

## THE TECHNOLOGY OF THE ISOLATION OF THE ALKALOIDS DUBINIDINE AND HAPLOPHYLLIDINE

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Khimiya Prirodnikh Soedinenii, Vol. 2, No. 2, pp. 143-144, 1966

The dihydrofuranoquinoline alkaloid dubinidine has been isolated from Haplophyllum foliosum and Haplophyllum dubium [1]. It possesses valuable pharmacological properties [2] and is in the stage of introduction into medical practice.

We have carried out the isolation and separation of the alkaloids from the epigeal part of H. foliosum under semiindustrial conditions. Various organic solvents have been tested, as well as aqueous and acid extractions using a number of cation-exchangers.

Of organic solvents, the best yield of total alkaloids was given by chloroform (1.1-1.2% of the dry weight of the raw material).

Extraction with water formed very dilute diffusion liquors, but 0.5-0.7% hydrochloric acid gave unstable results.

The alkaloids were absorbed completely from the acid extract by KU-1 cation-exchanger, and desorption was carried out with a 1.5% solution of ammonia in 85% ethyl alcohol with subsequent elimination of the solvent by distillation in a vacuum apparatus. The yield of total alkaloids was 1.1%; from the mixture dubinidine was obtained in the form of the hydrochloride (yield 0.11% of the weight of the plant). We have treated more than 1 ton of raw material by this method.

The furanoquinoline alkaloid haplophyllidine has been isolated from the seeds of H. perforatum [3]. It also possesses valuable pharmacological properties and is in the stage of introduction into medical practice [4].

We have developed a method of obtaining haplophyllidine by the continuous extraction of the seeds with extraction gasoline. The alkaloids were extracted from the gasoline solution with 10% sulfuric acid and technical haplophyllidine was precipitated with 25% ammonia. The yield of haplophyllidine was 0.22-0.24% of the weight of the seeds.

An industrial process of obtaining dubinidine and haplophyllidine has been developed on the basis of the investigation carried out.

The technology of isolating the other alkaloids is being developed at the present time.

### REFERENCES

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25 October 1965

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## THE POSSIBILITY OF PHENOL-DIENONE REARRANGEMENTS IN LIGNIN

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Khimiya Prirodnikh Soedinenii, Vol. 2, No. 2, pp. 144-145, 1966

When the thioglignin isolated from spent chips from the sulfate digestion of wood of coniferous species [1] is heated for 1 hr in 0.1 N caustic soda in an inert gas atmosphere, a marked reduction in the content of weakly acidic groups with pK values of 13-14 takes place. The same effect is observed (table) with a specially prepared fraction of thioglignin with a high content of these groups and also with lignins isolated under mild conditions by extracting spruce sawdust with dioxane [2] and digesting the latter in dimethyl sulfoxide [3].