# ORIGINAL ARTICLE



# Bioinspired green synthesis of copper oxide nanoparticles from *Syzygium alternifolium* (Wt.) Walp: characterization and evaluation of its synergistic antimicrobial and anticancer activity

Pulicherla Yugandhar<sup>1</sup> · Thirumalanadhuni Vasavi<sup>2</sup> · Palempalli Uma Maheswari Devi<sup>2</sup> · Nataru Savithramma<sup>1</sup>

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**Abstract** In recent times, nanoparticles are attributed to green nanotechnology methods to know the synergistic biological activities. To accomplish this phenomenon, present study was aimed to synthesize copper oxide nanoparticles (CuO NPs) by using Syzygium alternifolium stem bark, characterized those NPs using expository tools and to elucidate high prioritized antimicrobial and anticancer activities. Synthesized particles exhibited a color change pattern upon synthesis and affirmed its respective broad peak at 285 nm which was analyzed through UV-vis spectroscopy. FT-IR study confirmed that phenols and primary amines were mainly involved in capping and stabilization of nanoparticles. DLS and Zeta potential studies revealed narrow size of particles with greater stability. XRD studies revealed the crystallographic nature of particles with 17.2 nm average size. Microscopic analysis by using TEM revealed that particle size range from 5-13 nm and most of them were spherical in shape, non-agglomerated and poly-dispersed in condition. Antimicrobial studies of particles showed highest inhibitory activity against E. coli and T. harzianum among bacterial and fungal strains, respectively. The scope of this study is extended by examining anticancer activity of CuO NPs. This study exhibited potential anticancer activity towards MDA-MB-231 human breast cancer lines. Overall, these examinations relate that the S. alternifolium is described as efficient wellbeing plant and probabilistically for the design and

synthesis of nanoparticles for human health. This study paves a way to better understand antimicrobial and anticancer therapeutic drug potentials of nanoparticles to design and analysis of pharmaceuticals by in vivo and in vitro approaches.

**Keywords** *Syzygium alternifolium* · Stem bark · CuO NPs · Characterization · Antimicrobial activity · Anticancer activity

#### Introduction

In the present century research on nanotechnology gains much importance in the fields of biology, medicine, physics, chemistry and electronics. When coming to the biology nanoparticles are synthesized with the help of medicinal plants as reducing agents and is a fascinating research area by synthesizing different types of nanoparticles like, calcium, copper, gold, iron, silver and zinc (Kumar et al. 2016). Among these, copper nanoparticles have enormous considerations in the preparation of algaecides, fungicides, antifouling and antimicrobial agents. Thereafter which attract attention in various fields due to its catalytic, electric, optical, photonic and superconducting properties (Padil and Cernik 2013). The green synthesis method of CuO NPs with medicinal plants is eco-friendly, cost effective and stable. However, the research going on with this method is very seldom. Whereas the chemical- and physical-mediated methods were very familiar, but they arise some difficulties to humans as well as to the environment. In chemical-mediated synthesis, the copper salts were reduced with the help of reducing agents like, sodium borohydride, hydrazine, microemulsions and with evaporation/condensation methods. The use of sodium borohydride can irritate skin, eyes,



<sup>☐</sup> Pulicherla Yugandhar yugandharbotany@gmail.com

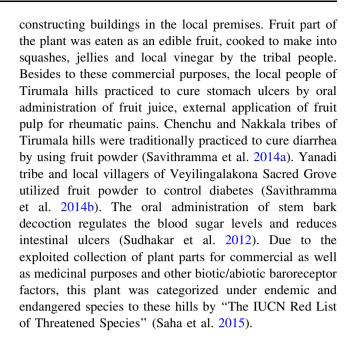
Department of Botany, Sri Venkateswara University, Tirupati, AP 517502, India

Department of Applied Microbiology, Sri Padmavati Mahila Visvavidyalayam, Tirupati, AP 517502, India

nose and throat while exposure. Higher exposures may cause pulmonary edema and affect the nervous system (Yugandhar et al. 2015). Hydrazine is a colorless, flammable inorganic compound, exposure to this chemical cause irritation of eyes, nose and throat. Higher exposures may lead to seizure, coma, kidney and liver damages. The disadvantage of microemulsion was its high expensive cost, which may adsorb on the surface of particles, separation and removal of them from final products was difficult, which ultimately reduces the usability of nanoparticles (Capek 2004). The evaporation/condensation method occupies large space, consumes more power and the obtained powders were in highly agglomerated state. These nano-powders were treated with high purity gas streams to cause compositional and structural modifications of nanoparticles, which arises hazardous waste materials to the environment (Tavakoli et al. 2007).

Coming to the physical-mediated familiar methods, the pulse laser ablation, microwave-assisted and pulsed/explosion wire discharge methods are in trend. The disadvantage of the laser ablation method was its high cost, high energy consumption and low product efficiency (Kalyanaraman et al. 1998). The microwave-assisted synthesis has radiation health hazards and the cost of microwave systems. The pulsed/explosion wire discharge method was not cost effective, limited production efficiency and contaminations of the final product were considered as limitations to this method (Umer et al. 2012). With these constraints, medicinal plants are employed for the alternative and future strategies to synthesize nanoparticles. These medicinal plants are acting as eco-friendly, safety, cost effective and green reducing sources for synthesis of copper nanoparticles. In recent past, the CuO NPs were synthesized from Acalypha indica (Sivaraj et al. 2014), Aloe vera (Vijay Kumar et al. 2015) and *Punica granatum* (Ghidana et al. 2016). These nanoparticles exhibit different biological activities like, anticancer (Sankar et al. 2014), anti-inflammatory, mosquito larvicidal (Angajala et al. 2014), wound healing (Tiwari et al. 2014), antimicrobial (Shende et al. 2015), anti diabetic and antioxidant activity (Ghosh et al. 2015). However, there is no report on green synthesis of CuO NPs from S. alternifolium and there is no validation of antimicrobial activity against microorganisms and anticancer activity against MDA-MB-231 breast cancer cell lines. Hence, the present work has been undertaken to synthesize CuO NPs and to study their synergistic antimicrobial and anticancer activities.

The plant *S. alternifolium* was grown on top hills of Tirumala forest, Chittoor District, Andhra Pradesh, India. It belongs to the family Myrtaceae and is locally known as mogi or adavinerudu. This is a medium sized perennial tree and has importance in both commercial and medicinal practices. The plant was exploited in a commercial way for



#### Materials and methods

# Collection of plant material

Fresh part of *S. alternifolium* stem bark was collected from Nagatheertham area of Tirumala hills, Chittoor District, Andhra Pradesh, India. Stem bark was washed thrice with running tap water followed by Milli Pak pure water. They may dry up to 15–20 days under shady conditions to evaporate residual moisture and finally ground with blender for further use.

#### Chemicals

Deionized Milli Q water (Merck Water Solutions, France), CuSO<sub>4</sub>·5H<sub>2</sub>O (Sisco Research Laboratories Pvt. Ltd., India), NaOH, 3-(4,5-dimethylthiazol-2-Yl)-2,5-diphenyltetrazolium bromide (MTT), Nutrient broth agar and Potato dextrose agar media (Sigma Aldrich, Bangalore, India), Fluconazole, Streptomycin and Doxorubicin (Thermo Fisher Scientifics, India) were mainly used in this work.

# Synthesis of nanoparticles

5 g of finely grounded plant powder was extracted with 100 ml of Milli Q water on boiling water bath for 30 min and were filtered with Whatmann no. 1 filter paper. An aliquot of 10 ml of aqueous plant extract was titrated with 100 ml of 5 mM  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  for reduction of CuO NPs at 50 °C for 2 h. Obtained mixture was centrifuged at 10,000 RPM for 15 min to separate agglomerated, broad sized particles as well as plant admixtures.



#### Characterization

The green synthesized nanoparticles were characterized by using different spectroscopic and microscopic tools. Initial confirmation of nanoparticles was done by UV-vis spectroscopy (Spectro UV 2080, Analytical Technologies, India) to know which of the metal nanoparticles were actually reduced and which of the phytochemicals were actually involved in reduction of nanoparticles by surface plasmon resonance method. To comprehend which of the phytochemicals were actually involved in capping and stabilization of nanoparticles, the FTIR spectroscopy (ALPHA interferometer, ECO-ATR, Bruker, Ettlingen, Karlsruhe, Germany) was utilized. Dynamic light scattering (DLS) and Zeta potential of synthesized nanoparticles were analyzed to know the average size and stability of particles (Nanopartica analyzer, Horiba SZ 100, Japan). XRD (Shimadzu, XRD-6000) was used to analyze crystalline nature and calculate the average size of particles. Microscopic analysis with TEM (HF-3300, 300 kV TEM/ STEM, Hitachi) instrument reveals the size, shape, dispersed nature and agglomerated pattern of nanoparticles. Along with these, the publicly accessed software tools like ImageJ and CrysTBox were used to know the average size and diffraction pattern of nanoparticles.

# Antimicrobial activity

The antimicrobial activity of green synthesized nanoparticles was carried out against seven selected bacterial strains include Bacillus subtilis ATCC 6633, Staphylococcus aureus ATCC 6538, Escherichia coli ATCC 25922, Klebsiella pneumonia ATCC 43816, Proteus vulgaris ATCC 13315, Pseudomonas aeruginosa ATCC 15442, Salmonella typhimurium ATCC 14028 and five fungal strains include Alternaria solani ATCC 32904, Aspergillus flavus ATCC 9643, Aspergillus niger ATCC 16404, Penicillium chrysogenum ATCC 11709 and Trichoderma harzianum ATCC 20476 by using disc diffusion assay method (Cruickshank 1986). To know the minimum inhibitory concentration of prepared nanoparticles, serial concentrations of 5, 10, 20, 40 and 80 µg/ml were tested. Finally, found that 20 µg/ml concentrations showed minimum growth inhibition and 80 µg/ml concentrations as almost lethal toxic to all the microorganisms. For this, 40 μg/ml concentrations were used as LD50 value for antimicrobial activity of prepared nanoaprticles. The plant extract (1 mg/ ml) and CuSO<sub>4</sub>·5H<sub>2</sub>O (5 mM) was used as positive and negative controls, respectively. Streptomycin and Fluconazole (10 µg/ml/disc) were used as standard controls for bacterial and fungal strains, respectively. An aliquot of 20 µl per disc was prepared for above-mentioned aliquots and an amount of 50 µl per petri plate microbial broth cultures were used for swabbing on culture medium. The triplicates of all these experiments were carried out under sterile conditions and were incubated at 37 °C for 24–48 h.

# **Anticancer activity**

MDA-MB-231 Human breast cancer cell lines were procured from National Centre for Cell Science, Pune, India. The 0.2 ml of Dulbecco's Modified Eagle's Medium was used to growing up  $1 \times 10^4$  cells per well in 96-well plate. These cell lines were incubated in 5% CO<sub>2</sub> atmosphere at 37 °C for 24 h supplemented with 2 mM/l glutamine, 10% Foetal Bovine Serum (FBS) with 10 µg/ml of ciprofloxacin (Jagadeesh et al. 2013). After that medium was expelled and refilled with 0.15 ml of 10, 25, 50 and 100 µg/ml concentrations of CuO NPs. The 0.1% of DMSO was prepared for dissolving sample nanoparticles as well as MTT dye crystals has been set as negative control and 1 µM doxorubicin treated cell lines were set as positive control. The initial experiment was maintained from 0 to 72 h of timeline period with 12 h of time gap period to check probability of cell toxicity. It provides specific time course period to allow functional cell mortality to understand the experiment in a flexible and adaptable way. According to that, less cell toxicity was observed at 12-, 24-, 36-h period and greater cell toxicity was observed at 60- and 72-h period. It reduces the chance of taking readings with ELISA reader. The 48 h of time period showed optimum reliability than other timeline periods. Due to this, the 48 h of incubation period was considered for nanoparticles cytotoxicity prediction analysis. Triplicates of experiments were carried out and incubated them up to 48 h at 37 °C. Further, at the end of incubation period the sample solutions were discarded and incubated for 4 h at 37 °C by adding 0.02 ml of MTT reagent (5 mg/ml) to each well. After that, MTT containing medium was discarded and refilled with 0.15 ml/well DMSO to dissolve formazan crystals. The viability of cell lines was read at 570 nm by an ELISA reader. The percentage of cell viability was calculated by the following formula (Kadirareddy et al. 2016):

Percentage of cell viability = 
$$\frac{\text{OD value of treated cell lines}}{\text{OD value of negative control}} \times 100.$$
 (1)

# Results and discussion

#### **UV-vis spectroscopy**

In green synthesis protocols, the matter of first importance is the visual characterization of the color change pattern upon synthesis of nanoparticles. Here, observed that stem



bark-mediated CuO NPs manifest a color change from brown to thick brown (Fig. 1). The color change mechanism of the reaction mixture is due to a reaction between copper(II) sulfate pentahydrate and sodium hydroxide to form copper(II) hydroxide (2), which reacts with the plant ascorbic acid to give dehydroascorbic acid and copper(I) oxide (3).

$$\begin{array}{cccc} CuSO_4 \cdot 5H_2O & + & 2NaOH & \rightarrow & Na_2SO_4 \\ & & + & Cu(OH)_2 + & 5H_2O, \end{array} \eqno(2)$$

$$2Cu(OH)_2 + \ C_6H_8O_6 \to \ Cu_2O \ + \ C_6H_6O_6 + \ 3H_2O.$$

When the reaction time was increased, the copper(I) oxide was formed to CuO nanoparticles by a series of reactions (4–6).

$$Cu_2O + OH^- + H_2O \rightarrow [Cu(OH)_2]^- + CuOH,$$
 (4)

$$2CuOH \rightarrow Cu_2O + H_2O, \tag{5}$$

$$4CuOH + O_2 \rightarrow 4CuO + 2H_2O.$$
 (6)

After the reduction of CuO material, they formed as spherical shaped CuO nanoparticles by the action of plant phytochemicals. Plant phytochemicals act as capping and stabilizing agents to give particular shape and to avoid agglomeration between the particles (Baoshun et al. 2016). When analyzed these samples with UV-vis spectrophotometer between the scan range 190-750 nm. Nanoparticles in the reaction mixture were excited by absorbing light at different wavelengths due to surface plasmon resonance (SPR) nature to give respective broad peak, represents which metal nanoparticles was actually reduced. Synthesized reaction mixture showed that a broad peak at 285 nm corresponds to copper oxide, which

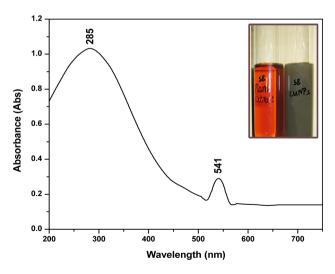


Fig. 1 UV-vis spectroscopic analysis of synthesized CuO NPs represents two peaks correspond to copper oxide and ascorbic acid. Inset photograph represents color change pattern upon synthesis



confirms the synthesized nanoparticles were CuO NPs. Another small peak obtained at 541 nm confirmed as ascorbic acid, acts as reducing agent for CuO NPs (Fig. 1). Similar type of result was observed from copper nanoparticles synthesized with the help of *Punica granatum* peel extract (Ghidana et al. 2016).

## FT-IR analysis

(3)

FT-IR study of biologically synthesized nanoparticles was analyzed within the scan range of 4000–500 cm<sup>-1</sup> to know the feasible phytochemicals responsible for capping and stabilization. Here, synthesized nanoparticles exhibit broad peaks at 3313 cm<sup>-1</sup> corresponds to O–H bond of phenols and 1640 cm<sup>-1</sup> corresponds to C=O stretching vibration of primary amines. Thus, consequences confirmed that phenols and primary amines of plant extracts were mainly responsible for capping and stabilization of CuO NPs (Fig. 2). These kind of results was obtained from copper nanoparticles synthesized with the help of *Commelina nudiflora* whole plant extract (Kuppusamy et al. 2017).

# Zeta potential and particle size determination

Synthesized nanoparticles were dispersed in 10 ml of distilled water to know the particle size with DLS by surpassing laser scattering light and set instrumental conditions like, 0.892 mPa s of dispersion medium viscosity, 173° of scattering angle, 25.2 °C of temperature. Whereas to know the zeta potential by adding few  $\mu$ l of dispersed nanoparticle solution to electrode and set the conditions like, 0.264 mS/cm of conductivity, 3.3 V of electrode voltage, 25.2 °C of temperature and the data acquisition time was set to 1 min. Synthesized

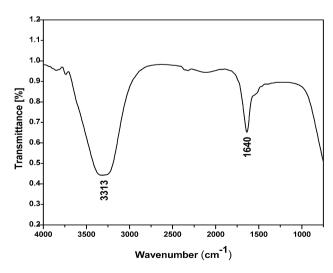


Fig. 2 FT-IR analysis of synthesized CuO NPs represents two broad peaks correspond to phenols and primary amines

nanoparticles exhibited polydispersed type of particles with 2.6 nm average size and -63.9 mV of zeta potential value (Fig. 3). These outcomes indicated that the plant *S. alternifolium* was a good source for reduction of narrow size of CuO NPs. The higher negative zeta potential value confirmed repulsion between particles to thwart agglomeration. This indicates synthesized nanoparticles have higher stability in a liquid medium.

# XRD analysis

XRD analysis of synthesized nanoparticles showed 08 Bragg reflections at  $2\theta$  of X-axis at  $32.4^{\circ}$ ,  $35.6^{\circ}$ ,  $38.8^{\circ}$ ,  $48.9^{\circ}$ ,  $53.3^{\circ}$ ,  $58.2^{\circ}$ ,  $61.6^{\circ}$ ,  $66.3^{\circ}$  corresponds to 110, 111, 200, -202, 020, 202, -113 and 022 integer 'hkl' planes, respectively (Fig. 4). This indicates end-centered monoclinic crystalline nature of nanoparticles. This result was affirmed by cross checking the obtained data with International centre for diffraction data (ICDD) file no. 80-1916. This type of results was obtained from an A. vera leaf extract mediated CuO NPs (Vijay Kumar et al. 2015). Highest Bragg reflection was obtained at  $2\theta$  of  $38.8^{\circ}$  to predict full width half maximum (FWHM) value, i.e., 5 and was calculated to 17.2 nm average size of the particle by using the following Debye–Scherrer equation:

$$D = \frac{k\lambda}{\beta\cos\theta}.\tag{7}$$

# TEM analysis

An ample resolution up to 2 nm is possible by high resolution transmission electron microscope (HR-TEM) analysis. Because, focusing of 300 kV of electron beam energy

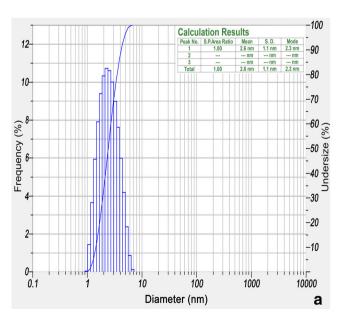


Fig. 3 a DLS, b Zeta potential analysis of synthesized CuO NPs

is possible on nanoparticles. The 20 nm resolution studies with TEM analysis revealed that the particles were spherical in shape, non-agglomerated, polydispersed with the size range from 5 to 13 nm (Fig. 5b). The 2 nm resolution studies of nanoparticles showed 0.28 nm size of 'd' space. This result indicates the synthesized nanoparticles were crystalline in nature (Fig. 5c).

# Particle analysis

The synthesized nanoparticles were analyzed by different image processing software tools like ImageJ and CrysT-Box. ImageJ (image processing and analysis in Java) is one of the public domain Java empowered programs widely

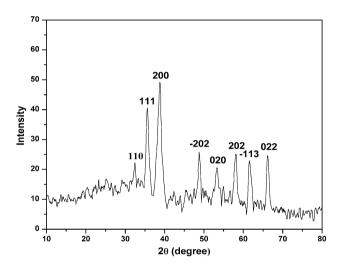
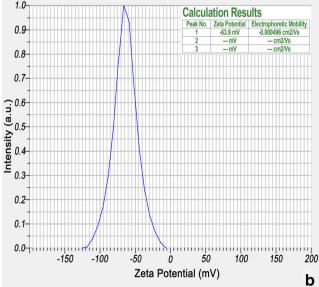


Fig. 4 XRD analysis of synthesized CuO NPs represents eight Bragg reflections corresponding to end-centered monoclinic crystalline nature of nanoparticles





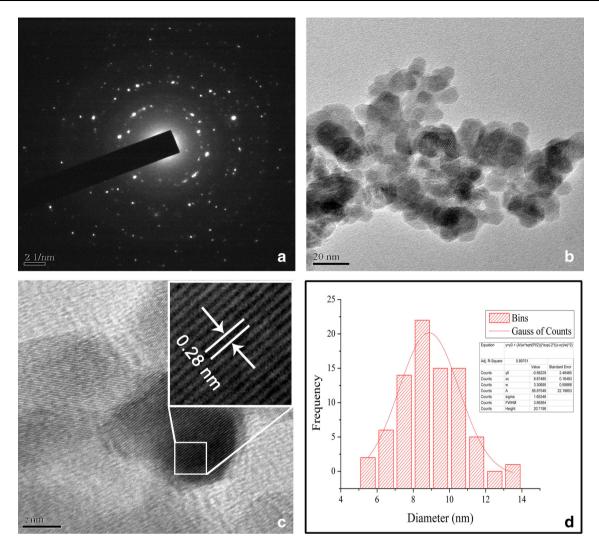


Fig. 5 a Selected area electron diffraction analysis of CuO NPs. b 20 nm resolution studies of synthesized CuO NPs. c Calculation of 'd' space with 2 nm resolution micrographs. d Particle size analysis with Gaussian fitting formula represents average size of the particles

used in the analysis of microscopic images. By this tool, synthesized nanoparticles were analyzed to know the size distribution of particles as well as the average size. Micrographs from TEM analysis showed 8.8 nm average size of the particles with 0.89 adjacent 'R' square value calculated by using the following Gauss fitting formula (Fig. 5d):

$$(y = y0 + (A/(w \times \text{sqrt}(PI/2))) \times \exp(-2 \times ((x - xc)/w)^2).$$
(8)

Crystallographic toolbox (CrysTBox) is open source software, to implicate set of three automated tools to analyze crystallography of TEM-SAED micrographs. Regarding to diffractGUI and ringGUI tools are used to detect electron diffraction patterns of TEM-SAED micrographs. Another program cellViewer is used for the visualization of diffracted planes in 3D pattern. TEM-SAED image of synthesized CuO NPs from stem bark

was imported into diffractGUI tool. That automatically determines the zone of axis from a diffraction pattern to assign crystallographic indices to the diffracted atoms to measure inter planner angle of two atoms (Fig. 6a). The ringGUI tool automatically determines crystallographic planes of synthesized material (Fig. 6b). The cellViewer, a visualization tool for depicting diffraction planes from diffractGUI displays, direct automatic lattice (cell view) (Fig. 6c) and reciprocal lattice (diffraction view) of the synthesized nanomaterials (Fig. 6d) (Klinger and Jager 2015). TEM-SAED pattern of synthesized nanoparticles was analyzed with this software displayed the planes of 'hkl' integers were exactly matched with Bragg reflections of XRD results and JCPDS card file no. 80-1916. CrysTBox software is astounding tool to correlate the characteristic crystallization pattern of XRD results with TEM-SAED pattern of synthesized CuO NPs.



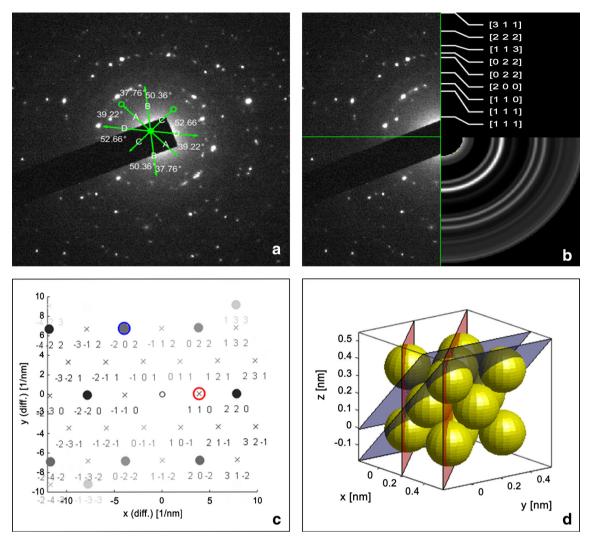


Fig. 6 a diffractGUI analysis of TEM-SAED pattern of CuO NPs. b ringGUI analysis of CuO NPs shows crystallographic planes. c Visualization of lattice planes in cell view. d Visualization of lattice planes in 3D diffraction view

# Antimicrobial activity

Antimicrobial study of green synthesized CuO NPs has been shown noteworthy growth inhibitory results on different microorganisms. A highest zone of inhibition was observed in E. coli followed by S. aureus, these are gram negative and gram positive bacterial strains, respectively. Same type of highest growth inhibitory result was reported in E. coli by Meghana et al. (2015). Whereas in fungi the highest zone of inhibition was observed in T. harzianum (Figs. 7, 8). In this study, the zone of inhibition was observed less in fungi when compared with bacteria. Among the bacteria, a zone of inhibition was lesser in gram positive bacteria when compared with gram negative bacteria. This may arise due to fungal cell walls are made up of chitin, a fibrous substance comprising of polysaccharides having N-acetylglucosamine and a nitrogen group, which is more firm to allow the passage of nanoparticles from the outer layer of the cell wall to the inner layer. In case of bacteria, cell membranes are made up of peptidoglycan a polymer having sugars and amino acids, which is less firm and passage of nanoparticles is easy when compared with fungi. Among the bacterial strains passage of nanoparticle through the cell membrane is easy in the case of gram negative bacteria when compared to gram positive bacteria due to variation in manufacturing of cell membrane (Yugandhar and Savithramma 2016a). Ren et al. (2009) elucidated that copper nanoparticles acts as inhibitory substances on microorganisms by the action of destroying enzymes of cell membranes. This might be possible by the small size and spherical shape of the particles have a high surface to volume ratio to interact with the cell membranes of microorganisms to exhibit the signs of growth inhibitory results (Agnihotri et al. 2014). Size, shape and agglomeration pattern of the nanoparticles depend on quantity-based presence of phytochemicals in medicinal plants. This might



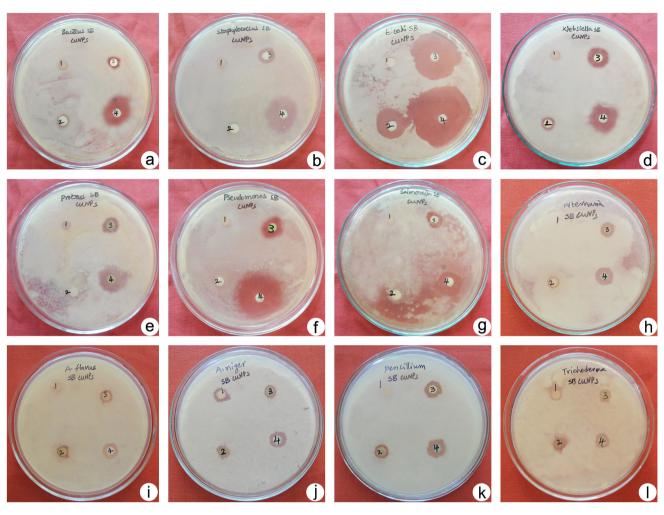
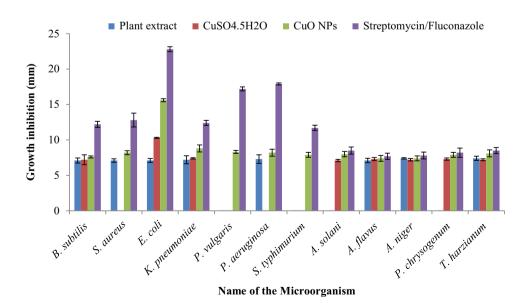


Fig. 7 Antimicrobial activity of synthesized CuO NPs. a B. subtilis, b S. aureus, c E. coli, d K. pneumoniae, e P. vulgaris, f P. aeruginosa, g S. typhimurium, h A. solani, i A. flavus, j A. niger, k P.

chrysogenum, 1 T. harzianum, 1 plant extract, 2 CuSO<sub>4</sub>·5H<sub>2</sub>O, 3 CuO NPs, 4 Streptomycin/Fluconazole

**Fig. 8** Antimicrobial activity of bioinspired green synthesis of CuO NPs from stem bark extract of *S. alternifolium* 





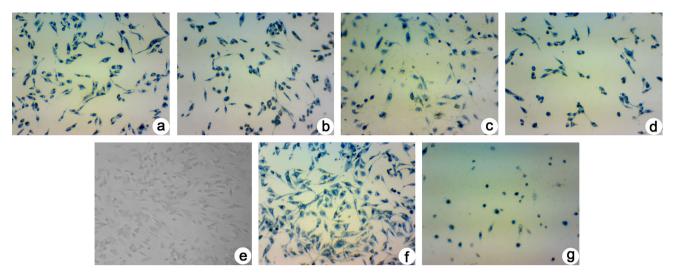


Fig. 9 Anticancer activity of synthesized CuO NPs. a 10 μg/ml, b 25 μg/ml, c 50 μg/ml, d 100 μg/ml, e untreated healthy cell lines, f DMSO (0.1%) negative control, g doxorubicin (1 μM) positive control

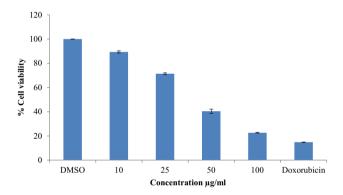


Fig. 10 Anticancer activity of bioinspired green synthesis of CuO NPs from stem bark extract of *S. alternifolium* 

be useful in reducing, capping and stabilization of nanoparticles to narrow size with spherical shape (Saif et al. 2016). In this study it is found that CuO NPs synthesized from stem bark exhibit narrow size particles with spherical shape. This is because of plant *S. alternifolium* is a good source for different phytochemicals especially phenols and proteins (Yugandhar and Savithramma 2016b). These are possibly helpful in this study for capping and stabilization of nanoparticles. A similar kind of result was obtained from copper nanoparticles synthesized from *A. indica* leaf mediated particles, resulted with spherical shape with size range from 26 to 30 nm to exhibit excellent antimicrobial activity against different bacterial and fungal pathogens (Sivaraj et al. 2014).

# Anticancer activity

The MDA-MB-231 human breast cancer cell lines were used for cytotoxicity analysis by reading formazan crystals formed by the reaction of mitochondrial dehydrogenase by MTT assay. At 48 h of time course incubation period, a significant abatement in cell viability was observed in the treated cell lines, while the concentration of CuO NPs was increased from 10, 25, 50 and 100 μg/ml. DMSO was used as a positive control to exhibit 100% of healthy proliferated cells (Figs. 9, 10). The 50  $\mu$ g/ml concentration (IC<sub>50</sub>) of CuO NPs may have the capability to reduce 50% of treated cell lines when compared with negative control. The cytotoxicity of nanoparticles may depend on the small size and spherical shape of the particles (Park et al.2011). In recent times, the A. indica leafmediated CuO NPs have the particle size ranging from 26 to 30 nm, which showed strong cytotoxic activity against MCF-7 breast cancer cell lines (Sivaraj et al. 2014). But, there is no report on CuO NPs synthesized from any medicinal plant to attribute anticancer activity against MDA-MB-231 breast cancer cell lines. From this study, the green synthesized CuO NPs from stem bark extract showed 5-13 nm size, spherical shaped particles, which exhibit strong cytotoxic activity against MDA-MB-231 cell lines.

## Conclusion

The present study, reported a cost effective, eco-friendly, green approach method for production of CuO NPs from stem bark extract of *S. alternifolium*. Due to constraints with chemical- and physical-mediated methods, the green approach method is best opted technique to synthesize CuO NPs. The analytical tools like UV–vis spectroscopy and FT-IR studies revealed that the reduction of nanoparticle by ascorbic acid and the presence of phytochemicals like phenols and primary amines of proteins become helpful in capping and stabilization of particles. DLS and Zeta potential analysis revealed 2.6 nm average size and



-63.9 mV of higher negative zeta potential value indicates greater stability of particles. The crystalline structure and 17.2 nm average size of the particles was revealed by XRD analysis. Microscopic analysis with TEM instrument showed spherical shaped particles with a size range from 5-13 nm. These particles were mostly settled in non-agglomerated and poly-dispersed condition. The software tools like ImageJ and CrystBox supports the calculation of average size and confirms the crystallographic nature of particles. The expository synergistic efficiency of CuO NPs showed growth inhibitory activity of microorganisms as well as MDA-MB-231 breast cancer cell lines. The presence of higher concentration of phenols and proteins in stem bark may be the reason behind for formation of narrow sized particles, bestowed to antimicrobial and anticancer activity. This study concludes that the plant S. alternifolium is an excellent green source for synthesis of copper oxide nanoparticles. This green approach method is very helpful in terms of endemic, endangered medicinal plant i.e., S. alternifolium. Here, the usage of plant material for synthesis is very limited and to produce greater quantities of nanoparticles in a non destructive way. This is the first report on bioinspired green synthesis of CuO NPs from the stem bark extract of S. alternifolium and too proved these nanoparticles as synergistic antimicrobial and anticancer agents. This study may pave a way for the futuristic synthesis of environmentally benign, cost effective CuO NPs from medicinal plants.

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