**ORIGINAL ARTICLE** 



# Petrography and origin of dedolomites of the Ordovician Majiagou formation in the southeastern Ordos Basin, China: implications for reservoir quality

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#### Abstract

Dedolomites, the replacement of dolomite by calcite, are widely distributed in the Ordovician Majiagou Formation of the southeastern Ordos Basin, China, which critically affects reservoir quality throughout the region. Two types of dedolomites were recognized in the upper 100 m of the Majiagou Formation, namely brecciated dedolomite and crystalline dedolomite. The petrographic and geochemical data reveal that the investigated dedolomites are often associated with clay minerals and dissolution-collapse breccia, indicating their close relationship with erosional unconformities. The relative negative  $\delta^{18}$ O and  $\delta^{13}$ C values and relative high Fe, Mn contents of dedolomite with respect to corresponding dolomite are interpreted as being result from meteoric phreatic water involvement. The dedolomitization process related to subaerial exposure is conventionally interpreted to be formed by dolomite dissolution and calcite precipitation. The rhombic shape crystals in the brecciated dedolomite was formed by an Mg<sup>2+</sup>-loss process under evaporite solution conditions, which retain the rhombic shape of the dolomite. The crystalline dedolomites, however, were formed by dissolution/precipitation and recrystallization process to form medium to coarse calcite mosaic with very low intercrystalline porosity. The resulting fabrics are mainly controlled by the paleo-topography. The precursor dolomite of the crystalline dedolomite is primarily recrystallized mudstone, which is more soluble and favorable for calcite precipitation. The crystalline dedolomite mostly occurs in the low paleotopographic locations, where the karst-saturated water with respect to CaCO<sub>3</sub> is more concentrated, occluding the remaining porosity when excess calcium is supplied. It's, therefore, recommended to implement drilling in tectonic highland and avoid low paleotopographic locations.

Keywords Dedolomitization · Majiagou formation · Diagenetic stages · Reservoir quality · Ordos Basin

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# Introduction

Dedolomitization is a process by which dolomite is replaced by calcite in the presence of a calcium-rich solution, which is often associated with evaporite minerals dissolution. The product of dedolomitization is referred to as dedolomite (Von Morlot 1847; Shearman et al. 1961; Evamy 1967; Flügel 2010). Dolomitization is relatively well documented, whereas little attention has been paid to dedolomitization. Dedolomitization occurs under a wide range of diagenetic conditions, such as during burial-diagenesis (Stoessell et al. 1987; Schoenherr et al. 2018), related to contact metamorphism (Faust 1949; Wood and Armstrong 1975), and associated with hydrothermal activity (Ronchi et al. 2004; Huang et al. 2013; Lyu et al. 2013). However, most of the reported dedolomite occurrences are indicative of near-surface conditions, and associated with the effects of meteoric water along unconformities (Kenny 1992; Nader et al. 2008; Fu et al. 2008; Reuning et al. 2009).

With respect to dedolomitization effects on the reservoir quality, a simulation performed by Escorcia et al. (2013) proposed that a porosity reduction is owing to calcitization. However, only when the extent of dolomite and evaporite dissolution exceeds that of calcite precipitated, the overall porosity can be increased (Canaveras et al. 1996; Ayora et al. 1998; Nader et al. 2008). In most cases, dolomite dissolution and calcite precipitation occur in different solutions at different time, owing to excess of calcium supplied to the system that leads to calcite cementation. Calcite crystals often mosaic contact tightly that reduces reservoir porosity (Fu et al. 2008). However, the porosity reduction by dedolomitization in aforementioned case study is more likely to be caused by calcite cementation. Therefore, the question of whether the reservoir quality is enhanced by dedolomitization or not remains controversial.

Dedolomitization can be controlled by several factors, such as the Ca<sup>2+</sup>/Mg<sup>2+</sup> ratio in the diagenetic fluid (Khalaf and Abdal 1993), dolomite stoichiometry (Nader et al. 2008), CO<sub>2</sub> partial pressure (De Groot 1967; Clark 1980; Hallenberger et al. 2018), temperature and salinity (Hardie 1967; Hallenberger et al. 2018) and the degree of fracture development (Al-Hashimi and Hemingway 1973; Budai et al. 1984; Zeeh et al. 2000; Reuning et al. 2009). The major factors that control dedolomitization in the Majiagou Formation are discussed in this study to provide further insight into the dedolomitization process.

Moreover, this study aims to provide new data on dedolomite fabrics, mechanisms, and effect of dedolomitization on reservoir quality of the Ordovician Majiagou Formation of the Ordos Basin in China as a case study, and to discuss its possible origin by means of detailed thin section investigation, scanning electron microscope (SEM), energy dispersive spectroscopy (EDS) analysis and stable isotope measurements.

# **Geological setting**

The Ordos Basin, the second largest basin in China with an area of  $2.5 \times 10^5$  km<sup>2</sup>, is a large scale intra cratonic basin (Fig. 1) (Yang et al. 2011). The basin was uplifted during the Caledonian orogeny in the early Paleozoic (He 2003) and subject to deep erosion for more than 140 My, resulting in



Fig. 1 Simplified structural map of the Ordos Basin showing the study area (He 2003) and the lithocolumn of main sampled Yican1 well

the missing of Silurian, Devonian and Early Carboniferous rock units (Yang et al. 2005) forming a significant regional unconformity above the Ordovician Majiagou Formation. The carbonate succession of the Ordovician Majiagou Formation has a thickness of > 500 m that is overlain by continental sediments of the Carboniferous Benxi Formation (Fig. 1). According to its structural framework, the Ordos basin can be subdivided into six tectonic units, these are the Jinxi flexible fold belt, Weibei uplift, a western margin thrust belt, Yishan slope, Tianhuan depression and Yimeng uplift (Feng et al. 1999; Fu et al. 2001; Wei et al. 2019). The study area is located in the southeastern Ordos Basin across the Yishan slope and Weibei uplift (Fig. 1).

The Ordovician Majiagou Formation can be subdivided into successive six members ( $Ma_1 \sim Ma_6$  from base to top). Owing to periodic sea level change, a set of transgression-regressive cycles are represented in the Majiagou Formation, in which the Ma<sub>1</sub>, Ma<sub>3</sub> and Ma<sub>5</sub> members are mainly composed of dolomite and anhydrite in evaporitic tidal flat facies and marine anhydrite lake facies, whereas the Ma2, Ma4 and Ma6 members are dominated by limestone and dolomite representing open platform facies (Fig. 1) (Li et al. 2008; Yang et al. 2011). The Ma<sub>6</sub> member consisting mainly of limestone is eroded in most regions and occurs in the southern part of the basin with a thickness ranging from 10 to 20 m (Bao et al. 2004). The  $Ma_1$  to  $Ma_5$  members are developed across the basin and can be laterally traced for several kilometers. Although the limestone of the Ma<sub>6</sub> is developed in southeast of the Ordos basin overlying the Ma<sub>5</sub> dolomite, due to its vicinity to the central paleouplift in Caledonian period, bedding karstification may occur in the  $Ma_5$  in slope region (Tian et al. 2017). Dolomite of the Ma5 was therefore subject to meteoric diagenesis and karstification in the study area. Up to 10 sub-members (Ma<sub>5</sub><sup>1</sup>,

Table 1 Co 1 well  $Ma_5^2...Ma_5^{10}$  from top to bottom) have been identified in the Ma<sub>5</sub> member with cyclic carbonate-evaporite intervals related to the short term sea-level variations (Fig. 1) (Bao et al. 2004). The  $Ma_5^1$  to  $Ma_5^4$  sub-members are mainly composed of microcrystalline dolostone, anhydrite, karst breccia and dolostone with anhydrite nodules. Dolostone with anhydrite nodules is interpreted to be formed by penecontemporaneous dolomitization mode of sabkha evaporative tidal flat (Bao et al. 2017). Because of the dissolution of anhydrite nodules in meteoric water, a favorable reservoir is developed in the upper four sub-members of the Ma<sub>5</sub>, which becomes major gas producing strata (Tian et al. 2017). For instance, the Yican 1 well, drilled in karst monadnock in 2014, produced  $3.5 \times 10^4$  m<sup>3</sup> gas daily in Ma<sub>5</sub><sup>1</sup> and Ma<sub>5</sub><sup>4</sup> submembers where the anhydrite moldic pores and fractures are developed. Whereas Yi 5, Yi 9, Yi 13, Yi 16 wells located in karst groove commonly have low production of oil and gas.

#### Sampling and methods

More than 240 core samples of the Ordovician Majiagou Formation were taken from 9 wells in the southeast of the Ordos basin and were examined by optical microscope (Fig. 1). More than 100 m cores are obtained from the Yican 1 well (Table 1). A total of 72 samples were analyzed for oxygen and carbon isotopes and trace element geochemistry. Prepared thin sections were stained with Alizarin Red S to differentiate between calcite and dolomite following Lindholm and Finkelman (1972). Core-plug porosity and permeability were determined from 72 samples using conventional 3020-062 helium porosimetry and Klinkenberg-corrected GDS-90F gas permeametry according to GB/T29172-2012 Core Analysis Method.

$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	ring table of Yican	Core barrel	Formation	Cores (m)	Drilling interval (m)	Core interval (m)	Core recovery (%)
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		5	Ma <sub>5</sub> <sup>1</sup>	2632.98~2636.68	3.70	3.66	98.92
7 $Ma_5^{3}$ 2652.07 ~ 2658.076.005.5091.678 $Ma_5^4$ 2677.02 ~ 2683.506.485.4584.109 $Ma_5^4$ 2683.50 ~ 2692.709.206.0565.7610 $Ma_5^6$ 2723.82 ~ 2733.039.219.21100.0011 $Ma_4$ 2806.74 ~ 2814.768.022.7033.6712 $Ma_4$ 2814.76 ~ 2823.969.209.1899.7813 $Ma_2, Ma_3$ 2943.00 ~ 2952.709.709.6699.5914 $Ma_2, Ma_3$ 2952.70 ~ 2962.409.709.6699.5915 $Ma_2$ 2962.40 ~ 2972.109.709.00100.0016 $Ma_1$ 3120.00 ~ 3129.009.009.00100.0017Sanshanzi3190.80 ~ 3200.009.209.1899.78		6	Ma <sub>5</sub> <sup>2</sup>	2636.68~2652.07	15.39	14.70	95.52
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$		7	Ma <sub>5</sub> <sup>3</sup>	2652.07~2658.07	6.00	5.50	91.67
9 $Ma_5^{44}$ 2683.50~2692.709.206.0565.7610 $Ma_5^{6}$ 2723.82~2733.039.219.21100.0011 $Ma_4$ 2806.74~2814.768.022.7033.6712 $Ma_4$ 2814.76~2823.969.209.1899.7813 $Ma_2, Ma_3$ 2943.00~2952.709.709.6699.5914 $Ma_2, Ma_3$ 2952.70~2962.409.709.6699.5915 $Ma_2$ 2962.40~2972.109.709.00100.0016 $Ma_1$ 3120.00~3129.009.009.00100.0017Sanshanzi3190.80~3200.009.209.1899.78		8	Ma <sub>5</sub> <sup>4</sup>	2677.02~2683.50	6.48	5.45	84.10
10 $Ma_5^{6}$ 2723.82 ~ 2733.039.219.21100.0011 $Ma_4$ 2806.74 ~ 2814.768.022.7033.6712 $Ma_4$ 2814.76 ~ 2823.969.209.1899.7813 $Ma_2$ , $Ma_3$ 2943.00 ~ 2952.709.709.6699.5914 $Ma_2$ , $Ma_3$ 2952.70 ~ 2962.409.709.6699.5915 $Ma_2$ 2962.40 ~ 2972.109.709.70100.0016 $Ma_1$ 3120.00 ~ 3129.009.009.00100.0017Sanshanzi3190.80 ~ 3200.009.209.1899.78		9	Ma <sub>5</sub> <sup>4</sup>	2683.50~2692.70	9.20	6.05	65.76
11 $Ma_4$ 2806.74~2814.768.022.7033.6712 $Ma_4$ 2814.76~2823.969.209.1899.7813 $Ma_2, Ma_3$ 2943.00~2952.709.709.6699.5914 $Ma_2, Ma_3$ 2952.70~2962.409.709.6699.5915 $Ma_2$ 2962.40~2972.109.709.70100.0016 $Ma_1$ 3120.00~3129.009.009.00100.0017Sanshanzi3190.80~3200.009.209.1899.78		10	Ma <sub>5</sub> <sup>6</sup>	2723.82~2733.03	9.21	9.21	100.00
		11	Ma <sub>4</sub>	2806.74~2814.76	8.02	2.70	33.67
13 $Ma_2, Ma_3$ 2943.00~2952.709.709.6699.5914 $Ma_2, Ma_3$ 2952.70~2962.409.709.6699.5915 $Ma_2$ 2962.40~2972.109.709.70100.0016 $Ma_1$ 3120.00~3129.009.009.00100.0017Sanshanzi3190.80~3200.009.209.1899.78		12	Ma <sub>4</sub>	2814.76~2823.96	9.20	9.18	99.78
14 $Ma_2, Ma_3$ 2952.70~2962.409.709.6699.5915 $Ma_2$ 2962.40~2972.109.709.70100.0016 $Ma_1$ 3120.00~3129.009.009.00100.0017Sanshanzi3190.80~3200.009.209.1899.78		13	Ma <sub>2</sub> , Ma <sub>3</sub>	2943.00~2952.70	9.70	9.66	99.59
15 $Ma_2$ 2962.40~2972.109.709.70100.0016 $Ma_1$ 3120.00~3129.009.009.00100.0017Sanshanzi3190.80~3200.009.209.1899.78		14	Ma <sub>2</sub> , Ma <sub>3</sub>	2952.70~2962.40	9.70	9.66	99.59
16         Ma <sub>1</sub> 3120.00~3129.00         9.00         9.00         100.00           17         Sanshanzi         3190.80~3200.00         9.20         9.18         99.78		15	Ma <sub>2</sub>	2962.40~2972.10	9.70	9.70	100.00
17 Sanshanzi 3190.80~3200.00 9.20 9.18 99.78		16	Ma <sub>1</sub>	3120.00~3129.00	9.00	9.00	100.00
		17	Sanshanzi	3190.80~3200.00	9.20	9.18	99.78
Total 114.50 103.65		Total			114.50	103.65	

Powder samples (~30–50 mg per single sample) of dolostone, brecciated dedolomite and crystalline dedolomite were extracted for carbon and oxygen isotope measurements. Samples are microdrilled with a handheld dental drill. The powdered samples were heated to remove organic materials and then reacted with anhydrous phosphoric acid, under vacuum condition, to release CO<sub>2</sub> at 25 °C for 24 h. The CO<sub>2</sub> was then analyzed for  $\delta^{13}$ C and  $\delta^{18}$ O value on a Finnigan MAT251 mass spectrometer. All  $\delta^{13}$ C and  $\delta^{18}$ O values were reported in ‰ units relative to the Vienna Pee Dee Belemnite (VPDB) standard.

Ultra-fabrics were studied using a Melin type Scanning Electron Microscope (SEM) (Carl Zeiss AG) operated at 15–20 kV with a 10 nA beam current and working distance of 10 mm. The elemental concentrations and spatial variation of micron-sized spots were obtained by Energy-Dispersive Spectrometry (EDS) at the China University of Petroleum (College of Geosciences), which could generate high-resolution, high-magnification images of carbonate textures.

#### Results

# Petrography

The Ordovician Majiagou Formation in the study area  $(Ma_1-Ma_6 \text{ members})$  are mainly composed of five lithofacies types including dolostone, calcitic dolostone, crystalline limestone, anhydrite dolostone and dolomitic anhydrite.

#### **Dolostone lithofacies**

Dolostone is the dominant lithology in the Majiagou Formation. Two types of dolostone were recognized in the Majiagou Formation, including dolostone with anhydrite nodules and dolostone without anhydrite nodules. The anhydrite nodule-bearing dolostone is restricted to the  $Ma_5^{-1}$  submember with a thickness less than 10 m. These rocks are mainly composed of dark gray dolomicrite with the size of  $10 \sim 30$  um and the nodules accounting for 5%–10% of the core sample with size of 1-2 mm (Fig. 2a–c). Most anhydrite nodules are dissolved (Fig. 2a, b), where pores are filled with fine crystalline dolomite, medium-coarse calcite, authigenic quartz and pyrite (Fig. 2c).

The dolostone without anhydrite nodules is mainly distributed below the  $Ma_5^4$  sub-member. The dolostone is composed mainly of dolomicrites. Some dolomicrites are argillaceous sediments showing banded structure. The crystals of dolomicrites are subhedral or anhedral, which are tightly interlocked and mosaic contacted.

#### Calcitic dolostone lithofacies (Brecciated dedolomite)

The brecciated dedolomite lithofacies is mainly composed of dark gray matrix and light gray clasts with size of  $0.1 \sim 1$  cm (Fig. 2d-f). The rock has been cracked with cmscale pockets of clays filled in the fractures (Fig. 2d). Some rocks exhibit brownish and reddish color in hand specimen (Fig. 2d). Detailed microscopic investigation reveals that the dark gray matrix is mainly composed of dolomicrite (Fig. 2g-i). The light gray clasts mainly contain very fine to medium subhedral to euhedral calcitized dolomite crystals with sizes ranging from 10 to 200 mm (Fig. 2g-i) and fine crystalline dolomite matrix. The calcitized dolomite crystals, which have been stained red by Alizarin Red S, are scattered in the micritic dolostone matrix or form crystal aggregates as the clasts of breccia (Fig. 2g). This type of calcitized dolomite crystals are identified as dedolomites. The fractures and veins are well developed, which cut both the matrix and clasts (Fig. 2h). The studied brecciated dedolomite mainly occurs in the uppermost 100 m of the Ma5, corresponding to the Ma<sub>5</sub><sup>1</sup>–Ma<sub>5</sub><sup>4</sup> sub-members below the paleo-karstic surface located at the top of the Majiagou Formation in the Yican 1 well (Fig. 1).

The SEM analyses show that the dark gray crystals in the large rhombic crystals (Fig. 3a, point 27) exhibit dolomite peak shapes in the EDS spectrum (Fig. 3b), whereas the light gray parts of the rhomb show calcite peak shapes (Fig. 3c, d, point 28). Four types of microfabrics are recognized in the dedolomite crystals: (1) cloudy cores of precursor dolomite rhombs are often preferentially replaced by calcite (Fig. 4a-c). Detailed BSE images reveal that the small dolomite crystals, which are found in the large dedolomite rhombs, are gathered at the rims of dedolomite crystals. Moreover, they seem to have the same size as the surrounding dolomitized matrix (Fig. 4c); (2) small dolomite crystals occurring in dedolomite crystals are scattered along the core/rim interfaces, but are not cut by the fracture across the dedolomite crystal (arrow). The small dolomite crystals are sub-euhedral to euhedral, differently oriented single crystals, which seems not relics of precursor dolomite, but another phase of dolomite growth that formed post-dated the dedolomitization (Fig. 4d-f); (3) dolomite rhombs pervasively replaced by calcite with dolomite patches scattered in calcitized dolomite (Fig. 4g-k). The dolomite crystals occur in dedolomite are sub-euhedral to anhedral with different sizes. (4) dedolomite fabric formed by internal partial or complete dissolution of dedolomite (Fig. 4j, k) rhombs. Progressive dissolution may lead to the emptying of dedolomite rhombs to form open rhombohedral pores (Fig. 41).

Detailed SEM images of zoned dedolomite and dolomite shows progressive dedolomitization process from core/rim interface to crystal core (Fig. 5a–c). The ghost of rhombic dedolomite outlined by the crystal lattice with gradual color



Fig. 2 Petrographic characteristics of the Ordovician Majiagou Formation of Yican 1 well in the southeastern Ordos Basin. Thin sections were stained with Alizarin Red-S. **a** Anhydrite nodule-bearing dolomite with anhydrite nodules dissolved to form dissolution moulds; **b** Anhydrite nodule-bearing dolomite with fractures filled with clay; **c** Anhydrite nodule-bearing dolomite with anhydrite nodules filled with calcite, dolomite and quartz; **d** Karst breccia with dark gray mudstone and dark brown clay (arrow) filled in the brec-

cia clasts; **e** and **f** Dolomitic breccia composed of light gray clasts (arrows) and dark gray micrite matrix; **g** Microphotograph of brecciated dedolomite showing dedolomite rhombs (arrows) floating in the micrite matrix or form an aggregates; **h** Open fractures across the dolomite matrix and dedolomite aggregates (arrow); **i** Brecciated dedolomite showing dedolomite rhombs (arrows) floating in the micrite matrix or form an aggregates (arrows)

change seems like erosional relics (Fig. 5b, the enlargement of top right of Fig. 5a), which are different from the small euhedral dolomite crystals surrounding it. In the zoned dolomite (Fig. 5c, the enlargement of bottom of Fig. 5a),  $Mg^{2+}$  content generally decreases from the dark gray cores to light gray rims (Fig. 5d). Although the value of point 4 shows some deviation, the general trend is consistent with the observed color changes.

#### Crystalline limestone (crystalline dedolomite)

The crystalline limestone shows various colors from gray to brownish red, or occurs as brownish red patchy in light gray dolostone (Fig. 6a–c). The patchy texture is more obvious due to clay concentrated among breccia. Some specimens are characterized by cm-scale layered structures, composed of gray dolomite and brownish red calcite (Fig. 6c). Open vugs or pores presumably formed by dissolution are observed in crystalline limestone (Fig. 6a, d). The vugs and pores, which are filled by calcite or dolomite, are found in multiple dedolomitized horizons with the pore size of  $0.5 \sim 2.0$  cm. The pores and vugs are common in the crystalline limestone (brownish red layer in Fig. 6e), mostly distributed in isolation or bead-like arrangement (Fig. 6e, f), but rarely observed in the dolomite horizons (gray layer in Fig. 6e).

Petrographically, the crystalline limestone is predominantly composed of xenotopic, medium to coarse, crystalline (mostly 200 to 1000um in diameter) calcite mosaics **Fig. 3** SEM images and EDS diagram of dedolomite crystal and dolomite rhomb in it. **a** Dark gray rhomb (point 27) represents dolomite identified from the (**b**) EDS diagram; **c** Enlargment of figure a in red box showing light gray background representing calcite (dedolomite in this study) identified from the (**d**) EDS diagram



with variable amount of dolomite impurities (Fig. 6g, h). The dolomite rhombohedra boundaries are obscured by calcite but retain part of the dark, cloudy core impurities, and appeared to be relics after replacement (Fig. 6g, h), which is identified as crystalline dedolomite. The medium-coarse calcite crystals show irregular intercrystalline boundaries and are interlocked with each other to make the rock dense with low porosity (Fig. 6g, h). The intercrystalline pores are enriched in clay minerals. In some samples, the clay minerals were found to have calcitic edges (Fig. 6i). The crystalline dedolomite mainly occurs in the  $Ma_5$  member of Yi 5, Yi 9, Yi 13, Yi 16 wells.

#### Anhydritic dolostone and dolomitic anhydrite lithofacies

This lithofacies is mainly composed of mm-scale lamination of dominantly dolostone and anhydrite layers (Fig. 7a). Petrographically, the dolostone and anhydrite layers are interbedded displaying an abrupt contact (Fig. 7b). The very-fine bedding fractures are filled with calcite parallel to the anhydrite and dolomite layers (Fig. 7b). In some cases, the fine-grained anhydrite layers are characterized by softsediment deformation fabrics such as convolute bedding and shear folds in wavy arrangement (Fig. 7c). Another common fabric is a mosaic of "chicken-wire" texture, which is often interpreted to be indicative of a sabkha environment (Fig. 7d, e) (Kendall 1979; Sallam et al. 2019a, b). The Majiagou Formation typically shows a lithofacies change from sabkha-type anhydrite to anhydrite-after-gypsum nodules. This change is regionally accompanied by an increasing frequency of anhydrite layer towards the lithologic base of the formation. With the increasing of anhydrite content, the anhydritic dolostone is gradually transitioned downward into dolomitic anhydrite (Figs. 7d–f). The massive and milky anhydrite is mainly distributed in and below the Ma<sub>4</sub> member (Fig. 7f).

#### Petrophysical data

The petrophysical data from the Majiagou Formation in the study area shows that the maximum porosity of the reservoir is up to 10.8% and the maximum permeability is 15.45 mD occurring in the anhydrite nodule-bearing dolomite of the  $Ma_5^1$  sub-member except the anomalously high permeability that caused by fractures (Table 2). The dominant pore types defining good reservoir facies are anhydrite moldic pores and fractures (Figs. 2a, b). The anhydrite moldic pores are 1–2 mm in diameter accounting for more than 10–20% of the cores, which mainly occurs in the  $Ma_5^1$  to  $Ma_5^4$  sub-members as shown in Fig. 2a. The fractures occur throughout the whole cores including structural fracture, gravity seam, cracking crack (Fig. 2b) etc. When the anhydrite is transformed to gypsum, the volume is usually increased by 30%, resulting in great swelling force on the surrounding



**Fig. 4** Thin section photomicrographs and SEM images of dedolomites from the Majiagou Formation of Yican 1 well in the southeastern Ordos Basin. Thin sections were stained with Alizarin Red-S to discriminate calcite and dolomite. All thin sections are shown in plane-polarized light. Dd: Dedolomite; Do: Dolomite. **a–c** Dedolomite mimics the rhombic shape of dolomite with dolomite grains in

the rims (arrows). **d–f** Dedolomite with dolomite grains at the core/ rim interface (arrows). Dedolomite crystals are cross-cut by fissures with small dolomite crystal not cut by the fissures. **g–i** Dedolomite with dolomite grains scattered in the romb. **j–l** Dedolomite crystals with rhombohedral pores

rock. After the gypsum is dissolved, the pressure will be released, and this process is repeated, resulting in cracking cracks in the surrounding rock matrix (Fig. 2b). Furthermore, petrophysical data show that most of the calcite samples (mainly dedolomite) have low porosity and permeability, whereas dolomite commonly correlates to high porosity and permeability except some anomalously low porosity and permeability of dolomicrite (Fig. 8). SEM images also reveal that the micropores are well developed in dolomite crystals, but rarely observed in dedolomite (Figs. 5b, c).

Petrophysical data show that crystalline dedolomites have poor reservoir quality (lower porosity and lower



Different points in the dedolomite zone

Fig. 5 a-c SEM image of zoned dedolomite showing progressive dedolomitization process from light gray rim to dark gray core (arrows); **d** Elemental composition of zoned dolomite showing the element changes from point 1 to point 6

permeability) than dolomite sections throughout the Majiagou Formation (Fig. 8). The crystalline dedolomites are well developed in the wells Yi 5, Yi 9, Yi 13, Yi 16 in the southeast of the Ordos basin, where the reservoir quality is poor and the gas content is low. The effective gas bearing layer was not found in either Yi 9 or Yi 16 wells. The average porosity of Yi 16 well is 1.04% and average permeability of 0.031 mD, which is quite lower than that in wells Yican 1 and Yi 8, where the crystalline dedolomites are not developed (Table2, Fig. 8).

#### **Geochemical results**

Geochemical data related to the investigated samples are listed in Table 3, including the Fe, Mn, Sr and MgO content, as well as  $\delta^{18}$ O and  $\delta^{13}$ C compositions. The brecciated dedolomite is characterized by negative  $\delta^{18}$ O, ranging from – 11.97‰ to – 8.69‰, average of – 10.14‰, which is lower than the dolostone, which ranges from – 11.79‰ to – 7.10‰, average of – 8.68‰. The  $\delta^{13}$ C values of the

brecciated dedolomite, ranging from – 5.14‰ to – 0.45‰, average of – 1.97‰ is also negative corresponding to dolostone, which ranges from – 3.39‰ to 1.20‰, average of – 0.75‰. The  $\delta^{18}$ O and  $\delta^{13}$ C values of crystalline dedolomite in the Majiagou Formation vary from – 10.30‰ to – 8.00‰ (average – 8.84‰) and from – 7.50‰ to – 2.00‰ (average of – 5.86‰). The  $\delta^{18}$ O value of crystalline dedolomite is more negative than those of dolostone and positive to those of the brecciated dedolomite. Whereas, the  $\delta^{13}$ C value of crystalline dedolomite is significantly lower than those of dolostone. The  $\delta^{13}$ C value of brecciated dedolomite is between those of dolostone and crystalline dedolomite (Fig. 9a). The  $\delta^{18}$ O and  $\delta^{13}$ C values of the anhydritic dolostone and dolomitic anhydrite are listed, but are not discussed in this work.

The respective Fe and Mn concentrations of both brecciated dedolomite and dolostone are shown in Table 3. The Fe content of brecciated dedolomite ranges from 2875 to 15,200 ppm (average of 8588 ppm) and Mn content of 33 to 225 ppm (average of 116 ppm), respectively, which is substantially higher



**Fig. 6** The crystalline limestone of the Ordovician Majiagou Formation in southeastern Ordos Basin. **a** Coarsely crystalline limestone with dissolution pores, Yi 13 well, 1384.57 m,  $Ma_5^{2}$ ; **b** Medium-coarsely crystalline limestone with patchy fabric by brownish or red brownish differences, Yi 11 well, 1920.64 m,  $Ma_5^{1}$ ; **c** Coarsely crystalline limestone with banded fabric by brownish or red brownish color differences, Yi 7 well, 2219.29 m,  $Ma_5^{1}$ ; **d** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **e** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **b** Crystalline limestone with dissolution pores, Yi 9 well, 1636.76 m,  $Ma_5^{1}$ ; **b** Crystalline limestone with dissolutine pores, Yi 9 well, 1636.

line limestone with dissolution pores filled with calcite, Yi 12 well, 1998.66 m,  $Ma_5^{1}$ ; **f** Crystalline limestone with dissolution pores filled with calcite and dolomite, Yi 13 well, 1369.92 m,  $Ma_5^{1}$ ; **g** Photomicrograph of crystalline limestone with dolomite impurities in the cores, Yi 12 well, 2030.16 m,  $Ma_4^{1}$ ; **h** Photomicrograph of crystalline limestone with dolomite impurities in the cores, Yi 7 well, 1922.86 m,  $Ma_5^{1}$ ; **i** Clay mineral with calcite rim injected in finely crystalline limestone, Yi 13 well, 1357.40 m,  $Ma_1^{1}$ 

than that of dolostone, ranging from 538 to 6288 ppm (average of 2811 ppm) and 24 to 201 ppm (average of 63 ppm) respectively (Fig. 9b). The Fe and Mn contents of anhydritic dolostone and dolomitic anhydrite show the largest spread with Fe concentration from 1663 to 23,113 ppm and Mn concentration from 44 to 418 ppm.

The average Sr content of the brecciated dedolomite varies from 101 to 195 ppm, average of 150, which is higher than that

of dolostone ranging from 61 to 222 ppm, average of 107 ppm (Fig. 9c). Whereas the Sr content of the anhydritic dolostone and dolomitic anhydrite ranging from 118 to 7363 ppm and 870 to 1197 ppm, is much higher than that in both brecciated dedolomite and dolomite (Fig. 9d).



**Fig. 7** Selected core photographs and microphotographs showing representative anhydrite fabrics of Yican 1 well in southeastern Ordos Basin. **a** Anhydrite (light gray) and dolomite (dark gray) interbedded in drill core, 3126.53 m,  $Ma_1$ ; **b** Lamellar anhydrite and dolomite (dark gray) and dolomite (dark gray) interbedded in drill core, 3126.53 m,  $Ma_1$ ; **b** Lamellar anhydrite and dolomite (dark gray) and dolomite (dark gray) and dolomite (dark gray) interbedded in drill core, 3126.53 m,  $Ma_1$ ; **b** Lamellar anhydrite and dolomite (dark gray) and dolomite (dark gray) and dolomite (dark gray) interbedded in drill core, 3126.53 m,  $Ma_1$ ; **b** Lamellar anhydrite and dolomite (dark gray) and dolomite (dark gray)

mite in thin section, 2962.81 m,  $Ma_2$ ; c Wavy arrangement of anhydrite in dolomite, 2817.51 m,  $Ma_4$ ; d Massive anhydrite in drill core, 2969.58 m,  $Ma_2$ ; e Lump anhydrite in thin section, 2725.03 m, Ma5; f Massive anhydrite in thin section, 2963.69 m,  $Ma_2$ 

# Discussions

#### Origin of the brecciated dedolomites

In dolostone and evaporite intervals, the dissolution of evaporite under leaching of meteoric water may cause the layered collapse of dolomicrite or finely crystalline dolomite to form collapse breccias. The brecciated dedolomite (mainly occurring in Yican 1 well) is formed by the collapse of dolostones and evaporite dissolution (Fig. 2d) indicated from the following four evidences: (1) The breccias are cracked (Fig. 2d), crumped (Fig. 2e) or collapsed (Fig. 2f), and sometimes filled with dark gray mudstones and dark brown clay (Fig. 2d); (2) The clasts are mostly angular and poorly sorted as in Fig. 2f, which are different from the syndepositional breccias, most of which may be rounded and well sorted in tidal flat environment; (3) The breccias mainly occur less than 60 m below unconformity, which is controlled by the hydraulic conductivity of the rocks below the unconformity.; (4)The sizes of the clasts are largely controlled by the thickness of the nonevaporative beds, which are often mm-scale as in Fig. 7a, b in the study area. The petrology characteristics of collapsed dolostone depend on the original lithology, which is mostly dolomicrite or finely crystalline dolostone in the study area.

In evaporite horizons, the breccias are the typical end result of evaporite dissolution (Swennen et al. 1990; Warren 2016). Collapse breccias resulting from evaporite dissolution by fresh water flushing has been genetically associated with dedolomitized sections in many previous case studies (Arenas et al. 1999; Rameil 2008). Brecciation is almost synchronous with the evaporite dissolution driven by meteoric-derived fluids undersaturated with respect to calcium sulphate following uplift in the Late Ordovician, because both the vugs and cavities in inter-clast pores are partly filled by yellow-brown clay (Figs. 2d, 6i). The anhydrite nodulebearing dolostone occurs at the top of the Majiagou Formation (Fig. 2a-c). The nodular morphology suggests precursor gypsum or anhydrite, which is dissolved by meteoric water during an episode of subaerial exposure during the Caledonian orogeny (He 2003).

Dedolomitization is characterized by rhombohedral replacement of dolomite by calcite, which has been stained by Alizarin Red S but remain the rhombic shape indicative of precursor dolomite (Figs. 2g-i). In the dedolomitization process, the rhombic shape of precursor dolomite commonly remained, which is due to different crystal lattice energy of dolomite and calcite. Mg<sup>2+</sup> has ionic radius of 0.75 Å, which is 74.26% that of Ca<sup>2+</sup>, with ionic radius of about 1.01 Å. Because of its smaller ionic radius, Mg-O interionic distance is smaller than that of Ca-O. The smaller an interionic distance is, the stronger the ionic strength effect

Table 2Petrophysical data ofOrdovician Majiagou formationof Yican 1 well, Yi8 and Yi 16wells

No	Samples	Depth	Lithology	Formation	Porosity (%)	Permeability (10-3µm2)	Note
1	68	2633.36	Brecciated Dedolomite	Ma <sub>5</sub> <sup>1</sup>	0.12	0.10	
2	72	2634.91	Dolomite	Ma <sub>5</sub> <sup>1</sup>	10.80	15.45	
3	78	2637.51	Dolomite	$Ma_5^2$	1.10	0.25	
4	82	2637.61	Brecciated Dedolomite	$Ma_5^2$	0.12	0.15	
5	84	2638.97	Brecciated Dedolomite	$Ma_5^2$	1.10	0.02	
6	86	2639.93	Brecciated Dedolomite	$Ma_5^2$	1.10	0.01	
7	88	2640.37	Dolomite	$Ma_5^2$	2.65	3.86	fracture
8	90	2640.9	Dolomite	$Ma_5^2$	2.10	0.29	
9	92	2641.44	Dolomite	$Ma_5^2$	3.60	14.64	fracture
10	104	2645.32	Brecciated Dedolomite	Ma <sub>5</sub> <sup>1</sup>	0.10	0.03	
11	108	2646.64	Brecciated Dedolomite	$Ma_5^2$	0.09	0.07	
12	110	2649.13	Brecciated Dedolomite	$Ma_5^2$	0.60	0.06	
13	112	2650.18	Brecciated Dedolomite	$Ma_5^2$	0.11	0.03	
14	113	2653.49	Dolomite	$Ma_5^3$	3.50	0.02	
15	114	2655.08	Brecciated Dedolomite	Ma <sub>5</sub> <sup>3</sup>	1.30	2.59	
16	118	2655.53	Brecciated Dedolomite	$Ma_5^3$	0.68	0.03	
17	122	2677.56	Brecciated Dedolomite	Ma <sup>4</sup>	2.70		fracture
18	124	2679.12	Dolomite	Ma <sup>4</sup>	0.30	0.01	
19	128	2680.24	Brecciated Dedolomite	Ma <sub>5</sub> <sup>4</sup>	0.79	0.02	
20	132	2681.13	Brecciated Dedolomite	Ma <sub>c</sub> <sup>4</sup>	0.06	0.02	
21	136	2683.75	Brecciated Dedolomite	Ma <sub>c</sub> <sup>4</sup>	0.77	0.42	
22	140	2684.86	Brecciated Dedolomite	Ma <sub>s</sub> <sup>4</sup>	0.19	0.04	
23	146	2686	Brecciated Dedolomite	Ma.4	0.06	0.02	
22	150	2688	Brecciated Dedolomite	Ma_4	1 48	27.75	fracture
25	154	2688.86	Brecciated Dedolomite	Ma <sub>c</sub> <sup>4</sup>	0.33	0.14	mueture
26	156	2000.00	Dolomite	Ma <sup>6</sup>	0.40	0.01	
27	158	2725 52	Dolomite	Ma_ <sup>6</sup>	0.12	5.81	fracture
28	162	2728.32	Anhydrite dolomite	Ma_ <sup>6</sup>	0.08	0.01	inactare
29	162	2732.93	Dolomite	Ma_ <sup>6</sup>	0.00	0.13	
30	178	2806.84	Dolomite	Ma.	0.09	0.02	
31	182	2807.57	Dolomite	Ma.	0.05	0.01	
32	186	2808.49	Dolomite	Ma.	0.05	0.01	
33	188	2809.1	Dolomite	Ma.	0.90	0.01	
34	190	2814.86	Dolomite	Ma.	0.08	0.01	
35	206	2818 31	Dolomite	Ma.	0.07	0.01	
36	214	2819.91	Dolomite	Ma.	0.06	0.02	
37	214	2821.46	Dolomite	Ma.	0.07	0.01	
38	226	2822.40	Dolomite	Ma	0.08	0.02	
30	220	2022.20	Dolomite	Ma	0.13	0.02	
40	250	2954.83	Dolomite	Ma.	0.48	0.03	
41	250	2057	Dolomite	Ma	0.13	25.51	fracture
41	250	2957	Dolomite	Ma	2.10	0.04	macture
42	202	2936.29	Dolomite	Ma	2.10	0.04	
43	200	2950.07	Dolomite	Ma	0.75	0.01	
44	270	2901.34	Dolomitic anhydrita	Mo	1.44	0.04	
45	274	2902.18	Dolomitic annydrite	Mo	0.78	0.04	
40 47	∠10 282	2904.94	Dolomitic annydrite	Ma	0.76	0.04	
-+/ /9	202	2303.97	Dolomita	Mo	0.50	0.02	
40 40	20 <del>4</del> 286	2900.07	Anhydrita dalamita	Ma	0.30	0.01	freeture
50	200	2968.52	Dolomitic anhydrite	Ma	1.10		fracture

Table 2 (continued)

No	Samples	Depth	Lithology	Formation	Porosity (%)	Permeability (10-3µm2)	Note
51	294	2969.38	Anhydrite dolomite	Ma <sub>2</sub>	0.59		fracture
52	298	2970.9	Anhydrite dolomite	Ma <sub>2</sub>	0.74	0.02	
53	302	3120.15	Anhydrite dolomite	Ma <sub>1</sub>	0.91	0.05	
54	306	3121.3	Anhydrite dolomite	Ma <sub>1</sub>	0.05	0.05	
55	310	3122.59	Anhydrite dolomite	Ma <sub>1</sub>	0.37	0.02	
56	316	3124.53	Anhydrite dolomite	Ma <sub>1</sub>	0.43		fracture
57	320	3125.87	Anhydrite dolomite	Ma <sub>1</sub>	1.05	0.24	
58	324	3127.16	Anhydrite dolomite	Ma <sub>1</sub>	1.46	0.01	
59	326	3127.83	Anhydrite dolomite	Ma <sub>1</sub>	1.03	0.02	
	Min				0.05	0.01	
	Max				10.80	27.75	
	Average				1.09	2.26	
60	Yi8	2248.58	Dolomite	Ma <sub>5</sub> <sup>1</sup>	6.22	0.062	
61	Yi8	2248.72	Dolomite	Ma <sub>5</sub> <sup>1</sup>	6.19	0.065	
62	Yi8	2248.89	Dolomite	Ma <sub>5</sub> <sup>1</sup>	4.42	0.054	
63	Yi8	2249.23	Dolomite	Ma <sub>5</sub> <sup>1</sup>	2.45	0.023	
64	Yi8	2249.57	Dolomite	Ma <sub>5</sub> <sup>1</sup>	6	0.062	
65	Yi8	2249.73	Dolomite	Ma <sub>5</sub> <sup>1</sup>	5.78	0.066	
	/	/	Average	/	5.2	0.055	
66	Yi16	2380.65	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	1.1	0.021	
67	Yi16	2381.94	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	1	0.018	
68	Yi16	2382.39	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	0.9	0.014	
69	Yi16	2382.59	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	1.1	0.015	
70	Yi16	2387.16	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	1.3	0.029	
71	Yi16	2388.46	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	1.2	0.109	
72	Yi16	2390.6	Crystalline dedolomite	Ma <sub>5</sub> <sup>1</sup>	0.7	0.013	
	/	/	Average		1.04	0.031	

is. The calcite, therefore, rarely destroys the dolomite fabric in dedolomitization process, but remains its rhombic shape.

Petrographically, the small dolomite crystals that are found in the large dedolomite rhombs at the rims or along core/rim interfaces are euhedral to subhedral, differentially oriented single crystals, which seem not to be remnants of a



Fig. 8 Crossplot of porosity and permeability in Yican 1 well

former, larger dolomite crystal, but another phase of dolomite growth that post-dated dedolomitization. If these are relics of precursor dolomite, they must be cut by the fractures together with calcitized dolomite that cross it (Fig. 4f). The small dolomite crystals are abundant and seem to have the same size with the surrounding dolomitized matrix, which would mean that dedolomitization was followed by a second, pervasive dolomitization event that only spares the cores of large dedolomite crystal.

Many case studies of dedolmitization have been documented in the literature (e.g., Sibley 1980; Khalaf and Abdal 1993; Rains and Dewers 1997; Nader et al. 2008). Most convincing and commonly known of dedolomitization is certainly a dolomite rhombohedron, being partially replaced by equicrystalline anhedral calcite. Schoenherr et al. (2018) summarizes the most common dedolomite microtextures: (1) the dolomite is totally replaced by anhedral calcite in mosaic texture; (2) the dolomite is selectively replaced by calcite at rims or cores of the rhombohedron; (3) the dolomite is partly replaced by calcite with dolomite poikilotopic inclusions in dedolomite rhombohedra. The above petrographic features

 Table 3
 Geochemistry data of Ordovician Majiagou formation in southeastern Ordos Basin

	Samples	Well	Depth (m)	Lithology	$\delta^{18}O(\%)$	$\delta^{13}C(\%)$	Fe(ppm)	Mn(ppm)	Sr(ppm)	MgO(%)
1	82	Yican 1	2637.61	Brecciated Dedolomite	- 10.62	- 5.14	15,200	116	152	1
2	84	Yican 1	2638.97	Brecciated Dedolomite	- 10.44	- 4.39	8363	156	169	4
3	104	Yican 1	2645.32	Brecciated Dedolomite	- 9.54	- 1.74	9038	170	135	12
4	136	Yican 1	2683.75	Brecciated Dedolomite	- 11.97		7275	46	141	1
5	94	Yican 1	2641.86	Brecciated Dedolomite	- 10.97	- 0.45				
6	108	Yican 1	2646.64	Brecciated Dedolomite	- 9.77	- 1.44	8313	225	102	18
7	150	Yican 1	2688.00	Brecciated Dedolomite	- 9.89	- 1.01	4775	115	165	18
8	122	Yican 1	2677.56	Brecciated Dedolomite	- 8.69	- 0.86	13,950	99	141	11
9	146	Yican 1	2686.00	Brecciated Dedolomite	- 10.15	- 3.31	13,750	140		10
10	154	Yican 1	2688.86	Brecciated Dedolomite	- 9.88	- 1.95	2875	33		
11	118	Yican 1	2655.53	Brecciated Dedolomite	- 10.24	- 1.09	4175	107	101	4
12	112	Yican 1	2650.18	Brecciated Dedolomite	- 9.90	- 1.94	6638	145	157	8
13	128	Yican 1	2680.24	Brecciated Dedolomite	- 9.62	- 0.52	10,375	81	189	2
14	132	Yican 1	2681.13	Brecciated Dedolomite	- 10.28	- 1.73	6913	79	195	11
		Yican 1	2688.86	Max	- 8.69	- 0.45	15,200	225	195	18
		Yican 1	2637.61	Min	- 11.97	- 5.14	2875	33	101	1
		Yican 1	2664.26	Average	- 10.14	- 1.97	8588	116	150	8
15	170	Yican 1	2731.40	Dolomite	- 10.05	- 2.77	2963	83	152	20
16	174	Yican 1	2731.83	Dolomite	- 8.85	- 2.43	3075	70	143	21
17	90	Yican 1	2640.90	Dolomite	- 11.79	1.20	4906	201	99	20
18	92	Yican 1	2641.44	Dolomite	- 8.04	- 0.59				
19	158	Yican 1	2725.52	Dolomite	- 7.88	- 3.39	4975	104	164	17
20	178	Yican 1	2806.84	Dolomite	- 8.34	0.42	2100	40	106	22
21	182	Yican 1	2807.57	Dolomite	- 7.97	0.37	2325	37	122	21
22	186	Yican 1	2808.49	Dolomite	- 10.32	- 1.93				
23	190	Yican 1	2814.86	Dolomite	- 8.77	- 0.27	1025	36	83	23
24	194	Yican 1	2815.81	Dolomite	- 8.63	- 0.35	3138	31	72	21
25	198	Yican 1	2816.60	Dolomite	- 8.53	- 0.13	1513	27	222	23
26	202	Yican 1	2817.46	Dolomite	- 8.81	- 0.21	1338	30	142	21
27	206	Yican 1	2818.31	Dolomite	- 8.42	- 0.09	825	27	67	23
28	214	Yican 1	2819.91	Dolomite	- 8.99	- 0.14	663	30	79	22
29	218	Yican 1	2820.71	Dolomite	- 8.53	- 0.15	988	24	75	23
30	222	Yican 1	2821.46	Dolomite	- 8.75	- 0.01	588	25	76	24
31	226	Yican 1	2822.26	Dolomite	- 8.56	- 0.04	538	26	75	20
32	228	Yican 1	2823.70	Dolomite	- 7.80	- 0.14				
33	232	Yican 1	2945.62	Dolomite	- 7.10	- 3.11				
34	234	Yican 1	2949.94	Dolomite	- 8.64	-0.88	4525	83	86	23
35	238	Yican 1	2950.72	Dolomite	- 8.37	- 0.77	3938	72	70	20
36	242	Yican 1	2951.52	Dolomite	- 8.52	- 0.92	6288	75	122	22
37	246	Yican 1	2953.57	Dolomite	- 8.21	-0.70	4188	67	68	20
38	250	Yican 1	2954.83	Dolomite	- 8.74	- 0.46	3388	85	180	22
39	254	Yican 1	2955.85	Dolomite	- 8.17	- 0.41	2575	59	61	20
40	258	Yican 1	2957.00	Dolomite	- 8.21	- 0.62	4288	78	70	22
41	262	Yican 1	2958.29	Dolomite	- 9.46	- 0.83	2825	81	157	21
42	266	Yican 1	2959.44	Dolomite	- 9.26	- 0.90	2425	88	120	23
43	270	Yican 1	2961.34	Dolomite	- 7.95	- 1.54	4888	84	69	20
		Yican 1	2961.34	Max	- 7.10	1.20	6288	201	222	24
		Yican 1	2640.90	Min	- 11.79	- 3.39	538	24	61	17
		Yican 1	2844.69	Average	- 8.68	- 0.75	2811	63	107	21
44	162	Yican 1	2728.72	Anhydrite dolomite	- 9.37	- 2.39	3750	71	7363	17

Table 3 (continued)

	Samples	Well	Depth (m)	Lithology	δ <sup>18</sup> O(‰)	δ <sup>13</sup> C(‰)	Fe(ppm)	Mn(ppm)	Sr(ppm)	MgO(%)
45	166	Yican 1	2730.60	Anhydrite dolomite	- 8.70	- 2.27	2550	70	291	18
46	286	Yican 1	2967.34	Anhydrite dolomite	- 7.63	- 2.41	7413	85	189	16
47	294	Yican 1	2969.38	Anhydrite dolomite	- 7.88	0.48	2200	52	209	21
48	298	Yican 1	2970.90	Anhydrite dolomite	- 7.59	-0.81	1663	44	877	14
49	302	Yican 1	3120.15	Anhydrite dolomite	- 8.21	- 3.26	8013	99	1635	10
50	306	Yican 1	3121.30	Anhydrite dolomite	- 7.77	- 3.30	9675	117	1320	10
51	310	Yican 1	3122.59	Anhydrite dolomite	- 7.84	- 3.25	17,450	139	832	11
52	316	Yican 1	3124.53	Anhydrite dolomite	-8.07	- 3.92	19,413	258	264	15
53	320	Yican 1	3125.87	Anhydrite dolomite	- 8.02	- 4.90	23,113	258	118	12
54	324	Yican 1	3127.16	Anhydrite dolomite	- 7.50	- 0.90	9913	411	137	19
55	326	Yican 1	3127.83	Anhydrite dolomite	- 7.55	- 0.83	10,963	418	269	17
		Yican 1		Max	- 7.50	0.48	23,113	418	7363	21
		Yican 1		Min	- 9.37	- 4.90	1663	44	118	10
		Yican 1		Average	-8.07	- 2.30	10,063	177	1499	15
56	274	Yican 1	2962.18	Dolomitic anhydrite	- 7.98	- 2.28	2088	26	1197	4
57	278	Yican 1	2964.34	Dolomitic anhydrite	- 7.55	-0.40	1427	29	870	8
58	282	Yican 1	2965.97	Dolomitic anhydrite	- 7.77	- 2.97	7288	58	984	5
59	290	Yican 1	2968.52	Dolomitic anhydrite	- 7.61	- 2.99	4425	38	1036	4
		Yican 1		Max	- 7.55	-0.40	7288	58	1197	8
		Yican 1		Min	- 7.98	- 2.99	1427	26	870	4
		Yican 1		Average	- 7.74	- 2.00	3990	39	1025	5
60	Yi13	Yi13	1356.02	Crystalline dedolomite	- 10.30	- 2.00				
61	Yi13	Yi13	1356.83	Crystalline dedolomite	- 9.20	- 6.90				
62	Yi13	Yi13	1362.32	Crystalline dedolomite	- 8.90	- 6.50				
63	Yi13	Yi13	1362.88	Crystalline dedolomite	- 8.70	- 5.60				
64	Yi13	Yi13	1367.88	Crystalline dedolomite	-8.00	- 6.70				
65	Yi13	Yi13	1369.92	Crystalline dedolomite	- 8.00	- 6.50				
66	Yi13	Yi13	1372.50	Crystalline dedolomite	- 8.50	- 4.90				
67	Yi13	Yi13	1384.57	Crystalline dedolomite	- 8.30	- 6.30				
68	Yi13	Yi13	1385.82	Crystalline dedolomite	- 10.00	- 7.30				
69	Yi13	Yi13	1386.96	Crystalline dedolomite	- 8.60	- 7.50				
70	Yi5	Yi5	2305.96	Crystalline dedolomite	- 9.29	- 6.16				
71	Yi5	Yi5	2306.69	Crystalline dedolomite	-8.08	- 5.32				
72	Yi5	Yi5	2305.71	Crystalline dedolomite	- 8.38	- 6.77				
				Max	- 8.00	- 2.00				
				Min	- 10.30	- 7.50				
				Average	- 8.84	- 5.86				

are quite different from those of this study. In previous study, the replaced rhombic calcite is composed of polycrystalline calcite mosaics, or the unaltered dolomite core is surrounded by multi-crystalline calcite at the rims or along the rim/core interfaces, whereas, in the present study, the replaced calcite core shows a homogeneous fabric with rims or rim/core interface filled with multi-crystalline dolomite instead of calcite. Previous case studies indicate that the dedolomitization is a process involving dolomite dissolution so that an open pore is formed, which is then cemented by euhedral calcite. The mosaic texture of the crystals in the end product, therefore, indicates an origin through cementation rather than direct replacement (Evamy 1967). Dedolomitization in brecciated dedolomite of this study, however, is a direct replacement process, which changes the rhomb composition but not destroying its original internal crystalline structure.

Previous studies reveal that the calcite solubility is 1.13 times of dolomite (Lu and Zhang 2007). Liu and Wolfgang (2007) also concluded that the primary solubility of calcite is  $2.5 \sim 7.5$  times of dolomite solubility according to the karst dynamics experiment. Experiments by Huang et al. (1993) show that calcite is much more readily dissolved than



**Fig.9** Geochemistry characteristics of brecciated dedolomite, dolomite, anhydrite dolomite and dolomitic anhydrite of the Majiagou Formation from Yican 1 well in the southeastern Ordos Basin. **a** Cross-plot of  $\delta^{18}$ O vs.  $\delta^{13}$ C values; **b** Cross-plot of Fe vs. Mn con-

tent; **c** Cross-plot of Sr content vs. depth for brecciated dedolomite and dolomite; **d** Cross-plot of Sr content vs. depth for brecciated dedolomite, dolomite anhydrite dolomite and dolomitic anhydrite

 Table 4
 Simulation of different texture dolomite dissolution with or without gypsum under surface environment (Huang et al. 1993; Fang and Hou 2013)

Lithology		Conidition	Ca release rate/	Mg release rate/	Total Ca and Mg	ω(Ca)/	
	ω(Ca)/ω(Mg)		(mg/L·h)	(mg/L·h)	release content/(mg/L)	ω(Mg) in solution	
Oolitic dolomite	1.98	Without gypsum	0.141	0.042	61.24	3.15	
		With gypsum	0.136	0.076	64.27	1.62	
Fine-powder dolomite	1.86	Without gypsum	0.141	0.045	58.05	2.60	
		With gypsum	0.139	0.075	65.72	1.62	

dolomite under acidic near-surface condition; even though dissolved  $CaSO_4$  is added. He simulates a karst dissolution condition under 40 °C at normal pressure with 0.0596 mg/L carbonate acid as solvent. The experiment reveals that under the influence of high concentration of calcium ions when dissolved  $CaSO_4$  is added to the solvent for dolomite dissolution, the release rates of Mg<sup>2+</sup> increases, while the Ca<sup>2+</sup> release is restrained (Fang and Hou 2013). The  $\omega_{(Ca)}/\omega_{(Mg)}$  ( $\omega$  represents the ionic content) in fluid dissolved from oolitic dolomite and fine-powder dolomite is

3.15 and 2.60, respectively without  $CaSO_4$  added. When the  $CaSO_4$  is added, the two values reduce to 1.62 and 1.62, respectively (Table 4). The experiment results indicate that under the influence of high concentration of calcium ions when dissolved  $CaSO_4$  is added to the solvent for dolomite dissolution, the release rates of  $Mg^{2+}$  increases, while the  $Ca^{2+}$  release is restrained. Dedolomitization is therefore an  $Mg^{2+}$ -releasing process, which changes the dolomite composition but not its fabric. The zoned dedolomite (Fig. 5b) and dolomite (Fig. 5c) fabric may indicate a probable



Fig. 10 A paragenetic sequence plot of carbonate in the Ordovician Majiagou Formation in southeastern Ordos basin

dedolomitization process. The ghost of rhombic dedolomite (Fig. 5b) seems like the erosional relic indicating a progressive dedolomitization process. The gradual  $Mg^{2+}$  content decrease from the dark gray cores to the light gray rims (Fig. 5d) confirm the above conclusion.

Because of the higher solubility of calcite than dolomite, it is to be expected that calcite will dissolve in preference to dolomite. Hence the dissolution pores are more developed in dedolomite crystals than in dolomite matrix (Fig. 4j, k). The rhombohedral pores are probably developed through the selective leaching of dedolomite rather than dolomite (Fig. 4l). The cementation documented in previous studies seems, therefore, to have concluded a rather complex diagenetic history involving dolomitization, dedolomitization, selective leaching of dedolomite or another phase of dolomitization, and finally calcite cementation (Evamy 1967; Sibley 1980; Nader et al. 2008) (Fig. 10).

It is suggested that the dedolomitization process might be an opposite of dolomitization (Huang et al. 1993). Because protodolomite is thought to be a possible precursor of stoichiometric dolomite (Oomori et al. 1988; Liu et al. 2019), it must be an intermediate product of dedolomitization. Under conditions of a fluid with high  $Ca^{2+}/Mg^{2+}$  ratio, ordered dolomite firstly transforms to proto-dolomite that is rich in  $Ca^{2+}$ , and then into calcite. The dissolution of Ca sulphate promotes dedolomitization by increasing the Mg ion release and constraining the Ca ion release. As the dolomite transforms into calcite, the anhydrite-supplied Ca ion is removed and Mg ion is released into the solution, thereby leading to a relative high Mg<sup>2+</sup> concentration in the vicinity of dedolomite and local re-precipitation of dolomite at the rims or along rim/ core interfaces. The dolomite crystals formed after dedolomitization are authigenic, and therefore, show euhedral to subhedral, differentially oriented fabric. They seem to preferentially occupy dedolomite rim or rim/core interface-probably within sites of abundant fluid inclusions.

# Origin of crystalline limestone (Crystalline dedolomite)

Another common microfabric of dedolomitization is characterized by anhedral calcite crystals, forming crystalline dedolomite with numerous poikilotopic inclusions or aggregates of dolomite. The medium to coarsely crystalline dedolomite with varying amount of dolomite in the Majiagou Formation are similar to those interpreted as the dedolomitization in the Winnipegosis Formation by Fu et al. (2008) and numerous other studies. Various evidences suggest that the crystalline dedolomite in the Majiagou Formation has formed by dedolomitization followed by recrystallization of very fine to fine dedolomite. Dolomite masses within the crystalline dedolomite horizon preserve the relic textures of precursor dolomite and display corrosion fabrics (Fig. 6g, h), strongly suggesting that dolomitization pre-dated the formation of medium to coarsely crystalline dedolomite. The patchy dolomite relics enclosed within the calcite crystals cores are dolomicrite aggregates, indicating that the precursor dolomite has very fine to microcrystalline texture. The crystalline dedolomite is massive and mosaic contact (Fig. 6g, h), indicating that its formation was followed by recrystallization of very finely to finely crystalline dedolomite. The textures of several relic dolomite crystals encased within single to multiple calcite crystals are probably caused by the growth of calcite crystals into the void space away from nucleation sites on the dissolving dolomite crystals (Stoessell et al. 1987; Fu et al. 2008). All the above petrographic observations suggest that the crystalline limestone consisting of massive calcite mosaics in the upper part of the Majiagou Formation has formed by the calcitization of dolomite and is best interpreted as dedolomite. Different from the direct replacement of rhombic dedolomite, the crystalline dedolomite involves a two-step process of dolomite dissolution and calcite precipitation. Because calcite has a higher solubility than dolomite, the calcite must have precipitated from a different solution at a different time (James et al. 1993). As a result of intense dedolomitization and recrystallization, the precursor dolomite is replaced by fine to medium calcite and then recrystallized to a medium to coarse calcite mosaic of newly formed calcite crystals with very low intercrystalline porosity.

Brecciated dedolomite is more common in Yican 1 well, whereas crystalline dedolomites are better developed in Yi 5, Yi 9, Yi 13 and Yi 16 wells. This phenomenon is partly attributed to the fact that the precursor dolomite in Yi 5, Yi 9, Yi 13 and Yi 16 wells are mostly very fine to microcrystalline fabric, which is more soluble than crystalline dolomite because of larger surface area. Furthermore, Yi 5, Yi 9, Yi 13 and Yi 16 wells are tectonically lower than Yican 1 well, where karst water saturated with respect to  $CaCO_3$  is more concentrated and facilitate the dissolution of dolomite and the precipitation of calcite.

Because of the relative high solubility of calcite formed by dedolomitization, the dissolution pores are more common in crystalline dedolomite (Fig. 6a, d). If a large section of dolomite of  $Ma_5$  has been dedolomitized, continuous dissolution pores may be formed in the longitudinal direction influenced by the heterogeneity of rock structure (Fig. 6e, f). In dolomite and calcite intervals, because of the fast dissolution rate of dedolomite, the ground water flow and dissolution are more and more concentrated in the dedolomite layers. Differential dissolution becomes more and more obvious, and a series of layered karst pores are then often formed in the dedolomite layer in the longitudinal direction (Fig. 6e, f). However, the dissolution pores are always partly or totally cemented by sparry calcite and/or dolomite (Fig. 6e, f).

# Paragenesis of dedolomitization

Dedolomitization occurs in a wide range of diagenetic environments, including the surface environment (Chilingar 1956), early burial environment (De Groot 1967), and deep burial environment (Stoessell et al. 1987). Most case studies conclude that an overwhelming majority of dedolomites form from meteoric waters in the near-surface environment (Ronchi et al. 2004; Nader et al. 2008; Reuning et al. 2009; Schoenherr et al. 2018). In the present study, the dedolomites are only observed in areas where the evaporite deposits are removed or residue-anhydritic breccias are present, suggesting that dedolomitization of the Majiagou Formation is probably related to the dissolution of anhydrite by fresh water, which is supported by the petrographic features and geochemical data of the dedolomite.

The dedolomites are restricted to the uppermost 100 m below a regional disconformity (Li et al. 2008; Yang et al. 2011), defined by clay minerals and karst dissolution-collapse breccia (Fig. 2d), which indicates its close relationship with erosional unconformities. The dedolomites are often found to be cross-cut by fractures (Fig. 2h). It could be argued that dedolomitization are pre-dated the fractures.

The  $\delta^{13}$ C values of Ordovician contemporaneous seawater varies between -2% - 2% (Lohmann 1988). The crystalline dedolomite in the Majiagou Formation is characterized by negative  $\delta^{13}$ C values (average of -5.86‰), much lower than both that of dolostone (average of -0.75%) and those of brecciated dedolomite (average of -1.79%). The  $\delta^{13}$ C value of dolomite is within the Ordovician contemporaneous seawater, indicating that the dolomite is formed in marine fluid. The strongly negative  $\delta^{13}$ C values of the crystalline dedolomite could be interpreted as resulting from fresh groundwater enriched in soil-derived CO<sub>2</sub> (Nader et al. 2008; Liu et al. 2017a) and/or bacterial sulphate reduction (BSR) (Sanz-Rubio et al. 2001; Fu et al. 2008). In combination with the negative  $\delta^{18}$ O value of crystalline dedolomite, the negative  $\delta^{13}$ C values of the crystalline dedolomite cannot be from bacterial sulphate reduction, as SRB will preferentially consume <sup>16</sup>O thus often enrich the  $\delta^{18}$ O value. Therefore, the negative  $\delta^{13}$ C values of the crystalline dedolomite are interpreted to be resulted from fresh groundwater involvement. The  $\delta^{13}$ C values of brecciated dedolomite (average of -1.97%) is between that of crystalline dedolomite (average of -5.86%) and dolomite (average of -0.75%), most likely due to varying calcite (dedolomite) content.

The  $\delta^{18}$ O value of Ordovician carbonate varied from -9.0% to -5.4%, average of -7.6% (Shields et al. 2005). The crystalline dedolomite from Yi 5 and Yi 13 wells, and the brecciated dedolomite from Yican 1 well shows relatively negative  $\delta^{18}$ O values (average of -8.84% and -10.14‰, respectively) with respect to dolomite (average of -8.68%). The negative O-isotope may be caused by meteoric condition or elevated temperatures, which drive the thermal fractionation of diagenetic carbonate (Hardie 1967; Choquette and James 1990; Liu et al. 2017b). The crystalline dedolomite and brecciated dedolomite shows a wider range of  $\delta^{13}$ C values but a narrower range of  $\delta^{18}$ O values (Fig. 9a), which is a distinctive feature of meteoric diagenesis (Lohmann 1988; Sanz-Rubio et al. 2001; Fu et al. 2008). The general isotope pattern of dedolomites in the Majiagou Formation is similar to that reported by Kenny (1992), Arenas et al. (1999) and Nader et al. (2008), who also interpreted the similar isotope composition in dedolomite to result from an influx of meteoric water.

The High Sr in the brecciated dolomite (average of 150 ppm) is because that it formed during dissolution of evaporites, which are naturally high in Sr (Carlson 1983). Stewart (1968) measured  $Sr^{2+}$  contents up to 2000 ppm in marine gypsum and 5000 ppm in anhydrite, which compares to the upper limit of the  $Sr^{2+}$  contents of anhydritic dolostone and dolomitic anhydrite of Yican 1 (maximum of 1499 ppm and 1025 ppm, respectively). The value is significantly higher than those in dolomite, average of 107 ppm. Because dedolomitization is driven by the sulphate solution as mentioned above, with the dissolution of gypsum and

anhydrite, the  $Sr^{2+}$  is released into solution and incorporated into the dedolomites that formed in such fluids with high  $Sr^{2+}$  content. However, it cannot be excluded that the dedolomite or dolomite samples may be polluted by anhydrite resulting in the elevated Sr content.

The dedolomites often show brownish or reddish color in hand specimen (Fig. 2e). The Fe and Mn content of brecciated dedolomite (average of 8588 ppm and 116 ppm, respectively) is much higher than that of dolomite (average of 2811 ppm and 63 ppm, respectively) (Table 3, Fig. 9b). Due to significant differences of Fe and Mn content from continental surface fluid (such as meteoric water) and marine fluid, and because of variable range of partition coefficient and dynamic effect of the two element in carbonate minerals, the carbonate affected by meteoric water commonly has higher Fe and Mn content (Huang et al. 2010). Karstification is the most important geological process of carbonate modification by atmospheric water, resulting in relatively high Fe and Mn content in the carbonate formed near the unconformity surface. Therefore, the content of Fe and Mn in carbonate becomes one of the methods to identify the paleo-karstification and determine the effective depth of atmospheric water during the karst process. During uplift, subaerial exposure and karstification through the circulation of meteoric water, Fe<sup>2+</sup> is incorporated into the dedolomite and oxidized in the open system to form hematite occurring as thin coating on the crystals. Many dedolomites are, therefore, heavily stained with iron oxides and hydroxides and exhibit brownish or reddish rock colors. All of the above features reveal that the dedolomites formed under the influence of relatively fresh water during subaerial exposure in a surface environment.

#### Impact on reservoir quality

It is most commonly suggested that the dedolomitization process will reduce the reservoir porosity (Li 1998; Hu et al. 2008; Shi et al. 2010; Tone et al. 2018). However, numerous studies have also shown an enhancement of reservoir quality by dedolomitization (e.g., Schmidt 1965; Evamy 1967; Braun and Friedman 1970; Frank 1981; Qin and Yang 1997). Theoretically, 12.9% void volume will be produced by the dolomitization reaction involving 1 mol of dolomite (64.4 cm<sup>3</sup>), which is 12.9% less than 2 mol of calcite by volume (73.9 cm<sup>3</sup>) ((73.9–64.4)/73.9\*100%) (Ayora et al. 1998; Hallenberger et al. 2018). The reservoir quality of dolostones is therefore significantly higher than that of limestones, such as in the Feixianguan Formation in the Sichuan Basin (Jiang et al. 2018) and the Ordovician carbonate in the Tazhong area of the Tarim Basin (Jia et al. 2016). Similarly, the solid volume will increase by 14.7% during the replacement of dolomite by calcite ((73.9-64.4)/64.4\*100%). Therefore, in brecciated dedolomite of Yican 1 well, the micropores are

well developed in dolomite matrix, but rarely observed in dedolomite crystals (Fig. 5b, c) due to the porosity shrinkage effect. The overall reservoir quality is, therefore, reduced in the dedolomitization process.

Whereas, because of higher solubility of calcite than dolomite, the dissolution vugs and pores are expected to be more developed in dedolomite (Fig. 4j, k), even form rhombohedral pores (Fig. 41). The pores from dedolomite dissolution are commonly isolated and less abundant, which contributes little to the overall porosity of the horizon. The pore space of Yican 1 well mainly originates from anhydrite dissolution pores and fractures, which mainly occurs in the uppermost 10 m of the Majiagou Formation. The crystalline dedolomite, however, are formed by dissolution/precipitation and recrystallization of precursor dolomicrite. As a result, the remaining pores may be occluded as long as excess calcium is supplied into a closed diagenetic system to enable dedolomitization, which will reduce the reservoir porosity. The dedolomitization will change both the composition and fabric of the precursor dolomite. The composition is changed from dolomite into calcite, whereas the fabric is changed from micrite crystals to medium-coarse crystals, even up to very coarse crystals. The calcite crystals are massive, anhedral and for a tight mosaic to make the rock low porosity. Yi 16 well, where the crystalline dedolomite is present shows low porosity of 1.04% (Table 2). Although the pores form by dissolution are developed in the crystalline dedolomite, the pores are always isolated in distribution (Figs. 6a, d), which is poorly connected for lack of fracture. Most of pores are cemented with calcite and/or dolomite, which occlude the pore space to reduce the reservoir porosity (Figs. 6e, f). Dedolomitization, therefore, can be a spatially very pervasive process, which has reduced reservoir quality, not only on a local but also on a regional scale.

#### Major controlling factors on the dedolomitization

# (1) $Ca^{2+}/Mg^{2+}$ ratio in the diagenetic fluid

Experiments have shown that the  $Ca^{2+}/Mg^{2+}$  ratio and fluid rate are the major factors that control dedolomitization (De Groot 1967). Dedolomitization in the study area occurs during subaerial exposure related to a regional unconformity. Based on the above discussion, the dissolution of Ca sulphate will increase the Mg<sup>2+</sup> release rate, while reducing the Ca<sup>2+</sup> release rate. In an open system, the produced Mg<sup>2+</sup> will be removed, which is favorable for the reaction to proceed. The dedolomitization reactions will be faster with higher Ca<sup>2+</sup>/Mg<sup>2+</sup> values.

A continuous profile across four wells in the southeast of the Ordos Basin is shown by the vertical distribution of well log correlations in Fig. 11. The top of the Taiyuan Formation is positioned at the same depth to compare the pre-Carboniferous paleo-topography. When the overlying Taiyuan and Benxi Formation is thin, the pre-Carboniferous paleo-topography is relatively high (e.g., Yican 1 and Yi 8 wells). When the Taiyuan and Benxi Formation is thick, the pre-Carboniferous paleo-topography is relative low (Yi 9 and Yi 16 well) (Fig. 11). An overall trend



Fig. 11 Profile across four wells with vertical distribution of the well log correlations indicating an increase of limestone proportion from the topographically higher towards lower parts of the basin (Modi-

fied from Liu et al. 2011). The limestone are all crystalline dedolomites in member  $Ma_5$ . Only the member  $Ma_6$  is composed of primary limestone

of increasing dedolomite proportion of  $Ma_5$  member is observed from the topographically higher towards lower parts of the basin (Liu et al. 2011). The lower part of the basin is the converging location of the karst fluid, where bears a higher  $Ca^{2+}/Mg^{2+}$  ratio and is oversaturated with respect to calcite. The rock-fluid reaction lasts for a considerable duration when  $Ca^{2+}/Mg^{2+}$  are high promoting the replacement of dolomite by calcite and subsequently recrystallization. It's therefore suggested to implement drilling in tectonic highland and avoid low locations in paleo-topography.

(2) Dolomite stoichiometry

Dedolomitization often begins within the core and progresses outwards toward the crystal edges (Fig. 5), because the cloudy cores of dolomite has stoichiometry about 53%- $56 \text{ M\% CaCO}_3$ , which is believed to be less stable. While as the clear rims are nearly to be stoichiometric with  $50\% \sim 53 \text{ M\% CaCO}_3$ , which are less prone to alteration. Therefore, the dedolomitization is partially controlled by their stoichiometry.

(3) Fracture development degree

Numerous studies have demonstrated that fractures acted as the preferred flow pathways for dedolomitizing fluids (Zeeh et al. 2000; Reuning et al. 2009; Schoenherr et al. 2018). Dedolomitization is much more developed in fractures or adjacent region in the southeastern Ordos basin. Tectonic in Caledonian movement is the root cause for the formation of a massive fracture network within the dolostone unit, which provides more paleo-conduits for the dedolomitizing fluid and causes the pervasive and large scale dedolomitization.

# Conclusions

(1) Two types of dedolomites were recognized in the upper 100 m of the Majiagou Formation, namely brecciated dedolomite and crystalline dedolomite.

(2) The brecciated dedolomite is mainly composed of dark gray matrix and light gray clasts, in which calcitized dolomite crystals are scattered in micritic dolostone matrix or form crystal aggregates as the clasts of breccia. Four types of calcitized dolomite are identified in brecciated dedolomite, including the dolomite crystals distributed at the rim, or along the rim/core interface, or scattered in the calcitized dolomite crystal, as well as the partial or complete dissolution of the crystal. The brecciated dedolomite is formed by  $Mg^{2+}$  release process, instead of a dolomite dissolution/ calcite precipitation process.

(3) Crystalline dedolomite is characterized by xenotopic, medium to coarsely, crystalline calcite mosaics with variable amount of dolomite impurities in crystal cores, which are considered to be formed by dolomite dissolution and calcite precipitation followed by recrystallization. As a result of intense replacement and recrystallization, the precursor dolomite is replaced by finely to medium calcite and then recrystallized to medium to coarsely calcite mosaic of newly formed calcite crystals with very low intercrystalline porosity.

(4) The dedolomites often occur with clay minerals and karst dissolution-collapse breccia, and are fractured with brownish or reddish color in hand specimens, strongly suggesting their close relationship with the erosional unconformities.

(5) Geochemical evidence further supports that the dedolomites formed in fluids of meteoric origin when uplifted in the Caledonian orogeny in lower Paleozoic, which are depleted in  $\delta^{18}$ O and  $\delta^{13}$ C values, and enriched in Fe and Mn. Because the dedolomite mainly occurs in anhydrite-dolomite intervals, we suggest that the dedolomitization is driven by evaporite dissolution when subject to exposure and erosion by meteoric water. The high Sr contents likely derive from the anhydrite dissolution, which has a high Sr content.

(6) The brecciated dedolomites are common in Yican 1 well, whereas the crystalline dedolomite mainly occurs in Yi 5, Yi 9, Yi 13 and Yi 16 wells, which is attributed to the precursor texture and tectonic locations. The precursor dolomites in Yi 5, Yi 9, Yi 13 and Yi 16 wells are mostly very finely to microcrystalline fabric, which is more soluble than crystalline dolomite because of larger surface area. The pre-Carboniferous paleo-topography of Yi 5, Yi 9, Yi 13 and Yi 16 wells is relative low, where karst water with saturated  $CaCO_3$  is more concentrated and facilitates the dissolution of dolomite and the precipitation of calcite. It's therefore suggested to implement drilling in tectonic highland and avoid low location in paleo-topography.

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#### **Declarations**

Conflict of interest The authors declare no conflict of interest.

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