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The impact of germination and thermal treatments on bioactive compounds of quinoa (*Chenopodium quinoa* Willd.) seeds

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Abstract

Quinoa is a highly nutritious crop with diverse applications in the food industry. The study assessed the impact of various processing techniques, including microwaving, boiling, roasting, steaming, flaking, and germination, on the crude protein content (CP), total phenolic content (TPC), antioxidant activity (AA), and 12 phenolic compounds in quinoa. CP was significantly affected by the heat treatments. Boiled quinoa flakes exhibited the highest average CP, boiling and roasting the lowest. Microwaving strongly enhanced the TPC and the content of six bioactive compounds (CFA, KMP, NAR, QCE, RUT, SA), while boiling and steaming had the most adverse effect. Germination improved the overall nutritional profile of quinoa. The most pronounced increase in the bioactive metabolites occurred between the third and fifth day of germination in a genotype-dependent manner. Six metabolites (NAR, SA, 4BA, IQ, PC, IH) were detected in germinated quinoa for the first time. The results emphasize the substantial influence of processing techniques and type of sample on quinoa nutritional quality and underscore the importance of proper consideration of those factors to obtain nutritionally optimal food products.

Keywords Germination · Phenolics · Pseudocereal · Seed · Thermal processing

Abbreviatior	۱S
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ANOVA	A two-way analysis of variance
AA	Antioxidant activity
4BA	4-Hydroxybenzaldehyde
FA	Caffeic acid
FC	Folin–Ciocalteau
CP	Crude protein
DPPH	2,2-Diphenyl-1-picrylhydrazyl
ESI	Electrospray ionization
GA	Gallic acid
G1D	1 Day of germination
G2D	2 Day of germination

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G3D	3 Days of germination
G4D	4 Days of germination
G5D	5 Days of germination
IH	Isorhamnetin
IQ	Isoquercetin
KMP	Kaempferol
NAR	Naringenin
PC	Pinocembrin
TE	Trolox equivalent
QCE	Quercetin
Q3G	Quercetin 3-O-glucuronide
RUT	Rutin
SA	Salicylic acid
TFC	Total flavonoid content
TPC	Total phenolic content

Introduction

Quinoa (*Chenopodium quinoa* Willd.) is a highly versatile crop with outstanding nutritional value, which was recognized even by ancient Andean populations, considering this pseudocereal a sacred food [1]. Although its cultivation has already spread worldwide, the biggest producers are still the countries of quinoa origin—Peru, Bolivia, and Ecuador [2]. Quinoa has been traditionally consumed in a form of grain or as an ingredient in many food products and dishes, such as soups, porridges, buns, and drinks [3]. Its growing popularity has led to the development of novel foods containing quinoa, in particular gluten-free, vegetarian, vegan, and dairy-free products [4, 5].

Although the quinoa's nutrient-rich profile has been a subject of extensive research, most of the studies were, however, realized on raw materials. Nonetheless, quinoa is usually not eaten raw but processed in order to decrease the content of anti-nutritional compounds, such as saponins and phytic acid [6, 7]. Studied have shown that commonly used processing methods, such as boiling, steaming, microwaving, and extrusion may alter the nutritional content and composition, as well as the overall bioavailability of nutrients. For example, microwaving and boiling under pressure have been reported as a suitable technique for preservation of polyphenols. In comparison, boiling caused the major loss of phenolic compounds and minerals [6, 8]. Although the protein content is not significantly affected by the common heat-utilizing preparations [6], it has been described that some methods like microwaving and fermentation may increase the protein digestibility of final quinoa product [9, 10].

Apart from heat-utilizing preparations, germination has emerged as an alternative and relatively cheap processing technique for improving the nutritional profile by promoting enzymatic activity and release of various bioactive chemical sand minerals [11, 12], while reducing the content of antinutritional factors like phytic acid and tannin [13, 14]. Germination may further improve biological value of quinoa protein and its overall digestibility [15, 16].

Comprehensive studies are needed to explore the impact of different preparation methods on those traits and their implications for further food processing. Therefore, the aim of this paper was to evaluate a spectrum of quinoa seed preparation methods and evaluate their impact on the content of protein, antioxidant activity, total phenolic content and 13 phenolic compounds. By providing an extensive analysis of these effects, this paper aims to raise the awareness around quinoa nutritional quality and the selection of appropriate processing techniques preserving the high-quality nutritional profile of quinoa food products.

Materials and methods

Plant material

A total of three quinoa samples were subjected to analysis. The original seeds of two quinoa samples (genotypes 'Besançon' and 'Faro') were provided from the U.S. National Plant Germplasm System operated by USDA. The seeds of these two genotypes were multiplied to provide sufficient material for further experiments in the experimental field of the Crop Research Institute in Prague in the Czech Republic during the year 2021. One commercial quinoa sample (Probio) was kindly provided by PRO-BIO Ltd, Czech Republic.

Procedure of germination

Germination was carried out on commercial Probio sample and genotypes 'Besançon' and 'Faro'. A total of 10 g of healthy and undamaged seeds of each sample was used for the experiment. In addition, seeds of genotypes 'Besançon' and 'Faro' were thoroughly rinsed in 30% (v/v) hydrogen peroxide for disinfection purposes to minimize microbiological contamination of the seed surface from the field condition. Then, seeds of all three samples were washed several times in distilled water. All three samples were soaked in distilled water for 4 h, drained and then placed in a sterile Petri dish lined with moist filter paper and covered with the lid. Hydrated quinoa seeds were allowed to germinate for 1 day (24 h), 2 days (48 h), 3 days (72 h), 4 days (96 h), and 5 days (120 h), respectively. Germination of Probio sample is shown in Fig. 1. Germination was performed at the room temperature under a 16/8 day/night regime and seeds were regularly watered with distilled water. Sprouted seeds were collected each day of germination and lyophilized before next use. Samples were stored in cold and dark place for following laboratory analyses. The non-germinated samples were indicated as control samples.

Processing techniques

The commercial sample Probio was subjected to several heat-utilizing processing techniques. All processing techniques were carried out under atmospheric pressure in room temperature. Prior to each thermal processing, Probio sample was soaked in distilled water for 24 h. The excess water was drained from the samples prior following treatments. For each treatment, the sampling intervals were established. After each sampling interval, quinoa seeds were immediately drained from any excess water, transferred to a sterile container, and labeled accordingly for subsequent analysis. After a cool-down, samples were lyophilized and stored in cold and dark place for following laboratory analyses. The non-processed samples were indicated as control samples.

In the case of boiling in plain water, grains were boiled in distilled water in the ratio of 1:2 (w/v) for 5, 10, and 15 min. The boiling in NaCl used the same proceeding, salt was at the concertation of 10 g/L (w/v). Microwaving was realized in the microwave oven (ETA 2209 90,000, ETA a.s., **Fig. 1** Germination of Probio sample for 1 day (24 h, **A**), 2 days (48 h, **B**), 3 days (72 h, **C**), 4 days (96 h, **D**), and 5 days (120 h, **E**)



Czech Republic) for 1, 2, and 3 min at the power of 1050 W. Another batch of samples was roasted on the pan for 5, 10, and 15 min at the temperature of 180 ± 20 °C. Lastly, steaming was carried out by placing the quinoa grains on a fine mesh sieve and set over boiling water, covered with a lid. The sample was steamed for 5, 10, and 15 min.

Further, raw Probio seeds were mechanically pressed to obtain flakes using a food processor (Jupiter Küchenmaschinen, System Drive Unit, Weimar, Germany) equipped with a flake roller. Raw quinoa flakes were boiled in distilled water in the ratio of 1:2 (w/v) for 1, 2, 3, 4, and 5 min following the same procedure as mentioned in the

first paragraph. Boiled flakes were subjected to chemical analysis.

Chemicals

Polyphenolic compounds, including 4-OH benzaldehyde, caffeic acid, gallic acid, isoquercetin, isorhamnetin, kaempferol, naringenin, pinocembrin, quercetin, quercetin 3-O-glucuronide, rutin, and salicylic acid, along with the internal standard probenecid were procured from Sigma–Aldrich (St. Louis, MO, USA). Methanol of LC–MS grade (\geq 99.9%) was sourced from Riedel de Haën (Seelze, Germany), while formic acid of LC–MS grade (99%) was obtained from VWR (Leuven, Belgium). Pure water was acquired from a Milli-Q purification system manufactured by Millipore (Bedford, MA, USA).

Standard and sample preparation

The preparation of reference stock solutions involved dissolving the methanol-dissolved reference standards of each phenolic compound to create stock solutions at a concentration of 0.5 mg/mL. These reference stock solutions were subsequently stored at -18 °C. To establish the calibration curves for quantifying the phenolic compounds, the stock solutions were diluted within a methanol to concentration range of 0.001–2.000 µg/mL. In addition, probenecid was dissolved in methanol at a concentration of 0.5 mg/mL to generate a stock solution of the internal standard. The internal standard was then added to the individual reference standard solutions or test samples, resulting in a final concentration of 0.1 µg/mL.

For the analysis using mass spectrometry, the lyophilized samples were milled using an IKA A11 basic mill (IKA-Werke, Staufen, Germany), and the resulting mixture was stored in well-sealed plastic bags in a dark, cold place at 4 °C. The extraction of sample followed the method described by Janovská et al. [17]. Briefly, 0.1 g of the milled mixture was extracted twice with 1 mL of extraction solvent (comprising 80% methanol with probenecid as internal standard at a concentration of 0.1 μ g/mL) in Eppendorf tubes. The extraction was performed using an ultrasonic bath for 60 min at 45 °C. After extraction, the samples were centrifuged for 10 min at 13,500 rpm. The obtained supernatants from each sample were then filtered through 0.2 μ m nylon syringe filters for further analysis.

UHPLC-ESI–MS/MS instrumentation

The chromatographic analysis was conducted using the Dionex UltiMate 3000 UHPLC system (Dionex Softron GmbH, Germering, Germany), comprising a binary pump (HPG-3400RS), an autosampler (WPS-3000RS), a degasser

(SRD-3400), and a column oven (TCC-3000RS). Detection of analytes was performed on the quadrupole/orbital ion trap Q Exactive mass spectrometer (Thermo Fisher Scientific, San Jose, CA, USA). The LC–MS system was equipped with a heated electrospray ionization source (HESI-II) and operated using Xcalibur software, version 4.0 (Thermo Fisher Scientific, San Jose, CA, USA).

UHPLC-ESI–MS/MS analysis

The analytes were separated on a reversed-phase C18 Ascentis Express column ($2.1 \times 100 \text{ mm}$, $2.7 \mu \text{m}$) from Supelco (Bellefonte, PA, USA). The chromatographic separation was performed using a gradient elution method. Solvent A consisted of 0.2% formic acid (v/v) in water, while solvent B comprised methanol with 0.2% formic acid (v/v). The LC gradient commenced with 99% of solvent A and 1% of solvent B at 0 min., followed by a linear gradient elution to 40% A and 60% B at 11 min. The column was then flushed with 100% solvent B for 2 min. Equilibration of the column was accomplished by washing with 99% A and 1% B for additional 2 min. The total analysis time was 15 min. The column temperature was maintained at 40 °C, and the flow rate was set to 0.35 mL/min. The injection volume was 1 μ L.

The mass spectrometer analysis was conducted in negative electrospray ionization (ESI) mode. The spray voltage was set at -2.5 kV, and the sheath gas flow rate, auxiliary gas flow rate, and sweep gas flow rate were 49, 12, and 2 arbitrary units, respectively. The capillary temperature was 260 °C, and nitrogen was used as the sheath, auxiliary, and sweep gas. The heater temperature was maintained at 419 °C, and the S-lens RF level was set to 30. Precursor ions in the inclusion list were isolated within a retention time window of ± 60 s, filtered in the quadrupole at the isolation window (target m/z ± 0.8 m/z), and fragmented in an HCD collision cell C-trap at a resolution of 17,500 FWHM (full width at half maximum). The AGC target value was 1×10^6 , and the maximum injection time was 50 ms.

The normalized collision energy (NCE) was optimized for each compound. Details of the precursor and daughter ions monitored, retention times, and NCE values can be found in Table S1. The precision and calibration of the Q Exactive Orbitrap LC/MS/MS instrument were assessed using a reference standard mixture provided by Thermo Fisher Scientific. The measurements were performed in triplicate, and the data were evaluated using Quan/Qual Browser Xcalibur software, version 4.0.

Determination of the phenolic compound concentration in quinoa samples

The identification of phenolic compounds in the quinoa samples relied on their retention times compared to

authentic standards and the analysis of mass spectral data obtained through LC–MS/MS. Accurate mass determination was employed to generate elemental compositions and fragmentation patterns of the molecular ions. Quantification was done on the basis of the transition from precursor ion $[M + H]^+$ to corresponding quantification ion (Table S1). Calibration curves were then established by plotting the peak area, adjusted with probenecid as internal standard, against the concentration of the corresponding reference standards.

Chemical analyses

All three quinoa samples were also investigated for the effects of germination on the protein content (CP), total phenolic content (TPC), and antioxidant activity (AA). The CP content of each sample was measured using the classical Kjeldahl mineralization method and calculated using a conversion factor of 6.25 [18]. For this method, 1 g of milled sample was utilized. The TPC was determined using Folin-Ciocalteau (FC) reagent with slight modifications based on the method [19]. The FC method employed 2 g of sample. The TPC results were expressed as grams of gallic acid equivalent (GAE) per kilogram of sample dw (GAE g/ kg dw). The AA of the samples was assessed using a DPPH assay [20], utilized 1 g of milled sample in this study. The results of the DPPH assay were expressed as millimoles of Trolox equivalent (TE) per gram of sample dw (µmol TE/g dw). Two replicates were performed for each protein content, TPC, and AA measurement.

Statistical analyses

Three biological replicates were measured for descriptors of interest. Statistical analysis was conducted using the R program [21]. Means and standard deviations were calculated for each sample type and processing method in individual traits. One-way analysis of variance (ANOVA) was performed to determine whether there was significant effect of preparation method or sample type on evaluated traits. For germination data, the method was also applied to confirm if there is significant difference between three evaluated cultivars. Tukey's honestly significant difference (HSD) test was employed to identify processing methods and its variants with significantly different means. To explore the association among samples, a principal component analysis (PCA) was conducted using scaled data for a set of 14 descriptors. Quality of representation of variables on the factor map was also assessed for the first two components with the largest variance. The routines within FactoMineR [22] and factoextra packages [23] were used for this task and to visualize PCA results.

Results and discussion

The effect of germination

The presented study investigated the effects of germination on the CP, TPC, AA, and the quantity of twelve metabolites in two quinoa genotypes 'Faro' and 'Besançon' and one commercial sample Probio (Fig. 2). Throughout the germination process, slight, but statistically significant differences (at p < 0.05) were reported for CP depending on the sample and the duration of germination. The major increase in CP was observed in the Probio sample on the fourth day of germination reaching $17.18 \pm 0.01\%$ dw in comparison to control $(15.47 \pm 0.21\% \text{ dw})$. In 'Faro' and 'Besançon', the peak CP values were indicated on the third day of germination reaching $13.93 \pm 0.02\%$ and $15.58 \pm 0.15\%$ dw, respectively. The increase in CP has already been recognized not only in quinoa, but also in other related species from family Amaranthaceae [14]. This increase could be attributed to the enhanced enzymatic activity of α -amylase and the release of packed proteins from starch granules [24] or synthesis of proteins de novo [14]. Further, loss of total dry weight of the seeds due to the respiration during germination may increase the CP percentage [14].

Germination resulted as the superior technique for the enhanced AA in quinoa among other processing methods examined in this study. Although the initial days of germination resulted in a decline in AA values for the 'Besançon' and 'Faro' by almost 30%, the subsequent days of germination stimulated a continuous increase in this trait. The samples presented the peak values in AA during the third, fourth, and fifth day of germination in 'Faro', 'Besançon', and Probio, respectively. It was previously stated that the increase of AA is likely due to the elevated activity of hydrolytic enzymes and/or biosynthesis of antioxidants of low molecular weight [7], however, the level of increment is influenced by germination conditions [7]. Besides, our results indicated that differences in AA increment were related to the studied sample/genotype, confirming the earlier reported research carried out on white and red quinoa [25]. In our case, the most promising sample was Probio, since it did not show any remarkable drop in the beginning of germination, and it further reached the highest AA values on the fourth day of germination among other studied samples.

Significant variations (p < 0.05) in TPC were indicated among the quinoa samples and germination days. The highest TPC was recorded for the 'Besançon' (25.77 ± 0.15 GAE g/kg dw) on the second day of germination, which is aligns with the findings of Guardianelli et al. [25], but conflicting with Bhinder et al. [25], who recognized



Fig. 2 The effect of germination time on selected nutritional parameters of three quinoa samples. Significant differences in means among control (C) and days of germination [1 day (G1D), 2 days (G2D), 3 days (G3D), 4 days (G4D), and 5 days (G5D)], are denoted by the different letters (Tukey HSD) above each column. Letters a-c indicates statistical differences within treatments, while letters A-E denotes statistical differences among treatments for individual cul-

tivars. The abbreviations for the selected descriptors are as follows: gallic acid (GA), 4-hydroxybenzaldehyde (C4B), caffeic acid (CFA), quercetin-3-O-glucoronide (Q3G), isoquercetin (IQ), rutin (RUT), salicylic acid (SA), quercetin (QCE), naringenin (NAR), kaempferol (KMP), pinocembrin (PC), isorhamnetin (IH), crude protein content (CP), antioxidant activity (AA), total phenolic content (TPC)

the peak values during the third and fourth day of germination. Detected contradictions may be attributed to the dynamic chemical changes during the germination including compound synthesis, release from bound form, or consumption [26]. In addition, specific germination conditions should be taken into consideration as factor influencing the TPC during germination [27]. As opposed to 'Besançon', 'Faro,' and Probio samples showcased their highest TPC values in the non-germinated state $(23.71 \pm 0.08 \text{ and } 22.46 \pm 0.88 \text{ GAE g/kg dw}$, respectively). Different rates of polyphenol accumulation in two different quinoa samples were published formerly [25], indicating the importance of proper genotype selection for obtaining the optimal quantity of phenolic compounds during germination.

The content of twelve studied metabolites as determined by UHPLC-ESI–MS/MS analysis is given in Fig. 2. The metabolite with the highest mean concentration in nongerminated samples was quercetin 3-O-glucuronide (Q3G), also known as miquelianin, whereas rutin (RUT) was the most abundant compound in germinated samples. RUT demonstrated an increasing accumulation with extended germination time, similar to what was presented in the study of Al-Qabba et al. [28] and Bhinder et al. [28]. The peak values of RUT content were recognized between the fourth and fifth day of germination but the degree of the increment varied among the samples. In our case, the most prominent increment in RUT content was reported for genotype 'Faro'.

As mentioned in the beginning, Q3G was the most abundant metabolite in non-germinated quinoa seeds, which is in agreement with Dostalikova et al. [29]. This metabolite has been primarily detected in aerial plant parts in various plant species [30-32], but research quantifying the content of Q3G in seeds is insufficient. During the germination, Q3G showcased an opposite pattern as RUT with 80% decline in the initial days of germination in 'Besançon' and 'Faro'. Contradictory results were published by Pilco-Quesada et al. [16] demonstrating a significant growth in the content of Q3G after 72 h of quinoa germination.

The isoquercetin (IQ) followed the same trend as discussed here in the case of Q3G. The drop in values was also noticed for salicylic acid (SA) and 4-hydroxybenzaldehyde (4BA) after first day of germination. Gallic acid (GA), naringenin (NAR), and caffeic acid (CFA) were presented in quinoa samples in a relatively trace concentrations, with respect to other studied compounds. The germination process improved their content, especially during the first 3 days of germination. To the best of our knowledge, NAR, SA, 4BA, and IQ have not been formerly quantified in germinated quinoa.

A small amount of kaempferol (KMP), pinocembrin (PC), quercetin (QCE), and isorhamnetin (IH) was detected in non-germinated samples. These metabolites were

rapidly synthetized during the fifth day of the germination process, but the degree of increment varied among the studied genotypes. Besides, the mean PC content was the highest in germinated quinoa contrasting to raw and heat-treated samples. While the increase in KMP and QCE concentrations during germination has been already published for quinoa [7, 28], it was not as prominent as observed in our study. To our knowledge, the presence of PC and IH in germinated quinoa has not been evaluated before, nonetheless, they have already been described in sprouted mung bean [33] and buckwheat [34].

Overall, the germination process led to the enhancement of several bioactive compounds, including GA, CFA, RUT, QCE, NAR, KMP, PC, and IH in comparison to the control sample, suggesting their potential role in the germination process. The most substantial increase in the content of these metabolites was reported between the third and fifth days of germination. Conversely, germination initiated a decline in the levels of 4BA, Q3G, IQ, and SA. The alterations in metabolite quantity occurred in genotype-dependent manner, with 'Besançon' and 'Faro' exhibiting the most intense synthesis of metabolites during germination. On the other hand, the changes in chemical content of the Probio sample were less prominent.

It has been suggested previously that various metabolic and enzymatic events occurring during germination may synthesize or consume the phenolic compounds, thus elevating or decreasing their overall content. In addition, those compounds play a non-negligible role in protection against free radicals generated during the germination process [26]. However, other factors like genotype, agronomic conditions, maturity level at harvest and postharvest storage conditions may considerably contribute to the variations in the polyphenol content of germinated quinoa [26].

Seed soaking

Quinoa seeds are prized for their superior nutritional quality, especially their high content of proteins and bioactive chemicals [35]. While this area has been researched extensively, most of the studies examined only raw materials, which might not give a full picture of quinoa potential and health benefits. Therefore, this paper evaluated the effect of various processing methods and processing time on the CP, TPC, AA (Fig. 3) and the content of selected bioactive compounds (Table 1) of Probio sample. Soaking was proven to be effective for minimizing the content of anti-nutritional compounds [36]. However, our results indicated that soaking in water worsened the majority of studied nutritional parameters. The exceptions were metabolites KMP, NAR, PC, and RUT where soaking led to a rise in their content. Presented alterations after soaking might be related to various factors.



Fig. 3 The effect of various processing methods and processing time on crude protein content (CP), antioxidant activity (AA), and total phenolic content (TPC) of commercial Probio sample. Significant differences in means (Tukey HSD) within individual treatments are indicated by different letters (a-c) above each column. The letters (A-E)

in the header of plot shows the difference among individual treatments. Dashed red line within each plot denotes overall mean of data for respective variable. The abbreviations for the selected processing methods are as follows: control (C), SK (soaking), B (boiling), B NaCl (boiling in NaCl) (min)

Table 1	The effect of various thermal	processing on the content	t of selected quinoa metabolites*
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	4BA	CFA	GA	IH	IQ	КМР
	$3.104 \pm$	$0.217 \pm$		$0.437 \pm$	1.593 ±	$0.27 \pm$
Control	0.119 D	0.007F	0.001 ± 0 B	0.013 BCD	0.044E	0.02 BF
	$0.326 \pm$	$0.147 \pm$		$0.48 \pm$	$0.579 \pm$	$1.293 \pm$
Soaking	0.01AC	0.003A	n.d.	0.012 BCD	0.02 AB	0.082AE
U	$0.36 \pm$	$0.102 \pm$		$0.114 \pm$	$0.632 \pm$	$0.751 \pm$
Boiling 5 (min)	0.068 bA	0.011 b E	n.d.	0.004 bBD	0.078 aA	0.023 bEF
8 ()	$0.269 \pm$	$0.09 \pm$		$0.873 \pm$	$0.536 \pm$	$0.691 \pm$
Boiling 10 (min)	0.028 bA	0.004 <mark>ab</mark> E	n.d.	0.041 cBD	0.02 aA	0.027 bEF
6 ()	$0.407 \pm$	$0.088 \pm$		$0.575 \pm$	$0.533 \pm$	$1.068 \pm$
Boiling 15 (min)	0.049 <mark>aA</mark>	0.008 <mark>a</mark> E	n.d.	0.011 aBD	0.041 aA	0.123 aEF
Boiling NaCl 5	$0.27 \pm$	$0.116 \pm$	n.d.	$0.055 \pm$	$0.57 \pm$	$1.311 \pm$
(min)	0.028 <mark>aA</mark>	0.004 bA		0.002 bA	0.017 aAB	0.024 bA
Boiling NaCl 10	$0.338 \pm$	$0.129 \pm$	n.d.	$0.079 \pm$	$0.62 \pm$	$1.687 \pm$
(min)	0.053 aA	0.004 cA		0.003 cA	0.008 bAB	0.061 aA
Boiling NaCl 15	$0.314 \pm$	$0.105 \pm$	n.d.	$0.05 \pm$	$0.611 \pm$	$1.649 \pm$
(min)	0.028 aA	0.002 aA		0.007 aA	0.046 aAB	0.063 aA
	$0.931 \pm$	$0.022 \pm$	n.d.	$0.026 \pm$	$0.554 \pm$	$0.153 \pm$
Flakes 1 (min)	0.034 dB	0.001eB		0.002eA	0.013dC	0.003eB
	$1.818 \pm$	$0.041 \pm$		$0.091 \pm$	$1.031 \pm$	$0.412 \pm$
Flakes 2 (min)	0.029 <mark>aB</mark>	0.001 <mark>aB</mark>	n.d.	0.006 <mark>aA</mark>	0.029 <mark>aC</mark>	0.015 <mark>aB</mark>
	$1.674 \pm$	$0.076 \pm$		$0.08 \pm$	$0.875 \pm$	$0.328 \pm$
Flakes 3 (min)	0.047 bB	0.003 bB	n.d.	0.003 bA	0.021 bC	0.021 bB
	$1.451 \pm$	$0.06 \pm$		$0.052 \pm$	$0.655 \pm$	$0.218 \pm$
Flakes 4 (min)	0.107 cB	0.003 cB	n.d.	0.008 cA	0.025 cC	0.032 cB
	$1.366 \pm$	$0.071 \pm$		$0.055 \pm$	$0.7 \pm$	$0.268 \pm$
Flakes 5 (min)	0.063 cB	0.002 dB	n.d.	0.002 dA	0.047 <mark>cC</mark>	0.028 dB
Microwave 1	$0.431 \pm$	$0.276 \pm$	$0.002 \pm$	$0.325 \pm$	$0.869 \pm$	$4.396 \pm$
(min)	0.046 aAC	0.005 <mark>aC</mark>	0.001 abA	0.008 cB	0.023cD	0.132 cC
Microwave 2	$0.412 \pm$	$0.271 \pm$		$0.48 \pm$	$0.825 \pm$	$3.277 \pm$
(min)	0.05 <mark>aAC</mark>	0.014 <mark>aC</mark>	0.001 ± 0 aA	0.055 <mark>aB</mark>	0.022 <mark>a</mark> D	0.097 <mark>aC</mark>
Microwave 3	$0.491 \pm$	$0.361 \pm$	$0.003 \pm$	$0.63 \pm$	$1.007 \pm$	$5.876 \pm$
(min)	0.047 <mark>bAC</mark>	0.014 bC	0.001 bA	0.031 bB	0.023 bD	0.115 bC
Steaming 5	$0.458 \pm$	$0.11 \pm$		$0.179 \pm$	$0.718 \pm$	$1.145 \pm$
(min)	0.016 bC	0.003 bA	n.d.	0.009bD	0.02 bBC	0.051 bA
Steaming 10	$0.312 \pm$	$0.091 \pm$		$0.478 \pm$	$0.628 \pm$	1.928 ±
(min)	0.029 cC	0.004 cA	n.d.	0.066cD	0.036 cBC	0.119 cA
Steaming 15	$0.696 \pm$	$0.181 \pm$		$1.629 \pm$	$0.762 \pm$	$2.401 \pm$
(min)	0.03/aC	0.005 aA	n.d.	0.038aD	0.022 aBC	0.05 aA
	$0.35/\pm$	$0.145 \pm$	0.001 ± 0.0	$0.026 \pm$	$0.6/2 \pm$	$1.859 \pm$
Roasting 5 (min)	0.0260	0.00300	0.001 ± 0.001	0.002DAC	0.03500	0.068DD
(min)	$0.391 \pm$	$0.1 / / \pm$	$0.001 \pm$	$0.390 \pm$	$0.797 \pm$	$2.339 \pm$
(IIIII) Deacting 15	0.003	0.00400	0.001	0.015CAC	1.206	
(min)	$0.710 \pm$	$0.213 \pm$	$0.001 \pm$	$0.137 \pm$	$1.290 \pm$	0.261 pD
(IIIII)		0.0084D	0.0014D	0.0554AC	DUT	0.2014
			<u>Q3G</u>	QCE	12.00 L	SA
Control	$0.04 \pm$	0.098 ±	$23.03/\pm$	$0.48/\pm$	$12.09 \pm$	$0.999 \pm$
Control	0.003	0.002	1.200D	0.028AB	0.21/ Br	0.077
Soaking	$0.097 \pm 0.007 \mathbf{P}$	$0.642 \pm$	$7.137 \pm$	$0.743 \pm$	$50.118 \pm$	$1.023 \pm$
SUAKIIIg	0.007D	0.023 L	$7517 \pm$	0.03 AD $0.550 \pm$	0.005D 11 188 ±	0.034 DC
Poiling 5 (min)	$0.039 \pm$	$0.234 \pm$	$7.317 \pm 1.411 \text{hC}$	$0.339 \pm$	11.100 ± 1.124	$0.374 \pm$
Lining 5 (iiiiii)	$0.002a$ \square	0.0220r	1.41100 $1.92 \pm$	0.032a0AD	0.1240A	0.0350D
Boiling 10 (min)	$0.019 \pm$	$0.22 \pm$	$-7.903 \pm$	$0.325 \pm$	$7.401 \pm$	$0.330 \pm$
Doming TO (mm)	0.00301	0.02301	5.207 a	0.03 + 0.04	0.172 40/1	$0.368 \pm$
Boiling 15 (min)	$0.0+3 \pm$	0.271 ± 0.007	$0.220 \pm$	$0.02 \pm$	9.04 ± 0.61	$0.000 \pm$
Boiling NaCl 5	0.000an	0.375 +	$7.673 \pm$	0.000 AAD	7.07 ± 0.01 aA	$0.795 \pm$
(min)	$0.070 \pm$	0.014hA	0 53164	0.75 + 1	0.346hAR	0.044h
(mm)	0.0000/1	0.0140/1	0.3310/1	0.0 <i>32<mark>a</mark>m</i>	U.JTUUAD	0.0 + 10

Table 1 (continued)

Boiling NaCl 10	$0.074 \pm$	$0.375 \pm$	$8.755 \pm$	$1.789 \pm$	$11.712 \pm$	$0.727 \pm$
(min)	0.004 bA	0.018 bA	0.176 cA	0.279 bA	0.305cAB	0.026 cA
Boiling NaCl 15	$0.062 \pm$	$0.347 \pm$	$7.866 \pm$	$1.166 \pm$	$10.381 \pm$	$0.526 \pm$
(min)	0.003 aA	0.011 aA	0.088 <mark>aA</mark>	0.118 <mark>aA</mark>	0.63 aAB	0.021 aA
()	$0.053 \pm$	$0.143 \pm$	$11.364 \pm$	$0.06 \pm$	$5.347 \pm$	$0.776 \pm$
Flakes 1 (min)	0.004 dBC	0.013 bB	0.311 dB	0.007 dB	0.154cC	0.03 bA
	$0.095 \pm$	$0.217 \pm$	$19.24 \pm$	$0.265 \pm$	$9.17 \pm$	$1.192 \pm$
Flakes 2 (min)	0.006 <mark>aBC</mark>	0.005 aB	0.467 <mark>aB</mark>	0.018 <mark>aB</mark>	0.174 <mark>aC</mark>	0.055 aA
	$0.091 \pm$	$0.132 \pm$	$17.799 \pm$	$0.282 \pm$	$7.268 \pm$	$0.728 \pm$
Flakes 3 (min)	0.007 aBC	0.005 bB	0.373 bB	0.02 <mark>aB</mark>	0.086 <mark>bC</mark>	0.027 bA
	$0.076 \pm$	$0.111 \pm$	$12.854 \pm$	$0.184 \pm$	$5.35 \pm$	$0.493 \pm$
Flakes 4 (min)	0.002 bBC	0.02 cB	0.378 cB	0.049 bB	0.217cC	0.013 cA
	$0.112 \pm$	$0.079 \pm$	$13.17 \pm$	$0.197 \pm$	$5.458 \pm$	$0.507 \pm$
Flakes 5 (min)	0.008 cBC	0.004 dB	0.65 cB	0.007 cB	0.261 cC	0.026 cA
Microwave 1	$0.141 \pm$	$0.517 \pm$	$13.011 \pm$		$30.746 \pm$	$1.138 \pm$
(min)	0.006 bD	0.018 <mark>aC</mark>	0.346 cB	5.512 ± 0.2 cC	1.129cD	0.032 cB
Microwave 2	$0.133 \pm$	$0.526 \pm$	$10.654 \pm$	$3.898 \pm$	$27.672 \pm$	$1.046 \pm$
(min)	0.007 <mark>aD</mark>	0.018 <mark>aC</mark>	0.659 aB	0.107 <mark>aC</mark>	0.841 <mark>aD</mark>	0.018 aB
Microwave 3	$0.139 \pm$	$0.503 \pm$	$18.169 \pm$	$8.292 \pm$	$36.635 \pm$	$1.351 \pm$
(min)	0.003 bD	0.013 <mark>aC</mark>	0.334 bB	0.297 <mark>bC</mark>	1.001 bD	0.076 bB
Steaming 5	$0.064 \pm$	$0.492 \pm$	$9.824 \pm$	$0.543 \pm$	$14.539 \pm$	$1.101 \pm$
(min)	0.006 bAC	0.024 bC	0.491 aA	0.041 bA	0.64 <mark>aEF</mark>	0.039 <mark>aBC</mark>
Steaming 10	$0.08 \pm$	$0.609 \pm$	$8.828 \pm$	$0.574 \pm$	$14.254 \pm$	$1.064 \pm$
(min)	0.003 cAC	0.017 <mark>cC</mark>	0.958 bA	0.028 bA	0.677 aEF	0.017 aBC
Steaming 15	$0.092 \pm$	$0.45 \pm$	$10.613 \pm$	$1.973 \pm$	$14.256 \pm$	$1.092 \pm$
(min)	0.005 <mark>aAC</mark>	0.014 <mark>aC</mark>	0.452 aA	0.081 aA	0.518 aEF	0.035 aBC
	$0.098 \pm$	$0.453 \pm$	$7.785 \pm$	$1.61 \pm$	$13.354 \pm$	$0.948 \pm$
Roasting 5 (min)	0.009 <mark>aB</mark>	0.029 bD	0.305 bA	0.051 bD	0.537 b E	0.025 bC
Roasting 10	$0.085 \pm$	$0.443 \pm$	$8.861 \pm$	$2.834 \pm$	$14.103 \pm$	$0.894 \pm$
(min)	0.005 <mark>aB</mark>	0.015 abD	0.512 cA	0.154 cD	0.842 b E	0.019 <mark>cC</mark>
Roasting 15	$0.096 \pm$	$0.464 \pm$	$9.88 \pm$	$5.587 \pm$	$19.98 \pm$	$1.166 \pm$
(min)	0.008 <mark>aB</mark>	0.013 aD	0.443 <mark>aA</mark>	0.383 aD	1.122 <mark>a</mark> E	0.035 <mark>aC</mark>

*Results are expressed as mean ± standard deviation. n.d.-not detected

Different lowercase letters a-g indicate statistical differences by Tukey's test (p < 0.05) within different treatments

Different uppercase letters A–F indicate statistical differences by Tukey's test (p < 0.05) among different treatments

The abbreviations for the selected descriptors are as follows: 4-hydroxybenzaldehyde (4BA), caffeic acid (CFA), gallic acid (GA), isorhamnetin (IH), isoquercetin (IQ), kaempferol (KMP), naringenin (NAR), pinocembrin (PC), quercetin-3-O-glucoronide (Q3G), quercetin (QCE), rutin (RUT), salicylic acid (SA)

The reduction in CP was attributed to a leaching of quinoa seed storage proteins into soaking water [14]. Similarly, a softening of cell-wall tissues could potentially facilitate the increased release of polyphenols into the soaking medium [37], thus possibly reducing the TPC and AA of soaked seeds. In addition, the variability in the metabolite content could be ascribed to the commencement of the seed germination processes, as discussed above.

Boiling in plain water and NaCl solution

When comparing two boiling solutions, boiling in plain water showed slightly better results in CP content and the AA, principally after 15 min of treatment, with respect to the boiling in NaCl. The values reported after 15 min of boiling in plain water were relatively similar to the control sample in both parameters. It was previously stated that no significant alterations in CP occurred after 15 min of boiling [6].Nonetheless, the AA could be, in part, influenced by thermally induced modifications in the phenolic content and composition [38].

Overall, boiling in both solutions was evaluated as the least suited method for the TPC enhancement among other examined processing methods. In addition, the TPC was reduced more intensively during the boiling in plain water than in boiling in NaCl solution (Fig. 3). The TPC might be influenced by various factors. Apart from sample variety, processing conditions, and method of analysis, the reasons for the decline in TPC might include the polyphenol leaching into boiling water and their thermal degradation. Conversely, the release of polyphenols and the inactivation of phenol oxidase may contribute to their propensity, as reported previously in buckwheat [37].

This was also reflected in the distinct behaviours of metabolite content during boiling. Boiling in plain water was not suited for the enhancement of 4BA, CFA, IQ, Q3G, and SA (Table 1), but it improved the content of KMP and PC, compared to control. Boiling in NaCl was considered more beneficial in contrast to boiling in plain water, since most of the studied metabolites reached higher mean values in their content. The presence of salt in the solution could possibly increase the boiling point and therefore induce a higher degree of thermal dissociation of bound molecules, as proposed for pulses [39].

Flaking

Flakes from whole quinoa seeds demonstrated a noteworthy reduction in the required boiling time, reducing it to a mere 5 min, with respect to the boiling of whole seeds. Therefore, further utilisation of quinoa flakes could be potentially advantageous in mitigating the heat-induced degradation of thermally unstable compounds. It is noteworthy that research in this specific domain for quinoa remains scarce. The shorter cooking time of quinoa flakes enhanced the TPC and the content of IO, NAR, and SA, in contrast to the boiled seeds. Moreover, among all the treatments explored in this study (Table 1), boiled flakes exhibited the highest mean content of 4BA and Q3G. In addition, boiled flakes achieved the highest CP values when compared to all the other heatutilizing methods investigated herein. Nonetheless, the mean AA of boiled flakes was lower than the values of boiled seeds. While the precise impact of boiling of quinoa flakes on the final nutritional quality has not been studied yet, it was previously concluded that flaking of ancient cereals and legumes may increase or decrease the TPC and AA depending on the type of sample. Contrary, the protein content was not significantly affected by flaking [40].

Microwaving

Microwaving was a relatively suitable method for enhancing the protein content since the mean CP of microwaved samples was the second highest among other studied treatments $(15.56 \pm 0.10\%)$. Furthermore, CP remained unaffected by varying microwave exposure times (Fig. 3). There is a lack of comprehensive studies elucidating the impact of microwaving on quinoa CP, however, studies conducted on other species, such as legumes and buckwheat, indicated quite variable outcomes in this area [41, 42]. While the mean AA values were statistically comparable to roasting and boiling in NaCl, the TPC values for microwaving were outstanding, reaching the peak in the third minute $(27.15 \pm 0.82 \text{ GAE g/}$ kg dw). In a parallel study, a similar reduction in AA with increasing time of processing was noticed, nevertheless, the highest TPC was detected after 5 min of microwaving [43]. Half of the studied metabolites, namely CFA, KMP, NAR, QCE, RUT, and SA, displayed the highest mean content during microwaving (Table 1), in comparison to other heat treatments and raw samples. This observation aligns with the PCA analysis results (Fig. 4), where microwave-treated samples distinctly cluster along the first principal component axis, revealing a strong influence from the mentioned traits. Including all heat treatments, GA was only found in microwaved and roasted samples. As concluded by Drulyte and Orlien [44], the heating effect of microwaving is more intense and faster than alternative cooking methods. This distinctive trait leads to a reduction in overall processing time and, notably, correlates positively with diminished losses of polyphenolic compounds [38]. Our results confirmed the conclusions of other studies that microwaving yields the highest number of polyphenols among other heat treatments, thereby increasing the overall antioxidant capacity [39, 43, 45, 46].

Steaming

Steaming is, together with boiling, a commonly employed methods of quinoa preparation. Even though both processing methods generally led to a reduction in the content of biologically active compounds compared to raw sample [47], steaming is recommended as more optimal method for better nutrient retention over boiling [6]. This preference stems from the fact that, during steaming, the quinoa seeds are not in direct contact with boiling water, thereby minimizing nutrient leaching into the water [8]. Our results confirmed this statement since the TPC in steamed sample was higher than in boiled samples. In addition, steaming did not affect protein content in quinoa seeds, aligning with previously published research [6, 48] although in contrast with Motta et al. [49], who reported a significant decrease in CP in studied pseudocereals (Amaranthus sp., quinoa and buckwheat). In terms of studied metabolites, their quantity was either comparable or lower than those observed in other heat treatments, except for IH, reaching the highest value in this study $(1.60 \pm 0.04 \,\mu\text{g/g dw})$ after 15 min of steaming (Table 1).

Roasting

Roasted quinoa seed did not reach any outstanding values for the content of protein and AA since both parameters were statistically comparable to boiling in plain water and boiling in NaCl, respectively (Fig. 3). Nevertheless, roasted seeds exhibited a great content of total polyphenols, reaching values comparable to the control after 15 min of roasting. The overall increment in TPC during roasting



Fig. 4 Principal component analysis biplot based on scaled data for set of 14 descriptors and 22 different culinary treatments. Two main components explaining 43.8 and 24.1% of total variance, respectively, are displayed. Individual points in plot stands for individual culinary treatments, highlighted by different colors and variants of those treatments, representing treatments duration in minutes (m.). The arrows within plot shows the quality of representation of individual descriptors on factor map and its contribution to first two axis. The abbre-

can be attributed to the release of bound chemicals due to heat and the formation of Maillard reaction products, but the yield of phenolics is also influenced by the roasting temperature and time used during processing [11]. This might explain the contradicting results of some studies, showing the TPC and AA of roasted seed with values even higher than control sample [43, 46] and others with significantly reduced polyphenolic content [8]. In case of metabolite content, roasting was a suitable technique for the enhancement of IQ, KMP, SA, RUT, and QCE, especially after 15 min of roasting time (Table 1). Similar metabolites were investigated previously in amaranth [50], nonetheless, the pattern of the changes during roasting was distinct from our results. For example, QCE and KMP significantly decreased after 15 min of roasting, whereas GA and CFA increased rapidly.

viations for the selected processing methods and descriptors are as follows: control (C), SK (soaking), B (boiling), B NaCl (boiling in NaCl), gallic acid (GA), 4-hydroxybenzaldehyde (4BA), caffeic acid (CFA), quercetin-3-O-glucoronide (Q3G), isoquercetin (IQ), rutin (RUT), salicylic acid (SA), quercetin (QCE), naringenin (NAR), kaempferol (KMP), pinocembrin (PC), isorhamnetin (IH), crude protein content (CP), antioxidant activity (AA), total phenolic content (TPC)

Conclusion

The present investigation was conducted to assess the impact of germination, soaking, boiling, flaking, microwaving, steaming, and roasting on the selected nutritional characteristics of quinoa. The quantitative analysis of 12 bioactive compounds was conducted in three distinct quinoa samples during a 5-day germination period. In all studied samples, GA, CFA, RUT, QCE, NAR, KMP, IH, and PC were enhanced compared to control, but the level of increment was contingent upon the type of sample. This underscores the importance of proper selection of genotype for optimum content of biologically active compounds in germinated quinoa. The most substantial increase in bioactive compounds was noticed between the third and fifth day of germination with the highest accumulation of metabolites occurring in the genotypes 'Besançon' and 'Faro'. Six compounds (NAR, SA, 4BA, IQ, PC, IH) were detected in germinated quinoa for the first time.

This study further examined a range of various heatutilizing methods. Statistically significant differences were observed in CP among heat treatments. Boiled quinoa flakes exhibited the highest average protein content and proved to be a time-efficient preparation method due to reduced boiling time. The lowest mean values of CP were associated with roasting and boiling. Most of the heat treatments caused a decrease in TPC and AA in comparison to raw sample. An exception to this trend was microwaving which strongly enhanced the overall TPC of quinoa sample and the content of several metabolites (CFA, KMP, NAR, QCE, RUT, and SA).

It can be concluded that different processing methods influenced the nutritional content and composition of quinoa differently. The specific effects varied depending on the processing technique, duration of treatment, compound measured, and the genotype. Nonetheless, further research is warranted to elucidate the underlying mechanisms driving these changes. The alterations observed in this study emphasize the importance of considering those variables in optimizing the processing methods used for quinoa to obtain the best nutritional profile of final food product. Therefore, this knowledge contributes to the development of processing techniques that preserve or enhance the nutritional value of quinoa and promote its utilization as a source of health-promoting compounds in human diets.

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Author contributions Conceptualization: PHČ; Methodology: PHČ, MJ, LD; Formal analysis and investigation: LD, MJ, PHČ; Data curation: PS; Visualization: PS; Writing—original draft preparation: LD; Writing—review and editing: PHČ, IV; Funding acquisition: DJ, VD; Resources: DJ; Supervision: PHČ, IV.

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Data availability All of the data have been listened in manuscript and Supplementary material.

Declarations

Conflict of interest The authors declare no conflict of interest.

Ethics approval This article does not contain any studies with human or animal subjects.

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