



## Correction to: Catalytic oxidation of hydrogen on platinum

### Thermochemical approach

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**Correction to: J Therm Anal Calorim (2013) 112:815–822**  
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Due to a fault (oversight) of the authors, L'vov BV, Galwey AK, Fig 1 of the article: 'Catalytic oxidation of hydrogen on platinum: thermochemical approach.' (2013) J Therm Anal Calorim 112: 815–822. <https://doi.org/10.1007/s10973-012-2567-0> is accompanied by an oxidation scheme based on the complete atomization of PtO<sub>2</sub>, while the caption for Figure 1 and the main text of the article relate exclusively to the oxidation mechanism based on the primary reduction of PtO<sub>2</sub> with hydrogen. A correct diagram of this mechanism is shown in Fig 1c in an article by the same authors in 'Toward a general theory of heterogeneous reactions:

thermochemical approach' (2013) J Therm Anal Calorim. 113: 561–568. <https://doi.org/10.1007/s10973-012-2754z> and in their article: (2013) Interpretation of the kinetic compensation effect in heterogeneous reactions: thermochemical approach> in Int Rev Phys Chem 32: 515–557. <https://doi.org/10.1080/144235X.2013.822109>.

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