

HETEROCYCLES IN FOCUS

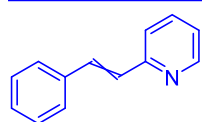
The synthesis of *ortho*-stilbazoles (2-styrylpyridines) (microreview)

Saveliy P. Sorokin^{1*}, Oleg V. Ershov¹

¹ Chuvash State University named after I. Ulyanov,
15 Moskovsky Ave. Cheboksary 428015, Russia; e-mail: ssp_9999@mail.ru

Translated from Khimiya Geterotsiklicheskih Soedinenii,
2022, 58(11), 582–584

Submitted July 27, 2022
Accepted after revision October 24, 2022



This microreview compiles methods for the synthesis of *ortho*-stilbazoles (2-styrylpyridines) described in the literature in 2017–2022. Depending on the synthons from which the target structure is formed, four main synthetic approaches can be distinguished: coupling reactions, Wittig reactions, condensation reactions, and pyridine ring formation reactions.

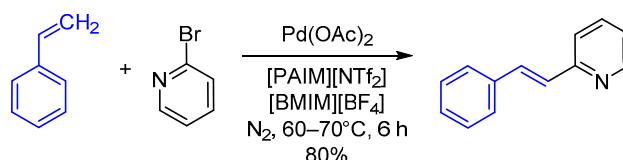
Introduction

Organic molecules based on the framework of *ortho*-stilbazole are widely employed in medicinal chemistry¹ and materials chemistry.² Such heterocycles exhibit high antioxidant,^{1a,b} antitumor,^{1c} and anti-inflammatory activity.^{1d} The possibility of inhibiting the replication of the SARS-CoV-2 coronavirus,^{1e} VEGFR-2 kinase,^{1f} and Mur ligase^{1g} has also been noted. Conjugated molecules based

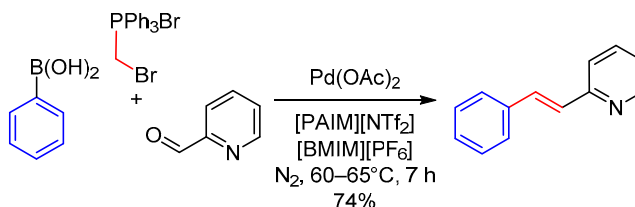
on *ortho*-stilbazole are used as chemosensors for the determination of $\text{Cr}_2\text{O}_7^{2-}$ and MnO_4^- ,^{2a} CN^- ,^{2b} F^- , and AcO^{2c} anions, Hg^{2+} ,^{2d} Zn^{2+} ,^{2e} 2,4,6-trinitrophenol^{2f} cations, as fluorescent probes and labels,^{2g} as candidate materials for photonic devices and optical switches.^{2h}

Coupling reactions

The coupling reactions are one of the principal methods for the synthesis of *ortho*-stilbazoles.^{1g,3} The Heck reaction based on the use of styrene and 2-bromopyridine can be provided as a classical example. A feature of the transformation is the use of ionic liquids.^{3a}



The [PAIM][NTf₂]/[BMIM][X] (X = PF₆, BF₄) system also demonstrated its efficacy in promoting the original tandem Wittig–Suzuki reaction.^{3a} Both transformations involving ionic liquids proceed stereoselectively with the formation of the *E*-isomers.



Saveliy P. Sorokin was born in 1999 in Alikovo, Chuvashia, Russia. He received the BSc degree at the Chuvash State University named after I. Ulyanov in 2020, and the MSc degree in 2022. He currently works as a laboratory assistant in the Department of organic and pharmaceutical chemistry. His research interests include cyano-substituted pyridines and pyridones, donor-acceptor chromophores, design of heterocyclic fluorescent molecules.

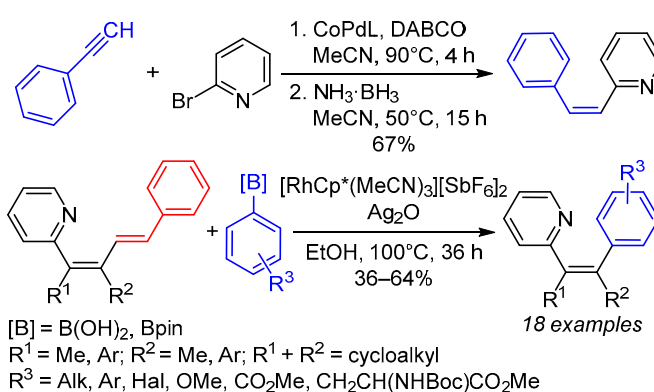


Oleg V. Ershov was born in 1975 in Petrovka, Odessa Oblast, Ukraine. He graduated from the Chuvash State University named after I. Ulyanov in 1997. He received the PhD degree in Chemistry in 2000. Currently, he serves as assistant professor at the Department of organic and pharmaceutical chemistry at the same University. His scientific interests include chemistry of polynitriles, heterocyclic fluorescent and biologically active compounds, donor-acceptor chromophores.

Coupling reactions (continued)

The elusive *Z*-isomer of *ortho*-stilbazole can be obtained by a one-pot synthesis.^{3b} For this, the sequence of the Sonogashira coupling in the presence of a cobalt-palladium catalyst and the semihydrogenation of alkynes with borazane is employed.

The original rhodium(III)-catalyzed Suzuki reaction of 1,3-dienes with arylboronic acids is accompanied by elimination of the styrene fragment and proceeds with excellent chemoselectivity.^{3c} The conversion is carried out in the presence of silver oxide to regenerate the catalyst and return it to the catalytic cycle.

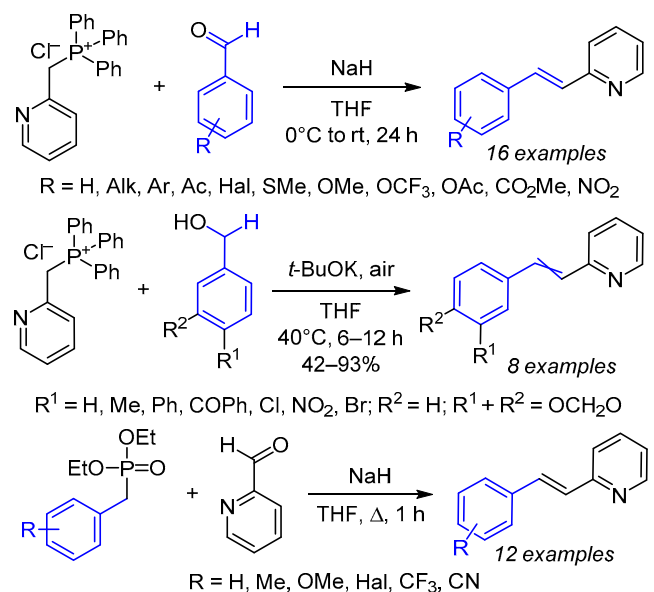


The Wittig reaction

The classical Wittig reaction remains one of the most accessible and versatile tool in the synthesis of styrylpyridines.^{1c,2f,4} One example of the application of the Wittig reaction in the synthesis of styrylpyridines is the reaction of benzaldehyde derivatives with 2-picolylphosphonium chloride in the presence of NaH.

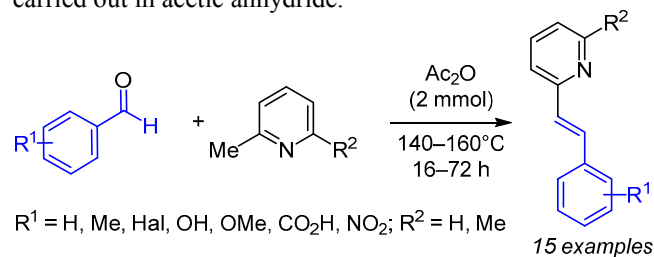
A method of direct Wittig olefination of alcohols was proposed. The reaction was carried out under aerobic conditions using air as an inexpensive and clean oxidizing agent. *ortho*-Stilbazoles were formed as a mixture of stereoisomers with a significant predominance of the *E*-isomer.^{4b}

A modified Wittig reaction, the Horner–Wadsworth–Emmons reaction, is also used for the synthesis of derivatives of *ortho*-stilbazole. In this case, the transformations proceed with the participation of pyridine-2-carbaldehyde and benzylphosphonate in the presence of a base.^{4c}

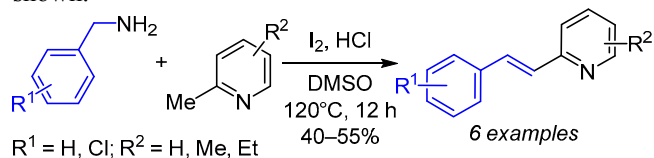


Condensation reactions

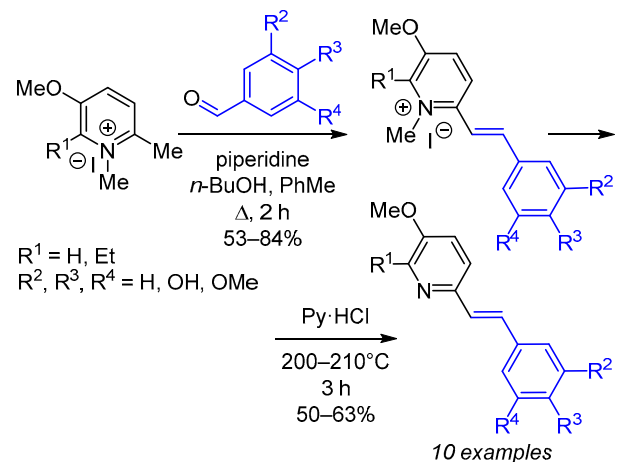
Condensation reactions are one of the simplest approaches to the synthesis of *ortho*-stilbazoles.^{1a,b,5} For example, reactions between *α*-picolines and benzaldehyde derivatives are carried out in acetic anhydride.^{5a}



An innovative synthesis of 2-styrylpyridines from benzylamines and *ortho*-picolines in DMSO medium with the addition of HCl and iodine as an oxidizing agent was shown.^{5b}

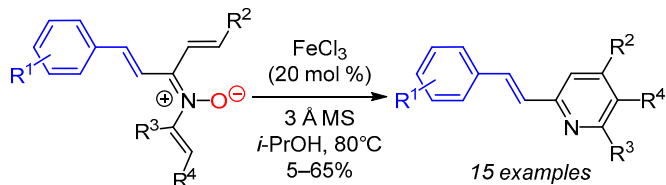


These transformations were also carried out using methylpyridinium salts in *n*-butanol^{1b} or a *n*-butanol-toluene mixture.^{1a} Demethylation of salts of the compounds was carried out by heating to 210°C with anhydrous pyridinium chloride. In this case, the time of the condensation step was significantly reduced.



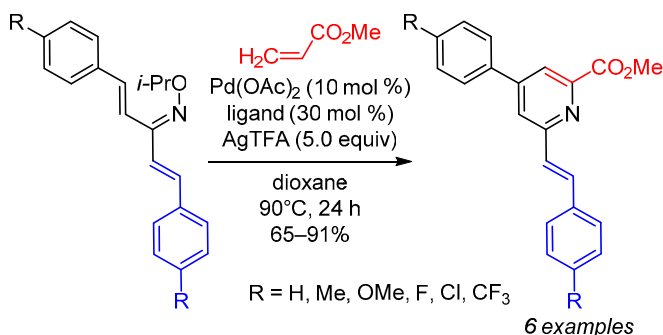
Pyridine ring formation reactions

Heterocyclization⁶ or trans-heterocyclization⁷ reactions are also widely employed for the preparation of *ortho*-stilbazole derivatives. A method of synthesis from *N*-vinyl- α,β -unsaturated nitrones in the presence of an iron catalyst and molecular sieves *via* cleavage of the N⁺-O⁻ bond was presented.^{6a}



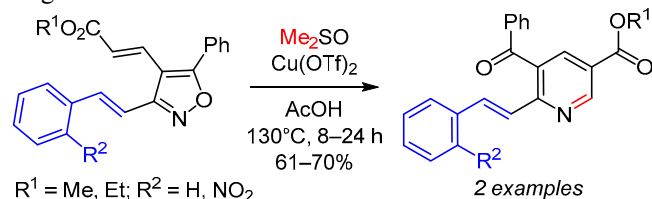
R¹ = H, Me, OMe, Br, CF₃; R², R³, R⁴ = H, Alk, Cy, Ar

It was shown that oximes, close structural analogs of nitrones, can be used instead by reacting them with methyl acrylates.^{6b} In this case, palladium(II) acetate with the ligand (2,6-bis-(adamantan-1-yl)oxypyridine) and a silver salt were used as the catalyst and as an oxidizing agent, respectively.



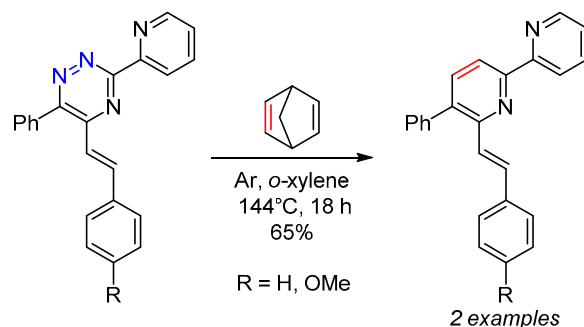
R = H, Me, OMe, F, Cl, CF₃

An example of a trans-heterocyclization reaction is the conversion of isoxazoles.^{7a} The copper catalyst initiates ring opening by cleaving the N–O bond. DMSO serves as a one-carbon synthon generating an active methylene group in the course of the reaction, which leads to the formation of two C–C bonds during the formation of the pyridine ring.



R¹ = Me, Et; R² = H, NO₂

The reaction of 1,2,4-triazine derivatives with norbornadiene proceeds according to the mechanism of the aza-Diels–Alder reaction.^{7b} The transformations result in the elimination of a nitrogen molecule and formation of the pyridine ring at the expense of two carbon atoms of the diene.



R = H, OMe

The study was supported by the Russian Science Foundation grant No. 22-13-00157, <https://rscf.ru/project/22-13-00157>.

References

- (a) Semenov, A. V.; Balakireva, O. I.; Tarasova, I. V.; Semenova, E. V.; Zulfugarov, P. K. *Med. Chem. Res.* **2020**, *29*, 1590. (b) Semenov, A. V.; Balakireva, O. I.; Tarasova, I. V.; Burtasov, A. A.; Semenova, E. V.; Petrov, P. S.; Minaeva, O. V.; Pyataev, N. A. *Med. Chem. Res.* **2018**, *27*, 1298. (c) Pugachev, M. V.; Pavelyev, R. S.; Nguyen, T. N. T.; Gabbasova, R. R.; Bulatov, T. M.; Iksanova, A. G.; Aljondi, B.; Bondar, O. V.; Grishaev, D. Yu.; Yamaleeva, Z. R.; Kataeva, O. N.; Nikishova, T. V.; Balakin, K. V.; Shtyrlin, Y. G. *Bioorg. Med. Chem.* **2021**, *30*, 115957. (d) Chen, G.; Shan, W.; Wu, Y.; Ren, L.; Dong, J.; Ji, Z. *Chem. Pharm. Bull.* **2005**, *53*, 1587. (e) Li, Y.-Q.; Li, Z.-L.; Zhao, W.-J.; Wen, R.-X.; Meng, Q.-W.; Zeng, Y. *Eur. J. Med. Chem.* **2006**, *41*, 1084. (f) Sun, W.; Fang, S.; Yan, H. *Med. Chem. Commun.* **2018**, *9*, 1054. (g) Hrast, M.; Frlan, R.; Knez, D.; Zdovc, I.; Barreateau, H.; Gobec, S. *Bioorg. Med. Chem. Lett.* **2021**, *40*, 127966.
- (a) Zhang, X.-D.; Zhao, Y.; Chen, K.; Jiang, Y.-F.; Sun, W.-Y. *Chem.–Asian J.* **2019**, *14*, 3620. (b) Guan, R.; Chen, H.; Cao, F.; Cao, D.; Deng, Y. *Inorg. Chem. Commun.* **2013**, *38*, 112. (c) Xie, P.; Guo, F.; Zhang, D.; Zhang, L. *Chin. J. Chem.* **2011**, *29*, 1975. (d) Zho, H.; Sun, L.; Chen, W.; Tian, G.; Chen, Y.; Li, Y.; Su, J. *Tetrahedron* **2016**, *72*, 2300. (e) Gabr, M. T.; Pigge, F. C. *Dalton Trans.* **2016**, *45*, 14039. (f) Zhang, X.-D.; Hua, J.-A.; Guo, J.-H.; Zhao, Y.; Sun, W.-Y. *J. Mater. Chem. C* **2018**, *6*, 12623. (g) Wang, M.-Q.; Ren, G.-Y.; Zhao, S.; Lian, G.-C.; Chen, T.-T.; Ci, Y.; Li, H.-Y. *Spectrochim. Acta A: Mol. Biomol. Spectrosc.* **2018**, *199*, 441. (h) Senthil, K.; Kalainathan, S.; Kumar, A. R.; Aravindan, P. G. *RSC Adv.* **2014**, *4*, 56112.
- (a) Savanur, H. M.; Kalkhambkar, R. G.; Laali, K. K. *Appl. Catal. A: Gen.* **2017**, *543*, 150. (b) Clauss, R.; Baweja, S.; Gelman, D.; Hey-Hawkins, E. *Dalton Trans.* **2022**, *51*, 1344. (c) Tan, G.; Das, M.; Maisuls, I.; Strassert, C. A.; Glorius, F. *Angew. Chem., Int. Ed.* **2021**, *60*, 15650.
- (a) Tian, J.-J.; Yang, Z.-Y.; Liang, X.-S.; Liu, N.; Hu, C.-Y.; Tu, X.-S.; Li, X.; Wang, X.-C. *Angew. Chem., Int. Ed.* **2020**, *59*, 18452. (b) Li, Q.-Q.; Shah, Z.; Qu, J.-P.; Kang, Y.-B. *J. Org. Chem.* **2018**, *83*, 296. (c) Cao, C.; Zeng, Z.; Cao, C. *J. Phys. Org. Chem.* **2022**, *35*(4), e4319.
- (a) Nguyen, T. B.; Nguyen, T. M.; Retailleau, P. *Chem.–Eur. J.* **2020**, *26*, 4682. (b) Sharma, R.; Abdullaha, M.; Bharate, S. B. *J. Org. Chem.* **2017**, *82*, 9786.
- (a) Chen, C.-H.; Wu, Q.-Y.; Wei, C.; Liang, C.; Su, G.-F.; Mo, D.-L. *Green Chem.* **2018**, *20*, 2722. (b) Yamada, T.; Hashimoto, Y.; Tanaka, K., III; Morita, N.; Tamura, O. *Org. Lett.* **2021**, *23*, 1659.
- (a) Kumar, P.; Kapur, M. *Org. Lett.* **2020**, *22*, 5855. (b) Khasanov, A. F.; Kopchuk, D. S.; Nikonov, I. L.; Taniya, O. S.; Kovalev, I. S.; Zyryanov, G. V.; Rusinov, V. L.; Chupakhin, O. N. *Russ. Chem. Bull.* **2021**, *70*, 999.