

ERRATA

The paper by C. A. Johnson and Glasser, Vol. **51**, 1–8, contained some errors in Tables 1 and 4. For the conversion of dissolved Zn, Co and Ni concentrations from g L^{-1} to mol L^{-1} the molecular weight of Mg was used unintentionally. The changes that are caused to Fig. 3 are minimal and the interpretation of the results still holds true.

Table 1: Under K_{SO} values:

for 20.80 read 19.94

for 22.88 read 22.08

for 20.03 read 19.19.

Table 4: Under K_{SO} values:

from the first value for Zn (21.95) to the fourth value for Ni (20.34) inclusive, insert the following list of values.

Zn 21.09

20.23

19.44

18.99

Co 23.56

20.15

22.27

22.32

Ni 18.62

18.95

19.60

19.57

The paper by Y. Nakashima, Vol. **51**, 9–22, contained misprints in equations 10 and 11.

On page 14, equation 10 should read:

$$\phi_{\text{clay}} = \frac{w}{\left(\frac{\rho_{\text{clay}}}{\rho_{\text{water}}}\right) + \left[1 - \left(\frac{\rho_{\text{clay}}}{\rho_{\text{water}}}\right)\right] \frac{w}{100}} \quad (10)$$

and equation 11 should read:

$$\frac{D}{D_0} = 1 - 0.0157 \frac{\chi w}{\left(\frac{\rho_{\text{clay}}}{\rho_{\text{water}}}\right) + \left[1 - \left(\frac{\rho_{\text{clay}}}{\rho_{\text{water}}}\right)\right] \frac{w}{100}} \quad (11)$$