



# Generalized olfactory detection of floral volatiles in the highly specialized *Greya-Lithophragma* nursery pollination system

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## Abstract

Volatiles are of key importance for host-plant recognition in insects. In the pollination system of *Lithophragma* flowers and *Greya* moths, moths are highly specialized on *Lithophragma*, in which they oviposit and thereby pollinate the flowers. Floral volatiles in *Lithophragma* are highly variable between species and populations, and moths prefer to oviposit into *Lithophragma* flowers from populations of the local host species. Here we used gas chromatography coupled with electroantennographic detection (GC-EAD) to test whether *Greya* moths detect specific key volatiles or respond broadly to many volatiles of *Lithophragma* flowers. We also addressed whether olfactory detection in *Greya* moths varies across populations, consistent with a co-evolutionary scenario. We analyzed flower volatile samples from three different species and five populations of *Lithophragma* occurring across a 1400 km range in the Western USA, and their sympatric female *Greya politella* moths. We showed that *Greya politella* detect a broad range of *Lithophragma* volatiles, with a total of 23 compounds being EAD active. We chemically identified 15 of these, including the chiral 6, 10, 14-trimethylpentadecan-2-one (hexahydrofarnesyl acetone), which was not previously detected in *Lithophragma*. All investigated *Lithophragma* species produced the (6*R*, 10*R*)-enantiomer of this compound. We showed that *Greya* moths detected not only volatiles of their local *Lithophragma* plants, but also those from allopatric populations/species that they not encounter in local populations. In conclusion, the generalized detection of volatiles and a lack of co-divergence between volatiles and olfactory detection may be of selective advantage for moths in tracking hosts with rapidly evolving, chemically diverse floral volatiles.

**Keywords** Coevolution · Floral scent · Pollination · GC-EAD · Olfaction · Olfactory receptors

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## Introduction

In specialized, interspecific interactions with reciprocal fitness consequences, coevolution between the interacting partners is often the consequence (Thompson 2005). A key model system for the investigation of coevolution in plant–insect associations has been the woodland star flower (*Lithophragma*) and *Greya* moths nursery pollination system (Thompson 2005; Thompson and Cunningham 2002; Thompson et al. 2017). In this system, *Greya* moths pollinate the flowers while ovipositing into them through the corolla or while nectaring, and *Greya* larvae feed on developing seeds. The *Greya-Lithophragma* pollination system is an open nursery system in which moths are highly specific to their local *Lithophragma* host species and are often the major pollinators of their hosts. The term “open nursery pollination system” refers to the fact that co-pollinators, that do not lead to destruction of seeds, also contribute to pollination in varying degrees among *Lithophragma* populations

and species (Thompson and Cunningham 2002; Thompson et al. 2017). Thus, whereas nursery pollination systems without co-pollinators are generally mutualistic, this system varies between mutualism and antagonism on a geographical mosaic scale, depending on the number of *Greya* species interacting with one *Lithophragma* species and the number of generalized co-pollinators present (Thompson and Cunningham 2002).

Like many insects (Bruce and Pickett 2011), *Greya* moths are thought to use plant odor as a cue of primary importance for host plant identification (Friberg et al. 2014). For olfaction, insects employ sets of highly sensitive, and often specifically tuned olfactory neurons (de Fouchier et al. 2017). To identify host plants, many insects use blends of common volatiles, but some use only specific compounds that serve as identification cue of their food plants (Bengtsson et al. 2011; de Fouchier et al. 2017; Schiestl 2015). In plant-pollinator interactions, examples of such olfactory “private channels” are spiroacetals emitted by bellflowers (Milet-Pinheiro et al. 2013), 1,4-benzoquinone produced by *Echium* flowers (Burger et al. 2012), 4-methyl-5-vinylthiazol in *Cyclocephala*-pollinated Araceae inflorescences (Maia et al. 2012), and diacetin from plants with oil as floral reward (Schaeffler et al. 2015). An additional aspect of such specificity is the use of specific enantiomers of chiral molecules. Several studies have shown that insects may respond only to specific enantiomers of a given volatile (Bohman et al. 2018; Borg-Karlson et al. 2003; Eltz et al. 2010), or to specific ratios of enantiomers (Chen and Song 2008), pointing out the need to better understand the enantiomeric composition of semiochemicals. Whereas geographic variation in volatile blends has commonly been documented, potential variation in enantiomeric composition of chiral molecules is very little understood but may contribute to population-specificity in chemical signals (Raguso 2016).

Specialization of insects to particular host plants may involve evolutionary change in the sensitivity of olfactory neurons towards different ligands. In principle, change in host plants or food sources should select for changes in the olfactory system to allow optimal detection of food. This may not always be the case, for example, if common volatiles are shared by different food sources (Bengtsson et al. 2011). Phylogenetic studies have indeed shown that related olfactory neurons often detect similar compound classes across large taxonomic groups of insects, suggesting conservative evolution of receptor sensitivity in these insects (de Bruyne et al. 2010; de Fouchier et al. 2017; Jones et al. 2005). Nevertheless, some examples of apparent rapid change in odor sensitivity are also known (Blazyte-Cereskiene et al. 2019; de Bruyne et al. 2010; Metcalf et al. 1979; Schaeffler et al. 2015).

A fascinating, yet almost unexplored question is whether and how rapidly receptor sensitivity may co-evolve with

plant volatile emission (Schiestl and Dötterl 2012). Floral scent in *Lithophragma* is chemically highly diverse, both among populations and species (Friberg et al. 2013, 2019), indicating rapid evolutionary divergence in this key floral signal (Ramos and Schiestl 2020). Such pattern of high variability has been found in other nursery pollination systems, too (Dötterl et al. 2005; Suinyuy et al. 2015; Waelti et al. 2008), but contrasts to the highly conserved odor bouquets found in yucca flowers (Svensson et al. 2005, 2006, 2011). Yucca flowers form a highly specialized nursery pollination association with yucca moths, which are closely related to *Greya* moths (Thompson 2010).

Given the specificity and tight dependence of *Greya* moths on *Lithophragma* host plants, adaptation to the region-specific floral volatile bouquets could be expected. Indeed, Friberg et al. (2014) found evidence that moths from three of the locations used here (TUR, SED, MBL), by using floral volatiles, prefer sympatric flowers for oviposition and often ignore flowers from allopatric species. In no-choice oviposition tests, more moths oviposited into sympatric host species, but a few also oviposited into allopatric ones. This result suggests differentiation in the moths’ olfactory perception, so that allopatric flowers are often not chosen, even though they would likely be suitable oviposition substrates, as all species within the *L. parviflorum* and *L. campanulatum* clades are suitable hosts within some part of the range of *G. politella* (Thompson et al. 2017). Hence, preference of *G. politella* for any particular *Lithophragma* is likely due to local specialization rather than to inherent major differences among *Lithophragma* species as suitable hosts.

Here we present data of the first study of electrophysiological detection of *Lithophragma* floral volatiles by *Greya* moths. This approach was chosen to determine whether *Greya* moths detect specific key compounds or respond broadly to many different floral volatiles. We also addressed the question of geographic variation in responses, namely whether *Greya* moths do/do not respond to floral volatiles of allopatric *Lithophragma* plants (i.e. those that they normally not encounter). A lack of (or notably reduced) response to allopatric volatiles would indicate co-divergence of olfactory sensitivity and volatile bouquets. Finally, we found a chiral volatile not-yet detected in *Lithophragma*, 6, 10, 14-trimethylpentadecan-2-one, which was highly EAD-active in *Greya* moths. This compound acts as a phagostimulant in the fall armyworm, *Spodoptera frugiperda* (Mohamed et al. 1992), has been reported as sex pheromone in some Lepidoptera (Nieberding et al. 2008) and is known to have enantiomer-specific behavioral activity in euglossine bees (Eltz et al. 2010). Hence, we characterized the enantiomeric composition of this compound as well as geographic variation thereof. If true, this may provide a previously undetected population- or species-specific chemical profile of flowers.

We used floral volatile samples from five populations, which included three closely related species of *Lithophragma* (Kuzoff et al. 1999), and their sympatric moths (see methods section and Table 1), occurring in contrasting habitats near the northern and southern edges of the geographic distribution of these interactions. For the assessment of electrophysiological activity and differences in activity between populations, we used gas chromatography coupled to electroantennographic detection (GC-EAD). This method broadly screens volatile samples for electrophysiologically active compounds, which is desirable in systems where olfactory detection of volatiles has not yet been investigated, in order not to miss any potential key compounds produced in small amounts. This method, however, may have a high detection threshold (depending on the sensitivity, specialization, abundance of antennal receptors, and background noise), and does not allow for a fine-tuned quantitative assessment of receptor sensitivity. Other, more quantitative/sensitive methods such as electroantennography (EAG) or single sensillum recording require previous knowledge of active compounds (Schiestl and Marion-Poll 2002) and were thus not applied in this study. Therefore, our comparison of population-specific detection of volatiles by moths was done only on a qualitative basis (response/no response), without any quantitative comparison of intensities of responses.

## Methods and materials

### Plants, insects

Experiments were conducted in spring 2015 and 2016. *Greya politella* moths (Lepidoptera: Prodoxidae) are completely dependent on the *Lithophragma* (Saxifragaceae) host plants, whereas the dependence of *Lithophragma* on *Greya* as pollinators can be reduced, at least at some sites, when generalized co-pollinators are abundant (Thompson and Cunningham 2002). *Greya* moth males search for females by flying among host flowers. Adults mate on and take nectar from flowers, and females pollinate the host flowers during nectaring and oviposition. The larvae feed on a fraction of the seeds before spending the dry summer months in diapause near or in the root system. In spring, larvae continue to feed on the leaves and pupate in a rolled-up leaf.

*Lithophragma* volatiles and *G. politella* moths were collected from the following populations in California and Washington State, USA: *Lithophragma cymbalaria* (SED: Sedgwick, University of California, Sedgwick Reserve, California: 34 42.871°N, 120 2.999°W), *L. bolanderi* (MBL: Marble Falls Trail, Kaweah River, California: 36 31.198°N, 118 48.024°W; WOO: Woody, California: 35 43.176°N, 118 47.907°W; KNG: King's River, California: 36 51.536°N, 119 06.202°W; no flower samples were available from KNG, and

no moths from WOO), and *L. parviflorum* (TUR: Turnbull National Wildlife Refuge, Washington State: 47 24.185°N, 117 34.103°W). The northern and southern edge populations were separated by approximately 1400 km, and the southern edge populations were separated, west to east, by up to approximately 230 km. The moth populations at these geographically extreme sites differed in the *Lithophragma* species they used as hosts. The *Lithophragma* species and populations differ dramatically, both quantitatively and qualitatively, in their floral scent bouquets (Friberg et al. 2013, 2014, 2017, 2019). Similarly, the *G. politella* moths from these locations are genetically diverged to an extent that they may be considered different cryptic species (Rich et al. 2008).

In 2015, only floral headspace samples were collected and analyzed; in 2016, headspace samples of SED, and flower extracts of all populations were collected/analyzed. Female moths were collected from the same populations by catching individuals sitting on flowers of each local *Lithophragma* species. Moths were kept in Eppendorf tubes with a hole punched in the lid, in a refrigerator at 4 °C, where they survived for up to 10 days without feeding. Nevertheless, moths were usually used within 2–5 days after capture for GC-EAD recordings.

### Headspace volatile collection

We collected floral scent from whole inflorescences for gas chromatography-electroantennographic detection (GC-EAD), using headspace collection with Porapak Q as adsorbent (Huber et al. 2005; Knauer and Schiestl 2015). Because GC-EAD requires highly concentrated samples in case of high detection thresholds, we included in each sample multiple, freshly cut inflorescences from multiple plants that collectively had up to 100 flowers. The inflorescences were enclosed in oven bags (Reynolds<sup>®</sup>, Richmond, VA, USA), and a glass tube filled with 4 mg of Porapak Q (Mesh size 80/100; Alltech Associates Inc., Deerfield, IL, USA) was inserted into the bag. The use of low amounts of adsorbent in a narrow tube allows for elution of collected volatiles with low volumes of solvent, thus avoiding the need to concentrate samples after collection. The Porapak was enclosed by a layer of quartz wool and glass beads (0.3 mm, Merck KGaA, Darmstadt, Germany) fused into the glass tube by heating on both sides. Before use, all Porapak tubes were cleaned with 400 µl dichloromethane. The tubes were connected with a silicon tube to a Micro Air Sampler (PAS-500 Micro Air Sampler, Spectrex, Redwood City, CA, USA), which pulled air through the filter with a flow of 100 ml min<sup>-1</sup> for 2–3 h. As a control for ambient contaminants, we collected scent from an empty bag using the same collection parameters. After sampling, we eluted the trapped volatile compounds with 50 µl of a hexane and acetone (9:1) mixture (for all

solvents, HPLC grade was used). All samples were stored in 2 ml glass vials at  $-30\text{ }^{\circ}\text{C}$  until analysis with GC-EAD and GC-MSD (GC-EAD; Schiestl and Marion-Poll 2002).

## Flower extract collection

We collected low-volatile compounds from flowers, by rinsing up to 300 cut flowers from ca. 30 plants for 30 s in 1–5 ml (depending on the number of flowers) solvent and then immediately removing them from the solvent. As solvent, we used a mixture of 9:1 hexane: acetone. The resulting crude extract was then either concentrated to 500  $\mu\text{l}$  volume under a stream of  $\text{N}_2$  and subsequently directly injected into the GC, or further purified using solid phase extraction (SPE) with 500 mg silica columns (Supelco), following Hedenström et al. (2015). Approximately 100  $\mu\text{l}$  of concentrated extract was applied on the column bed and the solvent evaporated under a stream of  $\text{N}_2$ . Elution of different compound fraction was then achieved with stepwise gradient elution with 0–10% ethyl acetate in pentane, in steps of 1% increase. Fraction 1 was pure pentane and fraction 11 was pentane with 10% ethyl acetate. 200  $\mu\text{l}$  solvent was used for each for elution; the solvent was slowly pulled through the silica bed with vacuum. Elution 1–3 (mostly alkanes), 4–8 (mostly ketones), and 9–11 (mostly alcohols) were pooled

and concentrated over an  $\text{N}_2$  flow to ca 200  $\mu\text{l}$ . The GC-EAD was then performed with the different fractions.

## Electrophysiology

Gas chromatographic analysis with electroantennographic detection (GC-EAD; Schiestl and Marion-Poll 2002) of volatile samples was performed using a gas chromatograph (Agilent 6890 N, Agilent Technologies, Palo Alto, CA, USA) equipped with a heated outlet for electroantennographic recordings (Effluent Conditioning Assembly, Syntech, Hilversum, the Netherlands). Antennal responses of female *Greya* moths were measured via EAD. For EAD recordings, the tip of the excised antenna was abscised and the antenna was mounted on an antennal holder using electrode gel. The antennal holder was connected via a signal interface box (Syntech, Hilversum, The Netherlands) to a personal computer. 5  $\mu\text{l}$  of the samples were injected splitless ( $250\text{ }^{\circ}\text{C}$ ) at an oven temperature of  $50\text{ }^{\circ}\text{C}$  (1 min) into the GC followed by heating to  $240\text{ }^{\circ}\text{C}$  at a rate of  $10\text{ }^{\circ}\text{C min}^{-1}$ . The GC was equipped with a DB-WAX (J&W Scientific, Folsom, CA, USA) column (0.325 mm diameter, 0.25  $\mu\text{m}$  film thickness, 30 m length) and a flame ionization detector (FID). Helium was used as carrier gas at a flow of  $2\text{ ml min}^{-1}$ . A GC effluent splitter (Agilent G2855 Deans Switching System, Agilent Technologies, Palo Alto, CA, USA) was used to direct

**Table 1** GC-EAD-active volatiles in *Lithophragma* headspace samples and synthetic blends

<i>Lithophragma</i> species/region (population)	<i>Greya politella</i> pop.	1 (Z)-3-Hexenyl acetate	2 <i>Ui</i> 1	3 Nonanal	4 Decanal	5 Linalool	6 <i>Ui</i> 2	7 Camphor	8 Acetophenone	9 <i>Ui</i> 3	10 1,4-Di methoxybenzene	11 Methyl salicylate	12 2-Phenylethyl acetate	13 2-Phenylethyl alcohol	14 <i>Ui</i> 4	Dimethyl salicylate	2-Amino benzaldehyde	Methyl anthranilate	Cinnamic alcohol	TMP	Number of moths
<i>L. cymbalaria</i> (SED)	SED	n.p.	n.p.	✓	low	✓	n.p.	n.p.	n.p.	n.p.	n.p.	✓	n.p.	n.p.	n.p.	✓	✓	✓	n.p.	n.p.	4
<i>L. bolanderi</i> (MBL)	MBL	✓	✓	✓	✓	✓	✓	✓	✓	✓	n.p.	✓	✓	✓	✓	n.p.	✓	✓	✓	n.p.	3
<i>L. bolanderi</i> (MBL)	KNG	✓	✓	✓	✓	✓	✓	✓ <sup>1</sup>	✓	✓	n.p.	✓	✓ <sup>1</sup>	✓	✓?	n.p.	✓	✓	✓	n.p.	3
<i>L. bolanderi</i> (WOO)	MBL	n.p.	✓	n.p.	✓	n.p.	✓	✓	n.p.	n.p.	✓	✓	✓	✓	✓	n.p.	n.p.	n.p.	n.p.	n.p.	4
<i>L. bolanderi</i> (WOO)	KNG	n.p.	✓	n.p.	✓	n.p.	✓	✓ <sup>1</sup>	n.p.	n.p.	✓ <sup>1</sup>	✓	✓ <sup>1</sup>	✓	✓	n.p.	n.p.	n.p.	n.p.	n.p.	3
<i>L. parviflorum</i> (TUR)	TUR	n.p.	n.p.	n.p.	n.p.	n.p.	n.p.	n.p.	n.p.	n.p.	n.p.	✓	✓	n.p.	n.p.	n.p.	n.p.	n.p.	n.p.	✓	3
Synth (SED mix1)	SED			✓		✓						✓									3
Synth. (SED mix2)	MBL			✓		✓						✓				✓	✓			✓	1
Synth (MBL mix)	MBL	✓			✓			✓	✓			✓	✓	✓							2
Synth (MBL mix)	KNG	✓			✓			✓	✓			✓	✓	✓							1
Synth (WOO mix 1)	SED			✓	✓						✓		✓								1
Synth (mix 2)	TUR										✓							✓	✓	✓	2

A check indicates reproducible EAD activity in female moths for a given compound; checks in red indicate EAD-responses to scent compounds that do not occur in the sympatric plant populations. Numbers refer to those peaks shown in Fig. 1. Compounds without numbers are not shown in Fig. 1. In all synthetic mixtures, all compounds were EAD active (i.e. the mixtures contain only those compounds that were indicated as being active)

SED Sedwick, CA; MBL Marble falls trail, CA; WOO Woody, CA; KNG King's river, CA; TUR Turnbull, WA (for GPS coordinates see Methods section). TMP 6,10,14-trimethylpentadecan-2-one, *Ui* unidentified compound, n.p. compound not present in sample

<sup>1</sup>Because KNG flowers were not analyzed in this study, the information on the volatiles emitted by these flower were taken from Friberg et al. (2019)

50% of the effluent over a heated transfer line (200 °C) into a purified and humidified air stream (250 ml min<sup>-1</sup>) in a stimulus delivery tube (10 mm diameter), which was directed over the excised antenna. EAD signals and FID responses were simultaneously recorded using Syntech software. Volatiles were assumed EAD-active if reproducible responses at the same retention time were seen in antennae of at least two individual *Greya* moths. For confirmation of responses with synthetic compounds, in case of clear responses, EAD responses of a single moth were considered sufficient (see Tables 1 and 2 for exact numbers). Volatiles were identified by gas chromatography with mass selective detection (GC-MSD, see below). All putatively EAD-active volatiles were also directly tested for EAD activity as synthetic standards using the same GC-EAD system as for the natural samples. Synthetic volatiles were prepared in blends of compounds with non-overlapping retention times, in concentrations of 100 µg ml<sup>-1</sup> in hexane.

### Compound identification

Volatiles were identified using GC-MSD with liquid injection. Samples were injected with pulsed splitless mode (200 °C) with 20 psi pulse pressure into a GC (Agilent 6890 N, Agilent Technologies, Palo Alto, CA, USA). The GC oven was programmed to rise from a starting temperature of 50 °C to 230 °C at a rate of 10 °C min<sup>-1</sup> and held at the final temperature for 20 min. The GC was equipped with a DB-WAX column (0.32 mm diameter, 0.25 µm film thickness, 30 m length) and helium was used as carrier gas at a flow rate of 1.5 ml min<sup>-1</sup>. The GC column was connected to a mass selective detector (Agilent MSD 5975, Agilent Technologies, Palo Alto, CA, USA). The ChemStation Enhanced Data Analysis program was used. Scanning was started after 4.6 min solvent delay, and ranged from m/z 40 to 400. Compounds were tentatively identified by comparison of MS spectra obtained from the samples, with those from a reference library (NIST 05 library). For final

identification and quantification, synthetic standards of all compounds were analyzed and retention times and total mass spectrum of synthetic standards were compared with those of natural samples. Those compounds for which no reference standards were available, but spectra comparison yielded a match of above 90% quality, were judged as tentatively identified. Several EAD active compounds gave no useful MS fragmentation pattern for library comparison and were hence judged as unidentified. All chemical analyses (GC-EAD, GC-MS) were conducted at the United States Department of Agriculture, Agricultural Research Services, Albany, California.

### Quantification

Quantification of volatiles was only tentatively possible, as most samples were collected from large bulk of flowers. To do so, synthetic reference standard were analyzed in two different concentrations to obtain calibration curves. Amounts of volatiles were then calculated from peak areas obtained from total ion chromatograms.

### Determination of stereoisomeric configuration of 6, 10, 14-trimethylpentadecan-2-one (TPM-one)

#### Purification of flower extracts prior to stereochemical analysis

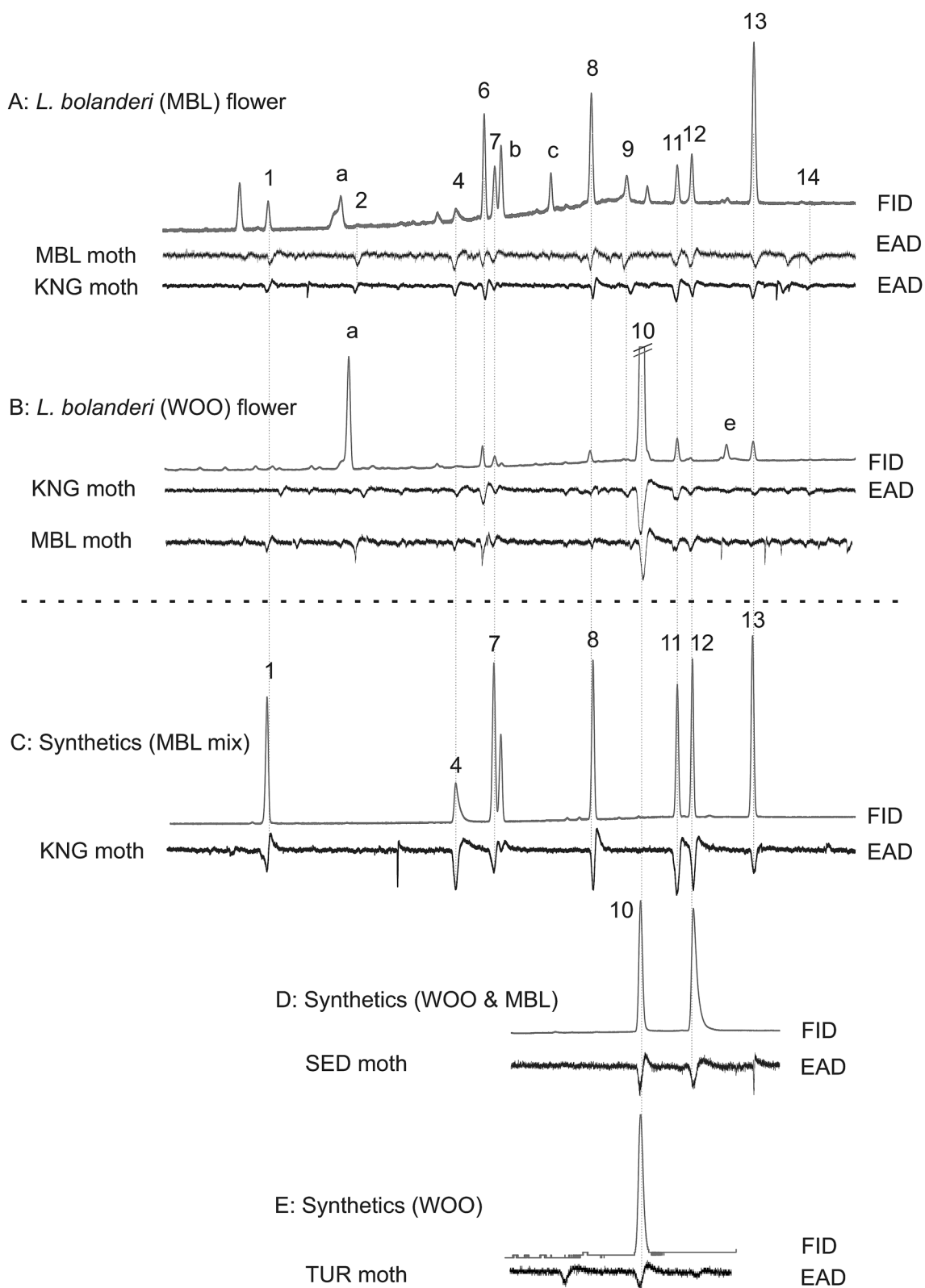
The flower extracts were purified by liquid chromatography (LC) on pre-conditioned 500 mg solid phase extraction column (SPE, Chromabond) using gradient elution in steps of 1% increase of ethyl acetate in pentane (0–15%), each fraction consisted of a volume of 750 µl. SPE conditioning was done with ethyl acetate (1.5 ml), 10% ethyl acetate in pentane (1.5 ml) and pentane (1.5 ml). All SPE fractions were analysed by means of GC-MS and fractions containing 6, 10, 14-trimethylpentadecan-2-one were combined, and solvent were reduced in volume prior to reduction and derivatization.

#### Reduction and derivatization of 6, 10, 14-trimethylpentadecan-2-one (TMP-one) fraction

For *L. bolanderi* and *L. parviflorum* ketone fractions 3–6 were collected (containing mostly ketones), pooled and reduced to alcohol with LiAlH<sub>4</sub> in Et<sub>2</sub>O (1 M solution) and subjected to SPE purification prior to derivatization. SPE fractions were analyzed and 6, 10, 14-trimethylpentadecan-2-ol was found in fractions 8–11. Due to the low concentration of the ketone in the *L. cymbalaria* extract, reduction to alcohol was performed without initial SPE purification. The SPE fractions containing reduced TMP-one were concentrated by means of evaporating solvent under a stream of

**Table 2** GC-EAD active volatiles from floral extracts of *Lithophragma*, and a synthetic, racemic sample of 6,10,14-trimethylpentadecan-2-one (TMP), analyzed in concentration of 100 and 1 µg ml<sup>-1</sup>; only female moths were used; n.p. compound not present in sample. Ui: unidentified compound

<i>Lithophragma</i> species (population)	Moth pop.	Ui 5	Ui 6	Ui 7 RT 16.6 (1-Indanone?)	Ui 8 RT 17.04	TMP	Number of moths
<i>L. cymbalaria</i> (SED)	SED	n.p.	n.p.	n.p.	n.p.	√	4
<i>L. parviflorum</i> (TUR)	TUR	√	√	√	√	√	4
<i>L. bolanderi</i> (MBL)	TUR	n.p.	n.p.	√	n.p.	√	2
<i>L. cymbalaria</i> (SED)	TUR	n.p.	n.p.	n.p.	n.p.	√	2
Synthetic TMP	TUR					√	2
Synthetic TMP	SED					√	2



Argon gas. Dry cyclohexane (50  $\mu$ l) was added to the vial containing the purified extract and subsequently was 1% pyridine in cyclohexane (65  $\mu$ l) and 1% (*S*)-2-acetoxypropionyl

chloride in dichloromethane (100  $\mu$ l) added, the resulting mixture was heated at 80  $^{\circ}$ C for 30–60 min under argon atmosphere (Figure S1). A needle was inserted into the

**Fig. 1** Examples of gas chromatographic analyses with electroantennographic detection (GC-EAD) with different *Lithophragma* floral headspace samples (a, b) and synthetic mixtures (c–e; 100 µg ml<sup>-1</sup>), and antennae of female *Greya politella* moths from different populations. **a** headspace sample of *L. bolanderi* (MBL) tested on sympatric (MBL) and allopatric (KNG) moths; **b** headspace sample of *L. bolanderi* (WOO) tested on allopatric moths (MBL, KNG). The analyses show that moths, irrespective of their origin, respond to the same compounds in the volatile samples. **c** Synthetic *L. bolanderi* (MBL) mix tested on KNG antennae, showing responses to all compounds, of which two (camphor and 2-phenylethyl acetate) do not occur in local KNG flowers (Friberg et al. 2019). **d** 1,4-dimethoxybenzene and 2-phenylethyl acetate tested on SED moths, and **e**: 1,4-dimethoxybenzene tested on TUR moths. In local SED and TUR flowers, both compounds do not occur. Numbers refer to names of EAD-active compounds in Table 1: letters refer to compounds without detectable EAD activity: (a) diacetone alcohol, (b) benzaldehyde, (c) caryophyllene, (d) methyl benzoate, (e) benzyl alcohol. FID flame ionization detector

septum of the vial and the solvent was allowed to evaporate before allowing the vial to reach room temperature. HCl (1 M, 200 µl) and pentane (1 ml) was added, the organic layer was subsequently dried over Na<sub>2</sub>SO<sub>4</sub> (anhydr.) in a Pasteur pipette. Solvent was either reduced to approximately 50 µl and the sample was analyzed as described below or the solvent was evaporated completely and the sample dissolved in cyclohexane prior to analysis.

### Stereochemical analysis of purified extracts

GC-MS in selected ion monitoring mode was utilized for chemical identification of reduced and derivatized stereoisomers of TMP-one. A GC (HP 6890, Agilent Technologies, Santa Clara, CA, USA) coupled to a mass selective detector HP 5973 N MS (Agilent Technologies, Santa Clara, CA, USA) and ChemStation Enhanced Data Program Analysis was utilized for the analysis. The sample (1 µl) was injected in splitless mode at 250 °C. The mass spectrometer was operated under electron impact (EI) ionization, the transfer line was maintained at 250 °C and the mass spectrometer at 250 °C. Selected ion monitoring (SIM) (*m/z* 105, 115, 133, 210, 252) was used for identification of the (*S*)-2-acetoxypropionyl derivatives of 6, 10, 14-trimethylpentadecan-2-ol (Hedenström et al. 2015; Nieberding et al. 2008; Wallin et al. 2020). The GC was equipped with an achiral FactorFOUR VF-23 ms polar GC column from Varian (0.25 mm diameter, 0.25 µm film thickness, 30 m long) for separation of all isomers and helium was used as carrier gas at a flow rate of 1 ml min<sup>-1</sup>. The GC oven was programmed to rise from a starting temperature of 50 °C by 10 °C min<sup>-1</sup> up to 110 °C, from 110 °C by 0.01 °C min<sup>-1</sup> up to 115 °C, and from 115 °C by 10 °C min<sup>-1</sup> up to 230 °C, and held at 230 °C for 10 min.

### Preparation of reference compounds. (2*R*,5*R*,6*R*,10*R*/5*S*)-6,10,14-trimethylpentadecan-2-ol

(2*E*,7*R*,11*R*)-(3,7,11,15)-Tetramethyl-2-hexadecen-1-ol ((2*E*,7*R*,11*R*)-phytol) (Sigma Aldrich) was oxidized at room temperature by sodium periodate and catalytic amount of ruthenium (III) chloride, the obtained ketone was reduced with lithium aluminum hydride in diethyl ether resulting in a mixture of eight isomers of TMP-ol (synthetic scheme Figure S1; Nieberding et al. 2008). The synthetic mixture was derivatized and analyzed according to the same method as the purified flower extracts.

### (2*S*,6*R*,10*R*)-6,10,14-Trimethylpentadecan-2-ol and (2*R*,6*R*,10*R*)-6,10,14-trimethylpentadecan-2-ol

(2*E*,7*R*,11*R*)-phytol (TCI America) was treated as described above, resulting in a mixture of (2*R*,6*R*,10*R*)- and (2*S*,6*R*,10*R*)-TMP-ol. The isomeric mixture was subjected to enzymatic resolution by *Candida antarctica* lipase-B (CALB, Roche Diagnostics batch 90750729) as catalyst and vinyl acetate (Sigma Aldrich) as acyl-donor. With this method the (2*R*)-alcohol is acetylated and therefore the two isomers are easily separated by means of flash chromatography on silica gel with gradient elution with increasing ratio of ethyl acetate in cyclohexane. The obtained (2*R*)-acetate was hydrolyzed in KOH/methanol furnishing the alcohol. A diastereomeric purity of 99.8% for (–)-(2*R*,6*R*,10*R*)-TMP-ol and 97.8% diastereomeric purity of (+)-(2*S*,6*R*,10*R*)-TMP-ol was obtained (Nieberding et al. 2012). The stereoisomerically pure isomers were derivatized and analyzed according to the same method as the purified flower extracts. (Synthetic scheme see Figure S1).

## Results

### EAD-active compounds

*Greya* moths detected a total of 23 volatiles from *Lithophragma* scent samples (headspace and extracts). In all headspace samples, 19 compounds were found to be EAD active (Table 1; Fig. 1). We chemically identified fifteen of these compounds and confirmed their EAD activity using synthetic standards. Four remained unidentified (Table 1).

In the floral extracts, five compounds were consistently EAD active, of which we identified one as 6,10,14-trimethylpentadecan-2-one (TMP), and another one tentatively as 2,3-dihydro-1*H*-inden-1-one (i.e. 1-indanone; Fig. 2; Table 2); three compounds remained unidentified. TMP was found in extracts of all species analyzed, and in the headspace samples of *L. parviflorum* (TUR). Whereas in extract samples it is not sure whether the presence of TMP is an

artefact from the wounding of the tissue, in headspace samples, only non-injured tissue was sampled and thus TMP can be considered as a volatile being emitted in small amounts from intact flowers.

Whereas TMP and 1-indanone have not been documented before in *Lithophragma*, most of the other active compounds represent a subset of the previously detected volatiles in this plant genus (Friberg et al. 2019). We found that most major compounds found in our headspace samples were active, with the notable exception of  $\alpha$ -pinene, which was a main compound in the headspace of *L. parviflorum* (TUR; data not shown; Friberg et al. 2013) but did not trigger EAD responses in moths from the same population. Headspace samples from TUR were not tested with moths from any other population (see Table 1), and thus we do not know whether  $\alpha$ -pinene was active in moths of other populations.

### Divergence of olfactory detection among populations

We did not find any evidence that *Greya* moths responded differently to volatiles from sympatric or allopatric *Lithophragma* species/populations (Table 3), despite the fact that *Lithophragma* species differed greatly in volatile composition (both quantitatively and qualitatively). Generally, all moths responded to all EAD-active major compounds, no matter whether they were present in sympatric *Lithophragma* species or not (Table 3). As the most obvious example, 1,4-dimethoxybenzene (compound no. 10 in Fig. 1) was only present in *Lithophragma* samples from WOO, but proved to be active when those samples were tested on moths from MBL and KNG, and as synthetic compound on moths from SED and TUR (Fig. 1B, D, E).

### Stereoisomeric composition of TMP

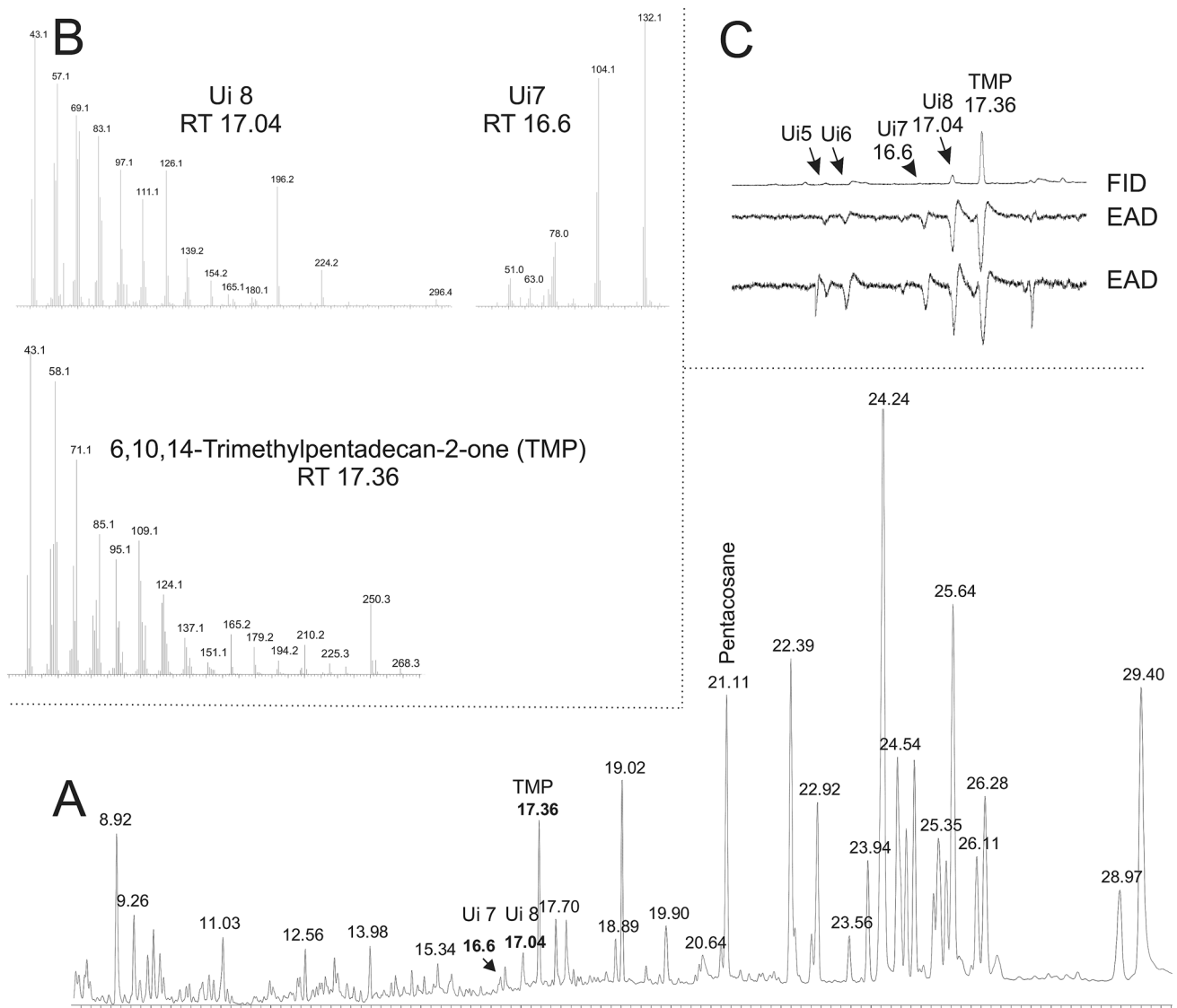
By applying methods developed by previous studies (Eltz et al. 2010; Hedenström et al. 2015; Wallin et al. 2020) we showed that all analyzed *Lithophragma* species produced the (6*R*,10*R*)-enantiomer of TMP-one. There was no difference in enantiomeric composition of TMP in the floral extracts of the different *Lithophragma* species (Fig. 3). The extra peak (RT 379.88) in the chromatogram of *L. bolanderi* (Fig. 3c) is not a match to any of the eight TMP-ol isomers when comparing retention time and evaluating the mass spectra fragmentation pattern (Figure S2). All eight isomers of the derivatized TMP-ol show the same fragmentation pattern in SIM mode, therefore we have strong evidence based on retention time and fragmentation pattern that the extra peak is not a TMP-ol isomer.

## Discussion

Despite the fact that insects specialized to host plants often use a narrow set of uncommon key volatiles for host location, we show here that olfactory detection of *Lithophragma* volatiles in *Greya* moths is generalized. This is evidenced by (i) the fact that moths responded to a large range of different compounds, including different compound classes, and (ii) that *Greya* moths from different regions detect even those volatiles that are not produced by their local *Lithophragma* host species. For example, moths from TUR respond to 1,4-dimethoxybenzene (Fig. 1E), which is not found in the local TUR species *L. parviflorum* (Friberg et al. 2013, 2014). The closest *Lithophragma* that emit this compound are *L. bolanderi* and *L. campanulatum* populations growing 800 km south of TUR. As another example, 2-phenylethyl acetate, which was found in *L. bolanderi* from MBL but not in *L. cymbalaria* from SED, was active in moths from SED (Fig. 1D). These results indicate that preferences in moths for local host species (Friberg et al. 2014) are not mediated by the olfactory detection of species-specific key volatiles in the moths' antennae, but are likely based on pattern recognition of volatile bouquets in the brain, which is a common principle in butterflies and moths (de Fouchier et al. 2017).

Very few studies have addressed variation in olfactory detection among populations or closely related species of insects, but the available studies show that both conservative maintenance of receptor sensitivity (Eltz et al. 2006), as well as rapid divergence (de Bruyne et al. 2010), is possible. Conserved receptor function was found, for example, in males of different euglossine bee species, where broad and overlapping detection of volatiles of tibial extracts was shown by GC-EAD, despite the fact that bees respond behaviorally in a highly specific manner to tibial extracts (Eltz et al. 2006). Similarly, two species of polyphagous fruit and flower feeding beetles of the genus *Pachnoda*, which live in different habitats, share olfactory receptors tuned towards the same volatiles (Bengtsson et al. 2011), and different species of bee- and moth pollinators respond similarly to *Salix*-volatiles (Jürgens et al. 2014). In the genus *Drosophila*, although ester-responsive olfactory neurons are generally conservative and show little functional change, some neuron classes evolve rapidly, showing different responsiveness among species (de Bruyne et al. 2010). Rapid evolution of olfactory receptor genes has also been shown to accompany host specialization in another lineage of *Drosophila* (McBride 2007), likely driven by adaptive evolution (McBride and Arguello 2007). In yet other insects, differences among species may evolve through changes in the number of compounds used for host detection. A recent study on two Oedemeridae beetles of the same genus has shown that EAD-detection of *Fragaria* volatiles differs





**Fig. 2** **a** Total ion chromatogram of the ketone fraction of *Lithophragma parviflorum* (TUR) floral extracts (300 flowers) obtained by SPE fractionation of the crude extract (see methods for details). Retention times of EAD-active compounds are given in bold. **b** MS spectra of three EAD-active compounds found in the sample. **c** The output of GC-EAD analysis of the same sample showing flame

ionization detector (FID) trace and electroantennographic detection (EAD) trace of two antennae of female *Greya politella* moths from TUR. The five active compounds are indicated by arrows and/or retention times. For UI 1 and UI 2 no proper MS spectra could be obtained with GC-MS analysis

among them, with one species detecting more compounds than the other (Blazyte-Cereskiene et al. 2019).

Our study shows that despite population-specific preferences of flower volatiles by moths (Friberg et al. 2014), the underlying qualitative olfactory receptor sensitivity (i.e. responding to compounds or not) of the moths is conserved and does not show population-specific divergence. This lack of divergence is in contrast to morphological traits in woodland stars and *Greya* moths, that diversify in a co-evolutionary manner depending on the nature of their interactions (Thompson et al. 2013). The lack of divergence

in qualitative olfactory receptor sensitivity may stem either from selection on broad olfactory sensitivity or from lack of suitable genetic variation. Selection for broad olfactory sensitivity in interactions may result from plant populations producing complex and variable floral volatile bouquets that may potentially evolve rapidly (Gervasi and Schiestl 2017; Ramos and Schiestl 2020). Under these conditions, reliance on one or a few compounds could be selectively disadvantageous. For example, some populations of *Lithophragma* show polymorphisms or patchy differences in compounds such as 1,4-dimethoxybenzene (Friberg et al. 2019).

**Table 3** Summary of EAD activity of *Lithophragma* volatiles in female *Greya politella* moths of different populations (only chemically identified volatiles are shown)

<i>Greya</i> pop.	Hexenyl acetate	Nonanal	Decanal	Linalool	Camphor	Acetophenone	1,4-Dimethoxybenzene	Methyl salicylate	Phenyl ethyl acetate	Phenyl ethyl alcohol	Dimethyl salicylate	2-Amino benzaldehyde	Methyl anthranilate	Cinnamic alcohol	TMP	No. of moths
SED	n.t.	✓	✓	✓	n.t.	n.t.	✓	✓	✓	n.t.	✓	✓	✓	n.t.	✓	8
MBL	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	10
KNG	✓	✓	✓	✓	✓	✓	✓	✓	✓	✓	n.t.	✓	✓	✓	✓	7
TUR	n.t.	n.t.	n.t.	n.t.	n.t.	n.t.	✓	✓	✓	n.t.	n.t.	n.t.	✓	✓	✓	5

The table shows the detected EAD active compounds, both in natural samples and synthetic compounds. A check indicates EAD activity, red checks indicate activity of compounds that normally do not occur in the plant populations co-occurring with the respective moths. Number of moths is the total number of moths used to detect the activity of all tested compounds; *n.t.* not tested. *TMP* 6,10,14-trimethylpentadecan-2-one

Alternatively, low genetic variability in receptor specificity may constrain the evolution of this trait in *Greya* moths. Our results suggest that the next problem to solve in understanding the mediation of these interactions is why *Lithophragma* species and populations make such complex and variable floral bouquets and why *Greya* remain so conservative in their olfactory receptor sensitivity.

One volatile not previously detected in *Lithophragma*, 6,10,14-trimethylpentadecan-2-one (TMP), was found here in low amounts in headspace samples and extracts, and was highly active in *Greya* olfactory neurons. This compound is a breakdown product of phytol, and commonly found in plant essential oils (Eltz et al. 2010), but rarely detected in headspace samples of flowers, likely because of its relatively low volatility. The enantiomeric composition of this compound has been shown to be important in its attractiveness for euglossine bees (Eltz et al. 2010). All studies that have assessed the enantiomeric configuration of this compound, have detected this compound only in the (6*R*,10*R*)-enantiomeric configuration (Schulz et al. 2011; Wallin et al. 2020), but none so far has investigated geographic variation in enantiomer composition. Our study shows that TMP occurs in the (6*R*, 10*R*)-enantiomeric composition in all studied *Lithophragma* species, suggesting it plays no role in the region-specific preferences of *Greya* moths. Quantitative studies on the variation of relative amounts of different enantiomers may nevertheless show differences among populations that may contribute to preferences of *Greya* moths.

An interesting aspect of TMP and its corresponding alcohol is its function as sex pheromone in diverse groups of Lepidoptera (Hedenström et al. 2015; Schulz et al. 2011). Our finding of EAD activity of this compound in *Greya politella*, a species of the ancestral Lepidoptera lineage of Monotrypsia, suggest that the use of this compound may have originated very early in the evolutionary history of this insect order. TMP may serve pheromonal functions in *Greya*, but our initial GC-EAD analyses with female and male body extracts using male and female antennae failed to detect the compound in the insect samples (data not shown). More detailed studies are deemed necessary, however, to investigate the potential production, or use of TMP

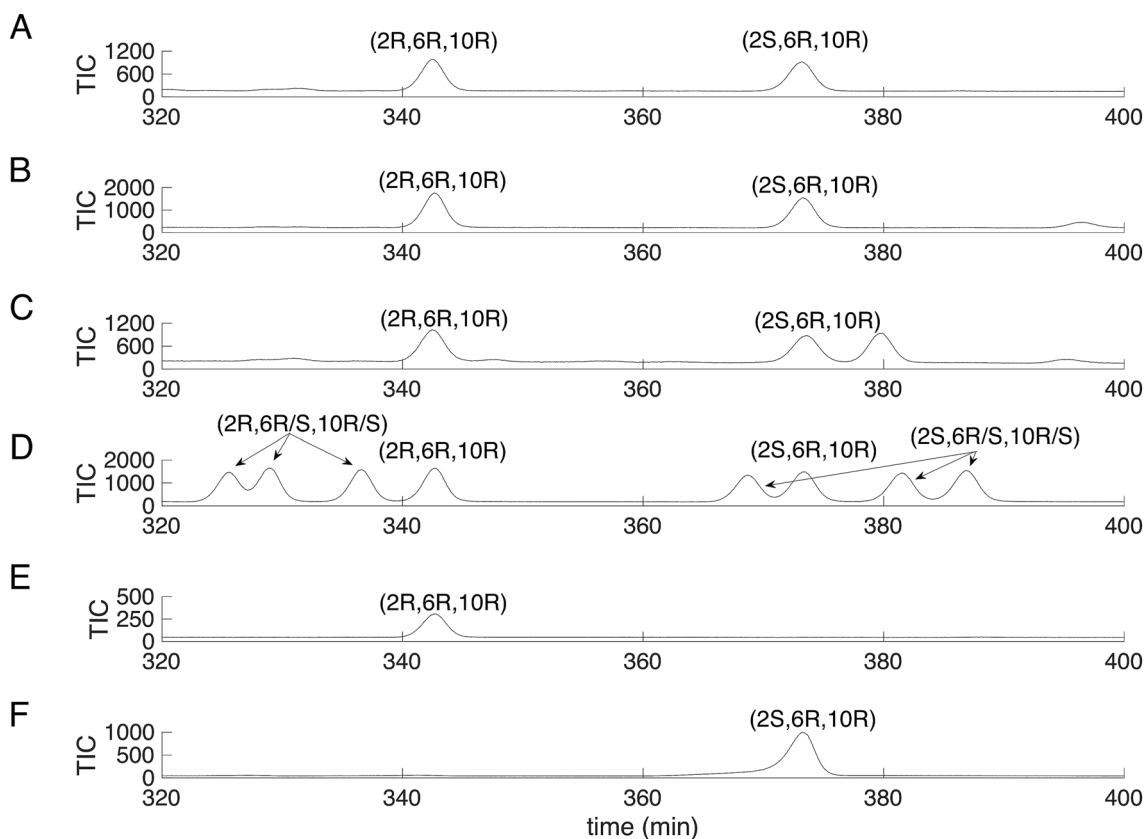
as pheromone by *Greya* moths. Earlier analyses have shown that plant volatiles and pheromones overlap chemically in many insect groups (Schiestl 2010), and the pheromonal functions of TMP in more derived Lepidoptera may have evolved from its initial function as a host detection cue in ancestral lineages.

In conclusion, our study shows that a moth species of the ancestral lineage of Monotrypsia, highly specialized on its host plant, is able to detect a broad range of volatiles, and seems to lack species- or region-specific divergence in olfactory neuron specificity, as detectable by GC-EAD. Our results using the GC-EAD method focus on qualitative differences in detection, but do not allow to precisely quantify differences in receptor sensitivity. Therefore, follow-up studies using more quantitative methods such as EAG or single sensillum recording are needed to assess differences in receptor fine-tuning or abundances of different receptor types in moth antennae. Future studies should also try to elucidate the behavioral activity of individual EAD-active compounds, to identify potential key compounds triggering attraction and oviposition behavior, and potentially mate finding.

**Supplementary Information** The online version contains supplementary material available at <https://doi.org/10.1007/s11829-021-09809-5>.

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**Fig. 3** Selected ion chromatogram of reduced and derivatized **a** *L. cymbalaria*, **b** *L. bolanderi*, **c** *L. parviflorum*, **d** derivatized synthetic mixture of 6,10,14-trimethylpentadecan-2-ol, **e** derivatized synthetic

(2*R*,6*R*,10*R*)-6,10,14-trimethylpentadecan-2-ol and **f** derivatized synthetic (2*S*,6*R*,10*R*)-6,10,14-trimethylpentadecan-2-ol

Jämtland/Härjedalen county and the Swedish Agency for Economic and Regional Growth.

### Compliance with ethical standards

**Conflict of interest** There are no conflicts of interest.

**Consent to participate** Consent is granted.

**Consent for publication** Consent is granted.

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