

High-entropy perovskite RETa_3O_9 ceramics for high-temperature environmental/thermal barrier coatings

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Abstract: Four high-entropy perovskite (HEP) RETa_3O_9 samples were fabricated via a spark plasma sintering (SPS) method, and the corresponding thermophysical properties and underlying mechanisms were investigated for environmental/thermal barrier coating (E/TBC) applications. The prepared samples maintained low thermal conductivity ($1.50 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$), high hardness (10 GPa), and an appropriate Young's modulus (180 GPa), while the fracture toughness increased to $2.5 \text{ MPa}\cdot\text{m}^{1/2}$. Nanoindentation results showed the HEP ceramics had excellent mechanical properties and good component homogeneity. We analysed the influence of different parameters (the disorder parameters of the electronegativity, ionic radius, and atomic mass, as well as the tolerance factor) of A-site atoms on the thermal conductivity. Enhanced thermal expansion coefficients, combined with a high melting point and extraordinary phase stability, expanded the applications of the HEP RETa_3O_9 . The results of this study had motivated a follow-up study on tantalate high-entropy ceramics with desirable properties.

Keywords: high-entropy ceramics (HECs); tantalates; thermal conductivity; nanoindentation; fracture toughness

1 Introduction

Oxide environmental/thermal barrier coatings (E/TBCs) have been extensively investigated to increase the working temperatures of aircraft engines, gas turbines, and hypersonic vehicles [1–6]. E/TBCs are applied to ceramic/alloy substrates to decrease surface temperatures and prevent damage from high-speed impact particles, molten oxides, and other environmental factors [2–12]. Rare earth (RE) tantalates (RETaO_4 , RE_3TaO_7 , and RETa_3O_9) are considered novel E/TBCs because of extraordinary thermophysical properties [4,8,13–15]. It

has been reported that tantalates have low thermal conductivity and high hardness, melting points, and thermal expansion coefficients. However, tantalates also have clear shortcomings, e.g., RETa_3O_9 has inadequate fracture toughness ($1.1\text{--}1.5 \text{ MPa}\cdot\text{m}^{1/2}$) [15]. Diverse methods are used to enhance the properties of RETaO_4 [8,13] and RE_3TaO_7 ceramics [14,16], and our aim is to stimulate the investigation and application of RETa_3O_9 with perovskite structures. Some scholars have identified perovskite oxides as promising E/TBCs, and high-entropy ceramics (HECs) are being studied in various fields [15,17–19]. Wright *et al.* [20] have reported that HECs of $\text{RE}_2\text{Zr}_2\text{O}_7$ exhibit low thermal conductivity and high stiffness, and Liu *et al.* [17] have proven that the thermal properties of perovskite-type

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(ABO₃) ceramics can be regulated by entropy engineering. Furthermore, HECs of RE₂Si₂O₇, RE₃NbO₇, and other RE tantalates have been studied for use as E/TBCs [21–24]. One of the most striking features of HECs is that multifarious properties can be successfully tailored to meet the demands of specific applications.

The structure of HECs is similar to that of high-entropy alloys (HEAs), that is, one or more Wyckoff sites in lattices are occupied by identical or near-identical atoms. In HECs, more than four types of atoms occupy one site, and the configurational entropy is higher than $1.5R$ (where R denotes the gas constant); the configurational entropy (ΔS_{conf}) is expressed as $\Delta S_{\text{conf}} = R \times \ln N$, where N is the number of elements in one sublattice [5,25–27]. The configurational entropy increases with the number of types of atoms in one or more Wyckoff sites in the lattice. Herein, we regulate the configurational entropy by controlling the types of atoms in the lattice. Theoretically, the highest configurational entropy is obtained for some materials when all the elements in a sublattice have the same atomic fraction [28,29]. To date, HECs have been fabricated with many excellent properties, including enhanced hardness and toughness [19], regulation of thermal transportation properties [20,23], increased corrosion resistance [21], and strengthened high-temperature phase stability [10,17]. HECs of RE₃TaO₇ [16], RE₂Zr₂O₇ [20,23], RE₃NbO₇ [22], and RE₃Al₅O₁₂ [30] have also been investigated for use as TBCs, whereas HECs of RE₂Si₂O₇ [21] and RE₂SiO₅ [31] have been investigated for use as EBCs. Currently, multifunctional thermal and environmental barrier integration coatings are used to improve the high-temperature performance of components of alloys and ceramic matrix composites. A few oxides meet the requirements of multifunctional E/TBCs. Perovskite-type ceramics typically have high melting points [17], low thermal conductivity [15,17,18], regulatable mechanical properties [17,32,33], etc. [34,35]. Accordingly, multifarious perovskite-type oxides have been fabricated to improve the properties and expand the applications [15,17,18,33–36]. In this study, we optimize the properties of HECs of RETa₃O₉ (RE_{0.33}TaO₃) perovskites as high-temperature E/TBCs. Differences in ionic radii and atomic weights of constituent elements affect the crystal structures and thermophysical properties of HECs. Therefore, a two-step sintering process is used to prepare four different equiatomic high-entropy perovskite (HEP) tantalates, including (La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta₃O₉

(5HEP-1), (Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-2), (La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-3), and (Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-4). The ionic radii and atomic weights of La and Pr are considerably different from those of Dy, Ho, and Tm in 5HEP-1, whereas Gd, Dy, Ho, Er, and Tm in 5HEP-2 have similar ionic radii and atomic weights. That is, differences in the ionic radius and atomic weight among the RE elements in 5HEP-1 are more prominent than those in the 5HEP-2 sample. The aforementioned designs are reproduced in the 6HEP-3 and 6HEP-4 samples to investigate the impacts of differences in ionic radii and atomic weights of various RE elements.

2 Experimental

2.1 Material fabrication

HEP RETa₃O₉ powders were synthesized by a high-temperature sintering process without pressure. We used RE₂O₃ (RE = La, Sm, Eu, Gd, Dy, Ho, Er, Tm), Pr₆O₁₁, and Ta₂O₅ with 99.9% purity as raw powders, which had particle sizes less than 10 μm. The weighed powders were mixed by a ball mill (300 r·min⁻¹, 12 h) using alcohol as the dispersion medium, and then placed in a drier at 90 °C for 6 h to evaporate alcohol. The dried powders were maintained at 1600 °C for 10 h to obtain high-purity RETa₃O₉ powders. The prepared powders were passed through a 500-mesh sieve, and approximately 2.5 g of the sieved powders were sintered by spark plasma sintering (SPS). The final SPS parameters were 1600 °C, 100 MPa, 10-min sintering time, and a heating rate of 50 °C·min⁻¹. For comparison, a dense yttria-stabilized zirconia (YSZ) bulk ceramic was synthesized and tested using the same methods.

2.2 Structure identification

The X-ray diffractometer (MiniFlex 600, Rigaku, Japan) was applied to confirm the crystal structures and lattice constants of the HEP RETa₃O₉. The X-ray diffraction (XRD) 2θ angle was ranged between 20° and 60°, and the scanning speed was 5 (°)·min⁻¹. The exact lattice constants, unit cell volumes, and theoretical densities of each sample were computed from the XRD patterns. The scanning electronic microscope (SEM, JSM-7800F, JEOL, Japan) and the energy-dispersive spectroscopy (EDS, 6751A-6UUS-SN, Thermo Scientific, USA)

were used to probe the surface structures. We used SEM and EDS to observe the grain sizes, pores, cracks, and elemental distributions.

2.3 Mechanical property testing

The hardness (H) and fracture toughness (K_{IC}) were determined by the lengths of indentation diagonal (d) and the indentation crack (c) using a hardness tester (DHV-1000Z-CCD, SZNG, China). Each sample was tested 10 times, the minimum and maximum results were removed, and the remaining results were used to calculate the average values. The fracture toughness increased with the surface fracture energy (Γ), and was calculated as below [37,38]:

$$H = 0.464 \frac{F}{d^2} \quad (1)$$

$$K_{IC} = 0.018 \sqrt{\frac{E}{H}} \cdot \frac{F}{c^{1.5}} \quad (2)$$

$$\Gamma = \frac{2\zeta^2 F d^2}{c^3} \quad (3)$$

where $F = 4.9$ N, $\zeta = 0.016$, and E denotes the Young's modulus measured using the ultrasonic equipment (UMS-100, TECLAB, France) [39]:

$$E = \frac{\rho V_t^2 (3V_l^2 - 4V_t^2)}{V_l^2 - V_t^2} \quad (4)$$

where ρ is the sample density, and V_l and V_t are the longitudinal and transverse acoustic velocities, respectively.

To eliminate the effect of the microstructure on the mechanical properties, the nano-Young's modulus (E_N) and nano-hardness (H_N) were simultaneously measured using the nanoindentation device (iNano instrument, iMicro, USA). During the nanoindentation test, we applied both E and H mapping (three-dimensional (3D) mapping) and advanced dynamic E and H (single-point test) models, where the highest test load was 100 mN. A total of $30 \times 30 = 900$ points were tested over a $300 \mu\text{m} \times 300 \mu\text{m}$ area at intervals of $10 \mu\text{m}$ during the 3D mapping measurement. The advanced dynamic E and H tests were used to determine how the hardness and Young's modulus varied with the indentation load and depth, and the results were analysed.

2.4 Thermal property testing

The thermal expansion coefficients (TECs) were calculated based on the elongation caused by increasing the temperature that was measured by a thermo-

mechanical analysis device (TMA 402 F3, NETZSCH, Germany). The thermal diffusivity (λ , 25–900 °C) was measured using the laser reflection device (LFA 457, NETZSCH, Germany), and the thermal conductivity (k) was calculated from λ , density (ρ), porosity (ϕ), and heat capacity (C_p) as follows [40]:

$$k = \frac{\lambda \cdot \rho \cdot C_p}{1 - 4\phi/3} \quad (5)$$

The phonon mean free path (l) was calculated using the relationship between the thermal diffusivity (λ) and the mean acoustic velocity (V_m) [40–42]:

$$l = \frac{3\lambda}{V_m} \quad (6)$$

The mean acoustic velocity (V_m) was calculated from V_l and V_t [39]:

$$V_m = \left[\frac{1}{3} \left(\frac{1}{V_l^3} + \frac{2}{V_t^3} \right) \right]^{-\frac{1}{3}} \quad (7)$$

3 Results and discussion

3.1 Crystal structures

Figure 1(a) shows that the main XRD peaks of the four HEP RETa_3O_9 samples coincide with PDF#52-0567 with the $P4/mmm$ space group. A RETaO_4 precipitate (PDF#24-0379) is detected in 5HPE-2, 6HEP-3, and 6HEP-4, whereas the minor peaks at 2θ angles of 25° – 30° for 5HEP-1 are associated with a different RETaO_4 phase (PDF#19-1013). The weak peaks of RETaO_4 indicate that the RETaO_4 content is too low to dominate the thermophysical properties of the prepared samples. In Ref. [43], the crystalline temperature of YTaO_4 is reported to be approximately 840 °C, and this low crystalline temperature may be the cause of the RETaO_4 precipitate in the prepared HEP samples. The lattice of RETa_3O_9 ($\text{RE}_{1/3}\text{TaO}_3$) with the $P4/mmm$ space group is treated as a defect perovskite ($\text{A}_{1/3}\text{BO}_3$ -type) structure when a $1/3$ vacancy is created in the RE cationic sublattice, as shown in Fig. 1(b). Each Ta is surrounded by six oxygens to form a $[\text{TaO}_6]$ octahedron, and the octahedra are connected to each other at the corners. These octahedrons are twisted structures, because the Ta–O bond lengths vary. The distortion degree of perovskites depends on the octahedral twist. Normally, the tolerance factor (t) is an indicator of the

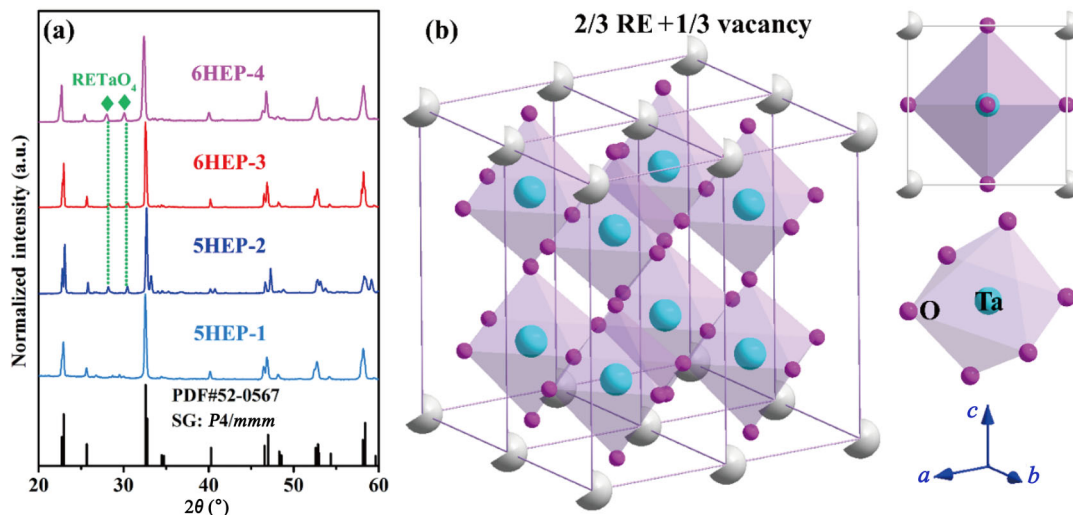


Fig. 1 Characterization of the crystal structure of (La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-1), (Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-2), (La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-3), and (Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-4): (a) normalized XRD patterns; (b) 2 × 2 × 1 unit cell, where the grey ball represents an A-site 2/3 RE atom with a 1/3 vacancy, the blue ball represents a B-site Ta atom, and the purple ball represents an oxygen atom.

Table 1 Design composition, sample name, lattice parameters (*a*, *b*, and *c*), unit cell volume (*V*), theoretical density (ρ_0), and porosity (ϕ) of HEP RETa₃O₉ ceramics

Composition	Sample	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	<i>V</i> (Å ³)	ρ_0 (g·cm ⁻³)	ϕ (%)
(La _{1/5} Pr _{1/5} Dy _{1/5} Ho _{1/5} Tm _{1/5})Ta ₃ O ₉	5HEP-1	3.878	3.878	7.850	118.081	7.89	1.5
(Gd _{1/5} Dy _{1/5} Ho _{1/5} Er _{1/5} Tm _{1/5})Ta ₃ O ₉	5HEP-2	3.854	3.854	7.864	116.796	8.06	1.3
(La _{1/6} Sm _{1/6} Eu _{1/6} Dy _{1/6} Ho _{1/6} Tm _{1/6})Ta ₃ O ₉	6HEP-3	3.902	3.902	7.803	118.834	7.86	2.0
(Eu _{1/6} Gd _{1/6} Dy _{1/6} Ho _{1/6} Er _{1/6} Tm _{1/6})Ta ₃ O ₉	6HEP-4	3.854	3.854	7.837	116.396	8.07	2.2

lattice stability and distortion degree of perovskite structures [29,32]:

$$t = \frac{r_A + r_O}{\sqrt{2}(r_{Ta} + r_O)} \tag{8}$$

where r_A , r_O , and r_{Ta} are the effective ionic radii of RE³⁺, O²⁻, and Ta⁵⁺ ions, respectively. Perovskite oxides have a stable lattice when *t* ranges from 0.8 to 1, and they may crystallize into an orthorhombic phase for *t* less than 0.9 [29,32]. The tolerance factors of the HEP RETa₃O₉ are proportional to the A-site ionic sizes, and are 0.871, 0.855, 0.866, and 0.856 for 5HEP-1, 5HEP-2, 6HEP-3, and 6HEP-4, respectively. Among the four prepared samples, 5HEP-2 has the maximum lattice distortion degree based on the minimum tolerance factor. The results presented in Table 1 and Fig. 1(b) indicate that the length of the *c*-axis is almost twice that of the *a*- and *b*-axes, because HEP RETa₃O₉ is an orthorhombic phase. Usually, the lengths of the *a*-, *b*-, and *c*-axes of ABO₃-type perovskites are almost equal (corresponding to a cubic phase), and the

structures of the HEP RETa₃O₉ are also called double-perovskite structures [18]. Herein, the 2 × 2 × 1 unit cell of HEP RETa₃O₉ is similar to that of cubic ABO₃ perovskites.

3.2 Microstructures

The four HEP bulk samples are polished using diamond sandpapers, and then held at 1400 °C for 3 h to observe the surface grains and elemental distributions by SEM and EDS. In Fig. 2, most grains are smaller than 5 μm, and the RE elements are homogeneously distributed in the prepared samples. The prepared HEP samples have smaller grains than the single-rare-earth RETa₃O₉, and grain refinement is achieved in the HEP RETa₃O₉ samples [15]. The mean grain sizes are 5–7 μm for the single-rare-earth RETa₃O₉ ceramics and less than 2 μm for the HEP samples. Thus, the corresponding reduction in the grain size is higher than 50%, and we investigate the effect of this reduction on the material properties. Some nanoscale grains are found in 5HEP-2, 6HEP-3, and 6HEP-4, and it is inferred that these grains

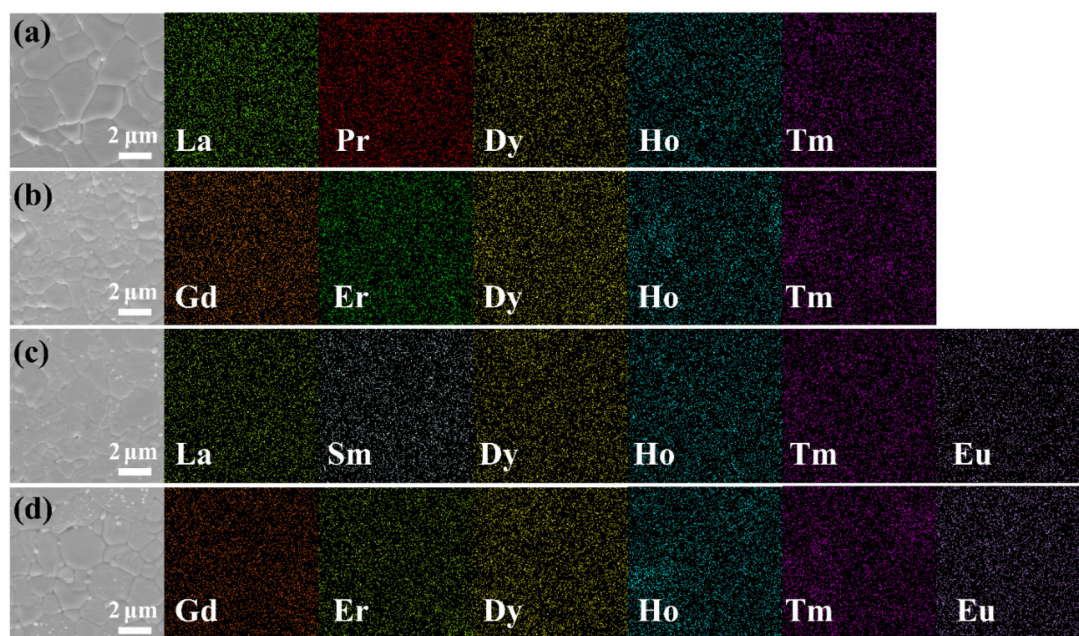


Fig. 2 Surface grains and RE elemental distributions of HEP $RETa_3O_9$ ceramics: (a) $(La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta_3O_9$ (5HEP-1), (b) $(Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta_3O_9$ (5HEP-2), (c) $(La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta_3O_9$ (6HEP-3), and (d) $(Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta_3O_9$ (6HEP-4).

constitute the $RETaO_4$ precipitate. We used the backscattered electron (BSE) model to analyse the grains, and no difference was found between nano- and micron-scale grains. The $RETaO_4$ and $RETa_3O_9$ phases are comprised of the same elements, and the grains of these phases have the same atomic contrast. Furthermore, the EDS results show no elemental segregation. The XRD, SEM, and EDS analyses imply that we can neglect the influence of $RETaO_4$ on the properties of the prepared samples. In Table 1, the porosities of the prepared samples are 1%–3%, which are lower than those of the samples synthesized via a solid-state reaction [15]. Hence, dense and high-purity HEP $RETa_3O_9$ ceramics can be fabricated via an SPS process. The grain size, grain boundary density, cracks, and pores affect the properties of structural ceramics, as discussed in the following sections.

3.3 Mechanical properties

As shown in Table 2, the HEP $RETa_3O_9$ ceramics have a Young's modulus (E), hardness (H), and toughness (K_{IC}) of 130–180 GPa, 7–10 GPa, and 1.8–2.5 $MPa \cdot m^{1/2}$, respectively. YSZ has a high Young's modulus (220–250 GPa), which results in inadequate deformation tolerance [44]. For comparison, a YSZ bulk sample is fabricated and tested using the aforementioned methods. A balance among different

mechanical properties is achieved in the HEP $RETa_3O_9$. Figure 3 shows that the four HEP samples have a lower Young's modulus than those of YSZ and $La_2Zr_2O_7$ (230 GPa) [6], and 6HEP-4 has the highest toughness ($2.5 \pm 0.4 MPa \cdot m^{1/2}$). The HEP ceramics and single-rare-earth $RETa_3O_9$ have similar Young's moduli (140–200 GPa) and hardness (7.0–10.0 GPa), whereas the toughness of the HEP $RETa_3O_9$ is 1.8–2.5 $MPa \cdot m^{1/2}$. Figures 4(a)–4(c) show that the hardness–Young's modulus, toughness–Young's modulus, and toughness–hardness curves are not linear. The results presented in Fig. 4(d) and Table 2 indicate that the toughness is proportional to the surface fracture energy. Equation (3) shows that the surface fracture energy is related to the lengths of indentation and cracks, and is an intrinsic feature of these materials. To study the ductile and brittle features of the four samples, we calculate the G/B ratio, where G and B represent the shear and bulk moduli, respectively [39,45]:

$$B = \frac{E}{3(1-2\nu)} \quad (9)$$

$$G = \frac{E}{2(1+\nu)} \quad (10)$$

where ν is the Poisson's ratio, as measured by ultrasonic equipment. Materials with a G/B ratio less than 0.57 are brittle, and materials with a G/B ratio higher than

Table 2 Young’s modulus and nano-Young’s modulus (E and E_N), hardness and nano-hardness (H and H_N), fracture toughness (K_{IC}), mean acoustic velocity (V_m), surface fracture energy (Γ), and G/B ratio of HEP RETa₃O₉ ceramics

Sample	E_N (GPa)	H_N (GPa)	V_m (m·s ⁻¹)	E (GPa)	H (GPa)	K_{IC} (MPa·m ^{1/2})	Γ (J·m ⁻²)	G/B
5HEP-1	178.9±8.7	10.8±0.6	3270.5±17.3	172.5±11.5	9.0±0.3	1.8±0.4	19.1±7.2	0.55
5HEP-2	143.3±12.7	10.0±1.3	2945.5±5.4	139.2±8.6	7.3±1.3	2.0±0.3	27.2±8.5	0.59
6HEP-3	156.7±5.3	10.4±0.6	3214.1±10.6	157.8±10.9	9.1±0.3	2.1±0.3	25.6±7.8	0.62
6HEP-4	180.5±13.6	10.6±1.1	3215.3±28.1	178.2±8.2	9.0±0.4	2.5±0.4	32.7±10.0	0.61

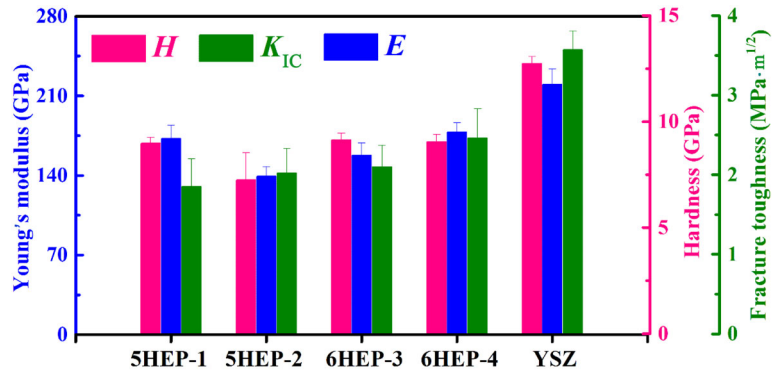


Fig. 3 Comparison of Young’s modulus, hardness, and fracture toughness of (La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-1), (Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-2), (La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-3), (Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-4) ceramics, and YSZ.

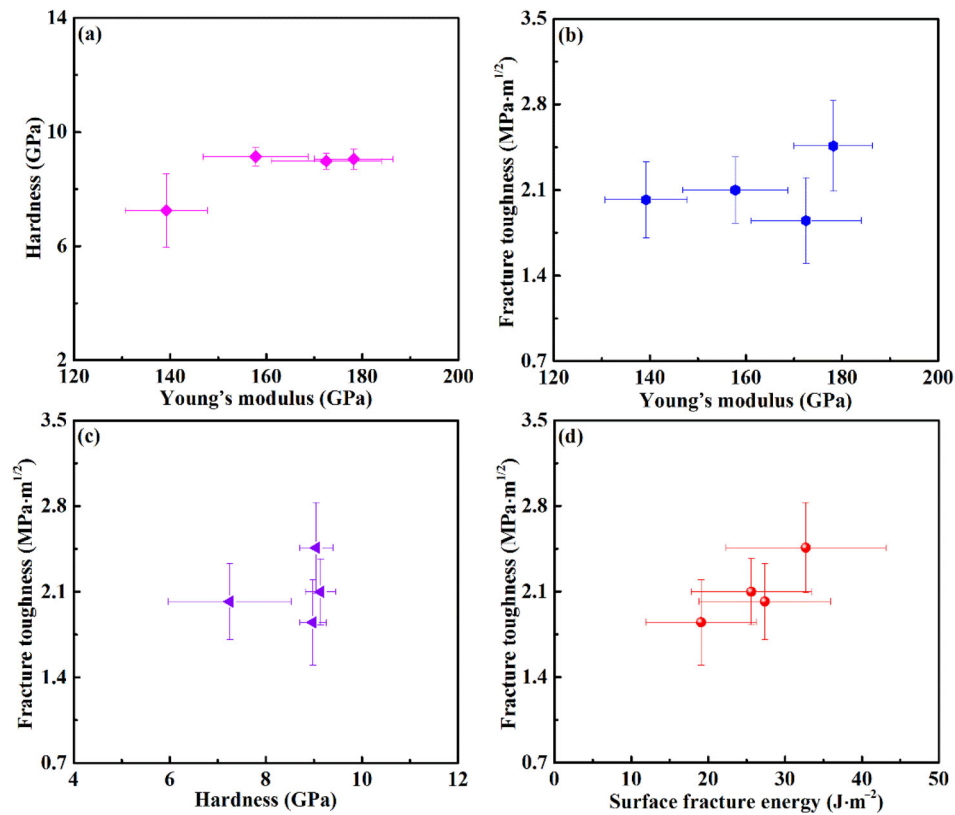


Fig. 4 Relationships among the mechanical properties of (La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-1), (Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-2), (La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-3), and (Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-4) ceramics: (a) hardness–Young’s modulus, (b) fracture toughness–Young’s modulus, (c) fracture toughness–hardness, and (d) fracture toughness–surface fracture energy.

0.57 are ductile [45]. Table 2 shows that the G/B ratio increases from 0.55 to 0.62, indicating that a brittle–ductile transition occurs with the difference in composition among the four HEP samples. The brittle–ductile transition is considered to increase the toughness. Small grains increase the hardness based on the Hall–Petch principle. The Young’s modulus of the HEP samples are lower than those of the HECs of RE_3NbO_7 (210 GPa) [22] and $\text{RE}_2\text{Zr}_2\text{O}_7$ (240 GPa) [20], and close to those of the HECs of RE_2SiO_5 (170 GPa) [31] and ABO_3 (180 GPa) [17]. The relatively low Young’s modulus of the HEP samples is attributed to their parent single-rare-earth RETaO_4 ceramics, where a comparatively low modulus promotes thermal stress reduction for high-temperature E/TBC applications [46].

3.4 Nanoindentation

The nano-Young’s modulus ($E_N = 140\text{--}180$ GPa) and hardness ($H_N = 10\text{--}11$ GPa) are shown in Table 2 and Fig. 5. The slightly higher mean values measured by nanoindentation than those tested by normal methods (Fig. 5(a)) may be due to the indentation-size effect [47–49]. Figure 6 shows the 3D mapping of E_N and H_N , where there are slight variations over the testing area

because microstructure defects (pores and cracks) and test error can produce fluctuations in E_N and H_N . In Fig. 7, most of the nano-Young’s modulus and hardness results are located around the mean values, and the hardness increases with the modulus. XRD patterns show that there are few precipitated RETaO_4 ceramics in the prepared samples, and SEM, EDS, and nanoindentation results jointly indicate the excellent compositional uniformity of the HEP RETa_3O_9 . Furthermore, the differences in the Young’s modulus of each sample tested by different methods in this study are less than 4%. The aforementioned results prove that the presence of a small quantity of precipitated RETaO_4 ceramics does not affect the Young’s modulus, hardness, and toughness of the HEP RETa_3O_9 . Normally, increases in the Young’s modulus and acoustic velocity reflect increases in the phonon propagation speed and thermal conductivity, which is discussed in Section 3.5 Thermal properties.

The results of the advanced dynamic E and H tests using nanoindentation are shown in Figs. 5(b)–5(f), where the investigated parameters include the hardness, Young’s modulus, indentation depth, and load. The indentation depth is proportional to the load, which has a maximum value of 100 mN in the current study. The H_N and E_N values for indentation depths less than 100 nm

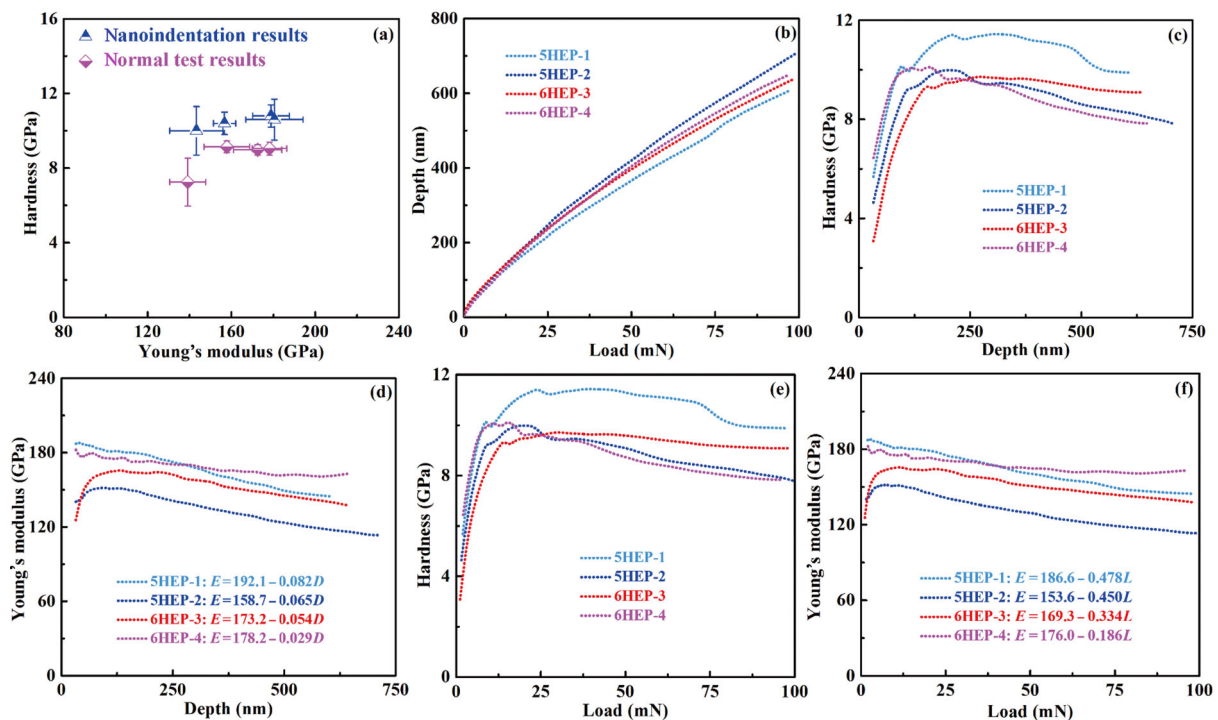


Fig. 5 Mechanical properties of $(\text{La}_{1/5}\text{Pr}_{1/5}\text{Dy}_{1/5}\text{Ho}_{1/5}\text{Tm}_{1/5})\text{Ta}_3\text{O}_9$ (SHEP-1), $(\text{Gd}_{1/5}\text{Dy}_{1/5}\text{Ho}_{1/5}\text{Er}_{1/5}\text{Tm}_{1/5})\text{Ta}_3\text{O}_9$ (SHEP-2), $(\text{La}_{1/6}\text{Sm}_{1/6}\text{Eu}_{1/6}\text{Dy}_{1/6}\text{Ho}_{1/6}\text{Tm}_{1/6})\text{Ta}_3\text{O}_9$ (6HEP-3), and $(\text{Eu}_{1/6}\text{Gd}_{1/6}\text{Dy}_{1/6}\text{Ho}_{1/6}\text{Er}_{1/6}\text{Tm}_{1/6})\text{Ta}_3\text{O}_9$ (6HEP-4) ceramics: (a) nano-Young’s modulus and hardness measured by different methods, (b–f) variations in different parameters observed during nanoindentation advanced dynamic E and H tests.

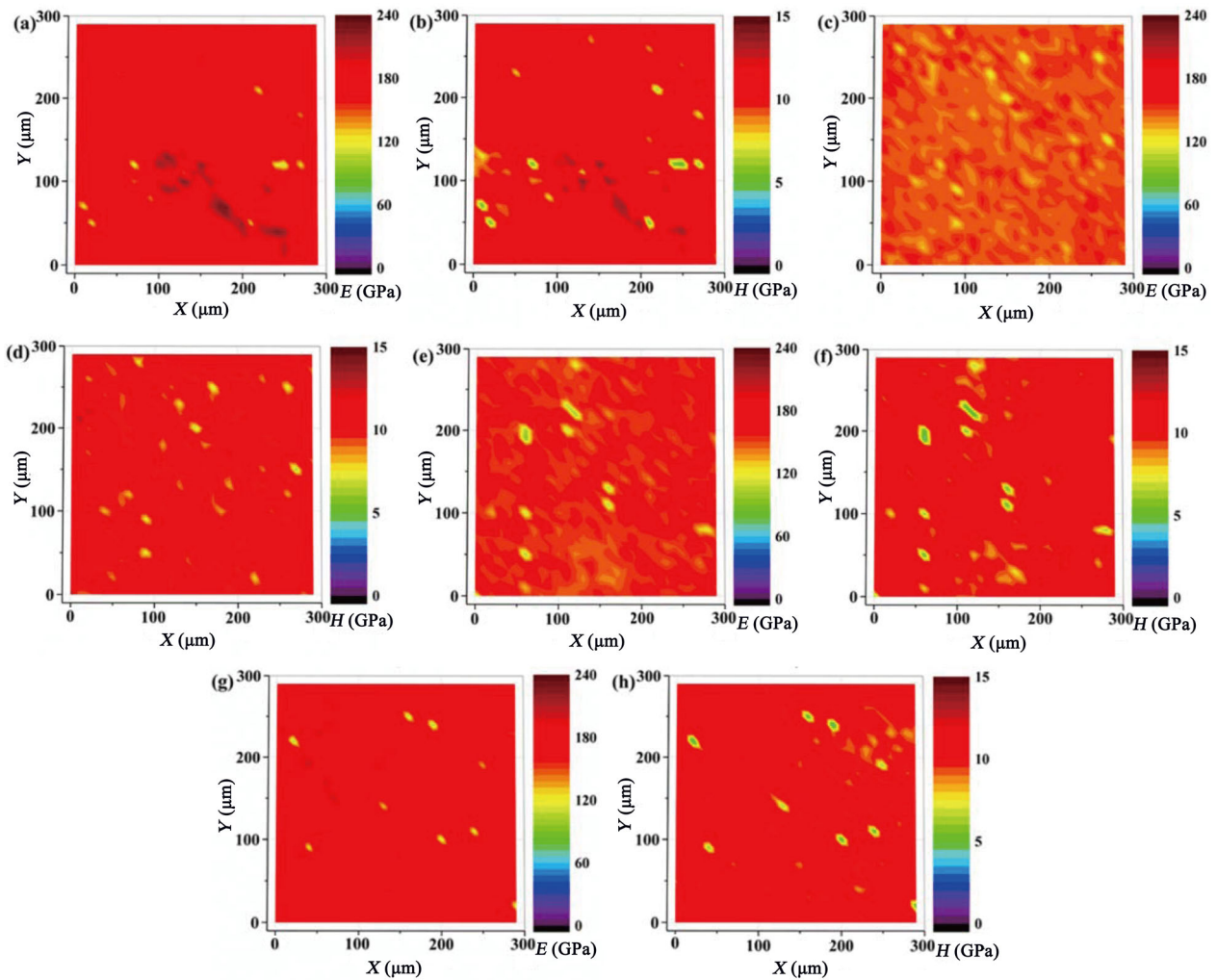


Fig. 6 3D mapping of Young’s modulus (E) and hardness (H) for HEP $RETa_3O_9$ ceramics: (a, b) $(La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta_3O_9$ (5HEP-1), (c, d) $(Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta_3O_9$ (5HEP-2), (e, f) $(La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta_3O_9$ (6HEP-3), and (g, h) $(Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta_3O_9$ (6HEP-4).

are not considered here because the indenter appearance is not self-similar, and the contact area is indeterminate at small indentation depths (low indentation loads) [47]. H_N and E_N decrease as the depth and load increase, for test depths beyond 100 nm or loads heavier than 15 mN. During the nanoindentation continuous stiffness test (using the advanced dynamic E and H model), the decrease in the strain gradient results in weakened hardening as the indentation depth and load increase [48–52]. In addition to the strain gradient, other factors affect the hardness and modulus of some materials. The grain size (the grain boundary density), pore, crack, and surface roughness may affect the hardness, toughness, and modulus. Considering the relative density and purity of the HEP $RETa_3O_9$, macro-defects should have little effect on the hardness and modulus, that is,

the strain gradient is the predominant factor. The relationships between the Young’s modulus and depth/load are shown in Figs. 5(d) and 5(f), respectively, and their slopes indicate the decreasing rate of the strain gradient in conjunction with increments of load/depth. In addition to the surface micro-defects, the internal micro-defect density affects the modulus and toughness of the prepared ceramics. The results of the advanced dynamic E and H tests show that the internal micro-defect density of the HEP $RETa_3O_9$ decreases in the order of 5HEP-1 > 5HEP-2 > 6HEP-3 > 6HEP-4, based on the decrease in the modulus with increasing depth.

3.5 Thermal properties

Figure 8(a) indicates that the thermal diffusivity of the HEP $RETa_3O_9$ samples initially decreases with the

increase of temperature, and then clearly increases at high temperatures (≥ 500 °C). Figure 8(b) shows that the thermal conductivity of the HEP RETa_3O_9 exhibits a similar temperature dependence to that of the corresponding single-rare-earth samples, which is caused by a high concentration of A-site cation vacancies and

thermal radiation [15,53]. Increases in thermal conductivity caused by thermal radiation have been reported for YSZ [34], $\text{La}_2\text{Zr}_2\text{O}_7$ [54], RETaO_4 [42], RE_2SiO_5 [31], and ABO_3 perovskite ceramics [17]. Evidently, the prepared samples have lower thermal conductivity than those of SrZrO_3 [55], YSZ [44], and many

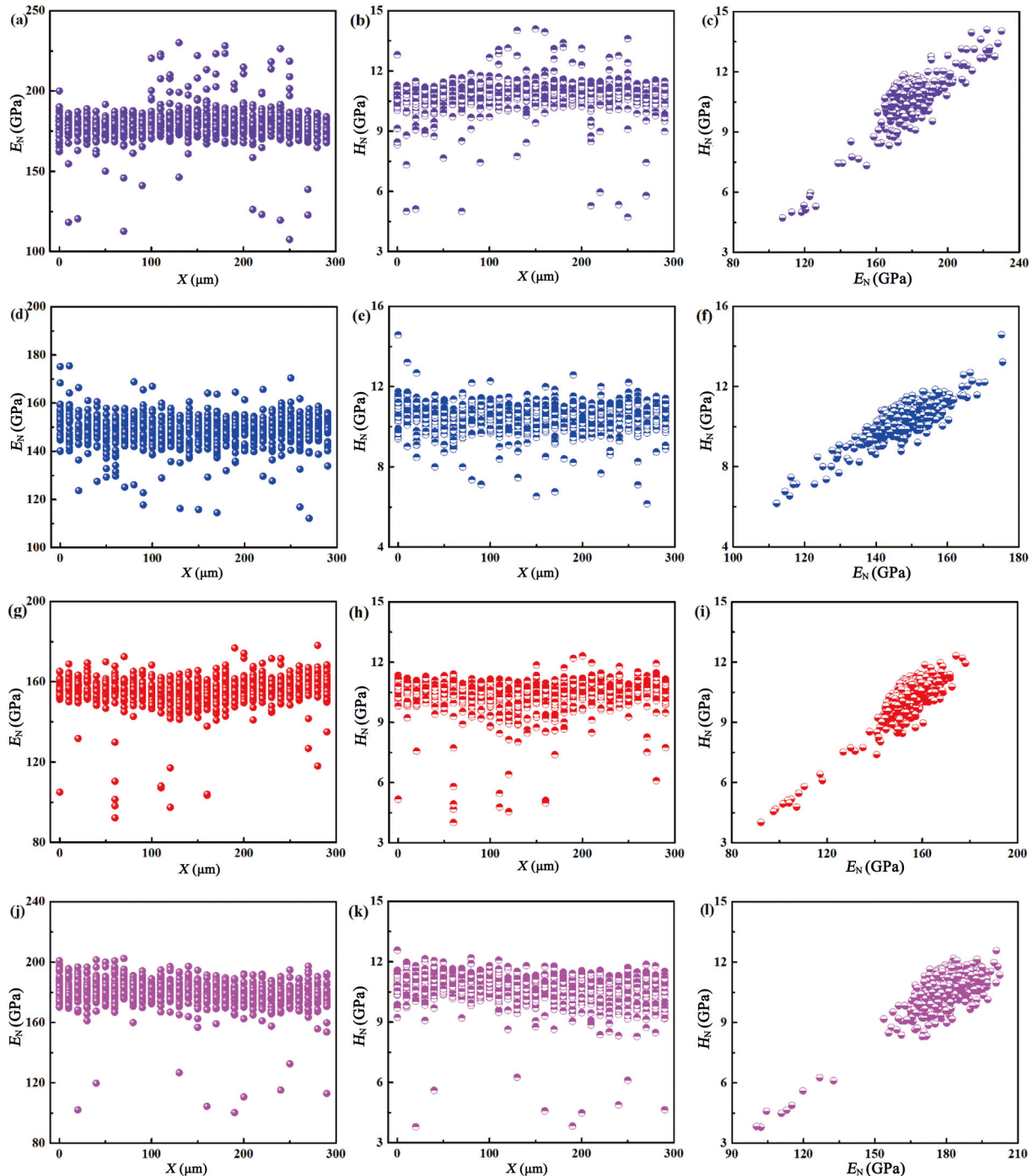


Fig. 7 Distributions of numerical values of nano-Young's modulus and hardness for HEP RETa_3O_9 ceramics: (a–c) $(\text{La}_{1/5}\text{Pr}_{1/5}\text{Dy}_{1/5}\text{Ho}_{1/5}\text{Tm}_{1/5})\text{Ta}_3\text{O}_9$ (SHEP-1), (d–f) $(\text{Gd}_{1/5}\text{Dy}_{1/5}\text{Ho}_{1/5}\text{Er}_{1/5}\text{Tm}_{1/5})\text{Ta}_3\text{O}_9$ (SHEP-2), (g–i) $(\text{La}_{1/6}\text{Sm}_{1/6}\text{Eu}_{1/6}\text{Dy}_{1/6}\text{Ho}_{1/6}\text{Tm}_{1/6})\text{Ta}_3\text{O}_9$ (6HEP-3), and (j–l) $(\text{Eu}_{1/6}\text{Gd}_{1/6}\text{Dy}_{1/6}\text{Ho}_{1/6}\text{Er}_{1/6}\text{Tm}_{1/6})\text{Ta}_3\text{O}_9$ (6HEP-4).

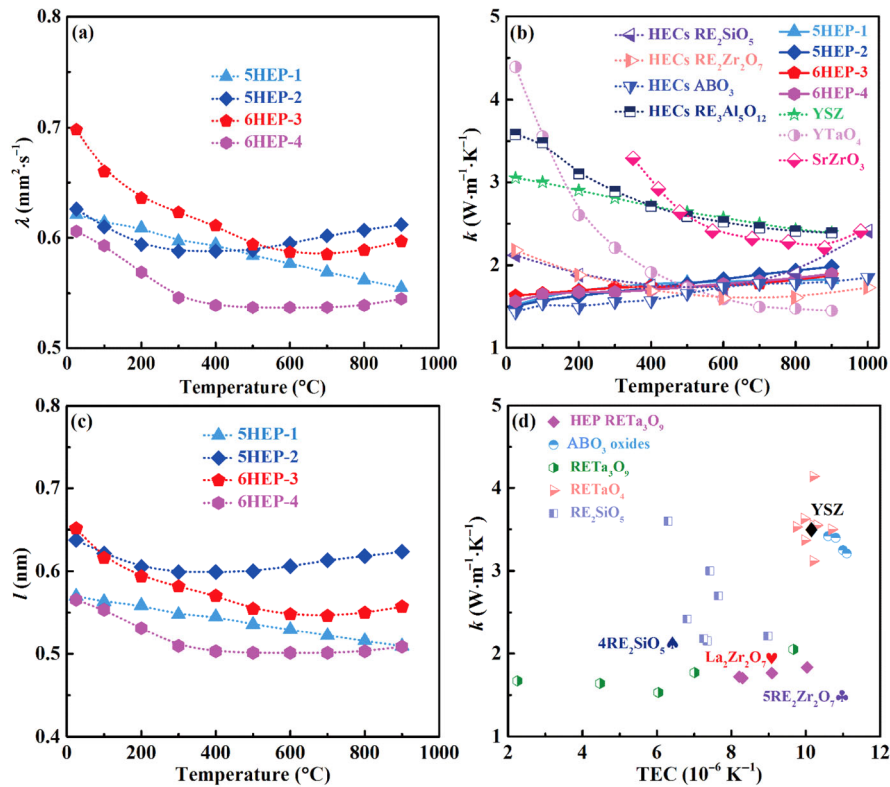


Fig. 8 Thermal properties of $(La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta_3O_9$ (5HEP-1), $(Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta_3O_9$ (5HEP-2), $(La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta_3O_9$ (6HEP-3), and $(Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta_3O_9$ (6HEP-4) ceramics: (a) thermal diffusivity, (b) thermal conductivity, (c) phonon mean free path, and (d) thermal properties compared with other oxides in Refs. [6,10,14,15,17,23,42,44,53–58].

other oxides [57,58]. Note that HEP $RETa_3O_9$ and HECs ABO_3 ($Zn_{0.1}Ca_{0.1}Sr_{0.4}Ba_{0.4}ZrO_3$) [17] exhibit similar thermal conductivity, which implies that amorphous-like thermal conductivity may be a common feature of HEP oxides. Furthermore, the thermal conductivity of the HEP samples is lower than those of HECs of RE_2SiO_5 [31] and $RE_3Al_5O_{12}$ [30], and close to those of HECs of $RE_2Zr_2O_7$ [20]. The lower the thermal conductivity of an HEP sample, the higher the thermal insulation performance is. The temperature dependence of the phonon mean free path (l) shown in Fig. 8(c) is similar to that of the thermal diffusivity, which further suggests the presence of thermal radiation. Figure 8(d) shows that the TECs range from 8.0×10^{-6} to $10.1 \times 10^{-6} K^{-1}$ (1473 K), which exceeds those of $GdTa_3O_9$ ($3.6 \times 10^{-6} K^{-1}$), $SmTa_3O_9$ ($5.0 \times 10^{-6} K^{-1}$), and $DyTa_3O_9$ ($7.5 \times 10^{-6} K^{-1}$), and lattice relaxation may be the predominant factor in the enhanced thermal expansion of the HEP samples [15]. The enhanced TECs of the HEP $RETa_3O_9$ can expand applications of these materials, e.g., as EBCs on the surface of Al_2O_3 fibre-enhanced Al_2O_3 ceramic matrix composites ($Al_2O_3_f/Al_2O_3$ with TECs of 8.5×10^{-6} – $9.0 \times 10^{-6} K^{-1}$)

because of matched TECs [10]. In addition, no phase transition is detected during the TECs test up to 1200 °C, which demonstrates the extraordinarily high-temperature phase stability of the prepared HEP $RETa_3O_9$ ceramics. Obviously, the TECs of the $RETa_3O_9$ are successfully regulated in the fabricated HECs, which exhibit outstanding phase stability.

Figure 8(d) shows that the HEP $RETa_3O_9$ oxides have higher TECs compared with those of RE_2SiO_5 (6.3×10^{-6} – $9.0 \times 10^{-6} K^{-1}$) [58] and $La_2Zr_2O_7$ ($9.0 \times 10^{-6} K^{-1}$) [6] and lower thermal conductivity compared with various E/TBCs [6,10,14,15,17,23,42,44,53–58]. HECs typically have lower thermal conductivity compared with their parent ceramics; however, this trend is not borne out for HEP $RETa_3O_9$. The point defect phonon scattering principle is not applicable for analysing the thermal transport properties of the prepared samples because of (1) the formation of random solid solutions, (2) severe lattice distortion, and (3) the cocktail effect of HECs [59,60]. As the E/k ratio reflects the phonon scattering rate, an increase in E/k causes a decrease in the thermal conductivity. The E/k ratio is 113.8, 92.6, 96.9, and 114.1 $GPa \cdot W^{-1} \cdot m \cdot K$ for 5HEP-1 (k :

1.52 W·m⁻¹·K⁻¹), 5HEP-2 (*k*: 1.50 W·m⁻¹·K⁻¹), 6HEP-3 (*k*: 1.63 W·m⁻¹·K⁻¹), and 6HEP-4 (*k*: 1.56 W·m⁻¹·K⁻¹), respectively. Herein, the prominent differences in the Young’s modulus, *E/k* ratio, and acoustic velocity among the four HEP ceramics do not noticeably affect the thermal conductivity. Grain refinements are found in the prepared samples: Fig. 2 shows that most grains are micron-sized and are therefore considerably larger than the nanoscale phonon mean free path. Herein, the phonon scattering strength from the grain boundary is not taken into account. The increasing degree of lattice distortion may decrease the thermal conductivity of perovskites, where the distortion degree is reflected by the tolerance factor (*t*). In the HEP RETa₃O₉, fluctuations in the disorder parameters for the electronegativity (δ_A^e), ionic radius (δ_A^r), and atomic mass (δ_A^m) in the A-site sublattice may change the thermal conductivity [17,61,62]:

$$\delta_A^e = \sqrt{\sum_i c_i \left(1 - \frac{e_A^i}{\bar{e}_A}\right)^2} \tag{11}$$

$$\delta_A^r = \sqrt{\sum_i c_i \left(1 - \frac{r_A^i}{\bar{r}_A}\right)^2} \tag{12}$$

$$\delta_A^m = \sqrt{\sum_i c_i \left(1 - \frac{m_A^i}{\bar{m}_A}\right)^2} \tag{13}$$

where *c_i* denotes the atomic ratio of the *i*th RE element; *e_Aⁱ* denotes the electronegativity of the *i*th RE element, and \bar{e}_A denotes the mean electronegativity of an A-site atom; *r_Aⁱ* denotes the ionic radius of the *i*th RE element, and \bar{r}_A denotes the mean ionic radius of an A-site atom; *m_Aⁱ* denotes the atomic mass of the *i*th RE element, and \bar{m}_A indicates the mean atomic mass of an A-site atom. Table 3 displays the values of the aforementioned parameters. The variation trends in the thermal conductivity with *t*, δ_A^e , δ_A^r , and δ_A^m are shown in Fig. 9. No regular variation trends are found

Table 3 Effective RE³⁺ ionic radius at the A-site (*r_A*), tolerance factor (*t*), electronegativity disorder parameter (δ_A^e), ionic radius disorder parameter (δ_A^r), and atomic mass disorder parameter (δ_A^m) of HEP RETa₃O₉ ceramics

Sample	<i>r_A</i> (nm)	<i>t</i>	δ_A^e	δ_A^r	δ_A^m
5HEP-1	0.109	0.871	0.082	0.072	0.050
5HEP-2	0.105	0.855	0.024	0.069	0.014
6HEP-3	0.108	0.866	0.065	0.065	0.041
6HEP-4	0.105	0.856	0.036	0.063	0.019

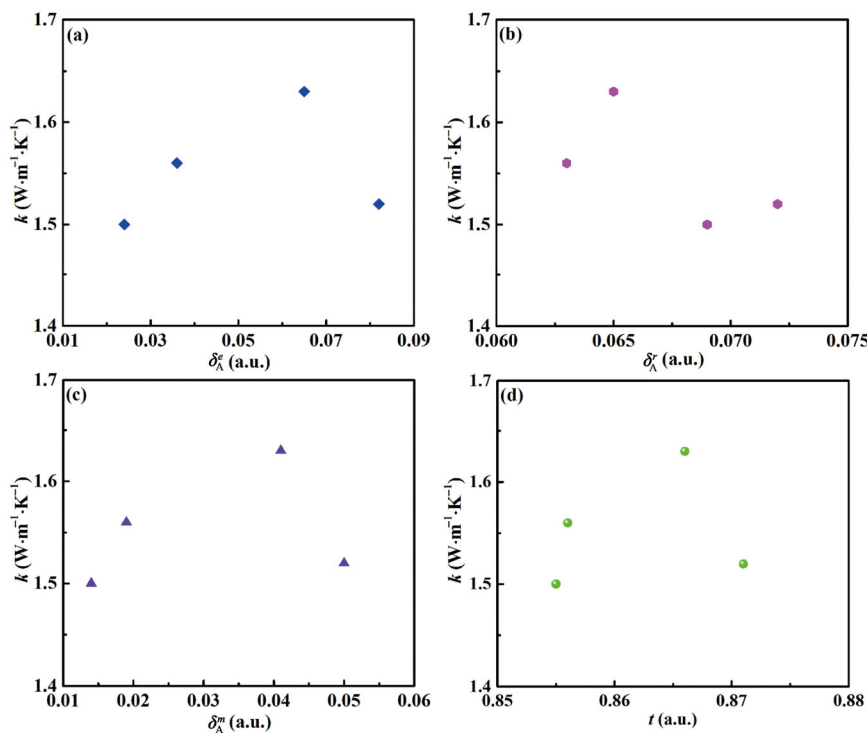


Fig. 9 Variation trends in thermal conductivity with different parameters of (La_{1/5}Pr_{1/5}Dy_{1/5}Ho_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-1), (Gd_{1/5}Dy_{1/5}Ho_{1/5}Er_{1/5}Tm_{1/5})Ta₃O₉ (5HEP-2), (La_{1/6}Sm_{1/6}Eu_{1/6}Dy_{1/6}Ho_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-3), and (Eu_{1/6}Gd_{1/6}Dy_{1/6}Ho_{1/6}Er_{1/6}Tm_{1/6})Ta₃O₉ (6HEP-4) ceramics: A-site atomic disorder parameters of (a) electronegativity, (b) ionic radius, and (c) atomic mass; (d) tolerance factor.

for the thermal conductivity with the aforementioned four parameters; thus, more HEP specimens should be fabricated and tested to identify variation laws. The four investigated samples clearly exhibit low thermal conductivity and high TECs, and they are therefore candidate E/TBCs.

4 Conclusions

Four novel A-site equiatomic HEP RETa_3O_9 ceramics are fabricated via a two-step sintering process, and the corresponding thermophysical properties are investigated and analysed as candidate E/TBCs for high-temperature applications. The contributions of this study are presented below:

1) Dense and high-purity HEP RETa_3O_9 bulk ceramics are successfully fabricated via SPS, and XRD, SEM, EDS, and 3D nanoindentation results demonstrate the excellent compositional homogeneity of all the samples.

2) The tolerance factor of the prepared HEP samples depends on the ionic radius of RE^{3+} , where the tolerance factor ranging from 0.8 to 0.9 shows that the samples have stable double-perovskite structures.

3) HEP RETa_3O_9 has a moderate Young's modulus (140–180 GPa) and hardness (7–10 GPa), and the brittle–ductile transition results in increased toughness ($1.8\text{--}2.5 \text{ MPa}\cdot\text{m}^{1/2}$).

4) Enhanced TECs ($8.0 \times 10^{-6}\text{--}10.1 \times 10^{-6} \text{ K}^{-1}$ at 1473 K) expand the applications of RETa_3O_9 , and a high melting point and extraordinary phase stability ensure high-temperature applications.

5) A low thermal conductivity ($1.5\text{--}2.0 \text{ W}\cdot\text{m}^{-1}\cdot\text{K}^{-1}$) indicates excellent thermal insulation by HEP RETa_3O_9 ceramics, and the results of this study have inspired a future study on tantalate HECs with desirable properties.

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