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TMS has forged cooperative agreements with several carefully selected organizations that actively work to benefit the materials science community. In this occasional series. JOM will provide an update on the activities of these organizations. This installment, by the Center for Resource Recovery & Recycling (CR³), focuses on the different recycling options for electric arc furnace dust and shines a light on a promising new developing technology. The CR³ is a research center established by Worcester Polytechnic Institute, Colorado School of Mines. and KU Leuven. More than 20 corporations. along with support from the U.S. National Science Foundation's Industry & University Cooperative Research Program, are sponsors of the center.

TMS Partners in Progress

Moving Towards Better Recycling Options for Electric Arc Furnace Dust

T. Suetens, K. Van Acker, B. Blanpain, B. Mishra, and D. Apelian

When recycling steel scrap in an electric arc furnace, 15–18 kg of dust (EAFD) is generated for each ton of produced steel. This dust essentially is an oxidized mixture of the volatile elements present in the furnace (Zn, Pb, Cd, and halides) and small steel melt and slag particles (Fe, Cr, Ca, Si, Mg). Worldwide, 7.1 Mt of dust were produced in 2010,¹ a number that continues to rise due to increasing recycling volumes and application of galvanization. Since the dust contains the spinel franklinite (ZnFe₂O₄), recycling requires drastic treatment conditions to recover the zinc.

U.S. Environmental Protection Agency legislation against landfilling the dust has caused an increase in recycling rates since the late 1980s. However, nothing changed in the regions where landfilling has not been restricted. Due to the large amount of dust generation in those regions, only 40% of the total amount of generated dust is recycled. This leaves a large potential for improvement in zinc recycling.

Since more than 50% of all used zinc

goes into galvanizing,² EAFD plays a key role in the life cycle of zinc. The growing awareness of depletion of metal reserves and increasing metal prices are becoming good motivators to start the treatment of the dust in favor of landfilling. In the regions where recycling rates were already good, these extra incentives resulted in a growing interest in the iron content of the dust for iron recovery.

A whole range of metallurgical technologies was developed to recycle the dust. So far, most of these have faced serious problems. For the hydro projects, the stability of the franklinite phase has always been a major issue. Since it does not dissolve under most leaching conditions, a maximum zinc leaching recovery of 70% has been observed.3 Another issue of the hydrometallurgical processes is that they are unable to recover iron economically. With increasing interest of the industry to also recover the iron, the appeal of hydro for further development has been dwindling. High temperature metal recovery (HTMR)



Figure 1. Traditional electric arc furnace dust treatment pathway.



Figure 2. Alternative EAFD treatment pathway. Implementation of IPS could replace the need for an energy and resource intense HTMR process.

Process Name Waelz Kiln ^{1,4} Temperature 1150–1200°CFurnace type Input Rotary kilnProducts Coal, lime/sand, natural gas, airScaleCapacity CO, Fe-slag (waste)(kt/y)RHF5 (low Zp EAED)1250–1300°CRotary hearth furnaceCoke, binder, Datural gasCZO, DRI/HBICommercial100–300						
Waelz Kiln ^{1,4} 1150–1200°C Rotary kiln Coal, lime/sand, natural gas, air CZO, Fe-slag (waste) Commercial 40–250 RHF ⁵ 1250–1300°C Rotary hearth furnace Coke, binder, natural gas CZO, DRI/HBI Commercial 100–300						
RHF ⁵ 1250–1300°C Rotary hearth furnace Coke, binder, CZO, DRI/HBI Commercial 100–300						
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RHF ⁵⁻⁹ 1250–1300°C Rotary hearth furnace Coke, binder, CZO, DRI/HBI Commercial* 200 (high Zn EAFD) natural gas						
Primus ^{10,11} 1000–1100°C Multiple hearth Coal, air CZO, pig iron Commercial 100 furnace + EAF						
Ausmelt ¹²⁻¹⁴ 1250–1300°C Top Submerged Coal, CZO, Fe-slag Commercial 100 Lance furnace O2-enriched air (waste)						
ESRF ^{15,16} 1300–1500°C EAF Binder, air, CZO, Pig iron, Commercial 36 electricity and slag						
Submerged1300–1400°CSubmerged plasmaCoke, natural gas,CZO, Fe-slag (waste)Commercial40 – 60Plasma17-19reactorfluxes, air, electricity						
PIZO ²⁰⁻²² 1300–1500°C Induction furnace Coal, air, electricity CZO, pig iron Commercial 50						
OxiCup ²³ 1500–1600°C Shaft furnace Coke, scrap, bricks CZO, molten Commercial 200 (dust (waste + cement + C) (waste + cement + C) metal, and slag and sludge)						
Coke Packed Bed ^{24,25} 1500–1600°C Shaft furnace Coke, fluxes, CZO, molten metal, Pilot Plant 10 O ₂ -enriched air and slag						
LAMS ^{26,27} 900-1100°C — CaCO ₃ + heat CZO, Ca ₂ Fe ₂ O ₅ for Lab-scale — use in blast furnace						
EAFD+PVC pellets ^{28,29} 800°C — PVC + heat ZnCl ₂ , Fe+C pellets Lab-scale —						

Table I. Overview of Currently	v Available Recycling	Options for Electri	c Arc Furnace Dus
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* First large proof-of-concept plant is still ramping up to full production.

processes, on the other hand, have known a certain degree of success. Currently, the Waelz kiln process is recognized as the best available technique to handle the dust. It is used to treat roughly 80% of all recycled EAFD. Traditionally, the recycling of EAFD can be seen as described in Fig. 1. As an alternative to the Waelz kiln process, a variety of different technologies have been developed. These use different furnace designs, working temperatures, reducing agents,



Figure 3. Exergy efficiencies of the Waelz kiln, RHF, and IPS technologies. additives, etc. An overview of these alternatives can be found in Table I. Most of these processes produce a crude zinc oxide (CZO) product that requires some further processing (halide removal) before it can be used by zinc smelters.

In a previous exergy efficiency analysis study for CR³

the two most applied technologies (Waelz kiln and RHF) have been compared.³⁰

Starting from the EAF off-gas, the optimized Waelz kiln process (Table I, first entry) underperforms compared to the RHF process as described by ZincOx (Table I, third entry).

However, this study clearly indicated that HTMR processes might not be the ultimate solution for EAFD recycling since their total exergy efficiencies are rather low. The study also included calculations for a new, conceptual alternative to HTMR processes: treatment of the EAF off-gas in the EAF plant itself to prevent the formation of EAFD and the need for its costly recycling processes.

Recently, N. Ma proposed a radical new approach, the In-Process Separation (IPS) technology,^{31,32} in which zinc is removed from the EAF off-gas before it can react with iron-containing particles to form the $ZnFe_2O_4$ phase in the combustion chamber (Fig. 2). The dust collected in the baghouse filters can then be sent directly to the CZO treatment plant as it will no longer contain Fe. Depending on the atmospheric conditions in the off-gas treatment system, iron or iron oxide particles can be returned to the EAF. Thermodynamically, this treatment step drastically outperforms any HTMR process since it has an exergy efficiency of 53% (Fig. 3) while the exergy efficiency of the current EAF off-gas system is 24%. Further treatment of the dust in a HTMR process will only reduce this value. Economically, the technology would transform the traditional EAF off-gas treatment system from a waste collecting part of the plant to a zero-waste valuable by-product producing segment.

Recognizing the technology's potential impact on EAFD recycling, CR³ has a project running to experimentally evaluate the feasibility of the IPS technology. The major drawback of this new technology is that it requires implementation in each individual EAF plant. Unlike traditional HTMR processes, the starting product cannot be collected from a larger area to treat a lot of dust simultaneously. All plants operating without the IPS technology will continue producing EAFD, maintaining the need for HTMR plants.

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