



The Taste of Waste: The Edge of Eggshell Over Calcium Carbonate in Acrylonitrile Butadiene Rubber

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Abstract

Rubber technology experiences a new age by the use of biowaste or natural fillers. In this regard, taking properties of reinforcing agents from biowaste fillers remains as the challenging matter. Chicken eggshell (ES) biowaste has recently been introduced to substitute calcium carbonate (CaCO₃) due to its superior properties and low price. In this work, composites based on acrylonitrile butadiene rubber (NBR) reinforced with ES and CaCO₃ microfillers at various loading levels were prepared and characterized. To improve the interactions between fillers and the NBR matrix, ES and CaCO₃ were surface-functionalized using a terpolymer, namely poly(vinyl 2-pyrrolidone-*co*-maleic acid-*co*-acrylic acid). Fourier-transform infrared spectroscopy (FTIR) and thermogravimetric analysis (TGA) were used to characterize the modified fillers. The incorporation of the functionalized fillers resulted in a significant rise in the maximum torque according to the rheometric measurements. The Young's modulus of the ES-based and CaCO₃-based compounds showed a mild improvement over a wide range of filler contents. The elongation at break of the NBR composites, however, was dependent on the filler content. This work provides exciting opportunities for the design of novel and innovative coupling agents to be used in rubber applications.

Keywords Eggshell biowaste · Surface functionalization · Polymer composite · Rubber technology

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Introduction

Nowadays, disposal of polymer wastes in landfills is a serious challenge in the developing countries, for disposed polymer wastes are permanent contaminants to the local environment including soil and water. This underlines the need for the development of biodegradable polymer composites [1–5]. On the other hand, the competitive market demands a delicate balance between the cost and the performance of the new products. These goals can be partially fulfilled by using natural and waste fillers in polymeric matrix. Past studies have underlined the use of mineral fillers from bio-based and renewable resources including rice husk ash [6, 7], eggshell (ES) powder [8, 9], and cuttlebone [10, 11] to tackle the cost problem of products, at the same time as far as possible promote their properties (Fig. 1).

According to the statistical data from Food and Agriculture Organization of the United Nations (FAO), 6.4 million tons of ES are disposed in landfills worldwide in 2010 alone. This data shows an enormous potential of ES for treatment and reuse in polymer products. ES powder has been used as fertilizer and soil conditioner [12], the adsorbent of heavy metals [13, 14] and sorption site for CO₂ [15, 16]. With a growing interest in academia, ES powder has been employed as a biofiller in several polymer matrices including polypropylene (PP) [17], low-density polyethylene (LDPE) [18], and high-density polyethylene (HDPE) [19]. However, the industrial consumption of ES remained limited. According to the literature, ES contributes by

11% of the total weight of the egg with its primary component being calcite form of calcium carbonate crystal (~94%). Other components of ES include MgCO₃ (~1%), Ca₃(PO₄)₂ (~1%), and organic matter (~4%) [20]. This abundance of calcium carbonate (CaCO₃) in the waste ES introduces an excellent source of bio-mineral CaCO₃ with possible opportunities to replace mineral-based and synthetic CaCO₃ in polymer composites.

ES possesses unique microstructure. The skeleton of CaCO₃ in ES is found to have hierarchical structures; primary nano-particles of ~10 nm loosely to give a porous and rough texture. The observed morphologies include sub-microspheres, nano-rods, and rhombohedral-cleavage pattern and slices that are elaborately arranged in a surface layer by the nature. The palisade layer and mammillary layer lie them along the radial directions. From the Fig. 2 one can conclude that the pore system exhibits a three-level hierarchy: nano-scale pores (between nano-rods and primary nano-particles), sub-micro-scale pores (“bubble pores”), and micro-scale pores (opening of “gas pores”) [21]. Such a structure has a message that ES should have good interaction with polymer resins thanks to its porous microstructure.

The performance of reinforcing agents in polymer systems depends on their compatibility with polymer chains [22–24]. As such, the surface of the particles is usually modified to promote polymer–filler interactions [25–28]. Additionally, surface functionalization by using organic materials, may lower the surface energy value of the mineral phase and reduce the possibility of agglomeration of the particles [29, 30]. In CaCO₃, for example, stearic acid has been frequently used for such purpose [31]. Similarly,

Fig. 1 Common bio-based waste fillers used for the development of biodegradable polymer composites. Amongst various bio-filler resources, widely-used ES exhibits unique properties



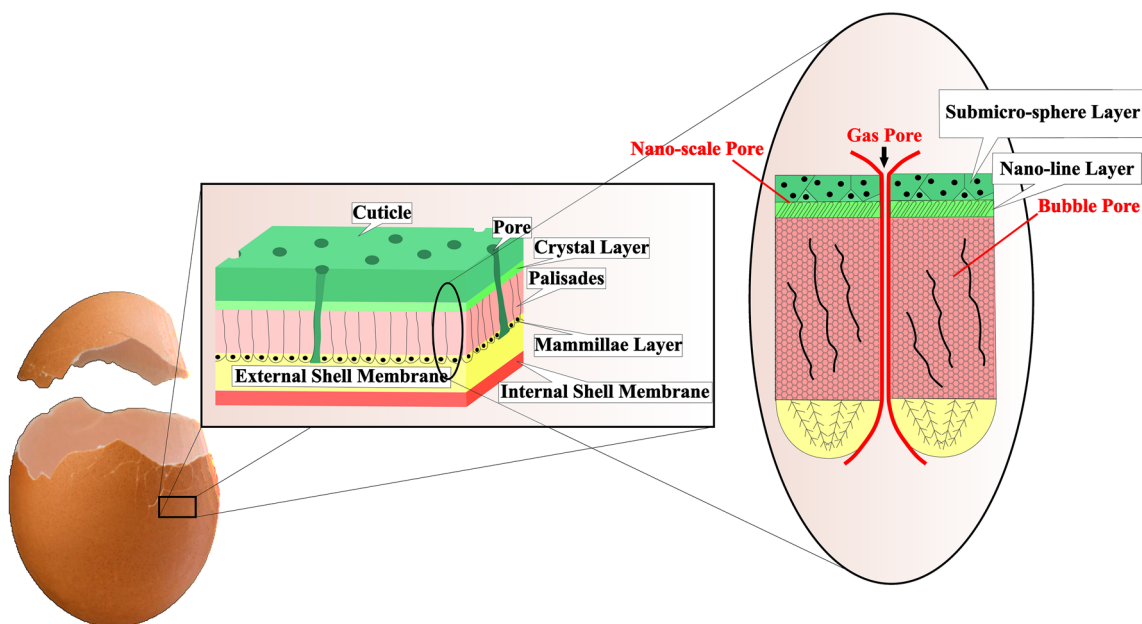


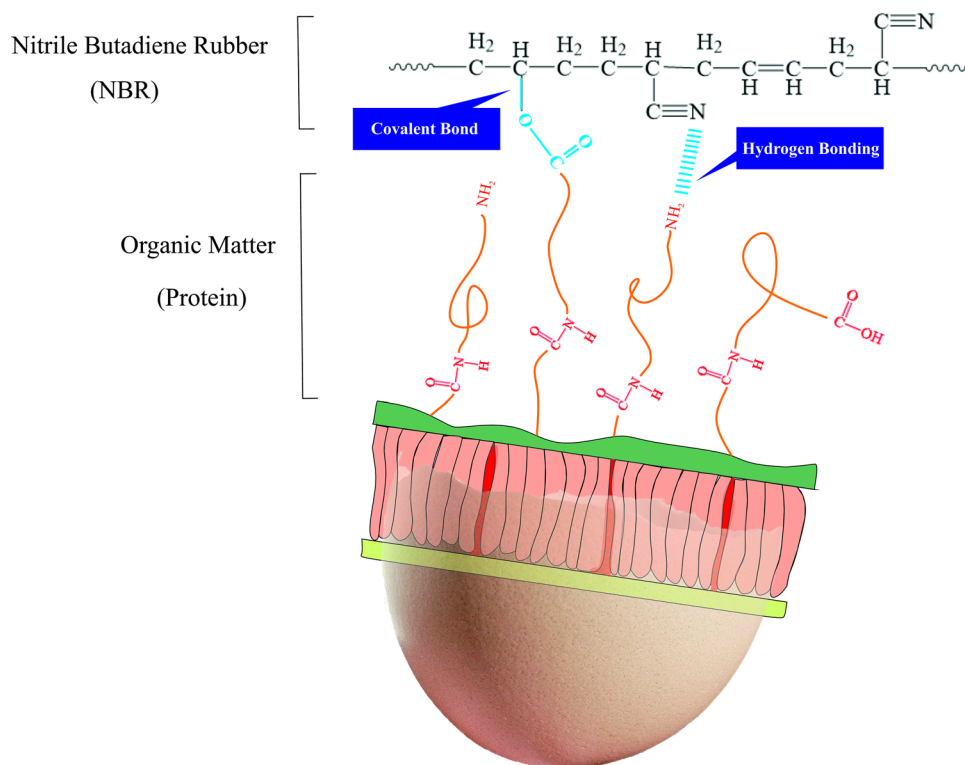
Fig. 2 Illustration of ES structure and its three-level pore system

ES powder has been treated with stearic acid to improve the mechanical properties of polypropylene (PP) composites. The Young's modulus of PP composites containing 40 wt% of treated ES was improved by ~200%, whereas the crystallinity of PP composites containing pristine ES powder remained almost unchanged [32]. Kumar et al. [33] used isophthalic acid in surface functionalization of ES and reported 3–18% improvement in tensile modulus of the PP composite compared to the one containing pristine ES or mineral-based CaCO_3 . Pimelic acid (PA) has also been used to modify ES powder. It was shown that ES/ β -PP biocomposite could be prepared through melt compounding of PP and ES. The modified ES caused high efficiency and selectivity for β -PP formation. The modification of PA promoted the dispersion state of ES in PP matrix by decreasing the interfacial tension. Although the addition of PA-modified ES has slightly deteriorated tensile and flexural properties of PP, it has increased the impact strength of PP by 228% compared to that of pure PP [17]. The ES powder has also been used as a reinforcing filler in elastomers such as natural rubber (NR), acrylonitrile butadiene rubber (NBR), styrene-butadiene rubber (SBR), and epoxidized natural rubber (ENR) [34]. In a previous work, curing characteristics and mechanical properties of NBR, NR, and SBR composites filled with ES powder were discussed [35]. It was revealed that incorporation of ES into rubber matrix decreases the optimum cure time and hereby improves the mechanical properties of NR, NBR, and SBR composites thanks to porous structure of ES

particles compared to CaCO_3 -filled composites [32]. The likely interactions between ES and NBR chains are depicted in Fig. 3.

So far, a growing body of work in the literature has sought to examine the possibility of replacing CaCO_3 with bio-waste ES. The results, although promising, are far from complete. It is important to note that studies in the past have mainly been focused on thermoplastics, and limited reports exist on thermosets and elastomeric composites. In recent studies, some important aspects of surface modification of ES powder towards the elastomeric compounds including NR, SBR, and NBR were discussed [32, 33]. Specifically, the effect of surface modification, filler loading, and particle size distribution on the curing and thermo-mechanical properties of NR, SBR, and NBR rubbers containing CaCO_3 and various types of ES have been comprehensively reported.

In this work, poly(vinyl pyrrolidone-*co*-maleic acid-*co*-acrylic acid) terpolymer was used instead of the conventional use of small molecules in surface modification of the fillers to improve curing behavior and mechanical properties of NBR composites. The efficiency of the surface functionalization procedure using FTIR and TGA measurements were discussed. The correlation between surface functionality, curing behavior and mechanical properties of the compounds were also studied. This work is an attempt towards the manifestation of biowaste ES-based rubber composites for engineering applications.

Fig. 3 Possible formation site for interaction of NBR with ES**Table 1** Average particle size and surface area of pristine and modified fillers studied here

Filler type	Trade name	Average particle size (nm)	Surface area (m ² /g) ^a
CaCO ₃	ALBAFIL® PCC	700	1.95
mCaCO ₃	–	660	2.40
ES	–	690	4.50
mES	–	570	5.55

^aMeasured using BET

Experimental

Materials

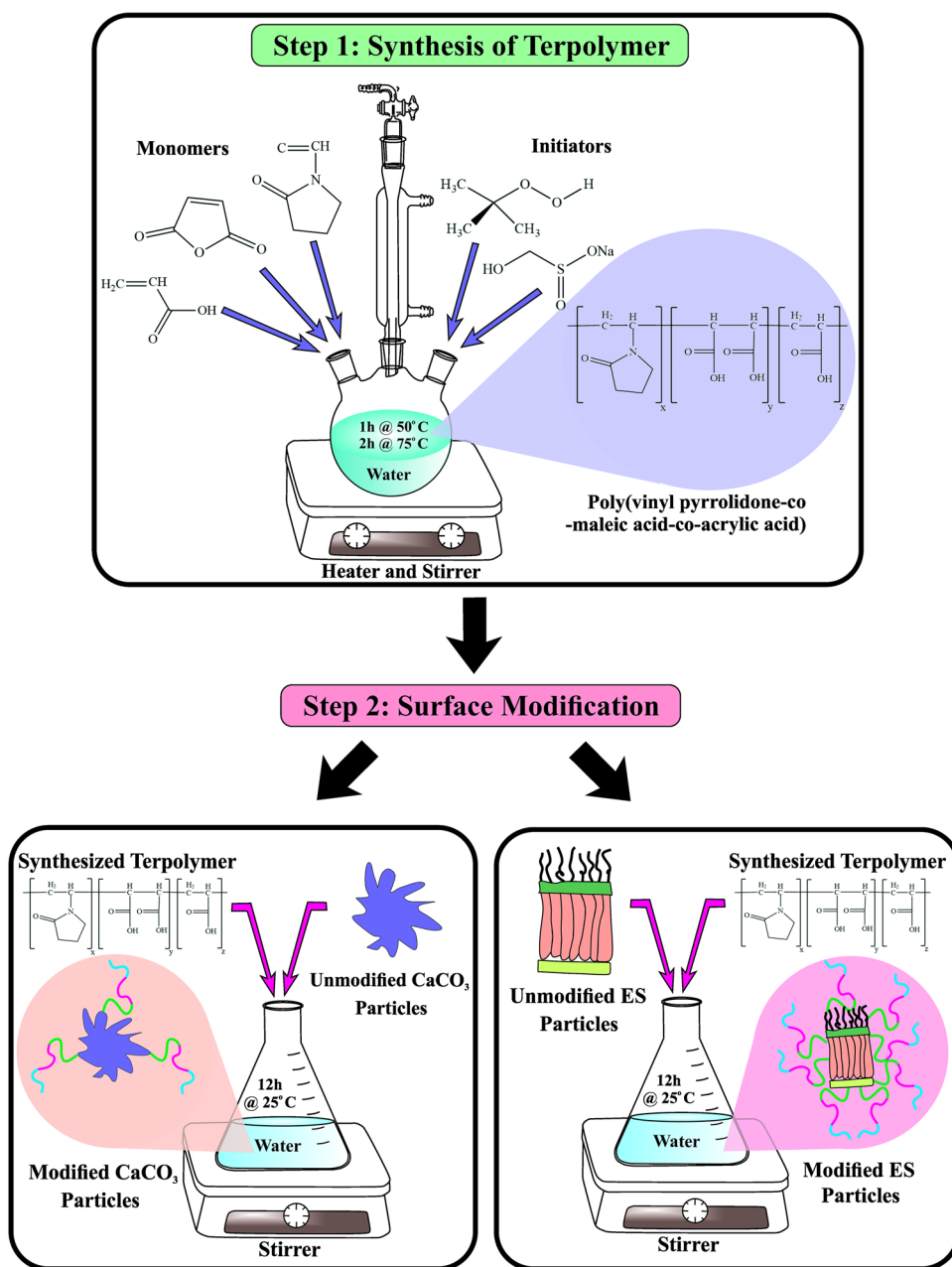
Acrylonitrile butadiene rubber (Perbunan 1846F) known took credit for seals at extremely low (down to $-40\text{ }^{\circ}\text{C}$) and high (up to $120\text{ }^{\circ}\text{C}$) temperatures. It contains 18 wt% acrylonitrile content was provided by Lanxess (Germany). The Mooney viscosity of rubber at $100\text{ }^{\circ}\text{C}$ was 41 ML (1 + 4), and the glass transition temperature (T_g) was $-35\text{ }^{\circ}\text{C}$. Sulfur and mercaptobenzothiazole sulfenamide (MBTS) were used as vulcanizing agent and accelerator, respectively, and obtained from Lanxess (Germany). The activating system including zinc oxide and stearic acid was supplied by Acros Organics (Belgium). The untreated

calcium carbonate GR102066 was purchased from Bayer Chemicals (Germany). The micro-size precipitated CaCO₃ particle with the trade name of ALBAFIL® PCC was provided by the CARY Co. (USA). Chicken ES wastes were collected from local farms in Tehran, Iran. Table 1 summarizes some properties of these fillers.

Surface Functionalization of Particles

Surface functionalization of fillers with terpolymer was done using a two-step procedure (Fig. 4). In the first step, 9.8 g (0.1 mol) maleic anhydride, 7.2 g (0.1 mol) acrylic acid, and 23 g (0.21 mol) vinyl pyrrolidone were added to 60 ml water at $50\text{ }^{\circ}\text{C}$ to prepare a terpolymer by aqueous radical polymerization. Then, 0.08 g TBHP (tert-butyl hydroperoxide) and 0.09 g SFS (formaldehyde sulfoxilate) were dissolved into the resulting solution and stirred for 1 h at $50\text{ }^{\circ}\text{C}$, followed by 2 h stirring at $75\text{ }^{\circ}\text{C}$. The obtained product was a clear solution of 40 wt% solid terpolymer, used for the surface functionalization of ES and CaCO₃. In the next step, 40 g of terpolymer and 100 g of CaCO₃ were dispersed in 450 g of water at room temperature for 12 h with mechanical stirring to graft the terpolymer on the filler surface. Subsequently, the mixture was heated at $80\text{ }^{\circ}\text{C}$ in a vacuum oven without filtration, and the obtained solids were grounded. The same procedure was followed for the ES powder as well. All chemical compounds used for surface treatment were supplied by Merck and used as received without further

Fig. 4 Schematic illustration of the synthesis of poly(vinyl pyrrolidone-*co*-maleic acid-*co*-acrylic acid) terpolymer. This polymer is subsequently used to modify ES and CaCO₃ fillers



purification. The functionalized ES and CaCO₃ will be denoted hereafter as mES and mCaCO₃, respectively.

Preparation of NBR Composites

The NBR matrix was first masticated for 1 min to reduce its viscosity in a laboratory size two-roll mixing mills (Polymix 110L, size: 203 × 102 mm Servitech GmbH, Wustermark, Germany) at 60 °C. Zinc oxide, stearic acid, and the fillers were added to the rubber and mixed for homogenization during 10–15 min depending on the quantity of the fillers. Next, the curing system was mixed and milled with the resulting composites at 40 °C for 10 min before curing at

Table 2 The compounding formulation of the samples

Ingredient	Loading (phr)
NBR	100
Zinc oxide	5
Stearic acid	2
Sulfur	0.5
MBTS	3
Filler	Variable

160 °C under pressure of 150 kN in a compression molding machine to obtain sheets of 2 mm thickness. Table 2 lists the formulation based on 100 phr NBR. We changed the filler

content from 5 to 50 phr to study the performance of the fillers at various contents.

Characterization and Testing

To verify the functionalization of fillers, Fourier transform infrared (FTIR) spectral analysis was utilized using Bruker Vertex 80 V spectrophotometer over the wavelength range of 400–4000 cm^{-1} . To prepare the samples for the test, filler powders were mixed with KBr and pressed into a pellet. Thermogravimetric analyzer TGA Q5000 from TA Instruments was also used to investigate the decomposition temperature and weight loss of the unmodified and modified fillers. The measurements were performed at a heating rate of 10 $^{\circ}\text{C}/\text{min}$ under nitrogen atmosphere up to 600 $^{\circ}\text{C}$. Tensile properties of the samples were determined using a universal testing machine (Zwick 1456, Z010, Ulm, Germany) with DIN S2 dumbbell samples. The strain was applied at a rate of 200 mm/min. In all cases, tensile tests were carried out on five dumbbell specimens. Average values measured and the standard deviations were reported. Investigations of curing characteristics were done using rheometric studies in a moving die rheometer Göttfert Elastograph Vario 67.03. Optimal curing time was considered as the time for the rubber to reach 90% of its maximum torque.

Results and Discussions

Assessment of the Surface Treatment of mES and mCaCO_3

FTIR spectra of the untreated and functionalized particles were measured in order to confirm the success of the surface treatment procedure (Fig. 5). The FTIR spectrum of

the CaCO_3 shows distinct adsorption peaks at 713, 862, 1164, and 1473 cm^{-1} , related to the in-plane bending vibration, out-of-plane bending vibration, symmetric stretch, and the asymmetric stretch of carbonate (CO_3^{2-}), respectively (Fig. 5a). In the case of mCaCO_3 , the peaks at 3400 cm^{-1} are attributed to the hydroxyl ($-\text{OH}$) functional groups of the terpolymer. This peak confirms the surface modification of the filler. The FTIR spectra of pristine ES (Fig. 5b) shows two adsorption peaks at 3400 and 2514 cm^{-1} corresponding to the stretching vibration of amide ($-\text{N}-\text{H}$) and thiol ($-\text{S}-\text{H}$) bonds of cysteine respectively. A typical protein that exists in the mammillary layer of ES. The peaks at 1418, 875, and 712 cm^{-1} correspond to other proteins in the membrane. Notably, FTIR spectra of mES reveal a strong adsorption peak at 3400 cm^{-1} due to the presence of $-\text{OH}$ groups on the filler surface.

Thermal behavior of the samples was performed to further evaluate the surface treatment of fillers. TGA thermograms of ES and CaCO_3 fillers along with a summary of the characteristic features including onset temperature of major decomposing (T_{onset}), first and second maximum decomposition temperature (T_{max}), and char content at 590 $^{\circ}\text{C}$ (%) are shown in Fig. 6. Dotted lines in this figure are derivative if TGA, which confirm occurrence of degradation by showing intensive peaks instead of shoulders in TGA thermograms. Accordingly, two steps in the TGA curve of unmodified ES are recognized: (i) weight loss of pristine ES in range of temperature from 260–400 $^{\circ}\text{C}$ with a maximum rate of weight loss at 300 $^{\circ}\text{C}$ corresponding to decomposition of organic components of the shell membrane, and (ii) de-carbonation of calcium carbonate around 500–600 $^{\circ}\text{C}$. A new peak, at about 160 $^{\circ}\text{C}$, appears for the surface-functionalized ES and CaCO_3 fillers due to decomposition of the terpolymer molecules attached on the surface. The TGA thermograms of mES shows a steeper weight loss compared to mCaCO_3

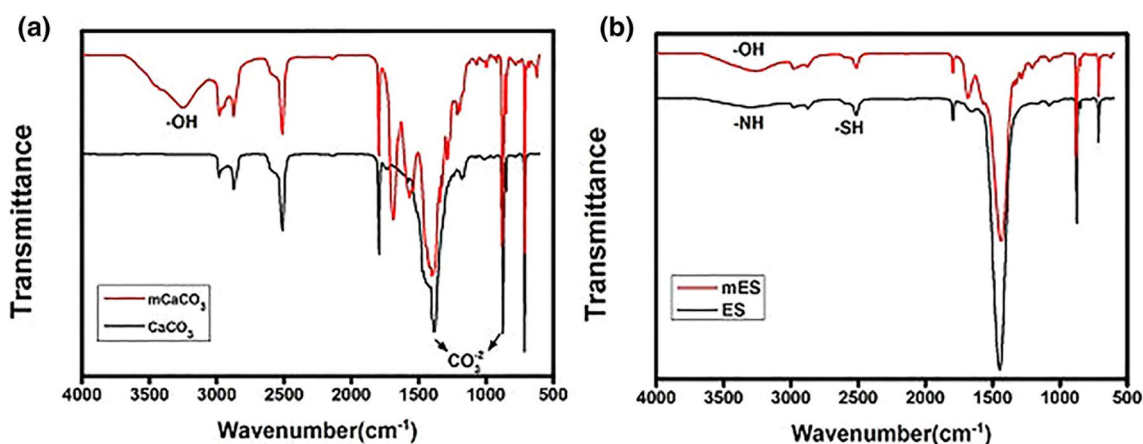
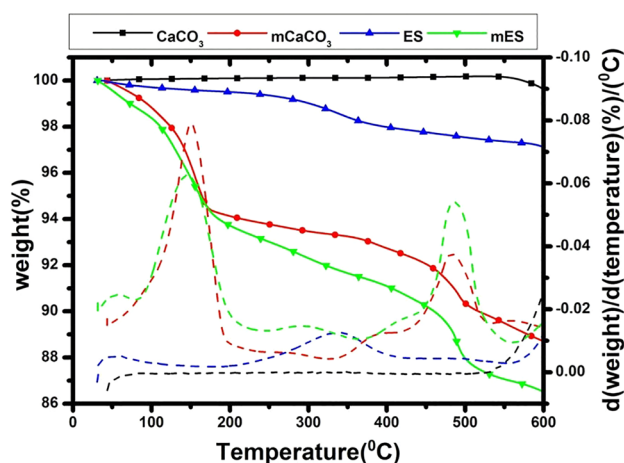


Fig. 5 FTIR spectra of **a** CaCO_3 and mCaCO_3 **b** ES and mES. The peaks around 3400 cm^{-1} correspond to $-\text{OH}$ functional group in mES and mCaCO_3 which confirm the success of the surface treatment procedure



Specimen	T_{onset} (°C)	Maximum Decomposition Temperature (°C)		Residue at 590 °C (%)
		First	second	
CaCO ₃	534			99.64
mCaCO ₃	43	150	483	88.72
ES	213	338		97.13
mES	83	144	486	86.52

Fig. 6 Thermogravimetric curves of ES, mES, CaCO₃, and mCaCO₃. The multi-step weight loss of modified fillers over the heating process confirms the attachment of terpolymer to the filler surface. Dotted lines are derivatives of TGA

due to decomposition of the organic membrane at low temperatures (Fig. 6). The char content of mES (or mCaCO₃) at 590 °C was also lower than pristine ES (or CaCO₃) due to decomposition of terpolymer molecules during the heating. On other words, chemical modification of ES residue along the ES structure which contribute in the decomposition during heating.

Rheometric Cure Characteristics

We hypothesize that the organic components of the ES and the terpolymer used for surface-modification of the filler may influence the cure characteristics of NBR-based composites rubber. To confirm these possible effects, we study the rheometric behavior of NBR composites containing various amounts of fillers (Fig. 7). Overall, the addition of 5 or 10 phr of fillers only marginally affects the cure behavior of the compounds, whereas higher contents of fillers considerably increase the maximum torque and decrease the curing time, especially for mES and mCaCO₃. Furthermore, in the case of mCaCO₃ or mES, the maximum torque increased compared to the unfilled NBR or composites filled with unmodified fillers. This behavior indicates an enhancement

of the cross-link density. It is worthy of note that the curing reaction of NBR matrix occurs faster in the presence of modified fillers, which show better vulcanization characteristics of the modified fillers.

The scorch time t_{10} (the corresponding time when torque reaches to 10% of the maximum value) and the optimum cure time t_{90} were analyzed to quantify the effect of filler type and the filler content on the cure characteristics of NBR composites (Fig. 8). The composites containing the mES and mCaCO₃ fillers show a lower scorch time and a lower optimum cure time compared to those containing unmodified fillers. These results indicate that the terpolymer would likely contribute in the vulcanization kinetics of the composites. In other words, the samples that contain modified fillers cause acceleration of the vulcanization process [36].

Young's Modulus

It is well-documented that the mechanical properties of the vulcanized rubbers are mainly controlled by the type of inclusions, the filler content, and particularly the strengths of the interfacial interactions between the rubber matrix and the fillers. To understand the role of ES and CaCO₃ on the Young's modulus of the NBR composites, we study the mechanical properties of the composites (Fig. 9). Overall, Young's modulus of the composites increase by increasing the filler content, independent of filler type or surface modification. The formation of a filler network above a percolation threshold further improves the Young's modulus at high filler contents [37]. While the Young's modulus of mES containing NBR composites is always higher than the ones filled with pristine ES, the Young's modulus of mCaCO₃ containing systems were lower than the ones filled with the unmodified CaCO₃ (Fig. 9). In general, the small and well-dispersed particles of the ES facilitate an improvement of the Young's modulus [8, 38]. Surface modification of the ES particles improves the filler-rubber interactions leading to higher Young's modulus. The different behaviors of the mES and mCaCO₃ fillers could be explained based on the nature of the interactions between the terpolymer and the surface of the fillers, and the surface topology of the fillers. Fig. 11 illustrates the possible interactions leading to different reinforcement behaviors of NBR composites filled by as-received and modified ES and CaCO₃ fillers. Accordingly, an enhancement in the Young's modulus of NBR occurred upon addition of mES, whereas modification of CaCO₃ adversely affected the Young's modulus. This can be explained by the possible mechanism of reinforcement of mES and mCaCO₃. When incorporating mES into the NBR, hydrogen and covalent bond between mES and NBR chains are formed

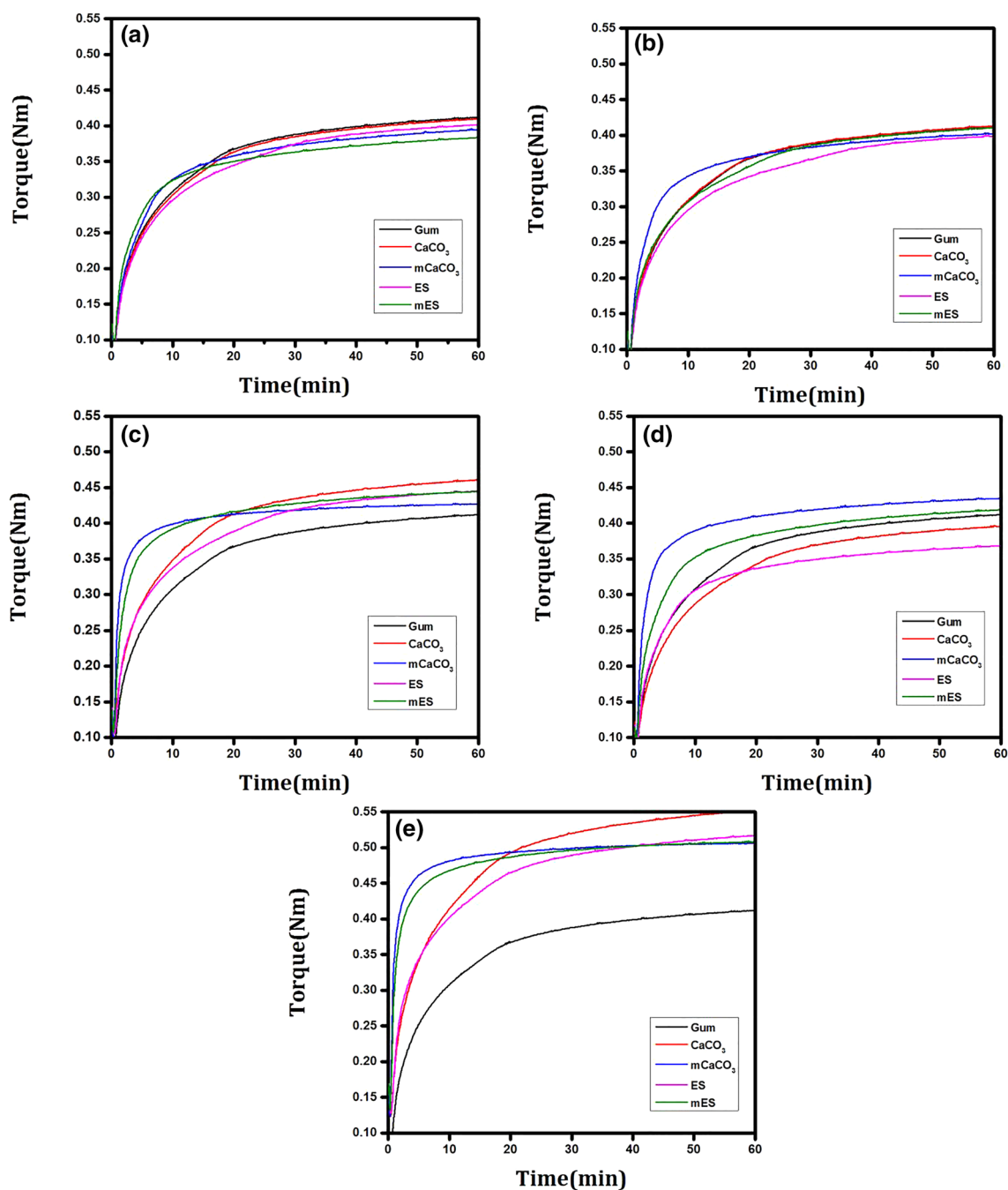


Fig. 7 Rheometric curves of NBR composites containing **a** 5, **b** 10, **c** 15, **d** 30, and **e** 50 phr of different fillers. The maximum torque increases upon introduction of the modified fillers to NBR due to possible intervention of functional groups of the terpolymer with the curing agents

expectedly, which improve the Young's modulus. However, modification of CaCO₃ impeded the formation of van der Waals bonds between the filler and elastomer chains, which inversely decreased Young's modulus compared to pristine CaCO₃. Moreover, it was likely that surface

functionalization of ES might improve the dispersion state of ES in NBR matrix, as evidenced by mechanical properties, while for mCaCO₃ more aggregated clusters were expectedly formed. Understanding the nature of these interactions will be the subject of future studies.

Fig. 8 Scorch times (t_{10}) of **a** CaCO_3 composites and **b** ES composites and optimum cure times (t_{90}) of **c** CaCO_3 composites and **d** ES composites as a function of filler content

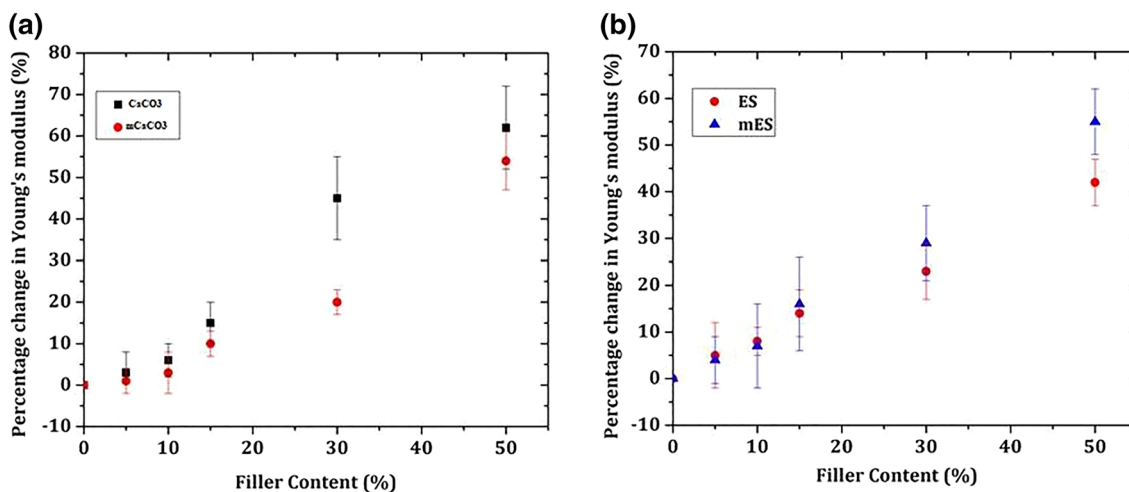
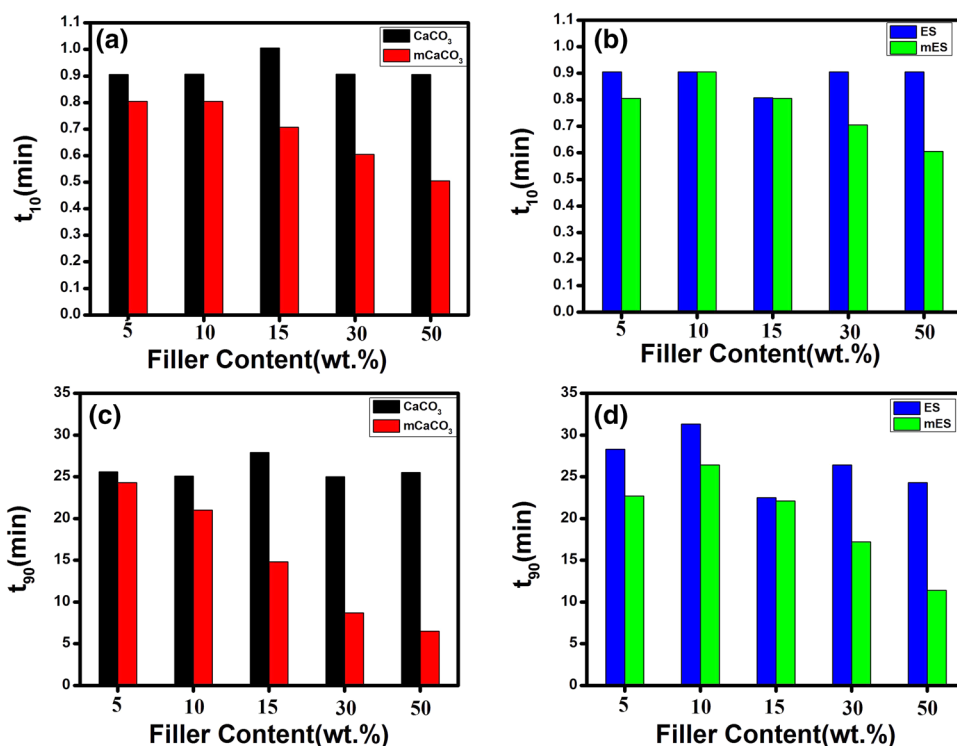


Fig. 9 Percentage change in Young's modulus of NBR composites filled with **a** modified and unmodified CaCO_3 **b** modified and unmodified ES

The reinforcing capability of fillers may be understood using Guth–Gold equation [39, 40], where the improvement in Young's modulus of filled composites is expressed as a function of filler volume fraction, and filler shape:

$$\frac{M}{M_0} = 1 + 0.67f + 1.62f^{2.2} \quad (1)$$

In this equation, M is the Young's modulus of the filled composite, M_0 is the Young's modulus of the unfilled vulcanizate, ϕ is the volume fraction of the filler, and f is the shape factor [35]. The shape factor f for non-spherical fillers is generally higher than unity. To obtain the shape factor f , we fit the Young's moduli of the compounds to the Guth–Gold equation (Fig. 10). Because the surface modification of m CaCO_3 worsens its reinforcing effect,

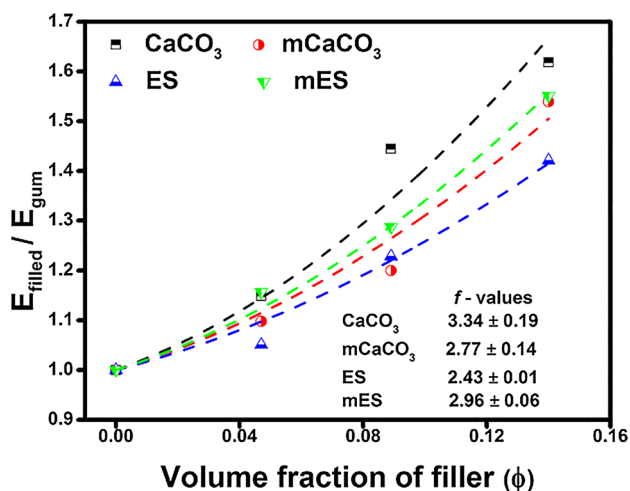


Fig. 10 Guth–Gold plot demonstrating the reinforcing effects of the used fillers. It is obvious that mCaCO_3 negatively influences the reinforcement, whereas mES shows a positive effect

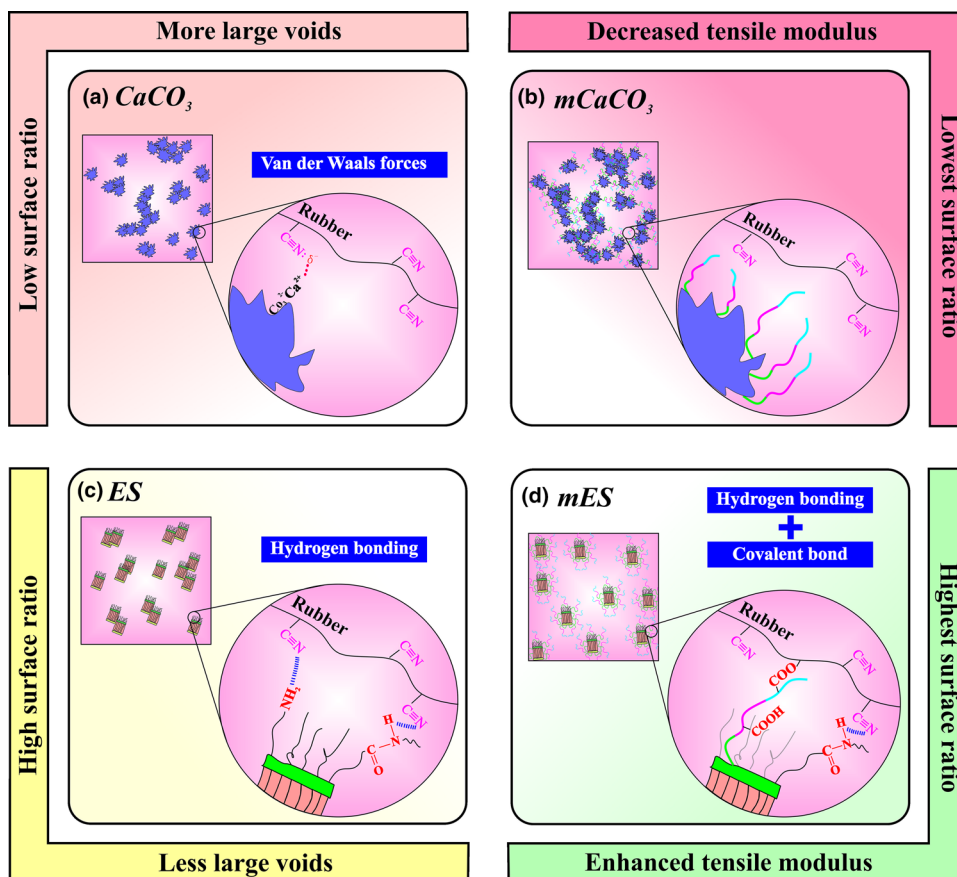
we obtain lower f values for mCaCO_3 compared to pristine filler. This fact agrees well with the trends observed in Fig. 9a. On the other hand, modification of ES has proved to be effective, as there is an increase in f value from 2.4 to 2.96 due to surface modification of filler with terpolymer.

This substantial improvement confirms an efficient dispersion of the mES in the NBR matrix.

Conclusions

In this work, biowaste eggshell and calcium carbonate powders were applied as fillers in preparation of NBR composites. It was observed that chemical modification of the surface of CaCO_3 and ES has a limited or no effect on the mechanical properties (tensile modulus and elongation at break) of NBR composites at low loading levels as confirmed by determination of shape factor of fillers that demonstrates reinforcing effect of the filler, while a completely different effect on high loadings. Modification of CaCO_3 deteriorates elongation at break of NBR, but eggshell surface modification by terpolymer increased interaction between the filler and polymer chains. Such contradictory behavior was attributed to the role of peptide groups in the eggshell and carboxylic groups of terpolymer, which synergistically enhance filler–polymer interaction in terms of hydrogen and covalent bonding in eggshell powder-filler NBR unlike their effects in the case of CaCO_3 -filled rubbers. Thus, it can be corroborated that surface functionalization of biowaste eggshell provides this additive with a very high

Fig. 11 Possible mechanisms for different reinforcement behaviors of composites filled with **a** CaCO_3 , **b** mCaCO_3 , **c** ES, and **d** mES fillers



potential for improvement of mechanical properties of rubber compounds.

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