

to precipitation of the  $\beta'$  phase of the Cu-Be system and that the subsequent hardening is explained by slow precipitation of the  $\epsilon$  phase of the Cu-Sn system.

In view of the high cost of beryllium, the fact that approximately the same properties can be obtained in a ternary alloy containing 0.75 pct Be as in a binary alloy containing 2.0 pct Be is noteworthy, and there seems no reason to believe that the cheaper alloy might not function satisfactorily in some of the applications where the binary alloy is now used.

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### References

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### Technical Note

## Lattice Parameter of InSb

by T. S. Liu and E. A. Peretti

**A**LITERATURE survey of the In-Sb system, which was made before a study of the binary diagram was undertaken, revealed that the intermediate phase InSb existed and that its crystal structure is face-centered cubic of the zinc blend type. In 1929 Goldschmidt<sup>1</sup> determined the lattice constant for InSb and obtained a value of 6.452Å. A more recent value, 6.461Å was reported by Iandelli<sup>2</sup> in 1941. The purity of the indium used was not very high, however. The indium used by Iandelli contained 2 pct Ge. In view of this fact it was decided to remeasure the lattice parameter using high-purity indium (99.97+ pct In) and spectrographically standardized antimony of the highest purity available (99.91 pct Sb).

Specimens corresponding to the InSb composition were made by carefully weighing the ingredients on an analytical balance and heating them in vacuo in a quartz tube at 700°C. As melting losses were negligible, the alloys were not analyzed chemically, but the intended analysis was taken as being accu-

Table II. Lattice Constant Calculation for Focusing Back-Reflection Pattern Taken at 25°C

Plane	Sin <sup>2</sup> θ	h <sup>2</sup> + k <sup>2</sup> + l <sup>2</sup>	δ*
711	0.976348	51	5.3
551			
711	0.972111	51	6.4
551			
444	0.922410	48	17.4
444	0.917841	48	18.4
731	0.913711	59	19.2
553			
642	0.874521	56	27.5
533	0.820171	43	38.6
533	0.816538	43	39.3

Normal Equations

$$184596K\uparrow + 2571.3D = 1097.96^{\dagger}$$

$$25171.3K + 4868.44D = 149.578^{\dagger}$$

$$K = 0.00596105 \quad D = -0.00009037 \quad a_0 = 6.4760$$

\*  $\delta = 2\phi \sin^2\phi$ , where  $\phi = 90^\circ - \theta$ .

†  $K = \frac{1}{4}a_0^3$ ,  $\lambda$  is incorporated into the equations.

Values of  $\lambda$  are based on those adopted at the conference of the X-ray Analysis Group of the Institute of Physics, London (1946).<sup>4</sup>

rate. Powder samples were prepared by either filing or crushing of the brittle alloy, followed by screening at 325 mesh. The powders were then annealed at 400°C in evacuated capsules for about an hour.

X-ray photographs were taken with a Debye and a back-reflection focusing camera, using characteristic cobalt  $K_{\alpha}$  radiation. Lattice parameters were calculated by Cohen's<sup>3</sup> method. The results of a Debye pattern and of a back-reflection pattern at 25°C are given in Tables I and II.

The lattice parameter at 25°C is 6.4760Å.

### References

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Table I. Debye Pattern of InSb\*

θ	Sin θ	d	Observed In-tensity	hkl	λ
81.18	0.9881	0.9054	m	711	α <sub>2</sub>
				551	
80.50	0.9863	0.9050	vs	711	α <sub>1</sub>
				551	
73.55	0.9590	0.9328	w	444	α <sub>2</sub>
73.23	0.9574	0.9324	w	444	α <sub>1</sub>
69.43	0.9362	0.8638	w	642	β
65.33	0.9086	0.9846	vw	533	α <sub>2</sub>
65.10	0.9070	0.9842	m	533	α <sub>1</sub>
61.23	0.8765	1.021	vw	620	α <sub>2</sub>
61.05	0.8750	1.020	m	620	α <sub>1</sub>
55.08	0.8200	1.089	s	531	α
51.68	0.7846	1.138	vw	440	α
48.13	0.7446	1.086	w	531	β
46.23	0.7221	1.236	m	333	α
				511	
42.90	0.6807	1.311	s	422	α
37.38	0.6070	1.471	m	331	α
33.93	0.5581	1.599	w	400	α
27.63	0.4637	1.925	s	311	α
25.08	0.4238	1.918	vw	311	β
23.40	0.3971	2.248	vs	220	α
21.18	0.3612	2.239	vw	220	β
14.28	0.2464	3.623	vw	111	α

\* Cobalt radiation with iron filter. s, strong; m, medium; w, weak; vw, very weak.