

Solution to trifluoroacetic acid NMR challenge

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Accepted: 10 May 2021 / Published online: 5 July 2021
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The winner of the trifluoroacetic acid NMR challenge (published in volume 413 issue 1) is:

Irina Ihnatenko, Technische Universität Braunschweig, Germany.

The award entitles the winner to select a Springer book of their choice up to a value of €100,-.

Our Congratulations!

In a previous Analytical Challenge about phosphine [1, 2], we saw that isotopic labeling brings a new level of complexity to NMR spectra of simple substances. The trifluoroacetic acid NMR challenge [3] invites readers to further explore this aspect of analytical chemistry. To understand the ¹⁹F-NMR spectrum of carbon-13 labeled trifluoroacetic acid (TFA), one has to consider all the interactions that fluorine-19 atoms can have with carbon-13 atoms, both of which have a non-zero nuclear spin, $I = 1/2$.

In TFA having only carbon-12 atoms, F₃[¹²C][¹²C]OOH, the three equivalent fluorine atoms have no neighboring atoms of non-zero nuclear spin. Thus, such a molecule will display a

single resonance with chemical shift in the vicinity of -76 ppm (relative to CFCl₃). In the singly labeled TFA, F₃[¹³C][¹²C]OOH, however, fluorine atoms will interact with the single adjacent carbon-13 giving rise to a doublet with a wide coupling constant (¹J_{FC} = 285 Hz) as shown in Fig. 1. Similarly for the other singly labeled TFA, F₃[¹²C][¹³C]OOH, fluorine atoms will interact with carbon-13 two bonds away, leading to a doublet albeit with much smaller coupling constant (²J_{FC} = 45 Hz) [4]. These two NMR spectra are depicted in Fig. 2.

Finally, in the doubly labeled TFA, F₃[¹³C][¹³C]OOH, fluorine atoms will interact with both carbon-13 atoms. First, the three equivalent fluorine atoms are coupling with the adjacent carbon-13 atom giving rise to a wide doublet (¹J_{FC}). But, in addition, this doublet is further split into doublets because the same fluorine atoms also interact with the distant carbon-13 atom (²J_{FC}) [5]. The result of this double splitting leads to a doublet of doublets with a wide first split and a much narrower second split as shown in Fig. 2 (corresponding to Fig. 1A of the trifluoroacetic acid NMR challenge [3]).

This article is the solution to the Analytical Challenge to be found at <https://doi.org/10.1007/s00216-020-02867-3>

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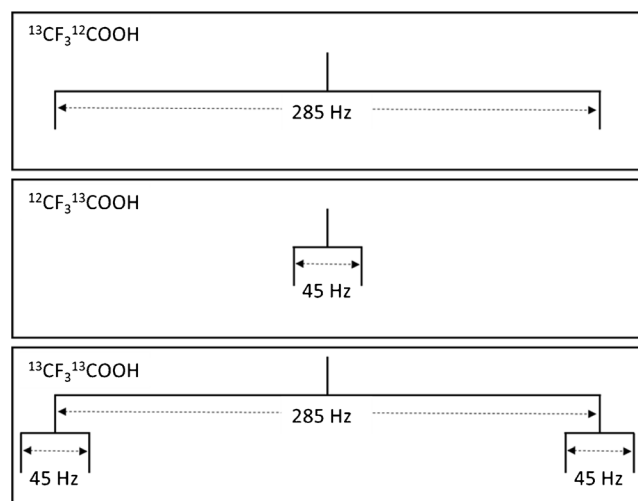


Fig. 1 Signal splitting patterns illustrating the ^{19}F -NMR spectra of carbon-13 labeled trifluoroacetic acids

Declarations

Conflict of interest The authors declare no competing interests.

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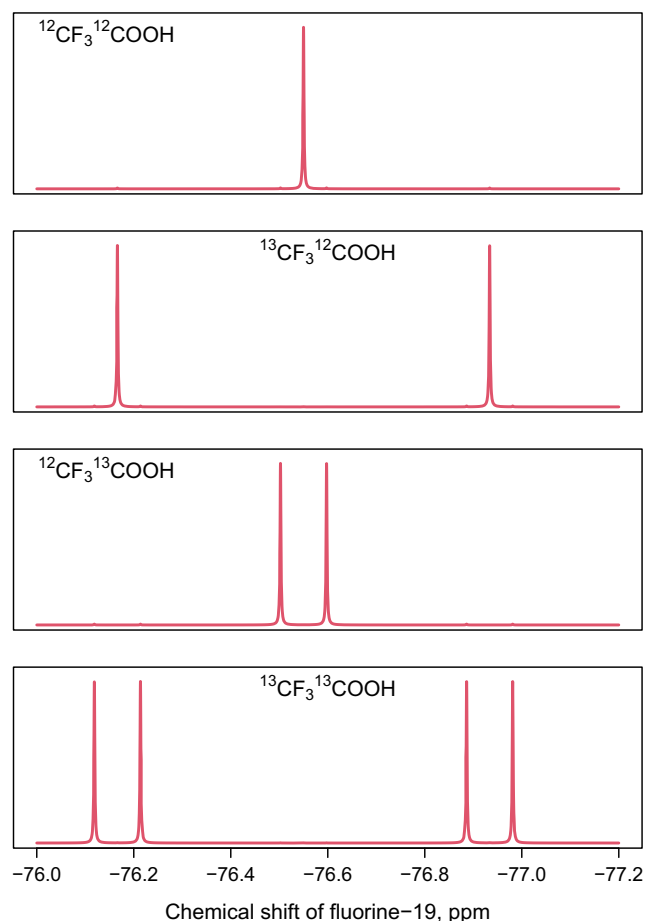


Fig. 2 ^{19}F -NMR spectra of carbon-13 labeled trifluoroacetic acids