

# Nd-Zn (Neodymium-Zinc)

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The Nd-Zn phase diagram in [Massalski2] was redrawn from [1972Mas].

Recently, this system was thermodynamically assessed by [2008Li], [2008Qi], and [2009Liu] independently. These assessments used the same experimental phase boundary data of [1972Mas] as the basis of their thermodynamic models. Accordingly the calculated phase diagrams were very similar except handling of the NdZn<sub>11</sub>-NdZn<sub>12</sub> phase. According to [1972Mas], NdZn<sub>12</sub> is the high-temperature modification of NdZn<sub>11</sub>, which has a solid solubility range of approximately 1 at.%. [2008Li] and [2009Liu] assumed only NdZn<sub>11</sub> exists as a line compound, whereas [2008Qi] assumed that NdZn<sub>11</sub> undergoes a polymorphic transformation at about 550 °C.

Figure 1 shows the Nd-Zn phase diagram taken from [2008Li] except the NdZn<sub>11</sub>-NdZn<sub>12</sub> phase area. In this phase diagram, both NdZn<sub>11</sub> and NdZn<sub>12</sub> are shown as line compounds, but the peritectoidal formation temperature of NdZn<sub>11</sub> and the eutectoidal decomposition temperature of

NdZn<sub>12</sub> are taken from [1972Mas]. Although this seems to be the most natural explanation of the existing experimental data, further confirmation is required.

## References

- 1972Mas:** J.T. Mason and P. Chiotti, The Nd-Zn Phase Diagram, *Metall. Trans.*, 1972, **3**, p 2851-2855
- 2008Li:** H. Li, X. Su, Y.L. Liu, Z. Li, and X. Wang, Thermodynamic Assessment of the Nd-Zn System, *J. Alloys Compd.*, 2008, **457**, p 344-347
- 2008Qi:** H. Qi, Z. Jin, L. Liu, and H. Liu, Thermodynamic Assessment of the Nd-Zn Binary System, *J. Alloys Compd.*, 2008, **458**, p 184-188
- 2009Liu:** X.J. Liu, X. Chen, and C.P. Wang, Thermodynamic Assessment of the Sm-Zn and Nd-Zn Systems, *J. Alloys Compd.*, 2009, **468**, p 115-121

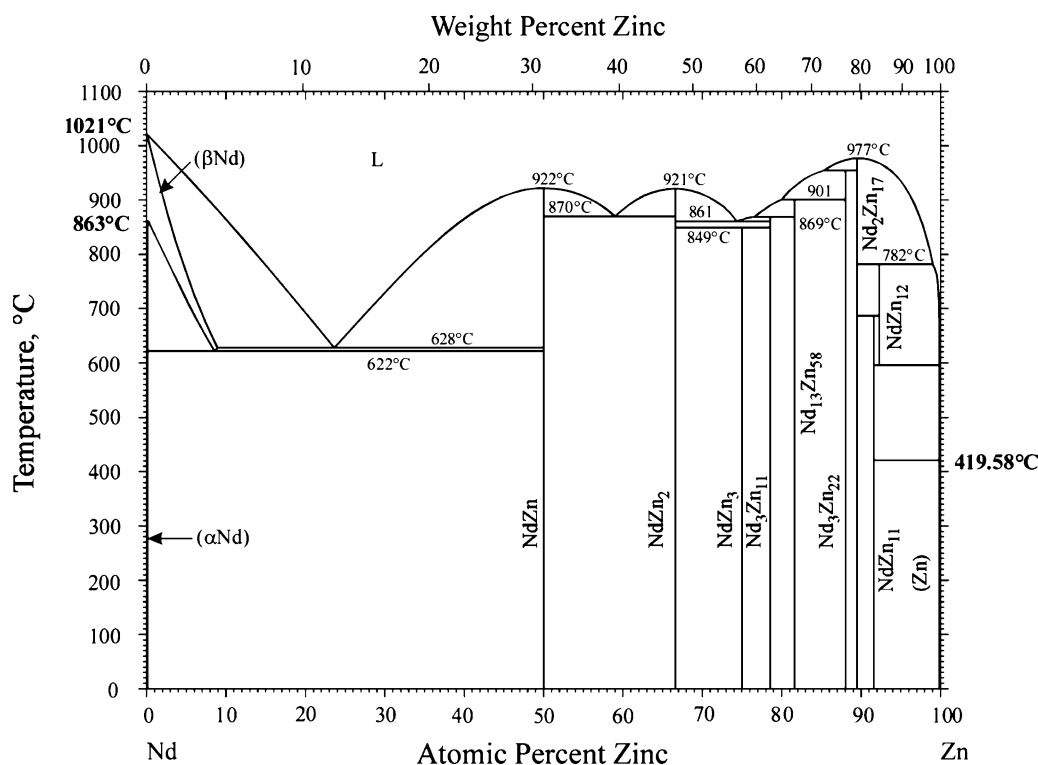


Fig. 1 Nd-Zn phase diagram