

Early Steps of Homogeneous Catalysis in Rennes: Carbon Dioxide Incorporation, Alkyne Activation and Ruthenium Catalysis

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Abstract This review reports the first discoveries in homogeneous catalysis in Rennes. Ruthenium(II) catalysts are shown to promote the activation of terminal alkynes and the synthesis of vinylcarbamates on addition of CO₂ and amines. The ruthenium–vinylidene are the key catalytic species. Applications to the addition of carboxylic acids to alkynes have opened the access to various enolesters and dienylesters via controlled catalyzed Markovnikov or anti-Markovnikov additions. The catalytic synthesis of α -methylene cyclic carbonates and oxazolidinones from propargylic alcohols and CO₂ is presented and shown to lead to allene derivatives via palladium catalysis. Cp**Ru*X(COD) catalyst promote the head to tail oxidative couplings of alkene with alkyne, and the synthesis of functional dienes via 1,4 addition of H-nucleophiles to the biscarbene-ruthenium intermediate arising from head to head couplings of alkynes, whereas Cp**Ru*L₂⁺ catalysts favour activation of alkynes via vinylidene intermediate.

Keywords Ruthenium catalysis · Vinylidenes · Carbon dioxide · Carbonates · Alkyne activation · Cp**Ru*XL₂ catalysts

Dedicated to Michael I. Bruce and Barry M. Trost who inspired our early steps in catalysis by their pioneered works, respectively on vinylidene-metal complexes and ruthenium catalysis for synthesis.

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1 Introduction

Homogeneous catalysis in Rennes started almost 30 years ago when we explored with an industrial company some ways to use CO₂ as a phosgene substitute. Our first steps were guided by the aim to discover selective combinations of simple accessible substrates, avoiding protecting groups and toxic reagents. This led us to discover an atom economy catalytic synthesis of vinylcarbamates and oxoalkylcarbamates from terminal alkynes, CO₂ and secondary amines with ruthenium catalysts. This catalytic synthesis was applied to the selective Markovnikov or anti-Markovnikov addition of carboxylic acids to alkynes affording enol esters and 1,3-dienol esters. These new catalytic transformations of alkynes questioned the involvement of vinylidene–ruthenium in catalysis which was proved, developed in Rennes and led to a new chemistry of the stoichiometric activation of terminal alkynes and preparation of vinylidene-, carbene-, allenylidene- cumulenyliidene- and indenylidene-ruthenium complexes and to the chemistry of carbon- rich organometallic complexes.

The direct incorporation of CO₂ as a functional group was developed via the organocatalytic activation with phosphine of propargyl alcohol derivatives to produce a variety of unsaturated cyclic carbonates which became the straightforward precursors of oxazolidinones, allenols, unsaturated ketones or heterocycles.

In parallel to the activation of alkynes to generate active vinylidene intermediates our Rennes team explored the selective transformation of alkynes by oxidative coupling with electron-rich ruthenium(II) catalysts, derivatives of Cp**Ru*XL₂ complexes, (Cp* = C₅Me₅), which was revealed first to be an excellent route to various functional dienes.

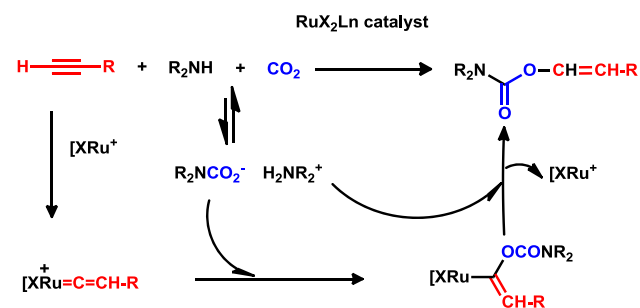
The above innovations in catalytic syntheses involving alkynes, by rising questions about mechanisms greatly contributed to the improvement of ruthenium organometallic chemistry.

At the time CO₂ chemistry is again the target of investigations to take profit of this stable, simple and abundant small molecule, and that selective catalytic transformations of alkynes is still contributing to both organic synthesis and molecular materials science, it might be of interest to tell this story on the first 10 years of catalysis in Rennes starting in 1985.

2 Ruthenium-Catalyzed Synthesis of Vinyl-carbamates

In cooperation with the French company SNPE, we were interested in finding solutions to replace toxic phosgene derivatives in some industrial processes. One target was the production of vinylcarbamate, a varnish precursor, previously produced from R₂NCOCl and a metal-enolate derivative. As a secondary amine and carbon dioxide are in equilibrium with ammonium carbamate we explored the activation of alkynes with potential electrophilic ruthenium catalysts to favour nucleophilic addition of carbamates to readily available alkynes. This work was initiated by a post-doctor fellow from Tsukuba, Yoshiyuki Sasaki [1], and a Rennes PhD student, Roger Mahé [2].

Our first attempt with Ru₃(CO)₁₂ as a catalyst for activation of phenylacetylene led to moderate yields of *Z* and *E*-PhCH=CH-O₂CNEt₂ [1]. The use of more electrophilic ruthenium(II) catalysts drastically improved the reaction of secondary amine with alkyne under 50 bar of CO₂. RuCl₂(py)₂(nbd), (py = pyridine, nbd = norbornadiene) and RuCl₂(PMe₃)(C₆Me₆) catalysts led to the formation of a variety of vinylcarbamates (Scheme 1) [2]. The *Z* isomer was always the major isomer and the use of RuCl₃·H₂O also led to the *Z* > *E* isomers of carbamates PhCH=CH-O₂CNR₂.



Scheme 1 Catalytic synthesis of vinylcarbamates via ruthenium-vinylidene

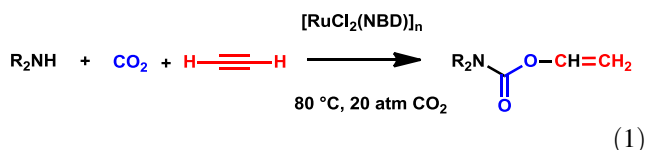
The only anti-Markovnikov addition products were observed. In addition the reaction was restricted to terminal alkynes. We could already propose [2] that this catalytic reaction proceeded via the in situ formation of a ruthenium(II)-vinylidene species [Ru(II)=C=CHR] for which the coordinated carbon was expected to be electrophilic. A metal-vinylidene species in catalysis was proposed for the first time suggesting that ruthenium(II)-vinylidenes are more active toward nucleophiles than ruthenium(II)- η^2 -alkyne intermediates.

This view was realistic as some metal-vinylidene complexes were already known. The first well-recognized vinylidene complex MCl(C₅H₅)(L)₂(C=C(CN)₂) (M=Mo, W) was obtained by R. B. King by chloride migration of an α -chloroalkenyl ligand [3]. Cationic (M=C=CHR⁺ vinylidene complexes were previously made by protonation of metal-alkynyl derivatives [4–7]. An early established method of preparation of metal vinylidene consisted also in the addition of an electrophilic group at the β -carbon of alkynyl-metal complexes [8–10]. The most straightforward method appeared to be the direct activation via 1,2-hydrogen migration of η^2 -coordinated terminal alkynes [11]. Only a few examples were shown later to proceed via oxidative addition of alkyne C–H bond to metal center followed by 1,3-hydride migration [12]. Very soon metal-vinylidene reviews appeared [13–15] pointing out the methods of synthesis and giving evidence for the electrophilic character of the carbon linked to the metal which favoured nucleophilic addition on that carbon atom.

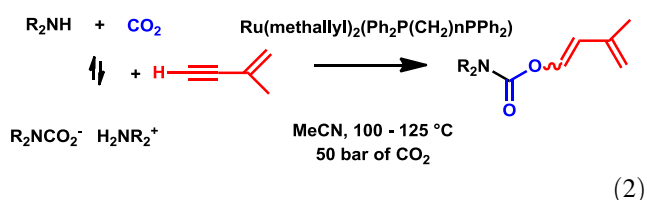
Le Bozec [16] demonstrated that the catalyst precursors RuCl₂(PR₃)(arene), used for catalytic vinylcarbamate synthesis, under very mild conditions, activate terminal alkynes via a vinylidene–ruthenium(II) intermediates to generate a variety of ruthenium–carbene complexes, simply on addition of alcohols [16]. Consequently the proposed mechanism as soon as 1986, based on vinylidene–ruthenium intermediate (Scheme 1) [2] was realistic and only the last step, the protonation of the Ru site or of the Ru–C bond could not be observed since.

This catalytic synthesis of vinylcarbamates was boosted when Christian Bruneau, a CNRS fellow, joined our group in the fall 1986. The reaction could be applied to a variety of secondary amines and terminal alkynes using catalysts of type RuCl₂(PR₃)(arene) and preferentially Ru(methyl)₂(diphosphine). It was shown that the simple vinylcarbamate could be prepared from acetylene itself [17, 18]. The simple catalyst RuCl₃·3H₂O could perform the vinylcarbamate synthesis from acetylene in up to 46 % yield in acetonitrile under 15 atm of CO₂ [17]. [RuCl₂(NBD)]_n, which in situ generates RuCl₂(HNR₂)₂(NBD) in the presence of amine was shown by C. Bruneau and S. Lécolier (SNPE) to be the most efficient catalyst precursor for this

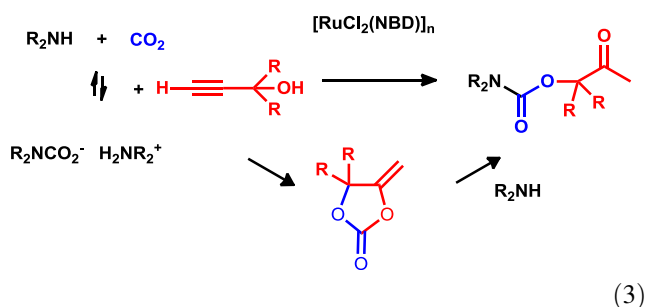
transformation of acetylene into the useful vinylcarbamates (20 atm CO₂, 80 °C) [18] (Eq. 1)



C. Bruneau and J. Höfer showed that the catalytic addition of ammonium carbamate to isopropenylacetylene led to the regioselective synthesis of *Z* > *E* *O*-1-(1,3-dienyl)carbamates (Eq. 2) [19]. For this reaction the best ruthenium(II) catalysts were of the type Ru(methallyl)₂(Ph₂P(CH₂)_nPPh₂). The complete evaluation of the previously cited ruthenium(II) catalysts for catalytic vinylcarbamate synthesis appeared in 1989 [20]



When the above reaction of secondary amines and CO₂ was applied to propargylic alcohols, it led to the selective formation of *O*-β-oxopropylcarbamates. In the presence of [RuCl₂(NBD)]_n catalyst precursor 30–40 % yields of carbamates were obtained (Eq. 3) [21, 22].

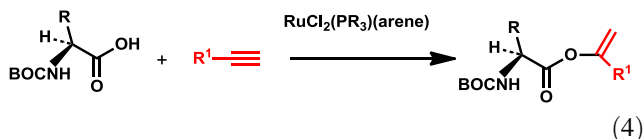


As the triple bond is transformed into an acetyl group, the reaction could proceed via initial carbamate addition at carbon C2 of the coordinated alkyne. However it will be shown later that when the reaction was performed in the presence of NEt₃, the unsaturated α-methylene cyclic carbonate was formed in small amount, thus the oxoalkyl carbamates can also arise from secondary amine addition to the cyclic carbonate (Eq. 3) (see Sect. 4.)

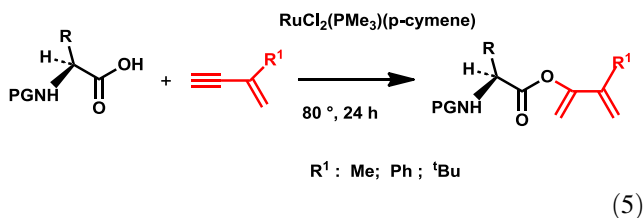
3 Ruthenium-Catalyzed Synthesis of Enol Esters from Terminal Alkynes

The regioselective synthesis of vinylcarbamates led us to evaluate the ruthenium(II) activation of alkynes towards carboxylic acids addition in an attempt to produce enol and dienol esters useful acylating reagents, such as vinylacetate or in peptide synthesis, or polymer precursors. According to the nature of both catalysts and substrates, both regioselective *Markovnikov* and *anti-Markovnikov* additions could be obtained.

The regioselective *Markovnikov* addition of carboxylic acids to terminal alkynes was first performed by C. Ruppin with RuCl₃·3H₂O/2 PR₃ but preferentially with RuCl₂(PR₃)(arene) [23]. The reaction was applied to the addition of *N*-protected aminoacids to butylacetylene and propyne [24] without racemization at 80 °C and the products were used in peptide synthesis by Z. Kabouche and C. Bruneau (Eq. 4) [24, 25]. These reaction conditions with RuCl₂(PR₃)(arene) catalysts were much milder than those with Ru₃(CO)₁₂, [Ru(O₂CH)(CO)₂]_n [26] or Ru(cyclooctadienyl)₂/PBu₃ [27]. The mechanism studies concluded that the reaction proceeded via external addition of carboxylate to the η²-coordinated triple bond.

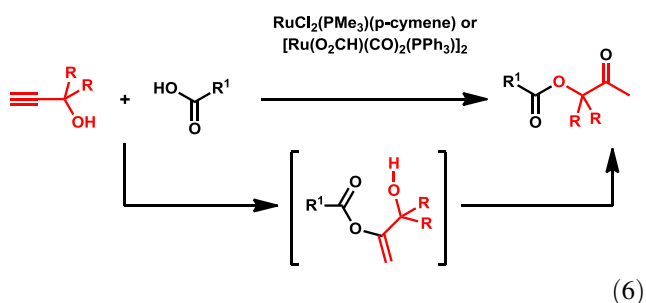


The addition of carboxylic acids to isopropenylacetylene, performed by K. Philippot, selectively afforded 2-acyloxy-1,3-dienes using the RuCl₂(PMe₃)(*p*-cymene) catalyst. The reaction can also be applied to optically active aminoacid derivatives without racemization (Eq. 5) [28].



D. Devanne showed that the same catalyst RuCl₂(PMe₃)(*p*-cymene) could perform the addition of carboxylic acids and aminoacids to propargylic alcohols to produce β-oxopropylesters without racemization (Eq. 6) [29].

The formation of these esters is consistent with external initial addition of the carboxylate to the substituted carbon of the coordinated triple bond.



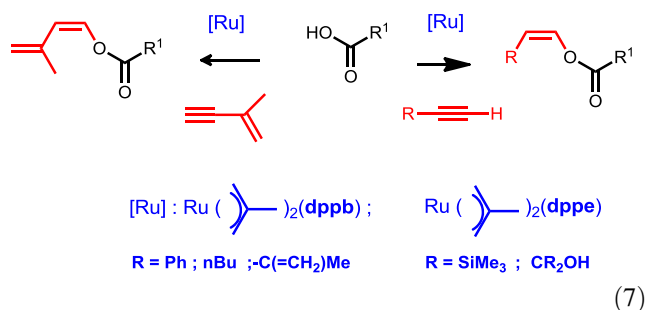
This synthesis of enol esters was applied by M. Neveux and C. Bruneau for the production of enol formates that were revealed as excellent formylation reagents under mild conditions [30]. They also showed that oxalic acid add to alkynes to form enol oxalyl esters which easily gave access to 1,2-dicarbonyl derivatives [31]. With B. Seiller they revealed that the $[\text{Ru}(\text{O}_2\text{CH})(\text{CO})_2(\text{PPh}_3)_2]$ catalyst could perform tandem catalysis offering the direct access to 1,3-dioxan-4-ones, via enol esters arising from addition of α -hydroxy carboxylic acids to alkynes followed by intramolecular catalytic addition of the hydroxyl group to the enol double bond [32, 33].

$[\text{Ru}(\text{O}_2\text{CH})(\text{CO})_2(\text{PPh}_3)_2]$ was found by C. Darcel and C. Bruneau to be the most efficient catalyst to activate propargylic alcohols for the formation of β -ketoesters in the presence of carboxylic acid and has allowed to produce optically active β -ketoesters with retention of configuration at the stereogenic carbon from chiral propargylic alcohols obtained by S. M. Roberts via enzymatic hydrolysis of their acetate [34]. Moreover they used the $[\text{Ru}(\text{O}_2\text{CH})(\text{CO})_2(\text{PPh}_3)_2]$ catalyst for the stereoselective transformation at C(17) of steroids of propargylic derivatives into steroids with β -oxopropyl ester functionality [35].

It is noteworthy that the ruthenium catalytic addition of carboxylic acids to terminal alkynes and propargylic alcohols actually inhibited ruthenium–vinylidene formation. It was suggested that this inhibition was brought by the concomitant protonation of the ruthenium center and external addition of the carboxylate to the coordinated triple bond [33, 34].

All the previous examples were consistent with *Markovnikov addition* of carboxylate to the triple bond of terminal alkynes. However, H. Doucet, B. Martin-Vaca, N. Derrien and C. Bruneau have demonstrated that the use of $\text{Ru}(\text{methallyl})_2(\text{diphosphine})$ catalyst precursors has allowed to produce *anti-Markovnikov addition* of carboxylic acids to terminal alkynes and to isopropenylacetylene

affording preferentially the *Z*-enol- or dienol-esters with high *Z*-stereoselectivity (>95 %) (Eq. 7) [36–39].



In that case H. Doucet showed that the catalyst precursors react with carboxylic acids to give $\text{Ru}(\text{O}_2\text{CR})_2$ (diphosphine) and isobutene inhibiting ruthenium protonation and favouring activation of alkynes via vinylidene intermediate. The nature of the diphosphine was crucial as bis(diphenylphosphino)butane (dppb) favours formation of *Z*-enol esters at low temperature whereas bis(diphenylphosphino)ethane (dppe) leads to better selectivity with bulky terminal alkynes (Eq. 7) [36–39]. These *Z*-enol esters were found to be useful for the access to (*E*)-enamines and to α -cyanoesters on reaction with amines and $\text{KCN}/\text{HSiEt}_3$, respectively [40].

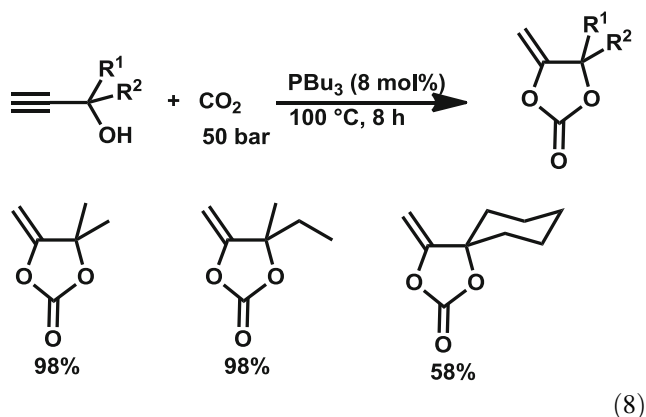
4 Catalytic Synthesis of Cyclic Carbonates and Carbamates

Our interest in directly incorporating CO_2 derivatives into alkynes, by routes avoiding the use of phosgene, led J. Fournier and C. Bruneau to discover an “organocatalytic” synthesis of cyclic carbonates and carbamates by a route different from the usual insertion of CO_2 into oxiranes and oxetanes.

During the study of the ruthenium catalyzed synthesis of β -oxoalkylcarbamates by reaction of CO_2 , HNR_2 and propargylic alcohol with $\text{RuCl}_2(\text{PBU}_3)(\text{arene})$ (Eq. 3), to understand the mechanism different ratios of $\text{Ru}(\text{II})$ catalyst and PBU_3 were used and by changing the amine by NEt_3 or H_2NR we observed that (i) PBU_3 alone catalyzed the reaction, (ii) with tertiary amine NEt_3 cyclic carbonate was partially formed, and (iii) with primary amine H_2NR α -methylene oxazolidinones were produced. Thus, tertiary propargylic alcohols with CO_2 (50 bar) in an inert autoclave with 8 mol % of PBU_3 led to the isolation (>95 %) of α -methylene cyclic carbonates, in the absence of other solvent than the propargylic alcohol. PBU_3 appeared to be more efficient than PPh_3 and PCy_3 (Eq. 8) [41, 42].

The formation of these cyclic carbonates can be explained by the addition of the phosphine to the triple

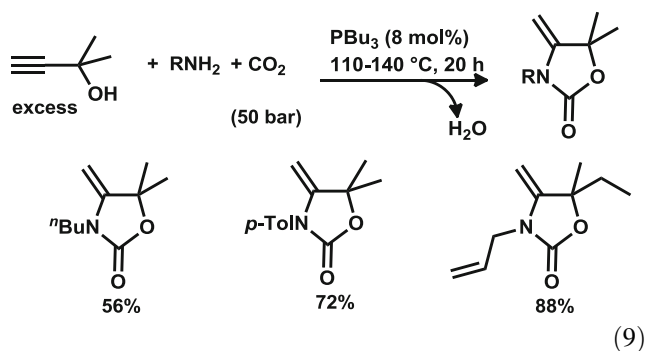
bond, transfer of alcohol proton to the alkenyl carbon, addition of CO₂ to the alcoholate and addition of the resulting carbonate oxygen to the alkenyl carbon with release of PBu₃.



However, under the same conditions with primary amines, but in the presence of both ruthenium(II) catalyst and PBu₃, the *N,N'*-disubstituted ureas were selectively formed [43].

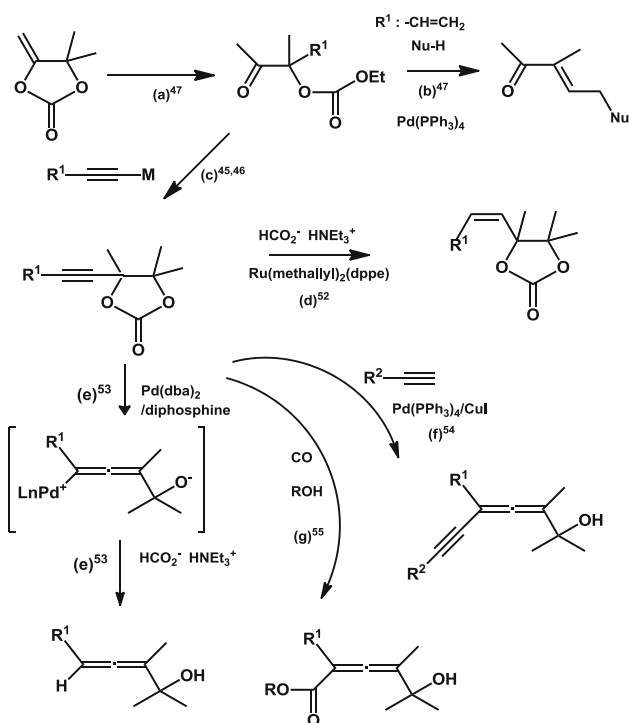
This easy formation of cyclic carbonates opened the route to many derivatives on addition of simple reagents such as primary amines leading to β-oxopropylcarbamates and to their oxazolidinones, as on addition of aminoacids [44], and to opened carbonates by addition of alcohols [45, 46] or to functional enones [47].

The PBu₃ catalyzed activation of propargylic alcohols was applied by J. Fournier and J. M. Joumier to the direct formation of α-methylene oxazolidinones via reaction of primary amine, CO₂ and a variety of tertiary propargylic alcohols (Eq. 9) [48].



The phosphine-catalyzed synthesis of α-methylene cyclic carbonates constituted actually a catalytic platform [49] for the fast access to functional allyl or alkynyl carbonates, and to allenyl alcohols and esters. C. Darcel developed this platform by palladium catalyzed transformations of these cyclic carbonates (Scheme 2).

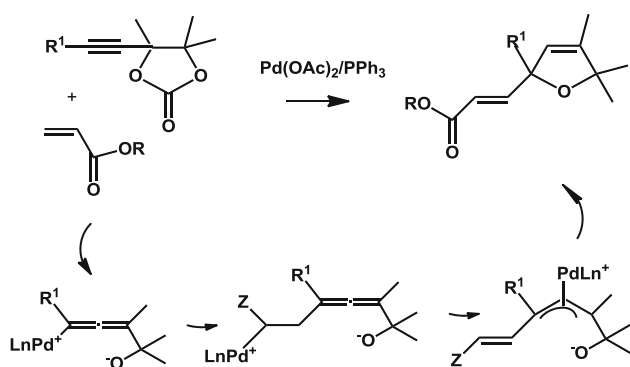
Cyclic carbonates were previously used for the Pd-catalyzed allylation of isocyanate and aldehydes leading to



Scheme 2 Alkynylcarbonate synthesis and Pd-catalyzed transformations

heterocycles [50, 51] We showed (Scheme 2) that these carbonates were straightforward precursors of allylcarbonates and of functional enones via Pd-catalyzed allylation of nucleophiles (*a,b*) [47]. The useful formation of cyclic alkynyl carbonates (*c*) [45, 46] offered the direct access to allylcarbonates on catalytic reduction with ammonium formate in the presence of Ru(methallyl)₂ (dppe) catalyst (*d*) [38], and to α-allenyl alcohols via catalytic hydrogenolysis with HCO₂⁻ HNEt₃⁺ and Pd(dba)₂/diphosphine catalyst (*e*) [52]. The reaction is expected to result from the formation of allenyl-Pd(II) species followed by hydride transfer from the coordinated formate. The cyclic alkynyl carbonates on reaction with terminal alkyne with Pd(PPh₃)₄/CuI catalyst offer a route to alkynyl allenyl alcohols (*f*) [53] which was applied to the modification of the C17 side chain of steroids [53]. C. Darcel showed that cyclic alkynyl carbonates can also be selectively carbonylated with carbon monoxide in alcohol to afford allenyl esters with Pd(dba)₂/PBu₃ catalyst (*g*) [54]. (Scheme 2).

A dihydrofuran synthesis was found by serendipity on Pd-catalyzed reaction of alkynylcarbonates with acrylic esters and acrylamides. However the reaction is very sensitive to the conditions to get excellent yield as it requires the use of Pd(OAc)₂/PPh₃ catalyst in the presence of NBu₄Br with NEt₃-H₂O in DMF or THF solvent. (Scheme 3) [55].



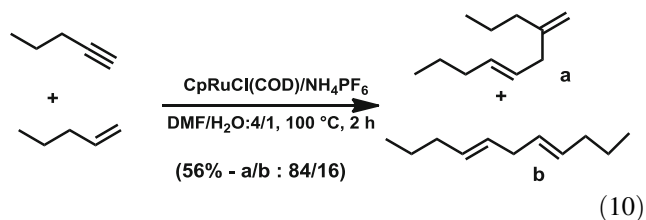
Scheme 3 Catalytic synthesis of functional dihydrofurans from alkynylcarbonates

5 Activation of Alkynes Via Oxidative Coupling Versus Vinylidene with $\text{Cp}^*\text{Ru}(\text{X})\text{L}_2$ Catalysts

5.1 Oxidative Couplings of $\text{C}\equiv\text{C}$ and $\text{C}=\text{C}$ bonds

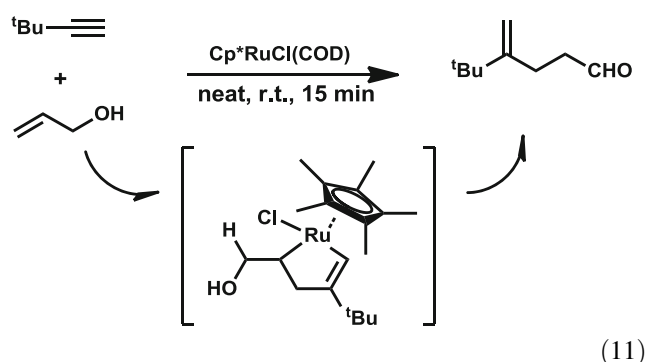
The catalytic synthesis of vinylcarbamates, via a vinylidene-ruthenium(II) active intermediate, resulted from alkyne activation with a 16 electron ruthenium(II) species for example of type $[\text{RuX}(\text{PR}_3)(\text{arene})]^+\text{Y}^-$. We decided to explore the activation of terminal alkynes with ruthenium(II) catalyst precursors capable to provide a coordinatively unsaturated 14 electron catalytic species, by displacement of weakly coordinating ligands, but at the same time more electron-rich ruthenium(II) sites to favour oxidative couplings between two functional unsaturated molecules.

Initial results by B. M. Trost already showed that $\text{CpRuCl}(\text{COD})$, ($\text{Cp}=\text{C}_5\text{H}_5$), in the presence of NH_4PF_6 allowed the regioselective oxidative coupling of an alkyne and a functional alkene and led to the formation of a 1,4-diene with branched derivative as a major isomer (Eq. 10) [56, 57].

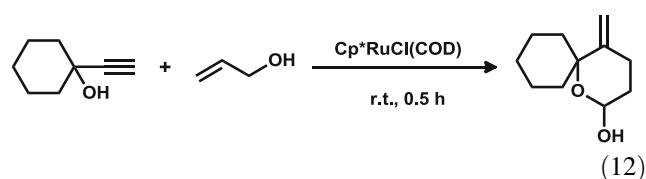


B. M. Trost showed also that the same catalyst $\text{CpRuCl}(\text{COD})$ could offer the access to unsaturated ketones from terminal alkyne and substituted allylic alcohols. In that case the oxidative coupling preferentially afforded the linear isomer [58].

Sylvie Dérien, as Rennes assistant professor, explored the reaction of terminal alkynes with simple allyl alcohol using a more sterically hindered and electron-rich catalyst $\text{Cp}^*\text{RuCl}(\text{COD})$ with labile COD which led to a high regioselective oxidative coupling. Only γ,δ -unsaturated aldehydes were selectively obtained (Eq. 11) [59, 60] and the branched isomer was only formed, due to the bulkiness of the $\text{Cp}^* = \text{C}_5\text{Me}_5$ ligand which favours the head-to-tail coupling of the triple and double bonds to give a metallacyclopentene. The exocyclic beta-elimination and reductive elimination explain the product formation. A vinylidene intermediate would have led to vinyl allyl ether. This reaction showed that the $[\text{Cp}^*\text{RuX}]$ moiety favours oxidative coupling versus vinylidene intermediate and that the bulkiness of the C_5Me_5 ligand drives the coupling regioselectivity.

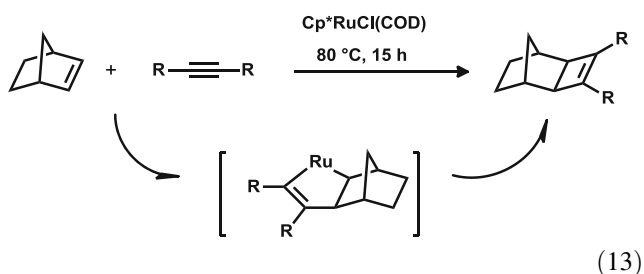


This reaction applied to a propargyl alcohol as terminal alkyne and the allyl alcohol with the same catalyst $\text{Cp}^*\text{RuCl}(\text{COD})$ led selectively to 2-alkoxy-5-methylene-tetrahydropyranes resulting also from head-to-tail coupling and formation of a hemiacetal (Eq. 12) [61, 62]



In the mean time Mitsudo used $\text{Cp}^*\text{RuCl}(\text{COD})$ for the catalytic coupling of a strained alkene (norbornene, norbornadiene) with an internal alkyne. In that case the ruthenacyclopentene intermediate cannot offer a beta-elimination process, thus the cyclobutene derivative was selectively obtained via reductive elimination (Eq. 13) [63]. Later this reaction was applied for the [2+2] cycloaddition of a variety of substituted norbornenes and

norbornadienes with alkynes [64–66]



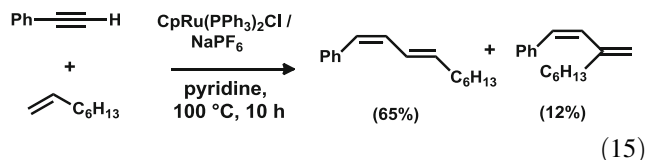
The catalytic precursor based on Cp^{*}RuXL₂ with labile L ligands offers till now the best catalytic system for regioselective oxidative couplings of alkyne with an alkene. Many useful applications in organic synthesis have been found later via intramolecular oxidative couplings of triple and double bonds by several groups especially by Trost's group [67–70] and gathered in a 2004 review [71].

5.2 Ruthenium–Vinylidene Intermediates with Cp^{*}RuXL₂ Catalysts

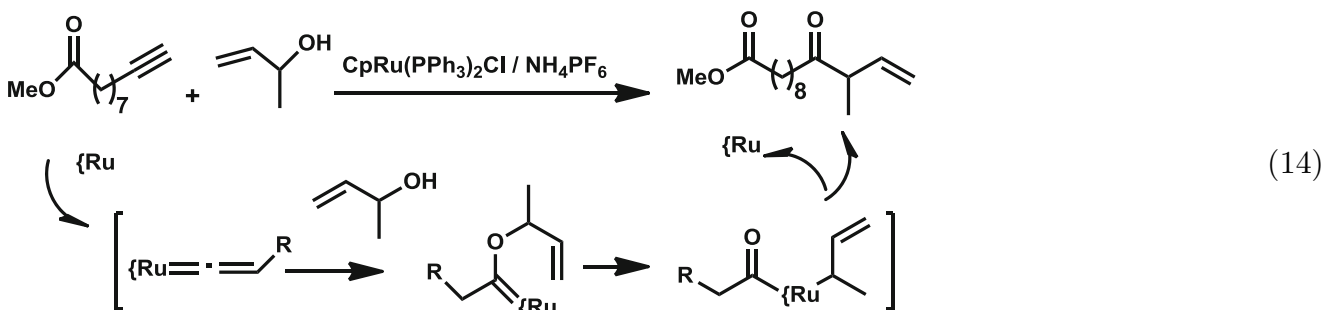
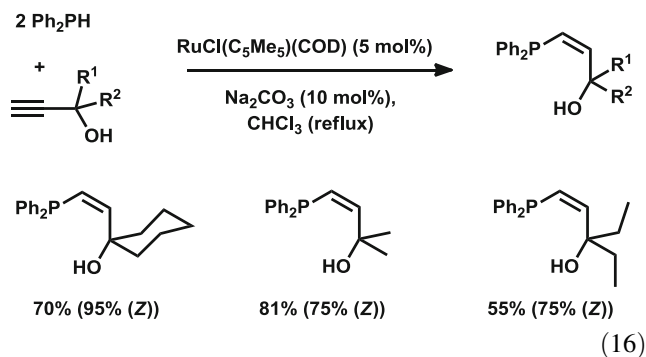
The development of vinylidene-metal intermediates has led to many applications in synthesis and we have gathered these discoveries in several reviews [15, 72, 73]. Special efforts have been also made to generate vinylidene active species arising from the terminal alkyne by activation with Cp^{*}RuXL₂ catalysts. However the examples were reported only with Cp^{*}RuXL₂ derivatives containing strong Ru–L bond such as Ru–PR₃ bond, thus affording the 16 electron cationic activating species Cp^{*}RuL₂⁺. First B. M. Trost showed that using the CpRu(PPh₃)₂Cl/NH₄PF₆ catalytic system, the addition of allylic alcohols to terminal alkynes was possible and took place affording β,γ-unsaturated ketones, via addition of allyl alcohol to the vinylidene intermediate and C–O bond cleavage and allyl migration (Eq. 14) [74–77].

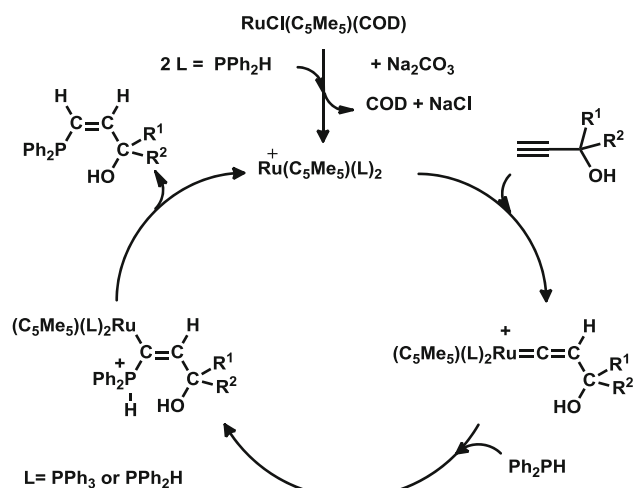
Ruthenium vinylidene intermediates were also proposed by M. Murakami using a similar catalytic system CpRu(PPh₃)₂Cl/NaPF₆ for the coupling of unactivated alkenes with terminal alkynes to afford 1,3-dienes as a mixture of two isomers, the linear one being favoured (Eq. 15) [78].

The same catalytic system has allowed later the useful regioselective ruthenium-catalyzed alkenylation at C2 position of pyridine with terminal alkyne [79].



We have discovered with F. Jérôme, a post doctor fellow, and PhD student J. Monnier another application of vinylidene intermediate arising from propargylic alcohol: the hydrophosphination of propargylic alcohols on addition of HPPH₂, in the presence of Cp^{*}RuCl(COD) catalyst. This reaction affords regioselectively the Z-hydroxyl alkenylphosphine as the kinetic product which isomerizes easily into the E isomer (Eq. 16) [80].





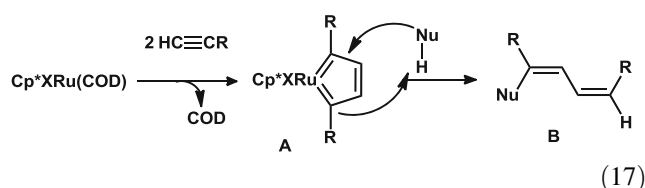
Scheme 4 Mechanism of catalytic hydrophosphination of propargylic alcohols

The formation of the vinylidene intermediate results from the initial displacement of the COD ligand by two HPPH₂ ligands and the activation of the terminal propargylic alcohol with the 16 electron intermediate before addition of the secondary phosphine to the vinylidene carbon (Scheme 4) [80].

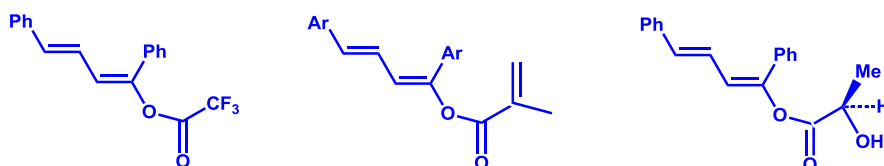
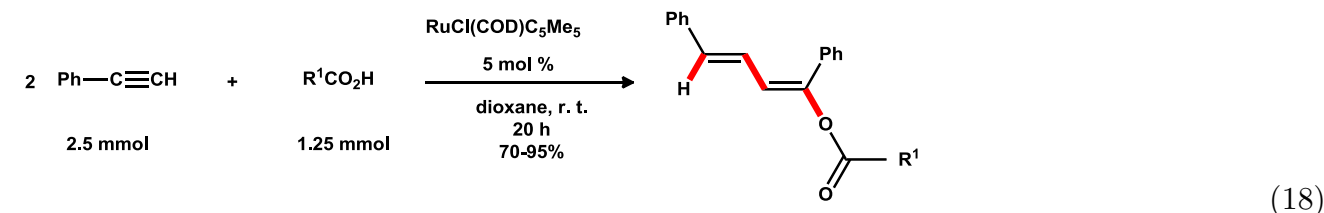
The power of CpRu(Cl)L₂ catalysts for the generation of vinylidene–ruthenium or alkylidene–ruthenium intermediates in synthesis has been illustrated in several reviews [71, 81].

as shown before (Sect. 5.1) [71]. However when the alkene is absent then head-to-head oxidative addition of two molecules of terminal alkyne takes place. The isolated intermediate surprisingly was shown by Singleton et al. as early as 1986 [82] to contain a “bis-carbene”-ruthenium structure **A** (Eq. 17), rather than the expected ruthenacyclopentadiene which is observed only on addition of a phosphine to the ruthenium [83].

S. Dérien decided to explore the “bis-carbene” properties of the in situ produced intermediate from terminal alkyne and Cp^{*}RuCl(COD) and addition of pronucleophiles to alkynes was attempted. The first example reported PhD student Jacques Le Paih showed the addition of a proton at one carbene carbon and the addition of nucleophile at the other carbene carbon, thus creating 3 bonds leading to the diene product **B** (Eq. 17) [84].



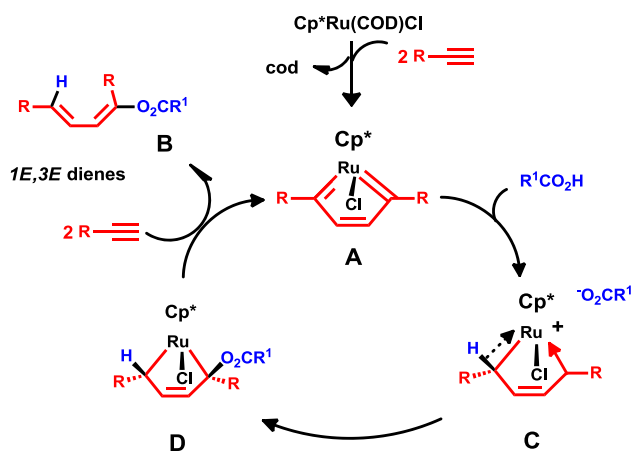
Thus the simple addition of carboxylic acids to aryl-alkynes quantitatively and stereoselectively led at room temperature in one step to 1,3-dienes, containing a carboxylate group at carbon C1 (Eq. 18) [84].



5.3 Oxidative Coupling of Terminal Alkynes with Cp^{*}RuXL₂ Catalysts as a Route to Functional 1,3-Dienes

It is noteworthy that the Cp^{*}RuXL₂ catalysts in the presence of both an alkyne and an alkene offer the cross oxidative coupling of the two different unsaturated bonds

The catalytic addition mechanism has been elucidated by DFT calculations with the help of Odile Eisenstein and Eric Clot (Scheme 5) [85]. It shows that the metallacyclic intermediate that behaves as a mixed Fischer-Schrock-type biscarbene-ruthenium complex, allowing protonation first at the Schrock type carbene carbon, not at the metal, to give the intermediate **C**, followed by nucleophilic addition



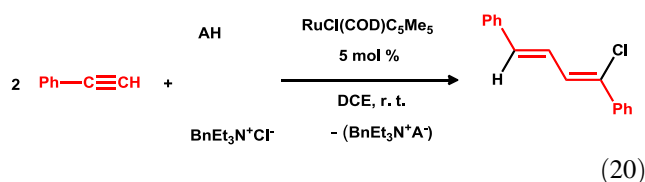
Scheme 5 Mechanism of addition of carboxylic acids to terminal alkynes

of the carboxylate at the Fischer type carbene carbon in **D**. The possibility to form the expected mixed alkyl allyl intermediate instead of **C** was ruled out. Another approach has been proposed recently for which the carboxylate coordinates to the ruthenium site before addition to the carbon C1 [86].

Later Min Zhang showed that the addition of alcohols and water to the head-to-head dimerized alkyne ruthenium species can also take place, however subtle catalyst modifications have to be done. As alcohols are less acidic than carboxylic acids the cationic catalyst $\text{Cp}^*\text{Ru}(\text{CH}_3\text{CN})_3^+ \text{PF}_6^-$ has to be used to form dienyl alkyl ethers (Eq. 19) [87]. For the addition of water, catalytic amount of a strong acid such as paratoluenesulfonic acid p-TSA needs to be used to protonate the biscarbene intermediate to lead to the quantitative formation of dienol and the unsaturated ketone (Eq. 19) [87].

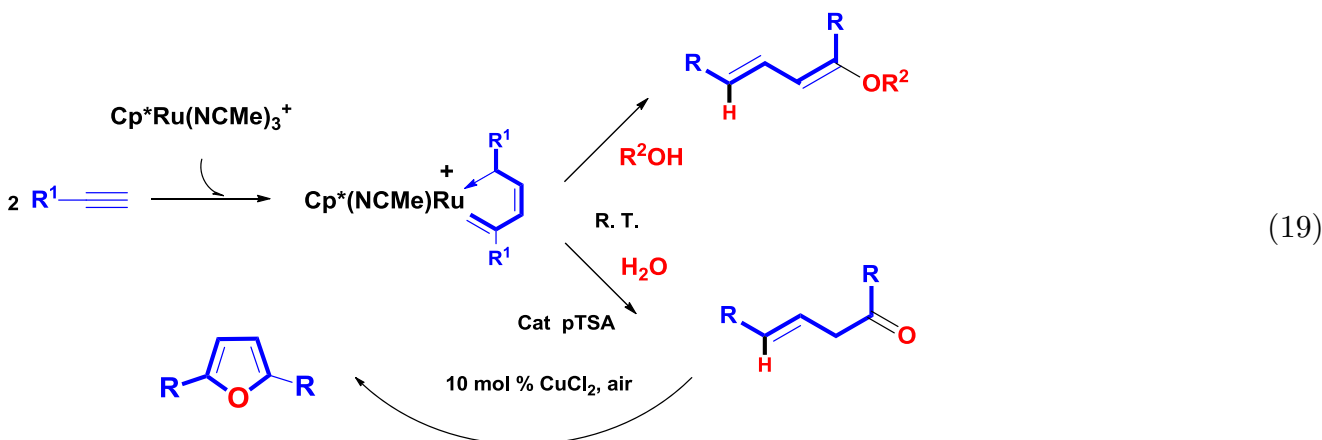
The easy synthesis of unsaturated ketones from alkynes allowed to perform sequential catalysis. Thus after formation of the ketone, addition of copper(II) catalyst in air atmosphere allowed its transformation into furan derivative (Eq. 19) [87]. Alternatively the ruthenium catalyzed formation of unsaturated ketone followed by silver catalysis allow the hydroarylation of the double bond with electron rich arenes [88].

Recently this catalytic 1,4-addition of a proton and anion on the biscarbene intermediate was used by S. Dérian for the preparation of 1,3-dienylchlorides and bromides by addition of camphorsulfonic acid (AH) and a chloride or bromide salt, using $\text{Cp}^*\text{RuCl}(\text{COD})$ catalyst (Eq. 20) [89]. This reaction led preferentially to the *1Z,3E* dienes

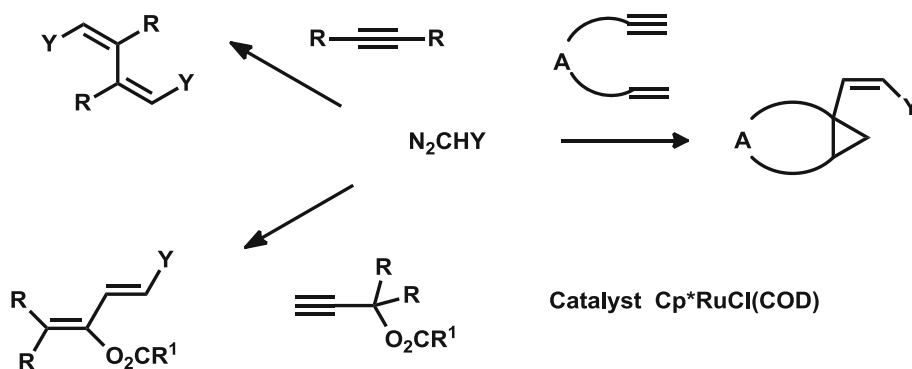


The above dimerization of alkynes via biscarbene-ruthenium intermediate has potential for the formation of a variety of functional dienes, via 1,4-addition, but also for molecular materials preparation, by catalytic modification of currently used alkyne group for electronic communication, and for oligomerization/polymerization of rigid conjugated bridged diynes.

The ability of Cp^*RuXL system to accommodate carbene has led to develop the chemistry of $\text{Cp}^*\text{RuX}(\text{=CHY})$ intermediates, often via the activation of diazoalkanes. These intermediates behave very differently of the iso-electronic Grubbs catalysts of type $\text{RuX}_2(\text{=CHY})\text{L}_2$ as they inhibit alkene metathesis processes [81].



Scheme 6 $\text{Cp}^*\text{RuX}(\text{COD})$ catalyzed of diazoalkanes to various alkenes



Profit of $\text{Cp}^*\text{RuX}(\text{=CHY})$ intermediates has recently been made for the synthesis of functional dienes from alkynes [90], and from propargylic carboxylates [91], and for the access to alkenyl bicyclic derivatives (Scheme 6) [92].

6 Concluding Remark

Our initial results in homogeneous catalysis presented above have deeply fed our further research topics till now. First the search for evidence of active vinylidene–ruthenium species in the ruthenium(II) activation of terminal alkynes has been the first step to develop the topic of carbon-rich ruthenium systems with D. Touchard [93] and A. Romero, via the characterization of vinylidene–allenylidene- and cumulenyliidene–ruthenium complexes [94–96]. Among these we reported in 1998 the evidence for the first allenylidene–ruthenium catalytic precursors for alkene metathesis [72, 97], and R. Castarlenas demonstrated that the resulting catalytic species was the corresponding indenylidene–ruthenium complex easily generated on protonation of the allenylidene–ruthenium parents [98, 99]. Indenylidene–ruthenium catalysts are now commonly used in catalysis based on their efficiency and the easiness of preparation from simple propargylic alcohols rather than from diazoalkanes [72, 73]. Moreover the Rennes allenylidene- and indenylidene–ruthenium catalysts led us to develop many applications in alkene metathesis [100–103] some of them in cooperation with IFP [103–105] and Arkema [106, 107] companies, and this topic is still a strong research topic in Rennes.

Our research topic on enantioselective catalysis, in cooperation with the company Oril-Servier, deeply involved the ruthenium-catalyzed hydrogenation of cyclic carbonates and oxazolidinones for the production of optically active diols from α -methylenedioloxanones [108] and of the optically active N-acyloxazolidinones, the Evans reagents, from α -methylene cyclic carbonates [109].

Finally the use of Cp^*RuXL_2 catalysts which revealed both the oxidative coupling of alkynes into biscarbene intermediate in catalysis, and the stabilization of Cp^*Ru -alkylidene and their vinylalkylidene– RuCp^* species has later allowed to discover novel syntheses of functional dienes and cyclopropyl and bicyclic derivatives [81].

The above examples of both activation of alkynes and ruthenium-catalyzed syntheses have contributed over the years to make of ruthenium an attractive and tamed metal for catalysis for the synthesis of complex architectures and molecular materials. It is no longer a matter of “connoisseurs”.

Acknowledgments All the students and researchers who started homogeneous catalysis in Rennes, whose the names appear in the references, are warmly acknowledged for their hard and enthusiastic work and for successfully paving the roads for their successors who innovated more recently in catalysis and organometallic chemistry. PHD thanks the industrial companies SNPE, Oril industrie, IFP, Firmenich, Arkema for their financial and scientific support. PHD also thanks the CNRS, French Ministry of research, University of Rennes, Region of Bretagne and European Networks who continuously supported his research on catalysis and Institut Universitaire de France for membership.

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