



# Groundwater recharge processes in the Lake Chad Basin based on isotopic and chemical data

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## Abstract

The Lake Chad Basin is Africa's largest endorheic basin. Because water supply for the rural population and most of the urban population depends on groundwater, assessment of groundwater recharge is crucial. Recharge sources for the upper Quaternary aquifer are precipitation, rivers, and swamps. Using water chemistry, and environmental (<sup>18</sup>O, <sup>2</sup>H, <sup>3</sup>H) and carbon (<sup>14</sup>C) isotopes, recharge processes can be assessed and groundwater ages roughly estimated. For this purpose, more than 1,000 samples from groundwater, surface water and precipitation were analysed for hydrochemistry and environmental stable isotopes. Furthermore, <sup>3</sup>H measurements and <sup>14</sup>C values of dissolved inorganic carbon for groundwater from the northeastern part of the Basin are included in the evaluation. The environmental isotope distribution shows recent recharge from precipitation north of Lake Chad (Kanem Region), where very low <sup>3</sup>H values indicate occurrence before the 1960s bomb peak. Focused recharge from fresh river water is typical for Salamat Region in south Chad and the Komadugu Yobe wetlands between Nigeria and Niger. Slightly high δ-values in water occur in the Waza Logone area between Chad and Cameroon. Groundwater along the Lake Chad shore and the Bahr el Ghazal corridor show high δ-values (δ<sup>18</sup>O –0.78 to 7.45‰, δ<sup>2</sup>H –13.6 to 30.8‰). Recharge is caused by surface water that undergoes evaporative processes before percolation. Groundwater ages of 600–4,150 years, estimated from <sup>14</sup>C analyses combined with high SO<sub>4</sub> concentrations, along the Bahr el Ghazal indicate that recharge was caused by residuals of the Mega Lake before it dried out completely.

**Keywords** Isotope hydrology · Groundwater renewal · Africa · Transboundary aquifer · Arid regions

## Introduction

Water availability worldwide is under threat because of rapid population growth and the consequent increase in withdrawal for domestic, industrial, and irrigation uses (Mohan et al. 2018). As a result, both surface water and

groundwater resources are experiencing overexploitation. Regarding groundwater, 37 of the world's largest aquifer systems were studied by Richey et al. (2015a, b), of which 21 were reported to be depleted. When adding the climate change factor, it becomes clear that there is an urgent need for adequate management of these resources to ensure social and economic development.

Groundwater recharge is often the limiting factor for groundwater withdrawal under sustainable conditions (Döll and Flörke 2005); thus, the estimation of present recharge is critical, especially in arid and semiarid areas. This is particularly true for the 400-km-wide band across Africa known as the Sahel region, where large aquifers are present, but data collection concerning climate, precipitation, and groundwater head is very scarce due to lack of monitoring stations.

The Lake Chad Basin (LCB) is an endorheic basin located in the central part of northern Africa. It has an area of 2.4 million km<sup>2</sup> and is inhabited by approximately 45 million people (LCBC 2016). Because of its location and extension, the LCB

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mirrors the climatological, hydrological, and societal situation of the semiarid Sahel region. It lies in an area of transition from perennial to ephemeral river systems and where reliable water supply using surface water suffers uncertainty. Therefore, it is often subject to conflicts among water uses (Ashton 2002), especially in the surroundings of the Lake Chad (Rizzo 2015). It is here where effective water management is needed the most.

Studies on groundwater recharge in arid to semiarid areas involve mainly local or semiregional evaluation. They often use the chloride mass balance technique (Gaye and Edmunds 1996; Edmunds et al. 2002; Edmunds and Tyler 2002) combined with stable water isotopes (Allison and Hughes 1983; Barnes and Allison 1988; Barnes et al. 1989; Leduc et al. 2000; Beyer et al., 2015; Adomako et al. 2010; Gaj et al. 2016; Tewolde et al. 2019; Goni et al. 2019; Mahamat Nour et al. 2022) and hydrochemistry (Njitchoua et al. 1995; Edmunds et al. 1999; Zagana et al. 2007a). Recharge has also been assessed by means of the mass balance equation (Zagana et al. 2007b; Ndubuisi 2007) and combination of empirical models (Adelana et al. 2006). Additionally, modelling of  $^3\text{H}$  and  $^{14}\text{C}$  (Favreau et al. 2002) and modelling of environmental isotopes by including evaporative fractionation in soil profiles (Zhou et al. 2021) or the analysis of aquifer hydrodynamic characteristics (Ngounou Ngatcha et al. 2007) have been used to determine recharge. Recharge distribution has also been mapped using satellite-based thermal data (Leblanc et al. 2003). Many of these studies were included in a global assessment of groundwater recharge in semiarid areas by Scanlon et al. (2006).

Using numerical flow modelling, diffuse recharge has been calibrated for the whole LCB (Boronina et al. 2005; Vaquero et al. 2021) and limited regions like the Waza Logone area (Candela et al. 2014; Nkiaka et al. 2018), the Chari Baguirmi region (Massuel 2001), and the Komadugu Yobe River region (Leblanc 1997; Zairi 2008; Genthon et al. 2015). Diffuse recharge distribution has been also determined based on chloride-36 analyses throughout the area (Bouchez et al. 2019).

This work will concentrate on the distribution of environmental isotopes combined with  $^3\text{H}$  values and chemical composition of groundwater as well as assessment of groundwater age through  $^{14}\text{C}$ , to attempt to define areas of present and past recharge throughout the whole LCB. A very valuable and extensive amount of data is available at the Lake Chad Basin Commission (LCBC) from a series of cooperation projects with the German government.

## Study area

### Location, climate, and hydrology

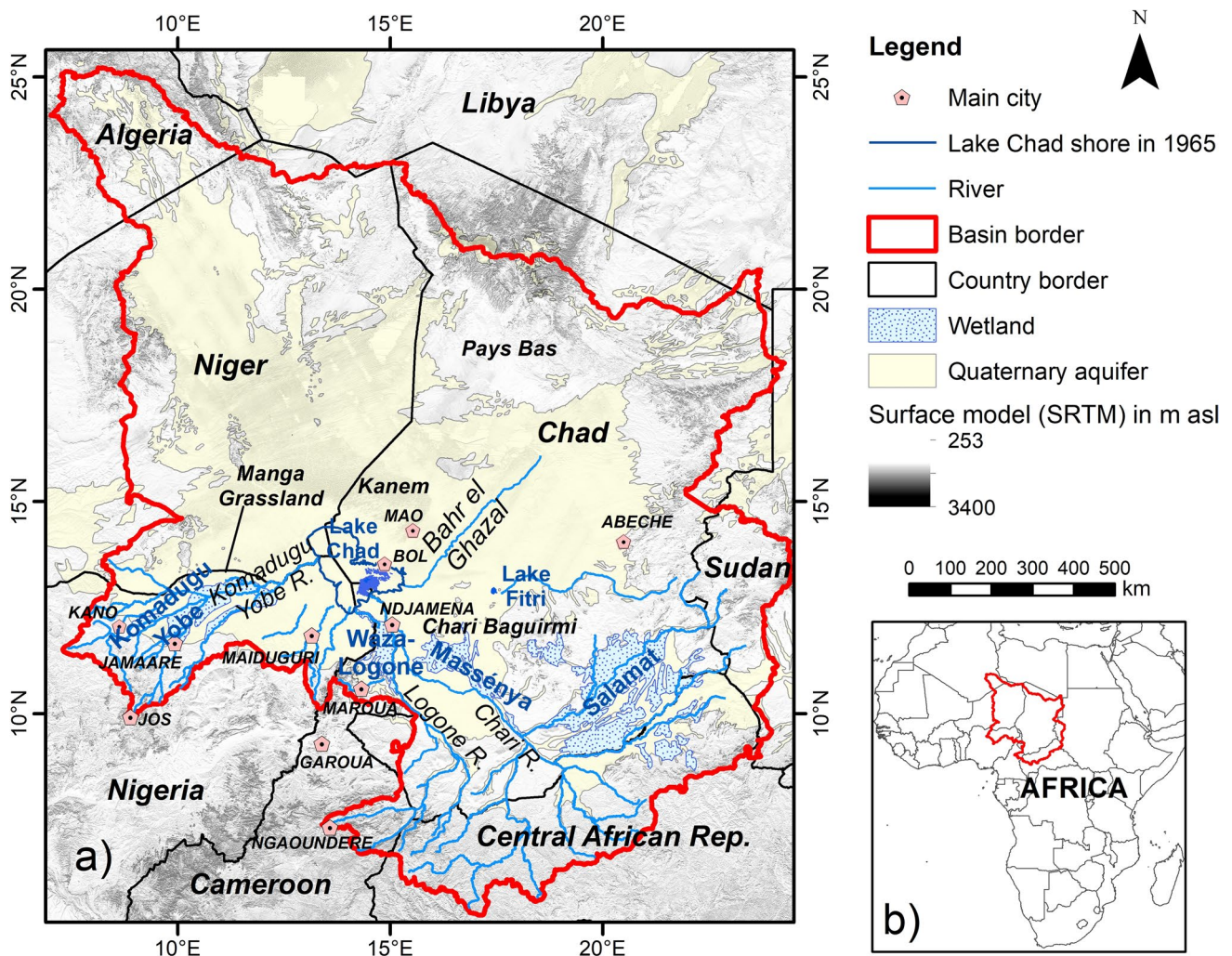
The LCB is located in the central part of northern Africa and extends from 5 to 25°N and from 7 to 25°E (Fig. 1).

Due to its size, the LCB expands over various climates and ecosystems. Following the updated Köppen-Geiger climate classification (Peel et al. 2007), the northern part of the basin corresponds to Saharan climate (hot arid desert) with very low precipitation, mean annual temperature above 18 °C, daily temperature amplitude greater than 35 °C, and annual temperature amplitude greater than 60 °C (White 1983). Southward lies the 400-km-wide Sahelian band (hot arid steppe). It is characterised by mean annual temperatures between 26 and 30 °C and precipitation between 150 and 500 mm (White 1983). Precipitation occurs over 3–4 months and is closely linked to the West African Monsoon, which is known for its extremely variable manifestation (Hall and Peyrillé 2006). To the south, the tropical region (tropical savannah) is encountered. Here mean annual temperature varies in a range of 24–28 °C and mean annual precipitation is from 500 to 1,400 mm (White 1983) or 1,700 mm according to ERA5 data for the period 1979–2005.

From a hydrological point of view, the most prominent feature in the region is Lake Chad, which receives most of its water from the tropical savannah through the Chari-Logone River system (Fig. 1). The Komadugu Yobe River usually provides small flow to the lake (Olivry et al. 1996), but it has become insignificant in recent decades because of flow retentions and diversions (Geerken et al. 2011). Several wetlands and periodically inundated areas spread over the LCB, like the lakes Chad and Fitri, the Waza Logone wetlands on the border between Chad and Cameroon, the Komadugu Yobe wetlands in Nigeria, the Salamet wetland between Chad and Central African Republic, and the Massénya wetland in Chad (Fig. 1).

## Hydrogeology

The upper unconfined to semiconfined aquifer is composed of Quaternary aeolian sands (Fig. 1) interbedded by fluvio-lacustrine silt and clay in areas prone to periodical flooding along the Lake Chad shore and the wetlands (Durant and Mathieu 1980; Schneider and Wolff 1992; Moussa et al. 2013). This large transboundary aquifer extends over most of the LCB and has an area of ~500,000 km<sup>2</sup> (Leblanc et al. 2007) with an average thickness of 100 m that varies from 190 m in the Kanem dunes to the north in Chad to a few meters towards the boundaries in the south (UNESCO 1969; Bouchez et al. 2019). It is known as the Upper Aquifer of the Chad Formation in Nigeria (Barber 1965; Eberschweiler 1993a, b) and is an important groundwater resource for Cameroon, Chad, and Niger, whereas in Nigeria it is used mainly for local rural water supply (Miller et al. 1968), due to low thickness and discontinuity (Barber 1965).



**Fig. 1** Location of the study area. **a** Map of the Lake Chad Basin (LCB) with the extension of the Quaternary aquifer (yellow-shaded area); **b** the LCB in the sub-Saharan northern part of Africa

The Quaternary aquifer is underlain by the upper Pliocene (Fig. 2), an aquitard composed of up to 300 m of greenish clay, and the lower Pliocene, represented by fine to coarse sand deposits, which often cannot be differentiated from the underlying Continental Terminal (CT) (Schneider and Wolff 1992). The lower Pliocene (Fig. 2), known as the Middle Aquifer of the Chad Formation in Nigeria (Barber 1965), is encountered at depths between 250 and 300 m and is artesian in the surroundings of the Lake Chad. It provides groundwater to Nigerian cities in the basin, like Maiduguri.

The Oligo-Miocene CT (Fig. 2), composed of fluvio-lacustrine deposits, is characterised by an alternation of sandstone and clay banks (Lang et al. 1990; Eberschweiler 1993a) of variable thicknesses that can reach up to 600 m, but on average is 100 m (Eberschweiler 1993b). This is the lowest aquifer that lies directly over the granitic basement complex (Arad and Kafri 1975) or in discordance on

Cretaceous formations (Massuel 2001). It is known as the Lower Aquifer of the Chad Formation in Nigeria (Barber 1965) and its groundwater quality is very poor due to mineralisation (Eberschweiler 1993b). Where the Cretaceous formations are present (Cameroon and Niger), they lay on the crystalline basin; however, it is not certain that they extend throughout the whole basin (Massuel 2001).

### Recharge

Presently, the Quaternary aquifer in the LCB receives recharge through seepage from wetlands (Tewolde et al. 2019) and perennial rivers (Bouchez et al. 2019), as well as along the Lake Chad shore (Isiorho et al. 1996); however, because of river scarcity in the LCB, the largest volume of recharge results from diffuse rainfall. Estimations vary considerably depending on the study area and the method used (Table 1). Values between 2 mm/annum have been





**Table 1** Summary of publications with estimated recharge in parts of the LCB

Region	Recharge (mm/annum)	Method	Publication
<i>Focused recharge</i>			
SW lakeshore (Nigeria)	745	Seepage meter	Isiorho et al. 1996
Sahelian LCB close to rivers	78±7	<sup>36</sup> Cl, Cl	Bouchez et al. 2019
Salamat wetland	111	Chloride mass balance	Tewelde et al. 2019
Waza Logone wetland	117–163	Chloride mass balance	Tewelde et al. 2019
<i>Diffuse recharge</i>			
SE Niger	2	Isotopes ( $\delta^{18}\text{O}$ , $\delta^2\text{H}$ , $^3\text{H}$ , and $^{14}\text{C}$ )	Leduc et al. 2000
Manga Grassland, NE Nigeria	49	Model	Carter et al. 1994
	44	Chloride mass balance	Edmunds et al. 1999
	132	Piezometric data	Goes and Zabudum 1999
	60	Chloride mass balance	Edmunds et al. 1999
NE Nigeria	16–30	Chloride mass balance	Edmunds et al. 2002
	43	Chloride mass balance and Cl in dug-wells	Edmunds et al. 2002
	19	Model of chloride profiles	Goni et al. 2005
	169 for Maiduguri to 837 for Kano	Soil moisture deficit	Ndubuisi 2007
N Nigeria			
N Cameroon	25–125	Aquifer hydrodynamic characteristics	Ngounou Ngatcha et al. 2007
Salamat region, S Chad	3–19	Chloride mass balance	Tewelde et al. 2019
Southern LCB	240±170	<sup>36</sup> Cl, Cl	Bouchez et al. 2019
Sahelian part of the LCB unconnected to rivers	16±27	<sup>36</sup> Cl	Bouchez et al. 2019

to precipitate and thus is not representative in groundwater chemistry.

Fluoride is present in the area (Fantong et al. 2010; Bura et al. 2018), which is of geogenic origin and not widespread. Nitrate pollution as a result of leaking pit-latrines and a general lack of hygiene has been reported from numerous cities within the basin (Ngounou Ngatcha and Daïra 2010; Goni et al. 2019; Bon et al. 2020). It is also a problem in agricultural regions, due to overuse of fertilizers and in pastoral areas because of mismanagement of water points (Vassolo and Daïra 2012; Huneau 2017).

## Materials and methods

Groundwater chemistry combined with environmental isotopes ( $\delta^{18}\text{O}$  and  $\delta^2\text{H}$ ),  $^3\text{H}$ , and  $^{14}\text{C}$  are used to determine the groundwater recharge of the Quaternary aquifer in the LCB. Furthermore, mapping the distribution of groundwater quality and isotopes allows for depicting areas that presently receive recharge as well as areas that do not participate in the actual water cycle.

Laboratory results from regular groundwater sampling throughout the basin over the last 12 years are available. In total, 1,267 full chemical analyses, 1,203 environmental isotopes analyses, 142 tritium analyses, and seven carbon-14 analyses exist at the LCBC from a series of cooperation

projects with the German Federal Institute for Geosciences and Natural Resources (BGR). Additionally, environmental isotopes in 386 event-based precipitation samples, collected at six stations scattered in the basin, as well as 110 samples for the Chari River in N'Djamena over the period July 2016–September 2021, are available. Because all analyses were performed in the same laboratories, relative comparability of the results is very high. Ion balance was less than 10% for 1,128 (97.4%) samples and less than 5% for 1,128 (89%) samples.

Coordinates of the sample points were recorded during sampling campaigns by means of a Garmin™ GPS device. Temperature, electrical conductivity (EC), and pH were measured on-site using a digital multi-parameter instrument (WTW™-Multi 3430 and HACH HQ40d with IntelliCAL™ electrodes). Groundwater was extracted either by pumping with the installed pump, a submersible pump (Grundfoss™MP1), or a bailer. Sampling from chemical analyses consisted in filling a 100-ml PVC bottle with filtered and HNO<sub>3</sub>-acidified water for cation analysis. Addition of HNO<sub>3</sub> lowers the pH value to ~2, avoiding precipitation, bacterial growth, adsorption on the bottle's wall, etc. Furthermore, a 500-ml PVC bottle was filled with unfiltered and nonacidified water for anion analysis. Samples were collected after rinsing the bottles three times with water from the sampling site and kept in a cool box during the field campaign. For isotope analyses, 30-ml amber glass bottles

**Table 2** Summary of rainfall collectors installed in the LCB

Location	Longitude (°)	Latitude (°)	Elevation (m asl)	Study period	No. of years (analyses)	Range of values		LMWL monthly weighted		Weighted average all data		
						$\delta^{18}\text{O}$ (‰)	$\delta^2\text{H}$ (‰)	Slope (‰)	Intercept (‰)	$\delta^{18}\text{O}$ (‰)	$\delta^2\text{H}$ (‰)	d-excess (‰)
N'Djamena	15.0013	12.12819	295	2016–2021	6 (184)	–9.9 to 3.8	–74.5 to 39.0	7.1	7.3	–5.0	–27.0	10
Mao	15.31217	14.13196	343	2018–2021	4 (80)	–10.4 to 7.3	–74.1 to 23.8	6.2	4.2	–4.8	–28.6	10
Bol	14.73753	13.44474	294	2018–2021	4 (68 <sup>a</sup> )	–11.0 to –0.6	–73.4 to 6.7	6.4	3.1	–5.9	–36.6	11
Abéché	20.83969	13.84453	536	2018–2021	4 (72 <sup>a</sup> )	–7.7 to 1.9	–46.6 to 18.4	6.3	2.7	–3.9	–21.5	10
Maiduguri	13.21856	11.86296	318	2018–2021	4 (40 <sup>b</sup> )	–10.4 to 3.3	–74.7 to 27.4	7.2	8.5	–5.1	–28.4	13
Kano	8.58167	11.96583	490	2018–2019	2 (9 <sup>b</sup> )	–6.9 to 2.7	–42.1 to 5.4	8.0	+13.1	–4.8	–25.4	13

Elevation was estimated from Shuttle Radar Topography Mission (SRTM) in m asl (meters above sea level), LMWL local meteoric water line, Coordinates in WGS 84 system

<sup>a</sup>2021 samples not yet analysed

<sup>b</sup>2019 to 2021 samples not yet analysed

with PE-seal caps filled with untreated and unfiltered water were used. In the case of tritium and  $^{14}\text{C}$ , 1,000-ml PE bottles were filled with unfiltered and untreated water. Samples were stored at 6 °C until the time of packaging, and were then airfreighted to the laboratory in Hanover (Germany) in cool boxes packed with cooling elements.

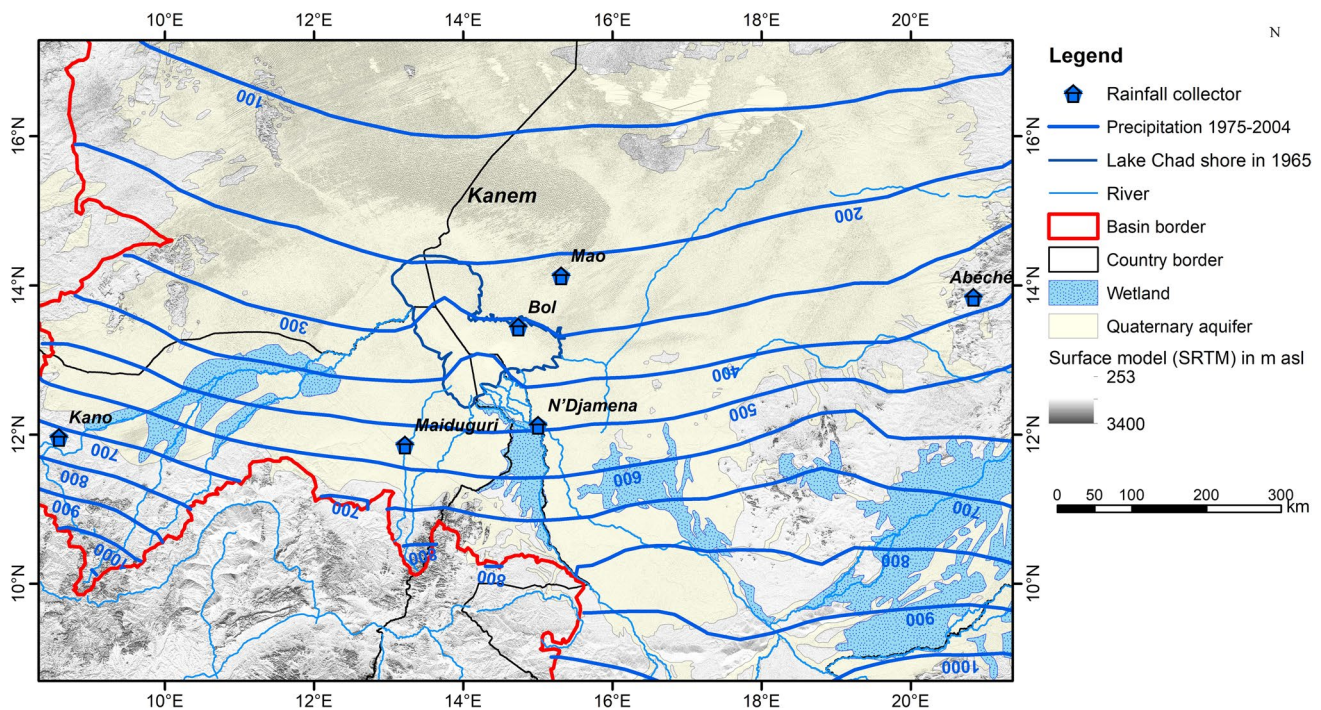
Concentrations of major and minor ions were measured using a DIONEX ICS-3000 ion chromatograph (Cl, Br, F,  $\text{NO}_3$ ,  $\text{NO}_2$ ,  $\text{SO}_4$ ), a SPECTRO ARCOS ICP-OES instrument (Na, K, Ca, Mg, P, Fe, Mn, Al, Si), and a UNICAM UV 300 photometer ( $\text{NH}_4$ ,  $\text{NO}_2$ ,  $\text{PO}_4$ ). Commercial standard solutions are used for daily calibration and limits of quantitation are determined by 10-point calibration according to the DIN 32645 standard.  $\text{HCO}_3$  and  $\text{CO}_3$  concentrations were determined with a SCHOTT Titroline Alpha Plus automatic titration system according to a modified DIN 38409-H7-2 standard by evaluation of titration curve shapes for the identification and quantitation of these species. The contribution of  $\text{PO}_4$  to alkalinity was estimated from ICP-OES concentration data.

All stable isotopes ( $^{18}\text{O}$  and  $^2\text{H}$ ) values were determined using a PICARRO L2130-i and L2140-i cavity ring-down laser spectrometers. Samples were measured at least four times and the reported value is the mean one. All values are given in delta notation in per mill (‰) vs. Vienna Standard Mean Ocean Water (VSMOW). Raw data were checked for organic contamination using the software ChemCorrect (unaltered software settings), then corrected for memory effects and drift. Furthermore, they were normalized to the VSMOW/SLAP scale. External reproducibility, defined as standard deviation of a quality check standard during all runs, was better than 0.20 and 0.8‰ for  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$ , respectively.

Tritium was measured at the Bremen University, Germany, by mass spectrometry of its decay product Helium-3. After degassing, each sample was stored for ~6 months in glass bulbs. Next, the glass bulbs were connected to the sample inlet system and unsealed under high-vacuum conditions. Accumulated Helium-3 was separated from water and analysed for Helium-3 and Helium-4 with a sector field mass spectrometer (MAP 215-50). Helium-4 serves to quantify sample contamination by air. The system is calibrated by defined amounts of atmospheric air and is able to detect 0.01 TU or 0.6 mBq/L in water.

Carbon-14 was measured by Beta Analytics Inc. (USA) by counting the number of  $^{14}\text{C}$  atoms present in the sample through accelerator mass spectrometry (AMS).

Deuterium-excess—d-excess (‰) =  $\delta^2\text{H}$  (‰) –  $8 \times \delta^{18}\text{O}$  (‰)—is used to determine evaporation effects in the recharge processes. It depends on evaporation conditions at the ocean and varies with cloud transport to the precipitation site (Bershaw 2018). According to the Global Network of Isotopes in Precipitation (GNIP), its global average value



**Fig. 3** Location of the rainfall collectors in the LCB and distribution of the long-term mean annual precipitation for the period 1975–2004 using data from the Climate Research Unit (CRU)

under oceanic evaporation conditions is 10‰ (Florea et al. 2017), when air humidity is above 85% (Fritz and Fontes 1980; Clark and Fritz 1997). Under evaporative conditions, the residual water is enriched in both  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$ , but, because the relative enrichment of  $\delta^{18}\text{O}$  is larger than for  $\delta^2\text{H}$ , d-excess results are lower than 10‰.

## Results and discussion

Groundwater recharge areas can be delineated using environmental isotopes and groundwater quality data. Furthermore, it is possible to describe recharge processes using the isotopic characteristics of groundwater.

### Isotopic composition

There are two groundwater recharge sources in the LCB: precipitation and surface water (rivers, lakes, and wetlands). These sources have dissimilar isotopic compositions and, thus, mark groundwater differently.

Concerning precipitation, the project has installed six rain collectors in the basin at N'Djamena, Mao, Bol, and Abéché in Chad as well as at Maiduguri and Kano in Nigeria (Table 2; Fig. 3). Isotope values for the Kano station are only available for the rainy season in 2018. In the case of Maiduguri, the study attained samples for four rainy

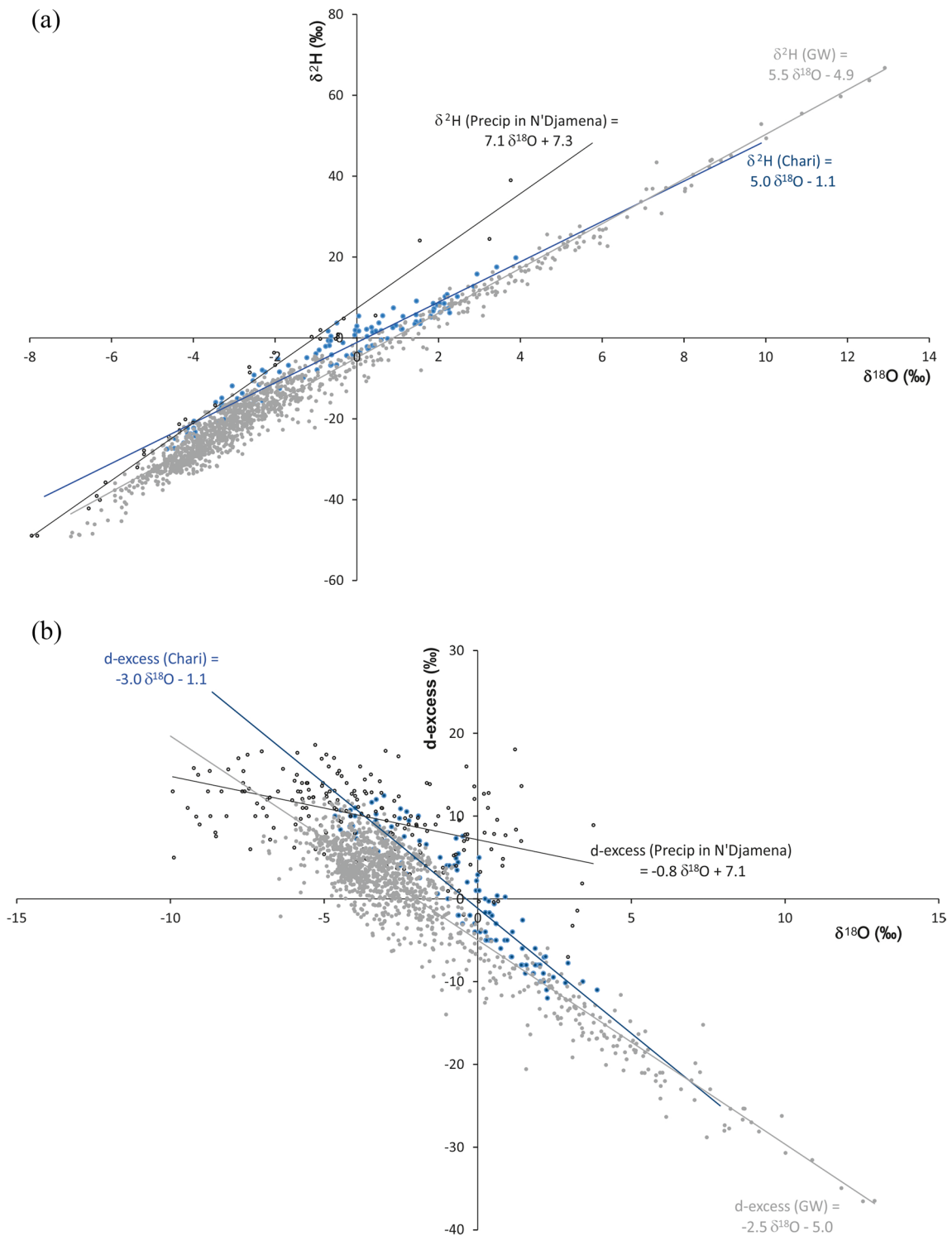
seasons (2018–2021), but analyses are available only for 2018. These two stations were not considered for further analyses. Analyses for Bol and Abéché do not include samples from 2021.

According to Table 2, groundwater presently recharged by precipitation should be close to the weighted average isotope values of precipitation with  $\delta^{18}\text{O}$  values between  $-3.9\text{‰}$  and  $-5.9\text{‰}$  and  $\delta^2\text{H}$  values in the range of  $-21.5$  to  $-36.6\text{‰}$ . Precipitation at Bol presents the most depleted concentrations for both isotopes, probably due to its proximity to the Lake Chad. The open surface water is able to supply depleted isotope values and, thus, change the composition of precipitation locally.

Since 2016, the most depleted values of  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  in the Chari River at N'Djamena are  $-4.6$  and  $-27.5\text{‰}$ , respectively, which correspond to the mean weighted average of precipitation in N'Djamena ( $-5\text{‰}$ ,  $-27\text{‰}$ ). This means that during high water, the flow originates from rainfall. The most  $\delta^2\text{H}$ -enriched values are  $3.9\text{‰}$  for  $\delta^{18}\text{O}$  and  $19.8\text{‰}$  for  $\delta^2\text{H}$ . The water line for the Chari River (Fig. 4a) results in  $\delta^2\text{H} = 5.0 \times \delta^{18}\text{O} - 1.1$ . Mean values for  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  are  $-0.62$  and  $-4.2\text{‰}$ , respectively. In N'Djamena, the Chari water line intercepts the local meteoric water line at  $\delta^{18}\text{O} = -4.0\text{‰}$  and  $\delta^2\text{H} = -21.1\text{‰}$ .

A regression line for groundwater is defined by  $\delta^2\text{H} = 5.5 \delta^{18}\text{O} - 4.9$  (Fig. 4a). Mean values for  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$  are  $-2.2$  and  $-17.2\text{‰}$ , respectively.  $\delta^2\text{H}$  varies between  $-7.0$  and





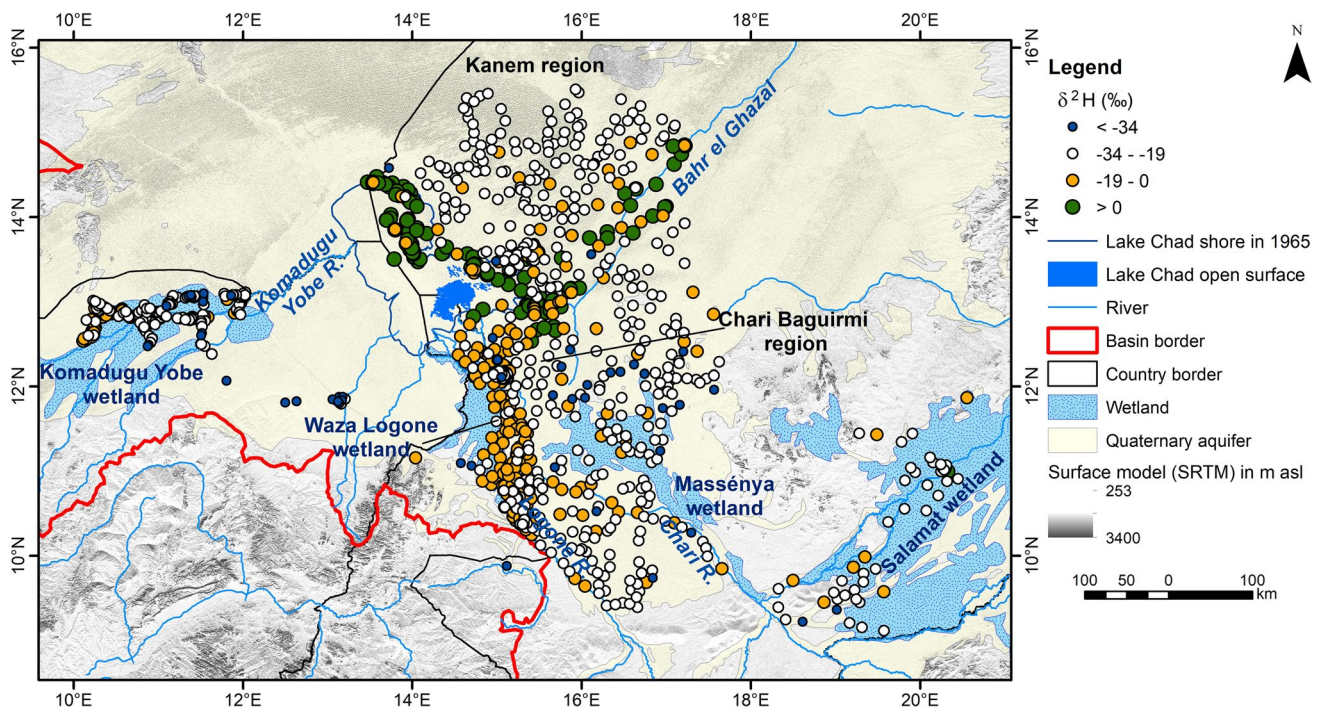
**Fig. 4** Meteoric water lines for precipitation in N'Djamena (black) and the Chari River in N'Djamena (blue), as well as groundwater (GW) in the LCB (grey): **a**  $\delta^{18}\text{O}$  vs.  $\delta^2\text{H}$  space and **b** d-excess vs.  $\delta^{18}\text{O}$

12.9‰, and  $\delta^2\text{H}$  between  $-49.1$  and  $66.8$ ‰. In the graph of Deuterium excess (d-excess) vs.  $\delta^{18}\text{O}$  (Fig. 4b), the regression lines for river water and groundwater are steeper compared to the local precipitation line. This is an indication

of evaporation effects experienced by both river water and groundwater.

Mahamat Nour (2019) reported that the Quaternary aquifer discharges into the Chari and Logone rivers during the





**Fig. 5** Distribution of  $\delta^2\text{H}$  (‰) in the LCB

dry season and exfiltrating groundwater had a mean isotopic composition of  $-1.5$  and  $-7.2$ ‰ for  $\delta^{18}\text{O}$  and  $\delta^2\text{H}$ , respectively for the years 2013, 2015, and 2016. Here, the  $\delta$ -values are more positive compared to the mean weighted precipitation values (Table 2), indicating evaporation effects during recharge.

Mapping the distribution of  $\delta^2\text{H}$  (‰) in the LCB allows for depicting the areas in which recharge takes place. Diffuse recharge from recent precipitation (white dots) occurs in the Kanem region (Fig. 5). The orange dots in this region indicate more positive  $\delta$ -values and thus, evaporation in the recharge process. Although precipitation in the area is very low, it appears that recharge can ensue because of the concentration of rainwater along the interdunal valleys. Present recharge also occurs in the Chari Baguirmi region in central Chad, where precipitation amounts are large enough to allow infiltration

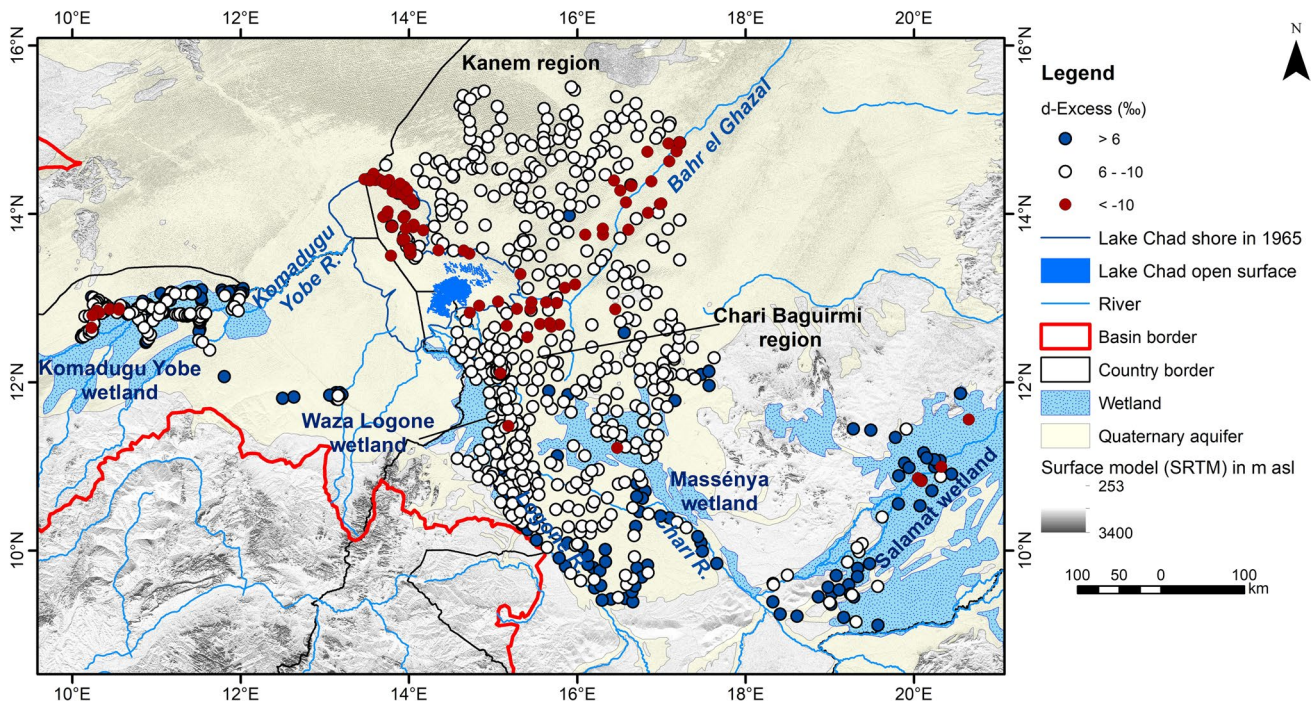
and percolation. Groundwater in the surroundings of the wetlands Salamat in south Chad and Komadugu Yobe at the border between Nigeria and Niger also show present recharge. In this case, it is either focused recharge during high water levels, when rivers are mainly fed by rainfall, or diffuse recharge through ponding water in the wetlands. The map (Fig. 5) indicates present groundwater recharge in the Waza Logone area with more positive  $\delta$ -values (orange dots), which is the result of recharge with surface water that has experienced some evaporation. Very positive  $\delta$ -values are encountered in the northern part and along the eastern shore of Lake Chad as well as along the lake’s shore and the Bahr el Ghazal (dark green dots). In these areas, recharge is caused by surface water that heavily evaporates before percolating.

Deuterium-excess values for precipitation in N’Djamena vary from  $-7$ ‰ (2017) to  $19$ ‰ (2018) with a value of  $10$ ‰ for both median and average (Table 3), which indicates an oceanic origin of precipitation that does not experience evaporation when transported across the continent. This confirms findings by Goni et al. (2021). They report that recharge in the LCB takes place in the middle of the rainy season, when precipitation events are strong and air humidity above  $85\%$  and thus, with d-excess values of approx.  $10$ ‰. Deuterium-excess values for the Chari River, the other possible source of recharge, are in the range of  $-12$  to  $12$ ‰ with a median of  $2$ ‰ and an average of  $1$ ‰, which clearly indicates evaporation effects (Table 3). In

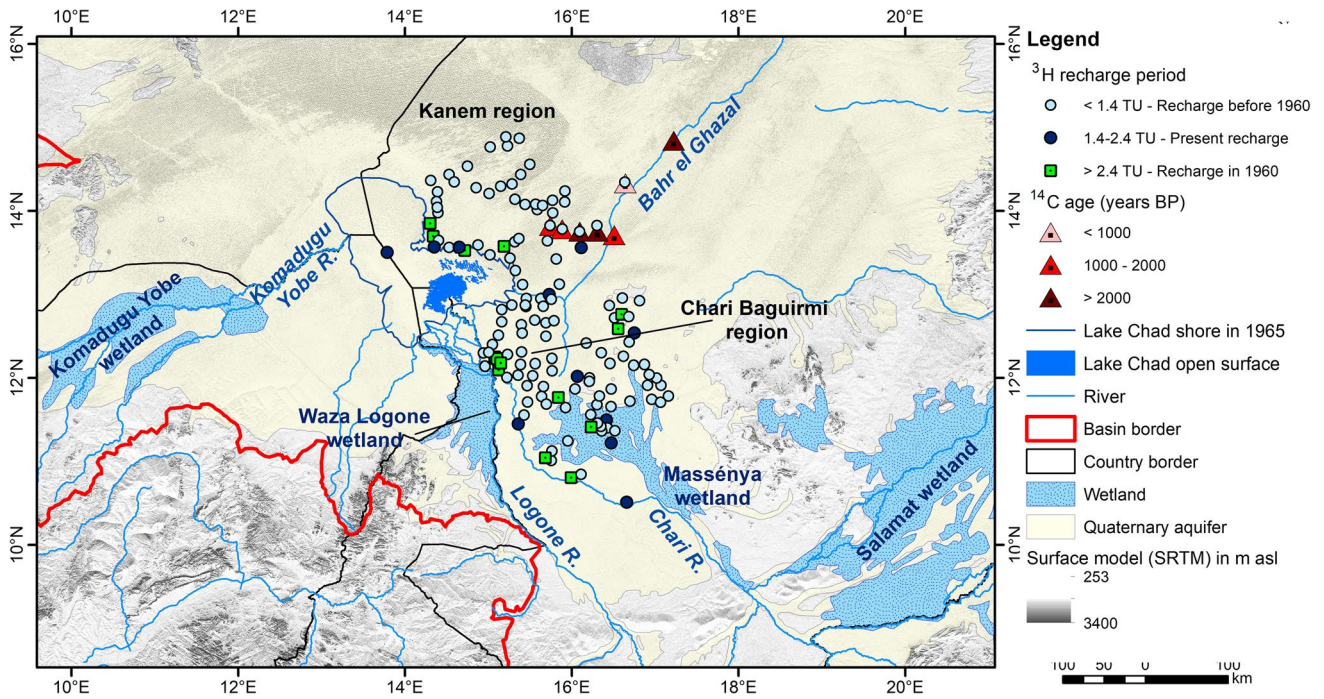
**Table 3** Statistics for d-excess measured in precipitation, Chari River and groundwater in the LCB

Statistic	Precipitation in N’Djamena	Chari River in N’Djamena	Groundwater in LCB
Min (‰)	-7	-12	-37
Max (‰)	19	12	12
Median (‰)	10	2	1
Mean (‰)	10	2	3





**Fig. 6** Distribution of d-excess (‰) in the LCB



**Fig. 7** Distribution of groundwater age in the LCB based on <sup>14</sup>C and <sup>3</sup>H results

groundwater, d-excess varies from -37 to 12‰ with a median of 1‰ and an average of 3‰. Because the lowest d-excess values in groundwater are much lower than

in precipitation or Chari River, it can be concluded that ponds or wetlands, where large evaporation is possible, also produce recharge.

**Table 4** Summary of recharge mechanisms in the LCB

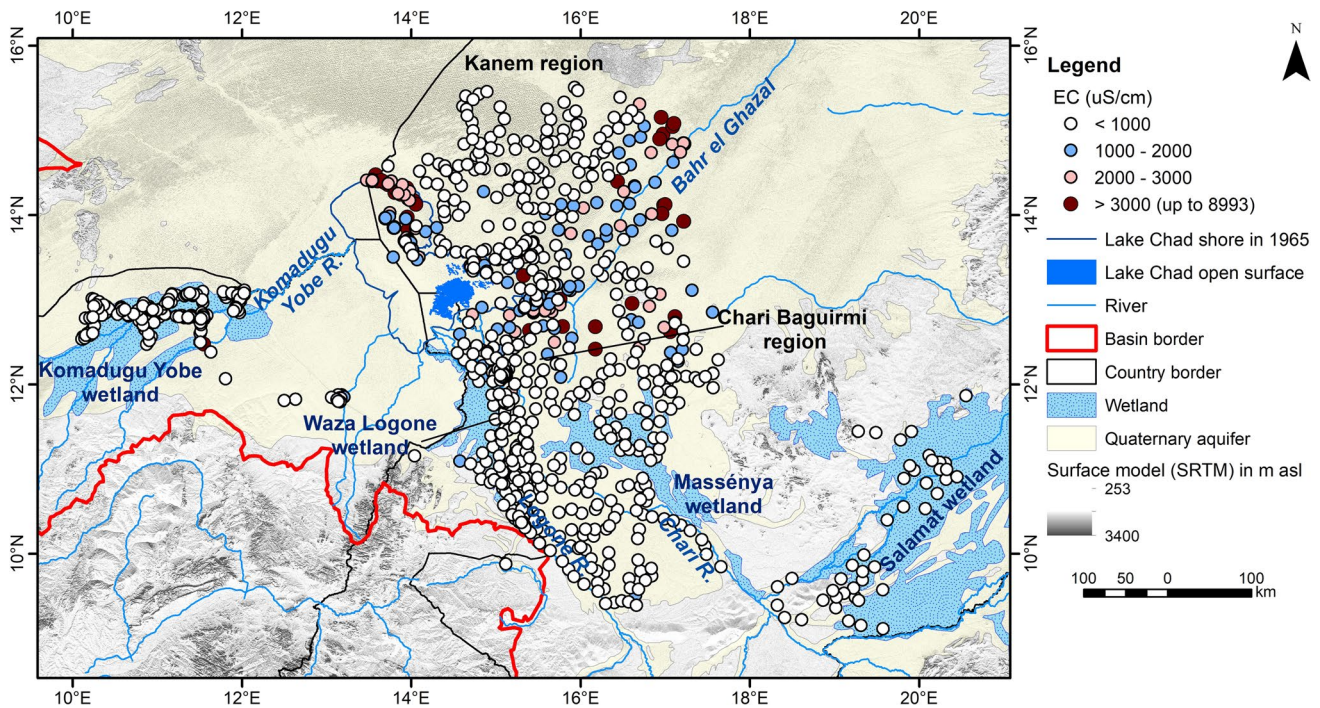
Region	Recharge type	Source of recharge	Recharge mechanism	Groundwater age
Kanem	Diffuse	Precipitation	Infiltration of rainfall water along interdunal valleys with evaporation	Recharged before 1960 or mixed (very old and present)
Chari Baguirmi	Focused	Lake Chad	Lateral inflow of heavily evaporated lake water along the southern rim	No dating
	Diffuse	Precipitation	Direct rainfall infiltration with evaporation	Recharged before 1960 or mixed (very old and present)
Komadugu Yobe wetland	Focused	Komadugu	Lateral inflow of river water during high water without evaporation	Recharged before 1960 or mixed (very old and present)
	Diffuse	Wetland	Infiltration of evaporated ponding water (heavily evaporated in the west)	No dating
Salamat wetland	Focused	River	Lateral inflow of river water during high water without evaporation	Recharged before 1960 or mixed (very old and present)
	Diffuse	Wetland	Infiltration of evaporated ponding water (heavily evaporated along the borders)	No dating
Waza Logone wetland	Diffuse	Wetland	Infiltration of evaporated ponding water	Recharged before 1960 or mixed (very old and present)
	Focused	Logone and Chari	Lateral inflow from river water during high water without evaporation	Recharged before 1960 or mixed (very old and present)
Southern Quaternary rim (in Chad)	Diffuse	Precipitation	Direct rainfall infiltration without evaporation	Recharged before 1960 or mixed (very old and present)
	Focused	Logone and Chari	Lateral inflow from river water during high water without evaporation	Recharged before 1960 or mixed (very old and present)
Northern part of Lake Chad	Focused	Lake water	Infiltration of heavily evaporated lake water	Recharged before 1960 or mixed (very old and present)
Bahr el Ghazal	Focused	River	Infiltration of heavily evaporated river water	Very old groundwater (500–4,150 years BP)

The d-excess distribution for the LCB (Fig. 6) shows direct recharge without evaporation at the southern rim of the Quaternary aquifer (blue dots) meaning that direct recharge from precipitation without or only little evaporation effects takes place here. Values of d-excess close to 10‰ appear also along the rivers; thus, they only recharge the Quaternary aquifer during high water levels, when river water is composed mainly of precipitation water. Groundwater in the northern pool of Lake Chad, as well as along the lake's shore and the Bahr el Ghazal, show very negative d-excess values (red dots), which confirms the fact that here recharge is produced by heavily evaporated water.

Tritium measurements from 142 samples in Kanem and the Chari Baguirmi region as well as seven  $^{14}\text{C}$  analyses were available to evaluate the time of recharge (Fig. 7).

Most of the samples show very low tritium values ( $<1.5$  TU, light blue dots) demonstrating that recharge occurred mainly before the bomb peak in the 1960s. However, samples along the Lake Chad shore show recharge from 55 years ago ( $^3\text{H} > 2.5$  TU, green squares) produced by lateral inflow from the lake at its “normal” state and confirm finding from Mahamat Nour et al. (2022). Lake Chad in the 1960s has been defined as “normal” (Lemoalle et al. 2012) and is characterised by water levels between 281 and 282 m above mean sea level (amsl), an extension of  $\sim 20,000$  km<sup>2</sup>, numerous dune islands, and occasional marshy vegetation on the shores (Lemoalle et al. 2012). According to tritium results, the present recharge by precipitation (values between 1.5 and 2.5 TU, dark blue dots) takes place in some locations scattered throughout the LCB (Fig. 6). Carbon-14 values along the Bahr el Ghazal estimate





**Fig. 8** Distribution of electrical conductivity (EC) in groundwater in the LCB

groundwater ages between 600 and 4,150 years (Fig. 7). These values confirm findings from Bouchez et al. (2019), who reported premodern recharge for this area.

In summary, water isotopes allow for defining recharge mechanisms within the LCB. Results are presented in Table 4.

### Chemical composition

The chemical composition of groundwater also reflects the recharge areas in a semiarid climate. Newly recharged groundwater is characterised by low values of EC, because precipitation shows generally low values of EC. In N'Djamena, 12 values of EC in precipitation have been recorded for May to July 2020. Values vary from 16 to 48  $\mu\text{S}/\text{cm}$  with an average of 27  $\mu\text{S}/\text{cm}$ .

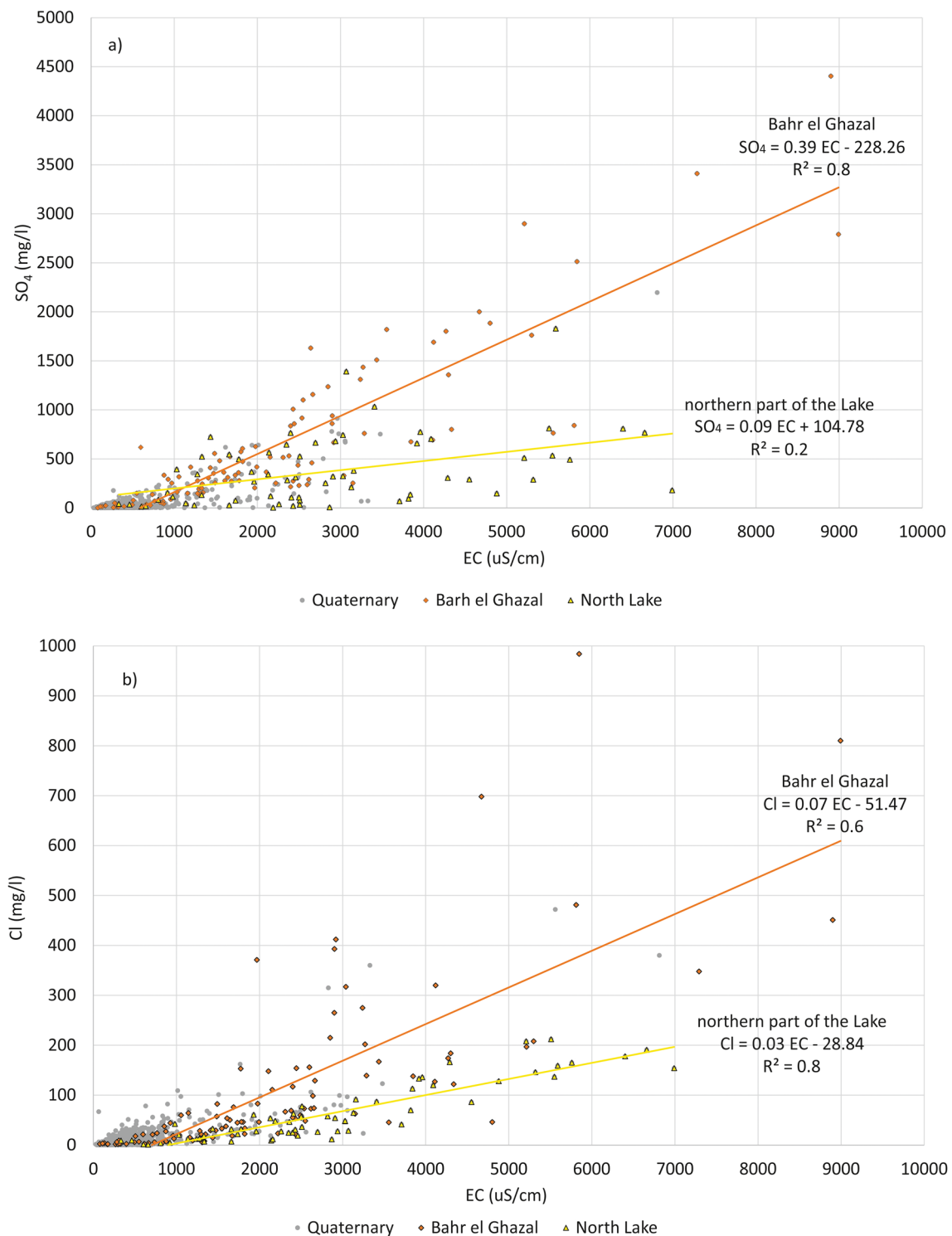
Electrical conductivity was measured in 1,263 groundwater samples in the LCB. Values vary largely from 31 to 8,993  $\mu\text{S}/\text{cm}$  with an average of 689  $\mu\text{S}/\text{cm}$  (Fig. 8). Altogether, 1,046 (83%) of the values show an EC value below 1,000  $\mu\text{S}/\text{cm}$  and thus reflect the general low groundwater mineralization of the Quaternary aquifer. High values of EC (>2,000  $\mu\text{S}/\text{cm}$ ) occur in the northern part and southern rim of Lake Chad, along the Bahr el Ghazal, and in the northern part of the Chari Baguirmi.

Sulphate concentration in 1,267 groundwater samples throughout the LCB vary from below the detection limit to

4,403 mg/L with an average of 98.5 mg/L. A large number of samples (1,197 samples, 94.4 %) have values below the WHO (2022) guideline of 500 mg/L. Samples with sulphate concentrations above the WHO limit also show high EC values.

It is possible to distinguish two different trends of EC– $\text{SO}_4$ . The increasing trend of low slope corresponds to the northern part of Lake Chad, where large values of EC do not correlate with sulphate concentration (Fig. 9a) but are the result of high chloride concentrations (Fig. 9b) due to evaporation. In this area, recharge is caused by lake water that experiences large evaporation before percolation. The steeper trend (Fig. 9a) belongs to the boreholes along the Bahr el Ghazal, where EC correlates with high sulphate concentrations that occur in the area. Eugster and Maglione (1979) postulated that sulphate could originate from oxidation of sulphides or organic matter trapped in clays deposited by the Mega Lake Chad (10,000–2,000 years BP). Seven groundwater samples along the Bahr el Ghazal area were dated by means of  $^{14}\text{C}$ . Calculated ages vary between 600 and 4,150 years with an average age of 2,176 years confirming that groundwater in the Bahr el Ghazal was recharged at the time of the Mega Lake Chad occurrence. This area does not receive recharge presently and is less affected by evaporation, which is depicted by a lower correlation between EC and Cl (Fig. 9b).





**Fig. 9** Relationship between electrical conductivity (EC) and **a** sulphate and **b** chloride, for groundwater in the LCB

### Conclusions

The combination of isotope and chemical analyses of groundwater allows one to delineate areas of recharge

in the LCB and to determine the age of the groundwater. Actual diffuse recharge occurs in the Kanem dunes. Although precipitation in the area is very low, recharge does occur because of the concentration of rainwater along

the interdunal valleys. The Chari Baguirmi area in central Chad, the wetlands of Salamat in south Chad, and Komadugu Yobe at the border between Nigeria and Niger also experience recent diffuse recharge from precipitation. However, in these areas focused recharge during floods prevails, when the rivers transport mainly rainfall. Tewelde et al. (2019) estimated diffuse mean annual groundwater recharge in the range of 3–19 mm and focussed recharge of 111 mm for Salamat wetlands.

The Waza Logone area is characterised by slightly enriched groundwater, which is the result of recharge with surface water that has experienced some evaporation. Here recharge has been estimated between 117 and 163 mm/annum (Tewelde et al. 2019). Heavily  $^2\text{H}$ - and  $^{18}\text{O}$ -enriched groundwater is found in the northern part of Lake Chad as well as along the lake's shore, where recharge is caused by lake water that severely evaporates before percolating. However, present recharge seems to be very low. Tewelde et al. (2019) estimated annual recharge between 0.6 and 0.8 mm/annum.

Groundwater along the Bahr el Ghazal is  $^2\text{H}$  and  $^{18}\text{O}$ -depleted with large negative values of  $d$ -excess, which points to a recharge with evaporated surface water. Groundwater also presents large contents of sulphate probably due to oxidation of sulphides or from organic matter trapped in clays deposited by the Mega Lake Chad (10,000–2,000 years BP). Measurement of  $^{14}\text{C}$  translates into groundwater ages varying between 600 and 4,150 years, which is defined as premodern by Bouchez et al. (2019).

In general, groundwater of the Quaternary aquifer has low electrical conductivity (EC), which is an indication of recently recharged water. Values above 2,000  $\mu\text{S}/\text{cm}$  are only present in the northern part and the southern rim of Lake Chad, the northern part of Chari Baguirmi, and along Bahr el Ghazal. The high EC values in the northern part and southern rim of Lake Chad are evaporation-induced and correlate with chloride. However, the elevated values along the Bahr el Ghazal correlate with  $\text{SO}_4$ , which could originate from oxidation of sulphides or organic matter trapped in clays deposited by the Mega Lake Chad.

### Recommendations to water resource managers

Although recharge occurs in most of the LCB, protection of the wetlands is urgently needed. The largest amount of recharge takes place in these wetland areas and they should be managed to maintain quantity and quality. It is highly recommended to avoid installation of dams or other engineering structures that could lead to their desiccation. If this occurs, not only recharge would reduce substantially in the basin, but also biodiversity would reduce

significantly. Furthermore, actions should be taken to avoid pollution, e.g. misuse of fertilisers or lack of effluent treatment facilities in towns and villages.

Use of groundwater along the Bahr el Ghazal could be of concern because of the high content of salts, especially sulphate. It is recommended not to install any wells for water supply before ensuring the good quality of groundwater.

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### Declarations

**Conflict of interest** The authors declare that they have no conflict of interest.

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