



# Moisture as key for understanding the fluorescence of lignocellulose in wood

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**Abstract** The fluorescence behaviour of lignocellulose in *Pinus sylvestris* L. was studied under the influence of moisture. Fluorescence excitation-emission-matrices (EEMs) of the solid wood surfaces were recorded. Two emission peaks were identified, one attributed to lignocellulose, the other to pinosylvins. The two peaks were successfully modelled with PARAFAC2-deconvolution. Lignocellulose showed excitation-dependent emission. Its emission was quenched and blue-shifted by moisture, while pinosylvins showed none of these properties. The quenching efficiency was proportional to the moisture content (linear Stern–Volmer plot), a phenomenon first demonstrated for wood in this study. Potential mechanisms for the moisture quenching are discussed, with clustering-triggered emission best explaining most of the observed peculiarities. The strong influence of moisture on the fluorescence of pine wood suggests that carbohydrates, or interactions between carbohydrates and lignin, play an important role in lignocellulose fluorescence.

**Keywords** Moisture · Lignocellulose · Fluorescence quenching · FRET · EEM · Clustering-Triggered emission

## Abbreviations

SD	Standard deviation
EEM	Excitation emission matrix
FRET	Förster resonance energy transfer
Ex	Excitation wavelength
Em	Emission wavelength

## Introduction

### Relevance of lignocellulose fluorescence

Given the increasing importance of biorefinery and the urgent need for sustainable resource utilization, it is highly relevant to understand the complexity of natural materials, in particular with regards to quality and process control. Fast, cheap and non-destructive analysis methods are needed to fully exploit the potential of the resources for the diverse applications (Galkin et al. 2017).

Several approaches have been made to use fluorescence spectroscopy for species identification on solid wood (Camorani et al. 2008; Ma et al. 2023; Moya et al. 2013; Piuri and Scotti 2010) and wood extracts (Pandey et al. 1998). Fluorescence imaging has been used for heartwood detection (Antikainen et al. 2012), photodegradation monitoring (Peters and Rapp 2021) and various identification and classification applications in microscopy (Babi 2017; Chimenez et al. 2014; Donaldson and Vaidya 2017; Maceda and Terrazas 2022; Mishra et al. 2018; Selig

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et al. 2012). Fluorescence lifetime imaging has been proposed for the classification of waste wood (Leiter et al. 2022) and for the analysis of lignin in compression wood (Donaldson and Radotic 2013) and sugar bagasse (Coletta et al. 2013).

In contrast to the above qualitative or semi-quantitative approaches, exact quantitative fluorescence-based methods on solid lignocellulosics are rare. Belt et al. (2021) proposed UV-excited fluorescence spectroscopy for quantitative measurements of stilbene content in pine wood. The correlation between fluorescence intensity and stilbene content was rather weak ( $r=0.79$ ). Werner and Pecina (1995) proposed fluorescence spectroscopy as a useful method for the quantitative monitoring of mass loss by fungal attack in beech wood, although their results showed high statistical variation.

It must be considered that fluorescence spectroscopy of solid surfaces is generally subject to higher errors than in liquids. These errors include internal filter effects, light absorbing extractives, quenching effects, surface roughness, and others. In general, it can be said that fluorescence spectroscopy of solids is less common, and some influencing factors are not yet fully understood. A better knowledge of the mechanisms of wood fluorescence will help to reduce these errors and increase the applicability of fluorescence methods in solid wood.

#### Attribution of lignocellulose fluorescence peaks to chemical substances

A fundamental question is the source of general wood fluorescence. In this context, the general fluorescence of lignocellulose has to be clearly separated from the strong fluorescence reported for specific wood species (Avella et al. 1988), originating from individual fluorophore extractives. Surprisingly, there is no consensus in the research community about the chemical compounds nor the mechanisms contributing to the general fluorescence spectra of lignocellulose. A number of studies have explained the general fluorescence of the wooden cell wall with the distribution and composition of lignin (Auxenfans et al. 2017; Chabbert et al. 2018; Djikanović et al. 2007; Donaldson 2013; Donaldson et al. 2015; Donaldson and Knox 2012; Ji et al. 2013). Also, a recent review of the autofluorescence in plants (Donaldson 2020) reports lignin to be a major contributor to wood

fluorescence, with highly excitation-dependent emission. Donaldson (2013) measured blue emission in pine and poplar wood upon UV (355 nm) excitation, green emission upon blue (488 nm) excitation, and red emission upon excitation with green (561 nm) light. This excitation-dependent emission behaviour is uncommon for most fluorophores (Kasha 1950).

On the other hand, cellulose and other cell wall carbohydrates have been shown to emit in a similar range to lignin, also with excitation-dependent emission (Ding et al. 2020). Fluorescence emission of cellulose nanofibres has been reported in the red and infrared range upon green (532 nm) excitation, as well as in the green range upon blue (450 nm) excitation (Khalid et al. 2019), and blue emission upon UV (320 nm) excitation (Castellan et al. 2007; Olmstead and Gray 1997). Also, lignin and cellulose have both been reported to show pH-dependent emission (Donaldson (2013) and Ding et al. (2020), respectively), with lowest emission at pH 7.

Since cellulose does not have delocalized pi-electrons or other conventional fluorescence systems, impurities of different kinds have been assumed to be the source of cellulose fluorescence (Atalla and Nagel 1972; Lloyd and Miller 1979), but it seems very unlikely that the same impurities exist in all kinds of cellulose of different origins and preparation processes. Note that Ding et al. (2020) and Olmstead and Gray (1997) used, among others, bacterial cellulose where lignin cannot at all occur, not even in trace amounts. Despite this striking evidence, the intrinsic fluorescence of cellulose and hemicelluloses often remains unconsidered, when all fluorescence is attributed to lignin only (Auxenfans et al. 2017; Chabbert et al. 2018; Chimenez et al. 2014; Donaldson and Radotic 2013; Hoque et al. 2023).

These contradictions in the above mentioned studies suggest that the true origin of lignocellulose fluorescence is still not fully understood. The apparent parallels between suggested lignin and cellulose fluorescence raise the question of their respective contribution to the fluorescence of native wood.

Unfortunately, lignin is always more or less modified when extracted from wood, and does not exist in pure form without cellulose and hemicelluloses. To examine the fluorescence of native lignin in wood, indirect methods have to be used. The approach of this study was to use moisture as a reversible and non-destructive way of modulating wood fluorescence,

since water is a known fluorescence quencher (Fürstenberg 2017; Maillard et al. 2020) and binds to hydroxyl groups in wood. These groups are especially present in hemicelluloses and cellulose, while lignin is much more hydrophobic (Christensen and Kelsey 1959; Fredriksson et al. 2023; Hou et al. 2022). The aim of this study was therefore to investigate the influence of moisture on the fluorescence of native wood. The results shall contribute to a deeper understanding of the origins of wood fluorescence, and to a better applicability of fluorescence based methods on wood.

### Material and methods

Seven samples (axial length  $\times$  tangential width  $\times$  radial depth =  $15 \times 15 \times 30$  mm) of commercial Scots pine (*Pinus sylvestris* L.) heartwood were produced with growth ring orientation along the measured surface. The samples were made in axial order from the same annual ring of a wooden strip. The tangential surface was cut down with a razorblade to exclusively examine latewood. Samples were marked one to seven in axial order and stored in darkness at 20 °C at relative humidities of 0, 33, 53, 75, 94, and  $2 \times 100$  %, respectively. After seven days, two replicate fluorescence excitation emission matrices (EEMs) were recorded from different spots of each sample. In a desorptive approach, this process was repeated for every moisture stage and sample. Photodegradation and other aging processes as cause of variation were ruled out by a final examination of all samples in the first moisture stage. This final examination did not differ significantly from the initial one. Altogether, 112 EEMs were recorded (2 replicates  $\times$  7 samples  $\times$  7 moisture stages + 14 final measurements).

All EEMs were taken in front-face geometry with a resolution of 5 nm with a Hitachi F-7100 fluorescence spectrometer. An in-house modified beam path setup was used, where the excitation beam impinged on the sample surface at an angle of 34° to the emission lens. The illuminated surface was app. 78 mm<sup>2</sup>. By this modification, the ratio of fluorescence signal to higher-order scattering was increased about sixfold, compared to unmodified setup, due to the reduced reflection into the emission lens. After the last measurement, the samples were again stored for 50 days in the respective humidity stages. Then the

equilibrium wood moisture was determined according to formula 1.

Formula 1: Wood moisture:

$$u = (m_1 - m_0) / m_0 * 100\%$$

where  $u$  is the wood moisture,  $m_1$  is the mass of the moist specimen, and  $m_0$  is the mass of the dried specimen.  $m_0$  was obtained by drying the samples at 60 °C until equilibrium, followed by drying at 103 °C for two hours. This method was reported by Rapp and Sudhoff (2020) to reduce the evaporation of resin in Scots pine and other species with volatile extractives.

The contributions of the individual peaks to the total spectrum were calculated with PARAFAC2 deconvolution in Matlab, using the PLS toolbox (version 9.0) from Eigenvector Research, Inc., Manson, USA. PARAFAC2 was used because of its ability to handle small deviations from trilinearity (Harshman 1972), e.g. retention time shifts in chromatography (Bro et al. 1999), while presenting chemically meaningful information, i.e. pure component spectra (Kiers et al. 1999). To the best of our knowledge, PARAFAC2 was not used before to deconvolute EEMs with compounds that do not follow Kasha's rule (Kasha 1950), i.e. have shifting emission spectra depending on the excitation wavelength. This was the case for the lignocellulose in pine wood, as observed in this study.

A two-component PARAFAC2-model was calculated with constraints set to nonnegativity in concentration and excitation modes. The data was preprocessed in the following steps: Using the EEM filtering tool in the software, Rayleigh scattering signals were removed in first and second orders in a half-width of 35 nm, replacing the missing values by interpolation. All emission signals below the respective excitation wavelength (sub-Rayleigh wavelengths) were set to zero. For the model calculations, the EEMs were clipped to a range of excitation = 250–500 nm and emission = 340–600 nm.

Before interpretation, PARAFAC 2 scores (the calculated "concentration" values) were divided by the peak intensity of kraft paper, stored at 20 °C and 65 % relative humidity, and recorded at the same day after the respective sample. This was done as reference to reduce the influence of long term drifts in lamp intensity and other machine parameters on the spectra. Kraft paper is relatively homogeneous and

**Fig. 1 a** Typical EEM of pine wood, stored at 0% relative humidity. One peak (blue arrow) showed excitation-dependent emission. The second peak (red arrow) did not show excitation-dependent emission. **(b)**: Emission spectra of **a**) at excitation wavelengths 395 (blue line) and 325 (red line) nm. **(c)** and **(d)**: PARAFAC2 loading surfaces of component 1 and 2, respectively. They represent the deconvolution of the two peaks in **a**)

shows fluorescence in a wide emission band, comparable to wood.

## Results

Identification and characterisation of the fluorescence peaks

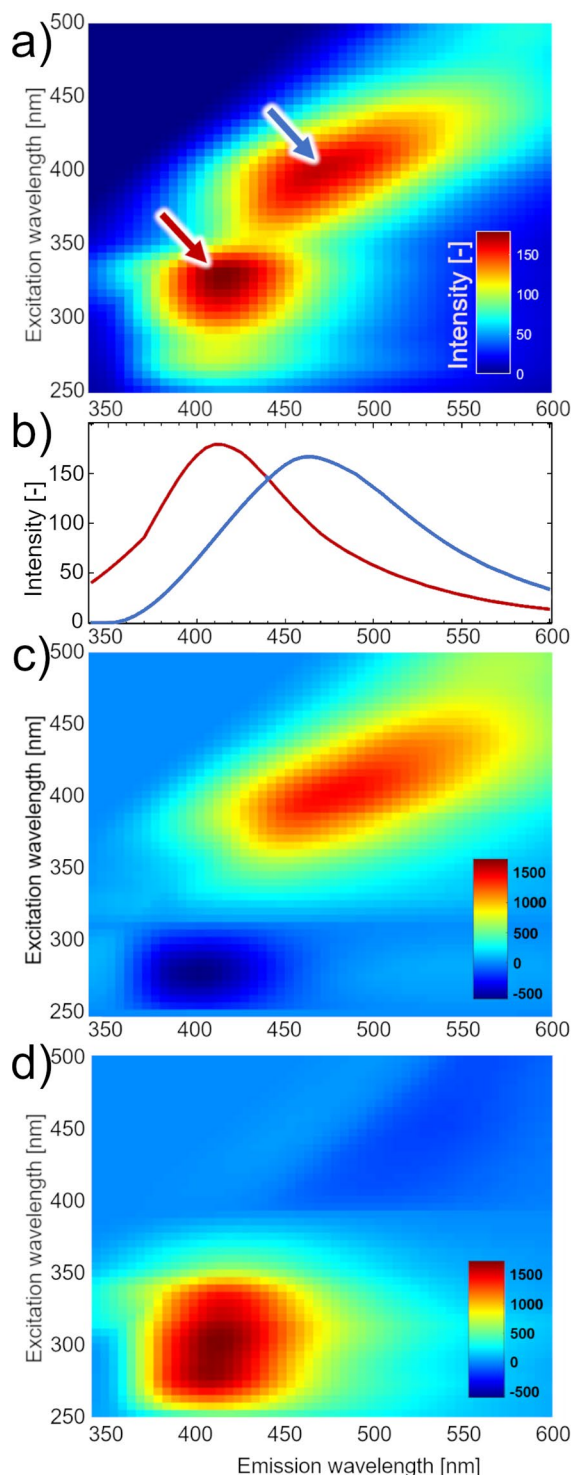
The EEMs of pine wood all showed two distinct emission peaks, as exemplified in Fig. 1a. One had a maximum at excitation ( $Ex$ ) =  $385 \pm 15$  nm and emission ( $Em$ ) =  $458 \pm 13$  nm (blue arrow/line in Fig. 1a/b) and showed a strong dependence of emission wavelength on excitation wavelength, which can be seen by its diagonal shape in the EEM (Fig. 1a). Note that this excitation-dependent fluorescence emission is unusual because, according to Kasha's rule (Kasha 1950), the emission wavelength of a fluorophore does not depend on excitation wavelength. The second fluorescence peak had a maximum at  $Ex = 323 \pm 8$  nm and  $Em = 413 \pm 3$  nm (red arrow/line in Fig. 1a/b).

The 2-component PARAFAC2-deconvolution showed promising model details (split-half-analysis: 99.6%; 100% Core consistency; 99.766% explained variance) and successfully deconvoluted the two peaks (Fig. 1c and d).

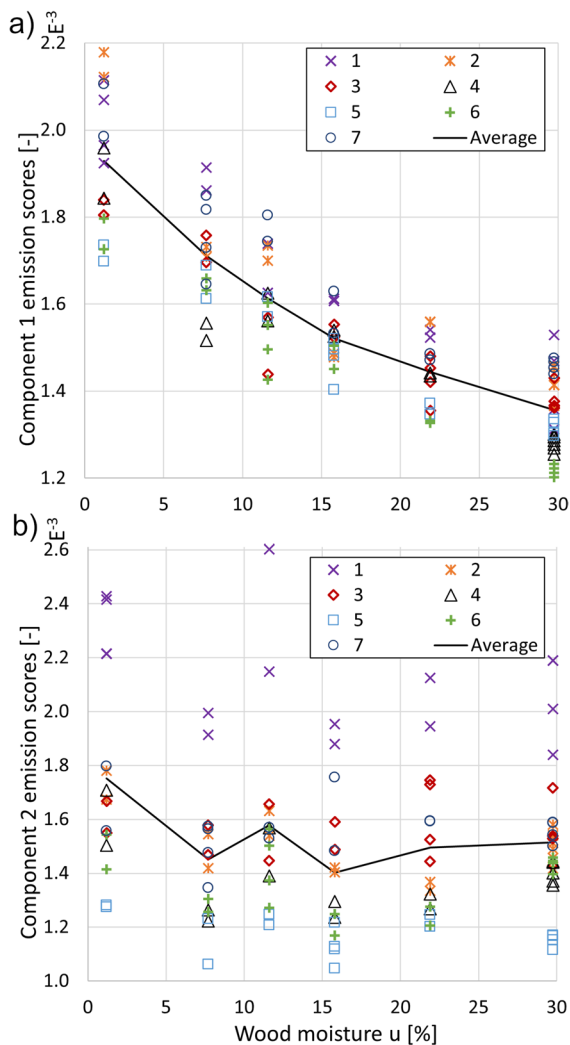
Changes in fluorescence spectra

Component 1 was considerably quenched by moisture, as displayed in Fig. 2a. Component 2 did not show any significant dependence on water content, but a higher between-sample variance (Fig. 2b). Sample 1, for example, consistently showed higher scores here, while sample 5 had lower scores than average. Additionally, component 2 also showed a high random variance upon repeated measurements.

The quenching efficiency ( $F_0/F$ ) of each sample was calculated by dividing the intensity in dry condition ( $F_0$ ) by the intensity of the same sample in the respective moisture stage ( $F$ ). Figure 3 a and



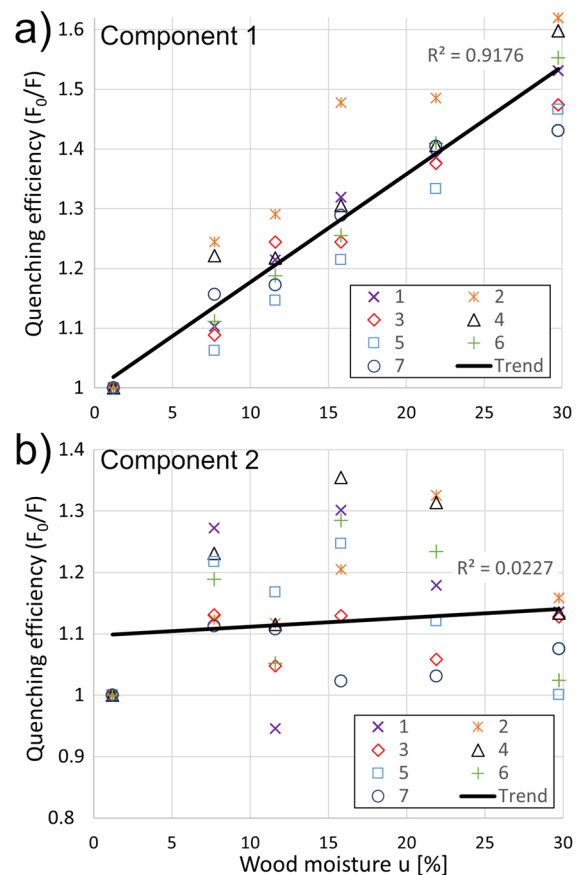
**b** show the quenching efficiency of the seven samples as a function of moisture content, also known as Stern–Volmer plot. Although some samples show



**Fig. 2** Fluorescence intensity scores of component 1 (a) and component 2 (b) in samples 1–7, plotted against wood moisture. Samples were numbered in longitudinal order. All values relative to the respective kraft paper reference peak intensity

high variation, the overall trend is clearly pronounced and linear for component 1, while there is no clear trend for component 2.

The quenching effect by water was stronger at longer wavelengths. Figure 4a–c show the EEMs of the pine samples at 1.2% ( $F_0$ ) and 29.8% ( $F$ ) moisture content, and the quenching efficiency ( $F_0/F$ ) EEM, respectively. The quenching efficiency (Fig. 4c) shows a clear maximum of 1.80 at  $Ex = 545$  nm and  $Em = 635$  nm, indicating that these fluorescence wavelengths were stronger quenched

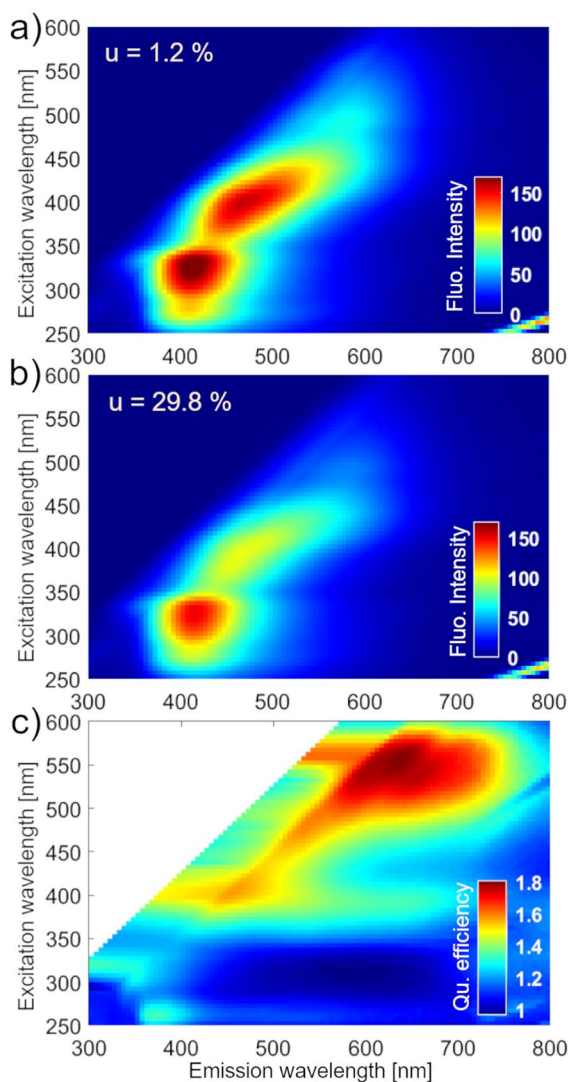


**Fig. 3** Quenching efficiency ( $F_0/F$ ) of components 1 (a) and 2 (b) as a function of moisture content, also known as Stern–Volmer plot. All values are mean values of the respective sample at the respective humidity stage

by water. At  $Ex = 400$ ;  $Em = 455$ , quenching efficiency had another maximum of 1.55, while the lower wavelengths ( $Ex < 370$  nm) were only marginally quenched.

The mean excitation and emission maxima were both blue-shifted with increasing moisture content (Fig. 5a and b, respectively). Although the effect of moisture on peak excitation and emission wavelengths was not strong ( $\rho = -0.41$  and  $-0.29$ , respectively) and subject to statistical variation, the effect was significant according to Spearman's correlation analysis ( $p < 0.05$  for Fig. 5a–c). Excitation and emission maxima of the lignocellulose peak were strictly correlated ( $\rho = 0.93$ , Fig. 5c). This was present across all samples and moisture stages and represents an interesting feature.

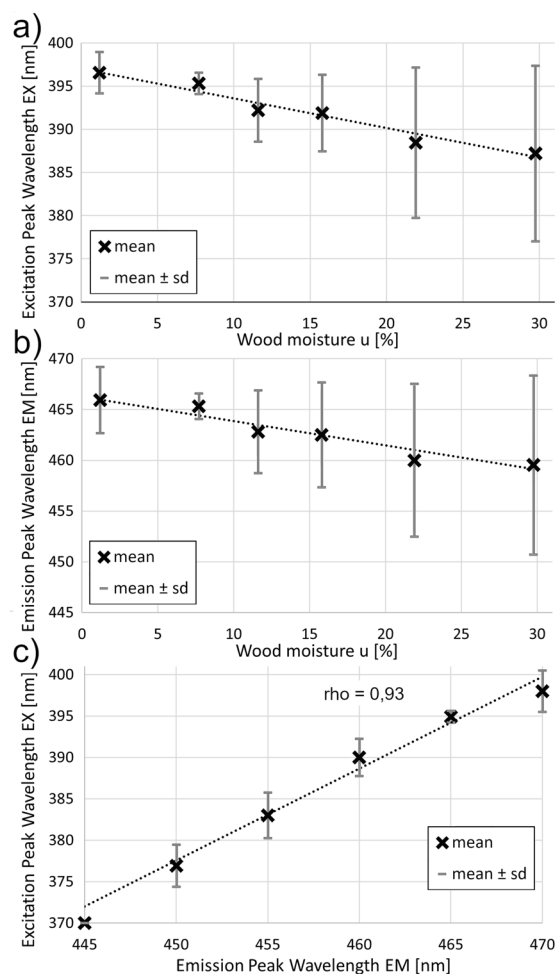




**Fig. 4** Mean EEMs of samples at (a) 1.2 % and (b) 29.8 % moisture content, and (c) mean quenching efficiency EEM, calculated by dividing a) by b). High values in (c) express a strong quenching effect by water (the white area represents missing values due to division by zero)

## Discussion

The peak at  $Ex=385$  and  $Em=458$  (component 1) was assumed to originate from the lignocellulose part of the wood, since wood fluorescence in this wavelength range was reported before in steam exploded lignocellulose (Auxenfans et al. 2017) and a variety of softwood and hardwood species like *Pinus radiata*, *Populus deltoides* (Donaldson 2013), and *Eucalyptus bosistoana* (Mishra et al. 2018). The peak at



**Fig. 5** Mean excitation (a) and emission (b) maxima of  $n=112$  EEMs, displayed against moisture content; (c) correlation between Ex and Em peak wavelengths. sd = standard deviation

$Ex=323$  nm/ $Em=413$  nm (component 2) was attributed to pinosylvin and its monoethyl ether, two heartwood extractives of *Pinus sylvestris* L., as reported by Antikainen et al. (2012) and Belt et al. (2021).

Both the high between-sample variance and the high variance upon repeated measurement of the pinosylvin peak (Fig. 2b) were explained by an uneven distribution of pinosylvins in the wood. As reported by Belt et al. (2017), pinosylvin is present in extractive deposits in certain cell wall areas and lumina in Scots pine. Repeated placement in the spectrometer in a slightly different sample position caused the pinosylvin-rich areas to be in or out of the range of the excitation beam. This led to noise error in the intensity of pinosylvin fluorescence.

The diagonal shape of the lignocellulose peak (Fig. 1c), also called excitation-dependent emission, is an interesting feature, and a violation of Kasha's rule (Kasha 1950). It could be explained, for example, by the fact that a large number of different fluorophores are present in lignocellulose, with individual, overlapping fluorescence peaks. The diagonal peak could then be regarded as a sum of multiple overlapping peaks with small deviations in their excitation and emission energies. This is quite plausible in view of the chemical multiformity of lignin and hemicelluloses, but could also be explained by e.g. crystallite size or chain length of fluorescing carbohydrates. The dependence of cellulose fluorescence on crystallinity and chain length have been reported by Jiang et al. (2021) and Ding et al. (2020), respectively.

Theoretically, the shift could also be explained by the inner-filter effect and reabsorption, which is reported to cause a wavelength shift in high fluorophore concentrations, even in front-face geometry (Bevilacqua et al. 2020). However, the inner-filter effect seems to be irrelevant in this study, because the pinosylvic peak ( $E_x=323$  nm,  $E_m=413$ ) did not show excitation-dependent emission. Because wood absorption is high in this wavelength range (Fengel and Wegener 2003), the inner-filter effect should be at least as strong for pinosylvic as for lignocellulose. However, it seemed to be not present for both.

The moisture quenching of lignocellulose (component 1 in Fig. 2a) indicates a strong interaction between the fluorescent substances and water.

Possible quenching mechanisms for water in solid wood are:

1. Quenching by energy transfer (energy loss) to water molecules
2. Suppression of Förster resonance energy transfer (FRET) between fluorophores through swelling
3. Suppression of clustering-triggered emission through swelling

#### Quenching by energy transfer (energy loss) to water molecules

In fluids, water is known to quench fluorescence of many organic dyes (Fürstenberg 2017; Maillard et al. 2020) by collisional quenching. In this mechanism, quenching efficiency is typically linearly correlated to

quencher concentration, according to the Stern–Volmer–Equation (Stern and Volmer 1919). The linear Stern–Volmer plot of moisture quenching in wood (Fig. 3a) found in this study might support the hypothesis of collisional quenching by water.

Although collisional quenching is well-explained in fluids, some questions remain concerning the transferability to solids like wood. Collisional quenching requires the quenching molecule (liquid water) to collide with the fluorophore, leading to energy transfer to the water. However, below fibre saturation, water is more and more firmly bound to the cell wall through H-bonding (Thybring et al. 2022), and the energetic states of both water and lignocellulose change (Guo et al. 2016). At low moisture levels, other mechanisms than energy transfer to water might prevail.

Maillard et al. (2020) explained water quenching in 42 organic fluorophores by energy transfer to vibrational overtones of water, whereby the energy becomes unavailable for fluorescence. They argue that only the first layer of water molecules around the fluorophore are likely to contribute to the quenching effect. If water binds to wood in multiple layers at high moisture, as proposed by widespread sorption isotherm models (Hailwood and Horrobin 1946; Dent 1977; Skaar 1988), the quenching efficiency would be expected to flatten out when the monolayer capacity is reached. This was clearly not the case in this study (Fig. 3a).

In the recent years, the physical interpretations of these multilayer models have been questioned (Thybring et al. 2021). The observed linearity of the Stern–Volmer plot in this study indicates that within the hygroscopic range, each additional portion of water has the same potential of quenching lignocellulose fluorescence. This seems to support Willems (2014) who rejects the multilayer theories and proposes a sorption site occupancy (SSO) model instead (Willems 2015). According to the SSO model, all water molecules in the hygroscopic range directly bind to a sorption site in the cell wall matrix.

Maillard et al. (2020) found quenching mainly above 600 nm, and attributed it to the known overtones of water. This fits to the findings shown in Fig. 4c (dark red area in the upper right corner). The lower peak in Fig. 4c at  $E_x=400$  and  $E_m=455$  cannot be explained by Maillard's hypothesis. Concludingly, energy transfer to vibrational states of water molecules can only explain part of the quenching observed in this study.

### Suppression of energy transfer (FRET) between fluorophores through swelling

An alternative explanation of the moisture quenching of lignocellulose is the suppression of Förster Resonance Energy Transfer (FRET) between fluorophores through moisture swelling of the wood cell walls (increased distance between donor–acceptor pairs). FRET describes the transfer of excitation energy between a donor (short-wavelength absorber) and an acceptor (long-wavelength emitter) molecule. This mechanism is heavily influenced by the distance between the two molecules (Lakowicz 2006). Swelling of the wood cell wall as an effect of moisture uptake would result in longer distances and hence suppress this energy transfer, resulting in quenching of the long-wavelength acceptor fluorescence. If FRET takes place, the short-wavelength emission of the donor molecules is reduced, and the long-wavelength emission of the acceptor molecule appears instead. In other words: if FRET is suppressed, the total excitation and emission wavelengths of the mixture will be shorter. The blue-shift of excitation and emission peaks upon moisture uptake observed in this study (Fig. 5a–b) supports this hypothesis.

FRET donor and acceptor groups are likely to be present in wood. Terryn et al. (2020) used native wheat straw as FRET donor, in combination with rhodamine acceptor molecules. Native lignin contains a number of supposedly fluorescing groups (Donaldson 2020), and native cellulose is a mixture of diverse crystalline and amorphous areas of varying chain length. Both crystallinity (Du et al. 2019; Gong et al. 2013; Jiang et al. 2021; Kalita et al. 2015) and chain length (Ding et al. 2020) have been reported to influence the fluorescence emission of cellulose and other carbohydrates. Therefore, it seems likely that FRET takes place between various donor and acceptor units in wood at various wavelengths. The fact that the fluorescence emission maximum of lignocellulose shifts with excitation wavelength supports the presence of various fluorescent states, since pure compounds typically show excitation-independent emission spectra, according to Kasha's rule (Kasha 1950), which is a basic principle in photochemistry.

One feature of lignocellulose fluorescence remains unexplained by the FRET hypothesis: Ideally, the excitation peak of a fluorophore should be

represented in its absorption spectrum, because the absorption of a photon is a requirement for fluorescence (Lakowicz 2006). However, according to literature (Sirviö et al. 2016; Zhang et al. 2020a; Ruwoldt et al. 2022), none of the wood cell wall polymers shows significant absorption peaks in the wavelength range around 385 nm, the excitation maximum observed in this study.

### Suppression of clustering-triggered emission through swelling

Clustering-triggered and aggregation-induced fluorescence emission is a young research field, recently reviewed by Tomalia et al. (2019). In the recent years, lignin and cellulose fluorescence have repeatedly been reported to depend on clustering of certain functional groups in the solid state. Gong et al. (2013) found that cellulose and starch emit only in the solid state. Du et al. (2019) attributed the emission of cellulose and its derivatives to the clustering of ether, hydroxyl and carbonyl groups. Based on model calculations, Jiang et al. (2021) even found electron delocalisation in the crystalline states of celluloses I and II. Interestingly, the delocalisation of excitation energy in form of exciton waves in crystals was already reported by Zander (1981). Concerning lignin, Xue et al. (2016) stated that the suppression of intramolecular rotation through clustering of carbonyl groups leads to the solid state fluorescence emission of lignin.

In a recent review, Zhang et al. (2020b) identified the following six characteristic features of clustering-triggered luminogens:

1. their luminescence is based on non-conjugated molecular motifs, where *n*- or *pi*- electron-based groups are separated by nonconjugated linkers
2. the intrinsic luminescence of molecule clusters appears at longer wavelengths and independently of the *n*- or *pi*-electron-based luminescence in isolated states,
3. the excitation spectra appear red-shifted in comparison with their absorption spectra,
4. the emission profiles are excitation-dependent,
5. the emission profiles are size-dependent, with a higher molecular weight leading to a red-shift and an increase in emission, and



6. clustering-triggered luminogens are capable of phosphorescence.

Although Zhang et al. (2020b) admit that not all named characteristics are necessary requirements for clustering-triggered emission, lignocellulose shows all six properties, as discussed below:

1. Cellulose and other carbohydrates are known to fluoresce (Ding et al. 2020), although they do not contain conjugated molecular motifs. Carbohydrates are rich in hydroxyl (and partly carbonyl) groups that contain non-binding  $n$ -electrons. Lignin consists of aromatic structures that are separated by non-conjugated linkers (Mai and Zhang 2023).
2. The fluorescence of isolated aromatic structures in lignin occurs at lower wavelengths than the lignocellulose fluorescence observed in this study. Coniferyl alcohol fluorescence, for example, was reported at  $E_x=310$  nm and  $E_m=416$  nm (Achyuthan et al. 2010), contrary to the lignocellulose peak at  $E_x=385$  nm and  $E_m=458$  nm in this study (Fig. 1).
3. Lignin absorption peaks at app. 280 nm (Ruwoldt et al. 2022), while cellulose absorption lies below 200 nm (Sirviö et al. 2016). Both is much lower than the measured excitation maximum of 385 nm in this study.
4. The excitation-dependent emission profile of lignocellulose is shown in Fig. 1a.
5. Through water uptake, hydrogen-bonds between cell wall composites in wood become weaker and some eventually break (Thybring et al. 2022), thus reducing the effective cluster size and clustering-triggered emission intensity. Although Zhang et al. (2020b) limited the term “size” (5) to generations of dendrimers, molecular weight of polymers, diameter of nanoparticles etc., we propose that the weakening of hydrogen bonds within the clusters has a comparable effect to size reduction. In lignocellulose, this leads to a blue-shift (Fig. 5c) and a decrease (Fig. 3a) in emission, as observed in the present study.
6. Phosphorescence was not examined in this study, but reported for cellulose by Grönroos et al. (2018).

Concludingly, all six characteristic features of clustering-triggered emission are present in

lignocellulose. Since clustering-triggered emission explains most of the characteristic (and uncommon) features of lignocellulose fluorescence, it seems reasonable to assume that its fluorescence originates from cluster-formation.

## Conclusions

The exact mechanism for moisture quenching in wood may not be fully proved yet, but this study has shown for the first time that moisture quenching exists for wood, and that it follows a linear Stern–Volmer relationship. Contrary to widespread assumptions, carbohydrates do exhibit their own, independent fluorescence (Olmstead and Gray 1997). Since they have much more sorption sites for water than lignin (Fredriksson et al. 2023; Hou et al. 2022), it seems likely that the general fluorescence and the moisture quenching originate to a significant part from carbohydrates or an interaction between carbohydrates and lignin.

Clustering-triggered emission seems to best explain most of the quenching and wavelength shifting by water observed in this study.

PARAFAC2 proved to be a useful tool for modeling fluorescence peaks with excitation-dependent emission, i.e. lignocellulose fluorescence. It might be applied for spectral unmixing in further studies, for example to quantify fluorescent extractives in solid wood EEMs.

**Author contributions** All authors contributed to the study conception and design. Material preparation, data collection and analysis were performed by Frank B. Peters. The first draft of the manuscript was written by Frank B. Peters and all authors commented on previous versions of the manuscript. All authors read and approved the final manuscript.

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**Data availability** The data used in the presented manuscript are available upon individual request from the authors.

## Declarations

**Ethical approval** Not applicable.

**Competing interests** The authors have no relevant financial or non-financial interests to disclose.

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