
Structural Characterization of the GM1 Ganglioside by Infrared Multiphoton Dissociation, Electron Capture Dissociation, and Electron Detachment Dissociation Electrospray Ionization FT-ICR MS/MS

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Gangliosides play important biological roles and structural characterization of both the carbohydrate and the lipid moieties is important. The FT-ICR MS/MS techniques of electron capture dissociation (ECD), electron detachment dissociation (EDD), and infrared multiphoton dissociation (IRMPD) provide extensive fragmentation of the protonated and deprotonated GM1 ganglioside. ECD provides extensive structural information, including identification of both halves of the ceramide and cleavage of the acetyl moiety of the N-acetylated sugars. IRMPD provides similar glycan fragmentation but no cleavage of the acetyl moiety. Cleavage between the fatty acid and the long-chain base of the ceramide moiety is seen in negative-ion IRMPD but not in positive-ion IRMPD of GM1. Furthermore, this extent of fragmentation requires a range of laser powers, whereas all information is available from a single ECD experiment. However, stepwise fragmentation by IRMPD may be used to map the relative labilities for a series of cleavages. EDD provides the alternative of electron-induced fragmentation for negative ions with extensive fragmentation, but suffers from low efficiency as well as complication of data analysis by frequent loss of hydrogen atoms. We also show that analysis of MS/MS data for glycolipids is greatly simplified by classification of product ion masses to specific regions of the ganglioside based solely on mass defect graphical analysis. (J Am Soc Mass Spectrom 2005, 16, 752–762) © 2005 American Society for Mass Spectrometry

Gangliosides are amphiphilic components localized mainly in the outer leaf of cell membranes. The structure of a ganglioside includes a ceramide tail (fatty acid N-linked to a sphingosine) of varying length, saturation, and hydroxylation linked to a polar carbohydrate head that contains sialic (neuraminic) acid (Figure 1). The complexity of the carbohydrate moiety makes for complex nomenclature; the current convention has been developed by Svennerholm [1]. The gangliosides are grouped according to

the number of their sialic acid residues: one (M), two (D), or three (T). The G1 series contains a tetraose chain (Gal β 1 \rightarrow 3GalNAc β 1 \rightarrow 4Gal β 1 \rightarrow 4Glc), whereas G2 glycolipids lack the terminal galactose and G3 lack the terminal disaccharide (Gal β 1 \rightarrow 3GalNAc). Therefore, the GM1 ganglioside (Figure 1) denotes a glycolipid that contains a monosialylated tetraose oligosaccharide (Gal β 1 \rightarrow 3GalNAc β 1 \rightarrow 4(NeuAca2 \rightarrow 3)Gal β 1 \rightarrow 4Glc β 1 \rightarrow 1'Cer).

The biological roles of gangliosides include cell–cell recognition, contact inhibition, cell adhesion, and tissue differentiation [2–4]. Although they are found in all tissues, they are particularly enriched in the central nervous system [5]. The carbohydrate portion of the molecule is directed toward the exterior of the cell and mediates molecular recognition. The ceramide composition is cell-specific [6] and provides a hydrophobic anchor. Ceramide modulates glycolipid function [3, 7–9], perhaps by changing the orientation of the carbo-

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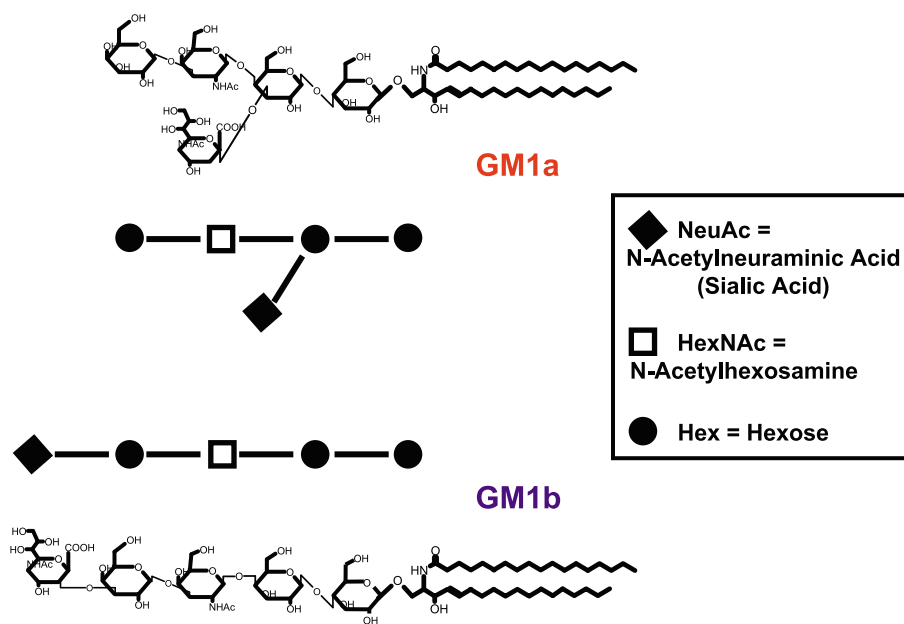


Figure 1. Structures of the ganglioside GM1. Gangliosides contain a hydrophobic ceramide moiety that consists of a long-chain amino alcohol, sphingosine, which is acylated with a fatty acid on the amino group. The polar head is an oligosaccharide that contains sialic acid. GM1 exists as two isomeric forms, GM1a and GM1b, differing only by the branched versus terminal position of the sialic acid.

hydrate relative to the cell surface. Therefore, structural studies of glycolipids should include characterization of both the lipid and the carbohydrate moieties.

Mass spectrometry has been employed extensively to study glycolipid structures. Fast atom bombardment ionization (FAB)-MS [10, 11] and FAB-MS/MS studies [12] of glycolipids successfully determined the structures of both the lipid and the carbohydrate portions of those molecules. However, FAB is a relatively harsh and insensitive ionization technique, which can cause extensive precursor ion fragmentation that complicates mixture analysis. More gentle ionization modes such as matrix-assisted laser desorption/ionization (MALDI) and electrospray ionization (ESI) have also been successful in the analysis of glycolipids. Juhasz and Costello studied underivatized and permethylated gangliosides containing one to five sialic acid residues and different lipid residues by positive- and negative-ion MALDI-TOF MS [13]. ESI has been employed in the analysis of gangliosides with different ceramides isolated from white blood cells [14], gangliosides from complex mixtures obtained from human granulocytes [15], and to obtain profiles of lipids extracted from human brain [16, 17].

Fourier transform ion cyclotron resonance (FT-ICR) mass spectrometry [18, 19] offers high resolving power and high mass accuracy. Dissociation techniques applied to FT-ICR MS analysis of biomolecules include sustained off-resonance collisionally-activated dissociation [20–22] (SORI-CAD), infrared multiphoton dissociation [23, 24] (IRMPD), electron capture dissociation [25, 26] (ECD), and electron detachment dissociation [27] (EDD). SORI-CAD has been employed to charac-

terize lipooligosaccharides from the pathogen *Haemophilus influenzae* [28] and galactosylceramide from pig intestine [29]. ECD has been used to localize sites of O-glycosylation [30] and fatty acid posttranslational modifications in peptides [31]. Recently, Budnik et al. combined ECD and CAD in order to characterize chitooligosaccharides from natural and synthetic sources [32]. We have previously characterized N-linked glycopeptides by ECD and IRMPD, demonstrating the complementarity of structural information obtained from each dissociation technique [33]. Here, we demonstrate the utility of ECD, EDD and IRMPD for the structural characterization of the ganglioside, GM1.

The slow heating IRMPD technique provides a variable duration of irradiation which can be employed to control the extent of secondary fragmentation, thereby accessing fragmentation pathways not available by single-pass CAD external to the ICR cell. The in-cell dissociation techniques of IRMPD, ECD, and EDD eliminate the loss of dynamic range, crucial to detection of single sugars, associated with transfer of externally generated product ions into the ICR cell and do so without the pumpdown delay necessary for in-cell SORI-CAD. For gangliosides, IRMPD produces extensive fragmentation of both positive and negative ions. The “nonergodic” ECD technique provides a more random and more extensive distribution of fragmentation sites. Consequently, ECD produces extensive primary fragmentation with short duration (milliseconds) electron irradiation and a better picture of connectivity between sugars. EDD provides an electron irradiation alternative for negative ions but more complex fragmentation patterns.

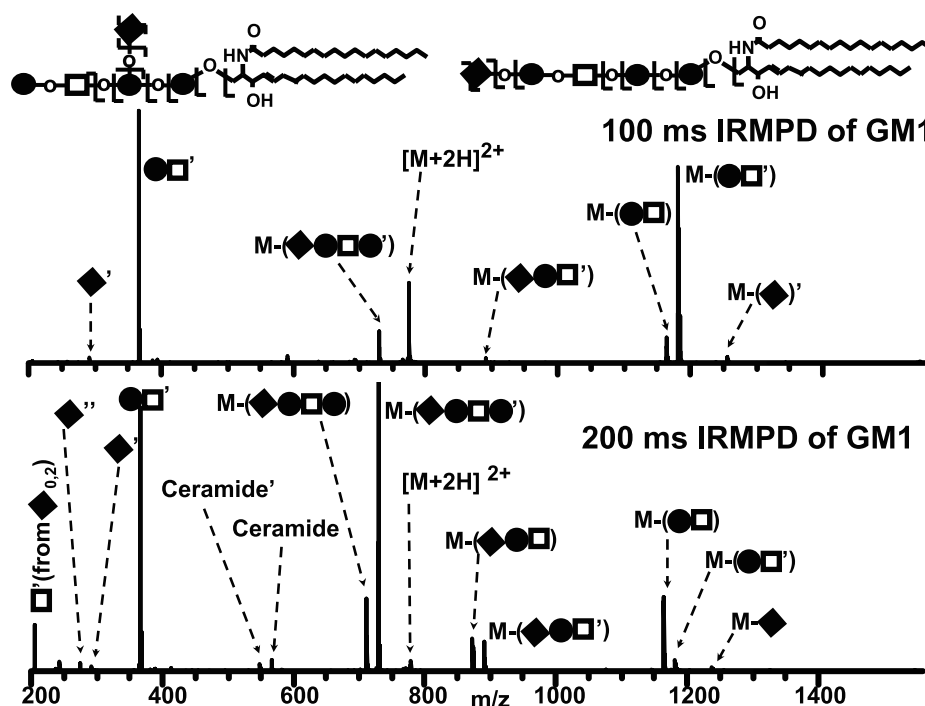


Figure 2. (Top) Product positive-ion spectrum following IRMPD (100 ms, 8W) FT-ICR MS/MS of the quadrupole isolated doubly protonated GM1 precursor (30 scans). This spectrum suggests that the dominant fragment, HexHexNAc' at m/z 366.139, is most likely the initial site of fragmentation of GM1a. All fragment ions are singly charged. (Filled diamond, NeuAc = N-acetylneuraminic acid or sialic acid; filled circle, Hex=hexose; open square, HexNAc = N-acetyl hexosamine). (The prime symbol denotes loss of H_2O). (Symbol with a through bar denotes cross-ring cleavage). (Bottom) Product positive-ion mass spectrum under the same conditions as Figure 3a, except that the duration of IRMPD irradiation has been lengthened to 200 ms (30 scans). Anomeric cleavage is seen, except at the nonreducing end of HexNAc. Much of the additional fragmentation at longer irradiation most likely arises from secondary fragmentation of the M-HexHexNAc' product ion, which makes the presence of the GM1b isomer ambiguous.

Methods

Sample Preparation

GM1-ganglioside from bovine brain was isolated and characterized at the Institute of Clinical Neuroscience [34, 35]. The GM1 sample contained no detectable amount of the lactotetraose series gangliosides 3'-LM1 and 3'-iso-LM1 [36]. Thin layer chromatography followed by chemical detection with resorcinol showed only the occurrence of GM1a. Due to the detection limit of the resorcinol reagent, the occurrence of GM1b can't be excluded but is $<0.2\%$. Dehydrated samples were resuspended to a concentration of 100 μM in methanol (Baker, Philipsburg, NJ). For electrospray, samples were diluted to a concentration of 3 to 5 μM in 1:1 water/methanol with 3% acetic acid for positive ionization or 100% methanol for negative ionization.

Fourier Transform Ion Cyclotron Resonance Mass Spectrometry

All experiments were performed with a homebuilt 9.4 T passively shielded ESI-Q-FT-ICR mass spectrometer [37] controlled by a modular ICR data acquisition

system (MIDAS) [38]. Samples were infused at a flow rate of 350 nL/min through a 50 μm i.d. fused silica capillary with a mechanically ground tip [39]. For external ion accumulation [40, 41], ions are held in a focusing octopole followed by precursor ion mass selection in a quadrupole mass filter and subsequent accumulation in a linear octopole ion trap (30 s for positive ions and 5 s for negative ions). Negative ions are further isolated in the ICR cell by stored waveform inverse Fourier transform (SWIFT) isolation [42, 43]. The accumulation octopole has been modified for improved ion extraction to include tilted wires that produce an axial potential gradient [44]. Prior to dissociation, ions are transferred through an octopole ion guide to an open-ended cylindrical Penning trap [45].

The instrument configuration for simultaneous access to both electron and photon irradiation has been previously described [46]. For IRMPD, an off-axis (~ 2.5 degrees) CO_2 laser (Synrad, Mukilteo, WA) fitted with a 2.5 \times beam expander irradiates the ion cloud through a BaF_2 window. For ECD and EDD, a 1 cm diameter dispenser cathode electron source (no. 1109; HeatWave, Watsonville, CA) is positioned on-axis approximately 73 cm behind the ICR cell. Normally the molybdenum grid

Table 1. Molecular assignments and neural masses of product ions of GM1 in IRMPD of the doubly protonated precursor ion

(z)	Measured m/z	Δ Precursor	Assignment	Formula	Calculated m/z	Error (ppm)
(1)	204.0864	1342.7976	HexNAc' or A1 of NeuAc'	$C_8H_{13}NO_5$	204.0867	+1.2
(1)	274.0919	1272.7921	NeuAc''	$C_{11}H_{15}NO_7$	274.0921	+0.5
(1)	292.1024	1254.7816	NeuAc'	$C_{11}H_{17}NO_8$	292.1027	+0.9
(1)	366.1392	1180.7448	HexHexNAc'	$C_{14}H_{23}NO_{10}$	366.1395	+0.7
(1)	530.5294	1016.3546	ceramide''	$C_{36}H_{67}NO$	530.5295	+0.1
(1)	548.5399	998.3441	cermide'	$C_{36}H_{69}NO_2$	548.5401	+0.3
(1)	566.5506	980.3334	ceramide	$C_{36}H_{71}NO_3$	566.5507	+0.1
(1)	710.5933	836.2907	M-HexHexNAcHexNeuAc	$C_{42}H_{79}NO_7$	710.5929	-0.6
(1)	728.6042	818.2798	M-(HexHexNAcHexNeuAc)'	$C_{42}H_{81}NO_8$	728.6035	-1.0
(1)	872.6475	674.2365	M-HexHexNAcNeuAc	$C_{48}H_{89}NO_{12}$	872.6457	-2.1
(1)	890.6572	656.2268	M-(HexHexNAcNeuAc)'	$C_{48}H_{91}NO_{13}$	890.6563	-1.1
(1)	1163.7429	383.1411	M-HexHexNAc	$C_{59}H_{106}N_2O_{20}$	1163.7412	-1.5
(1)	1181.7549	365.1291	M-(HexHexNAc)'	$C_{59}H_{108}N_2O_{21}$	1181.7517	-2.8
(1)	1237.7798	309.1042	M-NeuAc	$C_{62}H_{112}N_2O_{22}$	1237.7779	-1.6
(1)	1255.7867	291.0973	M-NeuAc'	$C_{62}H_{114}N_2O_{23}$	1255.7884	+1.3
(1)	1325.7984	221.0856	M-A1 of NeuAc	$C_{65}H_{116}N_2O_{25}$	1325.7939	-3.4
(1)	1528.8685	18.0155	$[M + H]^+ - H_2O$	$C_{73}H_{129}N_3O_{30}$	1528.8731	+3.0
(1)	1546.8825	0.0015	$[M + H]^+$	$C_{73}H_{131}N_3O_{31}$	1546.8838	+0.8

(80% transmission) is held at -200 V to impede the flow of electrons. For ECD (of positive ions) the cathode voltage is biased to -5 V, the grid to $+5$ V, and the ICR cell trap plates to $+10$ V. Following ECD, a 1 ms event in which the trap plates are dropped to $+2$ V, grid to $+5$ V, and cathode to $+10$ V is used to purge any remaining electrons

from the ICR cell. For EDD (of negative ions) the cathode voltage is biased to -8 V, the grid to $+200$ V and the trap plates to -5 V. Following dissociation ions undergo frequency-sweep excitation (72 to 960 kHz at 150 Hz/ μ s, m/z 150–2000) and broadband detection (512 Kword data points). The digitized time domain transient signal is

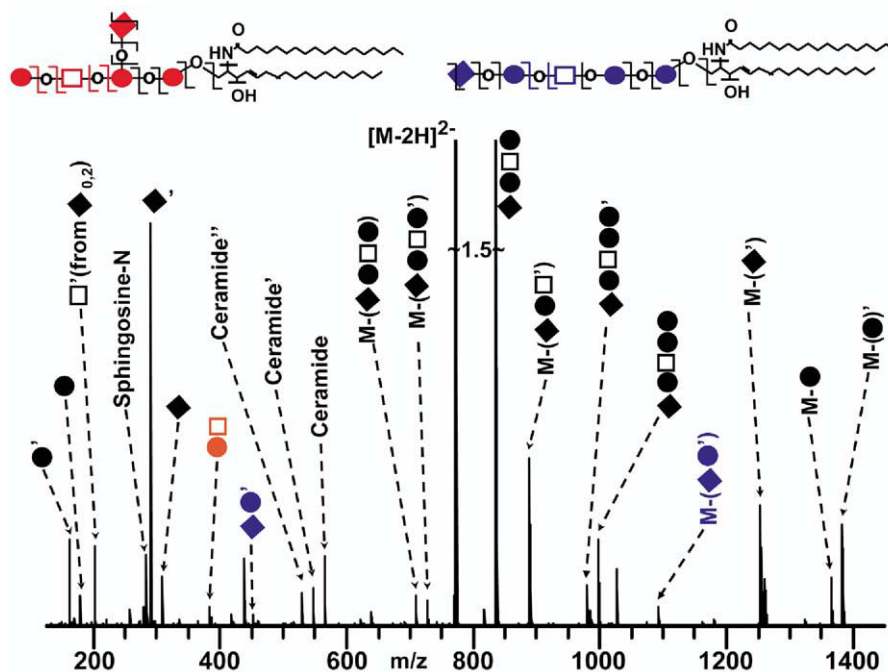


Figure 3. Product negative-ion mass spectrum following IRMPD (200 ms, 14 W) of doubly deprotonated GM1 precursor (100 scans). Fragmentation is more extensive than for IRMPD of the doubly protonated form. In contrast to IRMPD of the protonated form, these fragments include cleavage within the ceramide moiety associated with loss of the fatty acid. Product ions that can belong only to GM1a are shown in red; those which suggest GM1b are shown in blue. Presence of a GM1b as a minor component is evident from the fragment ion peak at m/z 1091.722 which corresponds to a loss of the disaccharide HexNeuAc' from the precursor and its disaccharide complement at m/z 452.141. (All fragment ions are singly charged.)

Table 2. Molecular assignments and neutral masses of product ions of GM1 in IRMPD of the doubly deprotonated precursor ion

(z)	Measured m/z	Δ Precursor	Assignment	Formula	Calculated m/z	Error (ppm)
(1)	161.0453	1383.8240	hexose'	C ₆ H ₁₀ O ₅	161.0456	+1.6
(1)	179.0558	1365.8135	hexose	C ₆ H ₁₂ O ₆	179.0562	+2.0
(1)	202.0718	1342.7975	HexNAc' or A1 of NeuAc'	C ₈ H ₁₃ NO ₅	202.0722	+1.7
(1)	283.2638	1261.6055	sphingosine-N	C ₁₈ H ₃₆ O ₂	283.2643	+1.6
(1)	290.0877	1254.7816	NeuAc'	C ₁₁ H ₁₇ NO ₈	290.0882	+1.6
(1)	308.0983	1236.7710	NeuAc	C ₁₁ H ₁₉ NO ₉	308.0988	+1.5
(1)	382.1348	1162.7345	HexHexNAc	C ₁₄ H ₂₅ NO ₁₁	382.1356	+2.0
(1)	452.1405	1092.7288	HexNeuAc'	C ₁₇ H ₂₇ NO ₁₃	452.1410	+1.0
(1)	528.5145	1016.3548	ceramide''	C ₃₆ H ₆₇ NO	528.5150	+0.9
(1)	546.5252	998.3441	ceramide'	C ₃₆ H ₆₉ NO ₂	546.5257	+0.8
(1)	564.5356	980.3337	ceramide	C ₃₆ H ₇₁ NO ₃	564.5362	+1.0
(1)	708.5784	836.2909	M-HexHexNAcHexNeuAc	C ₄₂ H ₇₉ NO ₇	708.5784	-0.1
(1)	726.5889	818.2804	M-HexHexNAcHexNeuAc'	C ₄₂ H ₈₁ NO ₈	726.5890	+0.1
(2)	771.9308	0.0004	[M-2H] ²⁻	C ₇₃ H ₁₃₁ N ₃ O ₃₁	771.9310	+0.3
(1)	817.2734	727.5959	HexHexNAcHexNeuAc'	C ₃₁ H ₅₀ N ₂ O ₂₃	817.2731	-0.4
(1)	835.2841	709.5852	HexHexNAcHexNeuAc	C ₃₁ H ₅₂ N ₂ O ₂₄	835.2837	-0.5
(1)	888.6421	656.2272	M-(HexHexNAcNeuAc')	C ₄₈ H ₉₁ NO ₁₃	888.6418	-0.4
(1)	979.3257	565.5436	glycan'	C ₃₇ H ₆₀ N ₂ O ₂₈	979.3259	+0.2
(1)	997.3375	547.5318	glycan	C ₃₇ H ₆₂ N ₂ O ₂₉	997.3365	-1.1
(1)	1073.7081	471.1612	M-(HexNeuAc)	C ₅₆ H ₁₀₂ N ₂ O ₁₇	1073.7106	+2.3
(1)	1091.7216	453.1477	M-(HexNeuAc')	C ₅₆ H ₁₀₄ N ₂ O ₁₈	1091.7211	-0.5
(1)	1161.7301	383.1392	M-(HexHexNAc)	C ₅₉ H ₁₀₆ N ₂ O ₂₀	1161.7267	-3.0
(1)	1179.7367	365.1326	M-(HexHexNAc')	C ₅₉ H ₁₀₈ N ₂ O ₂₁	1179.7372	+0.4
(1)	1253.7756	291.0937	M-NeuAc'	C ₆₂ H ₁₁₄ N ₂ O ₂₃	1253.7739	-1.4
(1)	1260.5987	284.2706	M-(fattyacid-H)-H ₂ O	C ₅₅ H ₉₅ N ₃ O ₂₉	1260.5978	-0.8
(1)	1323.782	221.0873	M-A1 of NeuAc	C ₆₅ H ₁₁₆ N ₂ O ₂₅	1323.7794	-2.0
(1)	1364.8076	180.0617	M-Gal	C ₆₇ H ₁₁₉ N ₃ O ₂₅	1364.8059	-1.3
(1)	1382.8166	162.0527	M-Gal'	C ₆₇ H ₁₂₁ N ₃ O ₂₆	1382.8165	-0.1

Hanning apodized and one zero-fill applied prior to fast Fourier transformation to yield a magnitude mode frequency spectrum that is converted to a mass-to-charge ratio spectrum with a two-term calibration equation [47, 48]. All precursor ions are doubly protonated or doubly deprotonated. All product ions are singly protonated or singly deprotonated. All spectra are externally calibrated with Agilent ESI tuning mix ions (m/z 622.02895, 922.00979, and 1521.97146, Agilent Technologies, Wilmington, DE). Molecular formulas were derived from calibrated masses with an in-house molecular formula calculator and theoretical masses were calculated with Isopro 3.1 (MS/MS software, <http://members.aol.com/mmssoft/>). All MS and MS/MS assignments have a mass accuracy of 3 ppm or better.

Results and Discussion

GM1 was analyzed as both positive and negative ions with a 9.4 T ESI-Q-FT-ICR mass spectrometer. GM1 exists as two major species in brain differing by ceramide chain lengths, d18:1/C18:0 and d20:1/C18:0 [49]. All present experiments derive from the quadrupole-isolated d18:1/C18:0 form.

Infrared Multiphoton Dissociation (IRMPD) of Positive Ions

Figure 2 (top) shows the product ion spectrum obtained from IRMPD (100 ms, 8W) FT-ICR MS/MS of

quadrupole-isolated doubly protonated GM1 d18:1/C18:0 precursor ions (30 scans). IRMPD is a slow heating technique and therefore cleaves the weakest bonds first [50]. The irradiation period of 100 ms represents the initial threshold for fragmentation; loss of the disaccharide HexHexNAc' (' = -H₂O) thus identifies the initial cleavage site. No fragments corresponding to terminal NeuAcHex and NeuAcHexHexNAc could be detected in this spectrum. This result confirms the GM1a isomer as the principal species. Because only the most labile bonds are broken, one learns only a little about the carbohydrate and nothing about the ceramide portion of the ganglioside.

Fortunately, the primary IRMPD fragment ions remain trapped in the path of the IR beam, and may thus continue to absorb IR power until secondary fragmentation (of stronger bonds) occurs. The extent of fragmentation may thus be adjusted by varying the duration of IR irradiation. Figure 2 (bottom) (Table 1) shows the FT-ICR MS/MS spectrum for a longer IR-irradiation period (200 ms, 8 W) under otherwise identical conditions (30 scans). Much more extensive information results from further fragmentation of the M-HexHexNAc' product ions, including anomeric cleavage between adjacent sugars, except at the non-reducing end of HexNAc. Fragmentation of both positive and negative ions produces a neutral loss equal to the mass of HexNAc m/z as well as the corre-

Table 3. Molecular assignments and neutral masses of product ions of GM1 in ECD of the doubly protonated precursor ion

(z)	Measured <i>m/z</i>	Δ Precursor	Assignment	Formula	Calculated <i>m/z</i>	error (ppm)
(1)	204.0863	1342.7977	HexNAc' or A1 of NeuAc	C ₈ H ₁₃ NO ₅	204.0867	+1.7
(1)	264.2682	1282.6158	sphingosine"	C ₁₈ H ₃₃ N	264.2686	+1.3
(1)	274.0917	1272.7923	NeuAc"	C ₁₁ H ₁₅ NO ₇	274.0921	+1.3
(1)	284.2943	1262.5897	fatty acid + NH ₂ *	C ₁₈ H ₃₇ NO	284.2948	+1.6
(1)	292.1023	1254.7817	NeuAc'	C ₁₁ H ₁₇ NO ₈	292.1027	+1.2
(1)	366.1389	1180.7451	HexHexNAc'	C ₁₄ H ₂₃ NO ₁₀	366.1395	+1.5
(1)	454.1549	1092.7291	HexNeuAc'	C ₁₇ H ₂₇ NO ₁₃	454.1555	+1.2
(1)	530.5286	1016.3554	ceramide" & -H	C ₃₆ H ₆₇ NO	530.5295	+1.6
(1)	548.5392	998.3448	ceramide' & -2H	C ₃₆ H ₆₉ NO ₂	548.5401	+1.6
(1)	566.5498	980.3342	ceramide	C ₃₆ H ₇₁ NO ₃	566.5507	+1.5
(1)	675.2449	871.6391	HexHexNAcNeuAc	C ₂₅ H ₄₂ N ₂ O ₁₉	675.2455	+0.8
(1)	710.5924	836.2916	M-(HexHexNAcHexNeuAc) & -2H	C ₄₂ H ₇₉ NO ₇	710.5930	+0.8
(1)	728.6029	818.2811	M-(HexHexNAcHexNeuAc)' & -2H	C ₄₂ H ₈₁ NO ₈	728.6036	+0.9
(2)	773.9451	0.0011	[M + 2H] ²⁺	C ₇₃ H ₁₃₁ N ₃ O ₃₁	773.9456	+0.6
(1)	819.2853	727.5987	HexHexNAcHexNeuAc' & -H	C ₃₁ H ₅₀ N ₂ O ₂₃	819.2877	+2.9
(1)	856.6499	690.2341	M-(HexHexNAcNeuAc)-O & -2H	C ₄₈ H ₈₉ NO ₁₁	856.6509	+1.1
(1)	872.6448	674.2392	M-(HexHexNAcNeuAc) & -2H	C ₄₈ H ₈₉ NO ₁₂	872.6458	+1.1
(1)	890.6561	656.2279	M-(HexHexNAcNeuAc)' & -2H	C ₄₈ H ₉₁ NO ₁₃	890.6564	+0.3
(1)	981.3394	565.5446	glycan' & -H	C ₃₇ H ₆₀ N ₂ O ₂₈	981.3405	+1.1
(1)	999.3502	547.5338	glycan	C ₃₇ H ₆₂ N ₂ O ₂₉	999.3511	+0.9
(1)	1075.7258	471.1582	M-(HexNeuAc)	C ₅₆ H ₁₀₂ N ₂ O ₁₇	1075.7252	-0.6
(1)	1163.7414	383.1426	M-(HexHexNAc)	C ₅₉ H ₁₀₆ N ₂ O ₂₀	1163.7412	-0.2
(1)	1181.7522	365.1318	M-(HexHexNAc)'	C ₅₉ H ₁₀₈ N ₂ O ₂₁	1181.7518	-0.4
(1)	1235.7618	311.1222	M-(NeuAc)-2H	C ₆₂ H ₁₁₀ N ₂ O ₂₂	1235.7623	+0.4
(1)	1237.7751	309.1089	M-NeuAc	C ₆₂ H ₁₁₂ N ₂ O ₂₂	1237.7780	+2.3
(1)	1253.7729	293.1111	M-(NeuAc)' -2H	C ₆₂ H ₁₁₂ N ₂ O ₂₃	1253.7729	-0.04
(1)	1255.7831	291.1009	M-NeuAc'	C ₆₂ H ₁₁₄ N ₂ O ₂₃	1255.7886	+4.3
(1)	1262.6091	284.2749	M-(fatty acid + NH ₂)	C ₅₅ H ₉₅ N ₃ O ₂₉	1262.6124	+2.6
(1)	1325.7949	221.0891	M-HexNAc or M-A1 of NeuAc	C ₆₅ H ₁₁₆ N ₂ O ₂₅	1325.7941	-0.6
(1)	1504.8743	42.0097	M-C ₂ H ₂ O (-acetyl)	C ₇₁ H ₁₂₉ N ₃ O ₃₀	1504.8734	-0.6
(1)	1528.8734	18.0106	[M + H] ⁺ - H ₂ O	C ₇₃ H ₁₂₉ N ₃ O ₃₀	1528.8734	-0.03
(1)	1546.8853	-0.0013	[M + H] ⁺	C ₇₃ H ₁₃₁ N ₃ O ₃₁	1546.8840	-0.9
(1)	1547.8869	0.0044	[M + 2H + e] ⁺ *	C ₇₃ H ₁₃₁ N ₃ O ₃₁	1547.8917	+3.1

charged product ions of GM1 would result in neutralization, so that any subsequent fragmentation will not be observable by mass spectrometry. However, multiple peptide backbone cleavages in cyclic peptides [52] and secondary side chain cleavages in linear peptides [53] have both been shown to be induced by capture of a single electron. Further investigation and a wider range of pure gangliosides will be required to establish whether capture of an electron results in single or multiple cleavages within the glycan.

Electron Detachment Dissociation (EDD) of Negative Ions

Figure 5 (see also Table 4) shows the negative product ion mass spectrum obtained by EDD (1 s electron irradiation) FT-ICR MS/MS of the quadrupole- and SWIFT-isolated doubly deprotonated d18:1/C18:0 GM1 precursor (100 scans). Little is known of the actual mechanism for EDD fragmentation, although it is postulated to involve electron-hole recombination [27]. Moreover, the lack of charge mediation in negative-ion fragmentation makes interpretation of the spectra difficult. As for negative-ion IRMPD, cleavage is seen between all sugars. However, no cleavage is seen within the ceramide. In addition, most water loss

fragments also show a loss of H[•] (not shown). Frequent loss of H[•] combined with the 2H loss associated with the loss of sialic acid makes assignment of novel fragmentation difficult. Several relatively abundant fragment ions remain unassigned. EDD does offer a negative-ion alternative to IRMPD but its low efficiency of precursor to product ion conversion combined with the ambiguity in novel assignments makes this technique difficult for structural analysis of gangliosides, at least until more is understood of this fragmentation technique. EDD of nonsialylated glycosphingolipids, which would eliminate the confusion of a loss of two hydrogens could prove more useful in deciphering novel fragmentation patterns such as possible cross-ring cleavages.

Mass Defect Analysis

Many species in the mass spectra of Figures 2–5 may be assigned by matching the accurately measured (to ~1 ppm) mass to a putative elemental composition, C_cH_hN_nO_o, corresponding to conventional bond cleavages (e.g., peptidic, glycosidic). Assignment of species corresponding to other cleavages, e.g., cross-ring, with the simultaneous complication of losses of H[•], 2H, H₂O, etc. is much more difficult. However, it is possible to distinguish lipid

Table 4. Molecular assignments and neutral masses of product ions of GM1 in EDD of the doubly deprotonated precursor ion

(z)	measured <i>m/z</i>	Δ precursor	assignment	formula	calc'd <i>m/z</i>	error (ppm)
(1)	290.0876	1254.7819	NeuAc-H ₂ O	C ₁₁ H ₁₇ NO ₈	290.0882	+1.9
(1)	308.0982	1236.7713	NeuAc and NeuAc-H [•]	C ₁₁ H ₁₉ NO ₉	308.0988	+1.8
(1)	450.1246	1094.7449	HexNeuAc'-2H	C ₁₇ H ₂₅ NO ₁₃	450.1253	+1.6
(1)	564.5353	980.3342	ceramide	C ₃₆ H ₇₁ NO ₃	564.5362	+1.5
(1)	655.2196	889.6499	HexHexNAcNeuAc' & -H [•]	C ₂₅ H ₄₀ N ₂ O ₁₈	655.2204	+1.1
(1)	706.2404	838.6291	HexHexNAcHexHex	C ₂₆ H ₄₅ NO ₂₁	706.2412	+1.1
(1)	726.5886	818.2809	M-(HexHexNAcHexNeuAc)'	C ₄₂ H ₈₁ NO ₈	726.5891	+0.6
(2)	771.9300	0.0022	[M-2H] ²⁻	C ₇₃ H ₁₃₁ N ₃ O ₃₁	771.9311	+1.4
(1)	833.2679	711.6016	HexHexNAcNeuAc-2H	C ₃₁ H ₅₀ N ₂ O ₂₄	833.2682	+0.3
(1)	978.3177	566.5518	glycan'-H [•]	C ₃₇ H ₅₉ N ₂ O ₂₈	978.3182	+0.5
(1)	995.3209	549.5486	glycan-2H	C ₃₇ H ₆₀ N ₂ O ₂₉	995.3210	+0.1
(1)	997.3359	547.5336	glycan	C ₃₇ H ₆₂ N ₂ O ₂₉	997.3366	+0.7
(1)	1091.7222	453.1473	M-(HexNeuAc)'	C ₅₆ H ₁₀₄ N ₂ O ₁₈	1091.7212	-1.0
(1)	1161.7244	383.1451	M-(HexHexNAc)	C ₅₉ H ₁₀₆ N ₂ O ₂₀	1161.7267	+2.0
(1)	1253.7738	291.0957	M-(NeuAc)' & -H [•]	C ₆₂ H ₁₁₄ N ₂ O ₂₃	1253.7740	+0.1
(1)	1324.7871	220.0824	M-(HexNAc) + H	C ₆₅ H ₁₁₇ N ₂ O ₂₅	1324.7873	+0.1
(1)	1382.8138	162.0557	M-(Hex-H ₂ O)	C ₆₇ H ₁₂₁ N ₃ O ₂₆	1382.8166	+2.0
(1)	1380.7992	164.0703	M-(Hex-O)	C ₆₇ H ₁₁₉ N ₃ O ₂₆	1380.8010	+1.3
(1)	1482.8345	62.0350	M-HexA2	C ₇₁ H ₁₂₅ N ₃ O ₂₉	1482.8327	-1.2
(1)	1542.8534	1.0088	[M-3H-2e] ⁻	C ₇₃ H ₁₃₀ N ₃ O ₃₁	1542.8538	+0.3
(1)	1543.8606	0.0016	[M-2H-e] ^{-•}	C ₇₃ H ₁₃₁ N ₃ O ₃₁	1543.8617	+0.7

from oligosaccharide from liposaccharide, and even determine the number of fatty acid chains and/or number of sugars, based simply on differences in fractional mass (i.e., the decimal portion of the exact mass). Mass defect graphical analysis [54–56] is particularly successful for structural analysis of glycolipids because of the significant compositional differences between sugars versus lipids

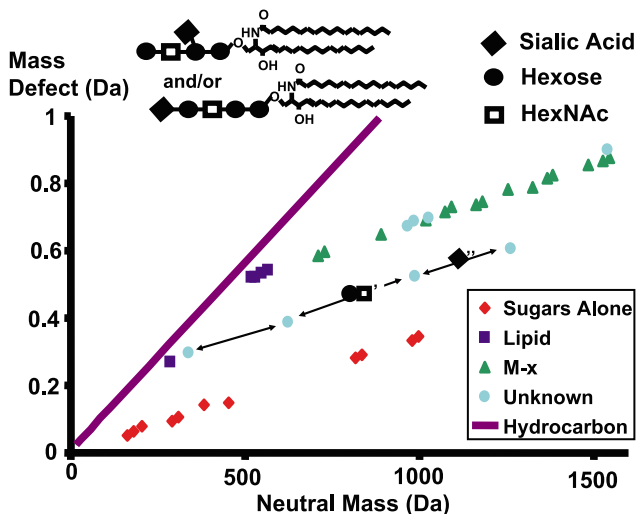


Figure 6. Mass defect graphical analysis of the IRMPD of doubly deprotonated GM1 spectra shown in Figure 3. The high mass resolving power of FT-ICR MS/MS allows localization of product ion masses to specific regions of the ganglioside based solely on mass defect graphical analysis. Product ion masses separate themselves into distinct lines, with those containing lipid eventually intercepting the theoretical hydrocarbon line. This graph shows the existence of a significant loss of hydrocarbon chain (most likely fatty acid) followed by a series of secondary sugar losses from the remaining long-chain base, a series that would not have been easy to pick out by product ion masses alone.

greatly simplifies assignment of MS-MS data. Figure 6 and Figure 7 illustrate the ease with which product ion masses can be assigned to specific regions of the ganglioside. Product ion masses separate themselves into distinct lines, with those containing lipid eventually intercepting the theoretical hydrocarbon line. Graphical analysis of product ion data for the negative-ion IRMPD of GM1 (Figure 6) makes evident the existence of a significant loss of hydrocarbon chain (most likely loss of fatty acid) followed by a series of secondary sugar losses from the remaining long-chain base, a series that would not have been easy to pick out from product ion masses alone. Trends in fragmenta-

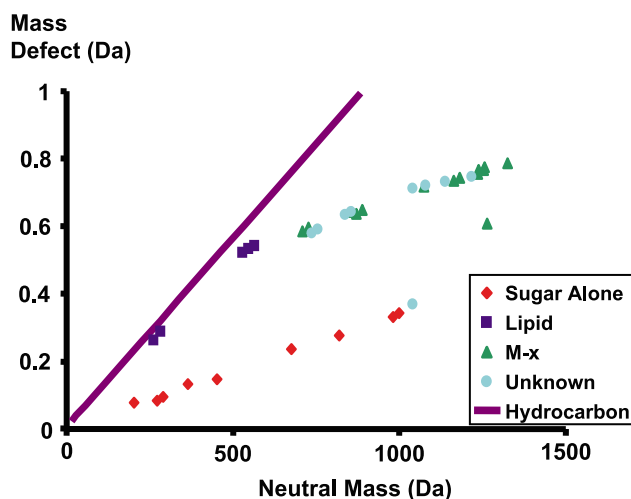


Figure 7. Mass defect graphical analysis of the ECD spectra shown in Figure 4. Assignment of the fragments corresponding to cleavage on either side of the nitrogen of the ceramide is confirmed, as well as identification of the presence of a complementary M - (loss of the fatty acid + NH₂) (not labeled in Figure 4).

tion patterns, such as the significant secondary fragmentation of IRMPD, are easily visualized and provide additional insight into the pattern of fragmentation. Similarly, mass defect graphical analysis for ECD of GM1 (Figure 7) supports the possibility that there may be little to no secondary fragmentation within the glycan following initial electron capture. Here we see a similar loss of a hydrocarbon chain as well as its complement but no intermediate sequence of sugar loss, suggesting primary fragmentation as the origin of sugar fragments. Graphical analysis of EDD fragment ion data (not shown) shows only a series of sugar loss from the parent and a series of sugars. All unassigned peaks fall into one of these two lines, suggesting cross-ring cleavage or extensive neutral loss for the yet unassigned fragments.

Conclusions

The FT-ICR MS/MS techniques of infrared multiphoton dissociation, electron capture dissociation, and electron detachment dissociation produce extensive fragmentation of the protonated and deprotonated GM1 ganglioside. ECD offers the most promise because its extensive fragmentation, as well as the possibility of little to no secondary fragmentation, allows for a more definitive determination of glycan sequence. In addition, ECD identifies both halves of the ceramide and cleaves the acetyl moiety to identify the presence of acetylated sugars. IRMPD of both positive and negative ions produces extensive fragmentation, but requires multiple experiments at different laser power, and also leads to secondary fragmentation, whereas a single ECD experiment provides comparable or more information. However, stepwise fragmentation by IRMPD is useful to map the relative lability among a series of cleavages. Cleavage between the two halves of the ceramide is seen by negative-ion (but not positive-ion) IRMPD, although we do observe such cleavage by positive-ion IRMPD of sulfatides. EDD provides extensive fragmentation of negative ions, but suffers from low efficiency as well as complication of data analysis by frequent hydrogen atom loss. Finally, mass defect analysis can help to distinguish lipid from glycan from glycolipid as an aid in assignment of MS/MS data.

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