## **ORIGINAL ARTICLE**





# Lignans and sesquiterpenoids from the stems of Schisandra bicolor var. tuberculata

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### Abstract

A pair of new tetrahydrofuran lignan enantiomers,  $(\pm)$ -schibiculatin A [ $(\pm)$ -1], a new enedione lignan, schibiculatin B (2), two new cadinane-type sesquiterpenoids, schibiculatins C (3) and D (4), along with two known seco-cadinanetype sesquiterpenoids (5 and 6) and seven known miscellaneous lignans (7-13) were isolated from the stems of Schisandra bicolor var. tuberculata. The structures of 1-4 were elucidated by comprehensive analysis of their spectroscopic data, guantum chemical calculations, as well as single-crystal X-ray diffraction. A few isolated compounds were tested for their protective activities against corticosterone-induced apoptosis in PC12 cells. Among them, compounds 5 and 6 showed moderate activities.

Keywords: Schisandra bicolor var. tuberculata, Cadinane-type sesquiterpenoids, Lignans, Quantum chemical calculation

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### 1 Introduction

The family Schisandraceae, including genera *Schisandra* and *Kadsura*, is a medicinally important plant taxon widely distributed in Southeastern Asia and Southeastern America. Previous phytochemical investigations on Schisandraceae species have led to the discovery of an array of structurally diverse natural products, including lignans [1, 2], triterpenoids [3–6], sesquiterpenoids [7, 8], etc. These compounds were found to exhibit various beneficial biological functions, including anti-HIV activity [9], hepatoprotective activity [1, 10], anti-hepatitis activity [11], neuroprotective activity [6, 12], inhibitory activity on proliferation of rheumatoid arthritis-fibroblastoid synovial cells [13], and so on.

Schisandra bicolor var. tuberculata is a climbing plant widely distributed in Guizhou, Hunan, and Jiangxi Provinces of China. Several studies have been undertaken to explore the chemical constituents and their bioactivities of S. bicolor var. tuberculata and its original variant, S. bicolor. For example, two new 3,4-seco-cycloartane triterpenoids from the stems of S. bicolor were reported by Qin et al. [14]; seven described cadinane-type sesquiterpenoids were isolated by Huang et al. from the stems of *S*. bicolor, and most of them were considered as chemotaxonomic markers of this species [7]; tetrahydrofuran-type [15], dibenzylbutane-type [16], and dibenzocyclooctanetype lignans have also been obtained from S. bicolor var. tuberculata or S. bicolor, and some of them were found to exhibit statistically significant neuroprotective activities [12]. Considering that S. bicolor var. tuberculata has rarely been phytochemically investigated, in the current research, S. var. tuberculata collected from Guizhou Province, China was subjected to phytochemical study to search for structurally and biologically fascinating molecules. As a result, a pair of new tetrahydrofuran lignan enantiomers,  $(\pm)$ -schibiculatin A  $[(\pm)-1]$ , a new enedione lignan, schibiculatin B (2), two new cadinane-type sesquiterpenoids, schibiculatins C (3) and D (4), along with two known *seco*-cadinane-type sesquiterpenoids (5 and 6) and seven known miscellaneous lignans (7–13) were obtained (Fig. 1). The structures of new compounds were elucidated by comprehensive analysis of their spectroscopic data, quantum chemical calculations, as well as single-crystal X-ray diffraction. This paper deals with the isolation, structural characterization, and bioactivity evaluation of these compounds.

### 2 Results and discussion

Compound 1 (schibiculatin A) was obtained as colorless crystals. It was assigned the molecular formula of  $C_{25}H_{34}O_9$  according to the ion peak at m/z 501.2102 ([M+Na]<sup>+</sup>, calcd for 501.2095) in the HR-ESI-MS analvsis and its <sup>13</sup>C NMR spectrum, suggesting nine degrees of unsaturation. The IR spectrum showed characteristic absorption peaks for hydroxyl group (3436  $\text{cm}^{-1}$ ) and aromatic moieties (1630, 1594, 1507, 1463  $\rm cm^{-1}$  ). The <sup>1</sup>H NMR spectrum showed two methyl groups ( $\delta_{\rm H}$  0.94 and 1.34); two methine protons ( $\delta_{\rm H}$  2.44 and 4.81); four aromatic protons [ $\delta_{\rm H}$  6.75 (2H), 6.83 (2H)]; and seven methoxy groups [ $\delta_{\rm H}$  3.23 (3H), 3.77 (6H), 3.86 (12H)]. Besides, the <sup>13</sup>C and DEPT NMR spectra showed signals for two methyls ( $\delta_{\rm C}$  8.8 and 19.4), two methines ( $\delta_{\rm C}$ 50.4 and 89.0), two non-protonated carbons ( $\delta_{\rm C}$  83.6 and 113.4), as well as eight resonances for aromatic



carbons ( $\delta_{\rm C}$  105.8, 107.2, 134.1, 138.7, 138.9, 139.0, 154.0, and 154.5). The above information (Table 1), in combination with the HMBC spectrum, suggested that compound 1 is a tetrahydrofuran lignan with two 3,4,5-trimethoxyphenyl groups. A comprehensive analysis of the  ${}^{1}H{-}^{1}H$  COSY and HMBC spectra (Fig. 2) indicated that the structure of 1 is closely related to that of kalongirin A [17], and the main difference was that the latter possessed only one symmetric trimethoxyphenyl ring. The HMBC correlations from H-7' to C-7 and C-8, from 7-OCH<sub>3</sub> ( $\delta_{\rm H}$  3.23) to C-7, from H<sub>3</sub>-9 to C-7 and C-8, in combination with the H-7'/H-8'/H-9' spin system revealed by the <sup>1</sup>H-<sup>1</sup>H COSY spectrum, confirmed the tetrahydrofuran moiety in 1. The HMBC correlations from six methoxy groups to C-3, C-4, C-5, C-3', C-4', and C-5, respectively, along with the HMBC correlations from H-7' to C-2' and C-6', confirmed the existence of two 3,4,5-trimethoxyphenyl groups and their attachment to C-7 and C-7', respectively. The key correlations between H-9 and H-8', H-9 and CH<sub>3</sub>O-7, H-7' and H-9' in the ROESY spectrum (Fig. 3), along with the  ${}^{3}J_{H-7'/H-8'}$  (10.2 Hz) revealed the relative configuration of 1 to be 7S\*,  $8S^*$ ,  $7'S^*$ , and  $8'R^*$ . Thus, 1 was determined as  $(7S^*, 8S^*, 7'S^*, 8'R^*)$ -8-hydroxy-3,4,5,7,3',4',5'-heptamethoxy-7,7'-epoxylignan. Crystals of 1 were obtained and subjected to X-ray diffraction analysis (Fig. 4), which not only proved the established structure, but also revealed that it was a racemic mixture. Chiral resolution of 1 afforded (+)-1 and (-)-1, and their absolute configurations were determined through TDDFT ECD calculation. According to the computational result (Fig. 5), the absolute configuration of (+)-1 was determined as 7*R*, 8*R*, 7'*R*, and 8'S, and that of (-)-1 was determined as 7S, 8S, 7'S, and 8′*R*.

<sup>a</sup> Recorded in methanol-d<sub>4</sub>

<sup>b</sup> Recorded in chloroform-d

<sup>c</sup> Recorded at 150 MHz

<sup>d</sup> Recorded at 600 MHz

Compound 2 (schibiculatin B) was obtained as a colorless oil. Its molecular formula was assigned as  $C_{23}H_{26}O_7$  according to the sodium adduct ion at m/z 437.1566 ([M+Na]<sup>+</sup>, calcd for 437.1571) in the (+)-HR-ESI-MS spectrum, indicating eleven degrees of unsaturation. The IR spectrum showed characteristic absorption peaks for aromatic moieties (1649, 1584, 1511, 1463, and 1416 cm<sup>-1</sup>), and carboxyl groups (1737 and 1722 cm<sup>-1</sup>). Its <sup>1</sup>H NMR spectrum (Table 1) showed two methyl groups [ $\delta_{\rm H}$  2.17 (6H)] and five aromatic protons [ $\delta_{\rm H}$  6.71 (2H), 6.74, 6.93, 7.18], and five methoxy groups ( $\delta_{\rm H}$  3.77, 3.78 (6H), 3.87, 3.90). The <sup>13</sup>C NMR, DEPT, and HSQC spectra exhibited 23 carbons (Table 1), including two methyl groups ( $\delta_{\rm C}$  17.3 and 17.5), fourteen olefinic carbons [ $\delta_{\rm C}$  106.6 (2C), 109.5, 110.2, 124.9, 130.4, 132.3, 138.0, 139.4, 142.3, 148.9, 152.7 (2C), 153.3], two carboxyl groups ( $\delta_{\rm C}$  197.7 and 197.9), and five methoxy groups [ $\delta_{\rm C}$  55.8, 56.0, 56.1 (2C), 60.9]. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **2** were similar to those of *threo*-2-methyl-3-oxo-1-(3',4',5'-trimethoxyphenyl) butyl-3",4"-dimethoxybenzoate [18], and the main difference was the substitution patterns of two aromatic rings. Careful analysis of the above information (Table 1) suggested the presence of a 3,4,5-trimethoxyphenyl group and a 3,4-dimethoxyphenyl group in **2**. The HMBC correlations from H-2, H-6 to C-7 ( $\delta_C$ 197.9), and from H-2', H-6' to C-7' ( $\delta_{\rm C}$  197.7) assigned the locations of two carboxyl groups. In addition, the HMBC correlations from H-9 to C-7 and C-8', from H-9' to C-8 and C-7' established the C-8/C-8' tetrasubstituted double bond. However, the geometry of C-8/ C-8' double bond can't be determined through ROESY correlations due to the overlap of H-9 and H-9' in <sup>1</sup>H NMR spectrum. Thus, two geometric isomers of 2, 8E-2 (2a) and 8Z-2 (2b) were subjected to quantum chemical calculation of NMR chemical shifts. According to the computational results, including R<sup>2</sup>, MAE, CMAE [19], and DP4+ probability [20] (Table 3), the C-8/C-8' double bond was determined to adopt E geometry. Thus, 2 was determined as (E)-1-(3',4',5'-trimethoxyphenyl)-4-(3",4"-dimethoxyphenyl)-2,3-dimethylbut-2-ene-1,4-

dione. Compound 3 (schibiculatin C) was obtained as a colorless oil. Its molecular formula was established as  $C_{15}H_{24}O_3$  by analysis of its (+)-HR-ESI-MS data ([M+Na]<sup>+</sup> m/z 275.1621, calcd for 275.1618), suggesting four degrees of unsaturation. The IR spectrum of 3 showed a characteristic absorption peak (3428 cm<sup>-1</sup>) for a hydroxy group. The <sup>1</sup>H spectrum showed four singlet methyl groups ( $\delta_{\rm H}$  1.17, 1.32, 1.36, and 1.64); two methine protons ( $\delta_{\rm H}$  2.76 and 4.44), two hydroxyl protons ( $\delta_{\rm H}$  5.72 and 6.37). The <sup>13</sup>C NMR and DEPT spectra, in combination with the HSQC spectrum showed fifteen carbon resonances, including four methyls ( $\delta_{\rm C}$  22.2, 24.5, 28.4, and 29.4), four methylenes ( $\delta_{\rm C}$  21.3, 22.2, 32.8, and 33.8), two methines ( $\delta_{\rm C}$  41.5 and 76.4), and five non-protonated carbons (including three oxygenated methines at  $\delta_{\rm C}$  71.8, 73.1, and 74.1; two olefinic carbons at  $\delta_{\rm C}$  135.6 and 140.2). The aforementioned NMR data of 3 (Table 2) indicated it is a cadinane-type sesquiterpene structurally similar to (4*R*,5*R*,10*R*)-10-methoxymuurol-1(6)-ene-4,5-diol [21]. The  ${}^{1}H-{}^{1}H$  COSY spectrum of **3** revealed the presence of three spin systems (Fig. 2) through H-2a ( $\delta_{\rm H}$  2.53)/H-3b ( $\delta_{\rm H}$  1.97), HO-5 ( $\delta_{\rm H}$  6.37)/H-5 ( $\delta_{\rm H}$  4.44), and H-7 ( $\delta_{\rm H}$ 2.76)/H-8b ( $\delta_{\rm H}$  1.52)/H-9a ( $\delta_{\rm H}$  1.76) correlations. In the HMBC spectrum of **3** (Fig. 2), correlations from  $CH_3$ -13 ( $\delta_{\rm H}$  1.17) to C-7 ( $\delta_{\rm C}$  41.5), C-11 ( $\delta_{\rm C}$  74.1), and C-12 ( $\delta_{\rm C}$ 28.4); from CH<sub>3</sub>-14 ( $\delta_{\rm H}$  1.36) to C-1 ( $\delta_{\rm C}$  135.6), C-9 ( $\delta_{\rm C}$ 32.8), and C-10 ( $\delta_{\rm C}$  73.1); from CH<sub>3</sub>-15 ( $\delta_{\rm H}$  1.64) to C-3 ( $\delta_{\rm C}$  33.8), C-4 ( $\delta_{\rm C}$  71.8), and C-5 ( $\delta_{\rm C}$  76.4) indicated that two methyls and one isopropyl group were attached to C-4, C-10, and C-7, respectively. Besides, the HMBC

Table 1 <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data for compounds 1 and **2** ( $\delta$  in ppm, J in Hz)

No.	(±)-1		2		
	$\delta_{C}^{a,c}$	$\delta_{\rm H}^{\rm a,d}$ , (mult <i>J</i> )	δ <sub>c</sub> <sup>b,c</sup>	$\delta_{\rm H}^{\rm b,d}$ , (mult J)	
1	138.7, C	_	132.3, C	-	
2/6	107.2, CH	6.83 (s)	106.6, CH	6.71 (s)	
3/5	134.1, C	-	152.7, C	-	
4	154.0, C	-	142.3, C	-	
7	113.4, C	-	197.9, C	-	
8	83.6, C	-	139.4, C	-	
9	19.4, CH <sub>3</sub>	1.34 (s)	17.5, CH <sub>3</sub>	2.17 (overlap)	
1′	138.9, C	-	130.4, C	-	
2′	105.8, CH	6.75 (s)	110.2, CH	6.93 (s)	
3′	139.0, C	-	148.9, C	-	
4′	154.5, C	-	153.3, C	_	
5'	139.0, C	-	109.5, CH	6.74 (d, 8.3)	
6′	105.8, CH	6.75 (s)	124.9, CH	7.18 (d, 8.3)	
7′	89.0, CH	4.81 (d, 10.2)	197.7, C	-	
8′	50.4, CH	2.44 (dq, 10.2, 6.8)	138.0, C	-	
9′	8.8, CH <sub>3</sub>	0.94 (d, 6.8)	17.3, CH <sub>3</sub>	2.17 (overlap)	
7-OCH <sub>3</sub>	50.7, CH <sub>3</sub>	3.23 (s)	-	-	
3/5-OCH <sub>3</sub>	56.6, CH <sub>3</sub>	3.86 (s)	56.1, CH <sub>3</sub>	3.78 (s)	
4-OCH <sub>3</sub>	61.0, CH <sub>3</sub>	3.77 (s)	60.9, CH <sub>3</sub>	3.87 (s)	
3'-OMe	56.6, CH <sub>3</sub>	3.86 (s)	55.8, CH <sub>3</sub>	3.77 (s)	
4'-OMe	61.1, CH <sub>3</sub>	3.77 (s)	56.0, CH <sub>3</sub>	3.90 (s)	
5'-OMe	56.6, CH <sub>3</sub>	3.86 (s)	-	-	







correlations (Fig. 2) from H-3a, H-3b, H-5, H-7, H-9a and H-9b to C-1; and from H-2, H-8, and HO-5 to C-6 ( $\delta_{C}$  140.2), suggested the presence of the C-1/C-6 double bond. The remaining one degree of unsaturation implied the existence of an epoxy ring, which was deduced to form between C-10 and C-11 according to molecular modeling. Assigning H-5 as  $\alpha$ -oriented, the H-7/H-5 $\alpha$ , CH<sub>3</sub>-13/H-5 $\alpha$ , CH<sub>3</sub>-15/H-5 $\alpha$  correlations (Fig. 3) in the ROESY spectrum indicated that both CH<sub>3</sub>-15 and the epoxy ring adopt  $\alpha$ -orientation. Subsequently, quantum chemical calculation of NMR chemical shifts succeeded in confirming the established the structure of **3** (Table **3**). However, TDDFT calculation failed to produce the experimental ECD curve of 3, despite that several independent methods have been tried. Thus, the absolute configuration of 3 was left undermined. The structure of 3 was established as 10,11-epoxy-cadin-1(6)-ene-4,5-diol.

Compound 4 (schibiculatin D) was isolated as a colorless oil. Its molecular formula was assigned as  $C_{15}H_{24}O_4$ according to the sodium adduct ion peak at *m/z* 291.1566

No.	3		4		
	$\delta_{C}^{a,c}$	$\delta_{\rm H}^{\rm a,d}$ , (mult J)	δ <sub>C</sub> <sup>b,c</sup>	$\delta_{\rm H}^{\rm b,d}$ , (mult <i>J</i> )	
1	135.6, C	_	159.6, C	_	
2	22.2, CH <sub>2</sub>	2.53 (m) 2.07 (overlap)	28.3, CH <sub>2</sub>	2.58 (overlap)	
3	33.8, CH <sub>2</sub>	2.07 (overlap) 1.97 (m)	36.3 CH <sub>2</sub>	1.99 (m) 1.92 (m)	
4	71.8, C	-	73.4, C	-	
5	76.4, CH	4.44 (s)	202.7, C	-	
6	140.2	-	137.2, C	-	
7	41.5, CH	2.76 (t, 2.7)	38.7, CH	2.58 (overlap)	
8	21.3, CH <sub>2</sub>	2.07 (overlap) 1.52 (m)	20.2, CH <sub>2</sub>	1.62 (m)	
9	32.8 CH <sub>2</sub>	1.76 (m) 1.33 (overlap)	31.3, CH <sub>2</sub>	2.22 (m) 1.41 (m)	
10	73.1, C	-	74.1, C	-	
11	74.1, C	_	30.7, CH	2.10 (dq, 13.6, 6.8)	
12	28.4, CH <sub>3</sub>	1.32 (overlap)	21.4, CH <sub>3</sub>	0.85 (d, 6.8)	
13	29.4, CH <sub>3</sub>	1.17 (s)	19.1, CH <sub>3</sub>	0.79 (d, 6.8)	
14	22.2, CH <sub>3</sub>	1.36 (s)	67.1, CH <sub>2</sub>	3.66 (d, 11.2) 3.49 (d, 11.2)	
15	24.5, CH <sub>3</sub>	1.64 (s)	24.1, CH <sub>3</sub>	1.24 (s)	
4-OH	-	5.72 (s)	-	_	
5-OH	-	6.37 (d, 5.7)	-	-	

**Table 2** <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data for compounds **3** and **4** ( $\delta$  in ppm, *J* in Hz)

<sup>a</sup> Recorded in pyridine- $d_5$ 

<sup>b</sup> Recorded in methanol-d<sub>4</sub>

<sup>c</sup> Recorded at 150 MHz

<sup>d</sup> Recorded at 600 MHz

([M+Na]<sup>+</sup>, calcd 291.1567) in the (+)-HR-ESI-MS spectrum, indicating four degrees of unsaturation. The IR spectrum showed characteristic absorption peaks for hydroxyl group (3427 cm<sup>-1</sup>), double bond (1462, 1611 cm<sup>-1</sup>), and carboxyl group (1663 cm<sup>-1</sup>). Its <sup>1</sup>H NMR spectrum showed three methyl groups ( $\delta_{\rm H}$  0.79, 0.85, and 1.24) and two oxygenated protons ( $\delta_{\rm H}$  3.49 and 3.66). The <sup>13</sup>C NMR, DEPT, and HSQC spectra exhibited 15 carbon resonances, including three methyl groups ( $\delta_{\rm C}$  19.1, 21.4, and 24.1), five methylenes ( $\delta_{\rm C}$  20.2, 23.8, 31.3, 36.3, and 67.1), two methines ( $\delta_{\rm C}$  30.7 and 38.7), five non-protonated carbons (including two oxygenated methines at  $\delta_{\rm C}$  73.4 and 74.1; two olefinic carbons at  $\delta_{\rm C}$ 137.2 and 159.6; one carboxyl carbon at  $\delta_{\rm C}$  202.7). The aforementioned NMR data of 4 (Table 2) revealed that it was also a cadinane-type sesquiterpene. The structure of 4 was found to be closely related to that of  $(4\alpha, 10\beta)$ -4,10-dihydroxycadin-1(6)-ene-5-one [22], with the main difference being that the CH<sub>3</sub>-14 in the latter was replaced by a hydroxy group-substituted methylene in 4. The above observation could be further confirmed by key

Parameters	<sup>13</sup> C data	<sup>13</sup> C data			<sup>1</sup> H data		
	R <sup>2a</sup>	MAE	CMAE	R <sup>2</sup>	MAE	CMAE	probability (all data) <sup>b</sup>
2a	0.9988	2.2	1.5	0.9959	0.45	0.09	100.00%
2b	0.9972	2.6	1.9	0.9914	0.20	0.12	0.00
3	0.9993	1.0	0.8	0.9506	0.33	0.15	-
4a	0.9992	1.7	1.1	0.9704	0.15	0.10	80.93%
4b	0.9988	1.8	1.6	0.9810	0.10	0.09	6.51%
4c	0.9977	2.1	2.1	0.9722	0.17	0.10	12.52%
4d	0.9989	1.7	1.5	0.9563	0.14	0.11	0.03%

Table 3 The results for NMR computation of compounds 2-4

MAE: Mean Absolute Error, CMAE: Corrected Mean Absolute Error

<sup>a</sup> R<sup>2</sup>: squared Pearson correlation coefficient between experimental and calculated NMR chemical shifts

<sup>b</sup> DP4+ probability (all data): DP4+ probability based on analyzing <sup>1</sup>H or <sup>13</sup>C data together. If <sup>1</sup>H chemical shifts of CH<sub>2</sub>, which were prone to error when assigning, are removed from analysis, the DP4+ probability can be more distinguishing (Figures S49 and S50)

HMBC correlations from H-14a ( $\delta_{\rm H}$  3.66) and H-14b ( $\delta_{\rm H}$ 3.49) to C-1 ( $\delta_{\rm C}$  159.6), C-9 ( $\delta_{\rm C}$  31.3), and C-10 ( $\delta_{\rm C}$  74.1). However, due to signal overlap in the <sup>1</sup>H spectrum (H-7 and H-2; H-8a and H-8b, for example), it was difficult to elucidate the relative configuration of 4 through ROESY correlations. Then, four possible diastereoisomers of 4,  $(4S^*, 7R^*, 10S^*)$ -4 (4a),  $(4S^*, 7R^*, 10R^*)$ -4 (4b),  $(4S^*, 7S^*, 10R^*)$ -4 (4b), (4S^\*, 7S^\*, 10R^\*)-4 (4b), (4S^\*, 7S^\*, 10R^\*) 10*R*\*)-4 (4c), and (4*S*\*,7*S*\*,10*S*\*)-4 (4d) were subjected to quantum chemical calculation of NMR chemical shifts. According to the computational results (Table 3), the relative configuration of 4 was determined as  $4S^*$ ,  $7R^*$ , and 10S\*. However, as the situation in 3, TDDFT calculation failed to produce the experimental ECD curve of 4. Thus, the absolute configuration of 4 was left undermined. Thus, the structure of 4 was determined as 4,10,14-trihydroxycadin-1(6)-en-5-one.

The structures of those known compounds, including (4R)-4-hydroxy-1,10-*seco*-muurol-5-ene-1,10-dione (5) [21], (4*S*)-4-hydroxy-1,10-*seco*-muurol-5-ene-1,10-dione (6) [21], kadlongirin A (7) [17], rel-(2*R*,3*R*,4*S*,5*S*)-tetrahydro-3,4-dimethyl-2,5-bis(3,4,5-trimethoxyphenyl) furan (8) [23], prinsepiol (9) [24], 8*α*-hydroxypinoresinol (10) [25], ginkgool (11) [26], massoiresinol (12) [27], acanthosessilin A (13) [28] were identified by comparing their spectroscopic data with those reported in literature.

Additionally, compounds (-)-1, (+)-1, 2, 3, 5, and 6 were evaluated for their protective effects against damage of PC12 cells induced by corticosterone. As a result, compounds 5 and 6 showed moderate activities (Table 4), which indicated their potential neuroprotective activities.

### **3** Experimental section

### 3.1 General

HRESIMS data were acquired on an Agilent 6540 QSTAR TOF time-of-flight mass spectrometer (Agilent Corp., America). A Tensor 27 spectrophotometer (Bruker

Table 4	Protective	activities	of	selected	compounds	against
corticosterone-induced apoptosis in PC12 Cells						

Compound	Survival rate (%)	Standard deviation (%)	Р
Blank	100.00	0.58	***
Negative control	59.71	0.52	
Desipramine	89.17	0.26	***
(—)- <b>1</b>	59.02	0.24	
(+)-1	59.71	0.43	
2	59.57	0.44	
3	59.73	0.28	
5	62.49	0.29	**
6	61.90	0.74	*

Data are analyzed by T test and F test, and presented as means  $\pm$  SD (\*p < 0.05, \*\*p < 0.01, \*\*\*p < 0.001)

Corp., Switzerland) was used for scanning IR spectroscopy with KBr pellets. UV spectra were obtained using a Shimadzu UV-2401 PC spectrophotometer (Shimadzu Corp., Japan). Experimental ECD spectra were measured on a Chirascan V100 instrument (Applied Photophysics Limited, Britain). Optical rotations were measured with a JASCO P-1020 polarimeter (Jasco Corp., Japan). Crystallographic data were obtained on a Bruker APEX DUO (Bruker Corp., Switzerland). 1D and 2D NMR spectra were recorded on Bruker AV III 500 MHz, Bruker DRX-600, or Bruker Ascend 800 MHz spectrometers (Bruker Corp., Switzerland) with TMS internal standard. Chemical shifts ( $\delta$ ) are expressed in ppm relative to the solvent signals. Semi-preparative HPLC was performed on an Agilent 1260 liquid chromatography (Agilent Corp., America) with a Zorbax SB-C18 (4.6  $mm \times 250 mm$ ) column and a CHIRALPAK IC (4.6 mm × 250 mm) column. Column chromatography (CC) was performed with silica gel (80-100, 100-200 and 200-300 mesh; Qingdao

Marine Chemical, Inc., Qingdao, People's Republic of China). MCI gel (75–150  $\mu$ m, Mitsubishi Chemical Corporation, Tokyo, Japan), Lichroprep RP-18 gel (40–63  $\mu$ m, Merck, Darmstadt, Germany), Sephadex LH-20 ((Pharmacia, Uppsala, Sweden) and SEPAFlash column (Spherical C-18, 20–45  $\mu$ m, 100 Åm). Fractions were monitored by thin layer chromatography and spots were detected with 10% H<sub>2</sub>SO<sub>4</sub> in EtOH.

### 3.2 Plant material

The stems of *Schisandra bicolor* var. *tuberculata* were collected in Tongren region, Guizhou Province, China, in April 2020, and identified by Prof. Heng Li, Kunming Institute of Botany. A voucher specimen has been deposited in the Herbarium of the Kunming Institute of Botany, Chinese Academy of Sciences.

### 3.3 Bioactivity evaluation

Poorly differentiated PC12 cells were maintained in DMEM medium supplemented with 10% fetal bovine serum (FBS), penicillin (100 U/mL), streptomycin (100 µg/mL), and incubated at 5% CO<sub>2</sub> and 37 °C. Poorly differentiated PC12 cells were divided into the following groups: untreated, CORT (150 µmol/L), CORT (150 mol/l) plus DIM (10 µmol/L), CORT (150 µmol/L) plus test compounds (20 µmol/L). Briefly, poorly differentiated PC12 cells were seeded into 96-well culture plates at a density of  $1 \times 10^4$  cells/well. After 24 h culturing, the wells were added compounds as previously described groups. 48 h later, MTS solution was added to each well. The absorbance was measured at 492 nm using a Thermo Multiskan FC [29].

### 3.4 Extraction and isolation

Air-dried and powdered stems (10.5 kg) of S. bicolor var. tuberculata were extracted with 70% aqueous Me<sub>2</sub>CO  $(3 \times 40 \text{ l})$  to yield an extract at room temperature to give a crude extract (1081 g), which was extracted with EtOAc. The EtOAc part (380 g) was separated by a silica gel column and eluted with a CHCl<sub>3</sub>/Me<sub>2</sub>CO gradient system (1:0 to 0:1) to afford five fractions A-E. Fractions B (45 g) and C (51 g) were successively decolorized on MCI gel with MeOH/H<sub>2</sub>O (90:10) to obtain B1 and C1. Fraction B1 (42 g) was subjected to ODS chromatography and eluted with MeOH/H<sub>2</sub>O (50:50 to 100:0) to give six fractions (B1a-B1f). Then B1a (18 g) was repeatedly subjected to silica gel column to obtain 8 (300 mg), other fractions further purified by preparative HPLC to afford 5 (30 mg) and 6 (20.0 mg). Fraction B1b (14 g) was repeatedly subjected to silica gel column, then purified by semi-preparative HPLC to give 1 (13.5 mg), 2 (3.0 mg) and 7 (2.9 mg). Fraction C1 (44 g) was subjected to ODS chromatography by eluting with MeOH/H<sub>2</sub>O (30:70 to 100:0) to give seven fractions. C1b (4.52 g) was chromatographed on flash column by eluting with MeOH/H<sub>2</sub>O (20:80 to 60:40) to yield 12 fractions bF1–bF12, and bF2 (850 mg) was then subjected to Sephadex LH-20 (CHCl<sub>3</sub>/MeOH, 1:1), and further purified by semi-preparative HPLC to give **3** (3.49 mg), and **4** (4.35 mg), **9** (10 mg), **10** (2.62 mg), **11** (3.71 mg). Fraction bF4 (420 mg) was subjected to Sephadex LH-20 (CHCl<sub>3</sub>-MeOH, 1:1), and further purified by semi-preparative HPLC to give **12** (5.40 mg) and **13** (3.71 mg).

# 3.5 Physical constants and spectroscopic data of compounds

(-)-Schibiculatin A [(-)-(1)]: colorless crystals, [ $\alpha$ ] – 37.45, (c 0.110, MeOH), UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 206 (4.96), 258 (3.24), 272 (3.40) nm. CD (CH<sub>3</sub>OH, c 0.05): 198 nm ( $\Delta \varepsilon$  = -11.19), 208 nm ( $\Delta \varepsilon$  = 31.90), 217 nm ( $\Delta \varepsilon$  = 0.73), 221 nm ( $\Delta \varepsilon$  = 1.60), 240 nm ( $\Delta \varepsilon$  = -7.31). IR (KBr)  $\nu_{max}$  3436, 2958, 2935, 1630, 1594, 1507, 1463, 1128 cm<sup>-1</sup>. ESIMS m/z 252 [M+Na]<sup>+</sup>, HRESIMS m/z501.2102 [M+Na]<sup>+</sup> (calcd for C<sub>25</sub>H<sub>34</sub>O<sub>9</sub>Na 501.2095). <sup>1</sup>H NMR (methanol- $d_4$ , 600 MHz) and <sup>13</sup>C NMR (methanol- $d_4$ , 150 MHz), see Table 1.

(+)-Schibiculatin A [(+)-(1)]: colorless crystals, [ $\alpha$ ] + 31.38 (c 0.087, MeOH), UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 206 (4.93), 259 (3.57), 268 (3.62) nm. ECD (CH<sub>3</sub>OH, c 0.05): 198 nm ( $\Delta \varepsilon$ =17.07), 208 nm ( $\Delta \varepsilon$ = -22.21), 217 nm ( $\Delta \varepsilon$ =4.14), 221 nm ( $\Delta \varepsilon$ =9.23), 240 nm ( $\Delta \varepsilon$ =3.03). IR (KBr)  $\nu_{max}$  3436, 2958, 2935, 1630, 1594, 1507, 1463, 1128 cm<sup>-1</sup>. ESIMS m/z 478 [M+Na]<sup>+</sup>, HRESIMS m/z501.2102 [M+Na]<sup>+</sup> (calcd for C<sub>25</sub>H<sub>34</sub>O<sub>9</sub>Na 501.2095). <sup>1</sup>H NMR (methanol- $d_4$ , 600 MHz) and <sup>13</sup>C NMR (methanol- $d_4$ , 150 MHz), see Table 1.

Schibiculatin B (2): colorless oil,  $[\alpha]$ +15.91 (*c* 0.093, MeOH), UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 202 (4.60), 256 (3.71), 281 (3.91) nm. IR (KBr)  $\nu_{max}$  2938, 2922, 1737, 1721, 1649, 1584, 1511, 1463, 1416, 1128 cm<sup>-1</sup>. ESIMS *m*/*z* 437 [M+Na]<sup>+</sup>, HRESIMS *m*/*z* 437.1566 [M+Na]<sup>+</sup> (calcd for C<sub>23</sub>H<sub>28</sub>O<sub>8</sub>Na 437.1571). <sup>1</sup>H NMR (chloroform-*d*, 600 MHz) and <sup>13</sup>C NMR (chloroform-*d*, 150 MHz), see Table 1.

Schibiculatin C (3): colorless oil,  $[\alpha] + 25.2$  (*c* 0.090, MeOH), UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 195 (3.74) nm. ECD (CH<sub>3</sub>OH, *c* 0.36): 199 nm ( $\Delta \varepsilon = 3.03$ ), 214 nm ( $\Delta \varepsilon = -2.01$ ). IR (KBr)  $\nu_{max}$  3428, 2966, 2921, 2859, 1631, 1560, 1384, 802 cm<sup>-1</sup>. ESIMS *m*/*z* 275 [M+Na]<sup>+</sup>, HRESIMS *m*/*z* 275.1621 [M+Na]<sup>+</sup> (calcd for C<sub>15</sub>H<sub>24</sub>O<sub>3</sub>Na 275.1618). <sup>1</sup>H NMR (pyridine-*d*<sub>5</sub>, 600 MHz) and <sup>13</sup>C NMR (pyridine-*d*<sub>5</sub>, 150 MHz), see Table 2.

Schibiculatin D (4): colorless oil,  $[\alpha] - 11.2$  (c 0.088, MeOH), UV (MeOH)  $\lambda_{max}$  (log  $\varepsilon$ ) 216 (3.53) nm, 248 (3.88) nm. ECD (CH<sub>3</sub>OH, c 0.26): 206 nm ( $\Delta \varepsilon = 2.69$ ), 248 nm ( $\Delta \varepsilon = -7.81$ ), 338 nm ( $\Delta \varepsilon = 1.95$ ). IR (KBr)  $\nu_{max}$  3427, 2958, 2930, 2859, 1663, 1611, 1463, 1383, 1060 cm<sup>-1</sup>. ESIMS m/z 268 [M+Na]<sup>+</sup>, HRESIMS m/z 291.1566 [M+Na]<sup>+</sup> (calcd for C<sub>15</sub>H<sub>24</sub>O<sub>4</sub>Na 291.1567). <sup>1</sup>H NMR (methanol- $d_4$ , 600 MHz) and <sup>13</sup>C NMR (methanol- $d_4$ , 150 MHz), see Table 2.

Crystal data for (±)-schibiculatin A [(±)-1]:  $C_{25}H_{34}O_9$ , M=478.52, a=7.9385(2) Å, b=9.5977(3)Å, c=17.1002(5) Å,  $\alpha=86.9070(10)^\circ$ ,  $\beta=84.8510(10)^\circ$ ,  $\gamma=66.3110(10)^\circ$ , V=1188.06(6) Å<sup>3</sup>, T=100.(2) K, space group *P*-1, Z=2,  $\mu$ (Cu K $\alpha$ ) = 0.843 mm<sup>-1</sup>, 39,839 reflections measured, 4665 independent reflections ( $R_{int}=0.0409$ ). The final  $R_1$  values were 0.0343 ( $I>2\sigma(I)$ ). The final  $wR(F^2)$  values were 0.0865 ( $I>2\sigma(I)$ ). The final  $R_1$ values were 0.0368 (all data). The final  $wR(F^2)$  values were 0.0882 (all data). The goodness of fit on  $F^2$  was 1.069. Crystallographic data for the structure of (±)-schibiculatin A [(±)-1] have been deposited in the Cambridge Crystallographic Data Centre (deposition number CCDC 2145390). Copies of the data can be obtained free of charge from the CCDC via www.ccdc.cam.ac.uk.

### Supplementary Information

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Additional file 1. It includes 1D NMR, 2D NMR, HRESIMS, UV, ECD, IR, OR, and computational data of compounds 1–4, and methods for quantum chemical calculations are available.

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### Author contributions

PTP conceived and designed the research; SMZ, KH carried out the experiment and wrote the manuscript; PTP, HDS and KH supervised the whole study and critically reviewed the manuscript; XNL determined crystal structure of ( $\pm$ )-**1** by X-ray diffraction analysis. All authors read and approved the final manuscript.

### Declarations

### **Competing interests**

No conflict of interest is declared.

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