Correction to: The calculation of the solubility of metal hydroxides, oxide-hydroxides, and oxides, and their visualisation in logarithmic diagrams

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We regret the misprint of the last two equations given in Table 2 in the original article.

The correct equations for a metal oxide of the specific stoichiometry M_2O_3 are as follows:

$$\log S_{M_2O_3} = \log 0.5 + 0.5 \log K_{\text{sol},M_2O_3} + 3pK_w - 3pH$$

$$\log S_{M_2O_3} = \log 0.5 + 0.5 \log K_{\text{sol},M_2O_3} + 42.0 - 3pH$$

The original article can be found online at <https://doi.org/10.1007/s40828-015-0006-0>.

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