ORIGINAL ARTICLE



Assessment of groundwater quality in Khulna city of Bangladesh in terms of water quality index for drinking purpose

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Received: 12 May 2020 / Accepted: 2 October 2020 / Published online: 13 October 2020 © The Author(s) 2020

Abstract

Valuation of water quality index (WQI) is one of the simplest, easily understandable, and efficacious techniques to evaluate the quality and suitability of water for drinking as well as other purposes. This research was aimed to investigate the drinking water quality of tube wells from different areas in Khulna City, Bangladesh, by developing the WQI. Water samples from 59 tube wells were collected from different locations during the pre-monsoon time. pH, electric conductivity (EC), dissolve oxygen (DO), total dissolved solid (TDS), chloride (Cl⁻), nitrate (NO₃⁻), and total hardness of the collected water samples were analyzed for the calculation of WQI. The mean value for pH, EC, DO, TDS, Cl⁻¹, NO₃⁻, and total hardness was 7.30, 1650 μS/cm, 1.60 mg/l, 1188.7 mg/l, 414.6 mg/l, 0.029 mg/l, and 52.03 mg/l, respectively. The calculated WQI values for individual places were distributed spatially through mapping by using ArcGIS software. Based on the WQI values, the drinking water was categorized into excellent, good, poor, very poor, and unfit for drinking purposes. The calculated WQI values ranged from 40.11 to 454.37 with an average value of 108.94. Among all the groundwater samples, 11.86% were excellent, 54.24% were good, 23.73% were poor, 1.69% were very poor, and 8.47% were unfit for drinking purpose based on WQI. The results showed that the groundwater quality of most of the studied areas of Khulna city could be considered safe and suitable for drinking barring the elevated EC and chloride content in some areas. Since Khulna city is situated in the southwestern part of Bangladesh and gradually approaches toward the base level of the Bay of Bengal which might be the source of salt concentration in the groundwater of Khulna city, Bangladesh.

Keywords Water quality index · Drinking water · Tube well · Salinity · Coastal area · GIS

Introduction

From the dawn of earth to the present, water is one of the essential factors that contribute to sustaining every form of life. Due to the growing demand for water worldwide, the assurance of adequate supply of drinking water by maintaining standard water quality has become a significant challenge (Mekonnen and Hoekstra 2016; Sawyer et al. 2003). Among the pure sources of water, groundwater (GW) is the most reliable and most extensive storage of freshwater on

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Soil, Water and Environment Discipline, Khulna University, Khulna 9208, Bangladesh earth which is used by one-third of the entire world's population for drinking purposes (Foster and Chilton 2003; Nickson et al. 2005). Despite the significance of GW, the quality of this natural resource is not maintained correctly. Various pollution sources, climate changes, GW recharge, subsurface geochemical reactions, surface water characteristics, geographical locations, atmospheric precipitation, and anthropogenic activities are the most critical factors that affect the quality and quantity of GW directly and indirectly (Collin and Melloul 2003; Kumar et al. 2014; Ramakrishnaiah et al. 2009). Furthermore, the intrusion of saline water into the coastal area is one of the leading causes responsible for the deterioration of GW quality (Karro et al. 2004; Kim et al. 2006).

Salinity in GW has become a significant problem in the south-west coastal region of Bangladesh, including Khulna City, located on the bank of *Bhairab* and *Rupsa* river. This city is gradually inclined to the base level of the Bay of Bengal (Hoque et al. 2003; Woobaidullah et al. 2006).



Here, some areas are facing enormous challenges to meet the increasing demand for freshwater due to the intrusion of excess saline water which impedes the supply of fresh drinking water and has become a very alarming issue for the city dwellers. The excess extraction of GW from the aquifer for drinking and irrigation purposes decreases the freshwater supply by lowering the GW table and eventually sea water enters into the GW zone to fill up the gap. Excess salt concentration also changes other water quality parameters and alters the natural composition of water (Abdalla et al. 2010; Adhikary et al. 2012; Van Camp et al. 2014). Side by side, continuous use of polluted water results in a detrimental effect on human health and socio-economic development (Milovanovic 2007). So, to control the use of polluted water especially for drinking purposes, regular monitoring of the GW quality is a very compatible step in recent times (Hasan et al. 2019; Saeedi et al. 2010).

For evaluating the water quality, estimation of WQI is one of the most suitable techniques which was first suggested and developed by Horton (1965). Usually, WQI is calculated to assess the water quality by considering its suitability for drinking purposes mainly. So, it has become a handy tool for the management of GW quality and ensures the utilization of pure drinking water (Ramakrishnaiah et al. 2009; Tiwari et al. 2014). This procedure helps to express the water quality stably and simply because this tool is very efficient to transform large quantities of complex data into a single number. Thus, the values derived from WQI can be considered as much more convenient and easily understandable to the researchers, general audiences, concerned citizens as well as national water policymakers for management and decision-making purposes. Though there are different ways of calculating WQI each system considers similar physical and chemical parameters of water and the only differences among the procedures are: the way of integration of the data and interpretation of the obtained results (Reza and Singh 2010; Sinha et al. 2004; Stambukgiljanovic 1999). The computed data were displayed by using a GIS mapping system, which is a very convenient and powerful tool that provides a guideline to understand, interpret and visualize the data easily through modeling or graphical presentation systems (Gupta and Srivastava 2010; Krishnaraj et al. 2015).

The present research work was aimed to investigate the quality and to justify the suitability of tube well water of different areas in Khulna City, Bangladesh for drinking purpose by calculating WQI; and to project the status of physicochemical parameters and WQI by preparing a proper map through GIS.

Materials and methods

Study area

The experiment was carried out in Khulna City, the third biggest city in Bangladesh. The city is situated in the southwestern part of the country, covering an area of approximately 59.57 square kilometres (Wikipedia 2020). The latitude and longitude of the city are 22° 46′ to 22° 58′ North and 89° 28′ to 89° 37′ East, respectively. The city is located on a riverine area where the tidal river *Rupsa* crosses the middle part of the city, the river *Bhairab* is on the northern, the *Pasur* river is on the southern side, and the *Mayur* river is located in the western area of the city (Adhikary et al. 2012; Roy et al. 2005).

According to Khulna Water Supply and Sewerage Authority (KWASA), the number of total tube wells in Khulna city was approximately 55 in 1996, then increased to 111 in 2011, and now the number is more (IWM 2011). In this experiment, a total of 59 water samples was collected from 59 different locations (Fig. 1). The sampling sites were predetermined and selected in such a way that the main populated areas of Khulna City were covered irrespective of the depth of the tube wells. Moreover, to ensure the permanent drinking water reservoir site, tube wells installed in schools, colleges, and religious places were selected as groundwater sources. The longitude and latitude values of the sample sites (Table 1) also denoted to specify the locations.

Water sampling

Water samples were collected from selective tube wells with intensive care. After continuous pumping of the tube well for 5 min, water samples were collected in new plastic bottles after washing the bottle properly to avoid any unwanted contamination of water. The bottles were filled up entirely, and the caps were closed very carefully to create the airtight condition. No air bubble was seen inside the bottle which ensured the absence of any trapped air in the water, and that helped to prevent the oxidation of the reduced substances of water samples during the time of transportation and storage. After collection, samples were immediately stored in a closed icebox. Then, the samples were brought to the laboratory for chemical analysis, and different parameters were analyzed as soon as possible to avoid different chemical and biological reactions that may take place over time and change the water quality. Side by side, for some timeconsuming, complex analysis, and samples were stored in a refrigerator.



Study Area

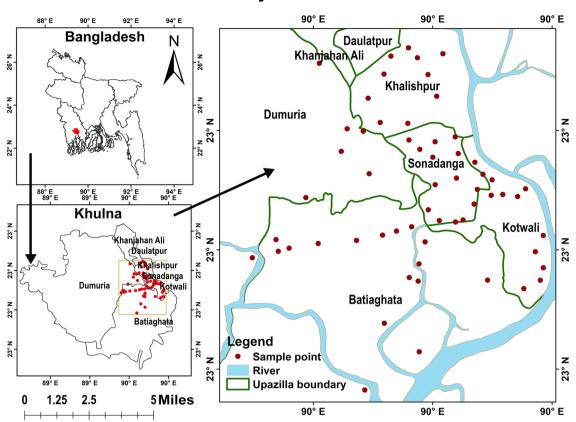


Fig. 1 The sampling locations of the study area

Chemical analysis of water samples

pH, EC, and TDS were measured by using a portable multimeter (Hanna HI9813-5 Portable pH/EC/TDS Meter). Before taking the reading, the meter was calibrated by using a calibration solution for a particular analysis. DO was determined by using DO Meter (HI98193). A zero oxygen solution (HI7040) was used to calibrate the probe of the device. The chloride of the water sample was determined by the titrimetric method (Mohr's Argentometric method, APHA 4500 Cl-B) where potassium was used to determine the concentration of chloride ions in solution (Mohr 1856). Total hardness was determined by estimating the Ca and Mg content of water samples through complexometric titration with EDTA, where Eriochrome Black T was used as the indicator of the visual endpoint (Harris 2010). Nitrate was determined by following the prescribed method of Cataldo et al. (1975).

The procedure of WQI estimation

WQI was calculated by following four steps. In step 1, according to the relative importance of each parameter

(pH, EC, TDS, DO, chloride, nitrate, and total hardness) in the quality measurement criteria of water for drinking purpose, a particular weight (*wi*) was assigned for each parameter (Table 2). EC, TDS, chloride, and nitrate were considered as the key elements, and the maximum weight (5.0) was set for each of those (Vasanthavigar et al. 2010). pH and DO were categorized under the weight of 4.0 according to its value on water quality determination. As total hardness (calcium and magnesium) is considered less important than the other parameters in measuring drinking water quality, minimum weight (2.0) was assigned for it (Batabyal and Chakraborty 2015).

In step 2, the relative weight (*Wi*) was calculated by using the following equation (eq):

$$Wi = wi / \sum_{i=1}^{n} wi \tag{1}$$

Here Wi the relative weight; wi the weight of each parameter; and n the total number of parameters.

Next, for each parameter, a quality rating scale (qi) was computed in step 3, which was obtained by dividing the



Table 1 Name of sampling sites and location's coordinate

Sl. no.	Sampling code	Sampling site name	Latitude (°)	Longitude (°)
1	GW1	Hope Polytechnic and Technical Institute Hostel	22.8056	89.5401
2	GW2	Sobujbagh Jame Masjid Complex	22.8144	89.5425
3	GW3	UCEP School, Sonadanga	22.8213	89.5404
4	GW4	Jamia Arbaria Fazlul Ulum Madrasha	22.8267	89.5372
5	GW5	Choto Boyra Kalibari Mandir	22.8300	89.5334
6	GW6	Khulna Public College	22.8358	89.5330
7	GW7	Home Economics College	22.8502	89.5283
8	GW8	Border Guard Public School, Khulna	22.8593	89.5271
9	GW9	South Kashimpur Mosque	22.8622	89.5332
10	GW10	Bongobashi School	22.8592	89.5375
11	GW11	Haji Shariatullah Bidyapith	22.8586	89.5369
12	GW12	Baitul Mamur Jame Mosque	22.8528	89.5376
13	GW13	Haji Muhsin College	22.8528	89.5401
14	GW14	Khulna Polytechnic Institute	22.8453	89.5431
15	GW15	Joragate Moslemia Mosque	22.8311	89.5496
16	GW16	New Market Kacha Bazar Mosque	22.8253	89.5506
17	GW17	PDB School	22.8167	89.5500
18	GW18	Khulna Government Girls' High School	22.8294	89.5425
19	GW19	Lions School and College	22.8019	89.5440
20	GW20	SOS Hermann Mainer School	22.8014	89.5496
21	GW21	Khulna Market Mosque	22.8021	89.5523
22	GW22	Baitul Muajjam Amtala Jame Masjid	22.8077	89.5561
23	GW23	Baitul Aman Jame Masjid	22.8128	89.5572
24	GW24	Jinnah Mosque	22.8180	89.5592
25	GW25	Dakbangla Jaame Masjid	22.8161	89.5624
26	GW26	B.K. Union Institution	22.8110	89.5621
27	GW27	Al Hera Jame Mosque	22.8112	89.5615
28	GW28	Govt. Coronation Secondary Girls' School	22.8110	89.5661
29	GW29	Khulna Zila School	22.8130	89.5741
30	GW30	Soburonnesha Mohila Shorkari Biddaloy	22.8102	89.5713
31	GW31	Rupsha Shorkari Biddaloy	22.7964	89.5805
32	GW32	Motia Khali Jame Mosque	22.7952	89.5805
33	GW33	Shipyard Govt. Primary School	22.7910	89.5775
34	GW34	Al-Amin Jame Mosque, Labonchora	22.7810	89.5792
35	GW35	Bokharia Jame Mosque	22.7781	89.5735
36	GW36	SSR School	22.7810	89.5608
37	GW37	Baitul Mokarrom Mosque	22.7820	89.5335
38	GW38	Khulna Zero Point Panjegana Masjid	22.7982	89.5290
39	GW39	Bangladesh Open University, Khulna Campus	22.7968	89.5241
40	GW40	Bismillah Nagar Madrasha	22.8182	89.5195
41	GW41	Baytul Shorif Jame Mosque	22.8339	89.5118
42	GW42	Baipas Jame Mosque	22.8568	89.5021
43	GW43	Hamid Nagar Haji Muhsin Prathomik Biddaloy	22.8331	89.5174
44	GW44	Rayer Mohol Madrasha and Etim Khana	22.8361	89.5233
45	GW45	Baytul Atik Jame Mosque and Madrasha	22.8260	89.5097
46	GW46	Samsul Rohman Mosque Complex	22.8099	89.4974
47	GW47	Baytul Hamid Jame Mosque	22.7952	89.4870
48	GW48	Islmabad Shahi Jame Mosque	22.7896	89.4815
49	GW49	Sahid Abul Kashem Degree College	22.7911	89.4878
50	GW50	Baytul Mamur Jame Mosque	22.7922	89.4916
51	GW51	Progoty Maddhomik Biddyaloy	22.7938	89.5017



Table 1 (continued)

Sl. no.	Sampling code	Sampling site name	Latitude (°)	Longitude (°)	
52	GW52	Madrasha Al Muhammad As Salafi	22.7949	89.5151	
53	GW53	Mouhammad Nagar Baytul Aman Jame Mosque	22.7985 89.5357		
54	GW54	Baytul Meraj Jame Mosque	22.7889	89.5371	
55	GW55	Mohammod Nagar Darus Salam Trust Mosque	22.7867	89.5369	
56	GW56	Al Modina Jobbariya Jame Mosque	22.7807	89.5366	
57	GW57	Baytul Mamur Jame Mosque Cokrakhali	22.7638	89.5272	
58	GW58	Baytul Korim Kendriyo Jame Mosque	22.7427	89.5180	
59	GW59	Haji Monoyara Jahangir Hafejiya Madrasha	22.7560	89.5370	

Table 2 Relative weight of chemical parameters (BIS 1991)

Chemical parameter	Standards (BIS)	Weight (wi)	Relative weight (Wi)
EC (μS/cm)	1000	5	0.1667
TDS (µg/l)	500	5	0.1667
Chloride (mg/l)	250	5	0.1667
Nitrate (mg/l)	45	5	0.1667
pН	8.5	4	0.1333
DO (mg/l)	13	4	0.1333
Total hardness (mg/l)	300	2	0.0667
		$\sum wi = 30$	$\sum Wi = 1$

Table 3 Classification of WQI range and category of water (Tiwari et al. 2014)

WQI Range	Category of water
< 50	Excellent water
50-100	Good water
100-200	Poor water
200-300	Very Poor water
> 300	Unfit for drinking purpose

concentration of each water sample by the respective standard of particular parameter mentioned in the guideline of Bureau of Indian Standards value, BIS 10500 (1991) and then the acquired result was multiplied by 100 (Batabyal and Chakraborty 2015; BIS 1991). So, the equation of *qi*:

$$q_i = \left(\frac{C_i}{S_i}\right) \times 100\tag{2}$$

Here *qi* quality rating of each parameter; *Ci* concentration (mg/l) of each chemical parameter in water sample; and *Si* Bureau of Indian Standards value for each chemical parameter in mg/l.

In the last step (step-4), at first *SI* for each chemical parameter was determined, and then the value the *SI* was used to calculate WOI by using the following equation:

$$SI_i = W_i \times q_i \tag{3}$$

$$WQI = \sum SI_{i-n}$$
 (4)

where SIi sub-index of the ith parameter; qi quality rating of the ith parameter; and n the number of parameters.

Classification of WQI range

The obtained WQI values were categorized under five categories. The ranges of WQI values and water categories are mentioned in Table 3.

Preparation of map

Several maps were prepared to delineate the spatial distribution of physicochemical parameters and WQI of groundwater quality by using GIS ArcGIS-version 10.5 software.

Results and discussion

General information and physicochemical data

The depth of tube wells and estimated values of the necessary parameters (pH, DO, EC, TDS, chloride, nitrate, and total hardness) to determine the WQI are given in Table 4. The pH value is one of the most vital indices that denotes the acidity or alkalinity and hydrogen ion concentration of groundwater (Murugesan et al. 2006). In our experiment, the pH value of the groundwater samples ranged from 6.40 to 7.90 with a mean value of 7.30 (Table 4) which indicates that water of the experimental areas was in optimal condition for drinking purpose as the values can be categorized within the permissible range (6.5–8.5) of drinking water according to WHO (2011). Akter et al. (2016) found the mean pH value



Table 4 Chemical characteristics of groundwater and WQI values

Sl.no.	Sample code	Depth (m)	pН	EC (μS/cm)	DO (mg/l)	TDS (mg/l)	Chloride (mg/l)	Total hardness (mg/l)	Nitrate (mg/l)	WQI	Comment
1	GW1	1000	7.7	690	1.89	457	106.5	35.68	0.0286	48.66	Excellent
2	GW2	800	7.5	790	1.51	499	71.00	24.8	0.0285	48.41	Excellent
3	GW3	1000	7.2	1630	1.38	1039	390.5	51.2	0.0285	101.70	Poor
4	GW4	1000	7.2	1480	1.24	923	319.5	48.0	0.0285	90.38	Good
5	GW5	800	7.6	1070	0.92	688	106.5	32.0	0.0290	61.46	Good
6	GW6	950	7.5	1540	2.04	950	426	36.8	0.0286	100.43	Poor
7	GW7	900	7.7	1130	1.76	704	142	32.8	0.0288	66.39	Good
8	GW8	150	7.1	2950	2.53	1958	745.5	62.4	0.0285	179.28	Poor
9	GW9	900	7.2	2680	1.35	1785	958.5	87.6	0.0285	182.72	Poor
10	GW10	950	7.2	3160	1.71	300	887.5	138.4	0.0285	137.97	Poor
11	GW11	960	7.5	2220	1.31	1433	532.5	84.0	0.0285	135.26	Poor
12	GW12	200	6.9	3690	1.50	3300	745.5	83.2	0.0287	235.44	Very poor
13	GW13	950	7.5	1760	1.05	1130	426	69.6	0.0288	109.81	Poor
14	GW14	1200	7.4	1720	1.55	1097	426	41.2	0.0285	107.76	Poor
15	GW15	150	7.1	5210	4.21	4740	1668.5	116	0.0287	374.14	Unfit
16	GW16	900	6.8	960	1.49	643	177.5	32.8	0.0285		Good
17	GW17	1000	7.4	950	1.44	610	142	31.2	0.0285		Good
18	GW18	800	7.5	1700	1.59	1080	426	44.0	0.0286	107.12	
19	GW19	1000	7.9	100	2.19	522	213	22.4	0.0289		Excellent
20	GW20	800	7.7	790	1.45	506	142	18.8	0.0285		Good
21	GW21	900	7.7	720	1.46	479	142	17.6	0.0286	51.41	Good
22	GW22	950	7.4	970	1.66	620	177.5	38.8	0.0287		Good
23	GW23	1000	7.5	1460	1.47	924	355	48.4	0.0290		Good
24	GW24	1000	7.3	1400	1.44	944	355	46.8	0.0287		Good
25	GW25	900	7.4	1640	2.81	1095	390.5	45.6	0.0288	105.39	
26	GW26	900	7.3	250	2.15	783	248.5	44.8	0.0294		Good
27	GW27	950	6.8	1000	1.68	632	142	33.2	0.0286		Good
28	GW28	1200	7.3	1150	1.41	716	355	29.2	0.0291		Good
29	GW29	1000	7.1	1450	1.56	959	355	54.4	0.0291		Good
30	GW29 GW30	950	7.0	4030	1.32	4200	1313.5	124	0.0251	309.86	
31	GW30 GW31	850	7.4	800	1.32	507	106.5	25.6	0.0285		Good
32	GW31 GW32	950		970	1.14	612	142	20.0	0.0283		Good
33	GW32 GW33	900		900	1.14	561	177.5	21.6	0.0341		Good
34	GW33 GW34	600		930	1.17	587	177.5	27.6	0.0283		Good
35	GW35	700	7.3	1000	1.54	633	177.5	36.4	0.0287		Good
36	GW36	1000	6.4	850	1.72	544	142	24.4	0.0285		Good
37	GW37	950	7.5		1.24	553	142	21.2	0.0285		Good
38	GW38	400		2190	0.59	1428	603.5	74.0	0.0285	138.21	
39 40	GW39	750		2350	1.02	1635	710	106	0.0285	155.09	
40	GW40	1260		1850	1.02	1232	319.5	21.2	0.0285	106.97	
41	GW41	150		6830	2.50	4170	2769	164.8	0.0285	454.37	
42	GW42	150	7.0	1630	1.49	1058	284	101.6	0.0285		Good
43	GW43	950			1.54	778	177.5	52.8	0.0285		Good
44 • -	GW44	950			1.28	651	106.5	24.8	0.0288		Good
45	GW45	900		690	1.99	459	106.5	55.2	0.0285		Excellent
46	GW46	450		1150	1.56	724	142	38.8	0.0285		Good
47	GW47	400		1120	1.93	678	213	71.6	0.0285		Good
48	GW48	450	6.9	1160	1.6	7.24	284	62.4	0.0286	52.37	Good



Sl.no.	Sample code	Depth (m)	pН	EC (μS/cm)	DO (mg/l)	TDS (mg/l)	Chloride (mg/l)	Total hardness (mg/l)	Nitrate (mg/l)	WQI	Comment
49	GW49	480	7.2	650	1.46	406	106.5	42.8	0.0287	45.22	Excellent
50	GW50	180	7.0	6090	1.48	5610	1917	90.4	0.0290	430.86	Unfit
51	GW51	450	6.7	4790	2.17	6450	1420	178.8	0.0286	406.26	Unfit
52	GW52	450	7.6	630	1.31	398	35.5	31.6	0.0293	40.11	Excellent
53	GW53	1060	7.0	820	1.18	512	177.5	30.0	0.0287	55.44	Good
54	GW54	950	7.3	840	1.87	548	106.5	16.0	0.0299	53.11	Good
55	GW55	950	7.4	550	1.36	537	106.5	20.8	0.0285	47.65	Excellent
56	GW56	1100	7.0	930	1.20	604	106.5	21.2	0.0294	55.43	Good
57	GW57	250	7.0	1900	1.19	1176	497	54.8	0.0296	117.44	Poor
58	GW58	500	7.3	1050	1.52	671	142	30.4	0.0297	63.03	Good
59	GW59	480	7.7	1110	1.66	688	177.5	27.2	0.0285	67.67	Good
	Max.	1260.0	7.9	6830	4.21	6450.0	2769	178.8	0.0341	454.37	
	Min.	150.00	6.4	100	0.59	7.240	35.50	16.00	0.0285	40.11	
	Avg.	775.76	7.3	1650	1.60	1188.70	414.6	52.03	0.0290	108.94	

of 7.4 in the drinking water of the Khulna area. Sikder et al. (2014) also observed an average pH value of 7.89 in the premonsoon season in Khulna coastal belt areas.

The total concentration of soluble salt, in the sampled water, expressed by EC, is one of the most prominent parameters to measure WQI, especially in coastal or salinity prone areas. EC value in this study varied from 100 to 6830 µS/ cm with an average value of 1650 µS/cm (Table 4), and around 37% water samples in Khulna City was included in the maximal allowable limit of EC (1000 µS/cm) given by BIS (1991) and NDWQS (2004) which presents the high amount of salts in the groundwater of the selected areas. The spatial distribution of sampling units could explain this wide variation in EC values. It was observed that some sampling units were situated very close to the tidal river (Fig. 1) and it was confirmed from Table 4 that the water collected from those points contained higher values of EC. The higher EC values in the groundwater studied in this research might be due to enrichment of salts in the water through saltwater intrusion since the areas are located near the coast as well as tidal rivers flow across the city. It might also be because of the interaction of rock and water and agricultural activities (Abbasnia et al. 2019). Sikder et al. (2014) experimented with assessing the groundwater quality of the western area of Khulna in the pre-monsoon season and obtained EC value of drinking water ranged from 789 to 1230 μS/cm, whereas Adhikary et al. (2012) observed an average EC value of 1777 μS/cm by researching six wards of Khulna City Corporation.

The calculated TDS values of groundwater samples fluctuated from 7.240 to 6450.0 mg/l, and the average value was 1188.7 mg/l (Table 4). Around 32% of the total samples exceeded the admissible limits (1000 mg/l) of TDS recommended by WHO (2006). Therefore, the water quality for the

selected tube wells in this study was categorized as brackish (1000 < TDS < 10,000 mg/L) (Logeshkumaran et al. 2015; Abbasnia et al. 2019). TDS in the groundwater studied in this research was high due to enrichment of salts in the water. It might also be because of the contact between rock and water and agricultural activities (Abbasnia et al. 2019). An average value of 1043 mg/l TDS in Khulna (Adhikary et al. 2012) and an average value of 1556.05 mg/l (Islam et al. 2017) in the groundwater of the coastal aquifer of Khulna city were also reported. Percolation of elements from rocks and gypsum rocks might be another reason for high EC and TDS in the water. Excessive withdrawal of groundwater, as well as excessive use of chemical fertilizers in the agricultural field contributing the higher concentration of ions into the groundwater, might be the other reasons for decreasing the water quality in terms of EC and TDS (Abbasnia et al. 2019).

Analysis of DO concentration in water is a significant step for WQI estimation, as oxygen is one of the essential regulatory factors that control the metabolic process of living substances in water and act as an indicator of the condition of water (Basavaraddi et al. 2012). The estimated maximum, minimum, and average values of DO were 4.21, 0.59, and 1.60 mg/l, respectively (Table 4). The estimated values were less than the standard limit of DO 6.0 mg/l, which is suggested by DPHE (2018) and ECR (1997). Sikder et al. (2014) also found a lower average DO value of 3.07 mg/l in the pre-monsoon season in Khulna. That means the observed results from different studies indicate that the condition of DO in this region is not at a satisfactory level and much lower than the standard condition.

The quantity of chloride concentration in the present study was in the range of 35.50–414.6 mg/l. The average value was 2769 mg/l (Table 4) indicating a higher value of chloride and



around 47% of the sampled water exceeded the permissible limit of chloride in drinking water which is 250 mg/l according to WHO (2006), and more than 64% of samples surpassed the acceptable limit (150 mg/l to 600 mg/l) mentioned in ECR (1997) of Bangladesh standard. The presence of high-chloride concentration in the groundwater might be due to contact between water, and soil and rock; and weathering as well as owing to anthropogenic contribution like effluent of wastewaters (Abbasnia et al. 2018). The higher amount of chloride (1776.74 mg/l) in shallow groundwater aquifer in the coastal region of Khulna was reported by Islam et al. (2017). Adhikary et al. (2012) also showed similar results of high chloride content in the groundwater of the Khulna region.

The total hardness of water represents the total dissolved calcium and magnesium content in water, is also an essential factor that helps to determine the suitability of water for drinking purpose (Howladar et al. 2018). The total hardness of the water in the Khulna city area changed from 16.00 to 178.8 mg/l, and the mean value was 52.03 mg/l (Table 4). The values lie under the allowable limit of hardness according to Bangladeshi standard, 200–500 mg/l (ECR, 1997), and WHO standard, 500 mg/l (WHO 2006). So, the total hardness of the sampled water was satisfactory for drinking purposes.

However, Adhikary et al. (2012) observed a higher value of total hardness (641 mg/l) from the groundwater sample of six wards of Khulna City Corporation, Bangladesh.

The amount of nitrate in the groundwater was minimum and under the allowable value of 10 mg/l and 45 mg/l, given by Bangladesh standard, ECR (1997), and WHO (2011), respectively. The results revealed that the nitrate concentration was in the range of 0.034–0.029 mg/l and the average value of nitrate was 0.026 mg/l (Table 4). Lower quantities of nitrate in the water sample of the Khulna region were also reported (Islam et al. 2017; Sikder et al. 2014).

The spatial analysis map of the depth of tube wells and physicochemical parameters of water

Depth wise spatial distribution of tube wells and physicochemical parameters of the water samples in the experimental areas of Khulna City are presented in Figs. 2, 3, 4, 5, 6, 7, 8, 9. The region covered by numerous water types is calculated from the WQI maps and given in Fig. 10. From the maps, it was clearly shown that the values of physicochemical parameters varied from place to place irrespective of the depth of the tube wells.

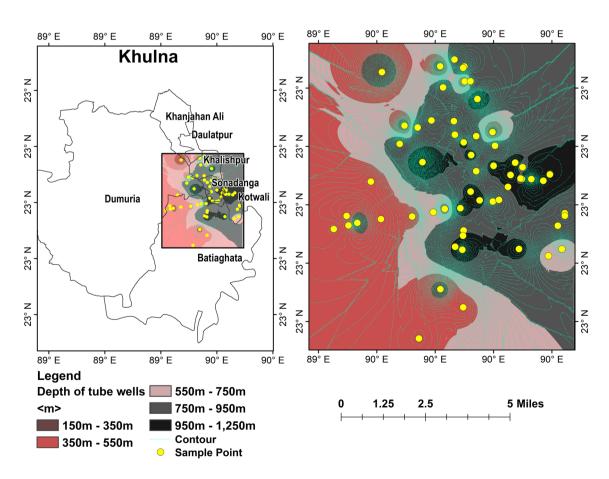


Fig. 2 Depth wise distribution of the tube wells selected for sampling



Fig. 3 Spatial distribution of pH in selected tube wells water

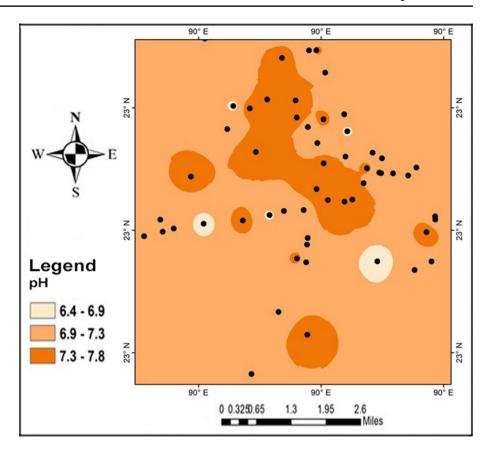
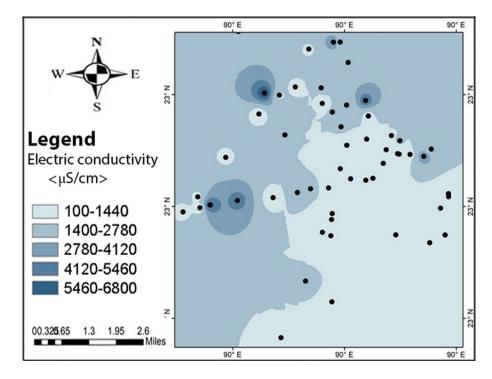


Fig. 4 Spatial distribution of EC in selected tube wells water



Water quality index (WQI)

The value of WQI of the sampled water with respective categories of each water sample is presented in Table 4.

In the present experiment, the computed values of WQI ranged from 40.11 to 454.37, and the average value was 108.94. Here, lower values of WQI (< 50) were excellent and higher values (> 300) were unfit for drinking purposes



Fig. 5 Spatial distribution of DO in selected tube wells water

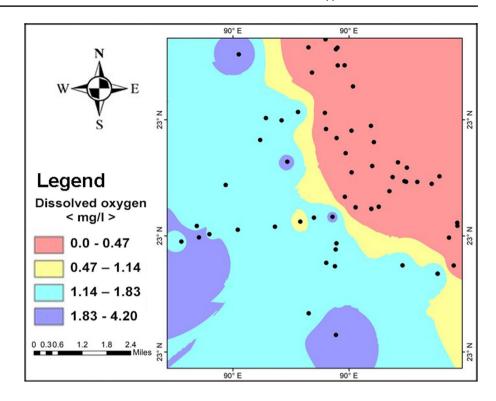
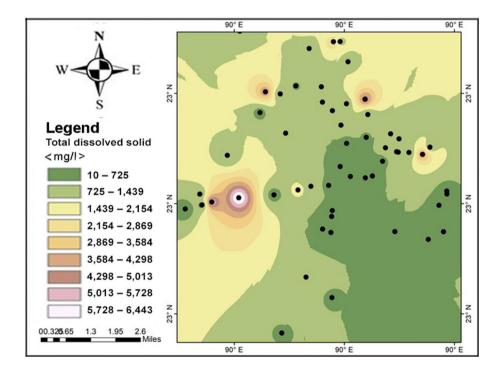


Fig. 6 Spatial distribution of TDS in selected tube wells water



(Table 3). Among all the groundwater samples, 11.9% of groundwater samples were "excellent", 54.2% were "good", 23.7% were "poor", 1.7% were "very poor", and 8.5% were "unfit for drinking purpose" based on WQI. So, more than half of the location falls in excellent to good categories. A suitable diagram was prepared (Fig. 10)

by spatial analysis system for clear visualization of the areas which are categorized under different water quality parameters by using the WQI values of 59 sampling points in Khulna city of Bangladesh. In the GIS map, the five categories were delineated with different color, and each color represents the information about the spatial



Fig. 7 Spatial distribution of chloride in selected tube wells water

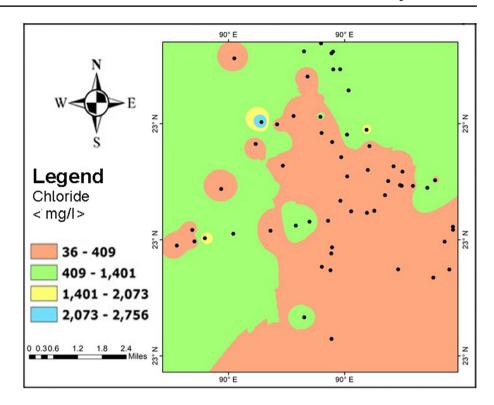
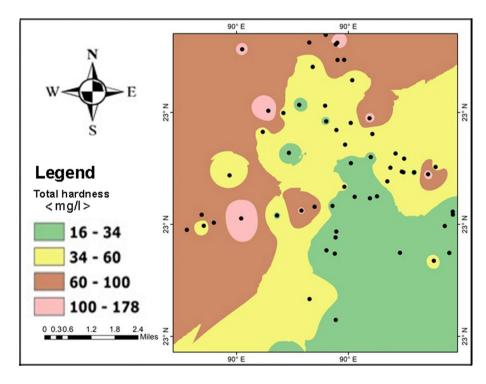


Fig. 8 Spatial distribution of total hardness in selected tube wells water

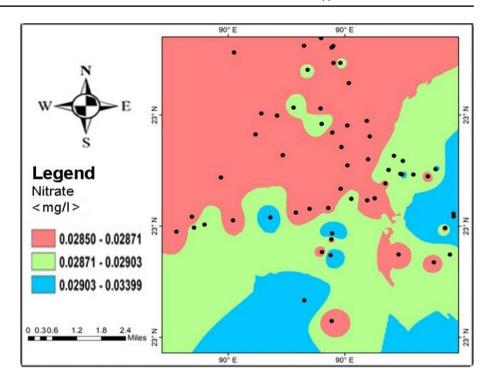


distribution of water of varying quality. From the map, it is visualized that, the tube wells of the south-east area of Khulna city supply better quality drinking water than those of the north-west side. It is also mentionable that the WQI values were not compatible with the values of depth of tube wells.

One vital fact is that there is no mentionable published research work to identify the suitability of groundwater for drinking purposes through investigating the WQI in the Khulna City area. So, it was not possible to compare our experimental data directly with any other studies. However, our findings were quite similar to the results of Akter et al.



Fig. 9 Spatial distribution of nitrate in selected tube wells water



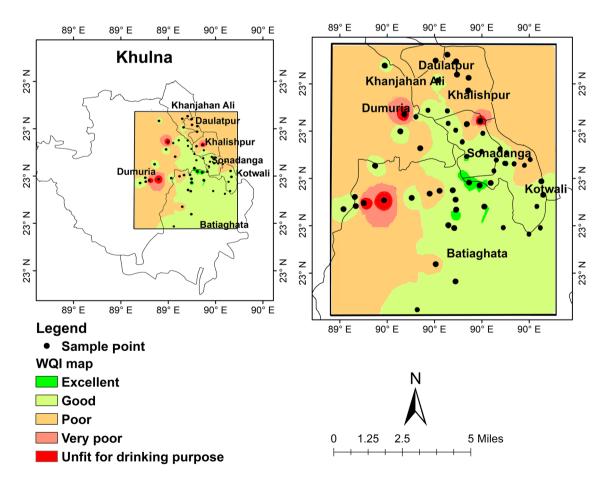


Fig. 10 Water quality index map for selected sampling sites



(2016) who conducted research to estimate different water quality parameters and WQI of different areas of Bangladesh including Khulna as a coastal area and observed that majority of groundwater sources were under the good category of drinking water. Though a considerable percentage of groundwater sources were suitable for drinking purpose, a significant number of tube wells supplies poor quality drinking water. It is conceivable that the higher values of EC, TDS, and chloride in the groundwater were the prominent factors of the higher values of WQI which indicates poor drinking water quality indicating high salinity of the water. Due to being a coastal area, the intrusion of saline water in some regions of Khulna can be considered as one of the critical reasons for the deterioration of water quality for drinking purpose in those areas (Adhikary et al. 2012; Akter et al. 2016).

Conclusion

The suitability of the groundwater, as a significant source of drinking purposes, was investigated in the Khulna city, the southwestern part of Bangladesh. WOI is the valuable and sole rating to describe the overall water quality status in a single term. However, WQI depicts the combined effect of different water quality parameters and transfers water quality information to the public and government decision-makers. WQI is a single value to represent the water quality of a source along with decreasing the larger number of parameters into a simple countenance resulting in easy understanding of water quality observing data. This present research work demonstrated the analytical data to assess the water quality and the utility of GIS, which combinedly represent the WQI of 59 selected tube wells of Khulna city area in Bangladesh through mapping. The WQI for groundwater samples ranged from 40.11 to 454.37, with an average value was 108.94. More than 60% of the samples were in the good and excellent category in water quality assessment, suitable for drinking water, and the other 40% of the samples were needed to be treated before using drinking purposes. The spatial distribution map of DO, EC, TDS, pH, chloride, nitrate, and total hardness showed that these parameters were not distributed uniformly throughout the study area. The high value of WQI at these samples had been found mainly from the higher values of salinity problem (high EC, TDS, and chloride) in the groundwater which might be due to seawater intrusion since Khulna city is situated in the coastal region of Bangladesh.

Compliance with ethical standards

Conflict of interest The authors declare that they have no conflict of interest.

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