b**, **6a–6l**). Electron-deficient styrenes, which are challenging for radical involved cross-coupling due to the propensity of radical oligomerization, are also good substrates for this reaction

(**6d** and **6e**). Vinylpyridine is tolerated in this reaction, furnishing γ -pyridinyl substituted γ -alkynylester **6m** in 52% yield. Cyclic and acyclic internal alkenes could be transformed into the desired products with γ - and β -substitutions in moderate yields (**6n** and **6o**). Emamine could be transformed to γ -alkynyl γ -aminoester **6p** in 33% yield. To further demonstrate the potential utility of this protocol, late-stage functionalization of natural product derivatives was examined. Isoflavone, estrone, tocopherol occurring alkenes could be tolerated under the reaction conditions and be further modified to deliver natural product containing γ -alkynylesters in good yields (**6q–6s**).

Synthetic application. We further demonstrated the synthetic application potential of this reaction using **4a** as the model substrate (Fig. 4). Trimethylsilyl group could be removed under basic

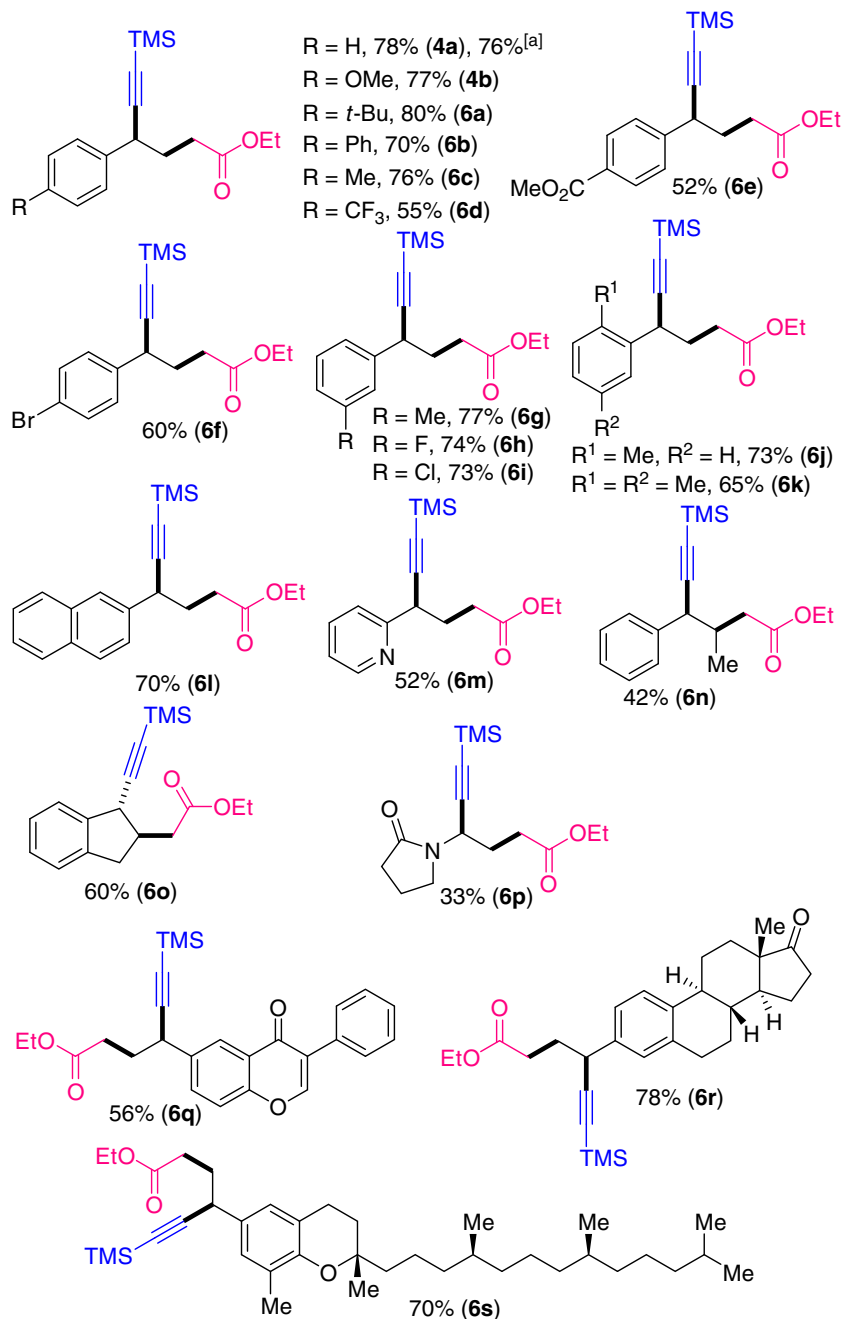


Fig. 3 Scope of alkenes. For reaction conditions, see Table 1, entry 1. When it is applicable, 1:1 dr is observed. ^[a]Isolated yield on 1.0 mmol scale

conditions, delivering γ -alkynyl ester (**7a**) or γ -allenyl ester (**7b**) in a controllable manner in 82% and 70% yield, respectively. In the presence of sodium hydroxide, **4a** was deprotected as well as saponificated, giving the formal γ -ethynyl carboxylic acid (**7c**) in 60% yield, which is the precursor for an inhibitor of serine proteases⁶². γ -Triazole substituted ester (**7d**) could be obtained in 54% yield via copper catalyzed [3+2] reaction. The triple bond could also be cleaved via ruthenium catalysis to deliver 1,4-dicarboxylic acid (**7e**) in 70% yield. In the presence of DIBAL-H, δ -alkenyl or δ -alkynyl alcohol (**7f** and **7g**) could be obtained in good yields depending on the reaction conditions, providing a divergent method to functionalize δ -position of alcohol which is inaccessible otherwise. It is noteworthy that γ -alkynylated aldehyde (**7h**) could be obtained in 62% yield, rendering this three-component radical cascade protocol accessible to γ -alkynylated acids, esters, amides, ketones, and aldehydes.

Mechanistic investigation. To further understand the reaction process, we carried out a series of experiments to probe the reaction mechanism (Fig. 5). In the presence of TEMPO, the reaction was completely shut down with the formation TEMPO adduct **8** in 60% yield, indicating the formation of radical intermediate **M2**. The use of 2-cyclopropyl styrene under the standard conditions delivered the radical ring-opening and cyclization product **9**, which further suggested the presence of radical intermediate **M4**. When phenylethynyl copper was used as the catalyst, the desired product **10** was obtained in 66% yield, proving alkynylcopper **M1** as one of the intermediates in the catalytic cycle.

Based on the results and literature, we proposed the following reaction mechanism (Fig. 6). First, ligand ligated copper (I) undergoes transmetalation with alkynyl silicate to generate alkynyl copper (I) species **M1**, which reduces α -bromomethyl

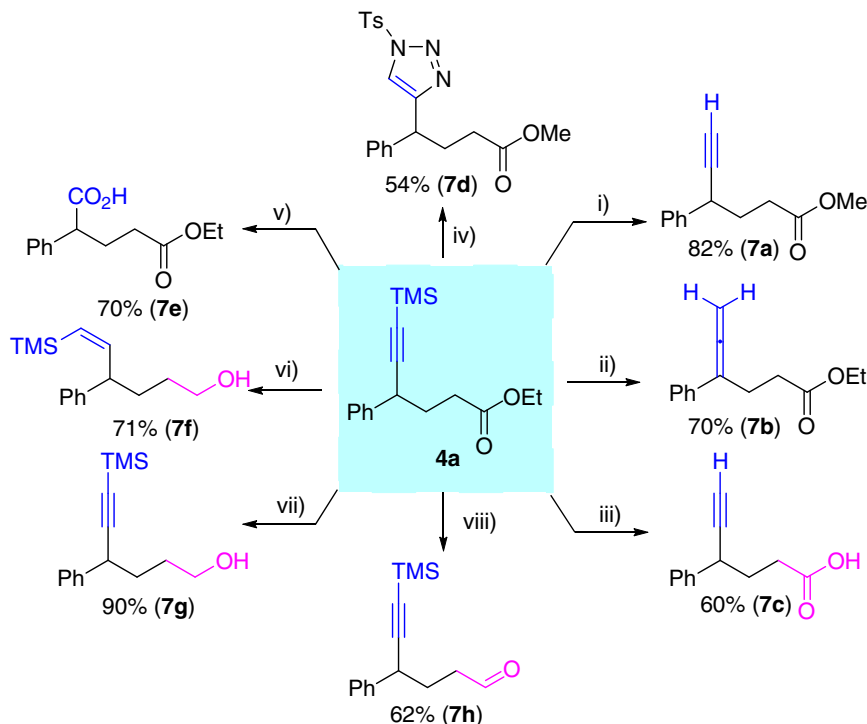


Fig. 4 Synthetic applications of the γ -alkynylated ester. **4a** Reaction conditions: (i) K_2CO_3 (3.0 equiv), MeOH (0.05 M), rt, 2 h. (ii) TBAF (3.0 equiv), THF (0.05 M), rt, 10 h. (iii) NaOH (2.0 M aq.), MeOH (0.1 M), rt, 3.5 h. (iv) K_2CO_3 (3.0 equiv), MeOH (0.05 M), rt, 2 h, then CuTc (10 mol%), TsN_3 (1.0 equiv), toluene, rt, 4 h. v) $RuCl_3$ (5 mol%), $NaIO_4$ (4.0 equiv), $CCl_4/CH_3CN/H_2O = 2/2/3$, rt, 2 h. (vi) DIBAL-H (5.0 equiv), Et_2O , 40 °C, 20 h. (vii) DIBAL-H (3.0 equiv), Et_2O , -78 °C, 2 h. (viii) DIBAL-H (1.0 equiv), Et_2O , -78 °C, 2 h

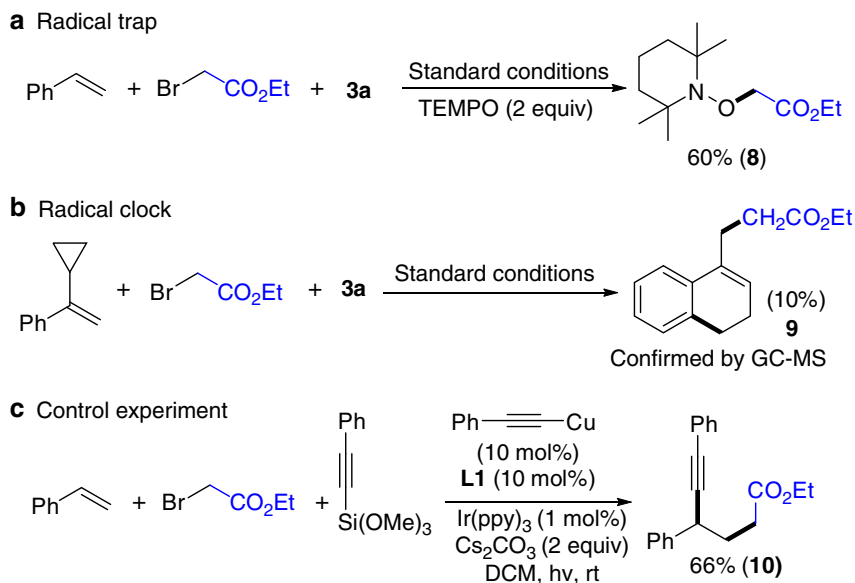


Fig. 5 Mechanistic investigations. **a** Radical trap experiment with TEMPO as radical scavenger. **b** Radical clock experiment using cyclopropyl styrene. **c** Control experiment using phenylethyne copper (I) as catalyst

carbonyl derivatives **1** to give radical intermediate **M2** and copper (II) intermediate **M3** by single electron transfer. **M2** could be trapped by alkene **2** to give a new carbon-centered radical **M4**^{39–46}, which recombines with **M3** to deliver intermediate **M5**. **M5** could be activated by energy transfer from excited Ir(III) photocatalyst to generate **M6**⁶³, which accelerates reductive elimination to furnish streamlined three-component assembly product and regenerate Cu(I) catalyst. At this stage, the possibility of undergoing direct reductive

elimination from **M5** to generate final product and Cu (I) species could not be ruled out^{64,65}.

Discussion

In summary, we have developed a unified γ -alkynylation of esters, ketones, secondary and tertiary amides enabled by the combination of copper- and visible-light catalysis at room temperature. This method features with the exclusive chemoselectivity of

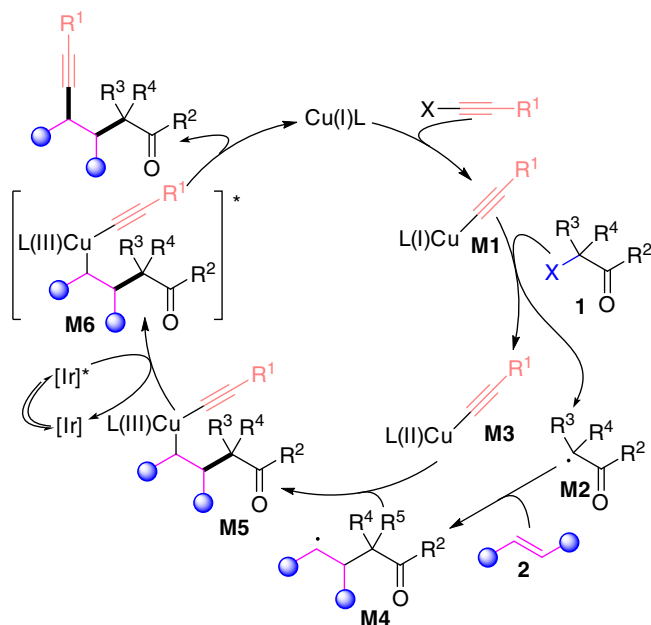


Fig. 6 Mechanistic hypothesis. Proposed mechanism of the copper- and visible-light catalyzed α -alkynylation reaction

alkenes over alkynes, and is applicable to a wide variety of carbonyl derivatives with excellent functional group compatibility. The merger of photocatalysis and copper catalysis along with judicious selection of ligand allows for the streamlined formation of $C_{sp^3-sp^3}$ bond and C_{sp^3-sp} bond to assemble the carbon chain skeleton and alkynylate α -position from alkene, α -bromomethyl carbonyl precursors, and alkyne silicates, circumventing the tedious synthesis of skeleton. Further synthetic extension provides access to α -alkynylated esters, acids, amides, ketones, and aldehydes. Mechanistic investigation suggests the reaction undergoes copper catalyzed radical cascade reaction, facilitated by energy transfer from excited photocatalyst.

Methods

Synthetic procedures. See Supplementary Methods.

Characterization. See Supplementary Methods, Supplementary Figs. 1–103 for ^1H , ^{19}F and ^{13}C NMR spectra of synthesized products.

Data availability

The datasets generated during and/or analyzed during the current study are available from the corresponding author on reasonable request.

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Author contributions

X.L.L. performed the experiments and collected the data. W.S. conceived the idea, supervised the project, and prepared the paper with contributions from X.L.L.; both authors contributed to discussions and data analysis.

Competing interests

The authors declare no competing interests.


Additional information

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