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CORRECTION

Correction to: Multiple Emulsions for Enhanced Delivery of Vitamins and Iron Micronutrients and Their Application for Food Fortification

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The original version of this article unfortunately contained some mistakes. In the published version of our article, Tables 1 and 2 are not published completely and only the first page of the tables is included in the article.

With this, here are the complete Tables 1 and 2.

The original article can be found online at https://doi.org/10.1007/s11947-021-02586-2.

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Table 1 Molecular structure and physical properties of vitamins (ChemAxon 2020; DrugBank 2020; Peinado et al. 2019)

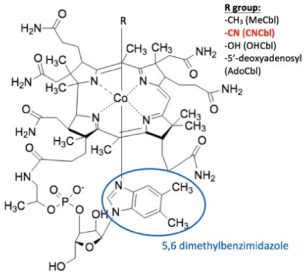
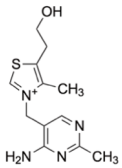
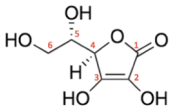
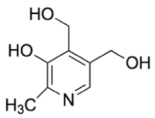
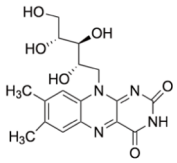
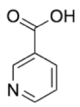
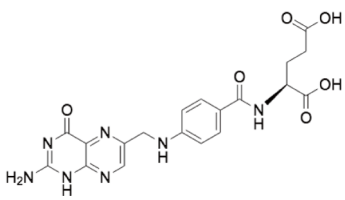
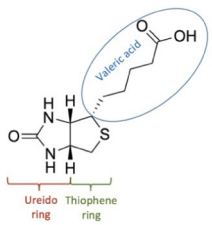
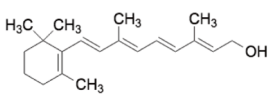
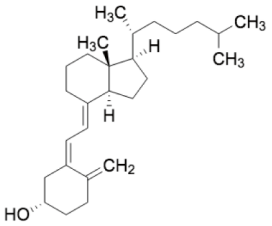
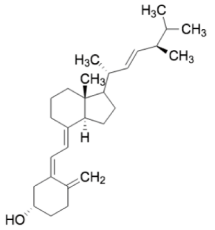
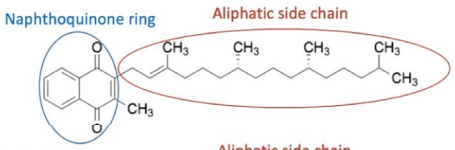
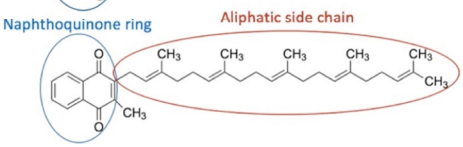
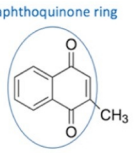
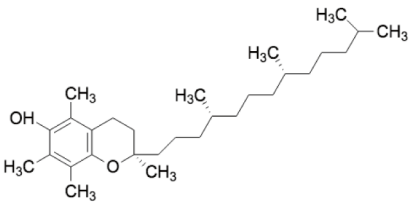
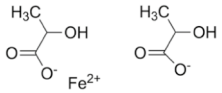
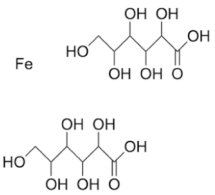
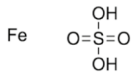
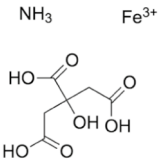
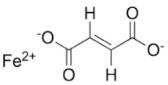
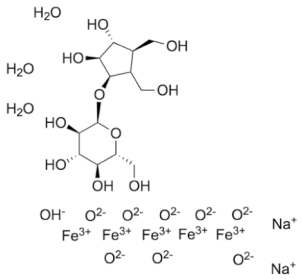
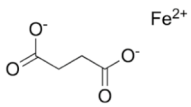
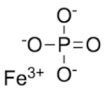
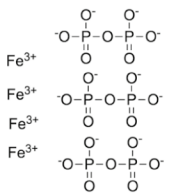
Vitamin	Molecular formula	Chemical structure	Molecular weight (g mol^{-1})	Water solubility (g L^{-1})	Log S[a]	Log P[b]
Vitamin B						
Cyanocobalamin (Vitamin B_{12})	$C_{63}H_{88}CoN_{14}O_{14}P$		1355.365	0.0202	-4.8	-3.2
Thiamine (Vitamin B_1)	$C_{12}H_{17}N_4OS$		265.355	0.0153	-4.3	-3.1
Ascorbic acid (Vitamin C)	$C_6H_8O_6$		176.124	245	0.14	-1.9
Pyridoxine (Vitamin B_6)	$C_8H_{11}NO_3$		169.178	16.1	-1	-0.95
Riboflavin (Vitamin B_2)	$C_{17}H_{20}N_4O_6$		376.364	0.657	-2.8	-0.92
Niacin (Vitamin B_3)	$C_6H_5NO_2$		123.109	83.1	-0.17	-0.17
Folic acid (FA) Vitamin B_9	$C_{19}H_{19}N_7O_6$		441.397	0.0761	-3.8	-0.04
Biotin (Vitamin H or B_7 or B_8)	$C_{10}H_{16}N_2O_3S$		244.311	1.22	-2.3	0.32
Retinol (Vitamin A)	$C_{20}H_{30}O$		286.452	0.00758	-4.6	4.69

Table 1 (continued)

Vitamin D						
Cholecalciferol (Vitamin D_3)	$C_{27}H_{44}O$		384.638	0.00038	-6	7.13
Ergocalciferol (Vitamin D_2)	$C_{28}H_{44}O$		396.648	0.000433	-6	7.05
Vitamin K						
Phylloquinone (Vitamin K_1)	$C_{31}H_{46}O_2$		450.696	5.92e-05	-6.9	9.7
Menaquinone (Vitamin K_2)	$C_{36}H_{48}O_2$		512.778	0.000328	-6.2	10.14
Menadione (Vitamin K_3)	$C_{11}H_8O_2$		172.18	0.504	-2.5	1.89
Vitamin E	$C_{29}H_{50}O_2$		430.706	7.4e-06	-7.8	10.51

[a] log S is the 10-base logarithmic measurement of solubility
[b] log P is the 10-base logarithmic measurement of the partition coefficient in octanol-water solution

Table 2 Molecular structure and physical properties of iron (ChemAxon 2020; DrugBank 2020)

Iron type	Molecular formula	Chemical structure	Molecular weight (g mol^{-1})	Water solubility (g L^{-1})	Log S[a]	Log P[b]
Ferrous lactate	$C_6H_{12}FeO_6$		233.98	165	-0.15	-0.47
Ferrous gluconate	$C_{12}H_{24}FeO_{14}$		448.15	36.6	-1.1	-3.4
Ferrous sulphate	FeO_4S		153.92	256	---	-0.84
Ferric ammonium citrate	$C_6H_8FeNO_7$		261.98	106	-0.26	-1.3
Ferrous fumarate	$C_4H_2FeO_4$		169.90	13.1	-1.2	-0.041
Sucroferric oxyhydroxide (Iron saccharate)	C18H27FeO24		866.546	9.52	-1.8	1.99
Ferrous succinate	$C_4H_4FeO_4$		171.917	31.8	-0.85	-0.4
Ferric orthophosphate (Ferric phosphate)	$FePO_4$		150.82	Anhydrous: insoluble Dehydrated: 0.642 g/100 ml (100 °C)	---	-1
Ferric pyrophosphate	$Fe_4O_{21}P_6$		745.21	---	---	-1.4

[a] log S is the 10-base logarithmic measurement of solubility
[b] log P is the 10-base logarithmic measurement of the partition coefficient in octanol-water solution