

## Preparation of rutile TiO<sub>2</sub> thin films by laser chemical vapor deposition method

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**Abstract:** TiO<sub>2</sub> thin films were prepared on Pt/Ti/SiO<sub>2</sub>/Si substrate by laser chemical vapor deposition (LCVD) method. The effects of laser power ( $P_L$ ) and total pressure ( $p_{\text{tot}}$ ) on the microstructure of TiO<sub>2</sub> thin films were investigated. The deposition temperature ( $T_{\text{dep}}$ ) was mainly affected by  $P_L$ , increasing with  $P_L$  increasing. The single-phase rutile TiO<sub>2</sub> thin films with different morphologies were obtained. The morphologies of TiO<sub>2</sub> thin films were classified into three typical types, including the powdery, Wulff-shaped and granular microstructures.  $p_{\text{tot}}$  and  $T_{\text{dep}}$  were the two critical factors that could be effectively used for controlling the morphology of the films.

**Keywords:** rutile TiO<sub>2</sub> thin film; laser chemical vapor deposition (LCVD); laser power; total pressure; microstructure

### 1 Introduction

The rutile TiO<sub>2</sub> films are widely investigated because they have many applications such as capacitors, sensors, antireflection coatings, and corrosion-resistant barriers [1–5]. The dielectric constant ( $\epsilon_r$ ) of the rutile TiO<sub>2</sub> crystal is anisotropic and has values of 170 in the  $c$  direction and 89 perpendicular to the  $c$  direction, which indicates that TiO<sub>2</sub> films have the possible application to future ultra-large-scale dynamic random access memory (DRAM) [6,7]. TiO<sub>2</sub> films have been prepared by many methods, including sputtering, conventional chemical vapor deposition, sol-gel

method, type-casting, and laser chemical vapor deposition (LCVD) [8–15]. Among these methods, LCVD is considered to be a promising process to prepare high-quality films with controllability of microstructure and orientation at high deposition rate ( $R_{\text{dep}}$ ) [16–21]. In LCVD process, the morphology and orientation of the films are affected by various parameters, such as laser power ( $P_L$ ), total pressure ( $p_{\text{tot}}$ ) in LCVD chamber, pre-heating temperature ( $T_{\text{pre}}$ ), and precursor evaporation temperature. In our previous study, the rutile TiO<sub>2</sub> thick films with random orientation were prepared by LCVD with  $\epsilon_r = 73$ , and the morphology and orientation of TiO<sub>2</sub> thick films were controlled by the variation of  $P_L$  [20,21].

In the present study, TiO<sub>2</sub> thin films are prepared on Pt/Ti/SiO<sub>2</sub>/Si substrate by LCVD, and the effects of  $p_{\text{tot}}$  and  $P_L$  on the orientation and microstructure of TiO<sub>2</sub> thin films are investigated.

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## 2 Experiment

TiO<sub>2</sub> thin films were prepared on Pt/Ti/SiO<sub>2</sub>/Si substrate by LCVD with a continuous-wave Nd:YAG laser (wavelength = 1064 nm). A schematic of LCVD apparatus has been reported elsewhere [16,19]. The substrate was heated on a hot stage at  $T_{\text{pre}} = 773$  K. A thermocouple was inserted at the bottom side of the substrate to measure the deposition temperature ( $T_{\text{dep}}$ ). A laser beam expanded to about 16 mm in diameter was introduced through a quartz window to irradiate the whole Pt/Ti/SiO<sub>2</sub>/Si substrate.  $P_L$  was changed from 48 W to 98 W. The titanium diisopropoxydipivaloylmethane (Ti(Oi-Pr)<sub>2</sub>(DPM)<sub>2</sub>, Toshiba Manufactory) precursor was heated at 423 K, and the vapor was carried into the chamber with Ar gas. O<sub>2</sub> gas was separately introduced into the chamber through a double-tube gas nozzle.  $p_{\text{tot}}$  in LCVD chamber was varied from 400 Pa to 950 Pa. The deposition was conducted for 300 s. Details of the deposition conditions are listed in Table 1.

**Table 1 Deposition conditions of TiO<sub>2</sub> thin films prepared by LCVD**

Ti(Oi-Pr) <sub>2</sub> (DPM) <sub>2</sub> evaporation temperature ( $T_{\text{Ti}}$ )	423 K
Substrate pre-heating temperature ( $T_{\text{pre}}$ )	773 K
Total chamber pressure ( $p_{\text{tot}}$ )	400–950 Pa
Gas flow rate	
Ar gas ( $FR_{\text{Ar}}$ )	$8.3 \times 10^{-7}$ m <sup>3</sup> /s
O <sub>2</sub> gas ( $FR_{\text{O}_2}$ )	$1.7 \times 10^{-6}$ m <sup>3</sup> /s
Laser power ( $P_L$ )	48–98 W
Deposition time ( $t$ )	300 s
Substrate–nozzle distance	30 mm
Substrate	Pt/Ti/SiO <sub>2</sub> /Si (10 mm × 10 mm × 0.5 mm)

The crystal structure of TiO<sub>2</sub> thin films was analyzed by X-ray diffraction (XRD, Rigaku RAD-2C) using Cu K $\alpha$  radiation. The surface and cross-sectional microstructures of TiO<sub>2</sub> thin films were observed by a field-effect scanning electron microscope (FE-SEM, JSM 6335F).

## 3 Results and discussion

Figure 1 shows the relationship between  $p_{\text{tot}}$  and  $T_{\text{dep}}$  of TiO<sub>2</sub> thin films prepared at different  $P_L$ . At the same  $p_{\text{tot}}$ , with increasing  $P_L$ ,  $T_{\text{dep}}$  also increases. As  $P_L$  is fixed,  $T_{\text{dep}}$  slightly changes with increasing  $p_{\text{tot}}$ , which

indicates that the relationship between  $T_{\text{dep}}$  and  $p_{\text{tot}}$  is very small. These results indicate that  $T_{\text{dep}}$  is mainly affected by  $P_L$ .

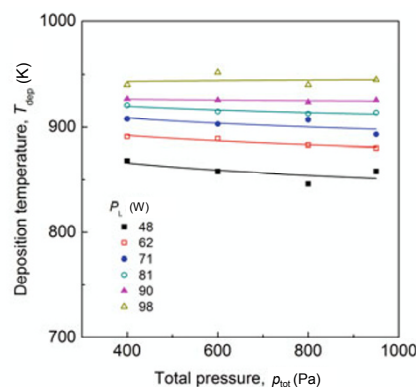


Fig. 1 Relationship between  $T_{\text{dep}}$  and  $p_{\text{tot}}$  of TiO<sub>2</sub> thin films prepared at different  $P_L$ .

Figure 2 depicts the typical XRD patterns of TiO<sub>2</sub> thin films prepared at  $P_L = 98$  W with different  $p_{\text{tot}}$ . The XRD patterns are indexed to the rutile TiO<sub>2</sub> phase (JCPDS 21-1276). The single-phase TiO<sub>2</sub> thin films are obtained, and the intensity of (110) peak decreases with increasing  $p_{\text{tot}}$ , which indicates that the crystallinity of the films becomes worse. In our previous study, the orientation of TiO<sub>2</sub> films changed from random to (110) orientation with increasing  $P_L$  from 48 W to 98 W at Ti(Oi-Pr)<sub>2</sub>(DPM)<sub>2</sub> evaporation temperature  $T_{\text{Ti}} = 433$  K [21]. However, in this study, all TiO<sub>2</sub> thin films prepared at different  $p_{\text{tot}}$  and  $P_L$  show the (110) orientation at  $T_{\text{Ti}} = 423$  K.  $T_{\text{Ti}}$  affects the pressure of Ti vapor, which further influences the gas-phase supersaturation. The supersaturation is one of the important parameters to control the structure of TiO<sub>2</sub> films [22].

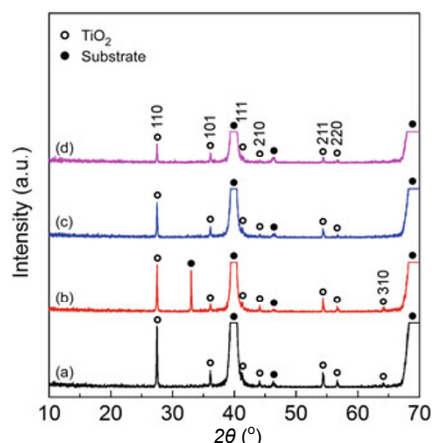


Fig. 2 XRD patterns of TiO<sub>2</sub> thin films prepared at  $P_L = 98$  W and different  $p_{\text{tot}}$ : (a) 400 Pa, (b) 600 Pa, (c) 800 Pa and (d) 950 Pa.

The surface and cross-sectional SEM images of TiO<sub>2</sub> thin films are shown in Fig. 3. In this study, the morphologies of all TiO<sub>2</sub> thin films are classified into three types. When TiO<sub>2</sub> thin film is prepared at  $p_{\text{tot}}=400$  Pa and  $P_L=90$  W ( $T_{\text{dep}}=927$  K), the film shows the powdery microstructure, as shown in Figs. 3(a) and 3(b). Kimura [23] reported that the powder was formed in gas phase due to the high  $T_{\text{dep}}$  when he prepared yttria-stabilized zirconia films by LCVD method. TiO<sub>2</sub> thin film prepared at  $p_{\text{tot}}=800$  Pa and  $P_L=98$  W ( $T_{\text{dep}}=940$  K) mainly consists of the Wulff-shaped grains with the columnar cross-section (Figs. 3(c) and 3(d)). Wulff construction gives the equilibrium crystal shape of a macroscopic crystal, and the Wulff-shaped rutile grains are commonly observed when the rutile nanocrystals are prepared [1,24]. The film prepared at  $p_{\text{tot}}=950$  Pa and  $P_L=48$  W ( $T_{\text{dep}}=858$  K) shows the granular microstructure and columnar cross-section, as shown in Figs. 3(e) and 3(f). Figure 4 displays the effects of  $p_{\text{tot}}$  and  $T_{\text{dep}}$  on the morphology of TiO<sub>2</sub> thin films. For TiO<sub>2</sub> thin films prepared at  $p_{\text{tot}}=400$  Pa and  $T_{\text{dep}}=921$ –940 K ( $P_L=81$ –98 W), and  $p_{\text{tot}}=600$  Pa and  $T_{\text{dep}}=952$  K ( $P_L=98$  W), they have the typical powdery microstructure. The powdery microstructure

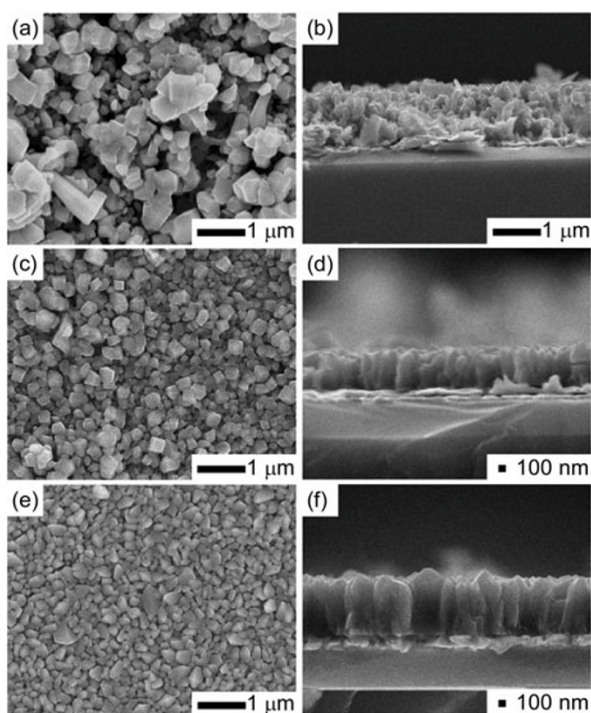


Fig. 3 Typical surface and cross-sectional images of TiO<sub>2</sub> thin films prepared at different  $p_{\text{tot}}$  and  $P_L$ : (a) and (b) 400 Pa and 90 W ( $T_{\text{dep}}=927$  K); (c) and (d) 800 Pa and 98 W ( $T_{\text{dep}}=940$  K); (e) and (f) 950 Pa and 48 W ( $T_{\text{dep}}=858$  K).

increases the surface area, which could result in superior photochemical properties of TiO<sub>2</sub> thin films. For TiO<sub>2</sub> thin films prepared at  $p_{\text{tot}}=400$  Pa and  $T_{\text{dep}}=891$ –908 K ( $P_L=62$ –71 W),  $p_{\text{tot}}=600$  Pa and  $T_{\text{dep}}=903$ –926 K ( $P_L=71$ –90 W),  $p_{\text{tot}}=800$  Pa and  $T_{\text{dep}}=924$ –940 K ( $P_L=90$ –98 W), and  $p_{\text{tot}}=950$  Pa and  $T_{\text{dep}}=926$ –945 K ( $P_L=90$ –98 W), they consist of the typical Wulff-shaped grains. When TiO<sub>2</sub> thin films are prepared at  $p_{\text{tot}}=400$  Pa and  $T_{\text{dep}}=868$  K ( $P_L=48$  W),  $p_{\text{tot}}=600$  Pa and  $T_{\text{dep}}=858$ –889 K ( $P_L=48$ –62 W),  $p_{\text{tot}}=800$  Pa and  $T_{\text{dep}}=846$ –913 K ( $P_L=48$ –81 W), and  $p_{\text{tot}}=950$  Pa and  $T_{\text{dep}}=858$ –914 K ( $P_L=48$ –81 W), they show the typical granular microstructure.

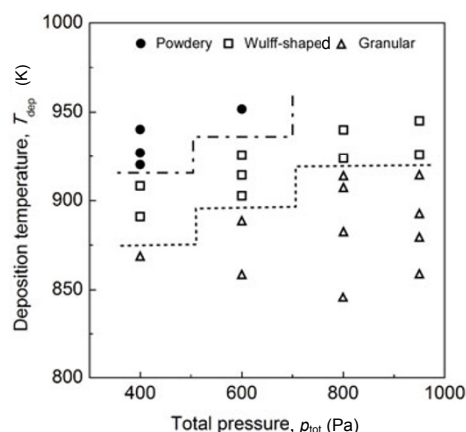


Fig. 4 Effects of  $p_{\text{tot}}$  and  $T_{\text{dep}}$  on the morphology of TiO<sub>2</sub> thin films.

Figure 5 demonstrates the relationship between  $R_{\text{dep}}$  and  $T_{\text{dep}}$  of TiO<sub>2</sub> thin films prepared at different  $p_{\text{tot}}$  in Arrhenius format.  $R_{\text{dep}}$  is obtained according to the cross-sectional SEM images of all TiO<sub>2</sub> thin films.  $R_{\text{dep}}$  slightly changes with the variation of  $T_{\text{dep}}$ , while  $R_{\text{dep}}$  obviously decreases with  $p_{\text{tot}}$  increasing.

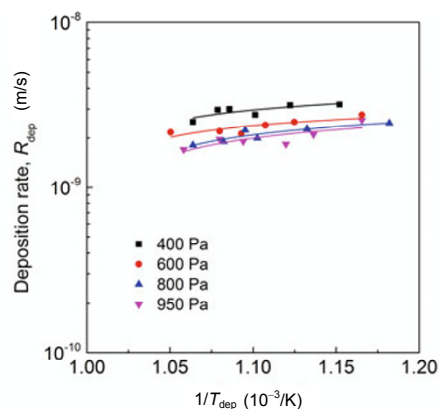


Fig. 5 Relationship between  $R_{\text{dep}}$  and  $T_{\text{dep}}$  of TiO<sub>2</sub> thin films prepared at different  $p_{\text{tot}}$  in Arrhenius format.

## 4 Conclusions

The single-phase rutile TiO<sub>2</sub> thin films were prepared on Pt/Ti/SiO<sub>2</sub>/Si substrate by LCVD. When the films were prepared at the same  $p_{\text{tot}}$ ,  $T_{\text{dep}}$  increased with increasing  $P_L$ . As  $P_L$  was fixed,  $T_{\text{dep}}$  slightly decreased with increasing  $p_{\text{tot}}$ . The morphologies of TiO<sub>2</sub> thin films were classified into three typical types, including the powdery, Wulff-shaped and granular microstructures.  $p_{\text{tot}}$  and  $T_{\text{dep}}$  were the two critical factors that could be effectively used for controlling the morphology of the films.  $R_{\text{dep}}$  slightly changed with the variation of  $T_{\text{dep}}$ , while  $R_{\text{dep}}$  obviously decreased with  $p_{\text{tot}}$  increasing.

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