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Summarv

The isomerisation of 5-hydroxy-benz(cd)indoline (I) to 5-keto-1,3,4,5-tetrahydro-benz(cd)indole (II) over a palladium catalyst has been studied systematically. Besides (II) varying amounts of the dehydrogenation product (III) are formed. This side reaction does not occur when the isomerisation is carried out in a hydrogen atmosphere.

The above isomerisation can be reversed by treatment of (II) with acetanhydride and potassium acetate.

A Synthesis of Kostanecki's 3:8:9-Trimethoxy-β-brazan

By heating trimethylbrazilone with hydriodic acid for a long period, Kostanecki and Lloyd obtained 3:8:9-trihydroxy- β -brazan which gave the trimethyl ether (I, m. p. 244–246°C) on methylation. The constitution of (I) was confirmed by converting the above-mentioned hydroxybrazan to β -brazan with the aid of zinc dust. The compound (I) has now been synthesised by an unambiguous method.

$$CH_3O = 3$$
 $O = 1$
 $O = 1$

$$\begin{array}{c} \text{CH}_3\text{O} & \text{OCH}_2 \\ \text{OC} & \text{OCH}_3 \\ \text{CN} \\ \text{(II)} \end{array}$$

$$CH_3O$$
 CH_3
 CH_2
 OCH_3
 OCH_3

The keto-nitrile (II) of Pfeiffer et al.² gave, on treatment with fuming hydrochloric acid in acetic acid

at 30°C, the isomeric amide (III, m. p. 162°C). This on prolonged boiling with 10% hydrochloric acid afforded 6-methoxy-3-(3:4-dimethoxybenzyl)-coumarone (IV, R = H, m. p. $61-62^{\circ}C$; picrate, m. p. $102^{\circ}C$). On Gattermann synthesis with the aid of zinc chloride IV (R = H) gave 6-methoxy-3-(3:4-dimethoxybenzyl)-2formylcoumarone (IV, R = CHO, m. p. 137°C; 2:4dinitrophenylhydrazone, m.p. 246-248°C), the orientation of which was established by oxidising it with potassium permanganate in acetone to the known acid1 (IV, $R = CO_2H$, m. p. 196°C). The aldehyde (IV, R = CHO) underwent quantitative cyclodehydration2 to I (m. p. 244-246°C) with glacial phosphoric acid at 100°C. The correctness of the synthesis was confirmed by typical colour reactions3 and oxidising (I) with chromic acid to trimethoxy-β-brazanquinone (m. p. 261-262°C) which was purified by sublimation in vacuum.

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Résumé

Une manière non équivoque de synthèse de 3,8,9-triméthoxy- β -brazan de Kostanecki se décrit dans cet article.

- ¹ K. W. Bentley and R. Robinson, J. Chem. Soc. 1950, 1355.
- ² C. K. Bradsher, Chem. Rev. 38, 492 (1946).
- ³ V. Kostanecki and L. Lloyd, Ber. Dtsch. chem. Gcs. 36, 2198 (1903).
- ⁴ V. Kostanecki and L. Lloyd, Ber. Disch. chem. Ges. 36, 2200 (1903).

7-Oxo-dehydro-iso-androsterone Acetate

During the purification of technical dehydro-iso-androsterone acetate by crystallization from methanol, the final mother-liquor showed no tendency to crystallization. The residue was dissolved in benzene and this solution subjected to chromatography over $\mathrm{Al_3O_3}$. Elution also took place with benzene.

The first fractions contained crystals which could not be identified. From the intermediate fractions crystals have been isolated with a melting range from $160-170^\circ$. By recrystallization first from methanol and subsequently from 40% aqueous methanol, the m. p. could be raised to $176-178^\circ$. The analysis yielded the following results: C: $73\cdot21$ and $73\cdot15\%$, H: $8\cdot40$ and $8\cdot33\%$. The values, calculated for an oxo-dehydro-iso-androsterone acetate are: C: $73\cdot12\%$, H: $8\cdot19\%$. A mixed m. p. with 7-oxo-dehydro-iso-androsterone acetate prepared according to BILLETER and MIESCHER¹ gave no depression. The U. V. absorption spectrum showed a maximum at $235\ \text{m}\mu$ with a log ε of $4\cdot09$. A similar maximum is observed in the spectrum of 7-oxo-cholesterol acetate.

The isolation of 7-oxo-dehydro-iso-androsterone acetate from the oxidation products of cholesterol has not yet been described. BILLETER and MIESCHER² isolated Δ -3,5-androstadienedione-7,17 from the mother-liquors of the dehydro-iso-androsterone semicarbazone and it is not unlikely that this compound has been formed from 7-oxo-dehydro-iso-androsterone acetate during the purification process.

¹ J. R. BILLETER and K. MIESCHER, Helv. chim. acta 31, 629 (1948).

 $^{^{\}rm 1}$ V. Kostanecki and L. Lloyd, Ber. Dtsch. chem. Ges. 36, 2198 (1903).

² P. Pfeiffer, K. Quehl, and F. Tappermann, Ber. Dtsch. chem. Ges. 63, 1304 (1930).