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PEGylated carbon dot/MnO₂ nanohybrid: a new pH/ H₂O₂-driven, turn-on cancer nanotheranostics

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ABSTRACT The effect of tumor-targeted photodynamic therapy (PDT) was improved by designing nanotheranostics to promote oxygenation in a tumor microenvironment (TME) wherein hypoxia, acidosis, and the elevated levels of H₂O₂ are three main characteristics. In this study, a carbon dot (CD) PDT agent recently developed by our group was firstly applied as reducing agent to react with potassium permanganate for fabricating CDs/manganese dioxide (CDs/MnO₂) composites, which were in turn modified with polyethylene glycol (PEG) to form water-soluble CDs/MnO₂-PEG nanohybrids. In a normal physiological environment, the as-prepared nanohybrids exhibited quenched fluorescence, weak singlet oxygen generation, and low magnetic resonance imaging (MRI) signal. However, given the high sensitivity of MnO₂ to the TME, the CDs/MnO₂-PEG nanohybrids changed from an "off" to an "on" state with synchronously enhanced fluorescence, singlet oxygen generation, and MRI signal in the TME. In vitro and in vivo analyses have revealed that CDs/MnO₂-PEG nanohybrids could be applied as TME-driven, turn-on nanotheranostics for the MR/fluorescence bimodal imaging-guided PDT of cancer. Moreover, complete clearance of CDs/MnO₂-PEG nanohybrids from the body of mice was observed, indicating their low long-term toxicity and good biocompatibility. This work offers a new nanotheranostic candidate for modulating the unfavorable TME, particularly for the targeted PDT of cancer through precise positioning and oxygen generation.

Keywords: tumor microenvironment, photodynamic therapy, carbon dots, turn-on theranostics, manganese dioxide

INTRODUCTION

Nanotheranostics that combine medical diagnostics and therapeutics into one nanosystem have been considered as the key toward realizing personalized nanomedicine for the precise treatment of cancer in the future [1-7]. In the past decade, numerous studies have explored various multifunctional theranostics nanosystems based on external (e.g., ultrasound, light, magnetic and electrical fields) or internal factors (e.g., pH value, redox, environment, and enzyme) [8–21]. Among them, pH/H₂O₂-driven nanotheranostics have been proven capable of markedly improving the O₂-dependent photodynamic therapy (PDT) of hypoxic tumors, as revealed by the interesting results obtained in pre-clinical animal experiments. The reason is that pH/H₂O₂-driven nanotheranostics can produce great amounts of oxygen in situ and modify the pH via a redox reaction with the acidic H₂O₂ in the tumor microenvironment (TME) [22-26]. For example, the extensively investigated pH/H₂O₂-driven nanotheranostics are MnO₂ nanoparticles and MnO₂-containing nanocomposites, which can improve tumor oxygenation to enhance the efficiency of PDT in the TME [27]. In addition, these nanoparticles and nanocomposites can also be rapidly reduced to water-soluble Mn²⁺ for T1-weighted magnetic resonance imaging (MRI) [28]. Thus, the study of MnO₂based nanotheranostics can offer insight into the development of next-generation intelligent stimuli-responsive nanomedicines for the efficient diagnosis and treatment of cancer in the future. However, MnO2-based nanotheranostics fabricated by current synthetic methods are hindered by the following limitations: 1) the complex and tedious preparation procedure; and 2) the need for drugs or photosensitizers for loading on MnO₂ nanoparticles [29]. Therefore, a simple method must be established for the preparation of MnO₂-based nanotheranostics.

Carbon dots (CDs), a new type of biocompatible fluorescent carbon material, have gained considerable interest in recent years because of their advantageous

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properties, including tunable emission, non-blinking fluorescence (FL) property, high photostability, excellent water solubility, and favorable biocompatibility [30-32]. These characteristics make CDs suitable for various applications, such as imaging agents [33,34], biosensors [35,36], and metal-free catalysts [37-39]. Meanwhile, nanotheranostics based on CDs have been recently developed because of the easy modification of CDs with drugs [40-48]. Moreover, the light-induced singlet oxygen $(^{1}O_{2})$ or heat generation of CDs have been developed recently by our group for PDT or photothermal therapy (PTT) of cancer, indicating their intrinsic capability for nanotheranostics [49-51]. According to the previous report [52], CDs also can function as reducing agents; therefore, the development of CDs with intrinsic photodynamic or photothermal properties as the reducing agent for preparing pH/H2O2-driven, CDs-based nanohybrids may provide new nanotheranostics candidates for simultaneous imaging and therapy of cancer.

Our group recently developed a novel CD PDT agent with a high ¹O₂ quantum yield exceeding 1.3, thus enabling it to act as a novel nanotheranostics for the FL imaging-guided PDT of cancer [49]. In this study, such a CD PDT agent was firstly utilized as the reducing agent to react with KMnO4. The resulting products were then modified with PEG to form CDs/MnO2-PEG nanohybrids. In a normal physiological environment (pH 7.4), the obtained CDs/MnO2-PEG nanohybrids exhibited quenched fluorescence, weak ¹O₂ generation, and low MRI signal. With the introduction of acidic H₂O₂ into the TME of solid tumors (pH 6.5-6.9), the quenched fluorescence of CDs was recovered/enhanced for diagnosis/ monitoring by the decomposition of MnO₂. The simultaneously generated Mn²⁺ can then be utilized as a contrast agent for MRI. Moreover, the MnO₂-H₂O₂ redox reaction at a reduced pH generated massive O₂ in situ, thereby significantly promoting the PDT efficacy of CDs (Scheme 1). Furthermore, complete clearance of CDs/ MnO2-PEG from the body of mice was achieved, indicating the low long-term toxicity and good biocompatibility of the fabricated nanohybrids. These findings highlight the promising potential of CDs/MnO2-PEG nanohybrids for modulating the unfavorable TME and enhancing the PDT response to cancer.

EXPERIMENTAL SECTION

Chemicals and materials

Potassium permanganate (KMnO₄) and hydrogen peroxide (H_2O_2 , 30 wt.% solution) were purchased from J&K Che-

mical Co. Amino group terminated glycol (mPEG-NH₂, M_w =2,000) was acquired from YareBio. All analytical-grade chemicals were commercially available and used without further purification unless indicated otherwise.

Synthesis of CDs/MnO₂

The strongly oxidizing KMnO₄ was one-step reduced by CDs to MnO₂. Then, the positively charged CDs were loaded on the formed MnO2 nanosheets to obtain CDs/ MnO₂ complexes. In brief, we prepared CDs by subjecting polythiophene (PT2) to hydrothermal treatment following our procedure in our previous study [49]. Then, 50 mL of KMnO₄ solution (0.02 mg mL⁻¹) was dropwise added to the activated CDs and stirred at room temperature for 6 h. Untreated KMnO4 and unbound CDs were removed by ultrafiltration at 18,000 rpm for 10 min and washed with water thrice. After ultrasonic dispersion was conducted for 5 min, the large particles of the CDs/ MnO₂ complexes were removed by centrifugation (7,000 rpm) and filtering through 0.22 mm membranes. The loading of the CDs onto the MnO₂ nanosheets was verified by measuring the samples with a UV-vis spectrophotometer. The quenching effect of MnO₂ on the CDs was observed with a luminescence spectrometer.

Synthesis of CDs/MnO₂-PEG

To prepare CDs/MnO₂-PEG NPs, approximately 25 mL of mPEG-NH₂ aqueous solution (25 mg, 1 mg mL⁻¹) was added to the solution under ultrasonication and then stirred vigorously at room temperature for 12 h to guarantee an efficient PEGylation. The resulting CDs/MnO₂-PEG was purified by a three-day dialysis procedure using a dialysis bag (MWCO 5,000 Da) to remove the unreacted mPEG-NH₂.

CDs loading ratio

The UV-vis-NIR spectra of CDs in CDs/MnO₂ nanohybrids were obtained after the nanohybrids were dissolved at pH 6.5 for 6 h. The UV-vis peaks at 492 nm were used to determine the concentration of CDs in the CDs/MnO₂ samples after the absorbance contributed by MnO₂ nanoparticles was removed by dissolving the samples at pH 6.5. The loading efficiency (\emptyset_{CDs}) of CDs can be calculated by the equation: $\emptyset_{CDs}=(Abs_{CDs}/Abs_{CDs/MnO_2-PEG})\times 100\%$, where Abs_{CDs} is the absorbance of CDs, and Abs_{CDs/MnO_2-PEG} is the absorbance of CDs/MnO₂-PEG with the same concentration.

Detection of singlet oxygen

The CDs/MnO₂-PEG samples were loaded in a quartz



Scheme 1 Schematic illustration of CDs/MnO2-PEG nanohybrids as a multimodal theranostics for the MR/FL imaging-guided PDT.

cuvette and exposed to laser at 635 nm with a power density of 100 mW cm⁻² for 10 min at room temperature. Given its high sensitivity to singlet oxygen $({}^{1}O_{2})$, 60 µL of disodium 9,10-anthracendipropionic acid (Na₂-ADPA, 1 mg mL⁻¹) was utilized to detect the $^{1}O_{2}$ generation by free acidic H₂O₂, free CDs/MnO₂-PEG, and CDs/MnO₂-PEG nanohybrids with or without the addition of H₂O₂ $(100 \mu mol L^{-1})$ at pH 6.5 under light irradiation. The produced ¹O₂ was measured based on the amount of reduced Na₂-ADPA absorbance (excitation: 378 nm). In addition, electron spin-resonance spectroscopy (ESR) was performed to qualitatively detect the ¹O₂ generation by CDs/MnO₂-PEG nanohybrids in the presence of acidic H₂O₂ under N₂ atmosphere. 2,2,6,6-Tetramethylpiperidine 1-oxyl was used as the ${}^{1}O_{2}$ capturing agent in the two groups.

Cellular experiments

CDs/MnO₂-PEG incubated *in vitro* with HeLa cells were subjected to imaging. All cells were cultured in normal DMEM medium containing 10% fetal bovine serum (FBS) and 1% penicillin/streptomycin at 37°C under 5% CO₂. The cells were seeded in 35 mm cell culture dishes and stimulated with a PBS solution of CDs/MnO₂-PEG (50 μ g mL⁻¹). After being incubated for 4 h, the cells were washed with PBS twice to remove non-specifically bound CDs/MnO₂-PEG. Changes in fluorescence were then observed after acidic H₂O₂ (100 μ mol L⁻¹) was added. Images were captured using a Nikon C1silaser scanning confocal microscope, which can provide different atmospheres (air or N_2).

To detect ${}^{1}O_{2}$ generation *in vitro*, two groups of HeLa cells were incubated with CDs/MnO₂-PEG (200 µg mL⁻¹) at 37°C under 5% CO₂. After 4 h, one group was mixed with PBS containing acidic H₂O₂ (100 µmol L⁻¹), and the other group was added with a PBS solution with a normal physiological pH. The cells were then incubated with 2,7-dichlorodi-hydrofluoresce in diacetate (DCFH-DA, 1 mg mL⁻¹, 2 µL) for 10 min prior to observation. Cells incubated only with DCFH-DA served as the control group.

For the cell cytotoxicity assay, the cells were seeded into a 96-well plate with a density of 5×10^4 cells per well until the cells adhered and were then incubated with increasing concentrations (12.5, 25, 50, 100, 150, and 200 µg mL⁻¹) of CDs/MnO₂-PEG with or without acidic H₂O₂. After being incubated for 24 h in the dark, the standard thiazolyl tetrazolium (MTT) test was conducted to evaluate the cell viabilities relative to the untreated cells.

For PDT, HeLa cells seeded in 96-well plates were incubated with various concentrations of CDs/MnO₂-PEG for 4 h. The 96-well plates were placed in a glove box that had been ventilated under N_2 atmosphere in advance. This setup was maintained for 30 min before being exposed to 635 nm irradiation at a power density of 100 mW cm⁻² for 30 min under circulating N_2 . Then, the cells were transferred into fresh media and further incubated for 24 h. Then the standard MTT test was conducted to evaluate the cell viability.

Calcein AM and propidium iodide (PI) assays were performed to verify the PDT effect of the CDs/MnO₂-PEG nanoparticles. Two groups of HeLa cells were mixed with CDs/MnO₂-PEG (200 μ g mL⁻¹) at 37°C under 5% CO₂ for 4 h. Then, the cells were stored in a glove box and exposed to N₂ for 30 min. One group was supplemented with PBS containing H_2O_2 (100 µmol L⁻¹) at pH 6.5, and the other group was added with PBS with a normal physiological pH. Afterward, both groups were exposed to 635 nm irradiation at a power density of 100 mW cm⁻² for 0, 3, 6 and 10 min under circulating N_2 . The cells were co-stained with AM/PI for 10 min and washed with PBS twice prior to observation. Cells were only irradiated with $635 \text{ nm} (100 \text{ mW cm}^{-2})$ laser served as the control group. Images were captured with a Nikon C1silaser scanning confocal microscope.

Animal modal

Female nude mice (15–20 g) were purchased from the Center for Experimental Animals, Institute of Process Engineering, CAS in Beijing, China. The procedures for the *in vivo* analyses were approved by the China Committee for Research and Animal Ethics in compliance with the law on animal experimentation. The tumor was induced through the subcutaneous injection of 2×10^6 murine breast cancer 4T1-luc cells in PBS (60 µL) into the right front paw of each mouse. The tumors were allowed to grow for 10 d to reach an approximate size of 100 mm³. For the *in vivo* MRI, CDs/MnO₂-PEG (1 mg mL⁻¹,

 $200 \,\mu\text{L}$) were intravenously (i.v.) injected into the 4T1 tumor-bearing nude mice. After 24 h, MRI of the mice was performed with a 7T MRI scanner (Micro MRI, Varian 7T) at different time points (0, 4, 12 and 24 h).

Fluorescence scans were then performed at various time points (0, 4, 8, 12 and 24 h) post-injection (p.i.) with the use of a Maestro 2 Multispectral Small-animal Imaging System with a 635-nm laser as the excitation source.

To quantitatively evaluate the biodistribution of CDs/ MnO₂-PEG, three tumor-bearing mice were sacrificed at different time intervals p.i. (4 h, 12 h, 24 h, 2 d, 7 d, 14 d and 28 d). The major organs and tissues (i.e., liver, spleen, kidney, heart, lung and the tumor) were weighed, digested, and solubilized in chloroazotic acid under heating for 2 h. After the organic compounds were completely oxidized, each sample was diluted to 10 mL with deionized water. The relative contents of Mn in the different samples were measured thrice by inductively coupled plasma mass spectrometry (ICP-MS).

In vivo O_2 generation was detected by using a photoacoustic (PA) tomography system with an excitation wavelength of 680 nm to measure the oxygenated hemoglobin (HBO₂) content at different time points (0, 4, 8, 12, and 24 h) after the i.v. administration of CDs/MnO₂-PEG (1 mg mL⁻¹, 200 µL). Oxygenation inside the tumor was reflected by the changes in HBO₂ content.

For *in vivo* PDT, the tumor-bearing nude mice were randomly divided into four groups (each group consisting of five mice): (1) without any treatment; (2) 635 nm laser irradiation (100 mW cm⁻²) only; (3) intravenous injection of CDs/MnO₂-PEG (1 mg mL⁻¹, 200 μ L) only; (4) i.v. injection of CDs/MnO₂-PEG (1 mg mL⁻¹, 200 μ L) and 635 nm laser (100 mW cm⁻², 10 min). After being to the different treatments, the tumor volumes were measured by a caliper every 2 d. Tumor size and body weight were monitored every 2 d after treatment. The tumor volume was calculated by the equation: $V = ab^2/2$, where *a* and *b* denote the length and width of the tumor, respectively.

Histopathological examination

The tissues (i.e., heart, liver, spleen, kidney and lung) at 30 d p.i. of CDs/MnO₂-PEG and the tumors of the four groups at 1 d post-treatment were harvested and fixed in 4% formalin solution. Histopathological examinations were performed in accordance with standard laboratory procedures. The tissue or tumor samples were numbered and given blindly to a pathologist for conventional processing and analysis. All samples were embedded in paraffin blocks, sectioned into 4 μ m slices, and mounted onto glass slides. The sections were stained with hematoxylin and eosin (H&E) prior to observation, and images were captured by an optical microscope.

RESULTS AND DISCUSSION

Characterization of CDs/MnO₂-PEG

After the CDs and KMnO₄ underwent a redox reaction, X-ray photoelectron spectroscopy (XPS) was performed to assess the compositions of the resulting samples. The survey spectrum in Fig. S1 indicated the presence of C, Mn and O. As shown in Fig. 1a, the deconvolution of the high-resolution XPS spectra of Mn 2p revealed two peaks at 642.2 and 653.9 eV, which corresponded to the Mn (IV) $2p_{2/3}$ and Mn(IV) $2p_{1/2}$ spin-orbit peaks, respectively, of Mn⁴⁺ with a spin-energy separation of 11.7 eV [53]. This phenomenon confirmed that the positively charged CDs reduced KMnO₄ into MnO₂. To improve their stability, the obtained CDs/MnO₂ composites were modified



Figure 1 Characterizations of CDs/MnO₂-PEG nanohybrids. (a) XPS spectrum of CDs/MnO₂ nanohybrids. (b) A TEM image of CDs/MnO₂-PEG nanohybrids (scale bar: 200 nm). (c) Zeta potentials of CDs, CDs/MnO₂, and CDs/MnO₂-PEG. (d) UV-vis spectra of KMnO₄, CDs, and CDs/MnO₂-PEG. (e) FL spectra of CDs/MnO₂-PEG and CDs. (f) Optical photographs of CDs/MnO₂-PEG in water, PBS, FBS, and DMEM at 0 d (top) and 7 d (bottom).

with aminoterminated PEG ($M_w \approx 2,000$) through Mn–N coordinate bonding (Fig. S2) [54]. As revealed by transmission electron microscopy (TEM) in Fig. 1b, the CDs/ MnO₂-PEG nanohybrids displayed an approximate average size of 180 nm. The high-resolution TEM image of the CDs loaded onto MnO₂ (Fig. S3a) revealed that the inter planar distance was approximately 0.31 nm, indicating that the reacted CDs retained their original crystallinity. Energy-dispersive X-ray spectroscopy spectrum also proved that the elements (i.e., Mn, C, O, etc.) were present in the CDs/MnO2-PEG nanohybrids (Fig. S3b). As shown in Fig. 1c, the zeta potential changed from +37.1 mV for CDs to -21.7 mV for CDs/MnO₂ because of the formation of hydrogen bonds in the aqueous solution and the transformation of the surface charge by the MnO₂ nanoparticles. After being stirred with aminoterminated PEG, the zeta potential of the CDs/MnO₂-PEG nanohybrids increased to -17.9 mV, indicating the successful formation of CDs/MnO2-PEG nanohybrids, which were favorable for long blood circulation.

UV-vis spectra (Fig. 1d) revealed that the four characteristic peaks of $KMnO_4$ (between 500 and 550 nm) disappeared after the reaction in the presence of the po-

sitively charged CD PDT agent. However, a new broad absorbance band appeared in the range of 300-800 nm, implying the surface plasmon band of generated MnO₂. As shown in Fig. 1e, the fluorescence of the CDs at the maximum emissive peak of 690 nm was significantly absorbed by the MnO₂ because of the significant overlap between the emission spectrum of CDs and the absorption spectrum of MnO₂. This phenomenon suggested the CDs/MnO₂-PEG nanohybrids displayed faint fluorescence signal under normal physiological conditions. The loading efficiency of CDs onto CDs/MnO2-PEG was evaluated by obtaining the UV-vis spectra of the CDs at 492 nm. The maximum loading efficiency reached up to approximately 65% (Fig. S4). In addition, the obtained CDs/MnO₂-PEG was highly dispersible and stable in water, PBS, FBS, and DMEM. No obvious aggregation was observed even after a prolonged time of 7d (Fig. 1f).

pH/H₂O₂-responsive CDs/MnO₂-PEG nanohybrids

In neutral solutions (pH 7.4), MnO_2 catalyzed the disproportionation reaction of H_2O_2 into H_2O and O_2 [27]. By contrast, in acid solutions of H_2O_2 (pH 6.5), H_2O_2 was oxidized to oxygen while MnO_2 nanoparticles were bro-

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Figure 2 pH/HO₂-response of CDs/MnO₂-PEG nanohybrids. (a) UV-vis-NIR spectra and (b) FL spectra of CDs/MnO₂-PEG nanohybrids at pH 6.5 and 7.4, respectively. (c) Simultaneous O₂ generation in acidic H₂O₂ solutions (10 mmol L⁻¹) after adding different concentrations of (MnO₂: 0, 50, and 100 mmol L⁻¹) of CDs/MnO₂-PEG. (d) Characteristic ¹O₂-induced signal in the ESR spectra after 635 nm laser irradiation (100 mW cm⁻²) under N₂ atmosphere, in the absence of presence of H₂O₂ (100 µmol L⁻¹). (e) T1-weighted MR images of various concentrations of CDs/MnO₂-PEG solutions incubated at different pH values (7.4 and 6.5) for 6 h prior to MRI. (f) Concentration-dependent T1 relaxation rates. The longitudinal relaxivities (r1) were determined to be 0.4763 and 7.0674 (mmol L⁻¹)⁻¹ s⁻¹ for CDs/MnO₂-PEG nanohybrids at pH 7.4 and 6.5, respectively.

ken down and Mn^{4+} was rapidly reduced into colorless Mn^{2+} [55–58], leading to the release of CDs, as the following reaction (1)

$$MnO_2 + H_2O_2 + 2H^+ \rightarrow Mn^{2+} + 2H_2O + O_2 \uparrow.$$
(1)

As shown by the comparison of the absorption and fluorescence spectra of CDs/MnO₂-PEG at pH 7.4 and 6.5 in Fig. 2a and b, the initial absorption spectrum of the CDs appeared, and the markedly enhanced FL intensity of the CDs was also observed. Then, dissolved O₂ in the reacted solution was quantitatively tested by an oxygen probe (ST300D, portable Dissolved Oxygen Meters, Chang zhou OHAUS Instrument Factory) with increasing concentrations of CDs/MnO₂-PEG nanohybrids (MnO₂: 0, 50, 100 mmol L^{-1}) which resulted in the rapid generation of oxygen in acidic H_2O_2 (10 mmol L⁻¹) as shown in Fig. 2c, and gas bubbles were clearly observed at 100 mmol L^{-1} of H_2O_2 (Fig. S5). These findings indicated that the CDs/MnO₂-PEG can be used for the quantitative detection of acidic H₂O₂ by examining the FL intensity of CDs and the production of oxygen. As shown in Fig. S6, the size of CDs/MnO2-PEG nanohybrids also changed from ~180 nm (at pH 7.4) to 5-20 nm (at pH 6.5), indicating the decomposition of MnO₂ nanoparticles in the acidic condition.

The capability of the CDs/MnO₂-PEG nanohybrids to produce ¹O₂ in response to acidic H₂O₂ was investigated. Specifically, the ¹O₂ generation of the CDs/MnO₂-PEG nanohybrids upon 635 nm irradiation with or without 100 μ mol L⁻¹ acidic H₂O₂ (pH 6.5) was tested by using Na₂-ADPA as an indicator. As shown in Fig. S7, without the addition of H₂O₂, slight ¹O₂ generation was observed because of the CDs quenching by MnO₂. However, the markedly decreased ADPA absorption at 378 nm implied that more ¹O₂ were generated after acidic H₂O₂ was added. In the acidic H₂O₂ solution (pH 6.5), the recovered emission of the CDs resulted in the further enhancement of ¹O₂ generation because of the dissociation and reduction of MnO₂ by the acidic H₂O₂. Then, ESR was used to measure the light-responsive ¹O₂ generation of the CDs/ MnO₂-PEG nanohybrids under N₂ atmosphere. As shown in Fig. 2d, the CDs/MnO₂-PEG nanohybrids exhibited a strong ESR signal only upon 635 nm laser irradiation (100 mW cm^{-2}) with the addition of acidic H₂O₂, indicating the efficient ¹O₂ production under such conditions. By contrast, no obvious ¹O₂ generation was observed under neutral conditions (pH 7.4), suggesting that the significantly improved PDT efficiency of the



Figure 3 *In vitro* imaging and PDT. (a) Confocal microscopy images of HeLa cells incubated with CDs/MnO₂-PEG nanohybrids (200 μ g mL⁻¹) for 4 h and added with acidic H₂O₂ at a bright field and excitation at 635 nm (scale bar: 10 μ m). (b) Confocal images of HeLa cells incubated with DCFH-DA, CDs/MnO₂-PEG nanohybrids at pH 7.4 and 6.5 at different irradiation times (0, 3, 6, 10 min). (c) Relative viabilities of HeLa cells after incubation with CDs/MnO₂-PEG nanohybrids for 24 h in the dark, at pH 7.4 and 6.5 upon light irradiation (635 nm,100 mW cm⁻², 10 min) under N₂ atmosphere, respectively. (d) Fluorescence images of calcein AM and PI co-stained HeLa cells incubated with CDs/MnO₂-PEG nanohybrids (200 μ g mL⁻¹) at pH 7.4 and 6.5 with 635 nm irradiation (100 mW cm⁻²) for 0, 3, 6 and 10 min (scale bar: 150 μ m).

 CDs/MnO_2 -PEG nanohybrids was due to the enhanced O_2 generation resulting from the MnO_2 -catalyzed decomposition of H_2O_2 under hypoxia.

The T1-weighted MRI of the CDs/MnO₂-PEG nanohybrids were further examined under neutral (pH 7.4) and acidic conditions (pH 6.5). As shown in Fig. 2e and f, the as-prepared CDs/MnO2-PEG nanohybrids displayed concentration-dependent T1-weighted MRI signals under both neutral and acidic conditions. However, the CDs/ MnO₂-PEG nanohybrids under acidic conditions (pH 6.5) exhibited a more strongly enhanced T1-weighted MRI than those under neutral conditions (pH 7.4) because of the reduction of MnO_2 into Mn^{2+} by the acidic H_2O_2 [59]. The r1 value of the CDs/MnO₂-PEG nanohybrids drastically increased from 0.4763 (mmol L^{-1})⁻¹ s⁻¹ under neutral conditions (pH 7.4) to 7.0674 (mmol L^{-1})⁻¹ s⁻¹ under acidic conditions (pH 6.5), thereby enhancing the r1 value by 15-fold in the presence of H_2O_2/H^+ . This finding indicated the potential of a MRI-guided PDT of solid tumors. Thus, CDs/MnO₂-PEG can act as a pH/H₂O₂-

driven, turn-on nanotheranostics for the multimodal imaging-guided PDT of cancer.

In vitro imaging and PDT

In vitro experiments were conducted to determine whether the as-prepared CDs/MnO₂-PEG nanohybrids can serve as an effective, pH/H2O2-driven, turn-on nanotheranostics for the multimodal imaging-guided PDT of cancer. Confocal images (Fig. 3a) showed that HeLa cells treated with CDs/MnO2-PEG nanohybrids under neutral conditions (pH 7.4) exhibited weak fluorescence because of the quenching effect of MnO₂ nanoparticles on CDs. However, the fluorescent signals of CDs were significantly increased after acidic H₂O₂ was added because of the MnO_2 nanoparticles were disintegrated [60], thereby releasing of the red emissive CDs. Z-Stack imaging further demonstrated that the CDs/MnO₂-PEG nanohybrids were localized in the cytoplasm rather than the cell membrane (Fig. S8). These results proved that the asprepared CDs/MnO₂-PEG nanohybrids can be applied as

a pH/H_2O_2 -responsive, turn-on fluorescence imaging agent *in vitro*.

To further investigate the *in vitro* PDT effect of the CDs/MnO₂-PEG nanohybrids under a TEM with acidic H_2O_2 conditions (pH 6.5), the capability of CDs/MnO₂-PEG nanohybrids for ¹O₂ generation was examined *in vitro*. 2,7-Dichlorodi-hydrofluorescein diacetate (DCHF-DA), which is a standard FL indicator for ROS, was used to track ¹O₂ generation under N₂ atmosphere. As shown in Fig. 3b, the cells incubated with CDs/MnO₂-PEG nanohybrids under neutral condition (pH 7.4) displayed week fluorescence under N₂, implying that scarce ¹O₂ production occurred. However, under acidic conditions, the green fluorescence was further enhanced by continuous incubation for 10 min, indicating that was effectively generated.

An in vitro viability assay was performed on HeLa cells under N₂ atmosphere to investigate the PDT effect of CDs/MnO₂-PEG nanohybrids. As shown in Fig. 3c, the CDs/MnO₂-PEG nanohybrids exhibited negligible adverse effects on the viabilities of the HeLa cells at the tested concentrations ranging from 6.25 to 200 μ g mL⁻¹ with or without acidic H_2O_2 , thus indicating the favorable biocompatibility of the CDs/MnO₂-PEG nanohybrids. However, in the presence of acidic H₂O₂ (pH 6.5), a significant concentration-dependent cytotoxicity of the CDs/MnO₂-PEG nanohybrids was observed, and a 95% inhibitory effect was achieved at 200 µg mL⁻¹ of CDs/ MnO₂-PEG nanohybrids upon 635 nm laser irradiation. By contrast, the cell viability slightly decreased at pH 7.4. The in vitro PDT effect of the CDs/MnO2-PEG nanohybrids under a TME with acidic H_2O_2 conditions (pH 6.5) under N₂ atmosphere was verified by conducting a calcein AM and PI co-staining assay. The observed green fluorescence indicated that few cell deaths occurred at pH 7.4 even after the irradiation time was prolonged to 10 min (Fig. 3d). However, at pH 6.5, a homogeneous red fluorescence was observed at 200 µg mL⁻¹ of the CDs/ MnO₂-PEG nanohybrids upon 635 nm laser irradiation, indicating complete cell death. These in vitro results suggested that CDs/MnO2-PEG can act as an effective pH/H₂O₂-driven, turn-on nanotheranostics for the imaging-guided PDT of cancer under an acidic and hypoxia TME.

In vivo MR/FL imaging and real-time biodistribution

To investigate the feasibility of CDs/MnO₂-PEG nanohybrids for *in vivo* MRI, mice bearing subcutaneous 4T1 tumors of approximately 100 mm³ were selected as models. *In vivo* MRI was performed at different times (0, 4, 12, and 24 h) with the use of a 7T MR scanner. As shown in Fig. 4a and b, a weak MR signal was observed in the tumor site at 4 h p.i. of CDs/MnO₂-PEG nanohybrids. This finding indicated that the CDs/MnO₂-PEG nanohybrids accumulated in the tumor area because of the enhanced permeability and retention (EPR) effect and the gradual reduction of MnO₂ into Mn²⁺ in the mildly acidic TME [61,62]. The MR signal intensity at the tumor site reached a fastigium at 12 h p.i. and decreased at 24 h p.i. The T1 signals observed in the kidneys reached their maximum at 12 h p.i. (Fig. 4c–e), suggesting the renal clearance of the Mn²⁺ decomposed from MnO₂. By contrast, no obvious T1 signal changes were observed in the liver because of its neutral environment.

Subsequently, the FL imaging of a tumor-bearing mouse was conducted in vivo at different time intervals after the i.v. injection of CDs/MnO2-PEG nanohybrids $(1 \text{ mg mL}^{-1}, 200 \mu\text{L})$. Fig. 4f and g show that the FL intensity at the tumor site rapidly increased within 12 h p.i., and then plateaued. Ex vivo imaging of major organs and tumors collected from mice at 12 h p.i. was further examined. As shown in Fig. 4h, the excised tumor tissue displayed strong FL intensity, whereas the liver and the kidney presented very low signals. These findings agreed well with the in vivo MRI behavior. Therefore, both MR and FL imaging indicated that 12 h p.i. was the optimal time for the subsequent PDT. And the CDs/MnO2-PEG nanohybrids were responsive only in the TME but maintained an "off" state in normal physiological environments, suggesting that CDs/MnO2-PEG nanohybrids can be used as a tumor-targeted, multifunctional theranostic for eliminating the damage toward normal tissues during PDT.

The real-time biodistribution of CDs/MnO2-PEG nanohybrids at different times was examined by ICP-MS, which is a sensitive method that can accurately detect the content of Mn in different organs. For ex vivo analyses, tumor-bearing mice were i.v. injected with CDs/MnO2-PEG nanohybrids (1 mg mL⁻¹, 200 µL) and then sacrificed at different time intervals (1 h, 12 h, 24 h, 2 d, 7 d, 14 d, and 28 d) to harvest the tumors. Fig. 4i illustrates the trends of biodistribution and metabolic clearance of the Mn ions in the tumors. Analysis of the statistical results revealed that CDs/MnO2-PEG nanohybrids accumulated in the tumor through the EPR effect. This accumulation increased within 1-12 h p.i., plateaued at 12 h p.i., and then steadily decreased over time. In addition, CDs/MnO2-PEG nanohybrids also accumulated in the liver, the lungs, the kidneys, the spleen, and the heart; the accumulation likewise steadily decreased over time,

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Figure 4 *In vivo* MR/FL imaging and biodistribution. (a) Time-dependent T1-weighted MR images (transverse section) and (b) T1-weighted MR signals of the tumor. (c) *In vivo* T1-weighted MR images (transverse section) of the kidney and liver in the same mouse. (d) T1-weighted MR signals of the liver in the same mouse. (e) T1-weighted MR signals of the kidney in the same mouse. (f) Time-dependent FL images of tumor-bearing nude mouse after an i.v. injection of CDs/MnO₂-PEG nanohybrids (1 mg mL⁻¹, 200 μ L). (g) FL signals in the tumor before, 4, 8, 12, and 24 h p.i. of CDs/MnO₂-PEG nanohybrids. (h) *Ex vivo* FL images of major organs and tumor dissected from the same mice at 12 h p.i. tumor, liver, spleen, kidney, heart and lung. (i) Time-dependent content of Mn in the *ex vivo* tumor and other organs after an i.v. injection of CDs/MnO₂-PEG nanohybrids. Inset is the blood circulation curve of the CDs/MnO₂-PEG nanohybrids in mice that was measured based on the content of Mn in blood at different time points p.i. Error bars are based on the standard deviation of three mice.

resulting in the complete removal of the CDs/MnO_2 -PEG nanohybrids from the main organs and tumors after 4 weeks.

To evaluate the blood circulation of the CDs/MnO₂-PEG nanohybrids, blood samples of tumor-bearing mice were extracted at various time points (15 min, 30 min,

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Figure 5 *In vivo* PDT. (a) PA images of 4T1 solid tumors measured based on HBO₂ at different time points p.i. of CDs/MnO₂-PEG nanohybrids. (b) Average enhanced tumor vascular saturated O_2 p.i. of CDs/MnO₂-PEG nanohybrids. (c) Time-dependent tumor growth curves of mice (n = 5) observed after various treatments. (d) Relative body weights of the nude mice after different treatments. (e) Representative photographs of the four groups after different treatments. (f) H&E-stained tumor slices from different groups collected 24 h after light irradiation. (scale bar: 100 µm).

1 h, 2 h, 4 h, 8 h, 12 h, 24 h) p.i. and then quantitatively measured by ICP-MS to precisely determine the Mn content in the blood. As shown by the inset in Fig. 4i, the

blood level of CDs/MnO₂-PEG decreased gradually over time, as in a two-compartment model. Then, secondary exponential fitting was utilized to calculate the first ($t_{1/2}$ (a)) and second ($t_{1/2}$ (b)) phases of the circulation halflives, which were approximately 0.57 and 8.15 h, respectively. Such long blood circulation of CDs/MnO₂-PEG nanohybrids should be favorable for effective tumor accumulation.

In vivo O₂-enhanced PDT

To further demonstrate the ability of CDs/MnO₂-PEG nanohybrids to ameliorate tumor hypoxia, PA imaging was performed to detect HBO₂ content in the tumors at different time points after the i.v. injection of the CDs/ MnO₂-PEG nanohybrids. As shown in Fig. 5a and b, the mean intensities of the HBO₂ signs increased with time and reached a maximum value at approximately 12 h p.i., indicating the successful alleviation of tumor hypoxia due to the MnO₂-triggered decomposition of H₂O₂ into O₂. Such enhanced oxygenation in the TME was favorable for increasing the efficiency of in vivo PDT. Then, the in vivo PDT efficacy of the CDs/MnO2-PEG nanohybrids was evaluated. Balb/c mice bearing subcutaneous 4T1 tumors were divided into four groups: Group 1: without any treatment; Group 2: laser irradiation only; Group 3: CDs/ MnO_2 -PEG nanohybrids (1 mg mL⁻¹, 200 µL) without laser irradiation; Group 4: CDs/MnO2-PEG nanohybrids $(1 \text{ mg mL}^{-1}, 200 \mu\text{L})$ with laser irradiation. At 12 h p.i., the mice in Groups 2 and 4 were irradiated by 635 nm laser (100 mW cm⁻², 10 min). The tumor volumes of all the mice were measured by a caliper every 2 d. As shown in Fig. 5c, the mice in Group 4 exhibited a significant suppression of tumor growth, demonstrating a remarkably improved therapeutic efficacy. By contrast, all tumor tissues of the mice in Groups 1, 2, and 3 continued to grow (Fig. 5e). The in vivo toxicity of the CDs/MnO₂-PEG nanohybrids was also investigated by monitoring the weight change of the mice during the study period. As shown in Fig. 5d, no abnormal changes were observed in any of the groups, indicating the absence of acute toxicity of the CDs/MnO₂-PEG nanohybrids in vivo.

The microscopy images of the H&E-stained tumor slices further revealed that the PDT triggered by CDs/ MnO₂-PEG nanohybrids severely damaged the tumor cells, whereas the tumor cells in the control groups mostly retained their regular morphology (Fig. 5f). The H&E-stained slices of the main organs (heart, liver, spleen, lung and kidney) were also evaluated for histology analysis at 30 d post i.v. injection of CDs/MnO₂-PEG nanohybrids. Compared with the cells of a healthy mouse, those in the examined tissues retained their normal morphology (Fig. S9), showing neither significant inflammation nor damage. Overall, these results illustrate that CDs/MnO₂-

PEG nanohybrids are a highly effective, scarcely biotoxic PDT agent *in vivo*.

CONCLUSIONS

In summary, pH/H₂O₂-responsive CDs/MnO₂-PEG nanohybrids were successfully synthesized via the redox reaction between the CD PDT agent and KMnO4 followed by PEGylation. The as-prepared CDs/MnO₂-PEG nanohybrids possessed the following characteristics: (1) quenched fluorescence, weak singlet oxygen generation, and low MRI signal in the normal physiological environment; (2) synchronously enhanced fluorescence, singlet oxygen generation, and MRI signal in the tumor microenvironment; and (3) low toxicity and complete clearance from the body. In vitro and in vivo analyses reveal that the low-toxic CDs/MnO2-PEG nanohybrids can be applied as pH/H₂O₂-driven, turn-on nanotheranostics for the concurrent bimodal MR/FL imaging and oxygen-elevated PDT of solid tumors if the TME is regulated. This work offers a new nanotheranostic candidate for modulating the unfavorable TME, particularly for the targeted PDT of cancer through precise positioning and oxygen generation.

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Author contributions Wang P and Ge J supervised the project. Chen S designed and carried out the experiments, analyzed the data and wrote the manuscript. Jia Q, Zheng X and Wen Y helped with the synthesis of the CDs/MnO_2 -PEG. Liu W and Zhang H helped with the photodynamic therapy.

Conflict of interest The authors declare that they have no conflict of interest.

Supplementary information Supporting data are available in the online version of the paper.

ARTICLES



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聚乙二醇化的碳点光敏剂/二氧化锰纳米复合物:一种新型酸/过氧化氢响应的纳米光诊疗剂

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摘要 缺氧、过酸和过量的活性氧(如过氧化氢)是肿瘤微环境的三个主要显著特征,针对这些特征可设计酸/过氧化氢响应的诊疗剂用于 增强肿瘤靶向光动力治疗效果.本文首次利用新型碳点光敏剂还原高锰酸钾制备碳点/二氧化锰,然后利用聚乙二醇修饰形成水溶性的多 功能复合物.在正常生理环境中,该聚乙二醇化的碳点光敏剂/二氧化锰纳米复合物的荧光大部分被淬灭,光生单线态氧的能力被抑制,也 不具备磁共振成像的能力.但是在肿瘤微环境中,由于二氧化锰对酸/过氧化氢的高灵敏响应,碳点荧光恢复,同时可产生单线态氧,能够 检测到强的磁共振成像信号.因此,该水溶性聚乙二醇化的碳点/二氧化锰复合物可用于肿瘤微环境响应的荧光/磁共振双模态成像介导的 光动力治疗,拓展了正电碳点作为新型光诊疗试剂在调节肿瘤微环境和增强光动力治疗方面的应用.