

# Biomass of a Psychrophilic Fungus as a Biocatalyst for Efficient Direct Esterification of Citronellol

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### Abstract

A biomass-bound lipase from psychrophilic *Chrysosporium pannorum* A-1 is an efficient biocatalyst for direct esterification of  $\beta$ -citronellol and acetic acid in an organic solvent. The biomass is effectively produced by fungal submerged culture at 20 °C, which results in lower energy consumption during the production of biocatalyst. Supplementation of the culture medium with calcium carbonate together with olive oil contributed to a significant increase in the active biomass of mycelium in one batch culture and increased the efficiency of the biocatalyst. Biomass-bound lipase showed high catalytic activity in a broad temperature range of 30–60 °C and stability up to 70 °C. A maximum molar conversion value of 98% was obtained at 30 °C in *n*-hexane using a 2:1 alcohol-to-acid molar ratio and 3% w/v of the biocatalyst within 24 h. The high equimolar concentration of the substrates (200 mM) did not have an adverse effect on mycelial biomass activity. Dry mycelium of *C. pannorum* is a promising biocatalyst for large-scale biosynthesis of citronellyl acetate, given its low-cost production, high activity at low temperatures, and reusability in a minimum of seven 24-h biocatalytic cycles.

Keywords Biomass · Biomass-bound lipase · Esterification · Chrysosporium · Citronellyl acetate · Psychrophiles

### Introduction

Esters of terpene alcohols, especially acetate esters, are valuable compounds in the flavor and fragrance industry, which was valued at 24.9 billion USD in 2014, with a predicted growth rate from 5 to 6% per year [1-3]. These compounds can be obtained via extraction from plant sources, chemical synthesis, or enzymatic biotransformation. It is difficult to obtain a large quantity of terpene esters from plant material due to the small amount of these compounds in essential oils and the hydrolysis of natural esters [4]. Chemical synthesis is

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<sup>3</sup> Department of Biophysics, Faculty of Environmental Biology, University of Life Sciences in Lublin, Akademicka 13, 20-950, Lublin, Poland characterized by formation of a racemic mixture of products and often generates adverse environmental conditions [5]. A good alternative for the preparation of "natural" terpene esters is the enzymatic esterification of terpene alcohols and carboxylic acids using lipases. Compounds produced in this way can be referred to as "natural" in accordance with the provisions of the European Union and the USA [6]. However, isolated and purified enzymes are quite costly even with large-scale production.

Lipases (EC 3.1.1.1) are an important class of serine hydrolytic enzymes. They are widely used in food, detergent, textile, pharmaceutical, leather, and cosmetic industry [7, 8]. They are synthesized by all living organisms: plants, animals, and microorganisms. Lipases have broad substrate specificity while maintaining their regioselectivity and stereoselectivity as well as a catalysis mechanism in which no cofactors are involved [9]. Lipases catalyze the hydrolysis of ester bonds in the water environment and a reverse reaction in an organic solvent and water deficiency conditions [10, 11]. This property is suitable for the biocatalytic production of terpene alcohol, glucose, or flavonoid esters [12–14]. Bacterial and fungal lipases, most often used in this bioprocess, are usually secreted outside cells [15]. Therefore, to be suitable for multiple applications, the enzyme from the post-culture liquid must be concentrated, purified, and often immobilized on an appropriate carrier. These operations are usually associated with high costs of obtaining a pure preparation, substantial energy consumption, and loss of biocatalytic activity. An alternative source of an immobilized biocatalyst for esterification reactions is the intracellular lipase linked to the cellular hydrophobic membrane [16–18]. It has the basic features of immobilized biocatalysts, including stabilization of the protein structure, which increases their thermostability and resistance to denaturing factors, and easy separation from the reaction mixture after the bioprocess [19, 20].

From the point of view of environmental protection, the biocatalytic esterification of alcohols with carboxylic acids is the best way to obtain esters, as only water is the by-product of the reaction. However, the water formed during the process changes the equilibrium of the reaction towards hydrolysis [21]. Also, according to some reports, free acetic acid has a negative effect on lipase activity [22-24]. Therefore, despite the availability of many commercial lipases, there is still a need for new sources of lipases that will be efficient biocatalysts of direct esterification of various alcohols with acetic acid [25]. The use of biomass-bound lipase supports the binding of the water formed in the esterification process, and it is not necessary to use additional water-binding agents, e.g., molecular sieves [26]. Despite these advantages and relatively high activity of the biocatalyst, a small amount of biomass-bound lipase is obtained in one batch culture, which makes it not profitable when used on a large scale [16, 17, 27].

The use of psychrotrophic microorganisms as biocatalysts offers numerous advantages. The high microbial growth rates and the high enzymatic activities in the temperature range of 10-25 °C may shorten the process times and minimize undesirable chemical reactions that may occur at higher temperatures [28], thus constituting considerable progress towards reduction of energy consumption.

The present paper reports for the first time that lyophilized biomass of psychrophilic fungus *Chrysosporium pannorum* A-1 is an efficient biocatalyst for esterification. The fungal culture conditions were successfully optimized to obtain a large amount of catalytically highly active biomass-bound lipase in one batch culture. The reaction conditions for esterification of  $\beta$ -citronellol and acetic acid were also optimized, and the possibility of using the biocatalyst in several reaction cycles was checked. The effect of the enzyme storage temperature on retention of the lipolytic activity of the biomass after 6 months and a year of storage was determined as well.

Tributyrin (97%) and calcium carbonate ( $\geq$ 99%) were

purchased from Sigma-Aldrich, USA. (S)-(-)-β-citronellol

### **Materials and Methods**

#### Chemicals

 $(\geq 98\%)$  was obtained from Merck, Germany. Acetic acid  $(\geq 99.5\%)$ , n-hexane, cyclohexane, hexadecane, acetonitrile, 1,4-dioxane, chloroform, ethyl acetate, 1-propanol, N-methyl-2-pyrrolidone (NMP), dimethylacetamide (DMAC), dimethylformamide (DMF), and tetrahydrofuran at a purity of  $\geq 99\%$  were obtained from POCH, Poland.

### **Microorganisms and Media**

Mesophilic *Rhizomucor miehei*, *Penicillium rhemazol*, and psychrophilic *Penicillium chrysogenum* 9' were used as reference strains with lipolytic activity [29–31]. *Chrysosporium pannorum* A-1 was shown to have lipolytic activity and was used in further studies. Fungal cells were isolated from soil samples collected in the Arctic tundra from West Spitsbergen [32, 33]. Prior to the experiments, the cells were maintained on malt agar slants at 4 °C and subcultured every month.

Preliminary determination of lipolytic activity was carried out on solidified agar medium (SM) composed of 0.5% butter, 0.1% K<sub>2</sub>HPO<sub>4</sub>, 0.03% MgSO<sub>4</sub>×7H<sub>2</sub>O, 0.01% CaCl<sub>2</sub>×6H<sub>2</sub>O, 0.2% (NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>, and 0.01% NaCl. Immediately before pouring onto Petri dishes, 0.15 mL of an alcoholic solution (0.04%) of bromothymol blue was added as a pH indicator to the tubes with dissolved agar (at a temperature above 60 °C) to obtain a clearly green color.

In the further step of the study, liquid medium (A) composed of 1% glucose, 0.6%  $KH_2PO_4$ , 0.1%  $MgSO_4 \times 7H_2O$ , 0.4% urea, 10% olive oil, 10 mg  $FeCl_3 \times 6H_2O$ , 8 µg D-biotin, 200 µg thiamin, and 4 µg myo-inositol per liter was used. Initial pH was adjusted to 6.0 by adding 0.1 M HCl [31].

The culture conditions were optimized on the basal medium (BM) composed of 0.5% Tween 80, 0.1% yeast extract, 0.5% (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 0.1% K<sub>2</sub>HPO<sub>4</sub>, and 0.02% MgSO<sub>4</sub>×7H<sub>2</sub>O (pH=6.0, adjusted by adding 0.1 M HCl) [16]. Modified BM media were also used. They were marked as BM1 (BM + 1% olive oil), BM2 (BM + 1% olive oil + 1% glucose), BMC (BM + 1% CaCO<sub>3</sub>), BMC1 (BM1 + 1% CaCO<sub>3</sub>), and BMC2 (BM2 + 1% CaCO<sub>3</sub>). The optimal concentration of Tween 80 was determined using the BMC medium.

## **Preliminary Determination of Lipolytic Activity**

The radial method was used to determine the initial lipolytic activity [34]. Discs with a diameter of 6 mm were cut with a sterile cork borer from malt agar plates with overgrown mycelium and then aseptically point-transferred to the SM medium. Plates with the mycelial discs on the SM medium were incubated for 24 h at 20 °C (*C. pannorum* A-1 and *P. chrysogenum* 9') and at 30 °C (*R. miehei* and *P. rhemazol*). After 24 h, the diameter of the discoloration zones in each strain was measured and expressed in mm. Lipase secreted

into the medium catalyzed the hydrolysis of butter fats and the release of fatty acids, which caused a decrease in pH and changed the color from green to yellow in the zones.

### **Operation at a Shake Flask Scale**

To optimize culture conditions, 50 mL of sterile basal medium (in 300-mL Erlenmeyer flasks) with the composition depending on the experiment was uniformly inoculated with 2 mL of *C. pannorum* A-1 spore suspension (at a concentration of  $2 \times 10^5$  mL<sup>-1</sup>) and cultivated at 20 °C on a rotary shaker (150 rpm) for up to 7 days. The fungal spore suspension was prepared as an inoculum using 7-day-old spores pre-grown at 20 °C on agar slants, which were then harvested, filtered through glass wool to remove hyphal fragments, and washed twice with sterile distilled water. The spore concentration was adjusted using a hemocytometer and an Olympus BX61 light microscope.

### Determination of Extracellular and Intracellular Lipolytic Activity

After completion of the culture, the mycelial biomass was separated from the post-culture liquid by centrifugation at 10,000 g for 15 min at 4 °C. The supernatant was used as a source of extracellular lipase. Part of the mycelium from each culture (0.2–0.3 g of wet weight) was frozen for further use as a source of intracellular lipase. The rest of the biomass was dried at 105 °C for 3 h to determine the dry weight of the mycelium expressed in g/L.

The determination of extracellular lipase activity was carried out on the basis of tributyrin hydrolysis using the following method: 0.5 mL of the supernatant was mixed with 1 mL of 50 mM phosphate buffer, pH 7.0, and 1.5 mL of tributyrin in a 25-mL Erlenmeyer flask. The samples were shaken (200 rpm) for 1 h at 30 °C and then 1 mL of 96% ethanol was added to extract free fatty acids. The concentration of the free fatty acids was determined by titration with 0.0500 M KOH. Control tests with an inactivated enzyme were carried out according to the procedure described above except that the supernatant was previously boiled for 15 min. One unit of lipase activity (U) was defined as the amount of enzyme releasing 1  $\mu$ mol min<sup>-1</sup> of free fatty acids. Extracellular lipase activity was expressed as U/mL of post-culture fluid [35].

To determine intracellular lipolytic activity, frozen mycelium was disintegrated using a potter homogenizer. The biomass was homogenized in an ice bath for 3 min with an addition of 0.5 mL of 50 mM phosphate buffer. After homogenization, phosphate buffer was added to the mixture to a final volume of 5 mL and centrifuged at 7000 g for 10 min at 4 °C to separate mycelium residues. The supernatant was used to determine intracellular lipase activity with the same procedure as for extracellular lipase [36]. The activity of intracellular lipase was expressed as U per mg of dry weight of the mycelium.

### Effect of Temperature and pH on Lipase Activity

The tributyrin hydrolysis reactions were carried out in a temperature range from 15 to 50 °C following the scheme described above. To determine the effect of pH, 0.2 M citrate phosphate buffer (pH range 2.2–8.0) and 0.2 M carbonate buffer (pH range 9.2–10.7) were used. Hydrolysis was performed at 20 °C [35].

# Preparation of the Biocatalyst for the Esterification Reaction

Esterification was carried out using both the extracellular and intracellular enzymes of the biomass. The post-culture liquid used as a source of extracellular lipase was frozen in liquid nitrogen and then lyophilized for 24 h. The lyophilizate was ground to powder and used in this form in the esterification reaction. The mycelial biomass used as a source of intracellular lipase was frozen at -20 °C overnight and next lyophilized for 24 h. The freeze-dried biomass was milled. To study the influence of storage temperature on biocatalyst stability, both preparations were stored for a year at -20 °C, 4 °C, and 20 °C and further used for esterification of terpene alcohol and carboxylic acid in optimal conditions.

### **General Esterification Procedures**

The esterification reaction was carried out in sealed 25-mL Erlenmeyer flasks equipped with magnetic stirring bars. In the standard esterification procedure, citronellol and acetic acid with an equimolar concentration (50 mM) were dissolved in 5 mL of hexane containing 2% (w/v) of the fungal biocatalyst and incubated at 40 °C for 8 or 24 h [16]. During other experiments, the esterification conditions depended on the optimized factors, i.e., the temperature (20–70 °C), the amount of the biocatalyst (1.0–3.5% w/v), the equimolar concentration of substrates (10–200 mM), the molar ratio of  $\beta$ -citronellol to acetic acid (1:2; 1:3; 1:4; 1:5; 2:1; 3:1; 4:1; 5:1), and the type of the organic solvent.

The reusability of the biomass-bound lipase from *C*. *pannorum* A-1 was tested in the optimal conditions of the esterification reaction. After each cycle, the reaction mixture was removed by centrifugation (7000 g for 10 min), and the mycelium was washed three times with 25 mL of *n*-hexane. Then, the biomass was dried in a desiccator and used in the next reaction cycle with fresh substrates in the same conditions.

### **Analysis of the Biocatalysis Efficiency**

After the specified esterification time, 50  $\mu$ L of the reaction mixture was taken, dried over anhydrous sodium sulfate, and diluted tenfold with *n*-hexane. The sample was subjected to GC-FID and GC-MS analyses. The GC-FID and GC-MS analyses of terpene alcohols and their esters were conducted using a DB-5 capillary column (30 m $\times$  0.25 mm inside diameter). The separation conditions were as follows: The carrier gas was helium at a constant flow of 1 mL/min and the injection temperature was 270 °C. The oven temperature programming was as follows: 100 °C for 1 min, which was then increased at a rate of 10 °C/min to 150 °C for 1 min and then increased at a rate of 20 °C/min to 260 °C for 1 min. The esterification efficiency was expressed as molar conversion (%) of alcohol to ester after a specified time and as an initial ester formation rate ( $\mu$ mol mg<sup>-1</sup> min<sup>-1</sup>). To determine the initial reaction rate, the samples were analyzed at 0.5, 1, 1.5, 2, 3, 4, 5, 6, 8, and 12 h of the biotransformation reaction.

### **Statistical Analysis**

The experiments were carried out in triplicate. The results were subjected to the analysis of variance (ANOVA) followed by Tukey's post hoc test for multiple comparisons. The correlation between pH and extracellular and intracellular activity was analyzed based on Pearson's correlation coefficient r. The statistical analysis was carried out using Statistica for Windows version 13.1 (TIBCO Software Inc., Palo Alto, CA, USA). All results are presented

BioEnergy Research (2022) 15:399-411

as mean  $\pm$  SEM (standard error of the mean). p < 0.05 was considered statistically significant with a confidence level of 95%.

## Results

The lipolytic activity of *Chrysosporium pannorum* A-1 was initially determined in a radial test on agar plates. The comparison of the discoloration zones with those formed by strains with confirmed lipolytic activity after 24-h incubation showed the largest zones in the *C. pannorum* A-1 (20 mm) and *P. chrysogenum* 9' cultures (17 mm). Lipolytic activity was also observed in *R. miehei* with a zone diameter of 14 mm, whereas *P. rhemazol* showed no activity after 24-h incubation.

To optimize the culture conditions for high lipase production, the BM medium (with Tween 80, 0.5%) was modified by adding 1% of olive oil (as a carbon source and an additional inducer of lipase synthesis) (BM1) and 1% of olive oil together with 1% of glucose (BM2). The highest extracellular lipase activity was obtained on the BM medium on the 5th day of culture  $(1.65 \pm 0.03 \text{ U/mL})$  and on the BM1 medium on the 3rd day of culture  $(1.43 \pm 0.11 \text{ U/mL})$  (Table 1). In the case of intracellular lipase, the highest activity was observed on the BM medium on days 3 and 5 ( $0.38 \pm 0.14$  and  $0.30 \pm 0.08$  U/mg, respectively) and on the 3rd day of culture on the BM1 medium ( $0.19 \pm 0.02$  U/mg). A high positive correlation between pH and extracellular and intracellular activity was noted (r=0.73) (Table 1).

The optimal temperature and pH were determined for the extracellular lipase activity. The highest hydrolytic activity

Table 1Effect of the mediumcomposition and cultureduration on lipase activity anddry weight of *C. pannorum* A-1mycelium (g/L) and pH of thepost-culture liquid

		Medium		
	Day of culture	BM	BM1 (BM+1% olive oil)	BM2 (BM+1% olive oil+1% glucose)
Extracellular lipase activity (U/mL)	3	$1.01 \pm 0.19^{ab}$	$1.43 \pm 0.11^{ac}$	0 <sup>d</sup>
	5	$1.65 \pm 0.03^{\circ}$	$0^{d}$	$0^{d}$
	7	$0.96 \pm 0.07^{\rm b}$	$0.13\pm0.02^{\rm d}$	$0^d$
Intracellular lipase activity (U/mg)	3	$0.38 \pm 0.14^{a}$	$0.19\pm0.02^{ab}$	$0.03 \pm 0.01^{b}$
	5	$0.30\pm0.08^{ab}$	$0.08\pm0.01^{\rm b}$	$0.03 \pm 0.01^{b}$
	7	$0.22 \pm 0.04^{ab}$	$0.09 \pm 0.03^{b}$	$0.05\pm0.01^{\rm b}$
pH of the post-culture media	3	$5.81 \pm 0.13$	$3.74 \pm 0.20$	$2.32 \pm 0.05$
	5	$6.35 \pm 0.04$	$2.59 \pm 0.07$	$2.22 \pm 0.03$
	7	$6.72 \pm 0.03$	$2.55 \pm 0.07$	$2.41 \pm 0.12$
Dry weight of mycelium (g/L)	3	$1.34 \pm 0.05$	$1.65 \pm 0.18$	$5.15 \pm 0.20$
	5	$1.09 \pm 0.11$	$5.17 \pm 0.02$	$6.30 \pm 0.75$
	7	$1.05 \pm 0.20$	$6.71 \pm 1.15$	$9.08 \pm 1.06$

Means for the extracellular and intracellular lipase activity with different superscript letters denote a significant difference at p < 0.05

Α

120

100

80

60

was obtained at 20 °C, while the tributyrin hydrolysis efficiency at 15 °C dropped by approx. 50% (Fig. 1A). At 25 °C and 45 °C, the enzyme retained 94% and 61% of its initial activity, respectively. Lipase exhibited lipolytic activity in the range of pH 4.0–9.0, with an optimal value of pH 7.0 (Fig. 1B).

The addition of 1% of calcium carbonate to the media resulted in stabilization of pH, which was kept around 6-7 on the BMC1 and BMC2 media (Table 2). This resulted in emergence of lipolytic activity on the BMC2 medium and minimum three times higher dry weight of the mycelial biomass compared to the media without CaCO<sub>3</sub>. The highest extracellular and intracellular lipolytic activity and the highest dry weight of the biomass were obtained on the 4th day of culture on BMC and BMC1.

The Tween 80 concentration was optimized in the range of 0-1% (w/v). The highest activity was recorded on day 4 of the culture at the inducer concentration in the range of 0.4–0.9%. At 0.7% Tween 80,  $3.70 \pm 0.79$  U/mL and



Fig. 1 Effect of temperature (A) and pH (B) on the hydrolytic activity (U/mL) of extracellular lipase from C. pannorum A-1. Means for the determined values with different superscript letters denote a significant difference at p < 0.05

 $0.28 \pm 0.04$  U/mg of extracellular and intracellular lipase activity were recorded, respectively. The low inducer concentrations (0.1-0.3%) resulted in the highest extracellular and intracellular lipolytic activity on day 3 (Fig. 2). The pH of the liquid medium was maintained at approx. 7.0 during the culture (Table 3).

To select the form of the fungal biocatalyst for the esterification of citronellol and acetic acid, C. pannorum A-1 was cultured on the OM medium (OM - optimal medium, i.e., BM medium with an optimal concentration of 0.7% Tween 80) in different variants for 4 days. Compared to the media without CaCO<sub>3</sub>, a sixfold to ninefold higher amount of the active mycelium biomass was obtained from the OM medium with the addition of calcium carbonate and/or olive oil. No esterification activity was observed when the lyophilized post-culture liquid was used. The C. pannorum A-1 mycelium cultured on the OM medium with addition of 1% of calcium carbonate and 1% of olive oil showed the highest biocatalytic activity (Table 4). After 8 h of biocatalysis in hexane at 40 °C, 75% molar conversion of substrate was obtained. The biomass obtained on the OM medium with addition of 1% of CaCO<sub>3</sub> and 1% of olive oil was used to optimize the esterification of citronellol and acetic acid. A total of 78% of molar conversion was noted at 20 °C after 24 h, while the highest molar conversion equal to 95% of the substrate to citronellyl acetate was observed at 40 °C for 24 h, with only a slight decrease at the higher temperatures (Fig. 3A).

Over 95% molar conversion was obtained when the substrates were used at the concentrations of 10-25 mM (Fig. 3B). The increase in the substrate concentration from 50 up to 200 mM was accompanied by a linear decrease in molar conversion. The maximum reaction yield (98%) was obtained using 3% w/v of biomass-bound lipase from C. pannorum A-1. The further increase in the amount of the biocatalyst did not increase efficiency (Fig. 3C).

Among the solvents used, the highest initial reaction rate and molar conversion were obtained using ethyl acetate, hexane, cyclohexane, and hexadecane (Fig. 3D). No biomass-bound lipase activity was observed when 1-butanol, DMAC, DMF, NMP, ethanol, tetrahydrofuran, and 1-propanol were used. Trace esterification activity of the biomass was obtained when acetonitrile, 1.4-dioxane, and chloroform were used. In the final optimization step, the impact of the molar ratio of the substrates on the initial reaction rate was determined. The use of excess alcohol led to an increase in the initial reaction rate of esterification of citronellol and acetic acid (Fig. 3E).

The C. pannorum A-1 biomass-bound lipase retained up to 70% of its initial activity after seven 24-h reaction cycles (Fig. 4). The highest decrease in biomass activity was recorded after the fourth biocatalytic cycle.

		Medium		
	Day of culture	BMC (BM+1% CaCO <sub>3</sub> )	BMC1 (BMC+1% olive oil)	BMC2 (BMC + 1% olive oil + 1% glu- cose)
Extracellular lipase activity (U/mL)	2	$0.32 \pm 0.01$	$0.32 \pm 0.04$	$0.39 \pm 0.01$
	3	$1.19 \pm 0.13^{ab}$	$0.53 \pm 0.08$	$0.42 \pm 0.1$
	4	$1.40 \pm 0.07^{b}$	$0.97 \pm 0.08^{a}$	$0.27 \pm 0.01$
	5	$0.25 \pm 0.05$	$0.33 \pm 0.05$	$0.25 \pm 0.05$
Intracellular lipase activity (U/mg)	2	$0.05 \pm 0.01^{a}$	$0.07 \pm 0.01^{\rm ac}$	$0.19 \pm 0.01^{abcd}$
	3	$0.10 \pm 0.01^{abc}$	$0.14 \pm 0.01^{abc}$	$0.25 \pm 0.05^{bde}$
	4	$0.24 \pm 0.01^{bde}$	$0.37 \pm 0.02^{e}$	$0.23 \pm 0.03^{bcde}$
	5	$0.17 \pm 0.07^{\rm abc}$	$0.34 \pm 0.04^{de}$	$0.16 \pm 0.03^{abc}$
pH of the post-culture media	2	$7.26 \pm 0.02$	$7.22 \pm 0.07$	$5.96 \pm 0.03$
	3	$6.92 \pm 0.04$	$6.46 \pm 0.13$	$5.86 \pm 0.01$
	4	$7.06 \pm 0.04$	$5.90 \pm 0.04$	$5.92 \pm 0.01$
	5	$7.25 \pm 0.01$	$5.90 \pm 0.06$	$6.07 \pm 0.02$
Dry weight of mycelium (g/L)	2	$3.44 \pm 0.11$	$6.20 \pm 0.35$	$16.60 \pm 0.23$
	3	$7.53 \pm 0.46$	$8.75 \pm 0.89$	$21.90 \pm 0.11$
	4	$7.28 \pm 0.16$	$15.17 \pm 0.32$	$21.80 \pm 0.11$
	5	$5.38 \pm 0.72$	$13.75 \pm 0.09$	$21.30 \pm 0.09$

 Table 2
 Effect of addition of calcium carbonate to the culture media on lipase activity and dry weight of *C. pannorum* A-1 mycelium (g/L) and pH of the post-culture liquid

Means for the extracellular and intracellular lipase activity with different superscript letters denote a significant difference at p < 0.05

After a year-storage period, the fungal biocatalyst lost 18% of its initial activity (Fig. 5). A significant reduction in activity was noticeable at 4 °C and 20 °C after half a year (by 32 and 36%, respectively) and after a year of storage (by 46 and 54%, respectively).

### Discussion

Psychrophilic microorganisms are arousing increasing interest due to their high catalytic efficiency at low temperatures, which ensures lower energy consumption during the bioprocess [37]. Lipases from psychrophilic microorganisms facilitate efficient biocatalysis of volatile and heat-labile compounds at low temperatures [38]. For enzyme-catalyzed esterification and transesterification, commercial lipases, e.g., from Candida antarctica, are most commonly used [39, 40]. This paper presents a study on psychrophilic biomass as a new biocatalyst for direct esterification of terpene alcohols and acetic acid. In the first stage of the research, it was shown that the psychrophilic Chrysosporium pannorum A-1 strain exhibited extracellular lipolytic activity. To the best of the authors' knowledge, there is no data on the lipolytic activity of the fungus C. pannorum A-1. This strain has been reported to be an efficient biocatalyst for oxidative biotransformation of  $\alpha$ -pinene and  $\beta$ -pinene [32, 33, 41].

Efficient biosynthesis of enzymes is achieved in appropriately selected culture media and depends on the temperature, pH, and culture duration. Psychrotrophic fungi are adapted to survive in cold and nutrient-poor environments [28]. The preliminary experiments on the growth of C. pannorum A-1 in submerged culture containing lipase inductors showed the best biomass growth rate in a temperature range of 15–20 °C but no growth at 30 °C (data not shown). The biomass production at lower temperatures offers economic benefits through energy savings, since the average temperature of the Earth's surface is about 14 °C [42]. When C. pannorum was cultivated on rich medium A (containing olive oil 10%), no extracellular and intracellular lipolytic activity was detected (data not shown). However, using a P. chrysogenum 9' strain and medium A, Bancerz et al. [31] obtained 68 U/mL of extracellular lipolytic activity. As reported by Molinari et al. [16], medium A was found to be suitable for lipase production by C. pannorum A-1. The presence of olive oil or a combination of olive oil and the easily assimilated carbon source (glucose) in the medium led to a significant decrease in the pH of the post-culture liquid along with the duration of the culture. When pH dropped to 3, inhibition of the enzyme by the emerging organic acids likely occurred (Table 1). In addition, in the case of the BM2 medium, there was no lipolytic activity probably because glucose is the preferred energy source for the cell and is primarily used for biomass production. This phenomenon resembling catabolic Fig. 2 Effect of the Tween 80 concentration and cultivation time on the hydrolytic activity of extracellular (U/mL) (A) and intracellular (U/mg) (B) lipase from C. pannorum A-1. Optimization of the Tween 80 concentration was carried out on the BMC medium. Means for the determined values with different superscript letters denote a significant difference at p < 0.05



 
 Table 3
 Effect of the Tween 80
 concentration on pH of postculture liquid and dry weight of C. pannorum A-1 mycelium (g/L)

0.05

0

0.1

0.2

0.3

0.4

0.5

0.6

Tween 80 concentration (%)

0.7

0.8

0.9

1

	pH of the post-culture media Day of culture			Dry weight of mycelium (g/L) Day of culture			
Tween 80 con- centration (%)	3	4	5	3	4	5	
0	$7.52 \pm 0.02$	$7.67 \pm 0.04$	$7.88 \pm 0.02$	$4.22 \pm 0.11$	4.81±0.04	$2.26 \pm 0.08$	
0.1	$7.52 \pm 0.01$	$7.58 \pm 0.01$	$7.65 \pm 0.03$	$3.71 \pm 0.16$	$3.51 \pm 0.07$	$2.14 \pm 0.13$	
0.2	$7.30 \pm 0.01$	$7.43 \pm 0.03$	$7.44 \pm 0.02$	$4.72 \pm 0.39$	$3.36 \pm 0.98$	$1.08 \pm 0.08$	
0.3	$7.12 \pm 0.01$	$7.29 \pm 0.03$	$7.33 \pm 0.03$	$5.99 \pm 0.27$	$3.75 \pm 0.55$	$2.06 \pm 0.18$	
0.4	$6.97 \pm 0.02$	$7.12 \pm 0.02$	$7.25 \pm 0.03$	$6.74 \pm 0.12$	$5.78 \pm 0.13$	$3.40 \pm 0.15$	
0.5	$6.84 \pm 0.01$	$7.03 \pm 0.01$	$7.19 \pm 0.04$	$6.98 \pm 0.16$	$4.85 \pm 0.55$	$4.18 \pm 0.86$	
0.6	$6.79 \pm 0.03$	$6.96 \pm 0.04$	$7.10 \pm 0.02$	$6.61 \pm 0.36$	$6.55 \pm 0.16$	$4.83 \pm 0.20$	
0.7	$6.73 \pm 0.02$	$6.87 \pm 0.02$	$7.07 \pm 0.02$	$7.20 \pm 0.26$	$5.02 \pm 0.66$	$2.92 \pm 0.10$	
0.8	$6.66 \pm 0.02$	$6.74 \pm 0.02$	$6.93 \pm 0.03$	$6.43 \pm 0.41$	$4.95 \pm 0.13$	$4.96 \pm 0.12$	
0.9	$6.66 \pm 0.02$	$6.66 \pm 0.01$	$6.91 \pm 0.01$	$7.12 \pm 0.20$	$5.09 \pm 0.17$	$6.33 \pm 0.23$	
1	$6.55 \pm 0.01$	$6.64 \pm 0.02$	$6.80 \pm 0.01$	$6.02 \pm 0.17$	$6.95 \pm 0.07$	$5.50 \pm 0.12$	

**Table 4** Influence of the culture medium on the biocatalytic activity of biomass-bound lipase from *C. pannorum* A-1 and the amount of its biomass obtained in one culture cycle. The culture was carried out at 20 °C for 4 days. The esterification reaction was carried out in *n*-hexane with 2% w/v of the biocatalyst and an equimolar 50 mM citronellol and acetic acid concentration at 40 °C for 8 h

Medium	Initial reaction rate ( $\mu$ mol mg <sup>-1</sup> min <sup>-1</sup> )	Reaction yield (%)	Dry mass of mycelium (g/L)
OM	$0.65 \pm 0.01^{a}$	64	$1.56 \pm 0.21$
OM + 1% olive oil	$0.32 \pm 0.01^{b}$	31	$2.43 \pm 0.06$
OM + 1% olive oil + 1% glucose	$0.11 \pm 0.01^{\circ}$	10	$6.30 \pm 0.09$
$OM + 1\% CaCO_3$	$0.62 \pm 0.01^{a}$	60	$6.82 \pm 0.11$
$OM + 1\%$ olive oil + 1% $CaCO_3$	$0.77 \pm 0.01^{d}$	75	$14.82 \pm 0.43$
$OM + 1\%$ olive oil + 1% $CaCO_3 + 1\%$ glucose	$0.72 \pm 0.01^{e}$	71	$22.40 \pm 0.33$

Means for initial reaction rate with different superscript letters denote a significant difference at p < 0.05

OM - BM medium with optimal concentration 0.7% of Tween 80

repression can limit the use of olive oil by the cell. In turn, Falony et al. [43] obtained the highest extracellular lipase activity from *Aspergillus niger* on a medium composed of 2% of olive oil and 2% of glucose as a carbon source. Therefore, the medium composition should be selected individually for each microorganism and biological macromolecule.

As expected, the extracellular lipase from psychrotrophic C. pannorum showed the maximal activity at 20 °C; however, it was also active in a broad temperature range of 15-50 °C. The results obtained in the study on the pH optimum of lipase activity (Fig. 1B) are consistent with the findings of earlier experiments (Table 1) and indicate that enzyme inhibition takes place at pH below 3. Therefore, pH stabilization was used in the variant with olive oil and glucose in the medium to avoid a significant drop in pH during culture and thus lipase inactivation. The maintenance of the pH value in the optimal range for the growth of C. pannorum prevented lipase inhibition. This had also a positive effect on the dry weight of the mycelial biomass obtained on all three media (Table 2). Some studies have described a positive effect of Ca<sup>2+</sup> ions increasing the biosynthesis of lipases, e.g., lipases from Bacillus cereus [44] and Penicillium camemberti [45]. However, in this study, the increase in lipase biosynthesis by C. pannorum A-1 does not seem to be caused directly by  $Ca^{2+}$  ions, because there were no significant differences in the activities between the BMC and BM media (without olive oil and glucose). The buffering property of CaCO<sub>3</sub> was most likely essential to maintain enzyme activity on the BMC1 and BMC2 media (Table 2). In the case of lipase biosynthesis by C. pannorum A-1, Tween 80 turned out to be a better inducer than olive oil. It was found that the optimal Tween 80 concentration was strongly dependent on the cultivation time. The significant decrease in lipolytic activity observed along the culture time, especially at the low inducer concentrations (0.1-0.3%), was probably related to the exhaustion of the inducer during prolonged cultivation. In turn, the significant decrease in mycelium biomass in the culture extended up to 5 days (Table 3), which was noticeable at the concentration of Tween 80 up to 0.7%, may have been caused by autolysis of the fungus on the BMC medium devoid of the carbon source. Therefore, 4 culture days and a medium containing 0.7% of the inducer seem to be an optimal variant for lipase production by *C. pannorum*.

For the further study of the biocatalyst for the esterification reaction, both the post-culture liquid media and the mycelium biomass were lyophilized and used in the esterification of citronellol and acetic acid. The C. pannorum A-1 mycelium showed the highest molar conversion of the substrates to the product and the highest initial reaction rate. It is worth noting that the biocatalyst obtained from the OM medium with the addition of calcium carbonate, olive oil, and glucose was characterized by high catalytic activity and a large amount of biomass obtained in one culture cycle (Table 4). An increase in mycelium biomass obtained in one culture cycle is particularly important when the lyophilized mycelium as a biocatalyst is intended to be used on an industrial scale. As opposed to the free enzyme, the dried biomass-bound lipase shows lower sensitivity to freezing and lyophilization and higher stability in an organic solvent. The use of lyophilized biomass-bound lipase significantly reduces the costs of production of the biocatalyst, as it is possible to skip the isolation and purification of the enzyme. Esterification of citronellol and acetic acid by the biomassbound lipase from C. pannorum A-1 is highly efficient in the temperature range of 30–70 °C (Fig. 3A). With a temperature rise up to 70 °C, the initial rate increased twofold  $(0.58 \ \mu mol \ mg^{-1} \ min^{-1} \ at \ 70 \ ^{\circ}C \ vs. \ 0.27 \ \mu mol \ mg^{-1} \ min^{-1}$ at 20 °C). Increasing temperature can increase the diffusion coefficient of substrates migrating to the active sites of enzymes, thus enhancing the reaction rate [46]. Also, at higher temperatures, thermal inactivation of the enzyme may occur with the duration of the reaction. Notably, the C. pannorum biomass showed stability up to 70 °C, as no



**Fig. 3** Effect of temperature (**A**), concentration of substrates (**B**), amount of dry mycelium (**C**), organic solvent (**D**), and molar alcoholto-acid ratio (**E**) on the esterification reaction of  $\beta$ -citronellol and acetic acid catalyzed by biomass-bound lipase from *C. pannorum* A-1. The reaction was carried out in n-hexane with 2% w/v of the biocata-

lyst and an equimolar 50 mM substrate concentration for 24 h. Bars represent the molar conversion, while the line and points in **D** indicate the initial reaction rate. Means for the determined values with different superscript letters denote a significant difference at p < 0.05

statistically significant decline in molar conversion was observed in relation to 40 °C. Generally, lyophilized mycelium biocatalysts are characterized by high thermostability in organic solvents [16, 27, 47] due to the low water content in the reaction environment, which enhances enzyme denaturation [48], and the natural immobilization of the enzyme into the biomass exerting a protective effect [49]. Dry mycelium of *Rhizopus oryzae* was used for synthesis of different flavor esters (hexyl acetate and butyrate, geranyl acetate and butyrate). After 24 h of reaction in heptane at 50 °C, over 95% molar conversion of geraniol to geranyl acetate was obtained [50]. In esterification of isoamyl alcohol with acetic acid by biomass-bound lipase from *A. niger* NCIM 1207, 100% conversion of the substrates was achieved in a reaction carried out in hexane at 30 °C for 96 h [51]. A high esterification yield (98%) of cis-3-hexen-1-ol and acetic acid using biomass-bound lipase from *Aspergillus oryzae* was obtained at 70 °C [47].

In this study, the initial rate did not decrease with the increasing in the substrate concentration above 50 mM, but remained at a constant level (ca.  $0.31 \ \mu mol \ mg^{-1} \ min^{-1}$ ). This indicates the maximum saturation of the active site of the enzyme at the concentration of 50 mM and higher (Fig. 3B). This also implies that the catalytic activity of the biomass is inhibited by neither citronellol nor acetic acid at equimolar ratio. The results obtained in this experiment

Fig. 4 Reusability of C. pannorum A-1 in a 24-h esterification reaction cycle. The esterification reaction was carried out in *n*-hexane at an equimolar 50 mM citronellol and acetic acid concentration with an optimal amount of the biocatalyst at 30 °C for 24 h. Prior to each 24-h reaction cycle, the biomass was filtered off, washed with hexane, and dried in a desiccator. Means for the determined values with different superscript letters denote a significant difference at p < 0.05



are different from other published results. An increase in the equimolar concentration of geraniol and acetic acid over 50 mM caused a decrease in the initial reaction rate and molar conversion of substrates to geranyl acetate catalyzed by biomass-bound lipases of A. oryzae [16] and R. oryzae [27]. The authors underlined the negative impact of high acetic acid concentrations on the catalytic activity of the enzyme. In the present study, a significant decrease in enzymatic activity of the biomass was observed only when an excess of acetic acid was used (Fig. 3E). The same relationship was observed by Mhetras et al. [50] in the process of esterification of isoamyl alcohol and acetic acid by A. niger mycelium. Excess acid that does not participate in the reaction can cause acidification of the micro-aqueous coating of the enzyme, which leads to its inactivation. The strong influence of acetic acid on enzymatic activity was also reported by other researchers [23, 24]. As expected, the increase in the amount of the *C. pannorum* biocatalyst resulted in greater molar conversion of the substrates within 24 h (Fig. 3C). The initial reaction rate, i.e., approx.  $0.34 \ \mu\text{mol mg}^{-1} \ \text{min}^{-1}$ , was independent of the amount of the enzyme-containing biomass. Maintenance of a constant reaction rate regardless of the amount of the biocatalyst allows a proper design of the catalysis process (selection of an appropriate substrate-biocatalyst ratio to minimize masstransfer limitation) [52]. In the case of the biomass-bound lipase from *A. oryzae*, the initial reaction rate depended on the amount of lyophilized biomass and was threefold lower at the concentration of 0.5% w/v than at 2% w/v [16].

The biomass-bound lipase from *C. pannorum* A-1 exhibited high activity of esterification of citronellol and acetic acid in the hydrophobic solvents with log P above 3.4.

Fig. 5 Effect of storage temperature on the activity of the biomass-bound lipase of *C. pannorum* A-1 in the esterification reaction of  $\beta$ -citronellol and acetic acid. Means for the determined values with different superscript letters denote a significant difference at *p* < 0.05



The results of these studies are consistent with the results obtained by You et al. [53]. The catalyst bioactivity is best correlated with the n-octanol-water partition coefficient P. Low yields are obtained when polar solvents with  $\log P < 2$ are used, due to the distortion of the water structure in the enzyme. In the log P range from 2 to 4, the effect of the solvent is unpredictable, and above 4, the protein structure is intact [9]. In the variant with ethyl acetate as a solvent, transesterification by C. pannorum A-1 occurred instead of the esterification reaction (Fig. 3D). This was confirmed in a reaction carried out without acetic acid (data not shown). In designing a biocatalysis process with immobilized enzymes, it is very important to determine the moment of biocatalyst deactivation. In industry, the immobilized enzyme is usually replaced when the residual activity is between 50 and 10% of the initial activity [20]. Less frequent biocatalyst replacement reduces the biocatalysis costs significantly. The C. pannorum biomass-bound lipase was found to retain up to 80% of its initial activity after five 24-h reaction cycles (Fig. 4). Kirdi et al. [47] reported that A. oryzae biomass-bound lipase retained approx. 60% of activity after five 24-h reaction cycles. The decrease in the activity of the biomass was probably caused by the filtration and drying process between the catalytic cycles and the frequent exposure to acid and alcohol. The biomass-bound lipase from C. pannorum A-1 was also characterized by high stability during storage at - 20 °C. Since biocatalytic activity dropped significantly after the year-storage period at 4 °C and 20 °C (Fig. 5), it is necessary to store the preparation in a frozen state. Also, biomass-bound lipase from mesophilic Geotrichum candidum reported by Loo et. al. [17] showed a decrease in the enzyme activity to 70% of the initial activity after 8 months of storage at 4 °C.

### Conclusions

The *C. pannorum* A-1 biomass-bound lipase is a promising biocatalyst for efficient esterification of citronellol and acetic acid. The stabilization of pH with CaCO<sub>3</sub> caused a 9.5-fold increase in the biomass of the mycelium in one batch culture and a 2.4-fold increase in its biocatalytic activity. In addition, the biomass is effectively produced at temperature range of 15–20 °C; hence, the cost of production of the high-performance biocatalyst can be significantly reduced. The biocatalyst is active at low temperatures, stable at relatively high temperatures, and resistant to high acetic acid concentrations in equimolar biocatalysis in hexane. Given its high performance in several biocatalytic cycles, it can be successfully used to design the process of industrially applied biocatalysis.

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Data Availability All relevant data are within the paper.

### Declarations

Conflict of Interest The authors declare no competing interests.

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411

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