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OPEN Potential of dredged bioremediated marine sediment for strawberry cultivation

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For the maintenance of the economic activity of the ports, it is necessary to dredge the marine sediments in order to guarantee their depth. These sediments, considered by European legislation as residues, present relevant limitations of use and generate environmental and economic problems concerning their final disposal. In this context, the present work aims to identify the phytoremediated dredged sediments potential as an alternative to the traditional substrate (peat) in horticultural growing through two-years controlled strawberry cultivation. The growing media mixes used were: (1) 100% peat (Pt) as a control substrate; (2) 100% dredged remediated sediment (DRS); (3) 50% each (Pt-DRS). The dredged sediment, plant drainage and strawberry plant parts (leaves, stems, roots, and fruits) were analyzed to mineral elements, heavy metal contents, and pesticide residues (polycyclic aromatic hydrocarbons, polychlorinated biphenyls and specific fumigants) during the experimental period. Only seven (Mn, Fe, Zn, Mo, Al, Mn and Ni) of the twenty-two metals and two (nitrates and fluorene) of the six hundred-thirteen pesticides analyzed were detected in the strawberry fruits. In all the cases, values detected were under the Spanish and European legal limit. The suitability of strawberry fruits for fresh and/or processed consumption with no risk was confirmed. Based on the results, can be affirmed that the dredged remediated sediment can be used as a culture substrate, alone or mixed with other substrates. Additional researches should be carried out to confirm the sediment characteristics and compare with other substrates to improve the physical and chemical properties.

The worldwide importance and economic influence of seaports are indisputable. However, to do this activity it is necessary to periodically dredge the marine sediments to guarantee the safe passage of large-draft boats, allowing preserve both traffic and port competitiveness¹.

Being an economical necessary activity, and due to the large volumes of marine sediments generated annually in both European Union seaports (around 200 million m³ every year)² and the rest world seaports, in recent years the need to investigate new options and alternative uses of dredged sediments have been identified to minimize the derived environmental impacts³. These impacts are mainly focused on: (1) changes in the physical, chemical and biological characteristics of marine and coastal ecosystems due to the movement of sediments⁴; (2) impacts derived from the transport and disposal of sediments in suitable places; and (3) specific requirements for management, as the sediments may contain pollutants from local anthropogenic activities and / or river contributions⁵. This last point has motivated current international legislation to consider dredged sediments as contaminated waste, limiting their use, application and disposal. However, specific studies, such as developed by Buceta et al.³ identified the "fine granular size" that corresponds to less than 10% in the investigated sediment, as the part generally contaminated by metals and organic pollutants. Therefore, it can be assumed that in many cases, and for the most part, these sediments could be considered a resource rather than a residue.

Peat has been considered the most used substrate in the last decade due to its physical-chemical characteristics and suitability for many species and cultivation systems⁶⁻⁸. Nevertheless, strong demand and unsustainable exploration of peatlands around the world is creating availability problems and signs of depletion. For this reason, it is increasingly necessary to identify and study new substrates that make it possible to reduce, or even replace, the use of peat in agriculture⁹⁻¹². Mattei et al.¹³ demonstrated the viability of phytoremediation dredged

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sediment as a substrate for plant growth. The term phytoremediation is derived from the Greek word "Phyto" which means plant and "Remedy" which is "equilibrium recovery", normally, it can be defined as a decontamination process through the use of green plants due to their ability to retain, metabolize and/or accumulate pollutants in water, soil and/or air through accumulation, immobilization, rhizodegradation mechanisms among others¹⁴. Phytoremediation is considered the most profitable and environmentally friendly soil recovery strategy¹⁵ that is in continuous development and improvement^{16,17}.

In this context, the present work focuses on the study of the phytoremediated dredged sediments potential and suitability as an alternative to the traditional substrate (peat) in horticultural growing media by monitoring the content of mineral elements, metals and pesticide residues in strawberry plant and fruits (*Fragaria x ananassa* Duch.), Variety Monterrey. Thus, the main objective of the research was to provide a sustainable alternative use and environmental appropriate management of dredged marine sediments that are currently considered as a major economic, landscape and environmental problem. Besides, the authors intended that the results of this study contribute to reducing the pressure on natural peatlands, guaranteeing their long-term permanence.

The present work started on the hypothesis that properly treated, dredged marine sediments may cease to be an economic and environmental issue, since they present potential for use as a safe agricultural substrate, and may even be used for other types of uses such as soil regeneration, recovery quarries and/or roads.

Materials and methods

Dredged remediated sediments. The dredged remediated sediments (DRS) used in this study came from Leghorn Port (Italy) and were previously phytoremediated for 3 years (**European Project AGRIPORT** - *Agricultural Reuse of Polluted dredged Sediments - ECO/08/239,065/S12.53226*). Before starting the experimental assays with strawberry plants, physical (apparent soil density) and chemical (electrical conductivity, pH, cation exchange capacity, total organic C and N, water-soluble and total removable carbon, humic substance fractions, P, Zn, Cd, Ni, Cu, Cr, Pb, Al, Mn, Fe, V, Hg, polychlorobiphenyls and polycyclic aromatic hydrocarbons) analyses were carried out (Table 1). DRS was used in a previous strawberry experiment conducted by the same research group¹⁸.

In addition, biochemical (dehydrogenase activity, hydrolytic enzyme activities, β -glucosidase, acid phosphatase, protease, phosphodiesterase activity, arylesterase activity, arylsulfatase activity, cellulase activity, protease activity, urease activity and dioxygenase activity) analyses were carried out (data pending publication). The soil toxicity was also evaluated according to the ISO standard method (ISO 11,348–3 1998)¹⁹.

After phytoremediation, the dredged sediment presented good nutrient content, good biological activity and a low level of contamination, with only a small residual organic contamination. The values complied with the Spanish legislation on substrates for agricultural uses²⁰.

Plant material and experimental design. The experimental trial was carried out during two seasons (2016 and 2017) in an experimental plot of the School of Engineering of Orihuela (Miguel Hernández University) located in Orihuela, SE Spain (38°04′N, 0°58′W, 26 m above sea level).

Five strawberry plants (*Fragaria x ananassa* Duch.), Monterrey cultivar, were planted in rectangular containers (150 cm \times 30 cm \times 15 cm) covered at the initial grow time with black polythene plastic. Each container was provided with a drainage system to collect excess water and prevent waterlogging. The entire assay was made in triplicate, i.e. three containers (10 plants each) were used for each substrate tested in a randomized complete block design (RCBD). In this sense, dredged remediated sediments and peat mixed in three different proportions were used as substrates: 100% peat (Pt) as a control, 100% dredged remediated sediments (DRS), and 50% each (Pt-DRS). Every year of this study, 90 strawberry plants were used. This same experimental design was used successfully in previous and parallel studies developed by the same research group¹⁸.

Crop irrigation and nutritional and microelements additives were scheduled according to the methodology developed by the same research group and previously published¹⁸.

Plant drainage and plant material characterization. The plant drainages were collected and measured daily. The pH and electrical conductivity were determined with a multiparameter analyzer (CONSORT C860 multiparameter analyzer). For chemical analysis (macro- and microelement concentrations, heavy metals and pesticide residues), aliquots of the drainage were separated in plastic bottles and stored in a refrigerator at 5 °C until used.

At the end of the cultivation, diverse plant parts (roots, stems and leaves) from the different treatments were sampled. All plant samples were analyzed to determine the heavy metal contents (manganese, iron, zinc, copper, molybdenum, lead, cadmium, nickel, chromium, mercury, cobalt, antimony, arsenic, selenium, aluminium, beryllium, tin, strontium, silver, thallium and vanadium) and the pesticide residues as explained below. The results were compared with the specific legal limits both in Europe and Spain.

Mineral element and heavy metal determination. The mineral elements were determined in aqueous solution samples. Nitrate, chloride, sulfate, nitrate, bromide and fluoride anions were determined by an ion chromatograph (883 COMPACT IC-PLUS METROHM) equipped with chemical suppression. Nitrates in the plant matrices were determined by water extraction of nitrate anions. A sample exchanger for pH and conductivity measurements (814 USB Sample Processor, METROHM) was used to determine the carbonates and bicarbonates. The results were processed with TIAMO 2.3. METROHM software. Ammoniacal nitrogen was determined by spectrophotometry (VARIAN CARY 50 UV-vis spectrophotometer) following the 4500-NH3 Phenate Method described by Standard Methods for the examination of water and wastewater²¹.

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Hydrocarbons C > 12 (mg kg ⁻¹) 207 ± 3.8 Hydrocarbons C < 12 (mg kg ⁻¹) $<$ L.Q PAHs (mg kg ⁻¹) 49.2 ± 5.3 PCB (mg kg ⁻¹) 0.039 ± 0.002 Cd _{total} (mg kg ⁻¹) 0.39 ± 0.002 Cd _{total} (mg kg ⁻¹) 378 ± 28 Mn _{total} (mg kg ⁻¹) 378 ± 28 Mn _{available} (mg kg ⁻¹) 13.8 ± 1.9 Al _{total} (mg kg ⁻¹) 22.744 ± 2319 Al _{available} (mg kg ⁻¹) 23.073 ± 1676 Fe _{total} (mg kg ⁻¹) 23.073 ± 1676 Fe _{total} (mg kg ⁻¹) 30.8 ± 3.7 Hg _{total} (mg kg ⁻¹) 0.075 ± 0.001 Cr total (mg kg ⁻¹) 34.6 ± 5.33 Cu _{total} (mg kg ⁻¹) 34.3 ± 4.3 Cu _{total} (mg kg ⁻¹) 17.4 ± 2.5 Cu _{total} (mg kg ⁻¹) 248 ± 11 Zn _{total} (mg kg ⁻¹) 37.2 ± 3.3 Pb _{total} (mg kg ⁻¹) 35.2 ± 3.7	P ₂ O ₅ (%)	0.11±0.02
Hydrocarbons C < 12 (mg kg ⁻¹) <l.q< td=""> PAHs (mg kg⁻¹) 49.2 ± 5.3 PCB (mg kg⁻¹) 0.039 ± 0.002 Cd_{total} (mg kg⁻¹) <l.q< td=""> Mn_{total} (mg kg⁻¹) 378 ± 28 Mn_{available} (mg kg⁻¹) 13.8 ± 1.9 Al_{total} (mg kg⁻¹) 22.744 ± 2319 Al_{available} (mg kg⁻¹) 3.24 ± 0.13 Fe_{total} (mg kg⁻¹) 23.073 ± 1676 Fe_{total} (mg kg⁻¹) 30.8 ± 3.7 Hg_{total} (mg kg⁻¹) 30.8 ± 3.7 Hg_{total} (mg kg⁻¹) 0.075 ± 0.001 Cr _{total} (mg kg⁻¹) 34.6 ± 5.33 Cu_{watlable} (mg kg⁻¹) 34.3 ± 4.3 Cu_{watlable} (mg kg⁻¹) 17.4 ± 2.5 Zn_{total} (mg kg⁻¹) 248 ± 11 Zn_{vatlable} (mg kg⁻¹) 37.2 ± 3.3</l.q<></l.q<>	P total (%)	495±33
PAHs (mg kg ⁻¹) 49.2 ± 5.3 PCB (mg kg ⁻¹) 0.039 ± 0.002 Cd _{total} (mg kg ⁻¹) $<$ L.Q Mn _{total} (mg kg ⁻¹) 378 ± 28 Mn _{available} (mg kg ⁻¹) 13.8 ± 1.9 Al _{total} (mg kg ⁻¹) 22.744 ± 2319 Al _{available} (mg kg ⁻¹) 3.24 ± 0.13 Fe _{total} (mg kg ⁻¹) 3.24 ± 0.13 Se _{vailable} (mg kg ⁻¹) 3.24 ± 0.13 V _{total} (mg kg ⁻¹) 3.24 ± 0.13 Fe _{available} (mg kg ⁻¹) 3.24 ± 0.13 Se _{vailable} (mg kg ⁻¹) 3.24 ± 0.13 Cu _{tal} (mg kg ⁻¹) 3.24 ± 0.13 Se _{vailable} (mg kg ⁻¹) 3.24 ± 0.13 Se _{vailable} (mg kg ⁻¹) 3.24 ± 0.13 Cu _{tal} (mg kg ⁻¹) 3.24 ± 0.13 Cu _{total} (mg kg ⁻¹) 3.8 ± 3.7 Hg _{total} (mg kg ⁻¹) 34.6 ± 5.33 Cu _{total} (mg kg ⁻¹) 34.3 ± 4.3 Cu _{total} (mg kg ⁻¹) 248 ± 11 Zn _{total} (mg kg ⁻¹) 37.2 ± 3.3 Pb _{total} (mg kg ⁻¹) 35.2 ± 3.7	Hydrocarbons C>12 (mg kg ⁻¹)	207±3.8
PCB (mg kg ⁻¹) 0.039 ± 0.002 Cd total (mg kg ⁻¹) $<$ L.Q Mn _{total} (mg kg ⁻¹) 378 ± 28 Mn _{available} (mg kg ⁻¹) 13.8 ± 1.9 Al _{total} (mg kg ⁻¹) 22.744 ± 2319 Al _{available} (mg kg ⁻¹) 3.24 ± 0.13 Se _{total} (mg kg ⁻¹) 3.24 ± 0.13 Fe _{total} (mg kg ⁻¹) 3.24 ± 0.13 V _{total} (mg kg ⁻¹) 3.24 ± 0.13 Se _{total} (mg kg ⁻¹) 3.24 ± 0.13 Cu _{available} (mg kg ⁻¹) 3.24 ± 0.13 Se _{total} (mg kg ⁻¹) 3.073 ± 1676 Fe avaitable (mg kg ⁻¹) 0.075 ± 0.001 Cr total (mg kg ⁻¹) 0.075 ± 0.001 Cr total (mg kg ⁻¹) 34.6 ± 5.33 Cu _{total} (mg kg ⁻¹) 34.3 ± 4.3 Cu _{total} (mg kg ⁻¹) 17.4 ± 2.5 Zn _{total} (mg kg ⁻¹) 248 ± 11 Zn _{total} (mg kg ⁻¹) 37.2 ± 3.3 Pb _{total} (mg kg ⁻¹) 35.2 ± 3.7	Hydrocarbons C < 12 (mg kg ⁻¹)	<l.q< td=""></l.q<>
Cd_{total} (mg kg ⁻¹) <l.q< td=""> Mn_{total} (mg kg⁻¹) 378 ± 28 $Mn_{available}$ (mg kg⁻¹) 13.8 ± 1.9 Al_{total} (mg kg⁻¹) 22.744 ± 2319 $Al_{available}$ (mg kg⁻¹) 3.24 ± 0.13 Fe_{total} (mg kg⁻¹) 23.073 ± 1676 Fe_{total} (mg kg⁻¹) 23.073 ± 1676 $Fe_{available}$ (mg kg⁻¹) 30.8 ± 3.7 M_{total} (mg kg⁻¹) 0.075 ± 0.001 Cr_{total} (mg kg⁻¹) 54.3 ± 1.2 Ni total (mg kg⁻¹) 34.6 ± 5.33 Cu_{total} (mg kg⁻¹) 17.4 ± 2.5 Zn_{total} (mg kg⁻¹) 248 ± 11 Zn_{total} (mg kg⁻¹) 37.2 ± 3.3 Pb_{total} (mg kg⁻¹) 35.2 ± 3.7</l.q<>	PAHs (mg kg ⁻¹)	49.2±5.3
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	PCB (mg kg ⁻¹)	0.039 ± 0.002
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Cd _{total} (mg kg ⁻¹)	<l.q< td=""></l.q<>
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Mn _{total} (mg kg ⁻¹)	378±28
$Al_{available}$ (mg kg ⁻¹) 3.24 ± 0.13 $Al_{available}$ (mg kg ⁻¹) 23.073 ± 1676 Fe_{total} (mg kg ⁻¹) 17.8 ± 3.3 V_{total} (mg kg ⁻¹) 30.8 ± 3.7 Hg_{total} (mg kg ⁻¹) 0.075 ± 0.001 Cr_{total} (mg kg ⁻¹) 54.3 ± 1.2 Ni total (mg kg ⁻¹) 34.6 ± 5.33 Cu_{total} (mg kg ⁻¹) 17.4 ± 2.5 $Cu_{available}$ (mg kg ⁻¹) 248 ± 11 $Zn_{available}$ (mg kg ⁻¹) 37.2 ± 3.3 Pb_{total} (mg kg ⁻¹) 35.2 ± 3.7	Mn _{available} (mg kg ⁻¹)	13.8±1.9
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Al _{total} (mg kg ⁻¹)	22.744±2319
$\begin{array}{llllllllllllllllllllllllllllllllllll$	Al _{available} (mg kg ⁻¹)	3.24±0.13
W_{total} (mg kg ⁻¹) 30.8 ± 3.7 Hg_{total} (mg kg ⁻¹) 0.075 ± 0.001 Cr_{total} (mg kg ⁻¹) 54.3 ± 1.2 Ni_{total} (mg kg ⁻¹) 34.6 ± 5.33 Cu_{total} (mg kg ⁻¹) 34.3 ± 4.3 $Cu_{available}$ (mg kg ⁻¹) 17.4 ± 2.5 Zn_{total} (mg kg ⁻¹) 248 ± 11 $Zn_{available}$ (mg kg ⁻¹) 37.2 ± 3.3 Pb_{total} (mg kg ⁻¹) 35.2 ± 3.7	Fe _{total} (mg kg ⁻¹)	23.073±1676
Hg _{total} (mg kg ⁻¹) 0.075 ± 0.001 Cr total (mg kg ⁻¹) 54.3 ± 1.2 Ni total (mg kg ⁻¹) 34.6 ± 5.33 Cu _{total} (mg kg ⁻¹) 34.3 ± 4.3 Cu _{available} (mg kg ⁻¹) 17.4 ± 2.5 Zn _{total} (mg kg ⁻¹) 248 ± 11 Zn _{available} (mg kg ⁻¹) 37.2 ± 3.3 Pb _{total} (mg kg ⁻¹) 35.2 ± 3.7	Fe _{available} (mg kg ⁻¹)	17.8±3.3
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	V _{total} (mg kg ⁻¹)	30.8±3.7
Ni $total$ (mg kg ⁻¹) 34.6 ± 5.33 Cu md_{gg} (mg kg ⁻¹) 34.3 ± 4.3 Cu md_{gg} (mg kg ⁻¹) 17.4 ± 2.5 Zn md_{gg} (mg kg ⁻¹) 248 ± 11 Zn md_{gg} (mg kg ⁻¹) 37.2 ± 3.3 Pb $total$ (mg kg ⁻¹) 35.2 ± 3.7	Hg _{total} (mg kg ⁻¹)	0.075 ± 0.001
$\begin{array}{c} \text{Cu}_{total} \ (\text{mg kg}^{-1}) & 34.3 \pm 4.3 \\ \text{Cu}_{available} \ (\text{mg kg}^{-1}) & 17.4 \pm 2.5 \\ \text{Zn}_{total} \ (\text{mg kg}^{-1}) & 248 \pm 11 \\ \text{Zn}_{available} \ (\text{mg kg}^{-1}) & 37.2 \pm 3.3 \\ \text{Pb}_{total} \ (\text{mg kg}^{-1}) & 35.2 \pm 3.7 \end{array}$	Cr total (mg kg-1)	54.3±1.2
Cu _{available} (mg kg ⁻¹) 17.4 ± 2.5 Zn _{total} (mg kg ⁻¹) 248 ± 11 Zn _{available} (mg kg ⁻¹) 37.2 ± 3.3 $2b_{total}$ (mg kg ⁻¹) 35.2 ± 3.7	Ni total (mg kg-1)	34.6±5.33
$Zn_{total} (mg kg^{-1})$ 248 ± 11 $Zn_{available} (mg kg^{-1})$ 37.2 ± 3.3 $Pb_{total} (mg kg^{-1})$ 35.2 ± 3.7	Cu _{total} (mg kg ⁻¹)	34.3±4.3
$Zn_{available}$ (mg kg ⁻¹) 37.2 ± 3.3 Pb_{total} (mg kg ⁻¹) 35.2 ± 3.7	Cu _{available} (mg kg ⁻¹)	17.4±2.5
$Pb_{total} (mg kg^{-1})$ 35.2±3.7	Zn _{total} (mg kg ⁻¹)	248±11
	Zn _{available} (mg kg ⁻¹)	37.2±3.3
$Pb_{available} (mg kg^{-1})$ 7.55 ± 0.74	Pb _{total} (mg kg ⁻¹)	35.2±3.7
	Pb _{available} (mg kg ⁻¹)	7.55 ± 0.74

Table 1. Initial DRS characterization. L.Q: Limit of quantification.

Heavy metal determination was carried out by ICP-MS (mass spectrometry with inductively coupled plasma) using AGILENT 7700-E and AGILENT 7700-X models. For the digestions, MILESTONE ETHOS and MILE-STONE ETHOS ONE microwaves were used.

Pesticide residue determination. The six hundred and thirteen different pesticides were determined according to the UNE-EN 1562: 2019 method by liquid chromatography with a triple quadrupole mass detector (LC–MS/QqQ, Bruker Evoq Elite) based analysis following acetonitrile (>99.9% SIGMA ALDRICH) extraction/partition and clean-up by dispersive solid-phase extraction (SPE – OASIS PRIME HLB 3cc/150 mg. WATERS MIL-FORD). The pesticide residues were identified by retention time comparison with standard calibration solutions (>99.7% SIGMA ALDRICH) and also by spectral libraries specific to the detected compounds. Besides, polychlorinated biphenyl (PCB) and polycyclic aromatic hydrocarbons (PAH) were analysed according to EPA 1668C 2010 and EPA 8015C 2007 methods, respectively, by gas chromatography with triple quadrupole mass spectrometry detector (GC–MS/QqQ Bruker Scion TQ model). All the pesticide, PAHs and PCBs tests were carried out in triplicate by an accredited Spanish laboratory (KUDAM Laboratory, Alicante).

Statistical analysis. Statistical analyses were performed using SPSS 24.0 for Windows (SPSS SCIENCE, Chicago, IL, USA). Basic descriptive statistical analysis was followed by an analysis of variance (ANOVA) test for mean comparisons. The method used to discriminate among the means (multiple range test) was Tukey's test at a 95.0% confidence level. Pearson correlation analyses were also performed.

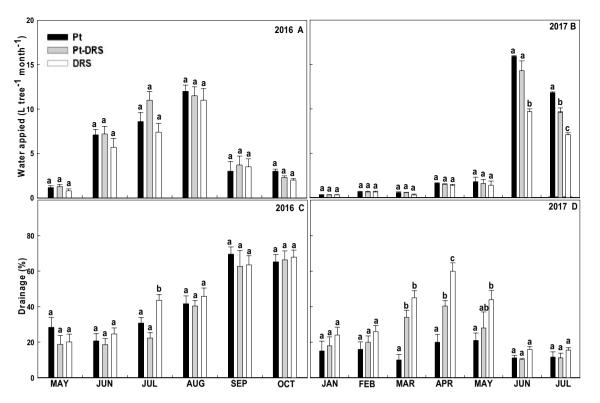


Figure 1. Monthly average of water applied in 2016 (**A**) and 2017 (**B**) and percentage of drainage in 2016 (**C**) and 2017 (**D**) for Monterrey strawberry cultivar. Each bar corresponds to the monthly average relative to each substrate, where Pt corresponds to peat; Pt-DRS 50% mix of peat and dredged sediment; and DRS dredged sediment. Different letters on top of the bars indicate significant differences according to Tukey's HSD test (p < 0.05).

Results

Plant drainage characterization. The strawberry plants were transplanted to the cultivation benches in May 2016. From this date until May 2017, the amount of water applied to the three substrates (Pt, Pt-DRS, and DRS) was kept essentially the same, but during the summer, the months during the year with greater evapotranspiration, more water was applied to Pt, followed by Pt-DRS and DRS. This difference may be due to the different textures, and therefore, different water retention capacities, of the substrates (Fig. 1A,B). Concerning the drainage volume (Fig. 1C,D), in general, no significant differences in drainage (%) were observed among the different substrates/months, except for the Pt-DRS drainage between March and May 2017, which was significantly higher (Fig. 1D). In volume terms, a constant tendency was identified with some specific variations, so in September and October 2016, drainages close to 60% were reached, while in 2017, the drainage percentage was approximately 20–30% for all substrates except Pt-DRS. This temporary increase in drained volume could have caused washing and dilution of salts and, therefore, a decrease in water drainage electrical conductivity.

Electrical conductivity and pH. The average monthly variations in plant drainage electrical conductivity (EC) and pH are shown in Fig. 2. Generally, the EC values were higher in 2016 than in 2017, reaching values above 4 dS m^{-1} (June 2016). In 2017, EC values close to 3 dS m^{-1} were maintained. On the other hand, in 2016, the Pt pH showed values slightly lower than those in the other two treatments (approximately 8), but in 2017, the average pH values of the three treatments were equalized at approximately 8, and no significant differences were observed (Fig. 2D). This was possibly due to progressive alkalization of the peat over the 2 years of the trial.

Macro- and microelements. The macro- and microelement concentrations were higher during the first year of the trial (2016) than during the second year (2017), probably due to the washing effect produced by irrigation (Table 2). The highest concentrations were observed for DRS drainage. However, no significant differences were observed for carbonates and bicarbonates. Ammoniacal nitrogen, manganese, iron and copper presented higher values during 2017 (Table 2). Macro- and micronutrient analyses were only performed with DRS because it was the study sediment; Pt is already commercially consolidated, and the Pt-DRS values were predictably lower than the DRS values.

Heavy metal contents. The heavy metal contents in the DRS drainage for 2016 and 2017 were below the maximum permissible level (MPL), as shown in Table 3. In addition, the sums of the fractions analyzed were considerably lower than the MPL, which means that the results were in accordance with the maximum legal limits.

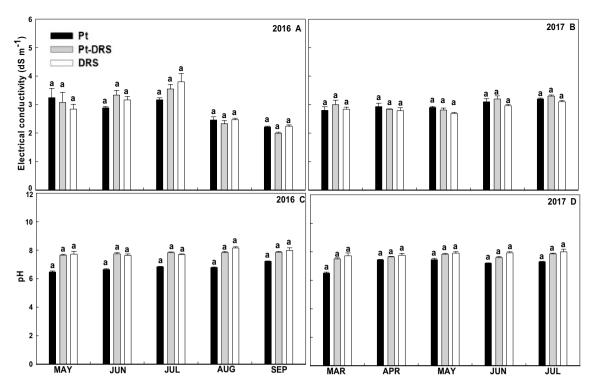


Figure 2. Monthly average electrical conductivity in 2016 (**A**) and 2017 (**B**) and pH in 2016 (**C**) and 2017 (**D**) for Monterrey strawberry cultivar. Each bar corresponds to the monthly average relative to each substrate, where Pt corresponds to peat; Pt-DRS 50% peat and dredged sediment mix; and DRS dredged sediment. Different letters on top of the bars indicate significant differences according to Tukey's HSD test (p < 0.05).

	2016	2017
Element (mg L ⁻¹)	2010	2017
Sodium (Na ⁺)	117	105
Potassium (K ⁺)	310	139
Calcium (Ca ²⁺)	198	142
Magnesium (Mg ²⁺)	49.4	35.5
Boron (B ³⁺)	0.791	0.636
Chlorides (Cl ⁻)	214	180
Sulfates (SO ₄ ^{2–})	310	259
Carbonates (CO ₃ ²⁻)	< 5.00	< 5.00
Bicarbonates (HCO ₃ ⁻)	105	108
Nitrates (NO ₃ ⁻)	681	346
Ammoniacal nitrogen (NH4+)	0.187	1.85
Phosphates (H ₂ PO ₄ ⁻)	56.4	26
Element (µg L ⁻¹)		1
Manganese (Mn)	103	180
Iron (Fe)	457	3970
Zinc (Zn)	774	435
Copper (Cu)	127	158
Molybdenum (Mo)	74.4	50.9

 Table 2.
 Drainage macro and microelements analysis. Sediments 100% (DRS).

Therefore, during the first trial year (total content of 0.1156), only 3.85% of the allowed threshold was reached (Σ of fractions < 3), and this value increased to 7.25% during the second trial year (total content of 0.2173). The values were in accordance with Spanish and European legal provisions^{22,23}.

Pesticide quantification. During the assay, 613 pesticide residues were analyzed. The pesticide residues detected in the DRS drainage are shown in Table 4. The full list of pesticides can be found in the *supplementary material*

	2016 2017 MPL			Fraction		
Element	$(\mu g L^{-1})$		2016	2017		
Arsenic (As)	ND	ND	ND	ND	ND	
Cadmium (Cd)	12.1	14.4	500	0.0242	0.0288	
Chromium (Cr)	<10.00	<10.00	500	0.02	0.02	
Nickel (Ni)	24	10.2	10,000	0.0024	0.00102	
Mercury (Hg)	< 0.200	12.6	100	0.002	0.126	
Lead (Pb)	7.8	< 2.00	500	0.0156	0.004	
Selenium (Se)	ND	ND	ND	ND	ND	
Copper (Cu)	127	158	10,000	0.0127	0.0158	
Zinc (Zn)	774	435	20	0.0387	0.02175	
			Σ total	0.1156	0.21737	
			MPL	3	3	

Table 3. Heavy metals contents and the sum of the fractions analysed in DRS drainage. MPL: MaximumPermissible Level (Established by Hydraulic Public Domain Regulation, approved by the Spanish Royal Decree849/1986); ND: Not Detected.

	Detected summary (mg L ⁻¹)			
Pesticide residues	2016	2017		
Acetochlor	0.01200	ND		
Anthraquinone	0.00014	ND		
Flutriafol	0.00006	ND		
Metalaxil (+ Metalaxil-M)	0.00020	ND		
Simazine	0.00053	ND		
Acenaftileno	ND	0.000024		
Phenanthrene	ND	0.000030		
Total amount	0.01293	0.000054		
MPL	0.05	0.05		

Table 4. SDR drainage pesticide quantification. MPL: Maximum Permissible Level (Established by HydraulicPublic Domain Regulation, approved by the Spanish Royal Decree 849/1986); ND: Not Detected.

attached to this paper. In general, five pesticides, acetochlor, anthraquinone, flutriafol, metalaxyl and simazine, were detected in 2016, while only two were present in 2017, acenaphthylene and phenanthrene. In all cases, the concentrations were below the Spanish MPL determined by the Hydraulic Public Domain Regulation²². Thus, in 2016, the sum of pesticide residues accounted for 25.86% of the MPL (0.05 mg L⁻¹), and this value fell to 0.11% in 2017 (Table 4). The results were also in accordance with European legal requirements²³.

Plant material characterization. *Heavy metal contents.* The heavy metal concentrations were determined in the different strawberry plant parts (root, stem, leaves and fruit) grown in Pt (as control) and in DRS (Tables 5, 6). The analyses were carried out during the two years of the experiment (2016 and 2017). The results for the studied substrates showed that for leaf, stem and root, the most abundant metals were Mn, Fe, Zn, Al, and Sr, independent of the year. Comparing the heavy metal concentrations in the leaf, stem and root, the root presented the highest values, which is logical since the root is the plant part that has direct contact with the substrate. The Zn, Cu Ni, Cr and Pb contents in the roots were above the maximal levels for food. From an alimentary perspective, since these plant parts are not used for food, heavy metal content could be considered not to be a problem. Even so, most of the metal contents were below the recommended limits for food^{24,25}.

In addition, for the strawberry fruits (Table 6), Mn, Fe, Zn, Mo, Al, Sr and Ni were the predominant metals in the two substrates, and Sn, Co, Sb, As, Be, Mn, Ag, Tl, V, Se, Al, Pb, Cd, Cr and Hg were below the detection limit. In no case did the metal contents reach the limits established by European Legislation²⁵⁻²⁹, by the Food and Agriculture Organization of the United Nations (FAO) and the World Health Organization (WHO) (1995 and later revisions), by the standards of the United States Environmental Protection Agency (USEPA) or by heavy metal legislation from Australia, New Zealand, Canada, and Brazil³⁰.

Pesticide quantification. Of the 613 pesticide residues analyzed (*see additional material*), only nitrates (101 mg kg⁻¹ Pt and 99.40 mg kg⁻¹ DRS) and fluorene (not detected in Pt and 4.10 μ g kg⁻¹ in DRS) were detected in the analyzed fruits. The fluorene content in strawberry fruit represents only 8.20% of the maximum allowed value of 0.05 mg kg⁻¹ for food³¹.

ROOT 2016		ROOT 2017		STEM 2016		STEM 2017		LEAF 2016		LEAF 2017		
Element	Pt	DRS	Pt	DRS	Pt	DRS	Pt	DRS	Pt	DRS	Pt	DRS
Manganese (Mn)	72.4 a	1200 b	171 a	338 b	54 b	20.5 a	83.3 b	21.2 a	86.5 b	<20.0 a	227 b	88.5 a
Iron (Fe)	325 a	4590 b	625 a	2830 b	47.2 a	64.5 a	17.7 a	60.3 b	184 a	205 a	120 a	203 b
Zinc (Zn)	90.1 a	128 a	80.5 a	104 a	43.3 a	32.8 a	39.1 a	30.4 a	26.6 a	25.3 a	22.8 a	20.1 a
Copper (Cu)	9.47 a	96.3 b	7.5 a	41.2 b	< 2.50	3.98	< 2.50	3.18	<2.50	4.91	< 2.50	3.38
Molybdenum (Mo)	17.5 a	43.5 b	4.08 a	196 b	0.575 a	1.08 a	0.165 a	0.333 a	0.33 a	2.02 b	0.557 a	1.16 b
Lead (Pb)	0.966 a	10.1 b	2.19 a	8.56 b	0.111 a	0.239 b	< 0.100	0.124	0.30 a	0.402 a	0.281 a	0.556 a
Cadmium (Cd)	0.193 a	1.3 b	0.327 a	1.51 b	< 0.050	0.089	0.061 a	0.22 b	0.154 a	0.275 b	< 0.050	0.173
Nickel (Ni)	0.683 a	7.18 b	0.964 a	8.23 b	0.147 a	0.266 a	0.246 a	0.360 a	0.487 a	1.32 b	0.424 a	1.16 b
Chromium (Cr)	0.74 a	7.17 b	1.35 a	7.84 b	0.209 a	0.225 a	0.323 a	0.316 a	0.693 a	1.15 a	0.567 a	0.425 b
Mercury (Hg)	<0.010 a	0.045 b	0.037 a	0.072 b	< 0.010	< 0.010	0.043 a	0.037 a	0.013 a	0.015 a	0.043 a	0.048 b
Cobalt (Co)	<0.200 a	6.38 b	0.258 a	2.46 b	< 0.200	< 0.200	< 0.200	< 0.200	< 0.200	< 0.200	< 0.200	0.312
Antimony (Sb)	<0.050 a	1.07 b	<0.05 a	0.283 b	< 0.050	< 0.050	< 0.050	< 0.050	0.024 a	0.027 a	< 0.0500	< 0.050
Arsenic (As)	0.129 a	2.19 b	0.222 a	1.54 b	0.0316 a	0.113 b	0.010 a	0.093 b	0.132 a	0.627 b	0.18 a	0.382 a
Selenium (Se)	<0.100 a	0.711 b	0.239 a	0.795 b	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100
Aluminium (Al)	188 a	1870 b	497 a	3150 b	38.4 a	82.5 b	10.5 a	58.7 b	260 a	285 a	110 a	177 a
Beryllium (Be)	<0.020 a	0.149 b	0.038 a	0.162 b	< 0.020	< 0.020	< 0.020	< 0.020	< 0.02	< 0.02	< 0.02	< 0.02
Tin (Sn)	<0.100 a	0.404 b	0.265 a	2.6 b	< 0.100	< 0.100	< 0.100	1.92	0.102 a	0.077 a	2.84 a	2.02 a
Strontium (Sr)	94.9 a	263 b	151 a	169 a	78.8 a	57.7 a	142 a	117 a	48.2 a	91.5 b	128 a	146 a
Silver (Ag)	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100	< 0.100
Thallium (Tl)	<0.020 a	0.221 a	0.038 a	0.176 a	< 0.020	< 0.020	0.040 a	0.040 a	< 0.020	< 0.020	< 0.020	< 0.020
Vanadium (V)	3.42 a	40.0 a	6.71 a	19.4 a	< 0.200	< 0.200	< 0.200	< 0.200	0.309 a	0.394 a	< 0.200	0.284

Table 5. Metal ion contents (mg kg⁻¹) in different strawberry plant parts grown in peat (Pt) and dredge remediated sediment (DRS). Different letters within a row for each plant part and each year indicate significant differences by Tukey's test. ($\rho \le 0.05$). (n = 3).

	2016		2017			
Element (mg kg ⁻¹)	Pt	DRS	Pt	DRS		
Total Manganese (Mn)	6.02 b	3.11 a	3.78	< 2.0		
Total Iron (Fe)	10.3 a	7.92 a	2.56 a	2.51 a		
Total Zinc (Zn)	3.67 a	4.31 a	1.11 a	1.41 a		
Total Copper (Cu)	< 0.5	0.566	< 0.5	< 0.5		
Molybdenum (Mo)	0.102 a	0.255 b	< 0.02	0.090		
Lead (Pb)	< 0.02	0.06	< 0.02	< 0.02		
Cadmium (Cd)	< 0.02	< 0.02	< 0.02	< 0.02		
Nickel (Ni)	< 0.100	< 0.100	< 0.100	< 0.100		
Chromium (Cr)	< 0.100	< 0.100	< 0.100	< 0.100		
Mercury (Hg)	< 0.005	< 0.005	< 0.005	< 0.005		
Cobalt (Co)	< 0.05	< 0.05	< 0.05	< 0.05		
Antimony (Sb)	< 0.05	< 0.05	< 0.05	< 0.05		
Arsenic (As)	< 0.05	< 0.05	< 0.05	< 0.05		
Selenium (Se)	< 0.05	< 0.05	< 0.05	< 0.05		
Aluminum (Al)	2.95 a	29.1 b	2.95	< 0.500		
Beryllium (Be)	< 0.005	< 0.005	< 0.005	< 0.005		
Tin (Sn)	< 0.2	< 0.2	0.374 a	0.669 b		
Strontium (Sr)	1.85 a	2.99 a	0.729 a	1.11 a		
Silver (Ag)	< 0.025	< 0.025	< 0.025	< 0.025		
Thallium (Tl)	< 0.005	< 0.005	0.070	< 0.005		
Vanadium (V)	< 0.05	< 0.05	< 0.05	< 0.05		

Table 6. Metal ion contents (mg kg⁻¹) in strawberry fruits grown in peat (Pt) and dredge remediated sediment (DRS). Different letters within a line indicate significant differences by Tukey's test. ($\rho \le 0.05$). (n = 3).

On the other hand, the nitrate concentrations obtained both in Pt and in DRS were much lower than that established by European Regulation (EC) n° 1881/2006 and Regulation (EU) n° 1258/2011 for the most restrictive situation (infant food for infants and young children), < 200 mg NO^{3-} kg⁻¹.

Discussion

The experiment began in May 2016, so for the following discussion, the period from May to December 2016 is considered the first year of cultivation, while that from January to July 2017 is considered the second trial year.

According to Fig. 1, during the first year, no significant differences were detected in the needs for irrigation and drainage among the three studied substrates (Pt, Pt-DRS and DRS), although during the second year, the irrigation needs for Pt were greater than those for DRS. This trend reaches significant differences beginning in the month of June (coinciding with the period of greater crop evapotranspiration), with the water applied during June and July in the Pt crop approximately 30% higher than that applied in the DRS crop. Pt-DRS showed intermediate water needs. The DRS drainage volume during the second year was greater than the Pt drainage volume, probably due to the DRS texture being heavier than that of Pt, which facilitated the appearance of channels and cracks in the sediment that increased the collected drainage volume. Regarding the drainage electrical conductivity, no significant differences were observed throughout the experiment; the pH showed similar behavior. These results agree with another study developed with the lettuce by the same research group².

Based on the analysis of mineral elements in DRS drainage, an increase of 2.60% in bicarbonates, 889.31% in ammoniacal N, 74.75% in Mn, 24.41% in Cu and 768.71% in Fe were observed during the second year (2017) when compared with the values in the first cultivation year (2016). For the rest of the analyzed elements (Table 2), during the second year, the concentration decreased by between 10.26% (for sodium) and 55.16% (for potassium), while carbonates remained below the detection limit. This decrease may be due to crop absorption and irrigation washing. Sodium, chloride, potassium, sulfate, nitrate, phosphate, bicarbonate, calcium and magnesium were the elements that contributed the most salinity. Similar results were obtained by Jayasinghe et al.³² in the evaluation of compost as an alternative to peat use. In addition, the results agree with Wang et al.³³ and Kuisma et al.³⁴ that studied more common alternatives growing media (rice husk, perlite, reed canary grass and coconut coir) for strawberry production, confirming the potential of DRS for agriculture media.

Analysis of the data in Table 3 shows that in general, a decrease in heavy metal concentrations in DRS drainage was observed during the second cultivation year; only the Cd, Cu and Hg contents were increased, but the values were below the limits established by the Spanish regulations on surface water quality (Royal Decree 849/1986) and to those indicated by Directive 2008/105/EC. Along the same lines, Jayasinghe et al.³² showed that the total and removable Cu, Zn, Cr, Mn and Pb contents in different media were significantly increased when compared to peat (control), but the values always remained below the limits recommended by the United States Environmental Protection Agency³⁰.

Of the 613 pesticide residues analyzed in the DRS drainage, during the experimental period, a total of 7 pesticides were detected, 5 in 2016 and another 2 in 2017. In both cases, at levels were well below the maximum limit authorized by the Hydraulic Public Domain Spanish Regulation (Royal Decree 849/1986)²².

The concentration of metals in plant tissues and fruits is of great interest, while some microelements, such as iron, copper, cobalt or manganese, have positive effects for humans, and high concentrations of arsenic, cadmium, lead and zinc can be dangerous to health. Therefore, the content of metals in fruits can be an indicator of the level of contamination in the soil in which they were grown^{35,36} since a proportional relationship between the increase in the extractable concentrations of Zn and Cu and the increase in their concentrations in plant tissues has been described³⁵. The contents of metal ions in the different inedible parts of the strawberry (stem and leaves) were below the legal limits. Only in the root were Zn, Cu, Ni, Cr, Cd and Pb values exceeded^{24,25}. However, the root is not used for human or animal food.

For the strawberry fruits, the toxic legal limits were not reached in any case. Thus, 15 of the 21 analyzed elements were below the detection limit, and the rest were below the permissible limits for human consumption^{25,27-29}. In a study on the concentrations of heavy metals in broccoli tissues grown in compost, Jayasinghe et al.³² reached similar conclusions. The analyses were made in plant tissues with both edible and higher concentrations of Zn. These results were also observed by Smith³⁷ and Fiasconaro et al.³⁵, who explained that Zn is easily transferred between tissues and usually has higher concentrations in sludge than other substrates. On the other hand, copper tends to be strongly absorbed by the soil, and its absorption by plants is regulated more effectively than that of Zn³⁸; thus, the Cu concentrations in the plant and fruit are generally smaller. High contents of Zn and Cu have also been described in wastewater sludge³⁹. In general, copper is not considered a toxic element. Therefore, the Cu content obtained for the strawberry fruits produced with DRS was 0.566 mg kg⁻¹ in 2016, and Cu was not detected in 2017. The value was considerably less than the amounts obtained by other authors for sweet orange mesocarp (0.9–3.9 mg kg⁻¹)⁴⁰ and banana peel (12 mg kg⁻¹)⁴¹. The value obtained was 18 times lower than the maximum limit tolerated by European regulations, 10 mg kg⁻¹.

In contrast, lead is considered a potential human carcinogen, with food being the main route of exposure to this metal⁴². Generally, passive Pb absorption occurs by plants, and Pb has difficulty translocating to the fruits. In this study, the Pb content in strawberry fruits was 0.06 mg kg⁻¹ in 2016 and below the detection limits (<0.020 mg kg⁻¹) during the second year of the experiment. European legislation has established a Pb concentration limit lower than 0.1 mg kg^{-125,27}. The Pb contents in the fruits were in accordance with those obtained by Oliva et al.³⁶ for bitter orange fruits.

Cadmium can accumulate in the human body and produce adverse health effects, and food intake is an important route of entry into the human body. Alimentary cadmium content is regulated by EC Regulation n° 1881/2006, which has established 0.050 mg kg⁻¹ as the maximum value in vegetables and fruits. In this case, Cd was not detected in the fruits (<0.02 mg kg⁻¹). Nickel was also not detected in the strawberry fruits. This

element is considered a health hazard due to its carcinogenic activity. Oliva et al.³⁶ found Ni values ranging from $0.15-1.33 \text{ mg kg}^{-1}$ in bitter orange mesocarp, while Markert⁴³ reported a normal Ni value in a reference plant of 1.5 ppm.

Fe and Mn may be considered essential elements for plants and are part of many enzyme systems. They are found in all plants in varying amounts depending on the plant species and the analyzed plant part. The Fe content in strawberry fruits did not show significant differences between those produced with DRS and those produced with Pt (control substrate).

Fruits produced with DRS showed lower values of Mn when compared to Pt fruits. In addition, during the second year, the Mn levels were below the detection limits (< 2.00 mg kg⁻¹). The rest of the studied metals (Cu, Pb, Cd, Ni, Cr, Hg, Co, Sb, As, Be, Ag, Tl, V, Se and Al) showed the same trend (Table 6). Therefore, in general, the strawberry fruits produced with DRS do not contain heavy metals, and the heavy metals that were present were below the limits established by European legislation.

Pesticide residue analysis of the strawberry fruits was also performed. As mentioned, 613 pesticide residues and other contaminants were analyzed (*see supplementary material*); only fluorene (DRS: 4.10 μ g kg⁻¹) and nitrates (Pt: 101 mg kg⁻¹ and DRS 99.40 mg kg⁻¹) were detected. The fluorene concentration is below the limit established by Regulation (EU) nº 10/2011 (0.05 mg kg⁻¹). The nitrate concentration was much lower than that established by European legislation²⁶. Therefore, from the point of view of the presence of pesticide residues, strawberry fruits produced with DRS are exempt and hence could be suitable for consumption in both fresh and processed form.

Based on the results obtained in this work, and based on other studies that have been carried out with different plant species and with sediments from other ports, the European Commission could consider the possibility of producing a legislative change that would allow the use of this resource (DRS, dredged remediated sediments) as a new agricultural substrate to reduce the environmental and economic problems generated by these sediments.

Conclusions

Based on the results, can be affirmed that the dredged remediated sediment used in this study can be used as a culture substrate. Its utilization can be alone or mixed with other substrates that improve its physicochemical properties, especially to achieve a lighter texture with major porosity and higher organic carbon content. This statement is supported and confirmed since the tests showed that the content of heavy metals and pesticides in both water drainage and in strawberry fruits was under the Spanish and European legal limit.

About strawberry fruits, only seven (Mn, Fe, Zn, Mo, Al, Mn and Ni) of the twenty-two metals and two (nitrates and flueorene) of the six hundred-thirteen pesticides analyzed were detected in the fruit validating its safe consumption. In all the cases, values detected were much lower than those established by specific European legislation. The results confirm the suitability of strawberry fruits for fresh and / or processed consumption with no risk. Therefore, the potential for the use of treated sediments as a viable agricultural substrate is confirmed. Thereby, it could also be used as a substrate for the regeneration of degraded soils, recovery of quarries and/or roads/accesses among other similar uses.

However, it is necessary to continue studying the treated sediment behavior, alone and/or mixed, quantitatively and qualitatively, with other substrates such as coconut fiber, biochar, perlite, among others, to improve its characteristics and to conclusively confirm the feasibility of safe use of the sediment.

The debate on the need to make substantial changes to European and Spanish legislation that allows the use of remediated marine sediment must be initiated aiming of minimizing this necessary environmental impact for the economic activity of seaports around the world.

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Author contributions

P.M., E.G., F.H., J.J.M.-N., P.L., and R.M.-F. carried out the experimental design and data collection. All authors wrote and reviewed the manuscript. D.N.-G. formatted the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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