

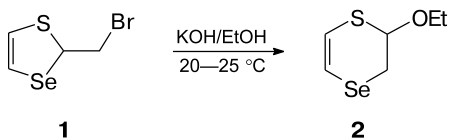
# A rearrangement in the reaction of 2-bromomethyl-1,3-thiaselenole with ethanol: synthesis of 2-ethoxy-2,3-dihydro-1,4-thiaselenine

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We are carrying out systematic investigations aimed at introduction of novel reagents such as selenium dichloride and dibromide into organic synthesis.<sup>1–10</sup> The reaction of selenium dihalides with divinyl sulfide leads to 2,6-dihalo-1,4-thiaselenanes.<sup>3–6</sup> When kept in chloroform solution at room temperature, 2,6-dihalo-1,4-thiaselenanes undergo spontaneous rearrangement with ring contraction to 5-halo-2-halomethyl-1,3-thiaselenolanes.<sup>3–5</sup> 2-Halomethyl-1,3-thiaselenoles were obtained by dehydrohalogenation reaction of 5-halo-2-halomethyl-1,3-thiaselenolanes in high yields.<sup>3,4,8</sup>

We found that treatment of 2-bromomethyl-1,3-thiaselenole (**1**) with ethanolic KOH at room temperature affords hitherto unknown 2-ethoxy-2,3-dihydro-1,4-thiaselenine (**2**) (yield 40%, not optimized) as a nucleophilic substitution and ring expansion rearrangement product.



It should be mentioned, that no formation of the expected 5-membered heterocycle, 2-ethoxymethyl-1,3-thiaselenole as the reaction product was observed. In addition to compound **2**, polymerization products formed.

The heterocycle **2** was isolated by column chromatography on silica gel (hexane as the eluent) as a light-brown oil. Its structure was established by <sup>1</sup>H, <sup>13</sup>C, and <sup>77</sup>Se NMR spectroscopy (including two-dimensional correlation spectroscopy) and chromato-mass spectrometry, its composition was proved by data from elemental analysis. The spin-spin coupling constant of selenium with the carbon atom of the CH<sub>2</sub> group (64.1 Hz) indicates this to be a direct constant and the selenium atom to be linked directly with the CH<sub>2</sub>-group.

The NMR spectra were recorded on a Bruker DPX-400 spectrometer in CDCl<sub>3</sub> at 400.13 (<sup>1</sup>H, HMDS), 100.61 (<sup>13</sup>C, HMDS) and 76.30 MHz (<sup>77</sup>Se, Me<sub>2</sub>Se). The EI-mass spectrum was recorded on a Shimadzu GCMS-QP5050A instrument with the electron energy of 70 eV.

**2-Ethoxy-2,3-dihydro-1,4-thiaselenine (2).** Found (%): C, 34.20; H, 5.00; S, 15.55; Se, 37.63. C<sub>6</sub>H<sub>10</sub>OSSe. Calculated (%): C, 34.45; H, 4.82; S, 15.33; Se, 37.75. <sup>1</sup>H NMR spec-

trum (400.13 MHz, CDCl<sub>3</sub>), δ: 1.25 (t, 3 H, CH<sub>3</sub>, <sup>3</sup>J = 7.0 Hz); 3.07 (dd, 1 H, CH<sub>2</sub>Se, <sup>2</sup>J = 12.1 Hz, <sup>3</sup>J = 5.6 Hz); 3.27 (dd, 1 H, CH<sub>2</sub>Se, <sup>2</sup>J = 12.1 Hz, <sup>3</sup>J = 2.08 Hz); 3.57 (dq, 1 H, OCH<sub>2</sub>, <sup>2</sup>J = 9.5 Hz, <sup>3</sup>J = 7.0 Hz); 3.92 (dq, 1 H, OCH<sub>2</sub>, <sup>2</sup>J = 9.5 Hz, <sup>3</sup>J = 7.0 Hz); 4.94 (dd, 1 H, SCH=, <sup>3</sup>J = 2.1 Hz, <sup>3</sup>J = 5.6 Hz), 6.38 (d, 1 H, SCH=, <sup>3</sup>J = 9.8 Hz); 6.46 (d, 1 H, SeCH=, <sup>3</sup>J = 9.8 Hz). <sup>13</sup>C NMR spectrum (100.61 MHz, CDCl<sub>3</sub>), δ: 14.85 (CH<sub>3</sub>); 25.23 (CH<sub>2</sub>Se, <sup>1</sup>J<sub>Se,C</sub> = 64.1 Hz); 64.46 (OCH<sub>2</sub>); 76.12 (SCHO); 110.74 (SeCH=, <sup>1</sup>J<sub>SeC</sub> = 116.5 Hz); 118.01 (SCH). <sup>77</sup>Se NMR spectrum (76.30 MHz, CDCl<sub>3</sub>) δ: 139 (m, <sup>2</sup>J<sub>Se,CH=</sub> = 51.7 Hz, <sup>3</sup>J<sub>SeCH=CH</sub> = 9.4 Hz). Mass spectrum, m/z (I<sub>rel</sub> (%)): 210 (69) [M]<sup>+</sup>, 164 (8) [M - C<sub>2</sub>H<sub>6</sub>O]<sup>+</sup>, 149 (14) [C<sub>4</sub>H<sub>5</sub>OSel]<sup>+</sup>, 138 (68) [C<sub>2</sub>H<sub>2</sub>SSel]<sup>+</sup>, 129 (49) [C<sub>6</sub>H<sub>9</sub>OS]<sup>+</sup>, 101 (100) [C<sub>4</sub>H<sub>5</sub>OS]<sup>+</sup>, 85 (20) [C<sub>4</sub>H<sub>5</sub>S]<sup>+</sup>, 72 (86) [C<sub>3</sub>H<sub>4</sub>S]<sup>+</sup>, 58 (67) [C<sub>2</sub>H<sub>2</sub>S]<sup>+</sup>, 45 (70) [C<sub>2</sub>H<sub>5</sub>O]<sup>+</sup>.

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