



Electrochemistry at Krakowian research institutions

Katarzyna Skibińska¹ · Piotr Żabiński¹

Received: 7 December 2022 / Revised: 8 January 2023 / Accepted: 9 January 2023 / Published online: 17 January 2023
© The Author(s) 2023

Abstract

The electrochemistry research team activity from Poland is marked by significant increase in the last 20 years. The joining of European Community in 2004 gives an impulse for the development of Polish science. The development of electrochemistry has been stimulated by cooperation with industry and the establishment of technology transfer centers, technology parks, business incubators, etc. and the mostly by simplified international collaborations. Five research institutions from Krakow reports work in the field of electrochemistry. The achievements of all teams are briefly described.

Introduction

The political transformation in 1989 strongly influences the development of Polish science, through its connection with industry and the establishment of technology transfer centers, technology parks, business incubators, etc. [1]. Then, joining the European Union by Poland in 2004 significantly simplified international collaborations. These days, there are many opportunities for Polish scientists to receive funds for projects in cooperation with foreign centers. Researchers can apply for a scientific stay abroad. It is possible thanks to the Polish Science Centre (NCN), the National Centre for Research and Development (NCBR), and the Polish National Agency for Academic Exchange (NAWA). Nowadays, there are almost 380 public and private universities in Poland divided into academic and vocational institutions [2]. Additionally, in 2010, due to the reform of science, research and development units were transformed into research institutes. Sources from 2015 reported that there are 115 research institutes, 58% of which work in the field of science and engineering and 34% in life sciences [3]. There are 70 research institutes of the Polish Academy of Sciences (PAN). Political transformation connected with opening to collaboration with foreign research groups and injection of funds to build or buy advanced research instrumentation results in blooming of research activity in a field of electrochemistry.

Only in Krakow, five research institutions report work in the field of electrochemistry. The achievements of all teams, over the last 20 years, are briefly described below. The research area of mentioned laboratory groups is strongly connected with the field of electrochemistry. The units are divided by the main research centers.

Electrochemistry at Krakowian research facilities

AGH university of science and technology

The AGH University of Science and Technology (the AGH UST) is a public university strongly involved in many national and international research projects. It is composed of sixteen faculties and one research institute. The research in the field of electrochemistry is performed at six of them. The biggest group works at the Department of Physicochemistry and Metallurgy of Non-Ferrous Metals at the Faculty of Non-Ferrous Metals. In the first years of the 2000s, the research teams guided by professors Lidia Burzyńska, Wanda Gumowska, and Krzysztof Fitzner performed intensive experiments in the field of, e.g., anodic dissolution [4–7], metal recovery [8, 9], and electrodeposition of transition group composites [10–12] and metal alloys [13, 14]. Then, with the growth of the research group, many new directions of interest have appeared, such as catalytic properties of transition group metals and alloys [15–19], tellurium thin films [20, 21], and synthesis of nanostructures in the anodic alumina oxide templates [22–24]. The influence of the magnetic field on the properties of the electrodeposited coatings has been investigated [25–27]. Nowadays, 3 groups

✉ Piotr Żabiński
zabinski@agh.edu.pl

¹ Faculty of Non-Ferrous Metals, AGH University of Science and Technology, A. Mickiewicza 30, 30-059 Kraków, Poland

guided by professors Ewa Rudnik, Remigiusz Kowalik, and Piotr Żabiński can be distinguished. Professor Żabiński is also a head of the department. Under their supervision, young scientists and Ph.D. students are working on the co-deposition of chalcogenides with noble [28] and transition metals [29, 30]. The synthesis of noble and transition metal alloys [31, 32] has been still developing. The synthesis of Zn–Ni [33], Zn–Mg [34], and Zn–Ni–Mg [35] from gluconate baths has been also investigated. Besides that, the metallization of 3D prints [36–38] and electrochemical synthesis of 1D nanostructures by the one-step method [39–42] have been studied. The superhydrophobic properties of coatings are also in interest of scientists [43]. Research groups have been collaborating with international groups from Germany [44–47], Japan [48–52], France [53], Algeria [54], Bulgaria [55, 56], Italy [57], Serbia [58, 59], and Turkey [32]. In 2015, one of the members of the Department, Professor Krzysztof Mech, moved to the Academic Centre for Materials and Nanotechnology to perform his own research. In his experiments, he mostly focuses on the electrodeposition of Ni coatings [60] and composites [61]. He took under the consideration also the influence of the magnetic field on, inter alia, the deposition rate, composition, current efficiency, structure, surface states, and morphology of the synthesized materials [62–64].

At the Faculty of Metals Engineering and Industrial Computer Science, Professor Tomasz Moskalewicz has been the head of the project entitled “Development of electrophoretic co-deposition of bioactive and antibacterial ceramics with biodegradable polymers to produce novel composite coatings for biomedical applications.” The project has been funded by NCN within the Beethoven competition. The research is focused on the development and characterization of novel composite coatings exhibiting bioactive and antibacterial functionalities for orthopedic and dental applications employing electrophoretic deposition (EPD) [65]. The experiments are performed in the collaboration with Professor Aldo R. Boccaccini from the Institute of Biomaterials at the University of Erlangen-Nuremberg.

Several research groups work at the Faculty of Materials Science and Ceramics in three different Departments. Professor Tomasz Brylewski, from the Department of Physical Chemistry and Modelling, is the co-author of the reviews about the hydrogen market, research, and development progress in central and Eastern European countries [66, 67]. With other researchers, he also performed experiments strongly connected with the solid oxide fuel cells (SOFC) operating in the higher temperatures [68–70]. The influence of applied solid electrolyte on the properties of obtained composites was investigated [71]. With the groups from German and Italian institutes, the first prototypes of dual membrane cell were electrochemically tested [72]. Another person, at the Department

of Physical Chemistry and Modelling, performing some experiments connected with the field of the electrochemistry is Professor Robert Filipek. With other co-authors, he modeled corrosion of steel [73] and electrodiffusion processes for ion-selective electrodes [74]. Generally, in his works, he focuses mostly on simulations of diffusion in various materials at higher temperatures [75–77]. At the Department of Analytical Chemistry and Biochemistry, the research teams were guided and formed by professors, inter alia, Bogusław Józef Baś, Władysław Wojciech Kubiak, Andrzej Lewenstam, Jan Migdalski, and Mieczysław Rękas. They work, e.g., on the design and calibration of measuring equipment for electrochemistry and applications of electrochemical sensors [78–81], including conducting polymers [82, 83] and ion-selective membrane electrodes [84]. The investigation of voltammetric methods and their improvement was performed [85, 86]. The development of new signal processing algorithms for analytical chemistry research methodologies [87–91] is also an important issue for scientists. The topic of SOFC [92] is explored as well. There are three research groups at the Department of Inorganic Chemistry related to electrochemistry. Scientists perform research in synthesizing of ceramic, metallic, and composite layers, as well as materials based on intermetallic phases [93, 94]. They determine the physical and chemical properties of the coatings, including the resistance to high-temperature oxidation [95, 96] and corrosion in electrolyte solutions [97]. They study the kinetics and mechanism of diffusion [98, 99]. Additionally, the research related to the development of materials such as compounds with the perovskite structure [100, 101], composite, including carbon–metal oxide systems [102], thermoelectric materials [103], and materials with catalytic properties [101], is performed. Also, the topics related to the materials, including transition metal oxides in the form of micro- and nanomaterials and thin layers, for the various applications, such as anodes for photoelectrochemical cells (PEC) [104–106], and semiconductor gas sensors [107, 108] are explored.

Researchers from the Faculty of Energy and Fuels at the AGH UST focus on the currently developed issues of Li-ion [109, 110] and Na-ion batteries [111, 112], SOFC [113, 114], oxygen storage [115, 116], etc. Professor Janina Molenda is the head of the Department of Hydrogen at this faculty. Besides the mentioned topics expanded at the faculty, her research interests relate to, among other things, catalytic properties of perovskite oxides [117, 118] and structural defects in solid electrolytes [119, 120]. Professor Konrad Świerczek, the Deputy Dean for Science at the Faculty of Energy and Fuels, also pays attention to oxide materials, their characterization, and measurements of their catalytic activity [121–123]. He works with ceramic membranes possessing mixed ionic–electronic

conductivity [124]. He has collaborated with scientists from, inter alia, China [125, 126], Japan [127], and the USA [128].

The last unit at the AGH UST where the research connected with the electrochemistry is performed is the Faculty of Foundry Engineering represented by the group from the Department of Metal Chemistry and Corrosion governed now by Professor Maria Starowicz. The research is carried out by great specialists in the field of corrosion on the micro- [129–131] and nanoscale [132], as well as in Krakow water supply network [133]. Professor Halina Krawiec was awarded inter alia by DCNS-French Embassy-French Institute for her scientific achievements in the field of corrosion. Professor Urszula Lelek-Borkowska focuses on, inter alia, the electrochemistry of metals and semiconductors in alcohol solutions [134]. The scientists at the department focus on synthesis of nanomaterials [135–137] and electrochemical behavior of metals [138, 139] and semiconductors [140, 141] in aqueous and organic electrolyte solutions. The research group also works on modern directions such as green chemistry and biomaterials [142, 143]. The team has collaborated with the group from, inter alia, France [144–146], Germany [147], UK [148], Austria [149], and Spain [150].

Faculty of chemistry jagiellonian university

Jagiellonian University is the oldest university in Poland and one of the oldest in Europe [151]. It consists of 16 faculties. The team specialized in Electrochemistry, previously guided by Professor Marian Jaskuła, and now by Professor Grzegorz Sulka, and has been working at the Faculty of Chemistry. The group has impressive achievements in fields of anodization of Al [152–155]; electrochemical synthesis of nanostructured oxides of, e.g., Sn [156, 157] and Zn [158, 159]; and nanoporous titanium dioxide [160–162]. The obtained anodic alumina oxide (AAO) templates are widely used for the fabrication of nanostructured electrodes of metals [163], alloys [164, 165], and polymers [166, 167]. Produced nanostructures and oxides can be applied in catalysis [168], photoelectrochemistry, and photocatalysis [169–174], as H₂O₂ sensors [175–178], semiconductors [158, 179], and in biomedical applications [180–184]. The research group worked also on the synthesis of Au [185, 186] and Au–Ag [187] for electrochemical epinephrine sensing. Furthermore, the team works on bioelectrodes [188] and materials for energy storage [189]. The scientists have attended many projects funded by NCN, NAWA, the Polish Ministry of Science and Higher Education (MNiSW), and the European Cooperation in Science and Technology (COST). They have collaborated with groups from, e.g., the Czech Republic [190], the USA [191], Portugal [192], and Ukraine [193].

Jerzy haber institute of catalysis and surface chemistry of the polish academy of sciences

The research aim for scientists from the Jerzy Haber Institute of Catalysis and Surface Chemistry is the application of obtained results in fields of protection of health, environment, and cultural heritage, as well as to improve technological processes [194]. One of the scientists working in the field of electrochemistry was Professor Alicja Drelinkiewicz. With other researchers, she performed experiments, inter alia, on the synthesis and characterization of mono- and bimetallic-supported catalysts [195–197]. Professor Paweł Nowak and his research group have worked on the synthesis of metal alloys and metal/ceramic composites by electrodeposition [198–201]. They have investigated the prevention of corrosion of metal and its alloys by using anti-corrosion coatings [202]. The research group is also a specialist in surface doping of titanium dioxide with transition metals and transition metal ions and their photocatalytic properties [203–206]. The group tests also materials for SOFC [207–209]. In 2000, the American Electroplaters and Surface Finishers Society awarded Professor Nowak for an outstanding work published in 1999 in the *Plating and Surface Finishing* magazine published by this society. He has collaborated with research groups from, inter alia, Germany [210, 211], the USA [212–214], Finland [215], and Switzerland [216].

Institute of metallurgy and materials science of the polish academy of sciences

In the Institute of Metallurgy and Materials, two research groups, under the supervision of Professors Ewa Bełtowska-Lehman and Piotr Ozga, can be recognized. Professor Bełtowska-Lehman is a specialist in the kinetics and mechanism of the processes occurring in complex electrolytes during electrocrystallization of metal, alloy, and composite coatings with increased tribological and anti-corrosion properties [217, 218]. The group performed research in electrochemical synthesis and micromechanical properties of nanocomposite coatings with a metal matrix containing the addition of a refractory metal, reinforced with a nanodisperse ceramic phase [219–221]. Professor Bełtowska-Lehman works also on photovoltaic silicon cells [222, 223]. In 2006, she was nominated by the MNiSW for the position of Polish representative in the Mirror Group of the European Union Technology Platform in the field of photovoltaics. Professor Ozga works on, inter alia, the electrochemical deposition of materials for different applications, such as lead-free solders (e.g., Sn–Ag and Sn–Ag–Cu [224], In–Sn [225], Sn–Zn–Cu [226]), and solder nanocomposites, alloys, and protective layers of high corrosion resistance, e.g., Zn–Ni [227, 228], are investigated by Professor and his team. He collaborates with researchers from Israel [229] and Ukraine [230].

Łukasiewicz research network — Krakow institute of technology

Krakow Institute of Technology is a part of the Łukasiewicz Research Network, the third largest research network in Europe [231]. At the Institute of Advanced Manufacturing Technology in Krakow, Doctor Maria Zybura-Skrabalak was working on electrochemical machining (ECM). The experiments were performed within the INNOTECH project. The aim of the project was the implementation of the proposed process. The investigation of the influence, inter alia, of the features of ECM processes on the surface roughness of the workpiece material was performed [232]. The modeling of the process considering the application of the flat rectangular universal electrode in the ECM [233] or current in the ECDM (electrochemical discharge machining) process [234] was made among others by Doctor Zybura-Skrabalak. In her work, she also investigated the possible application of the electrode, made by the milling process, for electrochemical smoothing [235].

Conclusions

Krakow is the headquarter of many research groups working in the field of electrochemistry. There are a few common topics for all, e.g., electrodeposition of metal, its alloys, and composites; the phenomenon of the corrosion and enhancement of anti-corrosion properties of coatings; and catalytic activity of the samples. The new directions, such as photovoltaics, biomaterials, and green chemistry, are in accordance with the interests of scientists all over the world. The teams collaborate with research facilities from various countries but also with each other. It results in high-quality results, new projects, and implementations.

Declarations

Competing interests The authors declare no competing interests.

Open Access This article is licensed under a Creative Commons Attribution 4.0 International License, which permits use, sharing, adaptation, distribution and reproduction in any medium or format, as long as you give appropriate credit to the original author(s) and the source, provide a link to the Creative Commons licence, and indicate if changes were made. The images or other third party material in this article are included in the article's Creative Commons licence, unless indicated otherwise in a credit line to the material. If material is not included in the article's Creative Commons licence and your intended use is not permitted by statutory regulation or exceeds the permitted use, you will need to obtain permission directly from the copyright holder. To view a copy of this licence, visit <http://creativecommons.org/licenses/by/4.0/>.

References

1. Historia nauki polskiej. https://pl.wikipedia.org/wiki/Historia_nauki_polskiej
2. Higher Education Institutions. <https://study.gov.pl/higher-education-institutions>. Accessed 12 Oct 2022
3. NIK o instytutach badawczych. <https://www.nik.gov.pl/aktualnosci/nik-o-instytutach-badawczych.html>
4. Burzyńska L, Zabiński P (2000) Analysis of the composition and morphology of products of the anodic dissolution of Cu₂S. Arch Metall 45:286–302
5. Burzyńska L (2001) Comparison of the spontaneous and anodic processes during dissolution of brass. Corros Sci 43:1053–1069. [https://doi.org/10.1016/S0010-938X\(00\)00130-X](https://doi.org/10.1016/S0010-938X(00)00130-X)
6. Burzyńska L, Gumowska W, Rudnik E (2004) Influence of the composition of Cu–Co–Fe alloys on their dissolution in ammoniacal solutions. Hydrometallurgy 71:447–455. <https://doi.org/10.1016/j.hydromet.2003.08.007>
7. Burzyńska L, Gumowska W, Rudnik E, Partyka J (2008) Mechanism of the anodic dissolution of Cu₇₀–Co₄–Fe₁₄–Pb₇ alloy originated from reduced copper converter slag in an ammoniacal solution. Hydrometallurgy 92:34–41. <https://doi.org/10.1016/j.hydromet.2008.01.009>
8. Burzyńska L, Rudnik E, Gumowska W (2004) The influence of phase structure on the dissolution of Cu–Co–Fe alloys in sulphuric acid solution and the metals recovery. Hydrometallurgy 71:457–463. [https://doi.org/10.1016/S0304-386X\(03\)00084-7](https://doi.org/10.1016/S0304-386X(03)00084-7)
9. Burzyńska L, Rudnik E, Barteczko P (2006) Recovery of copper and cobalt from low copper Cu–Co–Fe alloy. Arch Metall Mater 51:299–308
10. Burzyńska L, Rudnik E, Błaz L et al (2003) The influence of current density and bath composition on the electrodeposition of nickel and nickel/silicon carbide composite. Trans IMF 81:193–198. <https://doi.org/10.1080/00202967.2003.11871539>
11. Burzyńska L, Rudnik E, Koza J et al (2008) Electrodeposition and heat treatment of nickel/silicon carbide composites. Surf Coatings Technol 202:2545–2556. <https://doi.org/10.1016/j.surfcoat.2007.09.020>
12. Rudnik E, Burzyńska L, Jędruch J, Błaz L (2009) Codeposition of SiC particles with electrolytic cobalt in the presence of Cs⁺ ions. Appl Surf Sci 255:7164–7171. <https://doi.org/10.1016/j.apsusc.2009.03.053>
13. Burzyńska L, Rudnik E (2000) The influence of electrolysis parameters on the composition and morphology of Co–Ni alloys. Hydrometallurgy 54:133–149. [https://doi.org/10.1016/S0304-386X\(99\)00060-2](https://doi.org/10.1016/S0304-386X(99)00060-2)
14. Burzyńska L, Rudnik E (2000) The influence of saccharin and sodium lauryl sulfate on the electrodeposition process of Co–Ni alloys. Arch Metall 45:276–285
15. Zabinski PR, Jarek A, Kowalik R (2009) Effect of applied external magnetic field on electrodeposition of cobalt alloys for hydrogen evolution in 8MNaOH. Magnetohydrodynamics 45:275–280. <https://doi.org/10.22364/mhd.45.2.18>
16. Zabinski P, Górski M, Kowalik R (2009) Influence of superimposed external magnetic field onto electrodeposition of co–P alloys for hydrogen evolution. Arch Metall Mater 54:1157–1166
17. Kowalik R, Żabiński P, Fitzner K (2008) Electrodeposition of ZnSe. Electrochim Acta 53:6184–6190. <https://doi.org/10.1016/j.electacta.2007.12.009>
18. Zabinski P, Franczak A, Kowalik R (2012) Electrodeposition of functional Ni–Re alloys for hydrogen evolution. ECS Trans 41:39–48. <https://doi.org/10.1149/1.3702411>

19. Zabiński PR, Kowalik R, Piwowarczyk M (2007) Cobalt-tungsten alloys for hydrogen evolution in hot 8 M NaOH. *Arch Metall Mater* 52:627–634
20. Kowalik R, Kutyla D, Mech K, Żabiński P (2016) Analysis of tellurium thin films electrodeposition from acidic citric bath. *Appl Surf Sci* 388:817–824. <https://doi.org/10.1016/j.apsusc.2016.03.127>
21. Kowalik R, Kutyla D, Mech K et al (2015) Electrowinning of tellurium from acidic solutions. *Arch Metall Mater* 60:591–596. <https://doi.org/10.1515/amm-2015-0178>
22. Dobosz I, Gumowska W, Czapkiewicz M (2013) Magnetic properties of Co-Fe nanowires electrodeposited in pores of alumina membrane. *Arch Metall Mater* 58:663–671. <https://doi.org/10.2478/amm-2013-0052>
23. Gumowska W, Dobosz I, Wrzosczyk B (2014) The morphology of the alumina films formed in the anodization process of aluminium in the orthophosphoric acid solution. The Co-Fe alloys electrodeposition into obtained alumina pores. *Arch Metall Mater* 59:137–143. <https://doi.org/10.2478/amm-2014-0022>
24. Dobosz I, Kutyla D, Kac M et al (2020) The influence of homogeneous external magnetic field on morphology and magnetic properties of CoRu nanowire arrays. *Mater Sci Eng B Solid-State Mater Adv Technol* 262. <https://doi.org/10.1016/j.mseb.2020.114795>
25. Electrocatalytic properties of alloys deposited with superimposed magnetic field (2012) *Magneto hydrodynamics* 48:243–250. <https://doi.org/10.22364/mhd.48.2.1>
26. Zabiński PR, Franczak A, Kowalik R (2012) Electrocatalytically active Ni-Re binary alloys electrodeposited with superimposed magnetic field. *Arch Metall Mater* 57:495–501. <https://doi.org/10.2478/v10172-012-0051-2>
27. Magnetic field effect on properties of galvanostatically deposited Co-Pd alloys (2014) *Magneto hydrodynamics* 50:75–82. <https://doi.org/10.22364/mhd.50.1.8>
28. Wojtyśiak M, Jędraczk A, Stepien M et al (2021) Electrodeposition of Pd–Se thin films. *Electrochem Commun* 127:107053. <https://doi.org/10.1016/j.elecom.2021.107053>
29. Kwiecińska AM, Kutyla D, Kołczyk-Siedlecka K et al (2019) Electrochemical analysis of co-deposition cobalt and selenium. *J Electroanal Chem* 848:113278. <https://doi.org/10.1016/j.jelechem.2019.113278>
30. Kowalik R, Szaciłowski K, Żabiński P (2012) Photoelectrochemical study of ZnSe electrodeposition on Cu electrode. *J Electroanal Chem* 674:108–112. <https://doi.org/10.1016/j.jelechem.2012.03.002>
31. Kutyla D, Kołczyk-Siedlecka K, Kwiecińska A et al (2019) Preparation and characterization of electrodeposited Ni-Ru alloys: morphological and catalytic study. *J Solid State Electrochem* 23:3089–3097. <https://doi.org/10.1007/s10008-019-04374-7>
32. Kutyla D, Salci A, Kwiecińska A et al (2020) Catalytic activity of electrodeposited ternary Co–Ni–Rh thin films for water splitting process. *Int J Hydrogen Energy* 45:34805–34817. <https://doi.org/10.1016/j.ijhydene.2020.05.196>
33. Chat-Wilk K, Rudnik E, Włoch G, Osuch P (2022) Codeposition of zinc with nickel from gluconate solutions. *J Solid State Electrochem* 26:1715–1731. <https://doi.org/10.1007/s10008-022-05205-y>
34. Chat-Wilk K, Rudnik E, Włoch G, Osuch P (2022) Codeposition of zinc with manganese from different gluconate baths. *Russ J Electrochem* 58:168–183. <https://doi.org/10.1134/S1023193522030053>
35. Chat-Wilk K, Rudnik E, Włoch G (2022) Effect of chloride and sulfate ions on electrodeposition and surface properties of alloys produced from zinc-nickel-manganese gluconate baths. *J Electrochem Soc* 169:092515. <https://doi.org/10.1149/1945-7111/ac8eec>
36. Żabiński R, Skibiński K-SK, Kutyl K, Kwieciński D, Kowali A (2019) Influence of magnetic field on electroless metallization of 3D prints by copper and nickel. *Arch Metall Mater* 64:17–22. <https://doi.org/10.24425/amm.2019.126212>
37. Żabiński R, Zborowski KK, Kutyl W, Kwieciński D, Kowali A (2018) Investigation of two-step metallization process of plastic 3D prints fabricated by SLA method. *Arch Metall Mater* 63:1031–1036. <https://doi.org/10.24425/122438>
38. Kołczyk-Siedlecka K, Kutyla D, Skibińska K et al (2021) Well-ordered 3D printed Cu/Pd- decorated catalysts for the methanol electrooxidation in alkaline solutions. *Technologies* 9:6. <https://doi.org/10.3390/technologies9010006>
39. Skibińska K, Wojtaszek K, Krause L et al (2023) Tuning up catalytic properties of electrochemically prepared nanoconical Co-Ni deposit for HER and OER. *Appl Surf Sci* 607:155004. <https://doi.org/10.1016/j.apsusc.2022.155004>
40. Skibińska K, Kornaus K, Yang X et al (2022) One-step synthesis of the hydrophobic conical Co-Fe structures – the comparison of their active areas and electrocatalytic properties. *Electrochim Acta* 140127. <https://doi.org/10.1016/j.electacta.2022.140127>
41. Skibińska K, Kołczyk-Siedlecka K, Kutyla D et al (2021) Electrocatalytic properties of Co nanoconical structured electrodes produced by a one-step or two-step method. *Catalysts* 11:544. <https://doi.org/10.3390/catal11050544>
42. Skibińska K, Semeniuk S, Kutyla D et al (2021) Study on synthesis and modification of conical Ni structures by one-step method. *Electrochim Acta* 66:861–869. <https://doi.org/10.24425/amm.2021.136391>
43. Rudnik E, Chat K (2018) A brief review on bio-inspired superhydrophobic electrodeposited nickel coatings. *Trans IMF* 96:185–192. <https://doi.org/10.1080/00202967.2018.1475931>
44. Skibińska K, Huang M, Mutschke G et al (2022) On the electrodeposition of conically nano-structured nickel layers assisted by a capping agent. *J Electroanal Chem* 904:115935. <https://doi.org/10.1016/j.jelechem.2021.115935>
45. Huang M, Eckert K, Mutschke G (2021) Magnetic-field-assisted electrodeposition of metal to obtain conically structured ferromagnetic layers. *Electrochim Acta* 365:137374. <https://doi.org/10.1016/j.electacta.2020.137374>
46. Skibińska K, Kutyla D, Yang X et al (2022) Rhodium-decorated nanoconical nickel electrode synthesis and characterization as an electrochemical active cathodic material for hydrogen production. *Appl Surf Sci* 592:153326. <https://doi.org/10.1016/j.apsusc.2022.153326>
47. Koza JA, Mühlenhoff S, Żabiński P et al (2011) Hydrogen evolution under the influence of a magnetic field. *Electrochim Acta* 56:2665–2675. <https://doi.org/10.1016/j.electacta.2010.12.031>
48. Zabinski PR, Meguro S, Asami K, Hashimoto K (2006) Electrodeposited Co-Ni-Fe-C alloys for hydrogen evolution in a hot 8 kmol·m⁻³ NaOH. *Mater Trans* 47:2860–2866. <https://doi.org/10.2320/matertrans.47.2860>
49. Zabinski PR, Meguro S, Asami K, Hashimoto K (2003) Electrodeposited Co-Fe and Co-Fe-C alloys for hydrogen evolution in a Hot 8 kmol m⁻³ NaOH solution. *Mater Trans* 44:2350–2355. <https://doi.org/10.2320/matertrans.44.2350>
50. Zabinski PR, Nemoto H, Meguro S et al (2003) Electrodeposited Co-Mo-C cathodes for hydrogen evolution in a hot concentrated NaOH solution. *J Electrochem Soc* 150:C717. <https://doi.org/10.1149/1.1604788>
51. Zabinski PR, Meguro S, Asami K, Hashimoto K (2004) Electrodeposited Co-Fe and Co-Fe-C alloys for hydrogen evolution in a Hot 8 kmol m⁻³ NaOH solution. *J Japan Inst Met* 68:456–461. <https://doi.org/10.2320/jinstmet.68.456>
52. Hashimoto K, Kumagai N, Izumiya K et al (2013) The use of renewable energy in the form of methane via electrolytic hydrogen generation / Zastosowanie Odnawialnej Energii W Formie Metanu Na Drodze Elektrolitycznej Produkcji Wodoru. *Arch Metall Mater* 58:231–239. <https://doi.org/10.2478/v10172-012-0179-0>
53. Aaboubi O, Douglade J, Zabinski P, Chopart J (2012) Magnetic field effect on molybdenum-based alloys 48:271–278
54. Chouchane S, Levesque A, Zabinski P et al (2010) Electrochemical corrosion behavior in NaCl medium of zinc–nickel alloys

- electrodeposited under applied magnetic field. *J Alloys Compd* 506:575–580. <https://doi.org/10.1016/j.jallcom.2010.07.099>
55. Dobrovolska T, Krastev I, Zabiński P et al (2011) Oscillations and self-organization phenomena during electrodeposition of silver-indium alloys. Experimental Study. *Arch Metall Mater* 56: <https://doi.org/10.2478/v10172-011-0070-4>
56. Dobrovolska T, Kowalik R, Zabinski P, Krastev I (2008) Investigations of the surface morphology of electrodeposited Ag-In coatings by means of optical, scanning-electron and atomic-force microscopy. *Bulg Chem Commun* 40:254–260
57. Kolczyk-Siedlecka K, Bernasconi R, Lucotti A et al (2020) Self-assembled monolayers assisted all wet metallization of SU-8 negative tone photoresist. *J Electrochem Soc* 167:142506. <https://doi.org/10.1149/1945-7111/abc844>
58. Elezovic N, Zabinski P, Krstajic-Pajic M et al (2018) Electrochemical deposition and characterization of AgPd alloy layers. *J Serbian Chem Soc* 83:593–609. <https://doi.org/10.2298/JSC171103011E>
59. Lačnjevac U, Vasilčić R, Tokarski T et al (2018) Deposition of Pd nanoparticles on the walls of cathodically hydrogenated TiO₂ nanotube arrays via galvanic displacement: a novel route to produce exceptionally active and durable composite electrocatalysts for cost-effective hydrogen evolution. *Nano Energy* 47:527–538. <https://doi.org/10.1016/j.nanoen.2018.03.040>
60. Mech K (2017) Influence of organic ligands on electrodeposition and surface properties of nickel films. *Surf Coatings Technol* 315:232–239. <https://doi.org/10.1016/j.surfcoat.2017.02.042>
61. Mech K (2019) Electrodeposition of composite Ni-TiO₂ coatings from aqueous acetate baths. *Metall Mater Trans A* 50:4275–4287. <https://doi.org/10.1007/s11661-019-05325-7>
62. Mech K (2019) A novel magnetoelectrochemical method of synthesis of photoactive Ni-TiO₂ coatings from glycinate electrolytes. *Mater Des* 182:108055. <https://doi.org/10.1016/j.matdes.2019.108055>
63. Mech K, Gajewska M, Marzec M, Szaciłowski K (2020) On the influence of magnetic field on electrodeposition of Ni-TiO₂ composites from a citrate baths. *Mater Chem Phys* 255:123550. <https://doi.org/10.1016/j.matchemphys.2020.123550>
64. Mech K (2020) MHD supported electroreduction of formate nickel complexes with simultaneous incorporation of TiO₂ particles. *Arch Metall Mater* 65:219–227. <https://doi.org/10.24425/amm.2019.131118>
65. Moskalewicz T, Warcaba M, Łukaszczyk A et al (2022) Electrophoretic deposition, microstructure and properties of multicomponent sodium alginate-based coatings incorporated with graphite oxide and hydroxyapatite on titanium biomaterial substrates. *Appl Surf Sci* 575:151688. <https://doi.org/10.1016/j.apsusc.2021.151688>
66. Iordache I, Bouzek K, Paidar M et al (2019) The hydrogen context and vulnerabilities in the central and Eastern European countries. *Int J Hydrogen Energy* 44:19036–19054. <https://doi.org/10.1016/j.ijhydene.2018.08.128>
67. Stygar M, Brylewski T (2017) Hydrogen-based energy market in Poland. In: *Hydrogen in an International Context: Vulnerabilities of Hydrogen Energy in Emerging Markets*. River Publishers, pp 105–126
68. Brylewski T, Przybylski K (2008) Perovskite and spinel functional coatings for SOFC metallic interconnects. *Mater Sci Forum* 595–598:813–822. <https://doi.org/10.4028/www.scientific.net/MSF.595-598.813>
69. Kruk A, Stygar M, Brylewski T (2013) Mn-Co spinel protective-conductive coating on AL453 ferritic stainless steel for IT-SOFC interconnect applications. *J Solid State Electrochem* 17:993–1003. <https://doi.org/10.1007/s10008-012-1952-8>
70. Bednarz M, Molin S, Bobruk M, Stygar M, Długoń E, Sitarz M, Brylewski T (2019) High-temperature oxidation of the Crofer 22 H ferritic steel with Mn_{1.45}Co_{1.45}Fe_{0.104} and Mn_{1.5}Co_{1.504} spinel coatings under thermal cycling conditions and its properties. *Mater Chem Phys* 225:227–238. <https://doi.org/10.1016/j.matchemphys.2018.12.090>
71. Lubszczyk M, Wyrwa J, Wojteczko K et al (2021) Electrical and mechanical properties of ZrO₂-Y₂O₃-Al₂O₃ composite solid electrolytes. *J Electron Mater* 50:5933–5945. <https://doi.org/10.1007/s11664-021-09125-x>
72. Thorel AS, Abreu J, Ansar SA et al (2013) Proof of concept for the dual membrane cell. *J Electrochem Soc* 160:F360–F366. <https://doi.org/10.1149/2.051304jes>
73. Filipek R, Szyszkiewicz-Warzechka K, Szczudło J (2020) Corrosion of steel in concrete - modeling of electrochemical potential measurement in 3D geometry. *Arch Metall Mater* 65:117–124. <https://doi.org/10.24425/amm.2019.131104>
74. Szyszkiewicz K, Jasielc JJ, Danielewski M et al (2017) Modeling of electrodiffusion processes from nano to macro scale. *J Electrochem Soc* 164:E3559–E3568. <https://doi.org/10.1149/2.051711jes>
75. Jasielc J, Stec J, Szyszkiewicz-Warzechka K et al (2020) Effective and apparent diffusion coefficients of chloride ions and chloride binding kinetics parameters in mortars: non-stationary diffusion-reaction model and the inverse problem. *Materials (Basel)* 13:5522. <https://doi.org/10.3390/ma13235522>
76. Filipek R, Szyszkiewicz K, Dziembaj P et al (2012) Modeling of reactive diffusion: mechanism and kinetics of the intermetallics growth in Ag/Ag interconnections. *J Mater Eng Perform* 21:638–647. <https://doi.org/10.1007/s11665-012-0131-5>
77. Danielewski M, Filipek R (2000) Interdiffusion in oxide solid solutions, simulation of the process and calculation of intrinsic diffusivities. *Mol Cryst Liq Cryst Sci Technol Sect A Mol Cryst Liq Cryst* 341:277–282. <https://doi.org/10.1080/10587250008026153>
78. Kupis J, Mattinen U, Kisiel A et al (2012) Biomimetic membranes made of conducting polymers doped with adenosine diphosphate (ADP). *Electrochim Acta* 77:23–28. <https://doi.org/10.1016/j.electacta.2012.04.058>
79. Madej M, Fendrych K, Porada R et al (2021) Application of Fe(III)-exchanged clinoptilolite/graphite nanocomposite for electrochemical sensing of amitriptyline. *Microchem J* 160:105648. <https://doi.org/10.1016/j.microc.2020.105648>
80. Porada R, Fendrych K, Baś B (2021) Development of novel Mn-zeolite/graphite modified screen-printed carbon electrode for ultrasensitive and selective determination of folic acid. *Measurement* 179:109450. <https://doi.org/10.1016/j.measurement.2021.109450>
81. Blaz T, Migdalski J, Lewenstam A (2005) Junction-less reference electrode for potentiometric measurements obtained by buffering pH in a conducting polymer matrix. *Analyst* 130:637. <https://doi.org/10.1039/b418384c>
82. Kupis-Rozmysłowicz J, Wągner M, Bobacka J et al (2016) Biomimetic membranes based on molecularly imprinted conducting polymers as a sensing element for determination of taurine. *Electrochim Acta* 188:537–544. <https://doi.org/10.1016/j.electacta.2015.12.007>
83. Migdalski J, Błaż T, Lewenstam A (2014) Conducting polymers - mechanisms of cationic sensitivity and the methods of inducing thereof. *Electrochim Acta* 133:316–324. <https://doi.org/10.1016/j.electacta.2014.03.169>
84. Migdalski J, Lewenstam A (2022) Electrically enhanced sensitivity (EES) of ion-selective membrane electrodes and membrane-based ion sensors. *Membranes (Basel)* 12:763. <https://doi.org/10.3390/membranes12080763>
85. Węgiel K, Grabarczyk M, Kubiak WW, Baś B (2017) A reliable and sensitive voltammetric determination of Mo(VI) at the in situ renovated bismuth bulk annular band electrode. *J Electrochem Soc* 164:H352–H357. <https://doi.org/10.1149/2.1161706jes>
86. Piech R, Kubiak WW (2007) Determination of trace arsenic with DDTC-Na by cathodic stripping voltammetry in presence

- of copper ions. *J Electroanal Chem* 599:59–64. <https://doi.org/10.1016/j.jelechem.2006.09.004>
87. Jakubowska M, Baś B, Ciepela F, Kubiak WW (2010) A calibration strategy for stripping voltammetry of lead on silver electrodes. *Electroanalysis* 22:1757–1764. <https://doi.org/10.1002/elan.200900510>
 88. Górski Ł, Sordoń W, Ciepela F et al (2016) Voltammetric classification of ciders with PLS-DA. *Talanta* 146:231–236. <https://doi.org/10.1016/j.talanta.2015.08.027>
 89. Górski Ł, Jakubowska M, Baś B, Kubiak WW (2012) Application of genetic algorithm for baseline optimization in standard addition voltammetry. *J Electroanal Chem* 684:38–46. <https://doi.org/10.1016/j.jelechem.2012.08.014>
 90. Górski Ł, Kubiak WW, Jakubowska M (2016) Independent components analysis of the overlapping voltammetric signals. *Electroanalysis* 28:1470–1477. <https://doi.org/10.1002/elan.201501089>
 91. Jakubowska M, Piech R, Dzierwa T et al (2003) The evaluation method of smoothing algorithms in voltammetry. *Electroanalysis* 15:1729–1736. <https://doi.org/10.1002/elan.200302751>
 92. Rękas M (2015) Electrolytes for intermediate temperature solid oxide fuel cells. *Arch Metall Mater* 60:891–896. <https://doi.org/10.1515/amm-2015-0225>
 93. Godlewska E, Mars K, Mania R, Zimowski S (2011) Combustion synthesis of Mg₂Si. *Intermetallics* 19:1983–1988. <https://doi.org/10.1016/j.intermet.2011.06.013>
 94. Godlewska E, Mars K, Zawadzka K (2012) Alternative route for the preparation of CoSb₃ and Mg₂Si derivatives. 227–230
 95. Mitoraj M, Godlewska EM (2013) Oxidation of Ti–46Al–8Ta in air at 700 °C and 800 °C under thermal cycling conditions. *Intermetallics* 34:112–121. <https://doi.org/10.1016/j.intermet.2012.10.014>
 96. Mitoraj M, Mars K, Matula M, Godlewska E (2015) Hot corrosion behaviour of Cr-Si coated titanium alloys. *Ann Chim Sci des Matériaux* 39:141–148. <https://doi.org/10.3166/acsm.39.141-148>
 97. Godlewska EM, Mitoraj-Królikowska M, Czerni J et al (2020) Corrosion of Al(Co)CrFeNi high-entropy alloys. *Front Mater* 7. <https://doi.org/10.3389/fmats.2020.566336>
 98. Godlewska EM, Mars K, Drozd P et al (2016) Reaction and diffusion phenomena in Ag-doped Mg₂Si. *J Alloys Compd* 657:755–764. <https://doi.org/10.1016/j.jallcom.2015.10.174>
 99. Godlewska E, Mitoraj M, Morgiel J (2009) Reaction and diffusion phenomena upon oxidation of a (γ+α₂) TiAlNb alloy in air. *Mater High Temp* 26:99–103. <https://doi.org/10.3184/096034009X440263>
 100. Łańcucki Ł, Mizera A, Łącz A et al (2017) Development and chemical stability evaluation of enhanced surface LaFe_{1-x}Ti_xO₃ (LFT) perovskites using polystyrene nanospheres as templating agent. *J Alloys Compd* 727:863–870. <https://doi.org/10.1016/j.jallcom.2017.08.205>
 101. Łańcucki Ł, Lach R, Nieroda P et al (2019) Impact of calcium doping on structure, catalytic and conductive properties of lanthanum strontium iron oxide. *Process Appl Ceram* 13:411–417. <https://doi.org/10.2298/PAC1904411L>
 102. Popardowski A, Pasierb P (2022) Influence of electrolyte and redox active electrode materials properties on working mechanism and performance of manganese oxide-based supercapacitors. *Electrochim Acta* 416:140257. <https://doi.org/10.1016/j.electacta.2022.140257>
 103. Nieroda P, Kolezynski A, Leszczynski J et al (2019) The structural, microstructural and thermoelectric properties of Mg₂Si synthesized by SPS method under excess Mg content conditions. *J Alloys Compd* 775:138–149. <https://doi.org/10.1016/j.jallcom.2018.10.064>
 104. Trenczek-Zajac A, Kusior A, Radecka M (2016) CdS for TiO₂-based heterostructures as photoactive anodes in the photoelectrochemical cells. *Int J Hydrogen Energy* 41:7548–7562. <https://doi.org/10.1016/j.ijhydene.2015.12.219>
 105. Trenczek-Zajac A, Banas J, Radecka M (2016) TiO₂-based photoanodes modified with GO and MoS₂ layered materials. *RSC Adv* 6:102886–102898. <https://doi.org/10.1039/C6RA22979D>
 106. Radecka M, Rekas M, Trenczek-Zajac A, Zakrzewska K (2008) Importance of the band gap energy and flat band potential for application of modified TiO₂ photoanodes in water photolysis. *J Power Sources* 181:46–55. <https://doi.org/10.1016/j.jpowsour.2007.10.082>
 107. Kusior A, Klich-Kafel J, Trenczek-Zajac A et al (2013) TiO₂-SnO₂ nanomaterials for gas sensing and photocatalysis. *J Eur Ceram Soc* 33:2285–2290. <https://doi.org/10.1016/j.jeurceramsoc.2013.01.022>
 108. Zakrzewska K, Radecka M (2007) TiO₂-SnO₂ system for gas sensing—photodegradation of organic contaminants. *Thin Solid Films* 515:8332–8338. <https://doi.org/10.1016/j.tsf.2007.03.019>
 109. Redel K, Kulka A, Plewa A, Molenda J (2019) High-performance Li-rich layered transition metal oxide cathode materials for Li-ion batteries. *J Electrochem Soc* 166:A5333–A5342. <https://doi.org/10.1149/2.0511903jes>
 110. Molenda J (2011) Material problems and prospects of Li-ion batteries for vehicles applications. *Funct Mater Lett* 04:107–112. <https://doi.org/10.1142/S1793604711001816>
 111. Baster D, Maziarz W, Świerczek K et al (2015) Structural and electrochemical properties of Na_{0.72}CoO₂ as cathode material for sodium-ion batteries. *J Solid State Electrochem* 19:3605–3612. <https://doi.org/10.1007/s10008-015-2977-6>
 112. Molenda J (2017) Electronic structure ‘engineering’ in the development of materials for Li-ion and Na-ion batteries. *Adv Nat Sci Nanosci Nanotechnol* 8:015007. <https://doi.org/10.1088/2043-6254/aa5955>
 113. Gędziorowski B, Świerczek K, Molenda J (2012) La_{1-x}Ba_xCo_{0.2}Fe_{0.8}O_{3-δ} perovskites for application in intermediate temperature SOFCs. *Solid State Ionics* 225:437–442. <https://doi.org/10.1016/j.ssi.2012.05.025>
 114. Molenda J, Świerczek K, Zajac W (2007) Functional materials for the IT-SOFC. *J Power Sources* 173:657–670. <https://doi.org/10.1016/j.jpowsour.2007.05.085>
 115. Świerczek K, Klimkowicz A, Niemczyk A et al (2014) Oxygen storage-related properties of substituted BaLnMn₂O_{5+δ} A-site ordered manganites. *Funct Mater Lett* 07:1440004. <https://doi.org/10.1142/S1793604714400049>
 116. Klimkowicz A, Świerczek K, Takasaki A et al (2015) Crystal structure and oxygen storage properties of BaLnMn₂O_{5+δ} (Ln: Pr, Nd, Sm, Gd, Dy, Er and Y) oxides. *Mater Res Bull* 65:116–122. <https://doi.org/10.1016/j.materresbull.2015.01.041>
 117. Zajac W, Rusinek D, Zheng K, Molenda J (2013) Applicability of Gd-doped BaZrO₃, SrZrO₃, BaCeO₃ and SrCeO₃ proton conducting perovskites as electrolytes for solid oxide fuel cells. *Open Chem* 11:471–484. <https://doi.org/10.2478/s11532-012-0144-9>
 118. Kulka A, Hu Y, Dezanneau G, Molenda J (2011) Investigation of GdBaCo_{2-x}Fe_xO_{5.5-δ} as a cathode material for intermediate temperature solid oxide fuel cells. *Funct Mater Lett* 04:157–160. <https://doi.org/10.1142/S1793604711001737>
 119. Zajac W, Suescun L, Świerczek K, Molenda J (2009) Structural and electrical properties of grain boundaries in Ce_{0.85}Gd_{0.15}O_{1.925} solid electrolyte modified by addition of transition metal ions. *J Power Sources* 194:2–9. <https://doi.org/10.1016/j.jpowsour.2008.12.020>
 120. Zajac W, Molenda J (2011) Properties of doped ceria solid electrolytes in reducing atmospheres. *Solid State Ionics* 192:163–167. <https://doi.org/10.1016/j.ssi.2010.05.025>
 121. Zheng K, Świerczek K (2016) Evaluation of W-containing Sr_{1-x}Ba_xFe_{0.75}W_{0.25}O_{3-x} (x= 0, 0.5, 1) anode materials for solid oxide fuel cells. *Solid State Ionics* 288:124–129. <https://doi.org/10.1016/j.ssi.2015.11.022>

122. Zheng K, Klimkowicz A, Świerczek K et al (2015) Chemical diffusion and surface exchange in selected Ln–Ba–Sr–Co–Fe perovskite-type oxides. *J Alloys Compd* 645:S357–S360. <https://doi.org/10.1016/j.jallcom.2014.12.110>
123. Niemczyk A, Olszewska A, Du Z et al (2018) Assessment of layered La_{2-x}(Sr, Ba)_xCuO_{4-δ} oxides as potential cathode materials for SOFCs. *Int J Hydrogen Energy* 43:15492–15504. <https://doi.org/10.1016/j.ijhydene.2018.06.119>
124. Możdzierz M, Dąbrowa J, Stepień A et al (2021) Mixed ionic-electronic transport in the high-entropy (Co,Cu,Mg,Ni,Zn)₁-Li O oxides. *Acta Mater* 208:116735. <https://doi.org/10.1016/j.actamat.2021.116735>
125. Zhang Y, Zhang B, Zhao H et al (2021) Electrochemical performance and structural durability of Mg-doped SmBaMn₂O_{5+δ} layered perovskite electrode for symmetrical solid oxide fuel cell. *Catal Today* 364:80–88. <https://doi.org/10.1016/j.cattod.2020.05.057>
126. Olszewska A, Zhang Y, Du Z et al (2019) Mn-rich SmBaCo_{0.5}Mn_{1.5}O_{5+δ} double perovskite cathode material for SOFCs. *Int J Hydrogen Energy* 44:27587–27599. <https://doi.org/10.1016/j.ijhydene.2019.08.254>
127. Skubida W, Zheng K, Stepień A et al (2021) SrCe_{0.9}In_{0.1}O_{3-δ}-based reversible symmetrical protonic ceramic cell. *Mater Res Bull* 135:111154. <https://doi.org/10.1016/j.materresbull.2020.111154>
128. Klimkowicz A, Świerczek K, Kobayashi S et al (2018) Improvement of oxygen storage properties of hexagonal YMnO_{3+δ} by microstructural modifications. *J Solid State Chem* 258:471–476. <https://doi.org/10.1016/j.jssc.2017.10.037>
129. Kawalec M, Krawiec H (2015) Corrosion resistance of high-alloyed white cast iron. *Arch Metall Mater* 60:301–303. <https://doi.org/10.1515/amm-2015-0048>
130. Kozina I, Krawiec H, Starowicz M, Kawalec M (2021) Corrosion resistance of MgZn alloy covered by chitosan-based coatings. *Int J Mol Sci* 22:8301. <https://doi.org/10.3390/ijms22158301>
131. Krawiec H, Lelito J, Tyrała E, Banaś J (2009) Relationships between microstructure and pitting corrosion of ADI in sodium chloride solution. *J Solid State Electrochem* 13:935–942. <https://doi.org/10.1007/s10008-008-0636-x>
132. Krawiec H, Vignal V, Latkiewicz M, Herbst F (2018) Structure and corrosion behaviour of electrodeposited Co-Mo/TiO₂ nanocomposite coatings. *Appl Surf Sci* 427:1124–1134. <https://doi.org/10.1016/j.apsusc.2017.08.111>
133. Lelek-Borkowska U, Gruszka M, Banas J (2021) Effect of cathodic protection on corrosion of water-pipe network in kraków-case study. *Arch Foundry Eng* 21:59–64. <https://doi.org/10.24425/afe.2021.138666>
134. Bisztyga M, Lelek-Borkowska U, Proniewicz E, Banaś J (2016) Cathodic behaviour of nickel in alcohol solutions of electrolytes. *Electrochim Acta* 207:1–8. <https://doi.org/10.1016/j.electacta.2016.04.156>
135. Starowicz M, Stypuła B (2008) Electrochemical synthesis of ZnO nanoparticles. *Eur J Inorg Chem* 2008:869–872. <https://doi.org/10.1002/ejic.200700989>
136. Stypuła B, Banaś J, Starowicz M et al (2006) Production of nanoparticles of copper compounds by anodic dissolution of copper in organic solvents. *J Appl Electrochem* 36:1407–1414. <https://doi.org/10.1007/s10800-006-9233-9>
137. Starowicz M, Stypuła B, Banaś J (2006) Electrochemical synthesis of silver nanoparticles. *Electrochem Commun* 8:227–230. <https://doi.org/10.1016/j.elecom.2005.11.018>
138. Światowska-Mrowiecka J, Banaś J (2005) Anodic behaviour of zinc in methanol solutions of lithium perchlorate. *Electrochim Acta* 50:1829–1840. <https://doi.org/10.1016/j.electacta.2004.08.035>
139. Krawiec H, Vignal V, Banas J (2006) Macroscopic and local electrochemical studies of austempered ductile iron in perchlorate solutions. *J Electrochem Soc* 153:B231. <https://doi.org/10.1149/1.2197635>
140. Banas J, Lelek-Borkowska U, Starowicz M (2004) Electrochemical behaviour of p-Si in methanol solutions of chlorides. *J Solid State Electrochem* 8:422–429. <https://doi.org/10.1007/s10008-003-0475-8>
141. Lelek-Borkowska U, Banaś J (2006) Passivation and local corrosion of p-silicon in anhydrous organic solutions of chlorides. In: *Passivation of Metals and Semiconductors, and Properties of Thin Oxide Layers*. Elsevier 245–250
142. Proniewicz E, Tała A, Starowicz M et al (2021) Is the electrochemical or the “green chemistry” method the optimal method for the synthesis of ZnO nanoparticles for applications to biological material? Characterization and SERS on ZnO. *Colloids Surfaces A Physicochem Eng Asp* 609:125771. <https://doi.org/10.1016/j.colsurfa.2020.125771>
143. Proniewicz E, Tała A, Wójcik A et al (2020) SERS activity and spectroscopic properties of Zn and ZnO nanostructures obtained by electrochemical and green chemistry methods for applications in biology and medicine. *Phys Chem Chem Phys* 22:28100–28114. <https://doi.org/10.1039/D0CP03517C>
144. Krawiec H, Vignal V, Lelito J et al (2021) In-situ monitoring of the corrosion behaviour of austempered ductile iron (ADI) under cyclic salt spray exposure. *Corros Sci* 185:109437. <https://doi.org/10.1016/j.corsci.2021.109437>
145. Grevey A-L, Vignal V, Krawiec H et al (2020) Microstructure and long-term corrosion of archaeological iron alloy artefacts. *Herit Sci* 8:57. <https://doi.org/10.1186/s40494-020-00398-9>
146. Heintz O, Vignal V, Krawiec H, Loch J (2017) Passivity and corrosion behaviour of Ti-10Mo-4Zr and Ti-6Al-4V alloys after long-term ageing in Ringer’s solution at 37 °C. *J Solid State Electrochem* 21:1445–1455. <https://doi.org/10.1007/s10008-017-3506-6>
147. Krawiec H, Vignal V, Schwarzenboeck E, Banas J (2013) Role of plastic deformation and microstructure in the micro-electrochemical behaviour of Ti-6Al-4V in sodium chloride solution. *Electrochim Acta* 104:400–406. <https://doi.org/10.1016/j.electacta.2012.12.029>
148. Krawiec H, Vignal V, Akid R (2008) Numerical modelling of the electrochemical behaviour of 316L stainless steel based upon static and dynamic experimental microcapillary-based techniques. *Electrochim Acta* 53:5252–5259. <https://doi.org/10.1016/j.electacta.2008.02.063>
149. Major L, Krawiec H, Lackner JM et al (2020) Nanoscale characterization of corrosion mechanisms in advanced Zr/Zr₃N and Zr/Zr₃N+a-C:H nano-multilayer coatings for medical tools. *Mater Charact* 168:110565. <https://doi.org/10.1016/j.matchar.2020.110565>
150. Eiler K, Krawiec H, Kozina I et al (2020) Electrochemical characterisation of multifunctional electrocatalytic mesoporous Ni-Pt thin films in alkaline and acidic media. *Electrochim Acta* 359:136952. <https://doi.org/10.1016/j.electacta.2020.136952>
151. Overview. https://en.uj.edu.pl/en_GB/about-university/overview
152. Zaraska L, Kurowska E, Sulka GD, Jaskuła M (2012) Porous alumina membranes with branched nanopores as templates for fabrication of Y-shaped nanowire arrays. *J Solid State Electrochem* 16:3611–3619. <https://doi.org/10.1007/s10008-012-1795-3>
153. Zaraska L, Sulka GD, Jaskuła M (2011) Anodic alumina membranes with defined pore diameters and thicknesses obtained by adjusting the anodizing duration and pore opening/widening time. *J Solid State Electrochem* 15:2427–2436. <https://doi.org/10.1007/s10008-011-1471-z>
154. Zaraska L, Sulka GD, Jaskuła M (2010) Porous anodic alumina membranes formed by anodization of AA1050 alloy as templates for fabrication of metallic nanowire arrays. *Surf Coatings*

- Technol 205:2432–2437. <https://doi.org/10.1016/j.surfcoat.2010.09.038>
155. Sulka GD, Brzózka A, Zaraska L, Jaskuła M (2010) Through-hole membranes of nanoporous alumina formed by anodizing in oxalic acid and their applications in fabrication of nanowire arrays. *Electrochim Acta* 55:4368–4376. <https://doi.org/10.1016/j.electacta.2010.01.048>
156. Zaraska L, Bobruk M, Sulka GD (2015) Formation of nanoporous tin oxide layers on different substrates during anodic oxidation in oxalic acid electrolyte. *Adv Condens Matter Phys* 2015:1–11. <https://doi.org/10.1155/2015/302560>
157. Gawlak K, Knapik A, Sulka GD, Zaraska L (2022) Improving the photoelectrochemical performance of porous anodic SnOx films by adjusting electrosynthesis conditions. *Int J Energy Res* 46:17465–17477. <https://doi.org/10.1002/er.8414>
158. Zaraska L, Mika K, Hnida KE et al (2017) High aspect-ratio semiconducting ZnO nanowires formed by anodic oxidation of Zn foil and thermal treatment. *Mater Sci Eng B* 226:94–98. <https://doi.org/10.1016/j.mseb.2017.09.003>
159. Zaraska L, Mika K, Syrek K, Sulka GD (2017) Formation of ZnO nanowires during anodic oxidation of zinc in bicarbonate electrolytes. *J Electroanal Chem* 801:511–520. <https://doi.org/10.1016/j.jelechem.2017.08.035>
160. Sulka GD, Kapusta-Kołodziej J, Brzózka A, Jaskuła M (2010) Fabrication of nanoporous TiO2 by electrochemical anodization. *Electrochim Acta* 55:4359–4367. <https://doi.org/10.1016/j.electacta.2009.12.053>
161. Kapusta-Kołodziej J, Tynkevych O, Pawlik A et al (2014) Electrochemical growth of porous titanium dioxide in a glycerol-based electrolyte at different temperatures. *Electrochim Acta* 144:127–135. <https://doi.org/10.1016/j.electacta.2014.08.055>
162. Jarosz M, Pawlik A, Kapusta-Kołodziej J et al (2014) Effect of the previous usage of electrolyte on growth of anodic titanium dioxide (ATO) in a glycerol-based electrolyte. *Electrochim Acta* 136:412–421. <https://doi.org/10.1016/j.electacta.2014.05.077>
163. Zaraska L, Sulka GD, Jaskuła M (2012) Fabrication of free-standing copper foils covered with highly-ordered copper nanowire arrays. *Appl Surf Sci* 258:7781–7786. <https://doi.org/10.1016/j.apsusc.2012.04.148>
164. Zaraska L, Kurowska E, Sulka GD, Jaskuła M (2012) Template-assisted fabrication of tin and antimony based nanowire arrays. *Appl Surf Sci* 258:9718–9722. <https://doi.org/10.1016/j.apsusc.2012.06.018>
165. Hnida K, Mech J, Sulka GD (2013) Template-assisted electro-deposition of indium–antimony nanowires – comparison of electrochemical methods. *Appl Surf Sci* 287:252–256. <https://doi.org/10.1016/j.apsusc.2013.09.135>
166. Sulka GD, Hnida K, Brzózka A (2013) pH sensors based on polypyrrole nanowire arrays. *Electrochim Acta* 104:536–541. <https://doi.org/10.1016/j.electacta.2012.12.064>
167. Szuwarzyński M, Zaraska L, Sulka GD, Zapotoczny S (2013) Pulsatile releasing platform of nanocontainers equipped with thermally responsive polymeric nanovalves. *Chem Mater* 25:514–520. <https://doi.org/10.1021/cm303930y>
168. Brudzisz A, Rajksa D, Gajewska M et al (2020) Controlled synthesis and characterization of AgPd nanowire arrays for electrocatalytic applications. *J Electroanal Chem* 873:114373. <https://doi.org/10.1016/j.jelechem.2020.114373>
169. Zaraska L, Syrek K, Hnida KE et al (2016) Nanoporous tin oxides synthesized via electrochemical anodization in oxalic acid and their photoelectrochemical activity. *Electrochim Acta* 205:273–280. <https://doi.org/10.1016/j.electacta.2016.02.023>
170. Zych M, Syrek K, Pisarek M, Sulka GD (2022) Synthesis and characterization of anodic WO3 layers in situ doped with C, N during anodization. *Electrochim Acta* 411:140061. <https://doi.org/10.1016/j.electacta.2022.140061>
171. Sołtys-Mróz M, Syrek K, Pięta Ł et al (2022) Photoelectrochemical performance of nanotubular Fe2O3–TiO2 electrodes under solar radiation. *Nanomaterials* 12:1546. <https://doi.org/10.3390/nano12091546>
172. Mika K, Syrek K, Uchacz T et al (2022) Dark nanostructured ZnO films formed by anodic oxidation as photoanodes in photoelectrochemical water splitting. *Electrochim Acta* 414:140176. <https://doi.org/10.1016/j.electacta.2022.140176>
173. Gawlak K, Popiołek D, Pisarek M et al (2022) CdS-decorated porous anodic SnOx photoanodes with enhanced performance under visible light. *Materials (Basel)* 15:3848. <https://doi.org/10.3390/ma15113848>
174. Syrek K, Grudzień J, Sennik-Kubiec A et al (2019) Anodic titanium oxide layers modified with gold, silver, and copper nanoparticles. *J Nanomater* 2019:1–10. <https://doi.org/10.1155/2019/9208734>
175. Kurowska E, Brzózka A, Jarosz M et al (2013) Silver nanowire array sensor for sensitive and rapid detection of H2O2. *Electrochim Acta* 104:439–447. <https://doi.org/10.1016/j.electacta.2013.01.077>
176. Kurowska-Tabor E, Gawlak K, Hnida K et al (2016) Synthesis of porous thin silver films and their application for hydrogen peroxide sensing. *Electrochim Acta* 213:811–821. <https://doi.org/10.1016/j.electacta.2016.08.007>
177. Zaraska L, Gawlak K, Kurowska-Tabor E et al (2017) Template-assisted synthesis of rough Ag nanorods and their application for amperometric sensing of H2O2. *Comptes Rendus Chim* 20:693–696. <https://doi.org/10.1016/j.crci.2017.03.001>
178. Hnida KE, Socha RP, Sulka GD (2013) Polypyrrole–Silver composite nanowire arrays by cathodic Co-deposition and their electrochemical properties. *J Phys Chem C* 130916100825004. <https://doi.org/10.1021/jp4038304>
179. Ryzek K, Kozieł M, Wiercigroch E et al (2020) Fast fabrication of nanostructured semiconducting oxides by anodic oxidation of brass. *Mater Sci Semicond Process* 113:105035. <https://doi.org/10.1016/j.mssp.2020.105035>
180. Jarosz M, Pawlik A, Szuwarzyński M et al (2016) Nanoporous anodic titanium dioxide layers as potential drug delivery systems: drug release kinetics and mechanism. *Colloids Surfaces B Biointerfaces* 143:447–454. <https://doi.org/10.1016/j.colsurfb.2016.03.073>
181. Pawlik A, Rehman MAU, Nawaz Q et al (2019) Fabrication and characterization of electrophoretically deposited chitosan-hydroxyapatite composite coatings on anodic titanium dioxide layers. *Electrochim Acta* 307:465–473. <https://doi.org/10.1016/j.electacta.2019.03.195>
182. Golda-Cepa M, Syrek K, Brzywczy-Wloch M et al (2016) Primary role of electron work function for evaluation of nanostructured titania implant surface against bacterial infection. *Mater Sci Eng C* 66:100–105. <https://doi.org/10.1016/j.msec.2016.04.079>
183. Pawlik A, Jarosz M, Syrek K, Sulka GD (2017) Co-delivery of ibuprofen and gentamicin from nanoporous anodic titanium dioxide layers. *Colloids Surfaces B Biointerfaces* 152:95–102. <https://doi.org/10.1016/j.colsurfb.2017.01.011>
184. Pawlik A, Jarosz M, Socha RP, Sulka GD (2021) The impacts of crystalline structure and different surface functional groups on drug release and the osseointegration process of nanostructured TiO2. *Molecules* 26:1723. <https://doi.org/10.3390/molecules26061723>
185. Wierzbicka E, Szultka-Młyńska M, Buszewski B, Sulka GD (2016) Epinephrine sensing at nanostructured Au electrode and determination its oxidative metabolism. *Sensors Actuators B Chem* 237:206–215. <https://doi.org/10.1016/j.snb.2016.06.073>
186. Wierzbicka E, Sulka GD (2016) Fabrication of highly ordered nanoporous thin Au films and their application for electrochemical determination of epinephrine. *Sensors Actuators B Chem* 222:270–279. <https://doi.org/10.1016/j.snb.2015.08.066>

187. Wierzbicka E, Sulka GD (2016) Nanoporous spongelike Au–Ag films for electrochemical epinephrine sensing. *J Electroanal Chem* 762:43–50. <https://doi.org/10.1016/j.jelechem.2015.12.013>
188. Jarosz M, Grudzień J, Kamiński K et al (2019) Novel bioelectrodes based on polysaccharide modified gold surfaces and electrochemically active *Lactobacillus rhamnosus* GG biofilms. *Electrochim Acta* 296:999–1008. <https://doi.org/10.1016/j.electacta.2018.11.154>
189. Brzózka A, Fic K, Bogusz J et al (2019) Polypyrrole–nickel hydroxide hybrid nanowires as future materials for energy storage. *Nanomaterials* 9:307. <https://doi.org/10.3390/nano9020307>
190. Pawlik A, Socha RP, Hubalek Kalbacova M, Sulka GD (2018) Surface modification of nanoporous anodic titanium dioxide layers for drug delivery systems and enhanced SAOS-2 cell response. *Colloids Surfaces B Biointerfaces* 171:58–66. <https://doi.org/10.1016/j.colsurfb.2018.07.012>
191. Mika K, Socha RP, Nyga P et al (2019) Electrochemical synthesis and characterization of dark nanoporous zinc oxide films. *Electrochim Acta* 305:349–359. <https://doi.org/10.1016/j.electacta.2019.03.052>
192. Thomas R, Gurgul M, Xavier B et al (2022) Lithium and sodium storage performance of tin oxyphosphate anode materials. *Appl Surf Sci* 579:152126. <https://doi.org/10.1016/j.apsusc.2021.152126>
193. Gurgul M, Lytvynenko AS, Jarosz M et al (2020) Hierarchical nanoporous Sn/SnOx systems obtained by anodic oxidation of electrochemically deposited Sn nanofoams. *Nanomaterials* 10:410. <https://doi.org/10.3390/nano10030410>
194. Mission. <https://ikifp.edu.pl/en/institute/mission/>
195. Góral-Kurbiel M, Drelinkiewicz A, Kosydar R et al (2014) The effect of Nafion ionomer on electroactivity of palladium–polypyrrole catalysts for oxygen reduction reaction. *J Solid State Electrochem* 18:639–653. <https://doi.org/10.1007/s10008-013-2299-5>
196. Góral-Kurbiel M, Kosydar R, Gurgul J et al (2016) Carbon supported PdPt nanoparticles for oxygen reduction. The effect of Pd: Pt ratio. *Electrochim Acta* 222:1220–1233. <https://doi.org/10.1016/j.electacta.2016.11.096>
197. Góral-Kurbiel M, Drelinkiewicz A, Kosydar R et al (2014) Palladium content effect on the electrocatalytic activity of palladium–polypyrrole nanocomposite for cathodic reduction of oxygen. *Electrocatalysis* 5:23–40. <https://doi.org/10.1007/s12678-013-0155-0>
198. Socha RP, Nowak P, Laajalehto K, Väyrynen J (2004) Particle-electrode surface interaction during nickel electrodeposition from suspensions containing SiC and SiO₂ particles. *Colloids Surfaces A Physicochem Eng Asp* 235:45–55. <https://doi.org/10.1016/j.colsurfa.2004.01.011>
199. Socha R, Laajalehto K, Nowak P (2002) Influence of the surface properties of silicon carbide on the process of SiC particles codeposition with nickel. *Colloids Surfaces A Physicochem Eng Asp* 208:267–275. [https://doi.org/10.1016/S0927-7757\(02\)00153-X](https://doi.org/10.1016/S0927-7757(02)00153-X)
200. Socha RP, Laajalehto K, Nowak P (2002) Oxidation of the silicon carbide surface in Watts' plating bath. *Surf Interface Anal* 34:413–417. <https://doi.org/10.1002/sia.1329>
201. Pacuła A, Nowak P, Socha RP et al (2013) Preparation and characterization of the electroactive composites containing nickel nanoparticles and carbon nanotubes. *Electrochim Acta* 90:563–572. <https://doi.org/10.1016/j.electacta.2012.12.062>
202. Kopeć M, Szczepanowicz K, Warszyński P, Nowak P (2016) Liquid-core polyelectrolyte nanocapsules produced by membrane emulsification as carriers for corrosion inhibitors. *Colloids Surfaces A Physicochem Eng Asp* 510:2–10. <https://doi.org/10.1016/j.colsurfa.2016.08.056>
203. Haber J, Nowak P, Żurek P (2008) Charge transfer in photocatalytic systems: V and Mo doped TiO₂/Ti electrodes. *Catal Letters* 126:43–48. <https://doi.org/10.1007/s10562-008-9652-9>
204. Haber J, Nowak P, Socha RP, Żurek P (2008) Preparation of TiO₂ for photocatalytic applications - doping of TiO₂/Ti with transition metal ions. *Pol J Chem* 82:1753–1766
205. Haber J, Nowak P, Żurek P (2003) Electrodeposition of hedgehog-shaped gold crystallites on TiO₂ surface and their behavior in anodic oxidation of oxalic acid. *Langmuir* 19:196–199. <https://doi.org/10.1021/la020793y>
206. Dziedzic J, Wodka D, Nowak P, Warszyński P, Simon C, Kumakiri I (2010) Photocatalytic degradation of the humic species as a method of their removal from water - comparison of UV and artificial sunlight irradiation. *Physicochem Probl Miner Process* 45:15–28
207. Mosiałek M, Nowak P, Dudek M, Mordarski G (2014) Oxygen reduction at the Ag|Gd_{0.2}Ce_{0.8}O_{1.9} interface studied by electrochemical impedance spectroscopy and cyclic voltammetry at the silver point electrode. *Electrochim Acta* 120:248–257. <https://doi.org/10.1016/j.electacta.2013.12.071>
208. Mosiałek M, Dudek M, Nowak P et al (2013) Changes in the morphology and the composition of the Ag|Gd_{0.2}Ce_{0.8}O_{1.9} interface caused by polarization. *Electrochim Acta* 104:474–480. <https://doi.org/10.1016/j.electacta.2013.01.117>
209. Mosiałek M, Bielańska E, Socha RP et al (2012) Changes in the morphology and the composition of the Ag|YSZ and Ag|LSM interfaces caused by polarization. *Solid State Ionics* 225:755–759. <https://doi.org/10.1016/j.ssi.2012.03.011>
210. Szalaniec M, Hagel C, Menke M et al (2007) Kinetics and mechanism of oxygen-independent hydrocarbon hydroxylation by ethylbenzene dehydrogenase. *Biochemistry* 46:7637–7646. <https://doi.org/10.1021/bi700633c>
211. Tataruch M, Heider J, Bryjak J et al (2014) Suitability of the hydrocarbon-hydroxylating molybdenum-enzyme ethylbenzene dehydrogenase for industrial chiral alcohol production. *J Biotechnol* 192:400–409. <https://doi.org/10.1016/j.jbiotec.2014.06.021>
212. Choi H, Demeke D, Kang F-A et al (2003) Synthetic studies on the marine natural product halichondrins. *Pure Appl Chem* 75:1–17. <https://doi.org/10.1351/pac200375010001>
213. Cole DC, Olland AM, Jacob J et al (2010) Identification and characterization of acidic mammalian chitinase inhibitors. *J Med Chem* 53:6122–6128. <https://doi.org/10.1021/jm100533p>
214. Chen JJ, Cole DC, Ciszewski G et al (2010) Identification of a new class of small molecule C5a receptor antagonists. *Bioorg Med Chem Lett* 20:662–664. <https://doi.org/10.1016/j.bmcl.2009.11.058>
215. Nowak P, Laajalehto K (2000) Oxidation of galena surface – an XPS study of the formation of sulfoxy species. *Appl Surf Sci* 157:101–111. [https://doi.org/10.1016/S0169-4332\(99\)00575-9](https://doi.org/10.1016/S0169-4332(99)00575-9)
216. Wodka D, Socha RP, Bielańska E et al (2014) Photocatalytic activity of titanium dioxide modified by Fe₂O₃ nanoparticles. *Appl Surf Sci* 319:173–180. <https://doi.org/10.1016/j.apsusc.2014.08.010>
217. Bigos A, Wolowicz M, Janusz-Skuza M et al (2021) Citrate-based baths for electrodeposition of nanocrystalline nickel coatings with enhanced hardness. *J Alloys Compd* 850:156857. <https://doi.org/10.1016/j.jallcom.2020.156857>
218. Beltowska-Lehman E, Ozga P, Swiatek Z, Lupi C (2002) Influence of structural factor on corrosion rate of functional Zn–Ni coatings. *Cryst Eng* 5:335–345. [https://doi.org/10.1016/S1463-0184\(02\)00045-X](https://doi.org/10.1016/S1463-0184(02)00045-X)
219. Beltowska-Lehman E, Indyka P, Bigos A et al (2015) Ni–W/ZrO₂ nanocomposites obtained by ultrasonic DC electrodeposition. *Mater Des* 80:1–11. <https://doi.org/10.1016/j.matdes.2015.04.049>
220. Beltowska-Lehman E, Indyka P, Bigos A et al (2016) Effect of current density on properties of Ni–W nanocomposite coatings reinforced with zirconia particles. *Mater Chem Phys* 173:524–533. <https://doi.org/10.1016/j.matchemphys.2016.02.050>

221. Beltowska-Lehman E, Indyka P, Bigos A et al (2016) Effect of hydrodynamic conditions of electrodeposition process on microstructure and functional properties of Ni-W/ZrO₂ nanocomposites. *J Electroanal Chem* 775:27–36. <https://doi.org/10.1016/j.jelechem.2016.05.003>
222. Beltowska-Lehman E, Swiatek Z, Lipinski M et al (2000) The influence of surface modification on optoelectronic properties of monocrystalline silicon solar cells. In: Conference Record of the Twenty-Eighth IEEE Photovoltaic Specialists Conference - (Cat. No.00CH37036). IEEE 331–334
223. Lipinski M, Panek P, Beltowska E, Czternastek H (2003) Reduction of surface reflectivity by using double porous silicon layers. *Mater Sci Eng B* 101:297–299. [https://doi.org/10.1016/S0921-5107\(02\)00720-1](https://doi.org/10.1016/S0921-5107(02)00720-1)
224. Ozga P (2006) Electrodeposition of Sn-Ag and Sn-Ag-Cu alloys from thiourea aqueous solutions. *Arch Metall Mater* 51:413–422
225. Ozga P, Świątek Z, Michalec M, Onderka B, Bonarski J (2008) Phase structure and texture of electrodeposited InSn alloys on copper substrate. *Arch Metall Mater* 53:307–315
226. Slupska M, Ozga P (2014) Electrodeposition of Sn-Zn-Cu alloys from citrate solutions. *Electrochim Acta* 141:149–160. <https://doi.org/10.1016/j.electacta.2014.07.039>
227. Beltowska-Lehman E, Ozga P, Swiatek Z, Lupi C (2002) Electrodeposition of Zn–Ni protective coatings from sulfate–acetate baths. *Surf Coatings Technol* 151–152:444–448. [https://doi.org/10.1016/S0257-8972\(01\)01614-0](https://doi.org/10.1016/S0257-8972(01)01614-0)
228. Ozga P, Bielańska E (2003) Determination of the corrosion rate of Zn and Zn–Ni layers by the EDS technique. *Mater Chem Phys* 81:562–565. [https://doi.org/10.1016/S0254-0584\(03\)00075-0](https://doi.org/10.1016/S0254-0584(03)00075-0)
229. Kazimierzczak H, Świątek Z, Ozga P (2020) Electrodeposition of tin-zinc-bismuth alloys from aqueous citrate-EDTA baths. *Electrochim Acta* 338:135889. <https://doi.org/10.1016/j.electacta.2020.135889>
230. Pohrelyuk IM, Tkachuk OV, Proskurnyak RV et al (2020) Cyto-compatibility evaluation of Ti-6Al-4V alloy after gas oxynitriding. *J Mater Eng Perform* 29:7785–7792. <https://doi.org/10.1007/s11665-020-05265-z>
231. Łukasiewicz Research Network – Krakow Institute of Technology. <https://kit.lukasiewicz.gov.pl/en/company-presentation/>. Accessed 13 Oct 2022
232. Kozak J, Zybura-Skrabalak M (2016) Some problems of surface roughness in electrochemical machining (ECM). *Procedia CIRP* 42:101–106. <https://doi.org/10.1016/j.procir.2016.02.198>
233. Ruszaj A, Zybura-Skrabalak M (2001) The mathematical modelling of electrochemical machining with flat ended universal electrodes. *J Mater Process Technol* 109:333–338. [https://doi.org/10.1016/S0924-0136\(00\)00816-5](https://doi.org/10.1016/S0924-0136(00)00816-5)
234. Skrabalak G, Zybura-Skrabalak M, Ruszaj A (2004) Building of rules base for fuzzy-logic control of the ECDM process. *J Mater Process Technol* 149:530–535. <https://doi.org/10.1016/j.jmatprotec.2003.11.058>
235. Zybura-Skrabalak M, Ruszaj A (2000) The influence of electrode surface geometrical structure on electrochemical smoothing process. *J Mater Process Technol* 107:288–292. [https://doi.org/10.1016/S0924-0136\(00\)00696-8](https://doi.org/10.1016/S0924-0136(00)00696-8)

Publisher's Note Springer Nature remains neutral with regard to jurisdictional claims in published maps and institutional affiliations.