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Hot dense silica glass with ultrahigh elastic moduli

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Silicate and oxide glasses are often chemically doped with a variety of cations to tune for desirable properties in technological applications, but their performances are often limited by relatively lower mechanical and elastic properties. Finding a new route to synthesize silica-based glasses with high elastic and mechanical properties needs to be explored. Here, we report a dense SiO₂-glass with ultra-high elastic moduli using sound velocity measurements by Brillouin scattering up to 72 GPa at 300 K. High-temperature measurements were performed up to 63 GPa at 750 K and 59 GPa at 1000 K. Compared to compression at 300 K, elevated temperature helps compressed SiO₂-glass effectively overcome the kinetic barrier to undergo permanent densification with enhanced coordination number and connectivity. This hot compressed SiO₂-glass exhibits a substantially high bulk modulus of 361–429 GPa which is at least 2–3 times greater than the metallic, oxide, and silicate glasses at ambient conditions. Its Poisson's ratio, an indicator for the packing efficiency, is comparable to the metallic glasses. Even after temperature quench and decompression to ambient conditions, the SiO₂-glass retains some of its unique properties at compression and possesses a Poisson's ratio of 0.248(11). In addition to chemical alternatives in glass syntheses, coupled compression and heating treatments can be an effective means to enhance mechanical and elastic properties in high-performance glasses.

Glasses are indispensable materials in the industrial and technological applications owing to their unique and homogenous optical, mechanical, and chemical properties¹. The flexibility of designing and producing a wide variety of glasses with desirable properties and chemistries have made them of considerable interests in material science and engineering applications². In particular, mechanical properties, such as hardness, rigidity, and stiffness, are key to the performances of glasses and can be associated with their elastic moduli and atomic packing density (C_g)^{3–8}. C_g , defined by the ratio between the minimum theoretical and the corresponding effective volume of glass, reflects the atomic-scale structure of glass and was recently noted to be directly correlated to the Poisson's ratio (ν)^{9–11}. Elastic moduli reflect the bonding strength between atoms and strongly depend on the local atomic structures. Therefore, glasses with certain elastic moduli and ν can thus be indicative of their potential mechanical performances.

As the archetype of widely used oxide and silicate glasses and an analogue of felsic melts in nature, SiO₂-glass has a substantially high glass transition temperature (T_g) which is an important indicator for their mechanical integrities such as hardness and stiffness in high-temperature applications^{10,12–15}. Compared to the bulk metallic glasses, however, SiO₂-glass has a significantly low C_g of 0.45 ($\nu=0.15$), bulk modulus, K_s of 33 GPa, and Young's modulus, E of 70 GPa¹⁰. C_g and elastic moduli of SiO₂-glass can be tailored by performing thermal treatment or chemical doping of various cations^{14–17}. For example, SiO₂-glasses doped with rare-earth oxides can have C_g and elastic moduli comparable or even greater than most of the bulk metallic glasses^{18–20}. However, doping elements could lower the T_g of SiO₂-glass^{10,14–17}, which can greatly decrease the elastic moduli in the high-temperature applications^{10,14–17}. The relationship between the density and thermal history of a glass can be complex^{21,22}. Even with the same composition, a glass material could have different densities by varying the cooling rate and thermal history^{21,22}. Designing SiO₂- or SiO₂-based glasses with extremely high elastic moduli by retaining a great T_g value remains to be explored^{10,14–17}.

In addition to these two well-known methods, the treatment of glasses at high pressures (or compression) by shortening the interatomic distance and modifying the bonding pattern can be an effective means to induce irreversible densification and property modifications. The drawback in T_g variation in chemically-doped glasses

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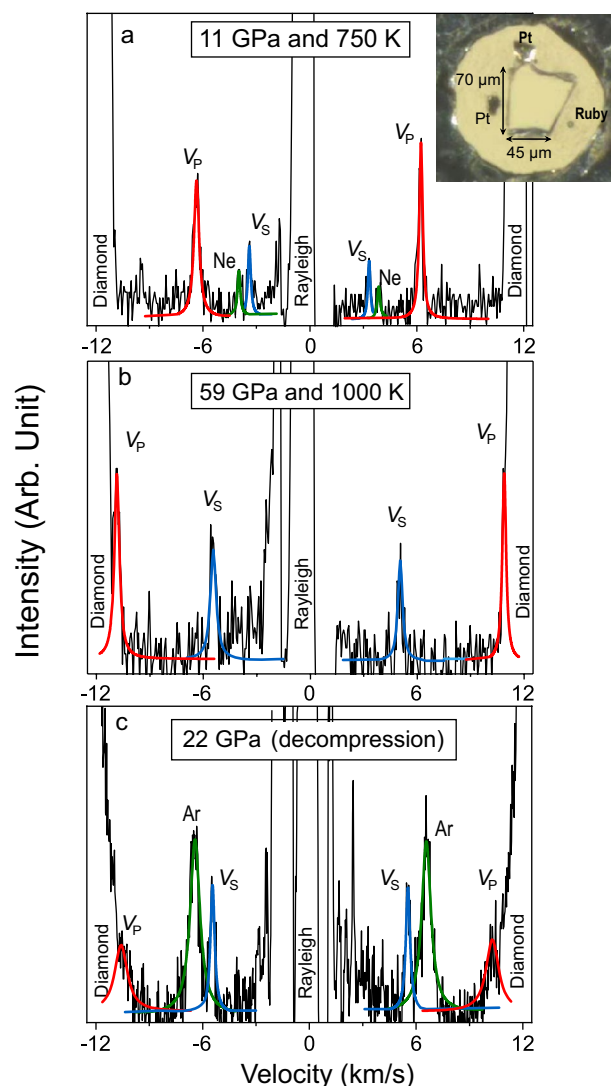


Figure 1. Representative Brillouin spectra of SiO_2 glass at high pressures and temperatures. (a) At 11 GPa and 750 K; (b) at 59 GPa and 1000 K; (c) quenched and decompression to 22 GPa and 300 K. Black lines: raw data; red lines: fitting results for the longitudinal modes, V_p ; blue lines: fitting results for the transverse modes, V_s . Insert figure is a representative sample photo taken at 7 GPa.

becomes a non-factor in this process^{23–30}. Since SiO_2 -glass is a key analogue to geologically abundant silicate melts, there have been a series of studies on its atomistic, elastic, and mechanical properties in compression and after high pressure–temperature (P – T) quenching^{28–32}. Previous studies have showed that increasing pressure to 20 GPa at 300 K can lead to irreversible permanent densification of SiO_2 -glass with a gradual increase in the fraction of SiO_5 -pentahedra, SiO_6 -octahedra, and shared edges³³. The disappearance of the SiO_4 -tetrahedra occurs between 20 and 45 GPa, accompanied by a dramatic increase in the number of SiO_6 -octahedra and shared edges and generating a more densified SiO_2 -glass^{33,34}. At pressures between 45 and 140 GPa, Si–O polyhedra with a greater coordination number and connectivity will be present, producing a high-density SiO_2 -glass^{28,30,33}. Simultaneous high P – T of SiO_2 -glass can be more effective in producing the aforementioned changes because the elevated temperature can help overcome the kinetic barrier of the transitions^{35–43}. However, in situ characterizations of the mechanical and elastic properties of glasses at high P – T remain technically challenging. As such, hot compression of SiO_2 -glass was only performed at pressures less than 20 GPa and many previous studies focused on characterizations of quenched samples^{35–43}.

One insightful approach to evaluate mechanical and atomistic behaviours of SiO_2 -glass at high P – T is to measure its sound velocities in situ for derivations of the elastic moduli and Poisson's ratio¹⁰. In this study, we have measured the compressional- (V_p) and shear-wave (V_s) velocities of SiO_2 -glass at simultaneously high P – T conditions up to 72 GPa at 300 K, 63 GPa at 750 K, and 59 GPa at 1000 K using Brillouin scattering in externally heated diamond anvil cells (Figs. 1 and S1, see “Methods” section for detail). The obtained results are applied to constrain ν and elastic moduli (K_s , G , and E) simultaneously. Our results show that compression at high temperatures leads to irreversible densification and produces a dense SiO_2 -glass with ultra-high elastic moduli. More

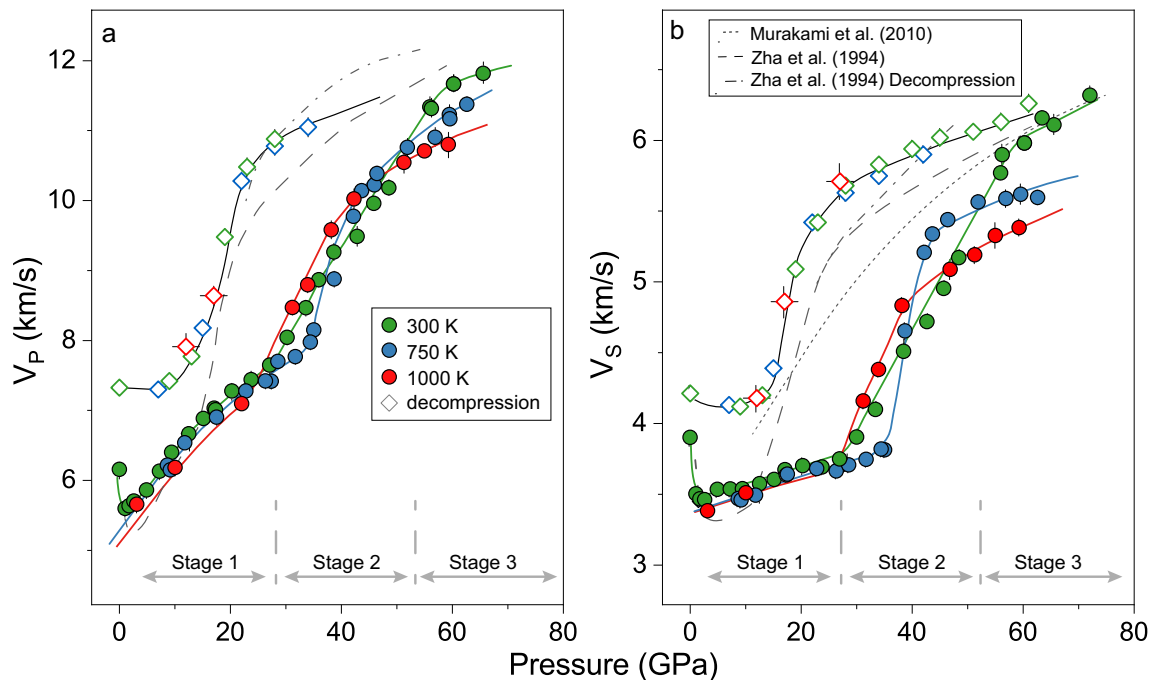


Figure 2. V_p and V_s of SiO_2 glass at given temperatures. (a) V_p of SiO_2 glass; (b) V_s of SiO_2 glass. Green, blue, and red circles represent the velocity data at 300 K, 750 K and 1000 K by compression, respectively. Green, blue, and red diamonds are our results in decompression for SiO_2 -glass quenched from 300 K, 750 K, and 1000 K, respectively. All the colour solid lines are shown for readers to follow the trend with pressure. Grey dotted lines: SiO_2 glass in Murakami et al. (2010)²⁸; grey long-dashed lines: SiO_2 -glass in compression in Zha et al. (1994)⁴⁹; grey dotted and dashed lines: SiO_2 -glass in decompression in Zha et al. (1994)⁴⁹.

importantly, the densification process by hot compression exhibits a strong dependence on the thermal path. A dense SiO_2 -glass with ultra-high elastic moduli can be achieved at lower pressures by compression at 1000 K than that under compression at 300 K. Regardless of its thermal history, a dense SiO_2 -glass with ν of 0.248(11) can be retained to ambient conditions after the treatment at extreme environments. Hot compression treatment can be used as an effective means for designing dense glasses with ultra-high elastic moduli in the future.

Results

The V_p and V_s of SiO_2 -glass were well constrained by our Brillouin measurements up to 72 GPa at 300 K, 63 GPa at 750 K, and 59 GPa at 1000 K (Figs. 1, 2, and S1). Analysis of our velocity results show that both V_p and V_s decrease with increasing pressure up to approximately 2.5 GPa at 300 K, while they start to increase at higher pressure. The most prominent feature of the velocity–pressure profile at 300 K is the dramatic increase in the V_p and V_s between 27 and 56 GPa (Fig. 2). Above 56 GPa, the V_p and V_s of the SiO_2 -glass exhibit a nearly linear increase with pressure. These pressure-dependent features are consistent with previous high-pressure studies for the first order^{26–30}.

On the other hand, high P–T experiments have revealed that sound velocities of the SiO_2 -glass have a weak dependence on temperature at pressures below 25 GPa (Fig. 2). The V_p and V_s of SiO_2 -glass at 1000 K are 3(1)% lower than those at 300 K between 1 bar and 25 GPa, respectively. At pressures between 35 and 43 GPa, the temperature effect on velocities is very different: a sudden increase in the V_p and V_s with increasing temperature was observed (Fig. 2). This anomalous change in the V_p and V_s at 750 K starts at a higher pressure but ends at a lower pressure than those at 300 K. Our velocity–pressure relation at 1000 K also exhibits an anomalous feature with a sudden increase in both V_p and V_s at 1000 K beginning at pressures as low as ~27 GPa and ending at 38 GPa. Interestingly, we observed a much stronger temperature effect on the reduction of the sound velocities of the SiO_2 -glass at higher pressures after the occurrence of the sudden increase: increasing the temperature from 300 to 1000 K at ~56 GPa leads to 6(1)% and 9(1)% reductions in the V_p and V_s of SiO_2 -glass, respectively, whereas the V_p and V_s below 25 GPa can only be lowered by 3(1)%.

After the high P–T measurements were completed at the maximum targeted pressures, electric power to the resistive heater on the diamond cell was cut off to quench the compressed SiO_2 -glass to 300 K for further decompression experiments. The temperature quenched SiO_2 -glass has V_p and V_s 39(2)% and 51(3)% larger than those at 750 K at 34 GPa, respectively (Fig. 2). More importantly, the V_p and V_s of SiO_2 -glass in decompression measurements, regardless of the previous heating temperature, were observed to consistent with each other and follow the same pressure-dependent trend within experimental uncertainties. During decompression, V_s followed a quasi-linear decrease with pressure down to 27 GPa. V_p , on the other hand, was too weak to be detected during decompression until the pressure was lowered to 33 GPa. The difference in the sound velocity between decompression and compression reaches a maximum of 55(2)% at 27 GPa, showing a large hysteresis loop. We observed a sudden drop in both V_p and V_s between 27 and 13 GPa by further lowering the pressure. It is worth

noting that the V_p and V_s in decompression between 1 bar and 13 GPa are 14(1)% and 15(1)% greater than those in compression at this pressure range, respectively.

Discussion

Our high P–T sound velocity data are used to constrain the ν and elastic moduli of SiO₂-glass. The obtained V_p and V_s can be directly used to calculate the Poisson's ratio, ν , using the following equation:

$$\nu = 0.5 \times \frac{(V_p/V_s)^2 - 2}{(V_p/V_s)^2 - 1} \quad (1)$$

Together with the literature experimental equation of state results and theoretical prediction for the influence of temperature on the density of silicate melts, we have also modelled the density (ρ) of SiO₂-glass at P–T conditions relevant to our measurements (see “Methods” section for details, Fig. S2). The calculated density and measured sound velocities of SiO₂-glass were used to compute the following elastic moduli at a given P–T:

$$\begin{aligned} K_S &= \rho \left(V_p^2 - \frac{4}{3} V_s^2 \right) \\ G &= \rho V_s^2 \\ E &= 2G(1 + \nu) = 3K_S(1 - 2\nu) \end{aligned} \quad (2)$$

We estimated that varying density by 5% will cause a $\pm 9, \pm 21, \pm 23$ GPa change in K_S , G , and E at ~ 60 GPa between 300 and 1000 K but has no influence on the calculated ν (Fig. S3).

These results reveal that elevating pressure can produce a dense SiO₂-glass with ultra-high elastic moduli and Poisson's ratio even at simultaneously high P–T (Fig. 3). Here the variation of elastic moduli and ν with pressure is discussed in three stages, which can be associated with change of atomistic behaviours of the SiO₂-glass at high P–T conditions. The first stage occurred in our glass from ambient to 27 GPa at 300 K, which should be caused by the formation of higher-coordinated polyhedron and the transition from corner sharing to a more complex network connectivity in the SiO₂-glass, represented by the total number of share edges per polyhedron, n_{ES} ^{21,33,44}. In particular, the weak decrease in the K_S , G , and E in the initial compression between 0 and 2.5 GPa should be related to the topological evolution of SiO₄-network with an increase in entropy⁴⁵. Further increasing pressure to 27 GPa results in a substantial increase in all the investigated elastic moduli, which are associated with the formation of SiO₅ and SiO₆ polyhedron and an increase in n_{ES} to 1.4³³. At this pressure range, compression causes a dramatic increase in ν , from 0.164(8) at 1 bar to 0.341(4) at 27 GPa and 300 K, and almost double the value of G to 53(1) GPa and E to 143(4) GPa, while quadrupling the K_S to 150(5) GPa.

The sudden jump in the G and E between 27 and 56 GPa at 300 K can be associated with the second stage of the densification. At this pressure range, K_S exhibits a continuous increase with pressure from 150(5) GPa at 27 GPa to 382(14) GPa at 56 GPa and 300 K. Previous theoretical study suggested that the second stage of the densification is represented by an increase in the fraction of SiO₆-octahedra with a jump. Meanwhile, n_{ES} gradually increases with pressure from 1.4 to 1.7³³. We thus conclude that the abnormal change in the G and E between 27 and 56 GPa at 300 K should be mainly caused by the substantial increase in the number of SiO₆-octahedra³³. In contrast, the K_S of the SiO₂-glass is hardly influenced by the change to a higher-coordinated structure but mainly controlled by the increase in the network connectivity. The second stage of the densification produces a SiO₂-glass with an ultra-high K_S of 382(14) GPa, G of 163(4) GPa, and E of 428(12) GPa at 56 GPa and 300 K³³. The K_S of this highly incompressible SiO₂-glass is only $\sim 20\%$ lower than its crystalline counterpart, stishovite and CaCl₂-phase SiO₂, at 56 GPa and 300 K (Fig. 4)⁴⁶.

Unlike the aforementioned elastic moduli, ν of SiO₂-glass in the second stage of the densification exhibits a weak decrease with pressure from 0.341(4) at 27 GPa to 0.325(4) at 56 GPa (Fig. 3). This anomalous reduction in ν should be a result of different responses of K_S and G to the structural changes of SiO₂-glass at high pressures. As noted above, the increase in K_S with pressure at 300 K is dominated by the continuous increase in the n_{ES} , whereas the variation in G is more sensitive to the variation in the coordination number. The dramatic increase in G related to the sudden increase in the fraction of the SiO₆-octahedra plays a major role in the observed softening of ν between 27 and 56 GPa at 300 K. From an atomistic perspective, the transition from the lower-coordinated polyhedron to the SiO₆-octahedra is associated with a reduction in the ionic radius of the bonding oxygen, leading to a lower minimum theoretical volume and atomic packing density, C_g ^{47,48}. Since C_g is directly correlated to ν , the dramatic increase in the SiO₆-octahedra fraction thus produces a reduction in ν .

Above 56 GPa at 300 K, all the investigated elastic moduli of SiO₂-glass follow a nearly linear increase with pressure. Most of the SiO₄-tetrahedra and SiO₅-pentahedra above 55 GPa have transformed into SiO₆-octahedra^{33,34}. SiO₂-glass, dominated by the SiO₆-octahedra, enters the third-stage of the densification. The observed increase in the investigated elastic moduli above 55 GPa should be related to the continuous increase in the n_{ES} to ~ 2 ³³. SiO₂-glass has a substantially great ν value of 0.318(6), K_S of 429(15) GPa, G of 178(4) GPa, and E of 470(13) GPa at 66 GPa and 300 K. It is interesting to note that the pressure dependence of ν for SiO₂-glass in the third stage of the densification is much weaker than that below 27 GPa when the initial compression promotes the formation of higher coordination SiO₅ and SiO₆ polyhedron and an increase of n_{ES} to 1.4³³.

We should also note that, although our measured velocities are in general agreement with previous experimental results below 12 GPa, the second stage of densification for our SiO₂-glass occurred at a greater pressure, leading to lower V_p and V_s between 12 and 56 GPa compared to literature results (Fig. S4)⁴⁹. The elastic moduli calculated using our velocity data in the second stage of densification are expected to be smaller than those computed using literature velocity data. The difference in the measured sound velocities between this work

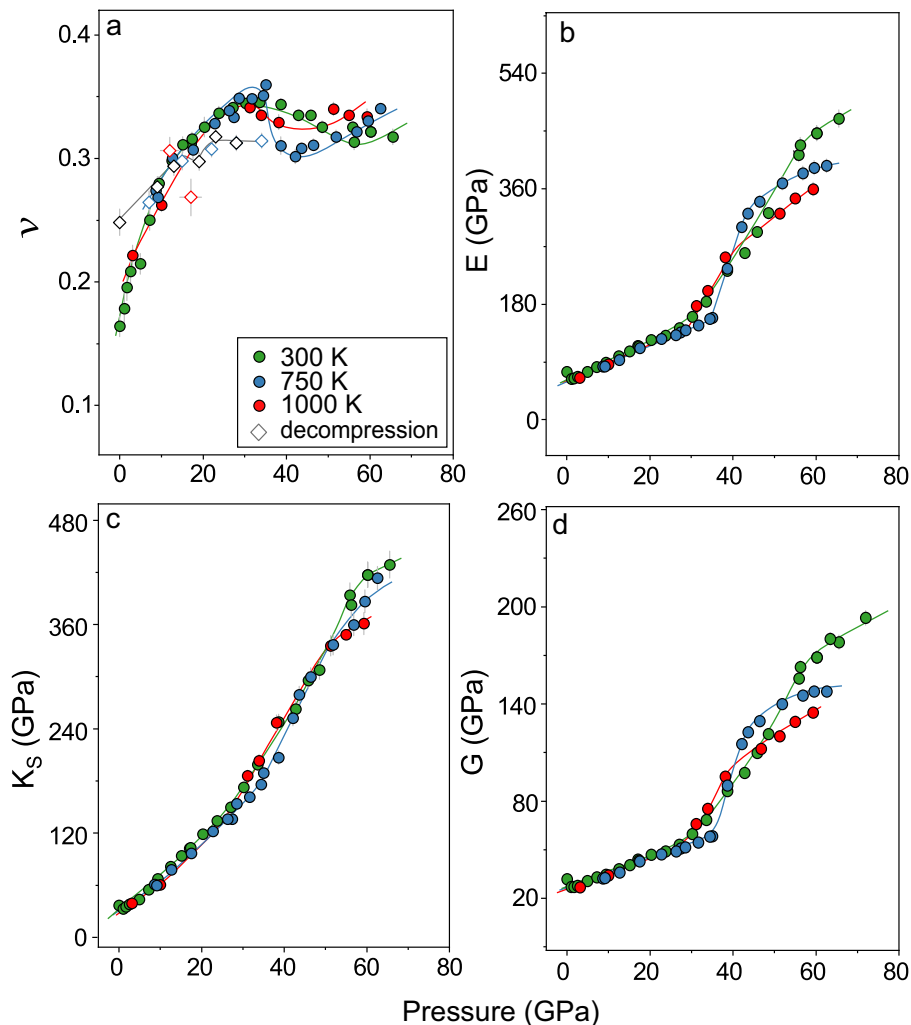


Figure 3. Poisson's ratio, bulk, shear, and Young's moduli of SiO₂ glass at high P–T conditions. **(a)** Poisson's ratio (ν); **(b)** Young's modulus (E); **(c)** bulk modulus (K_S); **(d)** shear modulus (G). Green, blue, and red circles represent the velocity data at 300 K, 750 K and 1000 K by compression, respectively. Green, blue, and red diamonds are our results in decompression for SiO₂-glass quenched from 300 K, 750 K, and 1000 K, respectively. All the colour solid lines are shown to guides the eyes for the pressure-dependent trend.

and previous studies could be caused by a lower initial density of 2.11 g/cm³ for our SiO₂-glass²⁹. This may be a result of different quench rate and thermal history when producing the SiO₂-glass in different companies^{21,44}. A higher pressure is thus required to boost the formation of SiO₆-octahedra in the second stage of the densification for a less dense starting SiO₂-glass. Once all the polyhedral transition into the SiO₆-octahedra, and SiO₂-glass enters the third stage of densification above 56 GPa, our measured sound velocities are indistinguishable from literature values. The initial state of SiO₂-glass also influences the densification process and elastic moduli during compression.

Compared to crystalline counterparts with a similar structure, investigated elastic moduli of SiO₂-glass exhibit a comparable temperature dependence of G and E below 25 GPa but a greater temperature dependence on K_S (Figs. 3, 4). Increasing temperature from 300 to 1000 K only lowers K_S by ~10(1)%, G by ~6(1)%, and E by ~6(1)% at 3–25 GPa. For comparison, ν of coesite (CN = 4) at 1000 K is ~3% lower than at 300 K, and K_S , G , and E are 1%, 5% and 5% lower, respectively. More importantly, our high P–T measurements suggest that varying the temperature can lead to a completely different densification path for SiO₂-glass. The second stage of the densification at high temperatures, represented by the sudden increase in the G and E and a decrease in ν , occurs between 35 and 43 GPa at 750 K and between 27 and 38 GPa at 1000 K. The transition to a structure dominated by the SiO₆ octahedra at high temperatures occurs at a much narrow pressure range than that at 300 K, showing that a higher temperature can help SiO₂-glass quickly overcome the energy barrier and transition to the higher coordinated structure. This also leads an abnormal temperature effect on the elastic moduli in the second stage of the densification. Between 43 and 46 GPa, K_S at 750 K becomes comparable to the value at 300 K, while G and E at 750 K are even greater. K_S at 1000 K is comparable to that at 300 K between 31 and 51 GPa, and G and E at 1000 K has a greater value. In particular, the second stage of densification at 1000 K completes at 38 GPa which

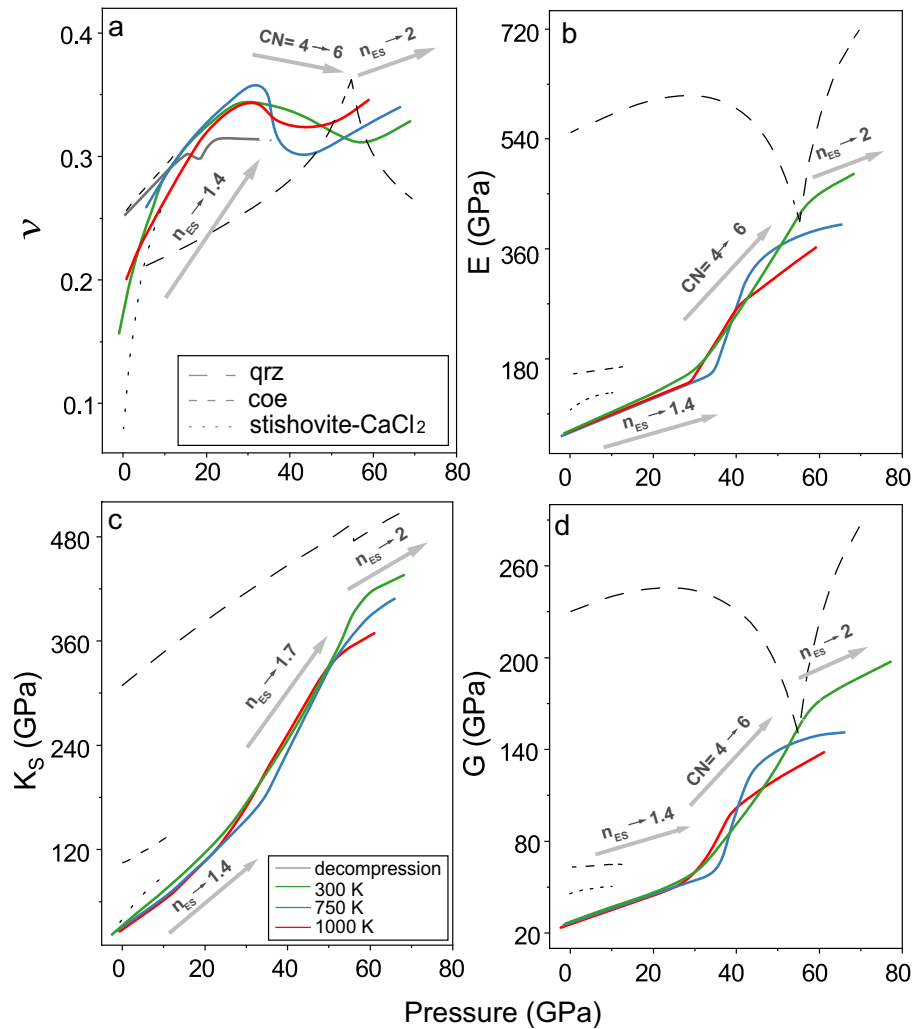


Figure 4. Elastic moduli of SiO₂ glass and crystals. (a) Poisson's ratio (ν); (b) Young's modulus (E); (c) bulk modulus (K_S); (d) shear modulus (G). Green lines: 300 K; blue lines: 750 K; red lines: 1000 K; grey solid line: decompression results. Grey dotted lines: quartz⁵¹; grey short-dashed lines: coesite⁵²; grey dashed lines: stishovite and CaCl₂-type SiO₂⁴⁶. CN: coordination number; n_{ES} : a total number of shared edges per polyhedron. Arrows with CN on the top indicated that the CN increasing from 4 to 6 was the major reason of the moduli increasing, while the arrows with n_{ES} on the top indicated the moduli increasing was mostly driven by the n_{ES} increasing.

is much lower than that at 300 K and 750 K. This indicates that a densified SiO₂ glass may be more preferred at higher temperatures.

In the third stage of the densification, once most of the polyhedra transform into SiO₆-octahedra, increasing the temperature causes a substantial decrease in the investigated elastic moduli. Temperature thus has a greater effect on the elastic moduli of densified SiO₂-glass above 56 GPa than that with similar fractions of SiO₄, SiO₅, and SiO₆ polyhedrons below 20 GPa³³. Increasing the temperature at ~56 GPa from 300 to 1000 K leads to a 10(1)% reduction in K_S , 18(1)% reduction in G , and 18(1)% reduction in E . The temperature effect on the K_S , G , and E of SiO₂-glass occupied by the SiO₆ octahedra is even greater than that of its crystalline counterpart, stishovite. K_S of stishovite at 1000 K is only 10% lower than at 300 K, while G and E are 5% and 6% lower, respectively. In contrast, there is an anomalous increase in ν with increasing temperature above 56 GPa. Elevating the temperature from 300 to 1000 K at ~60 GPa can increase the value of ν from 0.322(4) to ~0.334(6).

Owing to the lack of density information during decompression, elastic moduli of SiO₂-glass cannot be constrained after temperature quench. We thus only track the variation of V_p , V_s , and ν during decompression to the ambient conditions (Figs. 2, 3). By gradually lowering the pressure, a sudden drop in both V_p and V_s was observed between 27 and 13 GPa, which should be related to a fast increase in the fraction of SiO₄-tetrahedra and SiO₅-pentahedra during decompression. This structure transition has a weak effect on ν , represented by a weak softening. We speculate that, although most of the SiO₆-octahedra have transitioned back to the SiO₄-tetrahedra and SiO₅-pentahedra below 13 GPa, the interpolyhedra connectivity dominated by edge sharing may be largely retained during decompression, leading to a more densified SiO₂-glass with a greater V_p , V_s , and ν after the pressure was quenched to 1 bar. ν after decompression to 1 bar has a value of 0.248(11) which is much greater

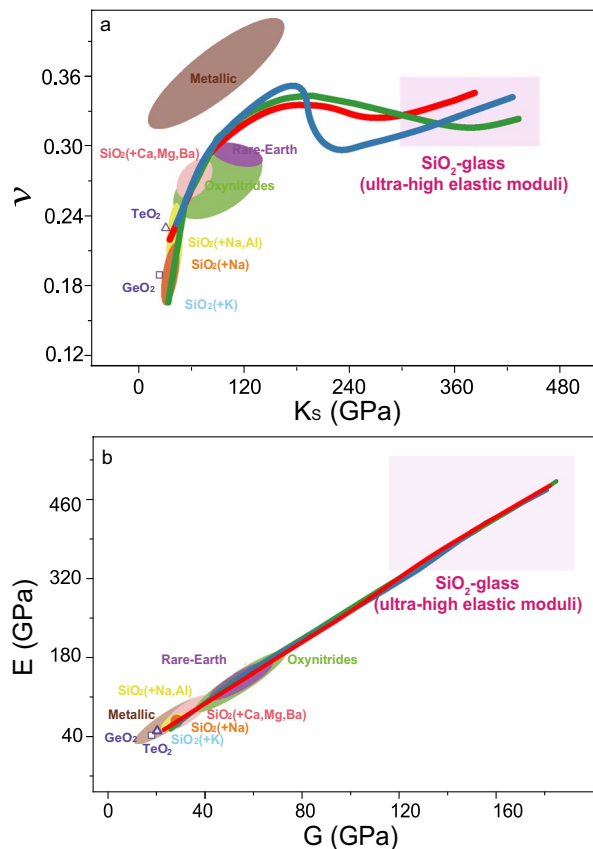


Figure 5. Comparison of Poisson's ratio and elastic moduli of oxide and metallic glasses. **(a)** Poisson's ratio (ν) versus bulk modulus (K_S). **(b)** Young's modulus (E) as a function of shear modulus (G). Green, blue, and red lines represent our results at 300 K, 750 K and 1000 K by compression, respectively. Triangle: TeO_2 glass¹²; square: GeO_2 glass¹²; light blue: SiO_2 glass with K¹⁶; orange: SiO_2 glass with Na¹⁰; SiO_2 glass with Na and Al¹⁵; pink: SiO_2 glass with Ca, Mg and Ba^{14,17}; light green: oxynitride glasses⁵³; purple: rare-Earth glasses^{18,54}; brown: metallic glasses^{19,55}.

than that without treatment at extreme environments. Similar irreversible permanent densification was also observed in a previous cold-compression Brillouin study^{49,50}. Although hysteresis in the V_p was observed during decompression from 57.5 to 26 GPa, most of V_S during decompression was indistinguishable from that upon compression considering the experimental errors (Figs. 2 and S4). Our high P–T data are unique in a way that they not only record the structural change of SiO_2 -glass by hot compression and decompression after temperature quench but also reveal high P–T paths to achieve ultra-high elastic moduli of SiO_2 -glass.

Implication. We further compare the elastic moduli and ν of SiO_2 -glass at high P–T conditions to metallic glasses in a broad range of composition and oxide glasses enriched in SiO_2 (Fig. 5). At ambient conditions, all the SiO_2 -based oxide glasses as well as TeO_2 and GeO_2 glasses have much smaller ν than the metallic glasses but have similar K_S , G , and E values. In contrast, high P–T treatment of SiO_2 -glass can cause permanent densification and also effectively increase the elastic moduli (Fig. 5). Here, our results have shown that great ν , K_S , G , and E are achieved when the structural network connectivity, n_{ES} , increases to 1.4 at 27–35 GPa between 300 and 1000 K accompanied by an increase in the SiO_5 and SiO_6 fraction (Fig. 3). At high P–T condition above ~ 15 GPa along constant experimental temperatures in the study, SiO_2 -glass has ν exceeding all the SiO_2 -based oxide glasses and comparable to some of the metallic glasses. The elastic moduli, K_S , G , and E are greater than the metallic glasses and comparable to rare-earth glasses and oxynitrides. Above 27–35 GPa at all the investigated temperatures, ν , K_S , and G exhibit a different dependence on the structure variation in compression, but the change of G and E follow a similar trend (Fig. 5). The transition to a “stishovite-like” local structure dominated by the SiO_6 -polyhedra with an increase in n_{ES} to 1.7 has a minor effect on ν at between 27 and 56 GPa at all the investigated temperatures. The increase in n_{ES} during this structure transition can cause a continuous increase in K_S up to 336–394 GPa. As a result, to achieve a ν greater than all the metallic glasses may require high-temperature treatment of SiO_2 -glass above 72 GPa. Application of higher temperature can also help achieve a greater ν at a relatively lower pressure.

Unlike ν , the densification process at high P–T causes a dramatic increase in the elastic moduli. As mentioned above, the increase of K_S in compression at all the investigated temperatures for SiO_2 -glass is mainly controlled by the increase in n_{ES} , while the change in G and E is dominated by the increase in the fraction of SiO_6 -octahedra

(Fig. 5). After the SiO₂-glass transitions to the “stishovite-like” local structure with $n_{ES} \approx 2$, its K_S , G , and E values have been as large as 361(9) GPa, 135(3) GPa, and 360(8) GPa at 59 GPa and 1000 K, respectively. These elastic moduli of SiO₂-glass at 59 GPa and 1000 K are two to four times greater than those of the metallic glasses and other oxide glasses. For comparison, GeO₂-glass follows a similar structure transition to SiO₂-glass up to ~ 11 GPa at 300 K (Fig. S5). Yet GeO₂-glass dominantly by the GeO₆-octahedra only has a bulk modulus of 256 GPa at 60 GPa and 300 K, which is much smaller than the K_S of 417(14) GPa for SiO₂-glass at the same P–T condition. Treatment of SiO₂-glass at extreme environments can thus generate a permanently densified glass which is highly hard, rigid, and stiff. Even after the high P–T quench to ambient conditions, the densified SiO₂-glass can partially retain a high network connectivity and has a ν value of 0.25, which is 55% greater than that of SiO₂-glass counterpart at ambient and SiO₂-glass doped Na, Al, and K as the additive^{10,12–15}.

In summary, we have measured the compressional- (V_p) and shear-wave (V_s) velocities of SiO₂-glass at simultaneously high P–T conditions. Hot-compression to 59 GPa and 1000 K produced a SiO₂-glass with an ultra-high elasticity, and the structure of SiO₂-glass during compression highly relies on the thermal path. At 59 GPa and 1000 K, SiO₂-glass has K_S 10 times greater than that at ambient conditions, while its G is 5 time greater. Additionally, the Poisson’s ratio, ν which is an indicator for the packing efficiency, of the silica glass at 59 GPa and 1000 K is comparable to the metallic glasses. On the other hand, this dense SiO₂-glass with ultra-high elasticity can be partially retained after decompression to the ambient conditions. The high P–T quenched SiO₂-glass has ν as great as 0.248(11). Treatment the SiO₂-glass at extreme environments is thus an effective way to increase its elasticity and enhance its mechanical performance.

Methods

Experimental details. SiO₂-glass used in the experiments was purchased from Kejing, Hefei Company in China. It was produced by burning SiCl₄ with O₂ at 2100 °C and then quickly quenched to 300 K. The purity of the SiO₂-glass was confirmed by electron microprobe to be 99.97% at the Material center of the University of Science and Technology of China. Measured Raman spectra of SiO₂-glass at ambient condition were consistent with previous experimental results (Fig. S6)^{11,23,26,56}. High P–T experiments were performed using the BX90 externally-heated diamond anvil cells (EHDACs) with 400 and 300- μ m culet diamonds at the GeoSoilEnviro Center for Advanced Radiation Sources (GSECARS) of the Advanced Photon Source (APS), Argonne National Laboratory (ANL) and at the High-Pressure Mineral Physics Laboratory, University of Science and Technology of China (USTC)⁵⁷. A piece of the SiO₂-glass of a few hundred μ m big was double-side polished into ~ 15- μ m thick platelet. The polished platelet was cut into small pieces in a diameter of ~ 100 μ m for 300- μ m culet cells and ~ 120 μ m for 400- μ m culet cells. Two Pt foils were placed near the sample and separated by 90° as the pressure calibrant⁵⁸. A ruby sphere was also loaded into the EHDAC as the pressure indicator during gas loading. Ne or Ar was loaded into the EHDAC at room temperature using the gas loading system at the University of Science and Technology of China as the pressure medium. The sound velocities were collected upon compression at a constant temperature (300 K, 750 K and 1000 K) (Figs. 1 and S1). For each EHDAC, we firstly increased the pressure to 4–5 GPa and then started heating, because heating the EHDAC at pressures lower than 4 GPa can easily lead to Ar or Ne leaking. Once we reached the target temperature, we waited for 20–30 min for the sample chamber’s pressure to become stable before we measured the pressure at this temperature. The sample was compressed up to 72 GPa at 300 K, while up to ~ 63 GPa at 750 K and 59 GPa at 1000 K. All the samples were quenched to 300 K after the high P–T measurements, and the velocities of the quenched sample were measured during decompression to 1 bar (Fig. S1).

Brillouin spectra were collected in a forward scattering geometry with an external scattering angle of 50° using a six-pass Sandercock tandem Fabry–Perot interferometer at GSECARS and 49.2° at High-Pressure Mineral Physics Laboratory, USTC (Fig. 1). The acoustic velocities (v) were calculated using the measured Brillouin frequency shift, $\Delta\nu_B$, in the following equation:

$$V = \frac{\Delta\nu_B \lambda_0}{2 \sin(\theta/2)} \quad (3)$$

where λ_0 is the incident laser wavelength (532 nm), and θ is the external scattering angle. The XRD patterns of Pt were collected both before and after Brillouin measurements at GSECARS of APS, ANL. XRD patterns of glass were also collected at 1000 K and 22 GPa to ensure the amorphous state of the sample (Fig. S8). The sample was also rotated by 90° to confirm its isotropy. The compression data at 300 K and all the decompression data were collected at the High-Pressure Mineral Physics Laboratory, USTC. Pressures of these data were determined by the Ruby fluorescence. Here, we used the Ruby pressure scale of Dewaele et al. (2004)⁵⁹ which is most consistent with the metal pressure scale of Pt in Fei et al. (2007)⁵⁸. The deviatoric stress inside the EHDAC at high P–T conditions was evaluated using the XRD peaks of Pt, which is less than 0.5 GPa (Fig. S7). Using Ar or Ne as the pressure medium can maintain a good quasi-hydrostatic experimental environment inside the sample chamber at high pressures and 300 K⁶⁰. The deviatoric stress for our high-pressure and 300 K measurements was estimated to be less than 0.8 GPa up to 65 GPa.

We have also performed Raman measurements of the glass to characterize its properties at high P–T conditions. The collected Raman spectra up to 24 GPa at 300 K were in good agreement with literature results (Fig. S5)^{11,23,26,56}. Below 6 GPa, the Raman spectra were characterized by a broad band between 188 and 554 cm^{–1} which turned into a sharp peak at 6.9 GPa with the wavenumber of 532 cm^{–1}. The change in the Raman spectra below 7 GPa could be associated with the increase in the Si–O bonding ionicity. Above 14.8 GPa, a new peak appeared at 631 cm^{–1} as the shoulder of the main peak at 580 cm^{–1}. The presence of this new peak should be caused by the change in the inter-polyhedra connectivity from purely corner-sharing to more complex connectivity. Above 25 GPa when the fraction of SiO₆-octahedra starts to increase with a jump, Raman signal was lost.

At 750 K, the variation of the Raman spectra with increasing pressure is the same as that at 300 K. We cannot detect the Raman signal of SiO₂-glass above 22 GPa at 750 K (Fig. S9).

Density calculation at high P–T conditions. Here, we followed the method of Petitgirard et al. (2017) to calculate the density of SiO₂-glass below 60 GPa at 300 K using a second-order polynomial function²⁹

$$\rho = -0.00053802 \times P^2 + 0.076204 \times P + 2.11 \quad (4)$$

where ρ is density, and P is pressure. For pressures above 60 GPa, the density was calculated using a third-order Birch–Murnaghan equation of state with the isothermal bulk modulus, $K_{T0} = 183$ GPa, pressure derivative of the bulk modulus, $K'_{T0} = 5$, and density at ambient conditions, $\rho_0 = 3.95$ g/cm³.

The effect of temperature on the density of SiO₂-glass was calculated following the method in Stixrude et al. (2005)⁶¹:

$$P = P_{T0} + P_{th} = P_{T0} + \gamma/\nu C_v(T - T_0) \quad (5)$$

and

$$\gamma/\nu C_v(T - T_0) = \alpha K_T(T - T_0) \quad (6)$$

where P_{T0} is pressure at reference temperature T_0 (300 K), P_{th} is the thermal pressure, γ is the Grüneisen parameter, ν is the Poisson's ratio, C_v is the constant molar heat capacity, α is the thermal expansion. Based on the curves of de Koker et al. (2009) for SiO₂ liquid, the value of αK_T is assumed to be independent of temperature.

Data availability

Experimental data are listed in Tables S1 and S2. They can also be downloaded online (<https://zenodo.org/record/6540997>).

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References

1. Mauro, J. C., Philip, C. S., Vaughn, D. J. & Pambianchi, M. S. Glass science in the United States: Current status and future directions. *Int. J. Appl. Glass Sci.* **5**, 2–15 (2014).
2. Wondraczek, L. et al. Towards ultrastrong glasses. *Adv. Mater.* **23**, 4578–4586 (2011).
3. Januchta, K. et al. Structural origin of high crack resistance in sodium aluminoborate glasses. *J. Non Cryst. Solids* **460**, 54–65 (2017).
4. Kapoor, S. et al. Permanent densification of calcium aluminophosphate glasses. *Front. Mater.* **6**, 63. <https://doi.org/10.3389/fmats.2019.00063> (2019).
5. Liu, A. Y. & Cohen, M. L. Prediction of new low compressibility solids. *Science* **245**, 841–842 (1989).
6. Novikov, V. N. & Sokolov, A. P. Poisson's ratio and the fragility of glass-forming liquids. *Nature* **431**, 961–963 (2004).
7. Smedskjaer, M. M., Mauro, J. C. & Yue, Y. Z. Prediction of glass hardness using temperature-dependent constraint theory. *Phys. Rev. Lett.* **105**, 115503 (2010).
8. Teter, D. M. Computational alchemy: The search for new superhard materials. *MRS Bull.* **23**, 22–27 (1998).
9. Ostergaard, M. B. et al. Revisiting the dependence of Poisson's ratio on liquid fragility and atomic packing density in oxide glasses. *Materials* **12**, 2439 (2019).
10. Rouxel, T. Elastic properties and short-to medium-range order in glasses. *J. Am. Ceram. Soc.* **90**, 3019–3039 (2007).
11. Rouxel, T., Ji, H., Hammouda, T. & Moreac, A. Poisson's ratio and the densification of glass under high pressure. *Phys. Rev. Lett.* **100**, 225501 (2008).
12. Bridge, B., Patel, N. D. & Waters, D. N. On the elastic-constants and structure of the pure inorganic oxide glasses. *Phys. Status Solidi A Appl. Res.* **77**, 655–668 (1983).
13. Rajendran, V., Begum, A. N., Azooz, M. A. & El Batal, F. H. Microstructural dependence on relevant physical–mechanical properties on SiO₂–Na₂O–CaO–P₂O₅ biological glasses. *Biomaterials* **23**, 4263–4275 (2002).
14. Askarpour, V., Manghni, M. H. & Richet, P. Elastic properties of diopside, anorthite, and grossular glasses and liquids—A Brillouin-scattering study up to 1400 K. *J. Geophys. Res.* **98**, 17683–17689 (1993).
15. Deriano, S., Rouxel, T., LeFloch, M. & Beuneu, B. Structure and mechanical properties of alkali-alkaline earth-silicate glasses. *Phys. Chem. Glasses* **45**, 37–44 (2004).
16. Schroeder, J., Mohr, R., Macedo, P. B. & Montrose, C. J. Rayleigh and Brillouin-scattering in K₂O–SiO₂ glasses. *J. Am. Ceram. Soc.* **56**, 510–514 (1973).
17. Soga, N. Elastic-moduli and fracture-toughness of glass. *J. Non Cryst. Solids* **73**, 305–313 (1985).
18. Johnson, J., Weber, R. & Grimsditch, M. Thermal and mechanical properties of rare earth aluminate and low-silica aluminosilicate optical glasses. *J. Non Cryst. Solids* **351**, 650–655 (2005).
19. Wang, W. H., Dong, C. & Shek, C. H. Bulk metallic glasses. *Mater. Sci. Eng. R* **44**, 45–89 (2004).
20. Yeganeh-Haeri, A., Ho, C. T., Weber, R., Diefenbacher, J. & McMillan, P. F. Elastic properties of aluminate glasses via Brillouin spectroscopy. *J. Non Cryst. Solids* **241**, 200–203 (1998).
21. Dyre, J. C. Colloquium: The glass transition and elastic models of glass-forming liquids. *Rev. Mod. Phys.* **78**, 953–972 (2006).
22. Mauro, J. C. & Smedskjaer, M. M. Minimalist landscape model of glass relaxation. *Phys. A Stat. Mech. Appl.* **391**, 3446–3459 (2012).
23. Champagnon, B. et al. High pressure elastic and plastic deformations of silica: In situ diamond anvil cell Raman experiments. *J. Non Cryst. Solids* **354**, 569–573 (2008).
24. Mackenzie, J. D. & Laforce, R. P. High-pressure densification of glass and effects of shear. *Nature* **197**, 480–481 (1963).
25. Martinet, C. et al. Permanently densified SiO₂ glasses: A structural approach. *J. Phys. Condens. Matter* **27**, 329501 (2015).
26. Hemley, R. J., Mao, H. K., Bell, P. M. & Mysen, B. O. Raman-spectroscopy of SiO₂ glass at high-pressure. *Phys. Rev. Lett.* **57**, 747–750 (1986).
27. Williams, Q. & Jeanloz, R. Spectroscopic evidence for pressure-induced coordination changes in silicate-glasses and melts. *Science* **239**, 902–905 (1988).
28. Murakami, M. & Bass, J. D. Spectroscopic evidence for ultrahigh-pressure polymorphism in SiO₂ glass. *Phys. Rev. Lett.* **104**, 025504. <https://doi.org/10.1103/Physrevlett.1104.025504> (2010).

29. Petitgirard, S. *et al.* SiO₂ glass density to lower-mantle pressures. *Phys. Rev. Lett.* **119**, 215701. <https://doi.org/10.1103/PhysRevLett.119.215701> (2017).
30. Prescher, C. *et al.* Beyond sixfold coordinated Si in SiO₂ glass at ultrahigh pressures. *Proc. Natl. Acad. Sci. U. S. A.* **114**, 10041–10046 (2017).
31. Sanloup, C. *et al.* Structural change in molten basalt at deep mantle conditions. *Nature* **503**, 104–109 (2013).
32. Mookherjee, M., Stixrude, L. & Karki, B. Hydrous silicate melt at high pressure. *Nature* **452**, 983–986 (2008).
33. Hasmy, A., Ispas, S. & Hehlen, B. Percolation transitions in compressed SiO₂ glasses. *Nature* **599**, 62–66 (2021).
34. Wu, M., Liang, Y. F., Jiang, J. Z. & Tse, J. S. Structure and properties of dense silica glass. *Sci. Rep.* **2**, 1–6. <https://doi.org/10.1038/Srep00398> (2012).
35. Smedskjaer, M. M., Rzoska, S. J., Bockowski, M. & Mauro, J. C. Mixed alkaline earth effect in the compressibility of aluminosilicate glasses. *J. Chem. Phys.* **140**, 054511 (2014).
36. Mackenzie, J. D. High-pressure effects on oxide glasses. 3. Densification in nonrigid state. *J. Am. Ceram. Soc.* **47**, 76–80 (1964).
37. Inamura, Y., Katayama, Y., Utsumi, W. & Funakoshi, K. Transformations in the intermediate-range structure of SiO₂ glass under high pressure and temperature. *Phys. Rev. Lett.* **93**, 015501. <https://doi.org/10.1103/PhysRevLett.1193.015501> (2004).
38. Lee, S. K. Effect of pressure on structure of oxide glasses at high pressure: Insights from solid-state NMR of quadrupolar nuclides. *Solid State Nucl. Magn.* **38**, 45–57 (2010).
39. Arndt, J. & Stoffler, D. Anomalous changes in some properties of silica glass densified at very high pressures. *Phys. Chem. Glasses* **10**, 117–124 (1969).
40. Deschamps, T., Margueritat, J., Martinet, C., Mermet, A. & Champagnon, B. Elastic moduli of permanently densified silica glasses. *Sci. Rep.* **4**, 1–7. <https://doi.org/10.1038/Srep07193> (2014).
41. Guerette, M. *et al.* Structure and properties of silica glass densified in cold compression and hot compression. *Sci. Rep.* **5**, 1–10. <https://doi.org/10.1038/srep15343> (2015).
42. Hofler, S. & Seifert, F. Volume relaxation of compacted SiO₂ glass—A model for the conservation of natural diaplectic glasses. *Earth Planet. Sci. Lett.* **67**, 433–438 (1984).
43. Poe, B. T., Romano, C. & Henderson, G. Raman and XANES spectroscopy of permanently densified vitreous silica. *J. Non Cryst. Solids* **341**, 162–169 (2004).
44. Mauro, J. C., Loucks, R. J. & Sen, S. Heat capacity, enthalpy fluctuations, and configurational entropy in broken ergodic systems. *J. Chem. Phys.* **133**, 164503 (2010).
45. Lee, S. K., Mosenfelder, J. L., Park, S. Y., Lee, A. C. & Asimow, P. D. Configurational entropy of basaltic melts in Earth's mantle. *Proc. Natl. Acad. Sci. U. S. A.* **117**, 21938–21944 (2020).
46. Zhang, Y. Y., Fu, S. Y., Wang, B. Y. & Lin, J. F. Elasticity of a pseudoproper ferroelastic transition from stishovite to post-stishovite at high pressure. *Phys. Rev. Lett.* **126**, 025701 (2021).
47. Ikuta, D. *et al.* First in situ X-ray identification of coesite and retrograde quartz on a glass thin section of an ultrahigh-pressure metamorphic rock and their crystal structure details. *Am. Mineral.* **92**, 57–63 (2007).
48. Ross, N. L., Shu, J. F., Hazen, R. M. & Gasparik, T. High-pressure crystal-chemistry of stishovite. *Am. Mineral.* **75**, 739–747 (1990).
49. Zha, C. S., Hemley, R. J., Mao, H. K., Duffy, T. S. & Meade, C. Acoustic velocities and refractive-index of SiO₂ glass to 57.5-GPa by Brillouin-scattering. *Phys. Rev. B* **50**, 13105–13112 (1994).
50. Weigel, C. *et al.* Pressure-induced densification of vitreous silica: Insight from elastic properties. *Phys. Rev. B* **100**, 094102 (2019).
51. Wang, J., Mao, Z., Jiang, F. M. & Duffy, T. S. Elasticity of single-crystal quartz to 10 GPa. *Phys. Chem. Miner.* **42**, 203–212 (2015).
52. Chen, T. *et al.* Anomalous elastic properties of coesite at high pressure and implications for the upper mantle X-discontinuity. *Earth Planet. Sci. Lett.* **412**, 42–51 (2015).
53. Ecolivet, C. & Verdier, P. Elastic properties and refraction indexes of nitrated glasses. *Mater. Res. Bull.* **19**, 227–231 (1984).
54. Zou, X. L. & Toratani, H. Compositional design of high modulus glasses for disk substrates. *J. Non Cryst. Solids* **290**, 180–188 (2001).
55. Peker, A. & Johnson, W. L. A highly processable metallic-glass—Zr_{41.2}Ti_{13.8}Cu_{12.5}Ni_{10.0}Be_{22.5}. *Appl. Phys. Lett.* **63**, 2342–2344 (1993).
56. Shen, G. Y. *et al.* Effect of helium on structure and compression behavior of SiO₂ glass. *Proc. Natl. Acad. Sci. U. S. A.* **108**, 6004–6007 (2011).
57. Kantor, I. *et al.* BX90: A new diamond anvil cell design for X-ray diffraction and optical measurements. *Rev. Sci. Instrum.* **83**, 125102 (2012).
58. Fei, Y. W. *et al.* Toward an internally consistent pressure scale. *Proc. Natl. Acad. Sci. U. S. A.* **104**, 9182–9186 (2007).
59. Dewaele, A., Loubeyre, P. & Mezouar, M. Equations of state of six metals above 94 GPa. *Phys. Rev. B* **70**, 094112 (2004).
60. Shim, S. H., Jeanloz, R. & Duffy, T. S. Tetragonal structure of CaSiO₃ perovskite above 20 GPa. *Geophys. Res. Lett.* **29**, 19-1-19-4. <https://doi.org/10.1029/2002GL016148> (2002).
61. Stixrude, L. & Karki, B. Structure and freezing of MgSiO₃ liquid in Earth's lower mantle. *Science* **310**, 297–299 (2005).

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Author contributions

Z.M. proposed and directed the project. N.S. performed the experiments. S.T. assisted the experiments. N.S. and X.Z. analyzed the experimental data. Z.M. and J.L. presented the physical implication. N.S., and Z.M., wrote the paper with inputs from everyone.

Competing interests

The authors declare no competing interests.

Additional information

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