The article, "Reactivity of Aromatic σ,σ -Biradicals Toward Riboses" by Anthony Adeuya, Linan Yang, F. Sedinam Amegayibor, John J. Nash, and Hilkka I. Kenttämäa which was published in the October issue, Vol. 17, No. 10, pages 1325–1334, had an error on **Table 2**. In the fourth column, the biradical **2a** reacts with 2-deoxy-D-ribose by addition (97%) as well as 2H-atom abstraction (3%). The corrected **Table 2** is shown below. The publisher apologizes for the error.

Table 2. Reaction efficiencies^a and products^b of the reactions of (bi)radicals **1a–4a** with sugars

	D-ribose	2-deoxy-D-ribose	1-O-methyl-2-deoxy-D-ribose
H abst. (2°) 2H abst.	H abst. (2°) 2H abst.	H abst. (2°) 2H abst.
1a	100%	100%	^d Trace amount of <i>m/z</i> 147 and 200
	Eff.	Eff.	Eff.
	42% ^c	46% ^c	56% ^e
Adduct	Adduct 2H abst.	Adduct 2H abst.
2a	100%	97% 3%	83% 17%
	Eff.	Eff.	Eff.
	1.3% ^f	0.70% ^g	0.45% ^c
Adduct	Adduct	Adduct
3a	100%	100%	100%
	Eff.	Eff.	Eff.
	1.0%	0.43% ^c	Very slow
H abst.	H abst.	H abst.
4a	100%	100%	100%
	Eff.	Eff.	Eff.
	32%	38%	34%

^aReaction efficiency (Eff.) = $k_{\text{reaction}}/k_{\text{collision}} \times 100\%$.

^bProduct branching ratios are given after each primary reaction product.

^cFrom nonlinear curve fit: ~20% of the reactant ion population was unreactive.

^dUnidentified products.

^eFrom nonlinear curve fit: ~5% of the reactant ion population was unreactive.

^fFrom nonlinear curve fit: ~10% of the reactant ion population was unreactive.

^gFrom nonlinear curve fit: ~40% of the reactant ion population was unreactive.