

# Microwave-assisted TiO<sub>2</sub>: anatase formation on cotton and viscose fabric surfaces

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**Abstract** The method of TiO<sub>2</sub>-anatase film preparation on cotton and viscose fabric surfaces using the sol-gel process and microwave treatment is presented. Microwave treatment was used to change the amorphous TiO<sub>2</sub> form to anatase directly on the fabrics. The influence of microwave treatment conditions on the obtainable polymorphic form of TiO<sub>2</sub> was examined. Fabrics were pretreated with low-temperature air plasma (30 min). The root mean square height in the selected area increased from 44 to 166 nm (cotton) and from 9 to 112 nm (viscose). Infrared analysis showed the new band at 1748 and 1732 cm<sup>-1</sup> corresponding to C=O stretching for plasma-treated cotton and viscose textiles, respectively. The plasma pretreatment also improved the wetting properties by TiO<sub>2</sub> sol and increased the surface free energy of fabrics. TiO<sub>2</sub> film thickness was 180 nm (12 %wg. Ti)

and 140 nm (3 %wg. Ti) for cotton and viscose, respectively. TiO<sub>2</sub>-modified cotton reduced the nicotine concentration three times more and TiO<sub>2</sub>-modified viscose was two times higher under sunlight compared to raw fabrics. No changes in strength were observed for TiO<sub>2</sub>-modified cotton, while the strength of TiO<sub>2</sub>-modified viscose decreased about 45 %. No effect of UV irradiation on cotton and a slight reduction of the strength of raw viscose (7 %) and TiO<sub>2</sub>-modified viscose (16 %) were observed. The Ti contents after washing decreased from 12 to 11 % (cotton) and from 3 to 2.6 % (viscose). The presented method allows obtaining TiO<sub>2</sub> film-anatase on the cotton and viscose fabrics, but its total effectiveness is better for cotton fabrics.

**Keywords** Microwave · TiO<sub>2</sub> · Sol-gel · Plasma · Cotton · Viscose

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## Introduction

Titanium dioxide (TiO<sub>2</sub>), especially in the anatase phase applied to the modification of textile materials, provides new properties such as photocatalytic, self-cleaning, bioactive, UV protective, superhydrophilic, antibacterial, flame retardant, etc. (Radetic 2013; Karimi et al. 2014; Morawski et al. 2013; Qi et al. 2006; Ibrahim et al. 2010; Karimi et al. 2010, Abidi et al. 2009). There are many methods to modify

materials with TiO<sub>2</sub>; some of them require post-heat treatment to induce crystallization, e.g., the sol-gel method. At low temperature, it can be distinguished hydrothermally (Zhang et al. 2012) or using the sonochemical method (Behzadnia et al. 2014a, b). Both methods allow obtaining TiO<sub>2</sub> film in crystalline form. In the hydrothermal method, a long aging period is necessary to obtain the crystalline TiO<sub>2</sub> form. The sonochemical method was used to produce multifunctional wool fabric using titanium isopropoxide or butoxide in acidic media. Another method is the application of a TiO<sub>2</sub> dispersion or Ag-doped TiO<sub>2</sub> in the anatase form to the modification of the fibers and wovens surface (Cieslak et al. 2009). The most popular, fast, effective and easy method of producing TiO<sub>2</sub> thin films is a sol-gel method, but one of the disadvantages is the annealing post-treatment at high temperature (near 500 °C) to transform amorphous TiO<sub>2</sub> to the anatase form (Cieslak et al. 2015; Tung and Daoud 2011; Kadziola et al. 2014). Such treatment limits the ability to create films on textile substrates with low thermal resistance, such as cotton, viscose, wool, polypropylene, polyamide, polyester fibers, etc. (Liu et al. 2007). The microwave treatment is one of the methods in which TiO<sub>2</sub> in anatase form can be enhanced at lower temperature (below 200 °C). The application of microwave treatment may be useful for the modification of low-thermal-resistance textile materials. Some approaches using microwave treatment to obtain crystalline TiO<sub>2</sub> have been reported (Liu et al. 2007; Wilson et al. 2002; Vigil et al. 2001). Liuxue et al. (2007) obtained TiO<sub>2</sub> anatase on cotton fibers prepared via a microwave-assisted liquid-phase process with hexafluorotitanate as precursor. Microwaving improved the hydrolysis of the precursor and led to highly crystalline anatase at low temperature. Similar studies were carried out by Peiro et al. (2001), but the TiO<sub>2</sub> anatase film was deposited on glass substrates and silicon wafers. Li et al. (2010) prepared silver-modified TiO<sub>2</sub> via a microwave-assisted method. All papers presented the process of TiO<sub>2</sub> preparation and/or deposition on film during the microwave treatment. In this study, we show another method in which fibers with TiO<sub>2</sub> film (prepared using the sol-gel technique earlier) were only treated with microwaves to transform TiO<sub>2</sub> to the anatase form.

## Materials

To prepare TiO<sub>2</sub> sol, titanium (IV) isopropoxide Ti(OC<sub>3</sub>H<sub>7</sub>)<sub>4</sub> (TTIP, 97 %, Aldrich) as a Ti precursor, isopropanol (99.5 %, POCH S.A.) and hydrochloric acid (Chempur) were used. TiO<sub>2</sub> sol was applied to plain-woven cellulose fabrics whose characteristics are given in Table 1.

TiO<sub>2</sub> film was also applied on silicon (Si) wafer 100 (Cemat Silicon S.A.) with a thickness of 320–380 μm as a reference substrate material.

## Methods

### TiO<sub>2</sub> thin film preparation

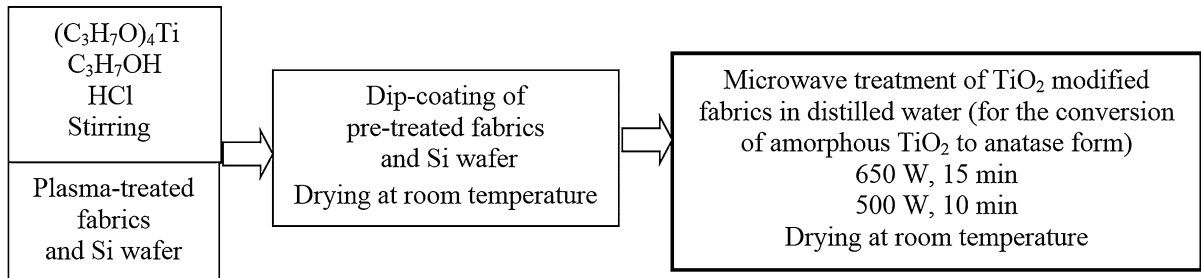
Titanium dioxide was prepared based on a sol-gel technique using Ti—titanium tetraisopropoxide as precursor. To the solution containing 1.2 g (0.00422 mol) TTIP in 13.2 g (0.22 mol) of isopropanol, 0.04 g (0.0011 mol) of hydrochloric acid (HCl 2 mol/l) was added dropwise. The prepared sol was stirred 30 min before deposition on the fabric samples and Si wafers. The surfaces of cotton and viscose fabrics were pretreated in low-temperature air plasma (45 W, 15.26 MHz, Zepto) for 30 min. TiO<sub>2</sub> sol was applied on the pretreated substrates by dip-coating technique using an NIMA dip coater (Technology LB, UK) with a constant rate of 25 mm/min. After the sol deposition, fabrics and Si wafers were dried at room temperature for 48 h. Next, the samples were placed in vessels filled with distilled water and treated with microwaves (MWs) using an Anton Paar Synthos microwave reactor. After the MW treatment, the samples were dried at room temperature. Different MW treatment conditions were tested (various powers and heating times), and the results of two selected conditions are presented in this article (MW power and heating time 650 W, 15 min at  $T = 180$  °C and 500 W, 10 min at  $T = 110$  °C) (Fig. 1). MW treatment was only used for conversion of the TiO<sub>2</sub> amorphous form deposited on the fabric surface to TiO<sub>2</sub> anatase.

### Characterization

A thermogravimetric analysis (TGA) of viscose and cotton fabrics was carried out using a TG 209F1 Libra analyzer (Netzsch, Germany) with a heating rate of

**Table 1** Characteristics of the studied fabrics

Fabric composition (%)	Weave	Mass per unit area (g/m <sup>2</sup> )	Pick and end densities (1/cm)	
			Warp	Weft
Cotton 100	Plain	108	43	32
Viscose 100	Plain	104	45	25

**Fig. 1** Flowchart of thin TiO<sub>2</sub> layer preparation on fabrics and Si wafers

10 °C min<sup>-1</sup> under a nitrogen flow rate of 20 ml min<sup>-1</sup> over the range of 30–700 °C. The samples (4.0 ± 2 mg) were tested in a ceramic crucible. The initial ( $T_{\text{Onset}}$ ) and final ( $T_{\text{End}}$ ) temperature of degradation, temperature at peak maximum degradation ( $T_{\text{Max Peak}}$ ) and loss of mass were evaluated. Fabrics were pretreated in low-temperature air plasma (50 W, Zepto) for 30 min. The surface morphologies of fabrics and Si wafers were studied with a scanning electron microscope (SEM) (Nova NanoSEM 450, FEI, USA). The thickness of the TiO<sub>2</sub> coating on the fiber surfaces was evaluated on the basis of the SEM images of fiber cross sections. The elemental composition analysis and maps of Ti distribution were performed with an X-ray microanalyzer (INCA Energy EDS, Oxford Instruments Analytical, UK) connected to a Vega 3 SEM (Tescan, Czech Republic). The 3D SEM images were prepared using Alicona MeX software connected to a Vega 3 SEM. FTIR analysis of textiles before and after plasma treatment using the Thermo Scientific Nicolet iS50 FT-IR equipped with a Golden Gate ATR accessory with 4-cm<sup>-1</sup> resolution was made. Raman spectra and mapping analyses of modified fabrics were performed using a Raman Renishaw InVia dispersive spectrometer (Renishaw, UK) with infrared  $\lambda = 785$ -nm laser excitation. The phase composition of the examined samples was measured using the X-ray diffraction technique. In the investigation, an Empyrean diffractometer (PANalytical) working in grazing incidence

(GIXRD) mode with Co K $\alpha$  (1.78901 Å) radiation was used. The incidence angle ( $\omega$ ) was set to 2°. Further data processing was done using the ICDD PDF 4 database and HighScore Plus software. To evaluate the impact of plasma treatment on the surface free energy (SFE) of fabrics and the wettability of fabrics by TiO<sub>2</sub> sol, the contact angle was measured using the Drop Shape Analyzer (Krüss GmbH DSA 100, Germany) and three non-reactive standard liquids (droplet size of 5  $\mu$ l): distilled water ( $\gamma_l = 72.8$ ;  $\gamma_l^d = 21.8$ ;  $\gamma_l^p = 51.0$ ), diiodomethane ( $\gamma_l = 50.8$ ;  $\gamma_l^d = 48.5$ ;  $\gamma_l^p = 2.3$ ) and formamide ( $\gamma_l = 58.0$ ;  $\gamma_l^d = 39.0$ ;  $\gamma_l^p = 19.0$ ). On the basis of the obtained contact angle values, the SFE was calculated using the Owens–Wendt model (Owens and Wendt 1969; Cieslak et al. 2012). In this model, the SFE is calculated from the geometric mean of dispersion and polar interactions between the solid and liquid, according to Eq. (1):

$$\gamma_l(1 + \cos\theta) = 2[(\gamma_l^d \gamma_s^d)^{1/2} + (\gamma_l^p \gamma_s^p)^{1/2}] \text{ and } \gamma_s = \gamma_s^d + \gamma_s^p \quad (1)$$

where  $\theta$  is the contact angle, and  $\gamma_s^d$ ,  $\gamma_s^p$ ,  $\gamma_l^d$  and  $\gamma_l^p$  are the dispersion (<sup>d</sup>) and polar (<sup>p</sup>) components of the surface free energies of the solid (<sub>s</sub>) and liquid (<sub>l</sub>), respectively. The reversible work of adhesion ( $W_a$ ) of the fabric and water was calculated using Dupre's Eq. (2) (Dupre 1869):

$$W_a = \gamma_s + \gamma_l - \gamma_{sl} \quad (2)$$

where  $\gamma_s$  is the surface free energy of the solid,  $\gamma_{sl}$  is the solid/liquid interface tension, and  $\gamma_l$  is the surface free energy of the liquid.

The photocatalytic properties were tested by the decomposition of nicotine adsorbed on cotton and viscose fabrics modified with TiO<sub>2</sub>-anatase. The testing was performed at  $21 \pm 2$  °C on the test stand as previously described (Cieslak et al. 2009; Cieslak et al. 2014). Briefly, ten samples of each fabric type were hung in a chamber (40 dm<sup>3</sup>) and the chamber air was evacuated; 5 µl of nicotine (Fluka 72290, CAS 54-11-5, puriss. p.a. >99.0 %) was injected into a U-shaped tube placed in the bath warmed up to 250 °C, and nicotine vapor was supplied in a stream of air into the chamber. The chamber was protected from light with a black cover. After 24 h, two samples of each fabric type were removed from the chamber and subjected to chromatographic analysis to assess the initial concentration of nicotine. The rest of the samples were placed on a frame attached to the window and exposed to sunlight. Two sensors (CMP 3 and CUV 4, Kipp and Zonen B.V.) were used to measure the light intensity, which was in the 310–2800 nm and 305–385 nm range, respectively. Periodically, two samples of each type were taken to determine the nicotine concentration. The total irradiation dose was 405.8 kJ/m<sup>2</sup>.

Tensile strength tests before and after modifications of the fabrics and UV irradiation were performed according to standard PN-EN ISO 13934-1:2013-07 using an Instron 3367 machine. The fabrics were UV irradiated (305–385 nm) with the use of a Xenotest Alpha HE apparatus (Atlas) with a total irradiation dose ( $P$ ) of 5184 kJ/m<sup>2</sup>. Five repetitions were carried out for each fabric type, and the average values of the breaking force and breaking elongation were determined.

Washing of the TiO<sub>2</sub>-modified samples was performed according to standard PN EN ISO 6330:2012, procedure 4 N, method A. The Ti contents of samples before and after washing were evaluated on the basis of the SEM/EDS analysis.

## Results and discussion

On the basis of the thermal analysis (Fig. 2), it was noticed that the thermal decomposition of cotton and viscose started at 320.9 °C ( $T_{\text{Onset}}$ ) and 293.5 °C ( $T_{\text{Onset}}$ ) with the maximum ( $T_{\text{Max Peak}}$ ) at 366.0 and

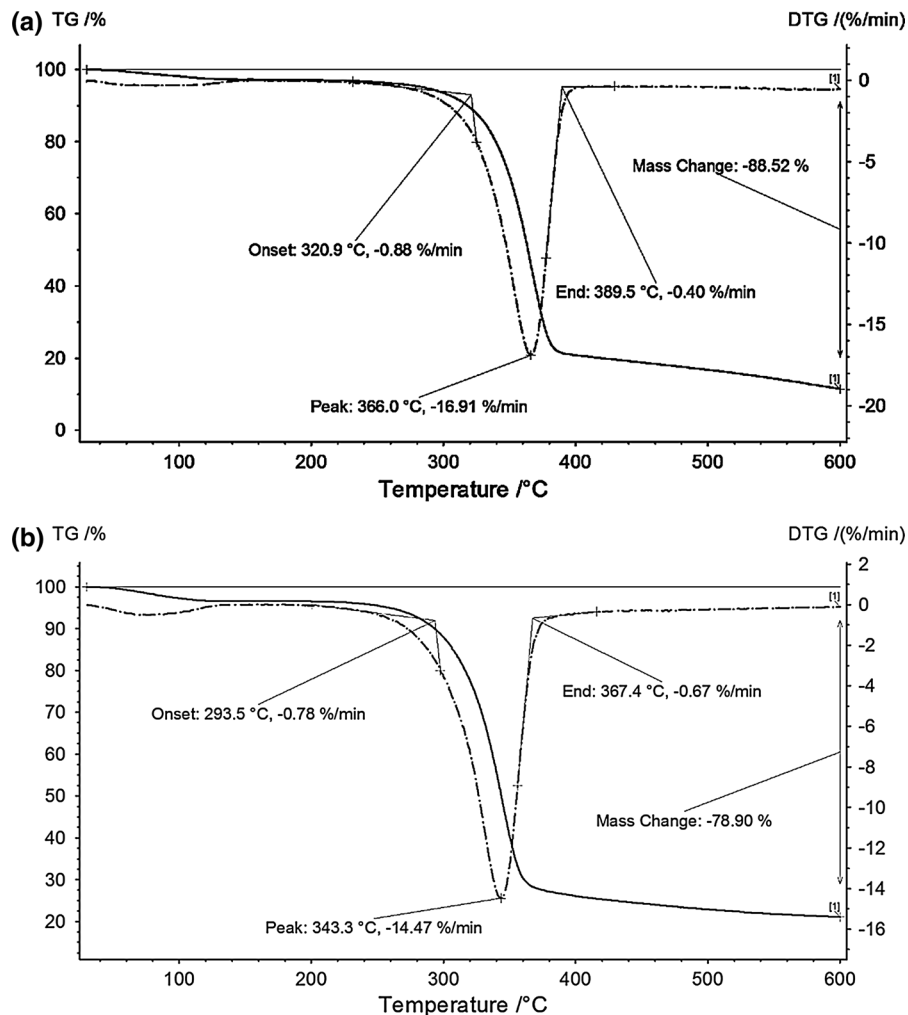
343.3 °C, respectively. Cotton fabric is thermally stable up to about 320 °C and viscose fabric up to 293 °C; above these temperatures, degradation of fabrics was observed. Therefore, the experiments were carried out at lower temperatures of 110 and 180 °C.

The nano-roughness of plasma-treated fabric surfaces was higher than that of the untreated fabrics (Figs. 3, 4, 5). An etching effect of plasma treatment and uniform surface structure of both fabrics without any cracks or damages were observed. The 3D SEM pictures (Fig. 5) clearly showed the changes in the surface structure of fibers after plasma treatment. A significant increase in their roughness was observed. The root-mean-square height of the measured area of 64 µm<sup>2</sup> increased from 44 to 166 nm and from 9 to 112 nm for plasma-treated cotton and viscose, respectively (Fig. 5). Such nano-topography changes on the fabric surfaces are also described in the literature (Navaneetha Pandiyaraj and Selvarajan 2008; Shishoo 2007; Mihailovic et al. 2011). The surface development can provide a new pathway for the TiO<sub>2</sub> sol and improve its application. According to Mirjalili et al. (2015), corona treatment of cotton improves the adsorption of TiO<sub>2</sub> nanoparticles and thus ensures stable properties of materials, e.g., self-cleaning. In addition, TiO<sub>2</sub> film on cotton with corona pretreatment is stable even after 50 washing cycles. In the research of Mihailovic et al. (2011), the amount of TiO<sub>2</sub> for air RF plasma-pretreated cotton increased about 31 % compared to untreated fabric. Corona discharge/air RF plasma resulted in an increase in the binding efficiency of TiO<sub>2</sub> nanoparticles.

The effect of plasma treatment of fabric in the studies differs for each type of fiber. In the case of cotton fibers (Fig. 3), these changes are characteristic of the morphology of the surface layer, including the primary topography (irregularities on the cotton fiber). The higher roughness of cotton allows increasing the surface area and the amount of TiO<sub>2</sub> deposition. The viscose fiber texture is very even, also in the longitudinal recesses, but more amorphous viscose causes susceptibility to plasma etching (Fig. 4) (Przyaszny et al. 2013).

The FTIR spectra of untreated and plasma-treated fabrics are demonstrated in Fig. 6. All spectra show characteristic broad bands around 3339, 3340 and 3392 cm<sup>-1</sup>, which correspond to alcoholic OH stretching, and bands around 2900 cm<sup>-1</sup> and in the region 1455, 1430 and 1429 cm<sup>-1</sup> due to CH<sub>2</sub>

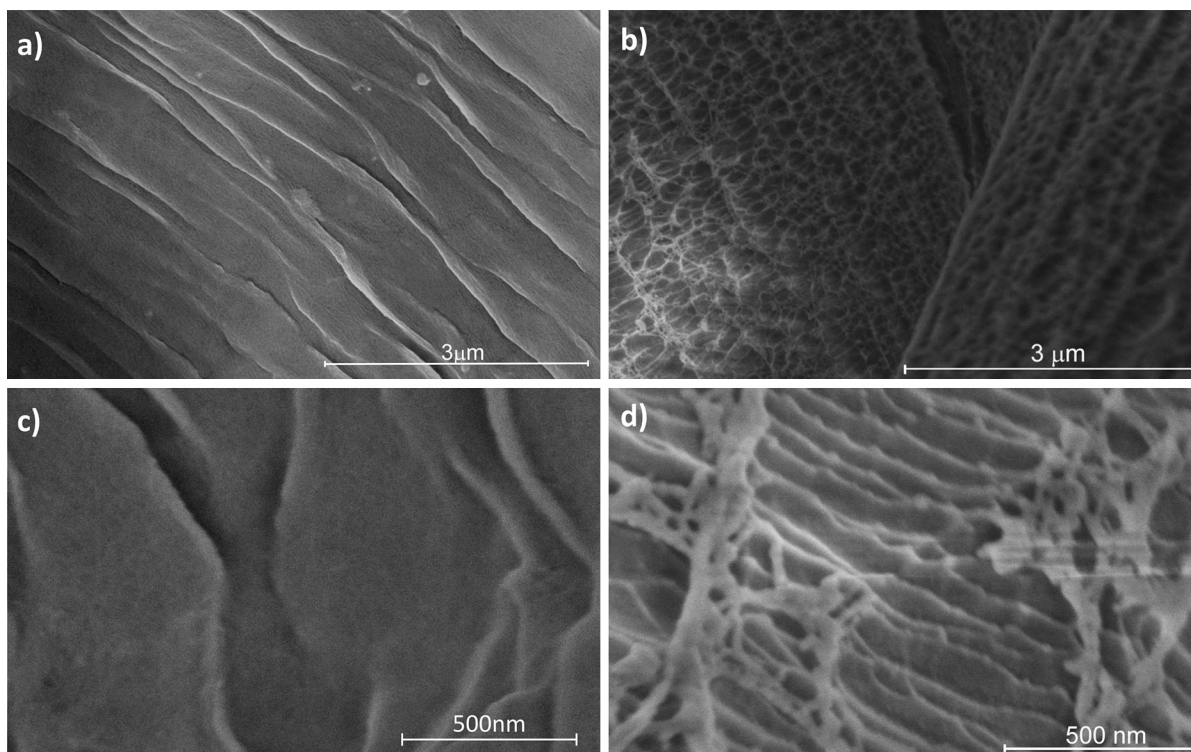
**Fig. 2** TGA analysis of **a** cotton and **b** viscose fabrics



stretching and  $\text{CH}_2$  bending, respectively. Carboxyl groups were noticed in the viscose and cotton plasma-treated samples with bands at  $1732$  and  $1748\text{ cm}^{-1}$  (Fig. 6b,d) corresponding to  $\text{C}=\text{O}$  stretching. This proves that besides etching (confirmed by SEM analysis), surface oxidation of both fibers by plasma occurs. Also a higher intensity of bands was observed in the  $1429\text{--}1061\text{ cm}^{-1}$  range. Bands near  $1455\text{--}1061\text{ cm}^{-1}$  correspond to  $\text{C}-\text{O}$  stretching and in the range  $1200\text{--}1400\text{ cm}^{-1}$  are  $\text{O}-\text{H}$  bending bands (Caschera et al. 2014; Ibrahim et al. 2012). Karahan and Özdoan (2008) did not obtain bands around  $1740\text{ cm}^{-1}$  by atmospheric plasma treatment; however, they obtained this band by Ar plasma treatment. The FTIR results clearly show an increase in the intensity band at  $3340\text{ cm}^{-1}$ , especially for cotton,

which indicates the greater number of OH groups after 30-min plasma treatment. The more polar OH groups and the formation of new polar groups such as  $\text{C}=\text{O}$  (both of cotton and viscose) after plasma etching cause an increase in the hydrophilicity and the value of the surface free energy and adhesion properties. Similar observations were made by Kiwi and Pulgarin (2010). The increase in polar groups and roughness of the cotton surface during RF plasma treatment improved the effectiveness of the modification of the cotton surface with the nanomodifiers  $\text{TiO}_2$  and Ag. After the plasma treatment of fabrics, a decrease in the contact angle was observed for all standard liquids. For water, the value of the contact angle decreased from  $55^\circ$  to  $10^\circ$  and from  $50^\circ$  to  $7^\circ$  for cotton and viscose fabric, respectively (Table 2). The values of SFE for cotton





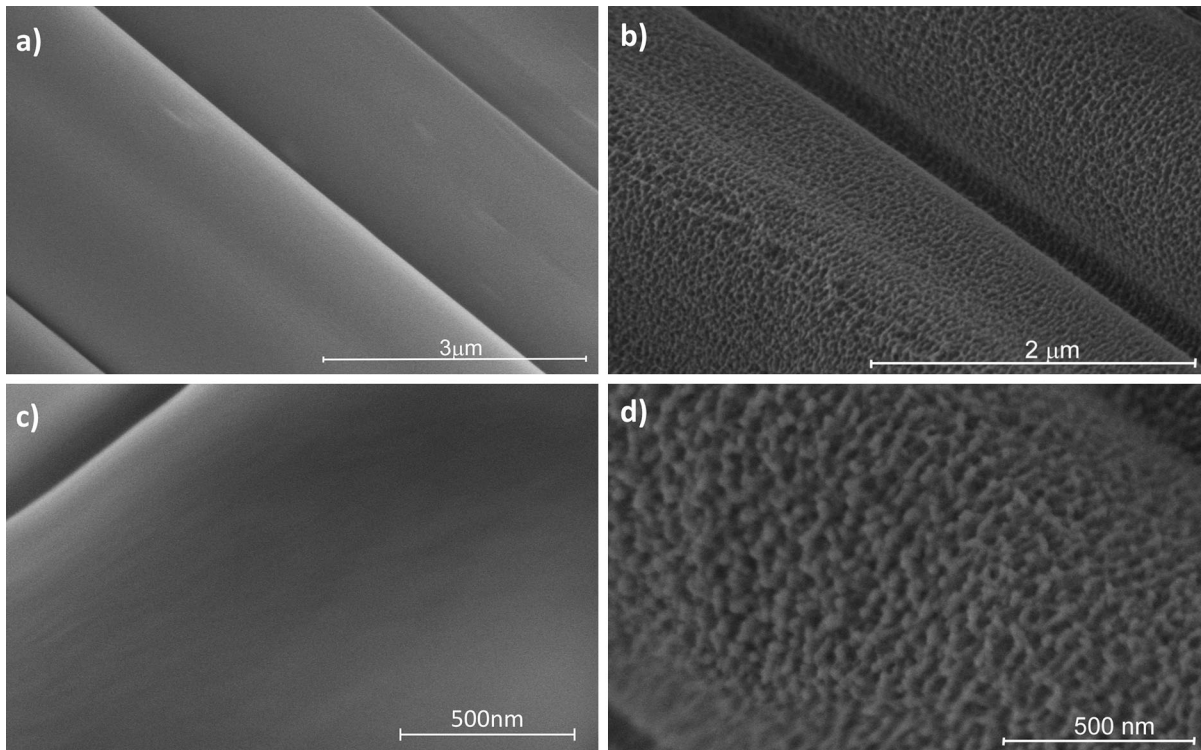
**Fig. 3** SEM images of raw cotton (**a**, **c**  $\times 30,000$  and  $\times 120,000$  magnifications) and 30-min plasma-treated cotton (**b**, **d**  $\times 30,000$  and  $\times 120,000$  magnifications)

and viscose increased by about 26 and 23  $\text{mJ/m}^2$ , respectively (Table 2). A larger increase in the polar component was observed for cotton (twofold) than for viscose (30 %). For the dispersion component, it rose by 25 and 35 % for cotton and viscose, respectively. This was mainly due to the polar groups, such as C=O, formed on the fabric surfaces. This resulted in an increase of the polar components of SFE and consequently improved the wettability. The  $W_{\text{adh}}$  for water increased to 145  $\text{mJ/m}^2$  for both fabrics. The adhesion properties of fabrics were higher, and for both fabrics the coating process with  $\text{TiO}_2$  sol should be more effective (Lam et al. 2011; Calvimontes et al. 2011; Caschera et al. 2014; Navaneetha Pandiyaraj and Selvarajan 2008). This is confirmed by the results of a wettability study with  $\text{TiO}_2$  sol (Fig. 7).

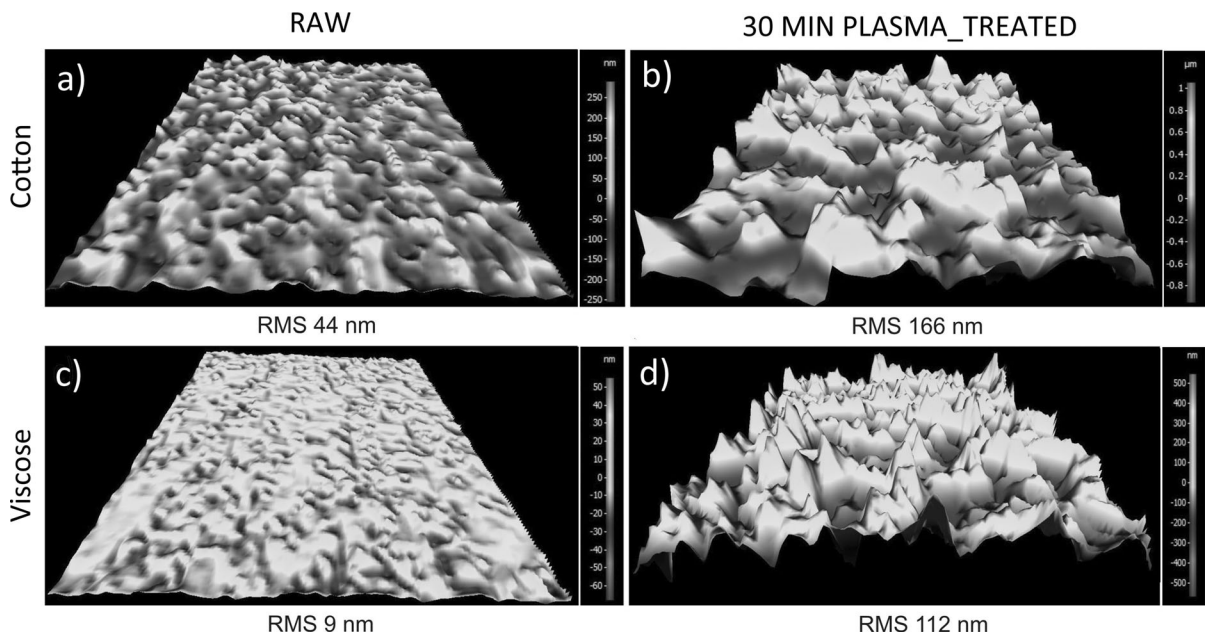
Generally, both fabrics are very well wettable by  $\text{TiO}_2$  sol, but there are some differences in the wettability process. For cotton, the drop of sol spreads on both samples wherein it penetrates immediately to the structure of the plasma-pretreated fabric. In the case of viscose, a flat drop of sol is formed, and the

contact angles before and after plasma treatment are  $23^\circ$  and  $18^\circ$ , respectively (Fig. 7). Under the same conditions of the sol application, the content of  $\text{TiO}_2$  is higher for cotton, which is confirmed by the results of the SEM/EDS (Figs. 14, 15).

The impact of MW treatment is presented on the basis of SEM images of cotton fabric (Fig. 8). No changes or damage of the cotton fiber surface morphology was observed. The same results were observed for the viscose fabric (not shown). The effect of microwave treatment on the fibers was evaluated by Raman analysis (Fig. 9). In the case of cotton (Fig. 9a), the characteristic bands are stretching COC at  $1098\text{ cm}^{-1}$ , a group of four bands ( $353, 411, 457$  and  $580\text{ cm}^{-1}$ ), bending COC and CCC, methylene bands ( $900, 1270, 1378, 1468\text{ cm}^{-1}$ ) and C–H stretching at  $2898\text{ cm}^{-1}$ . No changes after MW treatment were observed between raw cotton (spectrum I) and MW-treated cotton (spectrum II). Characteristic viscose bands included stretching COC at  $1100\text{ cm}^{-1}$ , bending COC and CCC at ( $332, 381, 438, 454$ ), methylene bands ( $920, 1297, 1341, 1384, 1482\text{ cm}^{-1}$ ) and C–H

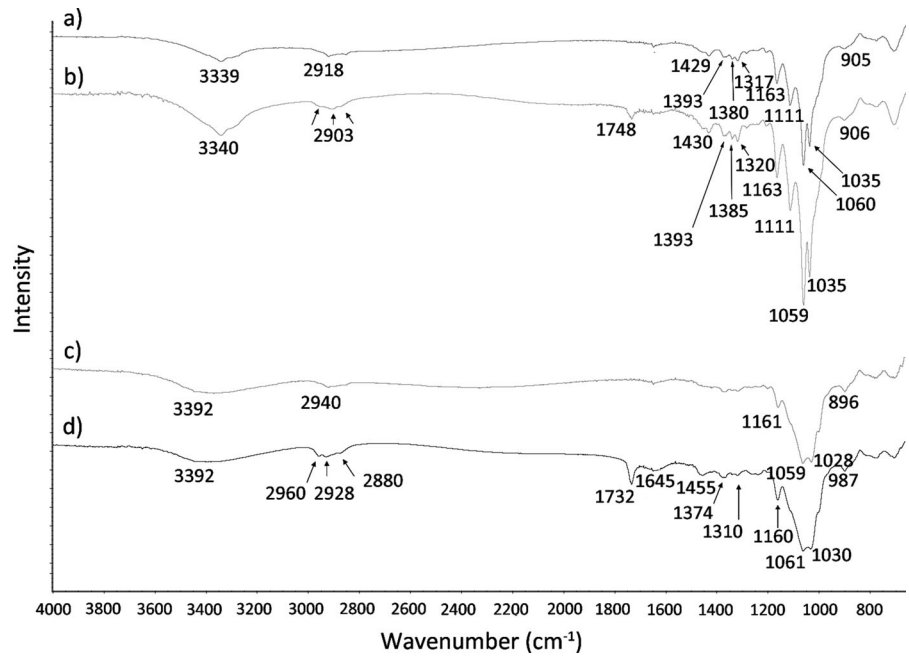


**Fig. 4** SEM images of raw viscose (a, c  $\times 30,000$  and  $\times 120,000$  magnifications) and 30-min plasma-treated viscose (b, d  $\times 30,000$  and  $\times 120,000$  magnifications)



**Fig. 5** The 3D SEM pictures of a raw cotton, b 30-min plasma-treated cotton, c raw viscose and d 30-min plasma-treated viscose with the root-mean-square-height value in the selected area

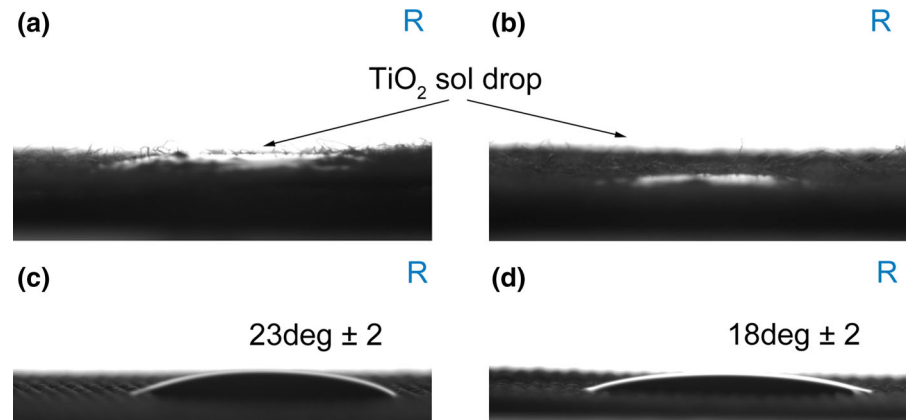
**Fig. 6** FTIR analysis of **a** raw cotton, **b** 30-min plasma-treated cotton, **c** raw viscose and **d** 30-min plasma-treated viscose



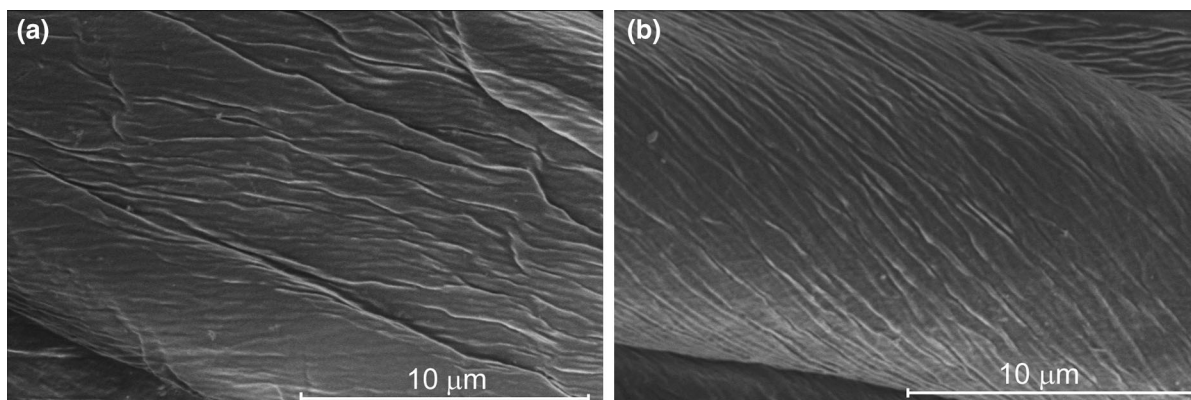
**Table 2** Contact angle, surface free energy and work of adhesion of cotton and viscose fabrics

Standard liquid	Cotton fabric		Viscose fabric	
	Raw	Plasma-treated	Raw	Plasma-treated
Contact angle (°)				
Water	55 ± 5	10 ± 2	50 ± 3	7 ± 2
Diiodomethane	40 ± 3	4 ± 2	56 ± 5	18 ± 4
Formamide	60 ± 7	8 ± 3	52 ± 4	16 ± 2
Surface free energy (SFE) (mJ/m <sup>2</sup> )				
SFE	47 ± 3	73 ± 2	49 ± 4	72 ± 2
γ <sup>d</sup>	31	41	26	40
γ <sup>p</sup>	16	32	23	32
Work of adhesion (Wa) (mJ/m <sup>2</sup> )				
Wa	115	145	120	145

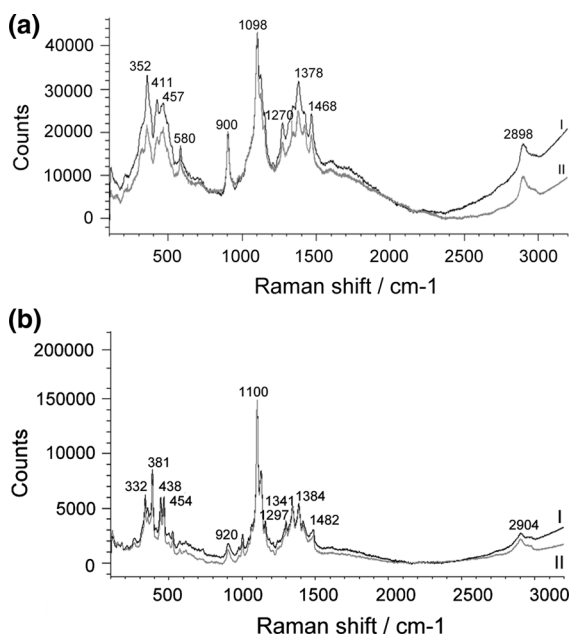
**Fig. 7** The wettability of **a** raw cotton, **b** 30-min plasma-treated cotton, **c** raw viscose cotton and **d** 30-min plasma-treated viscose







**Fig. 8** SEM images of **a** raw cotton fabric and **b** cotton fabric after MW treatment at 650 W, 15 min with  $\times 10,000$  magnification



**Fig. 9** Raman spectra of **a** cotton and **b** viscose fabrics: raw (I) and after MW treatment, 650 W, 15 min (II)

stretching at  $2904\text{ cm}^{-1}$  (Fig. 9b); this also did not reveal changes between raw viscose (spectrum I) and MW-treated viscose (spectrum II) (Kavkler and Demšar 2011).

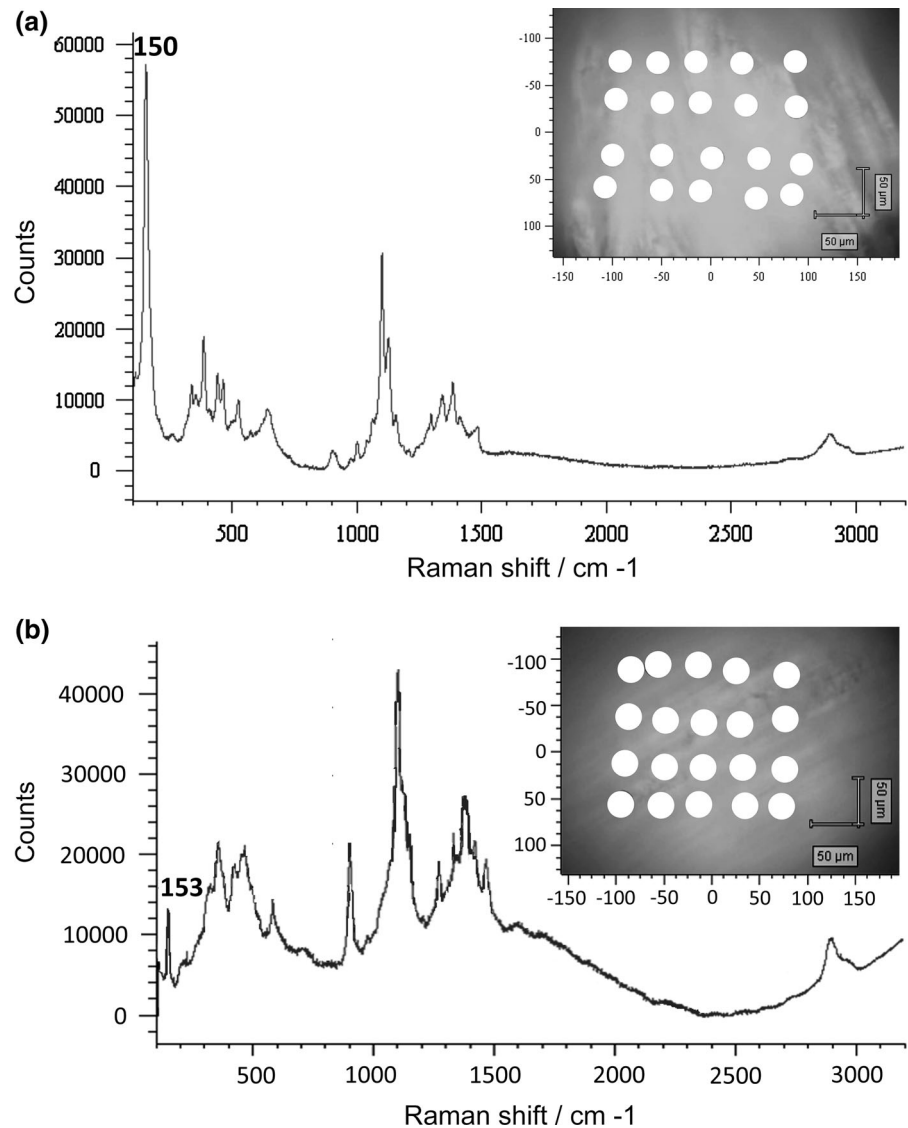
Raman spectra and map pictures obtained for cotton (Fig. 10a) and viscose (Fig. 10b) fabric modified with  $\text{TiO}_2$  sol and MW treated at 650 W for 15 min are presented. The detected spectra prove that on both fabrics  $\text{TiO}_2$  is present in the anatase form (band at

$150\text{ cm}^{-1}$  for cotton and  $153\text{ cm}^{-1}$  for viscose) (Fig. 10a, b, respectively). These strong sharp peaks are analogous to the  $142\text{-cm}^{-1}$  anatase O–Ti–O bending vibration (Ohsaka et al. 1978). As a result of the physical interaction between the titania and fabric, the main anatase band is shifted to larger wavelengths. Other anatase bands at 194, 393, 515 and  $637\text{ cm}^{-1}$  (Ohsaka et al. 1978) were not observed because there are strong bands of cotton and viscose fibers in this range. Raman map pictures show the distribution of  $\text{TiO}_2$ -anatase on the fiber surface. The dots on the pictures indicate where the Raman spectra were collected and the main anatase characteristic band was found.

The XRD results for  $\text{TiO}_2$  coatings on both fabrics confirmed the formation of crystalline  $\text{TiO}_2$ . Figure 11 presents the XRD spectra of cotton (Fig. 11a) and viscose (Fig. 11b) modified with  $\text{TiO}_2$ . The characteristic peaks centered at two theta around 29.50; three peaks that appeared as one wide one (43.22; 44.24; 45.11) and 56.00 are visible, being characteristic for the  $\text{TiO}_2$ -anatase phase (ref. pattern no. 00-064-0863). The intensity of the registered peaks is rather low; moreover, they are more pronounced in the case of  $\text{TiO}_2$ -modified cotton fabric. This is also consistent with the results of Raman spectroscopy.

After reducing the power and time of MW treatment to 500 W and 10 min, the  $\text{TiO}_2$  film on the fibers surface was a mixture of the anatase and amorphous form (Fig. 12). Besides the characteristic anatase bands, two bands at 440 and  $620\text{ cm}^{-1}$  assigned to the amorphous form of titania were found (Zhang et al. 1998), clearly visible for viscose

**Fig. 10** Raman spectra and Raman maps of **a** cotton and **b** viscose fabrics with TiO<sub>2</sub> after MW treatment at 650 W, 15 min (white dots on the map—TiO<sub>2</sub> anatase)



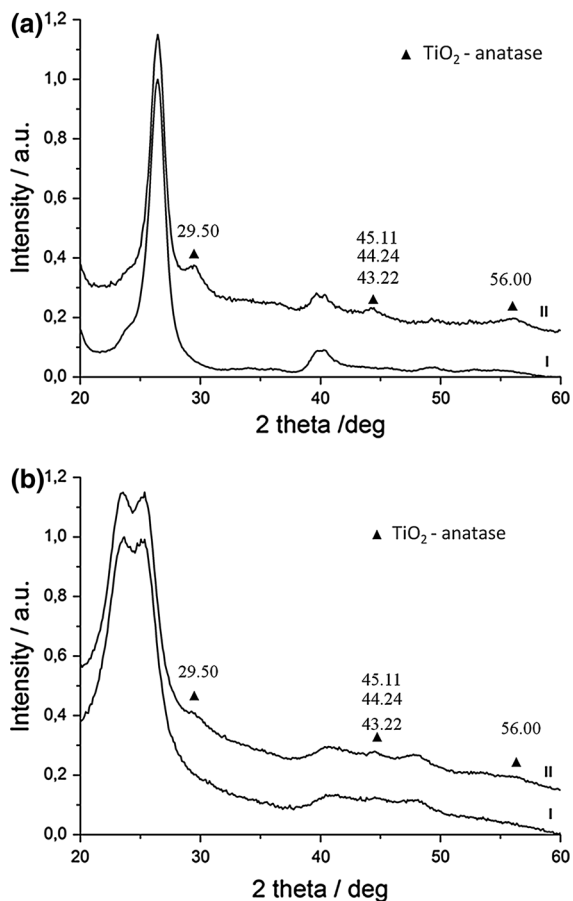
(Fig. 12b). In the case of cotton (Fig. 12a), the band at  $440\text{ cm}^{-1}$  is an overlap with the cotton bands present in this region. Results of Raman analysis show that for the transformation of TiO<sub>2</sub> to the anatase form, suitable MW treatment conditions must be selected.

SEM images of TiO<sub>2</sub> coatings in the anatase form on the Si wafers and fibers are presented in Figs. 13, 14 and 15.

TiO<sub>2</sub> film on the Si wafer (Fig. 13) creates a film with varied surface topography compared to smooth TiO<sub>2</sub> film—anatase on the Si wafer obtained by calcination for 2 h at 500 °C (Cieslak et al. 2015). TiO<sub>2</sub> forms a continuous coating on the cotton and

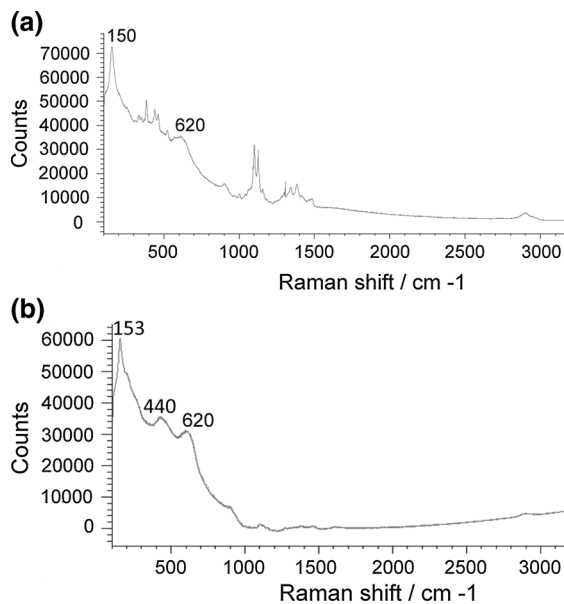
viscose fibers without any gaps or agglomerates (Figs. 14, 15). SEM/EDS maps show that Ti is distributed on all fibers, white dots—Ti (Figs. 14c, 15c). The thickness of TiO<sub>2</sub> film is  $180 \pm 7$  and  $140 \pm 8$  nm for cotton and viscose, respectively. The content of Ti evaluated on the basis of SEM/EDS analysis is nearly four times higher for cotton. It can be concluded that compared to the roughness of the fiber surface of viscose, the thickness of the TiO<sub>2</sub> layer and Ti content are higher for cotton fabrics.

The different behaviors of both of the studied cellulose fabrics toward plasma pretreatment and the effect of TiO<sub>2</sub> sol application are related to differences

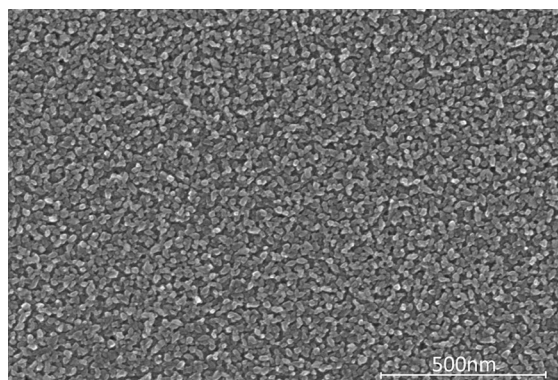


**Fig. 11** XRD spectra of **a** cotton and **b** viscose I raw and II with TiO<sub>2</sub> after MW treatment at 650 W, 15 min

in their morphology and topography. Surface ablation during plasma treatment is different for the cotton fiber and viscose fibers. The cotton fiber with a bean-shaped cross section has a concentric layered structure with an outer separable cuticle layer composed of wax and pectin materials, which are removed during the pretreatment processes (scouring and bleaching). The next, primary wall is composed of cellulosic crystalline fibrils, and the secondary three-layer wall consists of closely packed parallel cellulose fibrils. The degree of polymerization of cotton is 9000–15,000, and the crystallinity of cellulose is approximately 70 %. In the viscose fiber with a skin-core structure, the skin contains numerous small crystallites with a larger total surface area than the core composed of fewer but larger crystallites. Viscose is characterized by the more than 30 times lower degree of polymerization, and its crystallinity degree



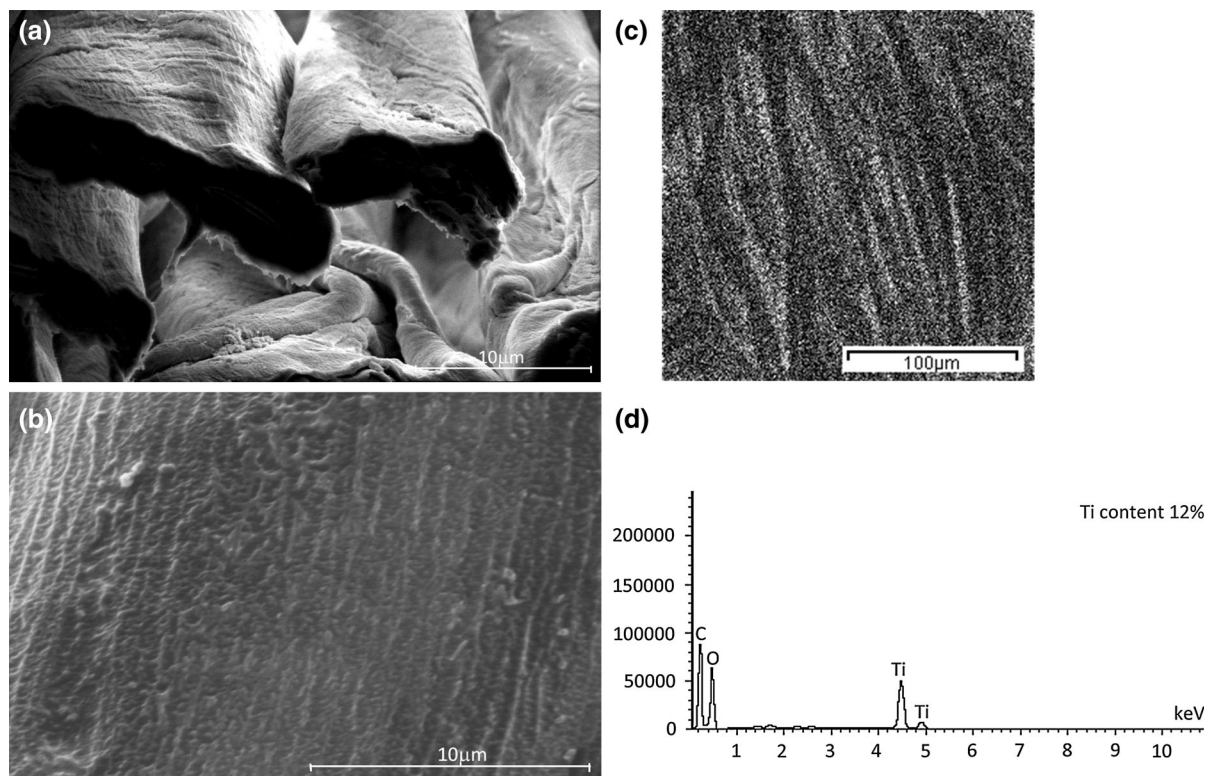
**Fig. 12** Raman spectra of **a** cotton and **b** viscose fabrics with TiO<sub>2</sub> film after MW treatment at 500 W, 10 min



**Fig. 13** SEM images of TiO<sub>2</sub> film on Si wafers (smooth substrate) after MW treatment at 650 W, 15 min ( $\times 120,000$  magnification)

is lower by almost 40 % in comparison to cotton (Duckett 1975).

The unmodified and modified cotton fabric showed higher susceptibility to nicotine sorption compared to viscose fabric. On the basis of exponential equations, the constant rate of nicotine decomposition was determined ( $k$ ). The ratio of constant  $k$  for modified and unmodified cotton and viscose fabrics is 3.3 and 2.0, respectively. After application of the total irradiation dose of 405.8 kJ/m<sup>2</sup>, the nicotine concentration decreased for unmodified and modified cotton and



**Fig. 14** SEM images of **a, b** TiO<sub>2</sub> coatings on cotton after MW treatment at 650 W, 15 min with  $\times 6500$  and  $\times 30,000$  magnifications, **c** SEM/EDS map of Ti distribution and **d** EDS spectrum of cotton textile modified with TiO<sub>2</sub> after MW treatment at 650 W, 15 min

viscose fabrics by about 21, 56, 45 and 64 %, respectively. A faster decrease in the nicotine concentration of modified fabrics results from their photocatalytic properties (Fig. 16).

The Ti contents before and after washing were evaluated on the basis of SEM/EDS analysis (Fig. 17). After washing, the Ti content on cotton fabric decreased from 12 to 11 % and on viscose fabric from about 3 to 2.6 %. Windler et al. (2012) investigated the release of titanium dioxide from consumer products (T-shirts and pants) of different compositions, also including cotton. The results indicated that functional textiles release some TiO<sub>2</sub> particles, but the amount is relatively low and mostly not in the nanoparticulate range, and Ti can occur.

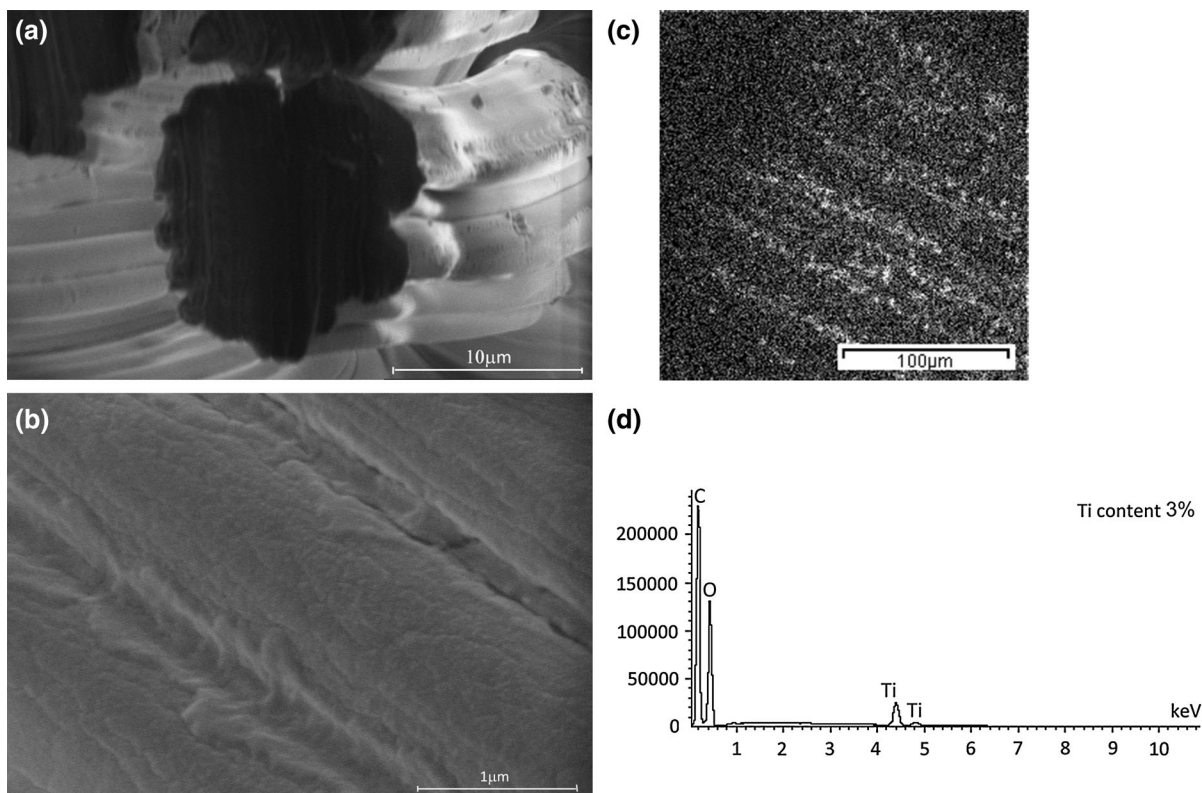
The average values of the breaking force and breaking elongation of both fabrics before and after applied modifications are listed in Table 3.

Slight changes of the tensile strength of cotton after each modification (plasma treatment, microwave treatment and plasma/TiO<sub>2</sub>/microwave treatment)

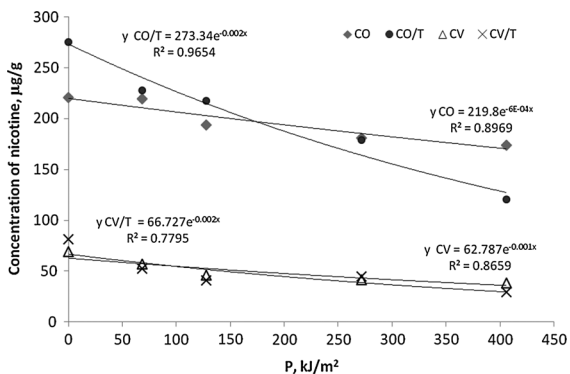
were observed. It can be concluded that cotton fabric retains its physical properties after modifications. Contrary to cotton fabric, the tensile strength of viscose fabric decreased after plasma treatment (about 14 %), microwave treatment (about 19 %) and the most for TiO<sub>2</sub>-modified fabric; the breaking force dropped from 499 to 275 N (about 45 %). Breaking elongation for cotton and viscose increased after the applied modifications. The developed method of modification with TiO<sub>2</sub> and microwave treatment tested on two cellulose fabrics is more suitable for cotton fabric because of the significant reduction of the tensile strength of viscose fabric.

The influence of UV irradiation on raw and TiO<sub>2</sub>-anatase-modified fabrics was evaluated (Fig. 18). The results show that raw and TiO<sub>2</sub>-modified cotton fabrics are stable under UV irradiation. In case of raw and TiO<sub>2</sub>-modified viscose fabrics, UV irradiation affects their tensile strength. The tensile strength of raw viscose after UV irradiation of 5184 kJ/m<sup>2</sup> decreased by nearly 7 %, and there were minor changes. For





**Fig. 15** SEM images of TiO<sub>2</sub> coatings **a, b** on viscose after MW treatment at 650 W, 15 min, with  $\times 6500$  and  $\times 50,000$  magnifications, **c** SEM/EDS map of Ti distribution and **d** EDS spectrum of viscose modified with TiO<sub>2</sub> after MW treatment at 650 W, 15 min



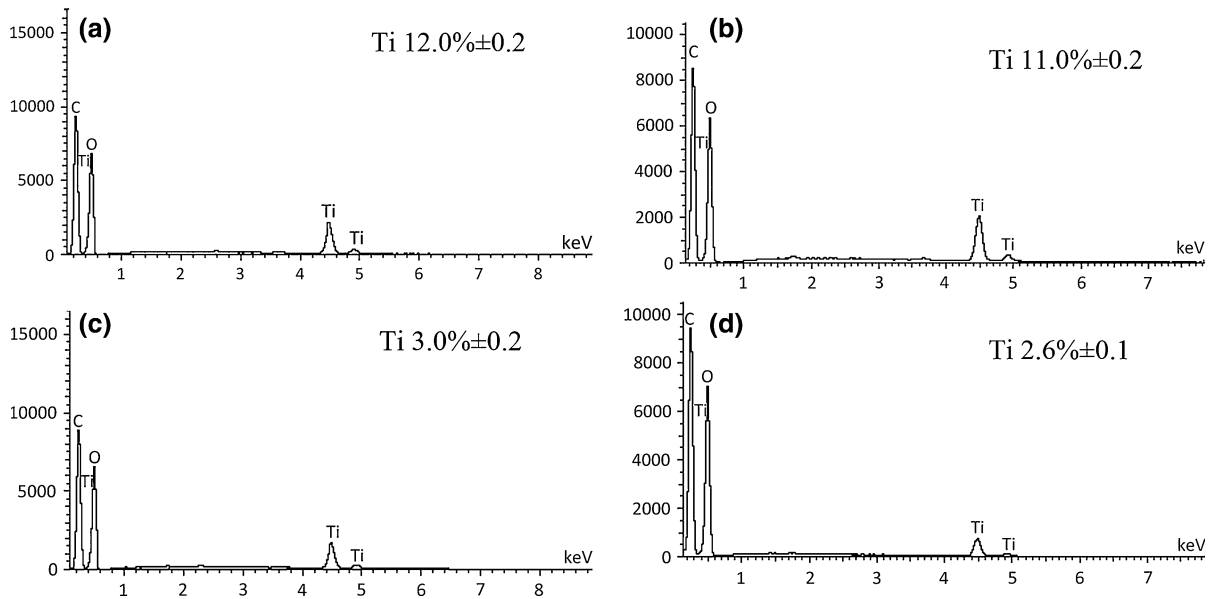
**Fig. 16** The influence of the sunlight irradiation dose ( $P$ ) on the nicotine concentration for unmodified and TiO<sub>2</sub> modified cotton and viscose fabrics. *CO* cotton, *CO/T* cotton modified with TiO<sub>2</sub>, *CV* viscose, *CV/T* viscose modified with TiO<sub>2</sub>

TiO<sub>2</sub>-modified viscose fabric, the tensile strength decreased by about 16 % (from 275 to 232 N) (Fig. 18a). Breaking elongation for cotton and viscose fabrics increased after UV irradiation (Fig. 18b).

## Conclusions

Application of microwave treatment to the formation a TiO<sub>2</sub>-anatase layer directly on cotton and viscose fabrics was reported. In this study, two selected conditions—650 W, 15 min and 500 W, 10 min—were investigated. The microwave conditions influenced the TiO<sub>2</sub> form; it obtained an anatase form (650 W, 15 min, 181 °C) and a mixture of the anatase and amorphous form (500 W 10 min, 110 °C). The conversion of TiO<sub>2</sub> to the anatase form takes place at temperatures below 200 °C, which opens up many possibilities for modifications of different fibers with low thermal stability. The microwave step replaces the calcination process at around 500 °C and can only be used for fibers with high thermal resistance. To improve the surface properties of fibers and the TiO<sub>2</sub> sol application, low-temperature air plasma pretreatment was used. This resulted in better wettability of both fabrics by TiO<sub>2</sub> sol and significant growth in nano-roughness. The value of the root mean square





**Fig. 17** Results of EDS analysis for **a** cotton fabric before washing, **b** cotton fabric after washing, **c** viscose fabric before washing and **d** viscose fabric after washing

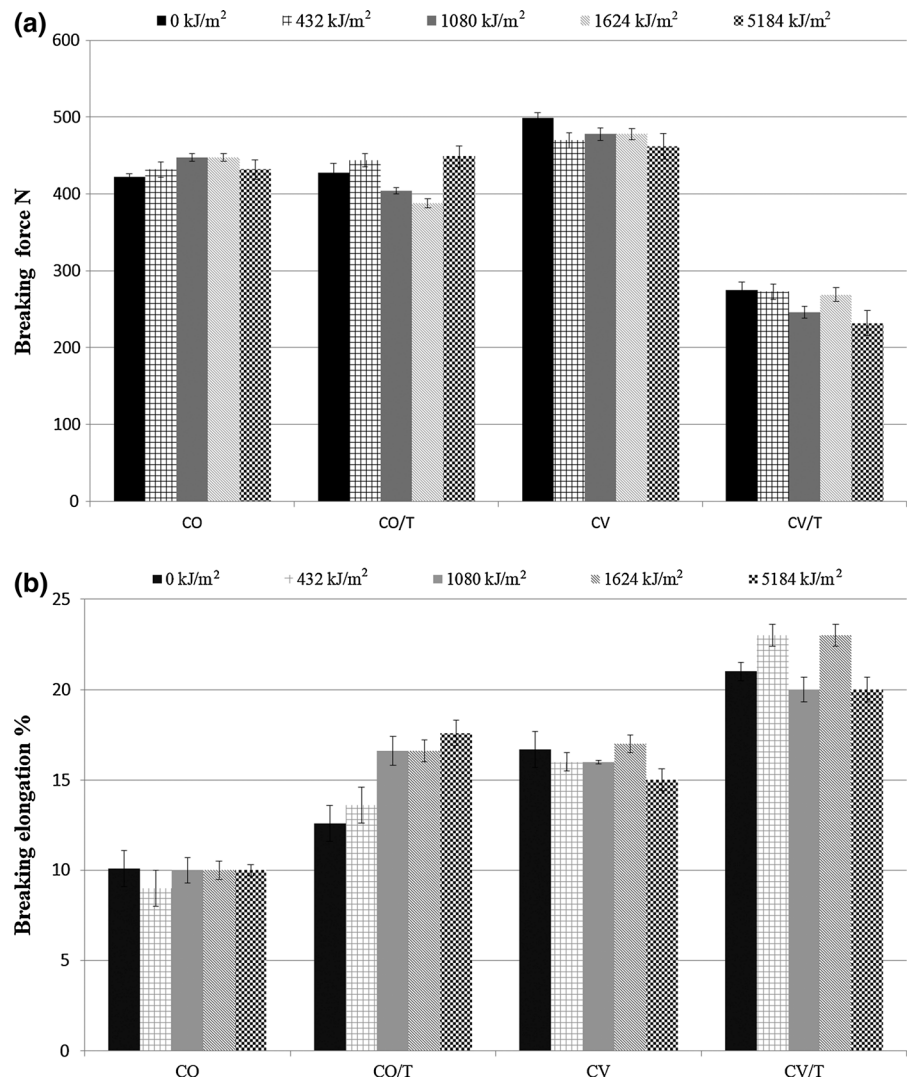
**Table 3** Results of the tensile strength of cotton and viscose fabrics before and after each step of modification

Fabric sample	Cotton		Viscose	
	Breaking force (N)	Breaking elongation (%)	Breaking force (N)	Breaking elongation (%)
Raw	422 ± 5	10.1 ± 0.1	499 ± 4	16.7 ± 0.5
Plasma treated	405 ± 7	10.7 ± 0.3	431 ± 6	17.5 ± 0.2
Microwave treated	450 ± 8	13.5 ± 0.8	404 ± 4	26.6 ± 0.9
Plasma/TiO <sub>2</sub> /microwave treated	428 ± 5	12.6 ± 0.6	275 ± 8	21.0 ± 0.2

(RMS) increased from 44 to 166 nm (cotton) and from 9 to 112 nm (viscose). The surface free energy increased by about 26 and 23 mJ/m<sup>2</sup> for cotton and viscose, respectively. In the FTIR spectrum, there were new bands at 1748 and 1732 cm<sup>-1</sup> corresponding to C=O stretching, which causes the increase in the hydrophilicity and adhesion properties. The final effect of modification differs for viscose and cotton fabrics. Generally, the wettability of cotton by TiO<sub>2</sub> sol is better, the nano-roughness and TiO<sub>2</sub> layer thickness are higher, and the amount of Ti is four times higher compared to viscose fabrics. TiO<sub>2</sub>-modified cotton and viscose fabrics accelerate the decomposition of the adsorbed nicotine and thus its accumulation in fibers. The ratio of constant k for modified and

unmodified cotton and viscose fabrics is 3.3 and 2.0, respectively. The nicotine concentration decreased for unmodified and modified cotton and viscose fabrics by about 21, 56, 45 and 64 %, respectively, after a UV irradiation dose of 405.8 kJ/m<sup>2</sup>. No significant strength reduction after modifications and UV irradiation of cotton fabric was observed. For viscose, a UV dose irradiation of 5184 kJ/m<sup>2</sup> decreased the tensile strength by about 7 and 16 % for raw and TiO<sub>2</sub>-modified viscose, respectively. The tensile strength of the viscose fabric after TiO<sub>2</sub> anatase modification reduced by about 45 %. The Ti contents after washing decreased from 12 to 11 % (cotton) and from 3 to 2.6 % (viscose). The durability of the TiO<sub>2</sub> film on the fabric surface was satisfactory. The developed method

**Fig. 18** Results of **a** breaking force and **b** breaking elongation for cotton and viscose fabrics before and after UV irradiation. *CO* cotton, *CO/T* cotton modified with  $\text{TiO}_2$ , *CV* viscose, *CV/T* viscose modified with  $\text{TiO}_2$



of modifying cotton and viscose with  $\text{TiO}_2$  to obtain the anatase form directly on the cellulose fabric surface is more effective for cotton fabric because of the reduction of the tensile strength of viscose fabric.

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