



# Effect of Silver-Graphene Oxide-Cobalt Oxide Nanocomposite on Cytotoxic Levels in MRC-5 and HepG2 Cell Lines and Molecular Docking Studies

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## Abstract

The cytotoxic properties of cobalt oxide (Co<sub>3</sub>O<sub>4</sub>) nanoparticles (NPs), in addition to graphene oxide (GO)-Co<sub>3</sub>O<sub>4</sub> and silver (Ag)-GO-Co<sub>3</sub>O<sub>4</sub> nanocomposites (NCs), were evaluated against both human healthy lung fibroblast (MRC-5) and hepatocellular carcinoma (HepG2) cell lines utilizing the XTT assay. The investigation revealed that synthesized Co<sub>3</sub>O<sub>4</sub> NPs and NCs (GO-Co<sub>3</sub>O<sub>4</sub> and Ag-GO-Co<sub>3</sub>O<sub>4</sub>) elicited significant cytotoxic responses in MRC-5 and HepG2 cell lines in a concentration-dependent manner. Through molecular docking analyses, it was observed that all fabricated nanomaterials exhibited DNA recognition *via* minor groove binding, with molecular affinities ranging from −4.82 to −11.66 kcal/mol. Furthermore, the docking outcomes illustrated that the angular conformations of GO-Co<sub>3</sub>O<sub>4</sub> and Ag-GO-Co<sub>3</sub>O<sub>4</sub> conferred ‘shape-selective’ characteristics as DNA minor groove binders, leading to heightened cytotoxicity, particularly in the HepG2 cell line compared to the normal MRC-5 cell line.

**Keywords** Cytotoxicity · Nanocomposite · Graphene · Molecular docking · Cobalt oxide · Silver

## Introduction

Recent advancements in nanotechnology have enabled the integration of nanomaterials into diverse fields ranging from technology to biology and medicine. This progress hinges on the precise engineering of nanoparticles (NPs) with tailored properties. The scientific pursuit within nanoscience

focuses on designing NPs with specific functionalities, miniaturizing complex structures, and ultimately, gaining deeper insights into life processes. Notably, the miniscule size of NPs endows them with unique properties that hold immense potential for modulating biological interactions [1]. Cobalt oxide nanoparticles (Co<sub>3</sub>O<sub>4</sub> NPs) are categorized among metal nanoparticles and are widely utilized in both industrial (gas sensors, solar selective absorbers, lithium-ion battery anodes, pigment and dye formulation, as well as in the development of electronic thin films and magnetoresistive devices) and biomedical applications (antibacterial, antiviral, antifungal, antileishmanial, therapeutic, anticancer, and drug delivery agents) [2–8]. Graphene oxide (GO), the oxidized variant of graphene, has garnered significant attention in recent times, particularly in biomedical applications owing to its water stability, biocompatibility, large specific surface area, extensive  $\pi$ -conjugated structures, and abundant functional groups [9, 10]. However, some researchers have found that silver may be hybridized in other ways with other useful materials to create better composite materials. Research efforts are underway to develop GO and reduced graphene oxide (rGO) nanocomposites (NCs) or silver in conjunction with Co<sub>3</sub>O<sub>4</sub> NPs for diverse

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biomedical applications due to their remarkable enhancements in electrical and mechanical properties. These composites exhibit improved stability, mitigate agglomeration issues, and enhance hydrophilicity in aqueous environments [11–15].

Over the past two decades, molecular docking has significantly emerged as a prominent bioinformatics method for predicting optimal binding conformations and energies in interactions between diverse receptors and ligands [16–18]. Despite the rapid proliferation of newly synthesized compounds across various domains of chemistry and industrial chemistry, there remains a lack of understanding concerning their interactions with DNA, the fundamental biomolecule accommodating cellular genetic information and crucial for life's continuation [19]. Thus, in the present study, the intermolecular interactions and binding affinities between the synthesized NPs ( $\text{Co}_3\text{O}_4$ ), and NCs ( $\text{GO-Co}_3\text{O}_4$  and  $\text{Ag-GO-Co}_3\text{O}_4$ ) against the DNA molecule have been investigated using molecular docking simulations in order to understand the molecular basis of induced cytotoxicity as well as to provide a link between theoretical calculations and experiments.

In the current investigation,  $\text{Co}_3\text{O}_4$  NPs and its graphene oxide-based NCs ( $\text{GO-Co}_3\text{O}_4$  and  $\text{Ag-GO-Co}_3\text{O}_4$ ) have been chemically synthesized. All synthesized nanomaterials were characterized using Fourier Transform Infrared Spectroscopy (FT-IR), X-ray Diffraction (XRD), and Transmission Electron Microscope (TEM) analyses. The cytotoxic impact of these nanomaterials was evaluated comparatively at physiologically relevant concentrations on both the human healthy lung fibroblast (MRC-5) cell line and hepatocellular carcinoma cell line (HepG2). Additionally, the interaction modes of synthetic nanomaterials with DNA, their affinities for DNA binding, and the ensuing intermolecular interactions were scrutinized at the molecular level employing molecular docking techniques. This analysis provided insights into the mechanistic underpinnings of experimentally-induced cytotoxicity, as well as the rationale behind utilizing these nanomaterials as potential anti-cancer agents (Scheme 1). Several nanoparticles are presently being assessed in our laboratory as part of our continuous efforts to find and comprehend the mechanisms of action of innovative and useful nanoparticles. Here, the mechanistic aspects of nanomaterials-induced cell death were reported in the MRC-5 and HepG2 cell lines.

## Materials and Methods

The materials, apparatus, and synthesis of nanomaterials used in this study are presented in the [Supplementary Material](#).

## XTT Cell Cytotoxicity Assay

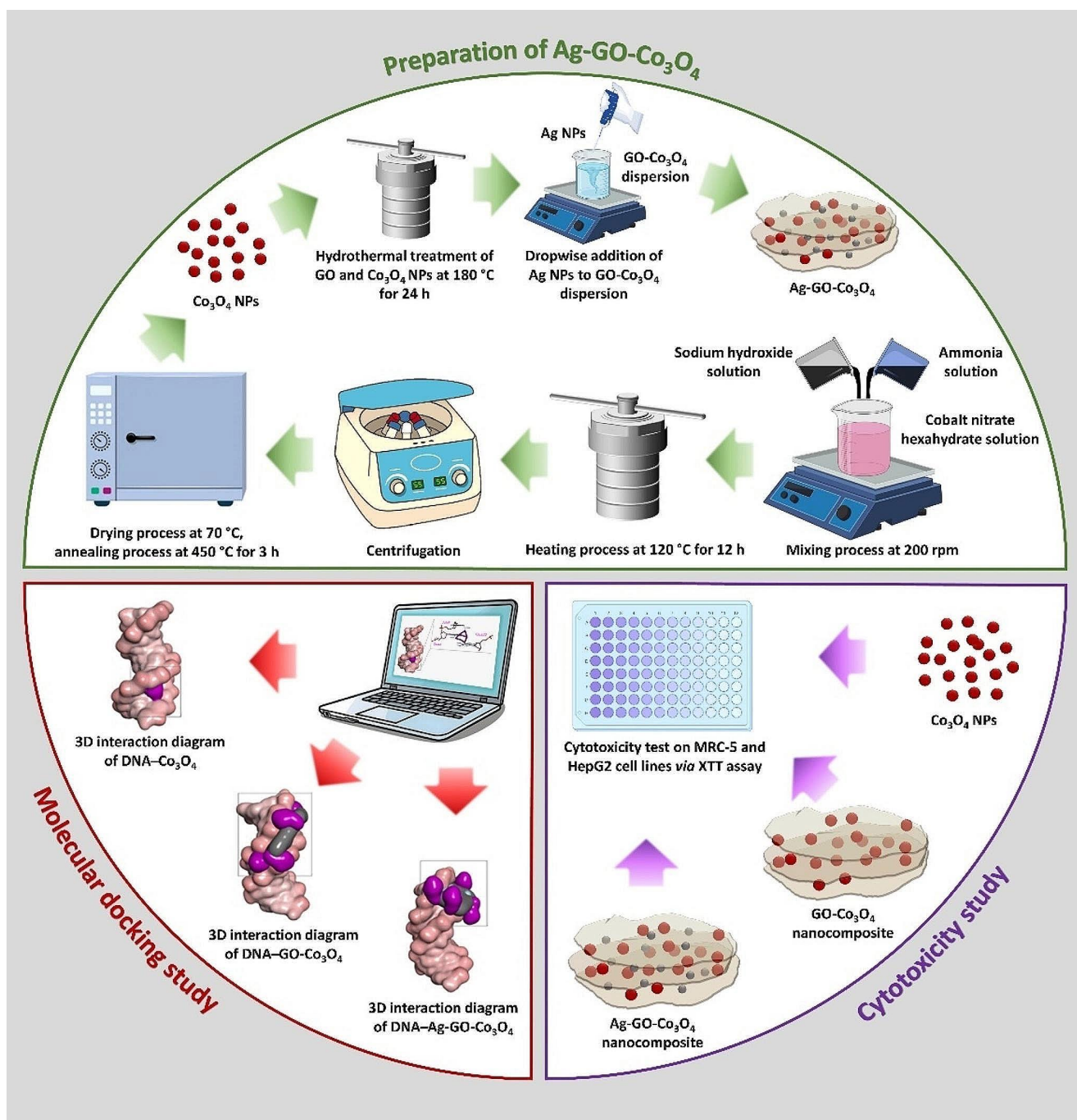
The MRC-5 and HepG2 cell lines were cultured in accordance with ATCC guidelines. They were cultured in Dulbecco's modified Eagle's medium (DMEM) supplemented with 0.1% streptomycin/penicillin (PS-B, Capricorn), L-Glutamine (Capricorn), and 10% fetal bovine serum. Using T75 culture flasks, the cultures were grown in a humidified incubator with a temperature control of 37 °C and 5%  $\text{CO}_2$ . Trypsin-EDTA was used for subculturing at preconfluent densities.

The standard 2,3-bis-[2-methoxy-4-nitro-5-sulfophenyl]-2-H-tetrazolium-5-carboxanilide inner salt (XTT) assay kit was used to quantify the metabolic activity of live cells in order to determine cell proliferation. Initially, 10,000 cells per well of 96-well microplates were used to seed MRC-5 and HepG2 cells. After allowing cells to attach overnight in growth medium, the medium was changed out for new one containing indicated concentrations of  $\text{Co}_3\text{O}_4$ ,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  nanoparticles and nanocomposites (10, 25, 50, 100, 200, and 300  $\mu\text{g}/\text{mL}$ ). To prevent particle aggregation, suspensions underwent sonication for 10 min at 40 W prior to cell treatment. Untreated cells served as controls. In the XTT assay, cell cultures were exposed to the samples for 24 h, followed by the addition of 50  $\mu\text{L}$  of XTT reagent for a further 4-hour incubation period. Metabolically active cells metabolize the XTT reagent, producing an orange formazan dye whose concentration correlates directly with cellular viability. When XTT was reduced extracellularly in untreated cells, the formazan dye product's absorbance level was determined to be 100% vitality. One-way ANOVA followed by the Duncan test was done for the significance study using a significance set at  $p < 0.05$ . Furthermore, the selectivity index (SI) values were computed in the manner described below to ascertain the relative efficacy of nanoparticles and nanocomposites in causing the death of cancer cells relative to the death of normal cells.

$\text{SI} = \text{IC}_{50}$  value of healthy human lung fibroblast cells /  $\text{IC}_{50}$  value of hepatocellular carcinoma cells.

## Receptor and Ligands' Retrieval and Execution of Molecular Docking Simulations Using AutoDock Vina

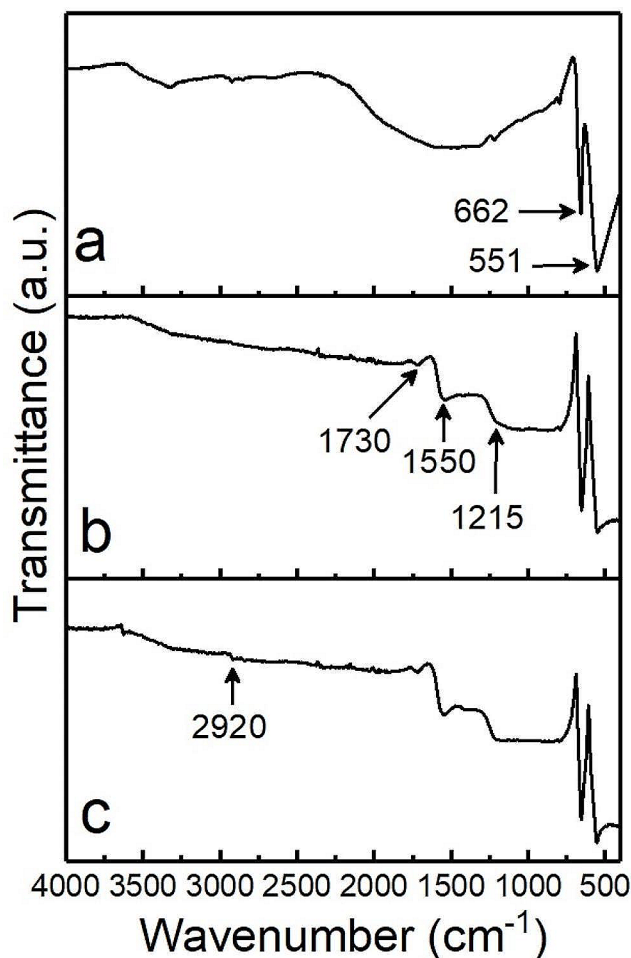
In this present research, molecular scale docking calculations were employed utilizing AutoDock Vina 1.2.3 (latest version) [20]. Therefore, the binding conformations as well as the binding free energy ( $\Delta G$ ) values of nanomaterials obtained as a result of the interactions of cobalt oxide ( $\text{Co}_3\text{O}_4$ ) nanoparticle (NP), graphene oxide—cobalt oxide ( $\text{GO-Co}_3\text{O}_4$ ) nanocomposite (NC), and silver—graphene



**Scheme 1** Schematic illustration of nanomaterials preparation and their application in cytotoxicity and molecular docking studies

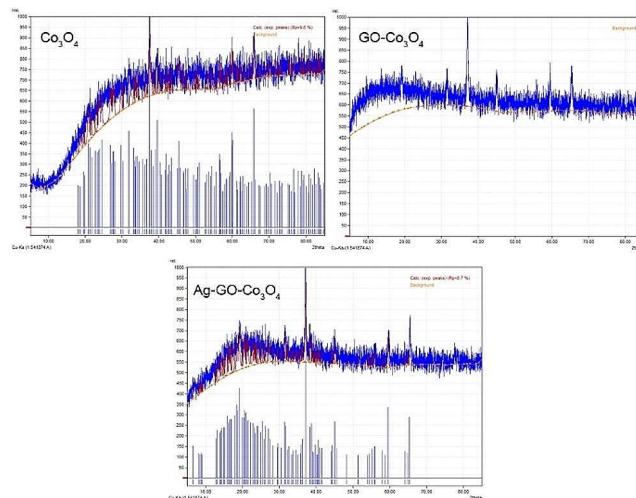
oxide—cobalt oxide (Ag-GO-Co<sub>3</sub>O<sub>4</sub>) NC complexes against the B-DNA dodecamer, crystal structure of a high resolution DNA conformation, were calculated. The crystallographic 12-mer B-DNA dodecamer (PDB ID: 7rqt; resolution: 1.26 Å), utilized as the receptor model in molecular docking experiments, was obtained from the Nucleic Acid Database (NDB, <http://ndbserver.rutgers.edu/>) in the pdb format, whereas the ligands Co<sub>3</sub>O<sub>4</sub>, GO-Co<sub>3</sub>O<sub>4</sub>, Ag-GO-Co<sub>3</sub>O<sub>4</sub> were sketched in 3D using CoreIDRAW Graphics

Suite (version 12.0) and subsequently saved in the mol2 format. Geometry optimization of all ligands were performed using UFF (united-atom force field) in the Avogadro program [21]. Prior to commencing molecular docking simulations, initial preparation involving the setup of both target (B-DNA) and ligand structures, alongside the adjustment of parameters related to docking, was conducted utilizing AutoDock Tools 1.5.6 [22]. During the docking calculations, polar hydrogen atoms were retained on both the target



**Fig. 1** FT-IR spectra of  $\text{Co}_3\text{O}_4$  (a),  $\text{GO-Co}_3\text{O}_4$  (b), and  $\text{Ag-GO-Co}_3\text{O}_4$  (c)

receptor and interacting ligands, whereas non-polar hydrogen atoms were eliminated. The DNA dodecamer, receptor structure, was assigned Kollmann charges, whereas the ligands were assigned Gasteiger charges. In this docking strategy, characterized as semi-flexible (rigid receptor-flexible ligand), the rotatable bonds, if existing, within ligands were permitted to rotate freely during docking trials. This semi-flexible approach enhances precision in forecasting ligand binding orientations and strengths. The dimensions of the grid box, spanning all the major and minor grooves of the target receptor, were constructed as  $62 \times 72 \times 116$  Å points ( $x=15.51$ ;  $y=20.95$ ;  $z=9.74$ ), with a grid resolution set at 0.375 Å. Subsequently, using the AutoDock Tools 1.5.6, the optimized structures of receptor and ligands were then saved in pdbqt format and submitted for docking simulations. For each ligand, we performed 20 distinct docking iterations against the target DNA, employing an *exhaustiveness* setting of '100'. Subsequently, the resultant docking configurations were sorted based on the binding free energy ( $\Delta G$ ; kcal/mol) assigned to each docking pose. Post-docking



**Fig. 2** XRD spectra of  $\text{Co}_3\text{O}_4$ ,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$ .

interactions between the DNA receptor and ligands were analyzed for the most favorable docking conformations, selected from among 20 randomly generated poses for each ligand against the DNA receptor, utilizing the DS Visualizer (version v16) software.

## Results and Discussion

### Characterization of Nanomaterials

The FT-IR spectrum of  $\text{Co}_3\text{O}_4$  displays two absorption peaks at  $662 \text{ cm}^{-1}$  and  $551 \text{ cm}^{-1}$ , which correspond to the Co-O stretching vibrations (Fig. 1a) [23, 24]. In the case of GO (Fig. 1b), the characteristic peak observed at  $1730 \text{ cm}^{-1}$  can be attributed to the C=O stretching vibration of carboxylic acid. Additionally, the bands seen at  $1215 \text{ cm}^{-1}$  and  $1050 \text{ cm}^{-1}$  are associated with the -C-H and C-O stretching vibrations, respectively. Furthermore, the peak observed at approximately  $3400 \text{ cm}^{-1}$  is due to -OH tensile vibration, as GO still contains a small amount of residual water molecules [25]. In the FTIR spectrum of  $\text{Ag-GO-Co}_3\text{O}_4$  nanocomposite (Fig. 1c), slight shifts in the peaks and decrease in peak intensities are thought to be due to the reduction of Ag nanoparticles [26].

Figure 2 shows the raw XRD patterns of the as-prepared  $\text{Co}_3\text{O}_4$ ,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  nanomaterials. As seen in the XRD pattern of  $\text{Co}_3\text{O}_4$  nanoparticles, there are three strong intense diffraction lines at around  $2\theta$  of  $38^\circ$  (311),  $60^\circ$  (511), and  $66^\circ$  (440), respectively [27]. The diffraction peak of GO shifted to approximately  $32^\circ$  in Fig. 2. This indicates a decrease in the  $\text{sp}^2$  carbon fraction [28]. In the composite of  $\text{Ag-GO-Co}_3\text{O}_4$  nanomaterial, the peaks observed at around  $32^\circ$ ,  $45^\circ$ ,  $47^\circ$ , and  $56^\circ$  corresponded to

the (122), (200), (231), and (142) crystallographic planes of the Ag nanoparticles phase [29].

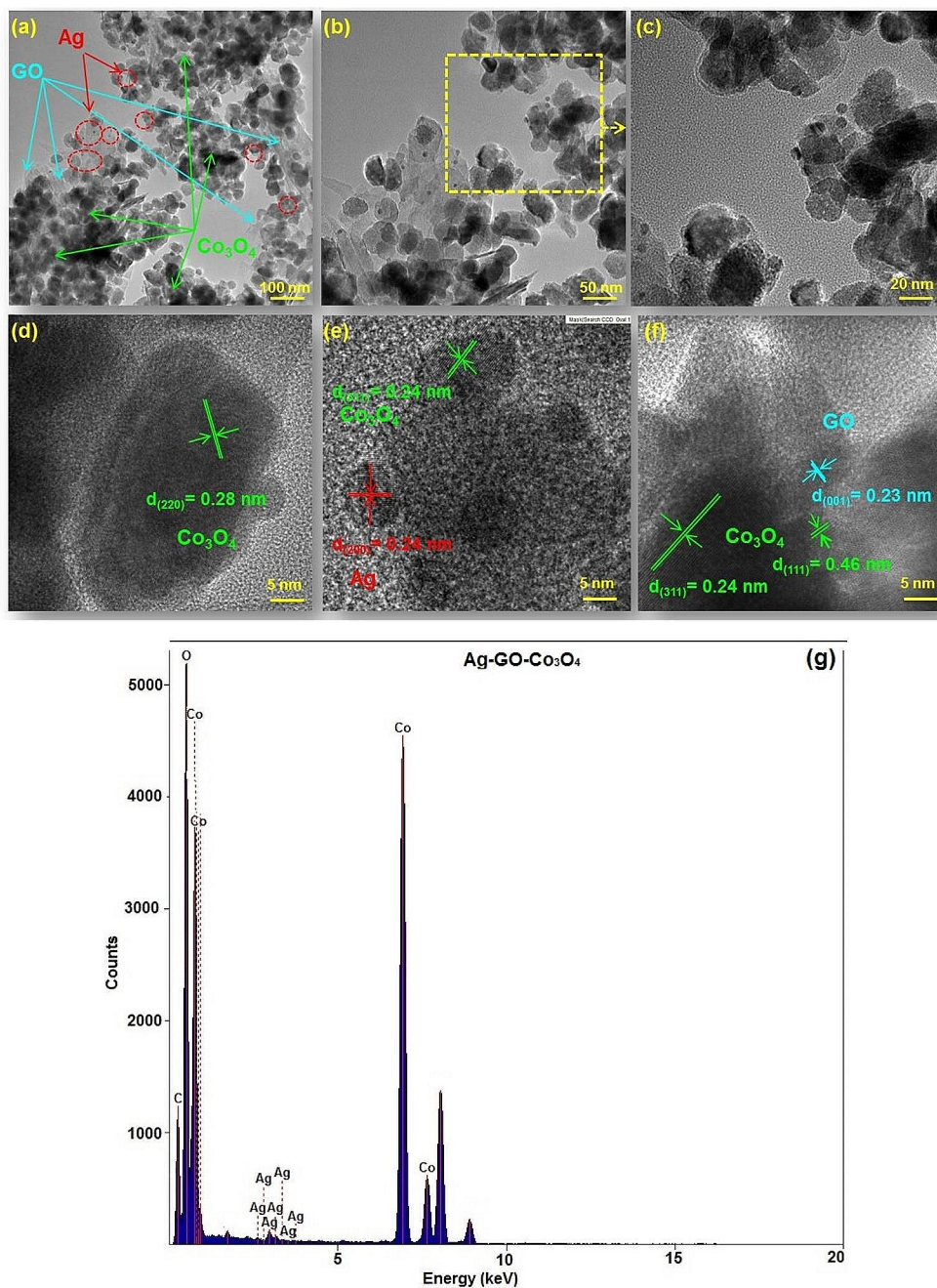
Figure 3 delineates TEM (a, b, c) and HRTEM (d, e, f) photographs of Ag-GO-Co<sub>3</sub>O<sub>4</sub>. The Ag nanoparticles were observed on the Co<sub>3</sub>O<sub>4</sub> nanoparticles and graphene surfaces (Fig. 3a,b) [30, 31]. As shown in Fig. 3d, the lattice plane spacing of 0.28 nm matches well with (220) of cubic Co<sub>3</sub>O<sub>4</sub> [32]. As illustrated in Fig. 3e, the lattice spacing of 0.24 and 0.24 nm correspond to the (200) and (311) crystal planes of Ag and the Co<sub>3</sub>O<sub>4</sub>, respectively [31, 33]. In addition, the d-spacing near the edge is about 0.46 nm and 0.23 nm, which corresponds to the (111) lattice planes of Co<sub>3</sub>O<sub>4</sub> and

the (001) plane of GO in Fig. 3f [34, 35]. Figure 3g indicates EDX spectrum of Ag-GO-Co<sub>3</sub>O<sub>4</sub>. In addition, the EDX result represents the existence of elements Ag, Co, O and C. Moreover, the EDX spectra and TEM photos of Co<sub>3</sub>O<sub>4</sub> and GO-Co<sub>3</sub>O<sub>4</sub> were illustrated in Figure S1 and Figure S2, respectively (Supplementary Information).

### Cytotoxicity Assay on MRC-5 and HepG2 Cells

Three types of nanomaterials (Co<sub>3</sub>O<sub>4</sub> NP, GO-Co<sub>3</sub>O<sub>4</sub> and Ag-GO-Co<sub>3</sub>O<sub>4</sub> NCs) were obtained successively based on cobalt using solvothermal method. The cytotoxic activity

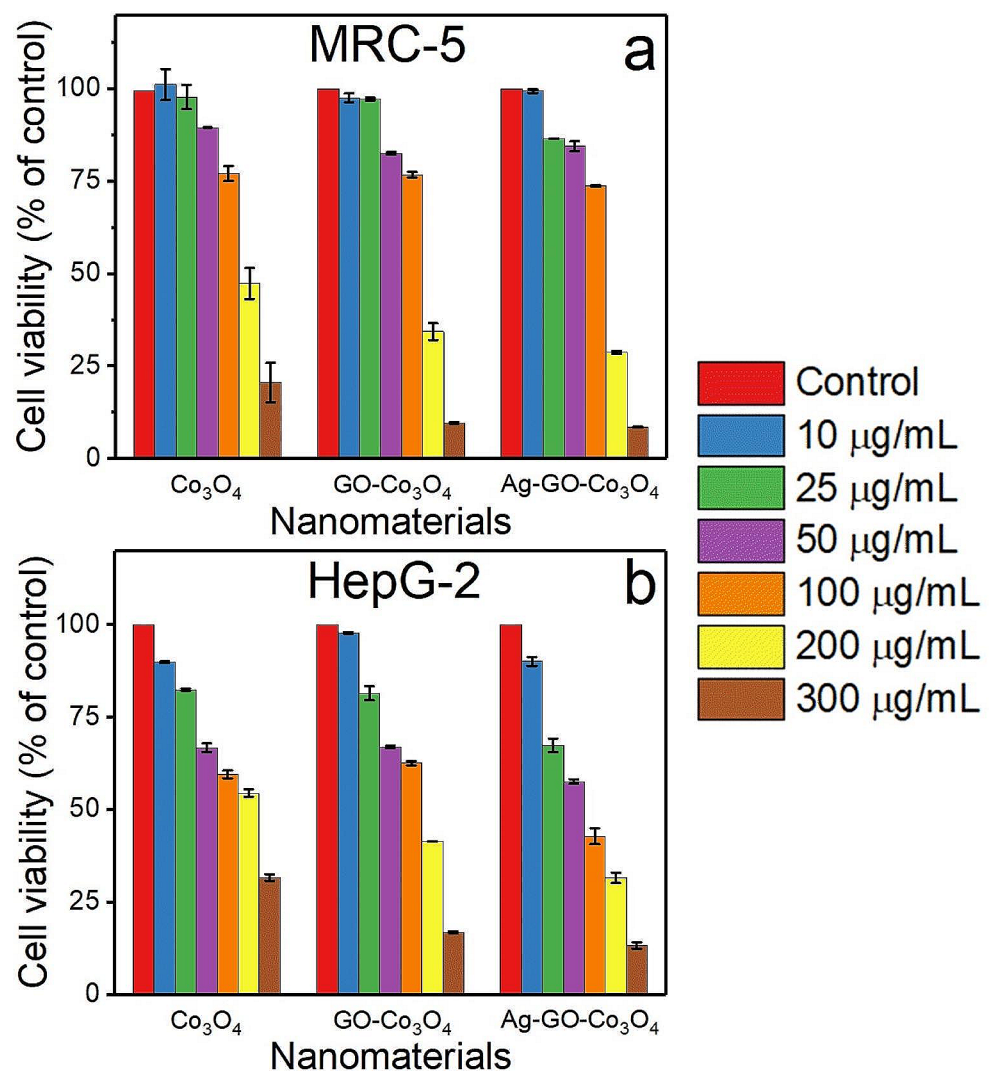
**Fig. 3** TEM (a, b, c) and HRTEM (d, e, f) photographs and EDX (g) profile of Ag-GO-Co<sub>3</sub>O<sub>4</sub>.



of the nanomaterials including  $\text{Co}_3\text{O}_4$ ,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  was evaluated *in vitro* using two human cell lines (MRC-5 and HepG2). Using the XTT assay, the cytotoxic effect of  $\text{Co}_3\text{O}_4$  NP,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  NCs was assessed by monitoring the metabolic activity in living cells. We suspected that  $\text{Ag-GO-Co}_3\text{O}_4$  NC is more cytotoxic on HepG2 cell line compared to MRC-5 lung fibroblast cell line after 24-h treatment. The results showed that all nanomaterials inhibited cell proliferation in a dose dependent manner above  $10 \mu\text{M}$  (Fig. 4). The results also demonstrated that, after a 24-hour treatment,  $\text{Ag-GO-Co}_3\text{O}_4$  NC is more cytotoxic on HepG2 cell line compared to MRC-5 lung fibroblast cell line. The induced cytotoxic effect on HepG2 cell line was found to be greater by  $\text{GO-Co}_3\text{O}_4$  and  $\text{Ag-GO-Co}_3\text{O}_4$  NCs treatments than in  $\text{Co}_3\text{O}_4$  NP treatment alone. In the case of the  $\text{Ag-GO-Co}_3\text{O}_4$  NC especially at high concentrations, a combined synergistic effect was found between Ag, GO and  $\text{Co}_3\text{O}_4$ , leading to the reduction of cell viability and induction of cancer cell death. Furthermore, alongside modifying cell morphology,

the  $\text{Ag-GO-Co}_3\text{O}_4$  NC may reduce cellular metabolic activity, elevates oxidative stress levels through the generation of reactive oxygen species or may induce ROS-independent mitochondrial dysfunction, which led to the reduction of energy metabolism and inhibition of proliferation [36–42, 15, 43–45]. The selective cytotoxic activity  $\text{Co}_3\text{O}_4$ ,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  in human hepatocellular carcinoma (HepG2) cells was compared using a healthy human lung fibroblast (MRC-5) cell line. The  $\text{IC}_{50}$  values of  $\text{Co}_3\text{O}_4$  NP,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  NCs were found to be  $196.92 \pm 9.64 \mu\text{g/mL}$ ,  $165.54 \pm 1.22 \mu\text{g/mL}$ , and  $157.27 \pm 0.58 \mu\text{g/mL}$  in the healthy cells, respectively. Similarly, in the liver cancer cells, the  $\text{IC}_{50}$  values were found to be  $196.92 \pm 0.18 \mu\text{g/mL}$ ,  $162.38 \pm 0.76 \mu\text{g/mL}$ , and  $127.40 \pm 1.13 \mu\text{g/mL}$ . The  $\text{IC}_{50}$  values of nanomaterials on MRC-5 cells were compared with HepG2 cells to evaluate the selectivity index (SI) values. A compound's ability to selectively inhibit proliferation in abnormal cells while avoiding harm in normal cells is shown by its SI value [46]. Among all the synthesized nanomaterials ( $\text{Co}_3\text{O}_4$ ,

**Fig. 4** Effect of nanomaterials on the cell proliferation of MRC-5 (a) and HepG2 (b) cell lines determined by XTT assay after 24-h incubation



**Table 1** IC<sub>50</sub> and SI values of nanomaterials on human normal lung fibroblast and hepatocellular carcinoma cell lines

Nanomaterials	MRC-5	HepG2	SI
	IC <sub>50</sub> (µg/mL)	IC <sub>50</sub> (µg/mL)	
Co <sub>3</sub> O <sub>4</sub>	196.92 ± 9.64a*	196.92 ± 0.18a	1.00
GO-Co <sub>3</sub> O <sub>4</sub>	165.54 ± 1.22b	162.38 ± 0.76bc	1.02
Ag-GO-Co <sub>3</sub> O <sub>4</sub>	157.27 ± 0.58c	127.40 ± 1.13d	1.23

GO-Co<sub>3</sub>O<sub>4</sub>, and Ag-GO-Co<sub>3</sub>O<sub>4</sub>) tested, Ag-GO-Co<sub>3</sub>O<sub>4</sub> NC had an SI value (1.23) which was higher than other tested nanomaterials for the HepG2 cell line (Table 1). We found that based on the calculated SI (> 1.00), Ag-GO-Co<sub>3</sub>O<sub>4</sub> NC could show anti-cancer activity against the HepG2 cells versus healthy MRC-5 cells; thus, according to the IC<sub>50</sub> values, Ag-GO-Co<sub>3</sub>O<sub>4</sub> NC presumed to be non-toxic and bioactive.

\* Different letters<sup>(a-d)</sup> on table are statistically significant at *p* ≤ 0.05. Data were depicted as the mean ± standard deviation (SD), with a sample size of *n* = 3. SI is equal to the IC<sub>50</sub> value of hepatocellular carcinoma cells divided by the IC<sub>50</sub> value of normal human lung fibroblasts.

**Interactions of Synthesized Nanomaterials against DNA**

Table 2; Fig. 5A, B present the target (DNA) affinity, binding conformation, and molecular contacts between the Co<sub>3</sub>O<sub>4</sub> nanoparticle (NP) and DNA, as revealed by the docking simulation. Co<sub>3</sub>O<sub>4</sub> exhibited a preferred binding pose wherein it interacted with the minor groove of DNA. This binding mode involved the formation of multiple hydrogen bonds with G4, A5, and G22, along with Van der Waals contacts with A6, C21, and C23 (Table 2). The binding affinity (Δ*G*; kcal/mol) of Co<sub>3</sub>O<sub>4</sub> NP against the DNA was found energetically favorable (Δ*G*<sub>best</sub> = - 4.82; Δ*G*<sub>average</sub> = - 4.39) (Table 2).

Δ*G*<sub>best</sub>: Binding free energy of the most favorable pose (binding mode).

Δ*G*<sub>average</sub>: The average of the binding free energy values obtained as a result of 20 independent docking runs.

GO-Co<sub>3</sub>O<sub>4</sub>: Graphene oxide—cobalt oxide NC.

Ag-GO-Co<sub>3</sub>O<sub>4</sub>: Silver—graphene oxide—cobalt oxide NC.

Table 2; Fig. 5C, D depict the target (DNA) affinity, binding conformation, and molecular contacts between the GO-Co<sub>3</sub>O<sub>4</sub> nanocomposite (NC) and DNA. It is evident that GO-Co<sub>3</sub>O<sub>4</sub> fits snugly within the minor groove of the DNA fragment in its optimal binding conformation. This conformation shows numerous hydrogen bond interactions with A6, T7, T20, G22, and C23, along with a limited number of Van der Waals contacts with G4, T19, and G24, electrostatic interactions with C21 and G22, and a hydrophobic pi-alkyl interaction with A5 (Table 2). The computed binding energy of GO-Co<sub>3</sub>O<sub>4</sub> NC with the DNA fragment demonstrates a thermodynamically highly favorable state (Δ*G*<sub>best</sub> = - 11.66; Δ*G*<sub>average</sub> = - 11.29) (Table 2).

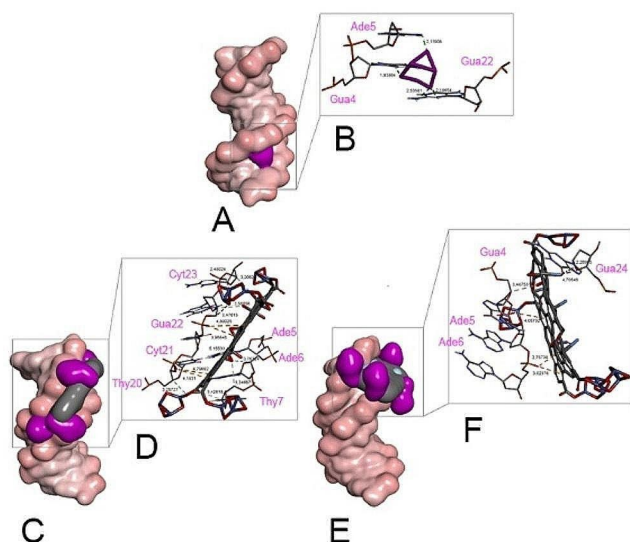
Table 2; Fig. 5E, F provide insights into the target (DNA) affinity, binding conformation, and molecular contacts of the Ag-GO-Co<sub>3</sub>O<sub>4</sub> nanocomposite (NC) with DNA. Similar to the GO-Co<sub>3</sub>O<sub>4</sub> NC, the Ag-GO-Co<sub>3</sub>O<sub>4</sub> NC fits snugly within the minor groove of the DNA fragment, predominantly positioning its Co<sub>3</sub>O<sub>4</sub> components upwards (Fig. 5E). It forms hydrogen bonding with G4, numerous Van der Waals contacts with G2, C3, G22, and C23, electrostatic interactions with A5 and A6, and ultimately, a metal-acceptor interaction with G24 (Table 2). Despite being slightly lower compared to the GO-Co<sub>3</sub>O<sub>4</sub> NC, the interaction energy derived from the docking of Ag-GO-Co<sub>3</sub>O<sub>4</sub> NC with DNA remains thermodynamically highly advantageous (Δ*G*<sub>best</sub> = - 11.51; Δ*G*<sub>average</sub> = - 9.51) (Table 2).

It can be inferred that there exists a certain degree of correlation between the calculated IC<sub>50</sub> values and docking scores for Co<sub>3</sub>O<sub>4</sub> NP, GO-Co<sub>3</sub>O<sub>4</sub> NC, and Ag-GO-Co<sub>3</sub>O<sub>4</sub> NC in the MRC-5 cell line. As the molecular weight of synthetic ligands increases (Table 3), there is an observed escalation in the cytotoxic response manifested in MRC-5 healthy cells, attributed to their DNA-binding strengths (Table 2), supported by the decrease in IC<sub>50</sub> values (Table 1).

In the HepG2 cell line, there is generally observed a lower correlation between the docking scores and IC<sub>50</sub>

**Table 2** The molecular docking analyses provided data on the binding free energies (Δ*G*) and the interactions between the DNA nucleotides and cobalt oxide (Co<sub>3</sub>O<sub>4</sub>) nanoparticles (NPs), as well as the GO-Co<sub>3</sub>O<sub>4</sub> and Ag-GO-Co<sub>3</sub>O<sub>4</sub> nanocomposites (NCs).

Compound	Receptor	Δ <i>G</i> <sub>best</sub> (kcal/mol)	Δ <i>G</i> <sub>average</sub> (kcal/mol)	Binding mode	H-bonds (classic and carbon-hydrogen)	Van der Waals	Electrostatic	Hydrophobic (pi-alkyl)	Other (metal-acceptor)
Co <sub>3</sub> O <sub>4</sub>	B-DNA	-4.82	-4.39	Minor groove	G4, A5, G22	A6, C21, C23	-	-	-
GO-Co <sub>3</sub> O <sub>4</sub>	B-DNA	-11.66	-11.29	Minor groove	A6, T7, T20, G22, C23	G4, T19, G24	C21, G22	A5	-
Ag-GO-Co <sub>3</sub> O <sub>4</sub>	B-DNA	-11.51	-9.51	Minor groove	G4	G2, C3, G22, C23	A5, A6	-	G24



**Fig. 5** The graphical representations in Figs. A, C, and E illustrate the molecular surface and binding conformations of the top-ranked docking complexes formed by various synthetic nanomaterials in complex with double-stranded (ds) DNA, accompanied by three-dimensional (3D) nucleotide interaction diagrams for each synthetic compound with DNA, as shown in B, D, and F. Specifically, Figs. (A, B) depict the binding mode and 3D nucleotide interaction diagram of  $\text{Co}_3\text{O}_4$  with DNA; (C, D) exhibit the binding mode and 3D nucleotide interaction diagram of  $\text{GO-Co}_3\text{O}_4$  NC with DNA; and (E, F) present the binding mode and 3D nucleotide interaction diagram of the  $\text{Ag-GO-Co}_3\text{O}_4$  nanocomposite with DNA. It is noteworthy that all synthetic nanomaterials snugly accommodate within the minor grooves of DNA, indicative of their shape-selective affinity towards DNA. Visualizations were generated using the DS Studio v16 software package

**Table 3** Calculated lowest energies of nanoparticles and nanocomposites based on the Universal Force Field (UFF).

Ligand	Molecular weight (g/mol)	Intramolecular energy (kJ/mol)
$\text{Co}_3\text{O}_4$	240.799	846.1
$\text{GO-Co}_3\text{O}_4$	2090.16	8547
$\text{Ag-GO-Co}_3\text{O}_4$	2627.49	12747.4

$\text{Co}_3\text{O}_4$ : Cobalt oxide nanoparticle

$\text{GO-Co}_3\text{O}_4$ : Graphene oxide—cobalt oxide nanocomposite

$\text{Ag-GO-Co}_3\text{O}_4$ : Silver—graphene oxide—cobalt oxide nanocomposite

values compared to the healthy (MRC-5) cell line (Tables 1 and 2). However, despite the  $\text{Ag-GO-Co}_3\text{O}_4$  ligand exhibiting slightly weaker DNA-binding energy compared to  $\text{GO-Co}_3\text{O}_4$  (Table 2), this data does not correlate with the  $\text{IC}_{50}$  value of the  $\text{Ag-GO-Co}_3\text{O}_4$  NC and the cytotoxic effect of  $\text{Ag-GO-Co}_3\text{O}_4$  NC in the HepG2 cell line was found greater than that of  $\text{GO-Co}_3\text{O}_4$  (Table 1; Fig. 4b). Moreover, an association between molecular energy and cellular toxicity levels (particularly in HepG2 cells) could be anticipated. The calculated intramolecular energy ranking

based on molecular configuration has been found as:  $\text{Ag-GO-Co}_3\text{O}_4 > \text{GO-Co}_3\text{O}_4 > \text{Co}_3\text{O}_4$  (Table 3). Hence, the nanocomposite  $\text{Ag-GO-Co}_3\text{O}_4$ , characterized by the lowest stability, demonstrates notably increased cytotoxic effects, particularly on the HepG2 cancer cell line. This could be expected since chemically unstable and bulky molecules often induce toxicity by binding to biological macromolecules, including DNA and proteins [47].

In the MRC-5 cell line, both the NP and the two NCs exhibit toxicity, starting from the higher concentrations of 200 and 300  $\mu\text{g}/\text{mL}$ . However, promisingly, in the HepG2 cell line, the NP and two NCs exert toxicity starting from concentrations as low as 50  $\mu\text{g}/\text{mL}$ , escalating at concentrations of 100, 200, and 300  $\mu\text{g}/\text{mL}$  (Fig. 4a, b). Hence, it can be inferred that the toxic effects of  $\text{Co}_3\text{O}_4$ ,  $\text{GO-Co}_3\text{O}_4$ , and  $\text{Ag-GO-Co}_3\text{O}_4$  on the HepG2 cell line could be differential and more potent. It is essential to highlight that owing to their rigid planar structures and continuous occupation of the minor grooves of DNA, the interaction between these engineered synthetic nanomaterials and DNA likely follows a *lock-and-key* mechanism. This mechanism primarily relies on three fundamental elements: hydrogen bonding, Van der Waals contacts, and electrostatic interactions, which align with our findings from docking interaction results (Table 2) [48, 49].

## Conclusion

In this study, it was observed that cell death increased in a dose-dependent manner across two examined cell lines (except 10  $\mu\text{g}/\text{mL}$  of  $\text{Co}_3\text{O}_4$  in MRC-5 cell line). The toxicity of  $\text{Ag-GO-Co}_3\text{O}_4$  nanocomposite especially at high concentrations tested was observed to be greater compared to other nanomaterials. In molecular docking experiments, it has been observed that synthetic nanoparticles ( $\text{Co}_3\text{O}_4$ ) and nanocomposites ( $\text{GO-Co}_3\text{O}_4$  and  $\text{Ag-GO-Co}_3\text{O}_4$ ) energetically form highly favorable complexes with the 12-nucleotide high resolution B-DNA dodecamer through minor groove recognition mode, and the DNA affinities, particularly for  $\text{GO-Co}_3\text{O}_4$  and  $\text{Ag-GO-Co}_3\text{O}_4$ , were found to be sufficiently strong to induce a high level of toxicity. Furthermore, when experimental results are combined with docking interaction analysis,  $\text{Ag-GO-Co}_3\text{O}_4$  may possess the intrinsic capability to elicit higher toxicity in HepG2 cancer cell lines. Nevertheless, it is imperative to reinforce these findings with additional *wet-lab* studies to comprehensively elucidate the mechanisms underlying the safe utilization and management of these synthetic nanomaterials across various applications.

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**Data Availability** No datasets were generated or analysed during the current study.

## Declarations

**Ethical Approval** Not applicable.

**Consent to Participate** The authors declare our consent to participate.

**Consent for publication** The authors declare our consent for publication upon acceptance.

**Competing Interests** The authors declare no competing interests.

**Conflict of interest** The authors declare that they have no conflict of interest. No potential conflict of interest was reported by the authors.

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