

# Demineralization of Gondwana coal with *Pseudomonas mendocina* strain B6-1: a case study of coal from Gopinathpur top and bottom seams of Mugma mine, Dhanbad, Jharkhand (India)

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**Abstract** In the present investigation an attempt has been made to demineralize the Gondwana coal of Gopinathpur top and bottom seams of Mugma mine, Raniganj coalfield, Dhanbad with the help of *Pseudomonas mendocina* strain B6-1. The change in the amount of ash yield and decrease in the concentration of selected minor elements like Na, K, Mn and Ca and environmentally sensitive selected trace elements such as Cd, Pb, Se, Ni, Mn, and Zn have been studied as a function of time of bacterial treatment as well as with variation in the bacterial biomass. After 28 days of bacterial treatment there was variable amount of decrease observed in ash yield as well as in the concentration of minor and trace elements. The removal of the elements was further enhanced with the increase in the bacterial biomass from 10 to 25 mg/mL. Due to over exploitation of superior grade coals in the country, the remaining coal resources, available for current use, are inferior in grade and contain high level of impurities and there is ample scope of bio-beneficiation of these coals using bacterial biomass.

Keywords Pseudomonas mendocina · Gondwana coal · Minor and trace elements · Demineralization

# **1** Introduction

Coal is an important source of energy and is mainly used to generate electricity in the thermal power plants. Gondwana coals of India contribute more than 99 % of total coal resource of the country and are with medium to high ash yield. Due to over exploitation of superior grade coals, the remaining resources are inferior in grade having high level of impurities. Moreover, the Gondwana coals being of drift origin contain relatively high inorganic matter. These inorganic matters get associated with coal during various stages and have different modes of occurrences such as (a) dissolved salts and other inorganic substances in the

Prakash K. Singh prakashbhu@rediffmail.com pore water of coal, (b) inorganic elements which are incorporated within the organic compounds of macerals, and (c) crystalline or non-crystalline discrete inorganic particles representing actual mineral components. During mining activities the minerals derived from the non-coal bands from intra-seam further add to the impurities. Removal of those inorganic matters which are chemically bonded with the organic matter is difficult through physical cleaning and has become a matter of concern for the scientists. Besides, there are several elements which occur in association of organic and inorganic constituents as major (>1 wt%), minor (1-0.1 wt%) or trace (<0.1 wt%) elements (Vassilev and Vassileva 1997). The elements like As, Be, Cd, Cr, Cu, Pb, Mn, Hg, Mo, Ni, Sr, U, V, and Zn are considered to be environmentally sensitive (Pickhardt 1989). Significant contributions on the geochemistry of trace elements have been made by several scientists (Finkelman 1995; Ward et al. 2003; Ren et al. 2004; Seredin 2004; Eskenazy 2009; Zhao et al. 2014; Singh et al. 2015a, b). Geochemistry of trace elements of Indian coals has been studied by Singh et al. (1983, 2015a) and Prachiti

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et al. (2011). Since studies on the demineralization of coal using bacterial biomass have not been carried out in details, there is much scope of such investigations. Contributions on limited numbers of coal samples have been made by Singh et al. (2012, 2014a, b, 2015c).

In the present investigation an attempt has been made to demineralize the Gondwana coal of Gopinathpur seam from Raniganj coalfield with the help of *Pseudomonas mendocina* strain B6-1. The change in the amount of the ash yield and change in the concentration of environmentally sensitive selected trace elements have been studied as a function of time of bacterial treatment as well as with variation in the bacterial biomass.

### 2 Methods of study

The coal samples for present investigation were collected from the Gopinathpur top and bottom seams of Mugma mine, Nirsa block. These Gondwana coals belong to Barakar Formation of Raniganj coalfield, Dhanbad district, Jharkhand. Two coal samples from Gopinathpur bottom seam (GB-1and GB-2) and one sample from Gopinathpur top seam (GT-1) have been subjected to bacterial treatment and various analyses.

### 2.1 Coal petrography

The maceral analysis has been carried out under Leitz Orthoplan Pol Microscope in Coal & Organic Petrology Lab, Banaras Hindu University. Oil recommended by Leitz (DIN 58889) has been used for the microscopy. The lineto-line and point-to-point spacings were maintained at 0.4 mm and more than 500 counts were taken on each sample. The methodology and recommendations given by Taylor et al. (1998) and ICCP (2001) were followed.

#### 2.2 Proximate analysis

The proximate analysis has been carried out as per ISI specifications (BIS 2003) to determine the contents of moisture, ash, volatile matter and fixed carbon. The analysis was carried out in Coal & Organic Petrology Lab, Banaras Hindu University.

#### 2.3 Minor and trace element analysis

One gram (dry weight) of Gopinathpur seam coal sample was transferred in a digestion vessel and 10 mL of digestion mixture (10 part conc.  $HNO_3$  and 1 part  $HClO_4$ ) was added to it. The mixture was refluxed for 30 min and the step was repeated thrice so that no brown fumes were given off by the sample. It was filtered using Whatman (No. 41)

filter paper. The digested samples were rinsed with 1 % Conc.  $HNO_3$  and transferred in a separate test tube. The volume was made up to 20 mL. Concentrations of Cd, Pb, Se, Ni, Mn, and Zn elements were determined using Perkin-Elmer Inductively Coupled Plasma optical emission spectrometer (ICP-OES), model Optima 7000 DV, in the School of Life Sciences, Banaras Hindu University. Concentrations of other elements like Na, K and Ca were determined using Systronics Flame Photometer 128 in the Department of Botany, Banaras Hindu University.

## 2.4 Bacterial treatment

## 2.4.1 Isolation and identification of bacteria

The bacteria used in the present investigation were isolated from Rajmahal coals and the detailed process has been discussed by Singh et al. (2014a, b). Identification of bacteria was carried out at Indian Institute of Vegetable Research (IIVR), Varanasi (Singh et al. 2014a, b). The bacterium was identified as *P. mendocina* strain B6-1.

# 2.4.2 Mass cultivation of Pseudomonas mendocina strain B6-1 bacteria

The isolated bacterial colonies have been grown in a growth medium and the growth of the colonies was measured individually at 660 nm with the help of spectrophotometer. The colony with maximum growth has been picked up, purified and transferred into liquid basal salt medium for mass cultivation. The medium had a composition of: glycerol 0.2 %, KH<sub>2</sub>PO<sub>4</sub> 0.244 %, NaHPO<sub>4</sub> 0.577 %, NH<sub>4</sub>Cl 0.2 % and MgCl<sub>2</sub> 0.02 %.

# 2.4.3 Immobilization of Pseudomonas mendocina strain B6-1 bacteria

The exponential phase bacterial cells have been collected through centrifugation at 10000 rpm for 10 min and repeated washing. The final pellet was collected for further use. Mixture of 5 % sodium alginate solution was poured into broth of basal salt medium and was sterilized. Bacterial beads were prepared by adding 10 mg/mL of bacterial biomass into sodium alginate solution and this mixture was pumped drop wise into 0.2 M CaCl<sub>2</sub> solution in sterilized condition under laminar flow. The bacterial beads thus prepared were harvested and resuspended in 100 mL growth medium in elementary flask of 250 mL. The same method was adopted for the preparation of bacterial beads using 25 mg/mL biomass.

SI.	Petrographic (	components (vol	1%)											
No.	Vitrinite				Liptinite				Inertinite					
	Collotelinite	Vitrodetrinite	Collodetrinite	Total vitrinite	Sporinite	Cutinite	Resinite	Total I liptinite	Macrinite	Semifusinite	Fusinite	Secretinite	Inertodetrinite	Total inertinite
GB1	17.0	0.0	34.1	51.1	9.9	0.0	0.6	10.5 2	2.2	12.4	3.7	0.0	1.9	20.1
	(20.8)	(0.0)	(41.7)	(62.5)	(12.1)	(0.0)	(0.8)	(12.9) (	(2.7)	(15.2)	(4.5)	(0.0)	(2.3)	(24.6)
GB2	11.2	0.0	35.6	46.8	6.1	0.0	0.3	6.4	1.6	11.2	5.4	0.3	2.6	21.2
	(15.1)	(0.0)	(47.8)	(62.9)	(8.2)	(0.0)	(0.4)	(8.6) (	(2.2)	(15.1)	(7.3)	(0.4)	(3.4)	(28.4)
GT1	22.4	2.6	20.6	45.6	8.2	1.5	0.3	10.0 (	0.0	9.7	7.4	0.3	7.1	25.3
	(27.6)	(3.3)	(25.5)	(56.4)	(10.2)	(1.8)	(0.4)	(12.4) (	(1.1)	(12.0)	(9.1)	(0.4)	(8.7)	(31.3)
Mean	16.9	0.9	30.1	47.8	8.1	0.5	0.4	0.6	1.6	11.1	5.5	0.2	3.8	22.2
	(21.2)	(1.1)	(38.3)	(60.6)	(10.2)	(0.6)	(0.5)	(11.3) (	(2.0)	(14.1)	(1.0)	(0.3)	(4.8)	(28.1)
SI.	Petrograph	ic components (	(vol%)		Proxir	nate comp	onents (wt	(%)						
No.	Mineral m	atter			Moist	ure As	h Vo.	latile matter	Fixed c.	arbon Vola	tile matter	(daf basis)	Fixed carbon	(daf basis)
	Sulphide	Carbonate	Argillaceous	Total mm										
GB1	0.0	0.6	17.6	18.3	0.33	23.	41 25.	9	50.66	33.5	7		66.43	
GB2	0.0	7.7	17.9	25.6	0.4	26.	.95 25.	62	47.03	35.20	9		64.74	
GT1	0.3	0.6	18.2	19.1	0.49	27.	30 20.	43	54.49	28.2	6		71.71	
Mean	0.1	3.0	17.9	21.0	0.40	25.	.88 23.	88	50.72	32.3′	7		67.62	

Table 1 Petro-chemical attributes of Gondwana coal samples of Gopinathpur seam, Mugma Mine, Nirsa block, Dhanbad



Fig. 1 Petrographic composition of Gopinathpur coal samples



Fig. 2 Mean values of petrographic elements (vol%) in Gopinathpur coal

#### 2.4.4 Method of treatment

Demineralization of coal samples of Gopinathpur seam was carried out with *P. mendocina* strian B6-1. These coal samples have been crushed to 70 mesh size. Two sets of coal samples (about 12 g) were taken (one for control and

the other for treatment) in 50 mL basal salt medium and the pH was adjusted at five which was maintained with the help of 1 N HCl. The experiment was performed at room temperature in Erlenmeyer flasks of 100 mL, with volume of 50 mL basal salt medium. Each coal sample was inoculated with 100 number of bacterial beads of P. mendocina strian B6-1 and all the treated samples had a respective control (Blank), i.e. an assay with the same operation conditions but without inoculation. The experiments have been performed with coal samples of Gopinathpur seam of Mugma mine. The treated samples of every 2nd, 4th, 8th, 12th, 16th, 20th, 24th, and 28th day were taken out for the analysis. The pH and oxidation reduction potential (ORP) were monitored throughout the experiment. The solid fraction obtained after filtering it with Whatman filter paper was dried at room temperature and subjected to proximate analysis.

# 3 Results and discussion

#### 3.1 Petro-chemical attributes

Gopinathpur coals are dominated by vitrinite group of macerals (45.6 %–51.1 %; 56.4 %–62.9 % mmf basis) which is contributed mainly by collodetrinite and little by collotelinite and vitrodetrinite. Inertinite (20.1 %–25.3 %; 24.6 %–31.3 % mmf basis) and liptinite (6.4 %–10.5 %; 8.6 %–12.9 % mmf basis) occur in low concentrations. Inertinite comprises of semifusinite, fusinite, macrinite and inertodetrinite while liptinite comprises of sporinite, resinite and cutinite in these coals. Mineral matter occurs in moderately high concentration (18.3 %–25.6 %) which comprises mainly of clay minerals while carbonates and sulphides occur in low concentration. The details of the petrographic composition of Gopinathpur coals are

Table 2 Selected minor and trace elements investigated in coal samples of G	Sopinathpur top and bottom seams (ppm)	)
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Element	World average/Clarke value	GB-1	GB-2	GT-1	Mean
Ca <sup>a</sup>	10000	918.2	1191.5	1721.0	1276.9
K <sup>a</sup>	100	1189.5	1189.5	5573.0	2650.7
Na <sup>a</sup>	200	5792.0	5618.0	4534.0	5314.7
$Cd^b$	$0.2 \pm 0.04$	Nd	1.8	Nd	1.8
Pb <sup>b</sup>	$9 \pm 0.7$	13.2	1.8	9.80	8.3
Se <sup>b</sup>	$1.6 \pm 0.1$	6.80	9.0	Nd	7.9
Ni <sup>b</sup>	$17 \pm 1$	18.50	8.2	32.0	19.6
Mn <sup>b</sup>	$71 \pm 5$	19.10	24.0	Nd	21.5
Zn <sup>b</sup>	$28 \pm 2$	9.00	13.3	31.0	17.8

Nd not determined

<sup>a</sup> World average (After Valković 1983)

<sup>b</sup> Coal Clarke values for bituminous coal (Ketris and Yudovich 2009)

Day	Ash yield (wt%) (GB-1)	Ash reduction (%)	Ash yield (wt%) (GB-2)	Ash reduction (%)
IC	23.41	_	26.95	_
After 2nd day	23.38	0.13	26.46	1.82
After 4th day	23.09	1.37	26.25	2.60
After 8th day	23.07	1.45	25.83	4.16
After 12th day	23.04	1.58	25.72	4.56
After 16th day	22.96	1.92	25.67	4.75
After 20th day	22.91	2.14	25.64	4.86
After 24th day	22.77	2.73	25.30	6.12
After 28th day	22.68	3.12	24.98	7.31

 Table 3
 Variation in the ash yield and ash reduction of coal samples of Gopinathpur bottom seam analyzed before and after bacterial treatment with 10 mg/mL biomass

IC initial concentration



Fig. 3 Variation in the ash yield and ash removal in coal samples of Gopinathpur bottom seam (before and after bacterial treatment using 10 mg/mL biomass)

furnished in Table 1 while the graphical representation of the group maceral and mineral matter is provided in Fig. 1. The mean values of the petrographic elements are shown in Fig. 2. Gopinathpur seam coals are low in moisture content (0.33 %–0.49 %; av 0.40 %) but the ash yield of these coals is moderately high and varies from 23.41 % to 27.3 %. The volatile matter (daf basis) varies from

28.29 % to 35.26 % and fixed carbon (daf basis) varies from 64.74 % to 71.71 % (Table 1).

#### 3.2 Distribution of minor and trace elements

Gopinathpur seam coals have high concentrations of K (1189.5–5573 ppm; av 2650.7 ppm) and Na (4534–5618 ppm; av 5314.7 ppm) among the minor elements as compared to the world average in coals as given by Valkovic (1983). Similarly Se (6.8–9.0 ppm; 7.9 ppm) and Pb (1.8–13.2 ppm; av 9.3 ppm) are high in concentrations in these coals when compared with the Clarke values of the elements in Bituminous coals as given by Ketries and Yudovich (2009). Other elements analysed in these coals such as Ni (8.2–32.0 ppm; 19.6 ppm), Zn (9–31 ppm; av 17.8 ppm) and Mn (19.1–24 ppm; av 21.5 ppm) occur within the limits. The details of the concentrations of the analyzed minor and trace elements are furnished in Table 2.

# 3.3 Demineralization using *Pseudomonas mendocina* strain B6-1

Metal removal from coal by bacteria occurs through leaching, adsorption and accumulation. Leaching mechanisms of metals through bacteria takes place in various ways such as, (i) the formation of organic or inorganic acids (protons); (ii) oxidation and reduction reactions; and (iii) the excretion of complexing agents (Berthelin 1983). The chief inorganic acid in leaching environments is sulphuric acid which is formed by sulfur-oxidizing microorganisms. However, a series of organic acids are produced by bacterial metabolism which leads to organic acidolysis, complex and chelate formation and facilitates metal removal (Berthelin 1983). After liberation of metals from coal or organic matter through microbial leaching, the processes like adsorption and

Day	Ash yield (wt%) (GT-1) treated with 10 mg/mL biomass	Ash reduction (%) (GT-1) by 10 mg/mL biomass	Ash yield (wt%) (GT-1) treated with 25 mg/mL biomass	Ash reduction (%) (GT-1) treated with 25 mg/mL biomass
IC	27.3	_	27.30	_
After 2nd day	26.2	4.03	25.51	6.56
After 4th day	25.3	7.33	23.30	14.65
After 8th day	24.21	11.32	21.12	22.64
After 12th day	22.9	16.12	19.59	28.24
After 16th day	22.1	19.05	18.12	33.63
After 20th day	20.12	26.30	17.97	34.18
After 24th day	18.45	32.42	17.42	36.19
After 28th day	18.02	33.99	17.08	37.44

Table 4 Variation in the ash yield and ash reduction of coal samples of Gopinathpur top seam (GT-1) analyzed before and after bacteria treatment (with 10 and 25 mg biomass of bacteria)



**Fig. 4** Variation in ash yield and ash reduction after treatment of GT-1 sample by *P. mendocina* strain B6-1 at two biomasses (10 and 25 mg/mL)

accumulation of metals by bacteria becomes pronounced. The metal binding in adsorption occurs on the bacterial surface. This may occur on the surface of both living as well as non-living bacteria. Numerous negatively charged groups are present on the bacterial cell surface where the positively charged cations bind with them through electrical attraction. Subsequently, the cations are removed. Metal accumulation, in bacteria, refers to translocation of the cations which is a metabolic dependent process and takes place across the wall and membrane into the cell. There is specific transport system for metal ions which involves channels and pores of cell membrane. Polyphosphate body in bacteria works as metal sink (Gadd 2009; Kornberg 1995).

In the present investigation three coal samples, two from Gopinathpur bottom seam (GB-1 and GB-2) and one sample from Gopinathpur top seam (GT-1), have been subjected to demineralization to see decrease in the content

**Table 5** Variation in concentration of minor and trace elements in sample GB-1, before and after treatment with 10 mg/mL of bacterial biomass (ppm)

Element	IC	After 2nd day	After 4th day	After 8th day	After 12th day	After 16th day	After 20th day	After 24th day	After 28th day
Ca	918.2	899.9	859.4	843.4	822.0	800.6	789.8	782.6	778.2
K	1189.5	1189.5	1169.8	1163.1	1144.45	1125.9	1094.1	934.6	926.6
Na	5792.0	4987.0	4934.0	4822.0	4789.0	3928.0	3072.0	2410.0	2134.0
Pb	13.2	12.1	11.5	11.5	9.1	8.7	8.2	8.0	7.7
Se	6.80	6.40	5.80	5.78	5.50	5.20	4.30	3.20	2.20
Ni	18.50	16.70	14.30	12.00	10.70	10.40	9.50	7.20	6.90
Mn	19.10	15.80	18.40	14.60	14.30	6.70	4.90	5.80	2.60
Zn	9.00	6.80	6.40	6.00	5.10	4.40	3.00	2.70	1.00

IC initial concentration

Table 6	Removal (%	) of 1	minor and	1 trace	elements	after	treatment	of sa	ample	GB-1	by	Р.	mendocina	strain	B6-1	with	10 mg	g/mL	biomas	s
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Day	Ca	K	Na	Mn	Pb	Se	Ni	Zn
2 day	1.99	0.00	13.90	17.28	8.33	5.88	9.73	24.44
4 day	6.40	1.66	14.81	3.66	12.88	14.71	22.70	28.89
8 day	8.15	2.22	16.75	23.56	12.88	15.00	35.14	33.33
12 day	10.48	3.79	17.32	25.13	31.06	19.12	42.16	43.33
16 day	12.81	5.35	32.18	64.92	34.09	23.53	43.78	51.11
20 day	13.98	8.02	46.96	74.35	37.88	36.76	48.65	66.67
24 day	14.77	21.43	58.39	69.63	39.39	52.94	61.08	70.00
28 day	15.25	22.10	63.16	86.39	41.67	67.65	62.70	88.89

 Table 7
 Variation in concentration of minor and trace elements in sample GB-2, before and after treatment with 10 mg/mL of bacterial biomass (ppm)

Element	IC	After 2nd day	After 4th day	After 8th day	After 12th day	After 16th day	After 20th day	After 24th day	After 28th day
Ca	1191.5	972.5	886.4	859.9	845.0	830.4	803.3	799.8	788.1
Κ	1189.5	1189.5	1169.8	1149.1	1134.5	1128.9	1109.0	1032.00	1019.8
Na	5618.0	4107.0	4098.0	3940.0	3898.0	3852.0	3501.0	2608.0	1769.0
Cd	1.8	1.69	1.60	1.8	1.53	1.50	1.40	1.3	1.2
Pb	9.0	8.80	8.60	8.2	8.0	7.8	7.5	7.2	6.8
Se	8.2	7.4	7.2	6.8	5.1	4.2	2.4	1.5	0.3
Ni	24.0	17.7	17.4	17.0	16.5	16.4	15.3	12.1	9.1
Mn	13.3	12.8	11.5	11.2	9.9	7.7	6.4	6.2	6.2
Zn	38.8	38.6	38.5	35.9	38.6	35.0	31.1	28.2	17.3

Table 8 Removal (%) of minor and trace elements after treatment of sample GB-2 by P. mendocina strain B6-1 with biomass 10 mg/mL

Day	Ca	Κ	Na	Cd	Pb	Se	Ni	Mn	Zn
2 day	18.38	0.00	26.90	6.11	2.22	9.76	26.25	3.76	0.52
4 day	25.61	1.66	27.06	11.11	4.44	12.20	27.50	13.53	0.77
8 day	27.83	3.40	29.87	0.00	8.89	17.07	29.17	15.79	7.47
12 day	29.08	4.62	30.62	15.00	11.11	37.80	31.25	25.56	0.52
16 day	30.31	5.09	31.43	16.67	13.33	48.78	31.67	42.11	9.79
20 day	32.58	6.77	37.68	22.22	16.67	70.73	36.25	51.88	19.85
24 day	32.87	13.24	53.58	27.78	20.00	81.71	49.58	53.38	27.32
28 day	33.86	14.27	68.51	33.33	24.44	96.34	62.08	53.38	55.41

of ash yield and the concentrations of minor/trace elements as a function of time (up to 28 days) and change in the concentration of bacterial biomass (10 and 25 mg/mL) comprising of *P. mendocina* strain B6-1.

# 3.3.1 Decrease in ash yield as a function of time

Coal samples (GB-1 and GB-2) from Gopinathpur bottom seam have been treated with 10 mg/mL bacterial biomass of *P. mendocina* B6-1 to see the decrease in the ash yield with time. In case of GB-1 sample there was a gradual decrease in the ash yield and after 28 days of bacterial treatment there was a reduction in the ash yield up to 3.12 %. However, in GB-2 coal sample there was relatively more reduction (up to 7.31 %) in the final ash yield. The details of the periodic reduction in the ash yield are furnished in Table 3 and are graphically shown in Fig. 3.

# 3.3.2 Decrease in ash yield as a function of change in concentration of bacterial biomass

In this case one coal sample from Gopinathpur top seam (GT-1) was treated with 10 and 25 mg/mL of *P. mendocina* strain B6-1. In this sample a substantial decrease in the ash yield was noticed. After 28 days of bacterial treatment there was 33.99 % decrease in the ash when 10 mg/mL of biomass was used while 37.44 % reduction in the ash yield was recorded when 25 mg/mL of biomass was used. The details of the periodic ash reduction are furnished in Table 4 and are graphically shown in Fig. 4.

# 3.3.3 Removal of minor and trace elements as a function of time

In case of GB-1 coal sample variable removal (%) of minor elements has been recorded after 28 days of bacterial treatment with 10 mg/mL of biomass. Ca and K have shown a relatively moderate removal of 15.25 % and 22.10 %, respectively while Na and Mn have shown significantly high removal of 63.16 % and 86.39 %, respectively. The trace elements such as Pb, Ni, Se and Zn have



Fig. 5 Comparative removal (%) of minor and trace elements in GB-1 and GB-2 coal samples, after 28 days of treatment with *P. mendocina* strain B6-1 using 10 mg/mL biomass

shown a high removal of 41.67 %, 62.70 %, 67.65 % and 88.89 %, respectively in these coals. The decrease in the content of minor and trace elements and their removal percentage are provided in Tables 5 and 6, respectively.

In case of GB-2 coal sample moderately high to very high removal of most of the minor and trace elements has been achieved after 28 days of bacterial treatment. Among the minor elements K, Ca, Mn and Na have given good response to bacterial treatment and removal was 14.27 %, 33.86 %, 53.38 % and 68.51 %, respectively while the trace elements like Pb, Cd, Zn, Ni and Se have been removed to a larger extent to the tune of 24.44 %, 33.33 %, 55.41 %, 62.08 % and 96.34 %, respectively. The details of decrease and percent removal are provided in Tables 7 and 8, respectively. The comparative removal of minor and trace elements in GB-1 and GB-2 coal samples of Gopinathpur bottom seam after 28 days of bacterial treatment using 10 mg/mL of biomass is shown in Fig. 5.

# 3.3.4 Removal of minor and trace elements as a function of change in the concentration of biomass

Coal sample (GT-1) from Gopinathpur top seam has been treated with *P. mendocina* B6-1 having 10 and 25 mg/mL

 
 Table 10
 Removal (%) of minor and trace elements after treatment in sample GT-1 by *P. mendocina* strain B6-1 with 10 mg/mL biomass

Day	Ca	K	Na	Pb	Ni	Zn
IC	0.00	0.00	0.00	0.00	0.00	0.00
2 day	4.94	3.63	3.80	2.97	34.38	3.23
4 day	13.25	7.36	10.86	5.94	43.75	9.68
8 day	15.63	11.67	16.63	9.90	62.50	22.58
12 day	18.01	15.09	25.02	11.88	68.75	29.03
16 day	19.70	18.29	34.80	16.83	74.69	45.16
20 day	22.31	26.42	39.25	19.80	80.31	51.61
24 day	23.35	31.42	48.78	21.78	82.50	58.06
28 day	22.88	31.96	50.43	23.76	84.06	61.29

 Table 9
 Variation in concentration of minor and trace elements in sample GT-1, before and after treatment with 10 mg/mL biomass of P.

 mendocina strain B6-1 (ppm)

Element	IC	After 2nd day	After 4th day	After 8th day	After 12th day	After 16th day	After 20th day	After 24th day	After 28th day
Ca	1721	1636	1493	1452	1411	1382	1337.1	1319.2	1327.3
Κ	5573	5371	5163	4923	4732	4554	4101	3822	3792
Na	4713	4534	4201	3929	3534	3073	2863	2414	2336
Pb	10.1	9.8	9.5	9.1	8.9	8.4	8.1	7.9	7.7
Ni	32	21	18	12	10	8.1	6.3	5.6	5.1
Zn	31	30	28	24	22	17	15	13	12

IC initial concentration

Element	IC	After 2nd day	After 4th day	After 8th day	After 12th day	After 16th day	After 20th day	After 24th day	After 28th day
Ca	1721	1593.3	1476.1	1423.2	1397.5	1343	1307.3	1262.6	1233.4
Κ	5573	5079	4654	4111	3545	3062	2876	2754	2589
Na	4713	4331	3910	3452	3012	2734	2556	2359	2310
Pb	10.1	9.4	9.1	7.4	7.1	6.2	5.6	5.1	4.9
Ni	32	18.1	16.3	14.3	11	8.3	5.4	4.1	2.2
Zn	31	25.1	19.2	15.1	13.2	11	10	9.1	4.5

Table 11 Variation in concentration of minor and trace elements in sample GT-1, before and after treatment of by *P. mendocina* strain B6-1 with 25 mg/mL biomass (ppm)

IC initial concentration

 Table 12
 Removal (%) of minor and trace elements in sample GT-1, after treatment by *P. mendocina* strain B6-1 with 25 mg/mL biomass

Day	Ca	K	Na	Pb	Ni	Zn
IC	0.00	0.00	0.00	0.00	0.00	0.00
2 day	7.42	8.86	8.11	6.93	43.44	19.03
4 day	14.23	16.49	17.04	9.90	49.06	38.06
8 day	17.30	26.23	26.76	26.73	55.31	51.29
12 day	18.80	36.39	36.09	29.70	65.63	57.42
16 day	21.96	45.06	41.99	38.61	74.06	64.52
20 day	24.04	48.39	45.77	44.55	83.13	67.74
24 day	26.64	50.58	49.95	49.50	87.19	70.65
28 day	28.33	53.54	50.99	51.49	93.13	85.48



**Fig. 6** Comparative removal (%) of minor and trace elements in GT-1 coal sample after 28 days of treatment with *P. mendocina* strain B6-1 using two biomasses (10 and 25 mg/mL)

of bacterial biomass to see decrease in the concentration of minor and trace elements and their removal percent. After 28 days of treatment using 10 mg/mL of biomass, there was moderately high removal. The minor elements such as Ca, K and Na have shown removal of 22.88 %, 31.96 % and 50.43 %, respectively. On the other hand trace elements like Ni and Zn have shown a high removal of

84.06 % and 61.29 %, respectively. However, Pb showed a medium removal of 23.76 %. The periodic decrease in the content of elements and their removal percent are provided in Tables 9 and 10, respectively. In case of bacterial treatment using 25 mg/mL biomass there was significant increase in the removal of elements. Ca, K and Na have shown a removal of 28.33 %, 53.54 % and 50.99 %, respectively after 28 days. Similarly Ni, Zn and Pb have shown a removal of 93.13 %, 85.48 % and 51.49 %, respectively. The periodic decrease in the concentration of these elements and their removal percent are provided in Tables 11 and 12, respectively. The comparative removal (%) of these elements in GT-1 coal sample of Gopinathpur top seam after 28 days of treatment using 10 and 25 mg/mL of bacterial biomass is shown in Fig. 6.

# 4 Conclusions

- Gopinathpur coals are dominated by vitrinite group of macerals (56.4 %-62.9 % mmf basis) which is contributed mainly by collodetrinite and little by collotelinite and vitrodetrinite. Inertinite (24.6 %-31.3 % mmf basis) and liptinite (8.6 %-12.9 % mmf basis) occur in low concentrations. Mineral matter occurs in moderately high concentration (18.3 %-25.6 %).
- (2) Gopinathpur seam coals have high concentrations of K (1189.5–5573 ppm) and Na (4534–5618 ppm) among the minor elements as compared to the world average in coals. Similarly, Se (6.8–9.0 ppm) and Pb (1.8–13.2 ppm) are high in concentrations when compared with the Clarke values.
- (3) Coal samples (GB-1 and GB-2) from Gopinathpur bottom seam have shown a gradual decrease in the ash yield and after 28 days of bacterial treatment there was a reduction in the ash yield up to 3.12 % and 7.31 %, respectively.

- (4) Coal sample (GT-1) from Gopinathpur top seam was treated with 10 and 25 mg/mL of biomass (*P. mendocina* strain B6-1) which indicated a substantial decrease in the ash yield up to 33.99 % and 37.44 %, respectively after 28 days.
- (5) In coal sample (GB-1) from Gopinathpur bottom, there is 15.25 % and 22.10 % removal of Ca and K respectively while Na and Mn have shown 63.16 % and 86.39 % removal respectively. The trace elements such as Pb, Ni, Se and Zn have shown a high removal of 41.67 %, 62.70 %, 67.65 % and 88.89 %, respectively. In case of GB-2 coal sample K, Ca, Mn and Na have given good response to bacterial treatment and removal was 14.27 %, 33.86 %, 53.38 % and 68.51 %, respectively while the trace elements like Pb, Cd, Zn, Ni and Se have been removed to a larger extent to the tune of 24.44 %, 33.33 %, 55.41 %, 62.08 % and 96.34 %, respectively.
- (6) When coal sample (GT-1) from Gopinathpur top seam, was treated with 10 mg/mL of bacterial biomass, the minor elements such as Ca, K and Na have shown removal of 22.88 %, 31.96 % and 50.43 %, respectively and the trace elements like Ni and Zn have shown a high removal of 84.06 % and 61.29 %, respectively. However, when 25 mg/ mL biomass was used, there was significant increase in the removal of 28.33 %, 53.54 % and 50.99 %, respectively while Ni, Zn and Pb have shown a removal of 93.13 %, 85.48 % and 51.49 %, respectively.

Thus bacteria may be used as an important tool for the beneficiation of coals which are rich in impurities and environmentally sensitive elements. Nevertheless, there is need of more research work in this field to scale-up this experiment on Pilot and commercial levels.

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