

Studies of enzymatic oxidation of TMP-fibers and lignin model compounds by a Laccase–Mediator–System using different ^{14}C and ^{13}C techniques

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Received: 29 November 2010 / Published online: 20 July 2011

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Abstract In this work, the results of the enzymatic oxidation of TMP-fibers (thermomechanical pulp) and a well-structured lignin model compound, the dehydropolymer (DHP), were investigated by different ^{14}C and ^{13}C methods, caused by a Laccase–Mediator–System (LMS). These methods are the nuclear magnetic resonance spectroscopy (^{13}C -NMR) with DHP (unmarked) and the determination of the $^{14}\text{CO}_2$ release of ^{14}C -marked DHP and TMP-fibers. The ^{13}C -NMR measurements were chosen to analyze the structural changes of the LMS-treated DHP model compounds and TMP-fibers qualitatively and quantitatively. The data of $^{14}\text{CO}_2$ release give an explanation of the demethylation of DHP and TMP-fibers. The effect of the LMS is shown by comparing the results in respect of DHP and TMP-fibers, which were only treated with laccase and of an inactivated LMS as the control. Comparing the results of the ^{13}C -NMR method, in particular the use of the Mediator during the enzymatical treatment, showed significant changes in the structure of the DHP. Also, the TMP-fibers were materially influenced by the LMS. The analysis of the $^{14}\text{CO}_2$ release data of the ^{14}C -marked DHP and TMP-fibers revealed that the rate of $^{14}\text{CO}_2$ increases in the ^{14}C -2 atom as well as in the O^{14}CH_3 group within the first hour of Laccase–Mediator incubation. Therefore, the $^{14}\text{CO}_2$ release from the DHP was higher than from the TMP-fibers.

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Introduction

Besides Ligninperoxidases (E.C. 1.11.1.7.) and Manganperoxidases (E.C. 1.11.1.13.), Laccases (Polyphenoloxidases; E.C. 1.10.3.2.) are the most important lignin-degrading and lignin-polymerizing enzymes (Thurston 1994). In nature, laccases are generated by many fungi, especially white-rot fungi (mostly basidiomycetes), plants, and some bacteria (Mayer and Staples 2002; Claus 2003, 2004; Baldrian 2006; Hoegger et al. 2006). Laccases are used in many technical processes such as in the biobleaching in the pulp and paper industry (Gutierrez et al. 2009; Zhang et al. 2007) and the textile industry (Galante and Formantici 2003). Those are mostly commercial laccases from saprophytic basiomycetes like the white-rot fungi *Trametes*, *Pleurotus*, or *Agaricus* species, which release high amounts of extra cellular enzymes (Minussi et al. 2002; Xu 2005; Rühl et al. 2007). Some former works report on the high potential of laccases for industrial applications (Wagenführ 1988; Wagenführ et al. 1989; Call and Mücke 1997; Couto and Toca-Herrera 2006; Riva 2006; Mikolasch and Schauer 2009). A disadvantage of the application of laccase is its low redox potential. Because of this low redox potential, laccases need a free phenolic group at the aromatic rings of the lignin for oxidation that normally inhibits their application in the lignin biotechnology because in natural lignin, the most phenolic groups are substituted during the radical reaction of the lignin synthesis (Lenowicz et al. 2001; Rochefort et al. 2004). Bourbonnais and Paice (1990) were the first who could show that, after the addition of low-molecular redox compounds (the Mediator), laccase is able to attack also non-phenolic lignin connections and to fasten and intensify the reactions. This is explained by the ability of the little Mediator molecules to reach the smaller reaction locations of the lignin, which are not accessible by the larger laccase molecules as well as the new reaction mechanisms which can be implemented by the Mediators during the enzymatic modification of the lignin (Bourbonnais and Paice 1992; Bourbonnais et al. 1995; Call and Mücke 1994). Thus, the Laccase–Mediator–System (LMS) offers an improvement on the low redox potential laccases.

In view of this, Mediators play a new and important role in the biochemistry of lignin degradation (Call and Mücke 1997; Goodell et al. 1997; Crestini and Argyropoulos 2001; Crestini et al. 2003). Mediators like 1-hydroxybenzotriazole, acetosyringone, and syringaldehyde were successfully adopted in combination with laccases in the biochemical bleaching of paper pulp and in the degradation of textile fibers. In contrast, the Mediator p-coumaric acid, for example, was less suitable for such applications (Moreira et al. 1998; Pandey et al. 2000; Lund and Felby 2001; Camarero et al. 2004; Chakar and Ragauskas 2004; Rochefort et al. 2004; Hakala et al. 2005; Ibarra et al. 2006; Camarero et al. 2007; Gutiérrez et al. 2007; Vivekanand et al. 2008). A positive effect of the Mediator 4-hydroxy benzoic acid (HBA) in the modifying of paper is described by Chandra et al. (2004). The LMS for the production of wooden fiberboards was first applied in Euring (2008) and then in Kharazipour and Euring (2010).

In this work, some positive effects of the HBA during the enzymatic oxidation of TMP-fibers (TMP = thermomechanical pulp) are described and verified by synthetic labeled lignin models.

Materials and methods

For the ^{13}C -NMR determination, unlabeled TMP-fibers from the wood species *Picea sp.* and *Pinus spec.* were used. For the other method to determine the $^{14}\text{CO}_2$ release, the lignin of TMP-fibers was ^{14}C marked in vivo by the method of Freudenberg and Reichert (1954), Freudenberg (1968) and Trojanowski and Hüttermann (1987).

For the DHP synthesis, coniferyl alcohol was used together with peroxidase from horseradish (Applichem) with an activity of 260 U/mg (Purpurogalin) and H_2O_2 according to Freudenberg's "Zutropfverfahren" (1968).

For the ^{13}C -NMR spectroscopy, the DHP was not labeled. For the $^{14}\text{CO}_2$ measurements, the DHP was labeled with ^{14}C in CH_3 -groups and in side chain-C-2 using the procedure described by Trojanowski and Hüttermann (1987).

The laccase for these studies was used from the genetically modified fungi *Aspergillus sp.* from Novozym, Denmark (Novozym 51003, E.C. 1.10.3.2). The activity of the laccase was routinely determined by monitoring the oxidation of diammonium salt of 2,2'-azinobis-3-ethylbenzothiazoline-6-sulfonic acid (ABTS) (Matsumura et al. 1986). The activity of the pure laccase was about 2200 U/ml (ABTS).

The Mediator 4-hydroxy benzoic acid (HBA) from Applichem had a purity of 99% and solubility of 9 g/l in water.

The reaction buffer was the McIlvain buffer, containing 0.2 M dipotassium hydrogen phosphate (K_2HPO_4) from *AppliChem*, 0.1 M citric acid monohydrate ($\text{C}_6\text{H}_8\text{O}_7 \cdot \text{H}_2\text{O}$) from *AppliChem* and $\text{H}_2\text{O}_{\text{bidest}}$ to dissolve both reagents. The pH value of this buffer was adjusted to 6.

TMP-fibers and DHP treatment with the LMS for ^{13}C -NMR studies

For the ^{13}C -NMR measurements, either 5 g TMP-fibers or 5 g DHP and 150 ml McIlvain buffer (pH 6), containing 200 U/ml laccase and 10 mM HBA, were incubated for 4, 60, and 180 h under constant shaking. In comparison, the same amount of the compounds was used without the Mediator to explain any effect of this Mediator.

As control samples, the TMP-fibers/DHP was incubated in McIlvain buffer with inactivated LMS. After the incubation times, the samples of TMP-fibers and DHP were washed with bidistilled water using a centrifuge with 12,000g. After that, the TMP-fibers and DHP were dried in a vacuum oven at 105°C for 12 h. After the drying process, the samples were measured by the ^{13}C -NMR method using a *Bruker Avance 400 MHz WB*.

^{14}C labeled DHP and TMP-fibers

As mentioned before, for the determination of $^{14}\text{CO}_2$ emission, the TMP-fibers and DHP were ^{14}C marked in the C-2 atom and in $^{14}\text{CH}_3$ methoxyl groups.

The lignin degradation was monitored using a modified radiorespirometric method based on Haider and Trojanowski (1975).

Each sample was placed in a glass flask and incubated in a vacuum in 7 ml McIlvain buffer, containing either 200 U/ml laccase and 10 mM HBA or only 200 U/ml laccase and inactivated LMS (control). The incubation solution was stirred and kept at between 30 and 45°C using a water bath for a total of 6 h.

The $^{14}\text{CO}_2$ released from ^{14}C atoms and $^{14}\text{CH}_3$ methoxyl groups was trapped in 2 ml 10% (w/v) NaOH, and the radioactivity was measured in the liquid scintillation counter C 400 (Fig. 1). The data were collected every hour.

Results and discussion

^{13}C -NMR spectroscopy

Table 1 shows the amounts of aromatic groups in TMP-fibers and DHP treated with laccase and LMS, respectively, at different incubation times. Besides the region of aromatic groups (160–95 ppm), many other regions in the ^{13}C -NMR spectra have been detected—such as aliphatic C compounds (45–10 ppm), methoxyl groups (60–45 ppm), and COOH/C=O groups (220–160 ppm). Those were present especially in TMP-fibers carbohydrates (95–10 ppm). By analyzing all these regions, the most noticeable differences appeared in the aromatic groups which are displayed in the Table. The differences between the values of detected aromatic groups in TMP-fibers and DHP are explained by the constituents of both materials. While DHP is a pure lignin model compound, TMP-fibers consist of many other components, such as cellulose, hemicelluloses, and extractives. For example, DHP does not contain carbohydrates, whereas TMP-fibers contain 78% of carbohydrates

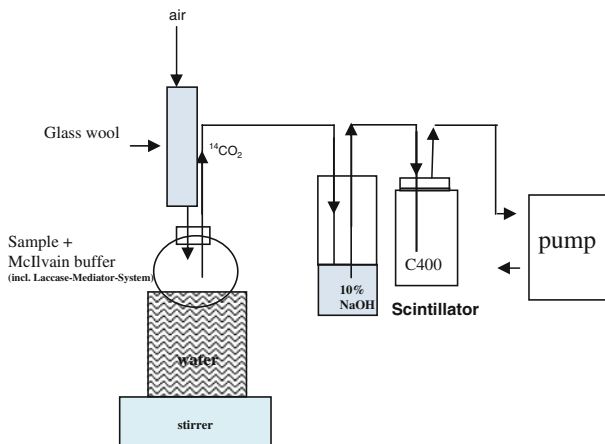


Fig. 1 Radiorespirometric method for TMP-fibers and DHP based on Haider and Trojanowski (1975)

Table 1 Amounts of aromatic groups of DHP and TMP-fibers in different treatments using the ^{13}C -NMR spectroscopy

	Hours of incubation	Samples	TMP-fibers region: 160–95 ppm (aromatic groups) (%)	DHP
Treatments of DHP and TMP-fibers: <i>control</i> incubation with inactivated LMS, <i>laccase</i> laccase in buffer, <i>LMS</i> Laccase–Mediator-System in buffer	0, 4, 60, 180	control	19.8	58.1
		Laccase	19.8	58.2
	0	LMS	19.6	58.1
		Laccase	18.3	52.9
	4	LMS	13.9	48.5
		Laccase	16.8	49.7
	60	LMS	13.0	47.9
		Laccase	16.5	48.6
	180	LMS	12.8	47.2

in relation to the other compounds. Therefore, the amount of aromatic groups in TMP-fibers is much lower than in DHP.

Nonetheless, during analysis of the ^{13}C -NMR spectra, it was shown that under the influence of the LMS, after 4 h of incubation, the aromatic groups (160–95 ppm) were noticeably degraded and transferred into the incubation solution. Prior to the incubation, with the LMS, 19.8% aromatic groups were found in the TMP-fibers and 58.1% in the DHP. After 4 h of incubation, just 13.9% were found in the TMP-fibers and 48.5% in DHP. In comparison, when using only laccase, after 4 h of incubation, the percentage of aromatic compounds in the TMP-fibers was of 18.3% and 52.9% in the DHP. At longer incubation times, a lower degradation value of aromatic groups was noticed. After 60 h of incubation, the aromatic compounds had a percentage of 13.0 in TMP-fibers and 47.9 in DHP, and after 180 h of incubation, a value of 12.8% (TMP-fibers) and 47.2% (DHP).

In reference to the other detected regions of the ^{13}C -NMR spectra during the complete incubation test for methoxyl groups (60–45 ppm), no essential changes were found due to a low degradation rate of methoxyl by the LMS. Equally low degradation values were recognized for aliphatic compounds. Apart from that, the percentile rate of COOH/C=O bonds was increased marginally from 0.7 to 1.3% in TMP-fibers and from 1.4 to 2.9% in DHP after 180 h of incubation. The increase of COOH/C=O bonds is caused by a structural change of TMP-fibers and DHP during the laccase degradation, which forces the formation of these bonds. Additionally, COOH/C=O bonds do not pass into the solution and can be released by the degradation and conversion of aromatic compounds. In the evaluation of the samples based on ^{13}C -NMR spectroscopy, it was found that laccase, especially laccase in a system with Mediator, affects chemical changes on the TMP-fibers and DHP. A lower absorption rate of the TMP-fibers and DHP treated with LMS during the ^{13}C -NMR measurements implied that the lignin-oxidized effect of laccase was increased by the Mediator HBA. Particularly, close to the aromatic compounds the oxidation of the TMP-fibers and DHP must have been strongly supported by the

Mediator. Furthermore, side chains (hydroxyl groups and ether groups) of the hydrocarbons were absorbed less after 4 h of incubation.

Also Milstein et al. (1993) recognized a clear chemical change in the lignin structure—mainly a release of aromatic hydrocarbons.

In contrast to the results of this paper, Kharazipour et al. (1997) performed ^{13}C -NMR measurements with lignin which was already dissolved in fiber suspension by incubation of the wood surface with laccase only. In that paper, the TMP-fibers were analyzed by themselves with the LMS.

In Fig. 2, the changes are illustrated, in comparison with the NMR spectra of the control sample of DHP (points): the DHP sample (broken line), which was incubated for 4 h with laccase, and the LMS-treated DHP (continuous line).

Degradation tests of ^{14}C marked DHP and TMP wood fibers

Analysis of the results showed that incubation supported by a LMS causes an oxidation and degradation of C-2 chains and CH_3 groups and additionally a $^{14}\text{CO}_2$ emission in all TMP-fibers and DHP samples as early as within the first 2 h. In comparison with TMP-fibers, higher emission rates emerged for DHP samples (Figs. 3 and 4). The measured $^{14}\text{CO}_2$ emission of the CH_3 groups of the DHPs, which

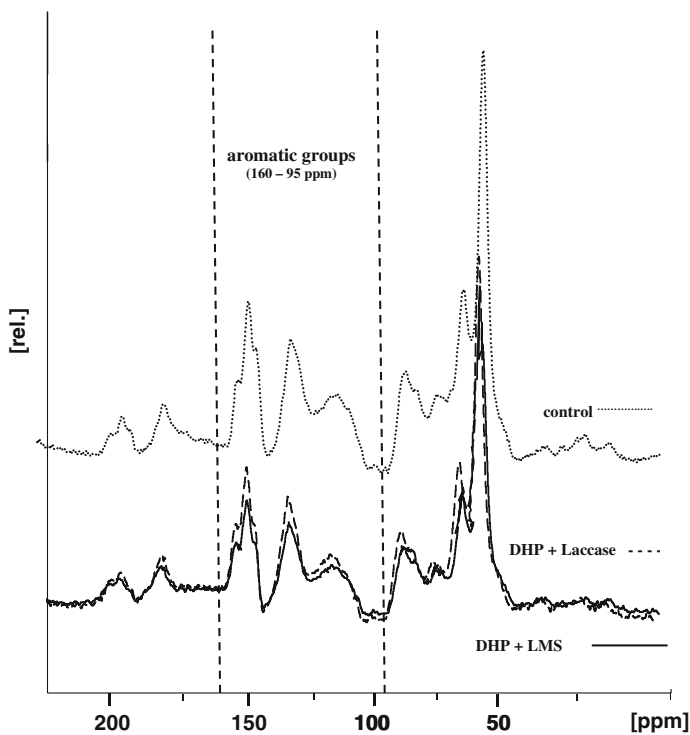


Fig. 2 ^{13}C -NMR spectra of differently treated DHP. Treatments: *control* incubation with inactivated LMS in buffer, *laccase* laccase in buffer, *LMS* Laccase–Mediator–System in buffer, [rel.] refers to relative peak height

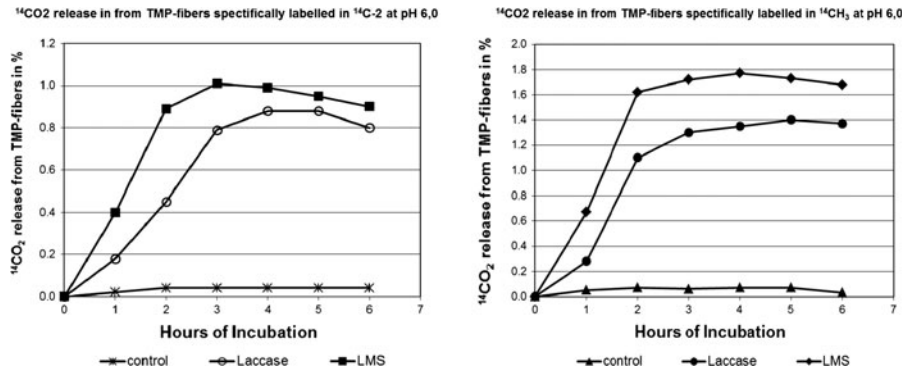


Fig. 3 $^{14}\text{CO}_2$ release from TMP-fibers, labelled in $^{14}\text{C-2}$ and $^{14}\text{CH}_3$. Treatments of TMP-fibers: *control* incubation with inactivated LMS, *laccase* laccase in buffer, *LMS* Laccase–Mediator–System in buffer

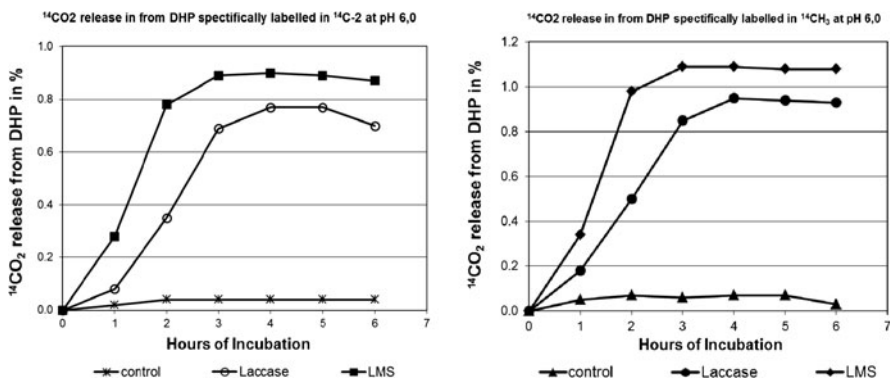


Fig. 4 $^{14}\text{CO}_2$ release from DHP, labelled in $^{14}\text{C-2}$ and $^{14}\text{CH}_3$. Treatments of DHP: *control* incubation with inactivated LMS, *laccase* laccase in buffer, *LMS* Laccase–Mediator–System in buffer

was treated with the LMS, was at 1.6% after 2 h of incubation; for the TMP-fibers, the value was below 1.0%. After 3 and 4 h of incubation for all treated samples, the maximal $^{14}\text{CO}_2$ emission was achieved. Afterward, a more or less continuous behavior for the illustrated incubation time was recognized (Fig. 4). It became obvious that reactions between laccase or the LMS and the TMP-fibers or the DHP proceeded within a few hours faster than in the tests by Haider et al. (1985) and Iimura et al. (1995). The faster degradation rate of the DHP lignin model in comparison with TMP-fibers is presumably closely related to the simpler structure of DHP, which is already a simple polymer. Consequently, the C atoms are easier to reach for the LMS. A lower $^{14}\text{CO}_2$ emission was counted for the TMP-fibers because other components besides lignin are inside of them, i.e., hemicelluloses, cellulose, and extractives.

The most essential aspect during the $^{14}\text{CO}_2$ emission was the different degradation rates of the C-2 chains and CH_3 - groups and the different treatments (LMS or only laccase). Again it becomes obvious that by application of a Mediator, a more intensive and accelerated lignin activation and degradation will be possible. This is an interesting

aspect for the industrial production of wood fiber-based materials, which are made of TMP-fibers that can form a fiber to fiber bond by the cross-linking molecules (Yamaguchi et al. 1992; Kharazipour and Hüttermann 1993, 1998; Kharazipour et al. 1993, 1994; Kharazipour et al. 1997; Kharazipour 1996; Hüttermann et al. 2001; Felby et al. 1997, 2002; Widsten 2002; Kües et al. 2007; Euring 2008).

Conclusion

This article illustrates coherencies between the influence of the LMS on both TMP-fibers and dehydropolymers (DHP). Furthermore, it provides insights into the oxidation mechanisms of the laccase enzyme in interaction with a Mediator in the lignin structure. The results showed that the use of a LMS can increase the lignin degradation rates, which were measured by a nuclear magnetic resonance spectroscopy (^{13}C -NMR) and which were caused by a $^{14}\text{CO}_2$ emission. Further investigations could show to what extent the LMS can be optimized. By a higher activation, and therefore a more intensive lignin degradation, the system could be used in different biochemical and technological applications, i.e., for paper bleaching, in the textile industry, or for the production of wood fiber products.

Acknowledgment Markus Euring is supported by the Deutsche Forschungsgemeinschaft (DFG), grant EU 124/2-1.

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